

Air Quality Criteria for Ozone and Related Photochemical Oxidants

Volume I of III

Air Quality Criteria for Ozone and Related Photochemical Oxidants

Volume I

National Center for Environmental Assessment-RTP Office
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PREFACE

National Ambient Air Quality Standards (NAAQS) are promulgated by the United States Environmental Protection Agency (EPA) to meet requirements set forth in Sections 108 and 109 of the U.S. Clean Air Act (CAA). Sections 108 and 109 require the EPA Administrator (1) to list widespread air pollutants that reasonably may be expected to endanger public health or welfare; (2) to issue air quality criteria for them that assess the latest available scientific information on nature and effects of ambient exposure to them; (3) to set “primary” NAAQS to protect human health with adequate margin of safety and to set “secondary” NAAQS to protect against welfare effects (e.g., effects on vegetation, ecosystems, visibility, climate, manmade materials, etc); and (5) to periodically review and revise, as appropriate, the criteria and NAAQS for a given listed pollutant or class of pollutants.

In 1971, the U.S. Environmental Protection Agency (EPA) promulgated National Ambient Air Quality Standards (NAAQS) to protect the public health and welfare from adverse effects of photochemical oxidants. The EPA promulgates the NAAQS on the basis of scientific information contained in air quality criteria issued under Section 108 of the Clean Air Act. Following the review of criteria as contained in the EPA document, Air Quality Criteria for Ozone and Other Photochemical Oxidants published in 1978, the chemical designation of the standards was changed from photochemical oxidants to ozone (O₃) in 1979 and a 1-hour O₃ NAAQS was set. The 1978 document focused mainly on the air quality criteria for O₃ and, to a lesser extent, on those for other photochemical oxidants (e.g., hydrogen peroxide and the peroxyacyl nitrates), as have subsequent revised versions of the document.

To meet Clean Air Act requirements noted above for periodic review of criteria and NAAQS, the O₃ criteria document, *Air Quality Criteria for Ozone and Other Photochemical Oxidants*, was next revised and released in August 1986; and a supplement, *Summary of Selected*

New Information on Effects of Ozone on Health and Vegetation, was issued in January 1992. These documents were the basis for a March 1993 decision by EPA that revision of the existing 1-h NAAQS for O₃ was not appropriate at that time. That decision, however, did not take into account newer scientific data that had become available after completion of the 1986 criteria document. Such literature was assessed in the next periodic revision of the O₃ air quality criteria document (O₃ AQCD) which has completed in 1996 and provided scientific bases supporting the setting by EPA in 1997 of the current 8-h O₃ NAAQS.

The purpose of this revised air quality criteria document for O₃ and related photochemical oxidants is to critically evaluate and assess the latest scientific information published since that assessed in the above 1996 O₃ AQCD, with the main focus being on pertinent new information useful in evaluating health and environmental effects data associated with ambient air O₃ exposures. However, other scientific data are also discussed in order to provide a better understanding of the nature, sources, distribution, measurement, and concentrations of O₃ and related photochemical oxidants and their precursors in the environment. The document mainly assesses pertinent literature published through 2004, but also includes assessment of a few additional important studies published or accepted for publication in 2005.

A First External Review Draft of this O₃ AQCD (dated January 2005) was released for public comment and was reviewed by the Clean Air Scientific Advisory Committee (CASAC) in May, 2005 to obtain. Public comments and CASAC recommendations were then taken into account in making revisions to the document for incorporation into a Second External Review Draft (dated August, 2005), which underwent further public comment and CASAC review at a December, 2005 public meeting. Public comments and CASAC advice derived from review of that Second External Review Draft were considered in making revisions incorporated into this final version of the document (dated February, 2006). Evaluations contained in the present document will be drawn on to provide inputs to associated O₃ Staff Paper analyses prepared by EPA's Office of Air Quality Planning and Standards (OAQPS) to pose options for consideration by the EPA Administrator with regard to proposal and, ultimately, promulgation of decisions on potential retention or revision, as appropriate, of the current O₃ NAAQS.

Preparation of this document was coordinated by staff of EPA's National Center for Environmental Assessment in Research Triangle Park (NCEA-RTP). NCEA-RTP scientific staff, together with experts from other EPA/ORD laboratories and academia, contributed to

writing of document chapters. Earlier drafts of document materials were reviewed by non-EPA experts in peer consultation workshops held by EPA. The document describes the nature, sources, distribution, measurement, and concentrations of O₃ in outdoor (ambient) and indoor environments. It also evaluates the latest data on human exposures to ambient O₃ and consequent health effects in exposed human populations, to support decision making regarding the primary, health-related O₃ NAAQS. Lastly, the document also evaluates ambient O₃ environmental effects on vegetation and ecosystems, surface level solar UV radiation flux and global climate change, and man-made materials to support decision making on secondary O₃ NAAQS.

NCEA acknowledges the valuable contributions provided by authors, contributors, and reviewers and the diligence of its staff and contractors in the preparation of this document.

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ABBREVIATIONS AND ACRONYMS

α	alpha; probability value
AA	ascorbic acid
ACh	acetylcholine
ADSS	aged and diluted cigarette smoke
AER	air exchange rate
AEROCE	Atmospheric/Ocean Chemistry Experiment
AHR	airway hyperreactivity
AHSMOG	Adventist Health Study on Smog
AIRPEX	Air Pollution Exposure (model)
AIRQUIS	Air Quality Information System (model)
AIRS	Aerometric Information Retrieval System
AM	alveolar macrophage
ANF	atrial natriuretic factor
AOP2	antioxidant protein 2
AOT40	seasonal sum of the difference between an hourly concentration at the threshold value of 40 ppb, minus the threshold value of 40 ppb
APEX	Air Pollution Exposure (model)
APHEA	Air Pollution on Health: European Approach (study)
AQCD	Air Quality Criteria Document
AQS	Air Quality System
ARIC	Atherosclerosis Risk in Communities (study)
ATS	American Thoracic Society
A/V	surface-to-volume ratio
β	beta-coefficient; slope of an equation
BAL	bronchioalveolar lavage
BALF	bronchioalveolar lavage fluid
BC	black carbon
BLD	below limit of detection

BS	black smoke
BSA	body surface area
BSA	bovine serum albumin
BMZ	basement membrane zone
BP	blood pressure
C	concentration
C3a	complement protein fragment
CAA	Clean Air Act
CADS	Cincinnati Activity Diary Study
CAPs	concentrated ambient particles
CAR	centriacinar region
CASAC	Clean Air Scientific Advisory Committee
CASTNet, CASTNET	Clean Air Status and Trends Network
CC16	Clara cell secretory protein
CCSP	Clara cell secretory protein
C_{dyn} , Cdyn	dynamic lung compliance
CDT	Central Daylight Time
CE	continuous exercise
CFCs	chlorofluorocarbons
CFD	computational fluid dynamics
CFR	Code of Federal Regulations
CH ₄	methane
C ₂ H ₅ -H	ethane
C ₅ H ₈	isoprene
C ₁₀ H ₁₆	terpene
CHAD	Consolidated Human Activities Database
CH ₃ -CCl ₃	methyl chloroform
CH ₃ -CHO	acetaldehyde
CH ₃ -CO	acetyl
CHO	Chinese hamster ovary (cells)

CH ₃ OOH	acetic acid
CI	confidence interval
CIE	Commission Internationale de l'Eclairage (International Commission on Illumination)
CINC	cytokine-induced neutrophil chemoattractant
CLM	chemiluminescence method
CMAQ	Community Model for Air Quality
CO	carbon monoxide
CO ₂	carbon dioxide
COD	coefficient of divergence
COP	Conference of Parties
COPD	chronic obstructive pulmonary disease
CRP	C-reactive protein
CTM	chemistry transport model
Δ	delta; change in a variable
3-D	three-dimensional
DHBA	2,3-dihydroxybenzoic acid
DNA	deoxyribonucleic acid
DOAS	differential optical absorption spectroscopy
DPPC	dipalmitoylglycero-3-phosphocholine
DTPA	diethylenetriaminepentaacetic acid
DU	Dobson units
ε	epsilon; convergence precision
ECG	electrocardiographic; electrocardiogram
EDU	ethylenediurea
EEG	electroencephalographic
ELF	epithelial lining fluid
ENA-78	epithelial cell-derived neutrophil-activating peptide 78
ENSO	El Niño-Southern Oscillation
EPA	U.S. Environmental Protection Agency
EST	Eastern Standard Time

ETS	environmental tobacco smoke
EVR	equivalent ventilation rate
F	female
F344	Fisher 344 (rat)
FA	filtered air
FACE	free-air carbon dioxide exposure
f_b	breathing frequency
FEF	forced expiratory flow
FEF ₂₅₋₇₅	forced expiratory flow between 25 and 75% of vital capacity; forced expiratory flow at 25 to 75% of vital capacity
FEF _x	forced expiratory flow after X% vital capacity (e.g., after 25, 50, or 75% vital capacity)
FEV ₁	forced expiratory volume in 1 second
FGF	fibroblast growth factor
FGFR	fibroblast growth factor receptor
FN	fibronectin
FR	Federal Register
FRM	Federal Reference Method
FVC	forced vital capacity
GAM	Generalized Additive Model
GCM	general circulation model
GEE	Generalized Estimating Equation
GEOS-CHEM	three-dimensional model of atmospheric composition driven by assimilated Goddard Earth Orbiting System observations
GHG	greenhouse gas
GLM	Generalized Linear Model
GLRAG	Great Lakes Regional Assessment Group
GM-CSF	granulocyte-macrophage colony stimulating factor
G6PD	glucose-6-phosphate dehydrogenase
GPx	glutathione peroxidase
GR	glutathione reductase

GSH	glutathione; reduced glutathione
GSHPx	glutathione peroxidase
GSTM1	glutathione S-transferase μ -1 (genotype)
GSTM1null	glutathione S-transferase μ -1 null (genotype)
H ⁺	hydrogen ion
HCFCs	hydrochlorofluorocarbons
H ₂ CO, HCHO	formaldehyde
HDMA	house dust mite allergen
HF	hydrofluoride
HFCs	hydrofluorocarbons
HLA	human leukocyte antigen
HNE	4-hydroxynonenal
HNO ₂ , HONO	nitrous acid
HNO ₃	nitric acid
HNO ₄	pernitric acid
HO ₂	hydroperoxyl; hydroperoxy
H ₂ O ₂	hydrogen peroxide
HO _x	hydrogen oxides
HR	heart rate
HRP	horseradish peroxidase
HRV	heart rate variability
H ₂ SO ₄	sulfuric acid
IARC	International Agency for Research on Cancer
IC	inspiratory capacity
ICAM	intracellular adhesion molecule
ICNIRP	International Commission on Non-Ionizing Radiation Protection
IE	intermittent exercise
IFN	interferon
Ig	immunoglobulin (e.g., IgA, IgE, IgG, IgM)
IL	interleukin (e.g., IL-1, IL-6, IL-8)

iNOS	inducible nitric oxide synthase; NOS-2
i.p.	intraperitoneal
IPCC	Intergovernmental Panel on Climate Change
IQR	interquartile range
IR	infrared
K_a	intrinsic mass transfer coefficient/parameter
K_g	mass transfer coefficient for gas phase
K_l	mass transfer coefficient for liquid phase
K_r	reaction rate constant
LDH	lactic acid dehydrogenase
LIDAR	LIght Detection And Ranging
LIS	lateral intercellular space
LLJ	low-level jet
LOEL	lowest-observed-effect level
LOESS	locally estimated smoothing splines
LOP	lipid ozonation products
LPS	lipopolysaccharide
LRT	lower respiratory tract; lower airways
LT	leukotriene (e.g., LTB ₄ , LTC ₄ , LTD ₄ , LTE ₄)
LT	local time
LT α	lymphotoxin- α
M	male
M	maximum number of iterations
MAP	mean arterial pressure
MARAT	Mid-Atlantic Regional Assessment Team
MCh	methacholine
MCP	monocyte chemotactic protein
MED	minimal erythema dose
MENTOR	Modeling Environment for Total Risk Studies
MET	metabolic equivalent

MI	myocardial infarction
MIP	macrophage inflammatory protein
MMEF	maximal midexpiratory flow
MONICA	Monitoring Trend and Determinants in Cardiovascular Disease (registry)
MPAN	peroxymethacryloyl nitrate; peroxy-methacrylic nitric anhydride
MPO	myeloperoxidase
mRNA	messenger ribonucleic acid
MSA	metropolitan statistical area
MSL	mean sea level
MT	metallothionein
n, N	number
NAAQS	National Ambient Air Quality Standards
NAD(P)H	reduced nicotinamide adenine dinucleotide phosphate
NAMS/SLAMS	National Ambient Monitoring Stations and State and Local Air Monitoring Stations
NARSTO	North American Regional Strategy for Atmospheric Ozone
NAS	Normative Aging Study
NCEA-RTP	National Center for Environmental Assessment Division in Research Triangle Park, NC
NCICAS	National Cooperative Inner-City Asthma Study
NCLAN	National Crop Loss Assessment Network
ND	not detectable; not detected
NEM	National Ambient Air Quality Standards Exposure Model
NERAG	New England Regional Assessment Group
NF- κ B	nuclear factor kappa B
NH ₃	ammonia
NHAPS	National Human Activity Pattern Survey
NH ₄ HSO ₄	ammonium bisulfate
(NH ₄) ₂ HSO ₄	ammonium sulfate
NIH	National Institutes of Health

NIST	National Institute of Standards and Technology
NK	natural killer (cells)
NL	nasal lavage
NM	national monument
NMHCs	nonmethane hydrocarbons
NMMAPS	National Morbidity, Mortality and Air Pollution Study
NO	nitric oxide
NO ₂	nitrogen dioxide
N ₂ O	nitrous oxide
NO ₃ ⁻	nitrate
N ₂ O ₅	nitrogen dioxide
NOS	nitric oxide synthase
NOS-1	neuronal nitric oxide synthase
NOS-2	inducible nitric oxide synthase; iNOS
NOS-3	endothelial nitric oxide synthase
NO _x	nitrogen oxides
NO _y	reactive nitrogen system components; sum of NO _x and NO _z ; odd nitrogen species
NO _z	difference between NO _y and NO _x ; reservoir and termination nitrogen species
NP	national park
NPP	net primary productivity
NQO1	NAD(P)H-quinone oxidoreductase (genotype)
NQO1wt	NAD(P)H-quinone oxidoreductase wild type (genotype)
NRC	National Research Council
NS	national seashore
NTP	National Toxicology Program
NTS	nucleus tractus solitarius
NWR	national wildlife refuge
O ₂	ground-state oxygen
O ₃	ozone

O ₃ *	electronically excited ozone
¹⁸ O ₃	radiolabeled ozone
O(¹ D)	electronically excited oxygen atom
OAQPS	Office of Air Quality Planning and Standards
OH	hydroxyl; hydroxy
8-OHdG	8-hydroxy-2'-deoxyguanosine
O(³ P)	ground-state oxygen atom
OPE	ozone production efficiency
OTC	open-top chamber
OVA	ovalbumin
OxComp	oxidative capacity of the atmosphere
p	probability value
P ₉₀	90th percentile
PAF	platelet-activating factor
PAN	peroxyacetyl nitrate; peroxyacetic nitric anhydride
P _a O ₂	partial pressure of arterial oxygen
PAR	proximal alveolar region
PBL	planetary boundary layer
PBPK	physiologically based pharmacokinetic (approach)
PCI	picryl chloride
PE	postexposure
PEF	peak expiratory flow
PEFR	peak expiratory flow rate
PEM	personal exposure monitor
P _{enh}	enhanced pause
PG	prostaglandin (e.g., PGD ₂ , PGE, PGE ₁ , PGE ₂ , PGF _{1α} , PGF _{2α})
6PGD	6-phosphogluconate dehydrogenase
PGP	protein gene product (e.g., PGP9.5)
PI	probability interval
PM	particulate matter

PM _{2.5}	fine particulate matter (mass median aerodynamic diameter $\leq 2.5 \mu\text{m}$)
PM ₁₀	combination of coarse and fine particulate matter
PM _{10-2.5}	coarse particulate matter (mass median aerodynamic diameter between 10 and 2.5 μm)
PMNs	polymorphonuclear leukocytes; neutrophils
pNEM	Probabilistic National Ambient Air Quality Standard Exposure Model
polyADPR	poly(adenosinediphosphate-ribose)
ppb	parts per billion
ppbv	parts per billion by volume
ppm	parts per million
PPN	peroxypropionyl nitrate; peroxypropionic nitric anhydride
PRB	policy relevant background
PSA	picryl sulfonic acid
PSC	polar stratospheric clouds
PUFA	polyunsaturated fatty acid
PWM	pokeweed mitogen
QCE	quasi continuous exercise
R	intraclass correlation coefficient
r	correlation coefficient
R ²	multiple correlation coefficient
Raw	airway resistance
R'CO	acyl
R'C(O)–O ₂	acyl peroxy
RH	relative humidity
R _L	total pulmonary resistance
RNA	ribonucleic acid
RO ₂	organic peroxy; organic peroxy
ROOH	organic peroxides
ROS	reactive oxygen species
RR	ribonucleotide reductase
RRMS	relatively remote monitoring sites

RT	respiratory tract
SAB	Science Advisory Board
SAC	<i>Staphylococcus aureus</i> Cowan 1 strain
SAMD	S-adenosyl methionine decarboxylase
S _a O ₂	oxygen saturation of arterial blood
SBUV	solar backscattered ultraviolet radiation
SC	stratum corneum
SD	Sprague-Dawley (rat)
SD	standard deviation
SES	socioeconomic status
sGAW	specific airways conductance
SHEDS	Simulation of Human Exposure and Dose System
sICAM	soluble intracellular adhesion molecule
SNAAQs	Secondary National Ambient Air Quality Standards
SNPs	single nucleotide polymorphisms
SO ₂	sulfur dioxide
SO ₄ ²⁻	sulfate
SOA	secondary organic aerosol
SOD	superoxide dismutase
SOS	Southern Oxidant Study
SP	substance P
SP	surfactant protein (e.g., SP-A, SP-D)
SPF	specific-pathogen free
sRAW, SR _{aw}	specific airways resistance
SRES	Special Report on Emissions Scenarios
STE	stratospheric-tropospheric exchange
STRF	Spatio-Temporal Random Field
SUM06	seasonal sum of all hourly average concentrations ≥0.06 ppm
SUM08	seasonal sum of all hourly average concentrations ≥0.08 ppm
SZA	solar zenith angle

T	time; duration of exposure
t	<i>t</i> -test statistical value; <i>t</i> statistic
T ₃	triiodothyronine
T ₄	thyroxine
TAR	Third Assessment Report
TAR WGI	Third Assessment Report of Working Group 1
TB	terminal bronchioles
TBARS	thiobarbituric acid reactive substances
^{99m} Tc-DTPA	radiolabeled diethylenetriaminepentaacetic acid
T _{CO}	core temperature
T _{CTL}	cytotoxic T-lymphocytes
TiO ₂	titanium dioxide
TLC	total lung capacity
TLR	Toll-like receptor
TNF	tumor necrosis factor
TNFR	tumor necrosis factor receptor
TOMS	Total Ozone Mapping/Monitoring Satellite; total ozone mapping spectrometer
TRIM	Total Risk Integrated Methodology (model)
TRIM.Expo	Total Risk Integrated Methodology Exposure Event (model)
TSP	total suspended particulate
TWA	time-weighted average
UA	uric acid
UNEP	United Nations Environmental Program
UNFCCC	United Nations Framework Convention on Climate Change
URT	upper respiratory tract; upper airways
USGCRP	U.S. Global Change Research Program
UV	ultraviolet
UV-A	ultraviolet radiation of wavelengths 320 to 400 nm
UV-B	ultraviolet radiation of wavelengths 280 to 320 nm
UV-C	ultraviolet radiation of wavelengths 200 to 280 nm

VC	vital capacity
V_D	volume of the anatomic dead space
\dot{V}_E	minute ventilation; expired volume per minute
$\dot{V}O_{2max}$	maximal oxygen uptake (maximal aerobic capacity)
VOC	volatile organic compound
\dot{V}_P	volumetric penetration
$\dot{V}_{P50\%}$	volume at which 50% of an inhaled bolus is absorbed
V_T	tidal volume
V_{TB}	terminal bronchiole region volume
V_{UA}	volume of the upper airways
W126	cumulative integrated exposure index with a sigmoidal weighting function
WMO	World Meteorological Organization
WT	wild type

EXECUTIVE SUMMARY

E.1 INTRODUCTION

Tropospheric or “surface-level” ozone (O₃) is one of six major air pollutants regulated by National Ambient Air Quality Standards (NAAQS) under the U.S. Clean Air Act. As mandated by the Clean Air Act, the U.S. Environmental Protection Agency (EPA) must periodically review the scientific bases (or “criteria”) for the various NAAQS by assessing newly available scientific information on a given criteria air pollutant. This document, *Air Quality Criteria for Ozone and Other Photochemical Oxidants*, is an updated revision of the 1996 Ozone Air Quality Criteria Document (O₃ AQCD) that provided scientific bases for the current O₃ NAAQS set in 1997.

E.1.1 Clean Air Act Legal Requirements

Clean Air Act (CAA) Sections 108 and 109 govern establishment, review, and revision of U.S. National Ambient Air Quality Standards (NAAQS).

- Section 108 directs the U.S. Environmental Protection Agency (EPA) Administrator to list ubiquitous (widespread) air pollutants that may reasonably be anticipated to endanger public health or welfare and to issue air quality criteria for them. The air quality criteria are to reflect the latest scientific information useful in indicating the kind and extent of all exposure-related effects on public health and welfare expected from the presence of the pollutant in the ambient air.
- Section 109 directs the EPA Administrator to set and periodically revise, as appropriate, two types of NAAQS: (a) *primary NAAQS* to protect against adverse health effects of listed criteria pollutants among sensitive population groups, with an adequate margin of safety, and (b) *secondary NAAQS* to protect against welfare effects (e.g., impacts on vegetation, crops, ecosystems, visibility, climate, man-made materials, etc.). Section 109 also requires peer review of the NAAQS and their underlying scientific bases by the Clean Air Scientific Advisory Committee (CASAC), a committee of independent non-EPA experts.

E.1.2 Chronology of Ozone NAAQS Revisions

In 1971, the U.S. EPA set primary and secondary standards for total photochemical oxidants. Based on a criteria review completed in 1978, the original NAAQS set in 1971 were revised in 1979 to focus on O₃ as the indicator for new primary and secondary standards that would be attained when the expected number of days per calendar year with maximum 1-h average O₃ concentrations >0.12 ppm did not exceed one. The NAAQS for ambient O₃ were revised in 1997 by replacing the 1-h standards with an 8-h primary standard that is met when the 3-year average of the annual fourth highest daily maximum 8-h average concentration is ≤0.08 ppm. The 1997 primary NAAQS was based on scientific data from controlled human exposure, laboratory animal, and epidemiological studies and associated analyses presented in the 1996 O₃ AQCD and in the 1996 O₃ Staff Paper.

- This revised O₃ AQCD, prepared by EPA's National Center for Environmental Assessment (NCEA), provides scientific bases to support the periodic review of O₃ NAAQS. This document assesses the latest available scientific information (published mainly through December 2004) judged to be useful in deriving criteria as scientific bases for decisions on possible revision of the current O₃ NAAQS.
- A separate EPA O₃ Staff Paper, prepared by EPA's Office of Air Quality Planning and Standards (OAQPS), will draw upon key findings/conclusions from this document, together with other analyses, to develop and present options for consideration by the EPA Administrator regarding review and possible revision of the O₃ NAAQS.

E.1.3 Document Organization and Structure

Volume I of this document consists of the present Executive Summary and eleven main chapters of this revised O₃ AQCD. Those main chapters focus primarily on interpretative evaluation of key information, whereas more detailed descriptive summarization of pertinent studies and/or supporting analyses are provided in accompanying annexes. Volume II contains the annexes for Chapters 4 through 7, whereas Volume III contains the annex for Chapter 9.

Topics covered in the main chapters of the present AQCD are as follows:

- This Executive Summary summarizes key findings and conclusions from Chapters 1 through 11 of this revised O₃ AQCD, as they pertain to background information on O₃-related atmospheric science and air quality, human exposure aspects, dosimetric considerations, health effect issues, and environmental effect issues.
- Chapter 1 provides a general introduction, including an overview of legal requirements, the chronology of past revisions of O₃-related NAAQS, and orientation to the structure of this document.
- Chapters 2 and 3 provide background information on atmospheric chemistry/physics of O₃ formation, air quality, and exposure aspects to help to place ensuing discussions of O₃ health and welfare effects into perspective.
- Chapters 4 through 7 then assess dosimetry aspects, experimental (controlled human exposure and laboratory animal) studies, and epidemiologic (field/panel; other observational) studies. Chapter 8 then provides an integrative synthesis of key findings and conclusions derived from the preceding chapters with regard to ambient O₃ concentrations, human exposures, dosimetry, and health effects.
- Chapter 9 deals with effects of O₃ on vegetation, crops, and natural ecosystems, whereas Chapter 10 evaluates tropospheric O₃ relationships to alterations in surface-level UVB flux and climate change and Chapter 11 assesses materials damage (these all being key types of welfare effects of relevance to decisions regarding secondary O₃ NAAQS review).

E.2 ATMOSPHERIC CHEMISTRY AND PHYSICS OF TROPOSPHERIC OZONE FORMATION

Key findings/conclusions from Chapter 2 regarding the chemistry and physics of surface-level O₃ formation include the following:

- Ozone (O_3) is a secondary pollutant formed by atmospheric reactions involving two classes of precursor compounds, volatile organic compounds (VOCs) and nitrogen oxides (NO_x). Carbon monoxide also contributes to O_3 formation.
- The formation of O_3 and associated compounds is a complex, nonlinear function of many factors, including the intensity and spectral distribution of sunlight; atmospheric mixing and other atmospheric processes; and the concentrations of precursors in ambient air.
- The photochemical oxidation of almost all anthropogenic and biogenic VOCs is initiated by reaction with hydroxyl (OH) radicals. At night, when they are most abundant, NO_3 radicals oxidize alkenes. In coastal and other select environments, Cl and Br radicals can also initiate the oxidation of VOCs.
- In urban areas, basically all classes of VOCs (alkanes, alkenes, aromatic hydrocarbons, carbonyl compounds, etc.) and CO are important for ozone formation. Although knowledge of the oxidative mechanisms of VOCs has improved in recent years, gaps in knowledge involving key classes, such as aromatic hydrocarbons, still remain. For example, only about half of the carbon initially present in aromatic hydrocarbons in smog chamber studies form compounds that have been identified.
- In addition to gas phase reactions, other reactions also occur on the surfaces of or within cloud droplets and airborne particles. Most of the well-established multiphase reactions tend to reduce the rate of O_3 formation in polluted environments. Direct reactions of O_3 and atmospheric particles appear to be too slow to reduce O_3 formation significantly at typical ambient PM levels.
- Oxidants other than O_3 are found in the gas phase and in particles. The chemistry occurring in particle bound-water and, hence, the mechanisms leading to the formation of reactive oxygen species in particles are largely unknown.

- Organic hydroperoxides produced in the oxidation of monoterpenes by O₃ could contribute substantially to secondary organic aerosol formation.
- Our basic understanding of meteorological processes associated with summertime O₃ episodes has not changed over recent years. However, the realization is growing that long-range transport processes are important for determining O₃ concentrations at the surface. In addition to synoptic scale flow fields, nocturnal low-level jets can transport pollutants hundreds of km from their sources in either the upper boundary layer or the lower free troposphere. Turbulence then brings O₃ and other pollutants to the surface.
- Even in the absence of photochemical reactions in the troposphere, some O₃ would be found near the earth's surface due to its downward transport from the stratosphere. Intrusions of stratospheric O₃ that reach the surface are rare. Much more common are intrusions that penetrate to the middle and upper troposphere. However, O₃ transported to the middle and upper troposphere can still affect surface concentrations through various mechanisms that mix air between the planetary boundary layer and the free troposphere above.
- Associations between daily maximum O₃ concentration and temperature vary across the United States and depend on location. In some areas (e.g., Baltimore, MD and surrounding areas), there is a strong positive association. In other areas (e.g., Phoenix, AZ), there is little association.
- Chemistry transport models are used to improve understanding of atmospheric chemical and physical processes, as well as to develop air pollution control strategies. Model evaluation does not merely involve a straightforward comparison between model predictions and observed concentration fields of a pollutant of interest (e.g., O₃). Such comparisons may not be meaningful because it is difficult to determine if agreement between measurements and model predictions truly represents an accurate treatment of physical and chemical processes in the model or the effects of compensating errors in model routines.

- The main methods currently used for routine monitoring of ambient ozone are based on chemiluminescence or UV absorption. Measurements at most ambient monitoring sites are based on UV absorption. Both of these methods are subject to interference by other atmospheric components. Studies conducted in Mexico City and in a smog chamber have found positive interference, but studies conducted in urban plumes did not find evidence for significant positive interference in the UV absorption technique.

E.3 ENVIRONMENTAL DISPERSAL, AMBIENT CONCENTRATIONS, AND HUMAN EXPOSURE TO OZONE

Key findings/conclusions derived from Chapter 3 with regard to ambient O₃ concentrations and human exposure are as follows:

- Ozone is monitored in populated areas in the United States during “ozone seasons,” which vary in length depending on location. All monitors should be operational from May to September. However, in many areas, O₃ is monitored throughout the year.
- The median of the mean daily maximum 8-h average O₃ concentration from May to September 2000 to 2004 across the U.S. was 0.049 ppm on a countywide average basis. Ninety five per cent of countywide mean daily maximum 8-h average O₃ concentrations were less than 0.057 ppm for the same period. Because most monitors are located in the East, these values should not be taken to represent conditions across the country.
- The daily maximum 1-h O₃ concentrations tend to be much higher in large urban areas or in areas downwind of large urban areas. For example, daily maximum 1-h O₃ concentrations in Houston, TX approached 0.20 ppm during the same period.
- Daily maximum 8-h average O₃ concentrations are lower than, but are highly correlated with, 1-h daily maximum O₃ concentrations. For example, in the Baltimore, MD area, the correlation coefficient between the two quantities was 0.98 for data obtained from May to September 1994 to 2004.

- Within individual metropolitan statistical areas (MSAs), O₃ tends to be well correlated across monitoring sites. However, there can be substantial spatial variations in concentrations. Ozone in city centers tends to be lower than in regions either upwind or downwind of the center, because of titration by NO emitted by motor vehicles.
- Ozone concentrations tend to peak in early- to mid-afternoon in areas where there is strong photochemical production and later in the day in areas where transport is more important in determining O₃ abundance.
- Summertime maxima in O₃ concentrations occur in areas in the United States where there is substantial photochemical activity involving O₃ precursors emitted from human activities. Maxima can occur anytime from June through August.
- Springtime maxima are observed in relatively remote sites in the western United States and at various other relatively unpolluted sites throughout the Northern Hemisphere. Relatively high O₃ concentrations can also be found during winter in several cities throughout the southern United States.
- Long-term trends in O₃ concentrations reflect notable decreases over time throughout the United States, with decreases nationwide of approximately 29% in 2nd highest 1-h O₃ concentrations from 1980 to 2003 and of about 21% in 4th highest 8-h O₃ concentrations during the same time period.
- These trends include dramatic decreases from peak 1-h O₃ levels of 0.4 to 0.6 ppm seen in the Los Angeles area at times in the late 1950's to 1970's to current peak levels of 0.17 ppm and 0.15 ppm (1-h and 8-h avg, respectively) seen in the Los Angeles basin during 2000-2003.
- Downward trends in the upper tail of the O₃ concentration distribution do not necessarily reflect trends for O₃ values towards the center of the O₃ concentration distribution at national parks. Concentrations toward the center of the distribution have remained more or less

constant, and O₃ values in the lower tail of the distribution show some evidence of slight increases on a nationwide basis.

- Policy relevant background (PRB) O₃ concentrations are used for assessing risks to human health associated with O₃ produced from anthropogenic sources in the United States, Canada and Mexico. Because of the nature of the definition of PRB concentrations, they cannot be derived from observations directly, instead they must be derived from model estimates.
- Current model estimates indicate that PRB O₃ concentrations in the United States surface air are generally 0.015 ppm to 0.035 ppm. Such concentrations decline from spring to summer and are generally <0.025 ppm under conditions conducive to high O₃ episodes. PRB O₃ concentrations may be higher, especially at high altitude sites during the spring, due to enhanced contributions from (a) pollution sources inside and outside North America and (b) stratospheric O₃ exchange.
- Only one model (GEOS-Chem) is documented in the literature for calculating PRB O₃ concentrations. Estimated PRB O₃ values are likely 10 ppbv too high in the Southeast in summer and are accurate within 5 ppbv in other regions and seasons.
- Sufficient data for other oxidants (e.g., H₂O₂, PAN) and oxidation products (e.g., HNO₃, H₂SO₄) in the atmosphere are not available for use in epidemiologic time series studies. Limited data for oxidants besides O₃ in the gas and particle phases suggest that their combined concentrations are probably <10 % that of O₃.
- Relationships between O₃ and PM_{2.5} are complex, in part because PM is not a distinct chemical species, but is a mix of primary and secondary species. For example, PM_{2.5} concentrations were positively correlated with O₃ during summer, but negatively correlated with O₃ during the winter at Ft. Meade, MD. Similar relationships were found for PM₁₀ and O₃ in data collected in a number of urban areas during the 1980s.

- Humans are exposed to O₃ either outdoors or in various microenvironments. Ozone in indoor environments results mainly from infiltration from outdoors. Once indoors, O₃ is removed by deposition on and reaction with surfaces and reactions with other pollutants. Hence, O₃ levels indoors tend to be notably lower than outdoor O₃ concentrations measured at nearby monitoring sites, although the indoor and ambient O₃ concentrations tend to vary together (i.e., the higher the ambient, the higher the indoor O₃ levels).
- Personal exposure to O₃ tends to be positively associated with time spent outdoors. Although O₃ concentrations obtained at stationary monitoring sites may not explain the variance in individual personal exposures, they appear to serve reasonably well as surrogate measures for aggregate personal exposures.
- Atmospheric reactions between O₃ and certain other ambient airborne contaminants, e.g., terpenes emitted by vegetation or wood products, contribute to generation of ultrafine particles, with formation of such particles being observed in both urban and rural areas. These reactions also occur in indoor environments and involve O₃ infiltrating from outdoors and terpenes emitted by household products (e.g., air fresheners). Gaseous products resulting from such reactions may also be toxic.

E.4 OZONE DOSIMETRY AND HEALTH EFFECTS

This section summarizes the main conclusions derived from the integrated synthesis of information regarding health effects associated with ambient O₃ exposures. The conclusions are based on O₃ dosimetry evaluations and human clinical, animal toxicologic, and epidemiologic studies which have evaluated health effects associated with short-term, repeated, and long-term exposures to O₃ alone or in combination with other ambient pollutants. The controlled human exposure (or “clinical”) studies provide the clearest and most compelling evidence for human health effects directly attributable to acute exposures to O₃ per se. The evidence from human and animal toxicologic studies presented in Chapters 4, 5, and 6 are further useful in not only providing insights into possible mechanisms of action underlying different types of O₃-related health effects but, also, in helping to provide biological plausibility for health effects observed in

epidemiologic studies assessed in Chapter 7. The studies have also been useful in identifying susceptible and vulnerable populations that are at potentially greater risk for effects of O₃ exposure. Overall, the new findings generally support and build further upon key health-related conclusions drawn in the previous 1996 AQCD, as summarized below.

1. Dosimetric Considerations

Chapter 4 discusses dosimetric issues, including factors that are important to consider in attempting animal-to-human extrapolations of experimentally-induced O₃ effects.

- Dosimetric studies seek to quantify dose and factors affecting the dose of O₃ and/or its active metabolites at specific lung regions, target tissues, or cells. In both humans and animals, the efficiency of O₃ uptake is greater in the nasal passages than the oral pathway. In the lower respiratory tract, increasing tidal volume increases O₃ uptake, whereas increasing flow or breathing frequency decreases O₃ uptake. However, O₃-induced rapid shallow breathing appears to protect the large conducting airways while producing a more even distribution of injury to the terminal bronchioles.
- In adult human females relative to males, the smaller airways and associated larger surface-to-volume ratio enhance local O₃ uptake and cause somewhat reduced penetration of O₃ into the distal lung. However, it is not clear from these findings if the actual anatomical location of O₃ uptake differs between males and females.
- Similarly exposed individuals vary in the amount of actual dose received, but O₃ uptake is not predictive of intersubject variability in measures of pulmonary function.
- The efficiency of O₃ uptake is chemical-reaction rate dependent and the reaction products (hydrogen peroxide, aldehydes, and hydroxyhydroperoxides) created by ozonolysis of lipids in epithelial lining fluid (ELF) and cell membranes appear to mediate O₃ toxicity.
- Ozone uptake in humans is increased by exposure to NO₂ and SO₂ and decreased during the O₃ exposure. This suggests that an inflammatory response during exposure to NO₂ and

SO₂ may elicit increased production of O₃-reactive substrates in the epithelial lining fluid and that these substrates are depleted by O₃ exposure but not by NO₂ and SO₂ exposures.

- Prior modeling studies have suggested the proximal alveolar and centriacinar regions as principal target sites of acute O₃-induced cell injury. New experimental work in rats suggests that the conducting airways are also a primary site of injury.
- In most clinical studies, humans are exposed to O₃ during exercise. Under these conditions, the switch from nasal to oral breathing, coupled with increases in respiratory flow (as occurs during exercise), causes a shift in the O₃ dose distribution, allowing O₃ to penetrate deeper into the lung and thereby increasing the potential for bronchiolar and alveolar damage.
- Comparisons of acute exposures in rats and humans suggest that, though both species have similar qualitative responses to O₃ exposure, there are interspecies mechanistic disparities that necessitate careful comparisons of dose-response relationships. Currently available data suggest that lowest observable effect levels in resting rats are approximately 4- to 5-fold higher than for exercising humans for some toxicological endpoints, e.g., increases in bronchoalveolar lavage (BAL) protein or neutrophil (PMN) levels (indicators of O₃-induced lung inflammation responses).
- Thus, a number of variables seem to affect O₃ uptake, notably including route of breathing, breathing pattern, gender, copollutants, and certain pre-exposure conditions. These differences are important in order to interrelate experimentally-demonstrated pathophysiological effects and epidemiologically-observed associations between ambient O₃ concentrations and health risks among human population groups.

2. Health Effects of Short-term Exposures to Ozone

The 1996 O₃ AQCD assessed a substantial body of evidence from toxicologic, human clinical, and epidemiologic studies. That AQCD concluded that short-term ambient O₃ exposure resulted in various respiratory health effects, including lung function decrements and increased respiratory symptoms in both healthy and asthmatic individuals exposed during moderate to

heavy exercise to O₃ concentrations ranging down to the lowest levels (0.12 ppm for 1 h; 0.08 ppm for 6.6 to 8 h) tested in the available controlled human exposure studies. Such experimentally demonstrated effects were consistent with and lent plausibility to epidemiologic observations highlighted in the 1996 AQCD of increases in daily hospital admissions and ED visits for respiratory causes. Epidemiologic evidence also provided suggestive evidence for an association between short-term O₃ exposure and mortality. However, there was essentially no evidence available in the 1996 O₃ AQCD regarding potential cardiovascular effects of short-term O₃ exposure. The newly-available evidence assessed in this revised O₃ AQCD notably enhances our understanding of short-term O₃ exposure effects, as summarized below, first in relation to respiratory morbidity endpoints and then cardiovascular effects and, lastly, mortality.

A. Respiratory Morbidity

Lung Function:

- Controlled exposure studies clearly demonstrate acute reversible decrements in lung function in healthy adults exposed to ≥ 0.08 ppm O₃ when minute ventilation and/or duration of exposure are increased sufficiently. On average, spirometric responses to O₃ exposure appear to decline with increasing age starting at approximately 18 to 20 years of age.
- There is considerable variability in responses between similarly exposed individuals, such that some may experience distinctly larger effects even when small group mean responses are observed. For example, healthy adults exposed to 0.08 ppm O₃ for 6.6 h with moderate exercise exhibited a group mean O₃-induced decrement in FEV₁ of about 6%, but a decrement of >10% was seen in 23% of these individuals. Also, exposure to 0.06 ppm O₃ caused >10% lung function decline in a small percentage (7%) of the subjects.
- Summer camp field studies conducted in southern Ontario, Canada, in the northeastern U.S., and in southern California have also reported lung function responses in pre-adolescent children associated with ambient O₃ levels.
- Repeated acute (1- to 6-h) O₃ exposures at 0.12 to 0.45 ppm over several days in controlled exposure studies typically find that FEV₁ response to O₃ is enhanced on the second of several

days of exposure, but spirometric responses become attenuated on subsequent days with these repeated exposures. However, this tolerance is lost after about a week without exposure.

- Animal toxicologic studies also provide extensive evidence that acute O₃ exposures alter breathing patterns so as to cause rapid shallow breathing (i.e., increased frequency and decreased tidal volume), an effect which attenuates after several days of exposure.
- Results from controlled human exposure studies and animal toxicologic studies provide clear evidence of causality for the associations observed between acute (≤ 24 h) O₃ exposure and relatively small, but statistically significant declines in lung function observed in numerous recent epidemiologic studies. Declines in lung function are particularly noted in children, asthmatics, and adults who work or exercise outdoors.

Respiratory Symptoms:

- Young healthy adult subjects exposed in clinical studies to O₃ concentrations ≥ 0.08 ppm for 6 to 8 h during moderate exercise exhibit symptoms of cough and pain on deep inspiration. An increase in the incidence of cough has been found in clinical studies as low as 0.12 ppm in healthy adults during 1 to 3 h with very heavy exercise and other respiratory symptoms, such as pain on deep inspiration and shortness of breath, have been observed at 0.16 ppm to 0.18 ppm with heavy and very heavy exercise. These O₃-induced respiratory symptoms gradually decrease in adults with increasing age. With repeated O₃ exposures over several days, respiratory symptoms become attenuated, but this tolerance is lost after about a week without exposure.
- The epidemiologic evidence shows significant associations between acute exposure to ambient O₃ and increases in a wide variety of respiratory symptoms (e.g., cough, wheeze, production of phlegm, and shortness of breath) in asthmatic children. Epidemiologic studies also indicate that acute O₃ exposure is likely associated with increased asthma medication use in asthmatic children.

- On the other hand, an effect of acute O₃ exposure on respiratory symptoms in healthy children is not as clearly indicated by epidemiology studies, consistent with diminished symptom responses seen in healthy children in human clinical studies.

Airway Inflammation:

- Inflammatory responses have been observed subsequent to 6.6 h O₃ exposures to the lowest tested level of 0.08 ppm in healthy human adults. Some studies suggest that inflammatory responses may be detected in some individuals following O₃ exposures even in the absence of O₃-induced pulmonary function decrements in those subjects.
- Repeated O₃ exposures over several days leads to an attenuation of most inflammatory markers. However, none of the several markers of lung injury and permeability evaluated show attenuation, indicating continued lung tissue damage during repeated exposure.
- Animal toxicologic studies provide extensive evidence that acute (1 to 3 h) O₃ exposures as low as 0.1 to 0.5 ppm can cause (1) lung inflammatory responses (typified by increased reactive oxygen species, inflammatory cytokines, influx of PMNs, and activation of alveolar macrophages); (2) damage to epithelial airway tissues, (3) increases in permeability of both lung endothelium and epithelium, and (4) increases in susceptibility to infectious diseases due to modulation of lung host defenses.
- Consistent with these experimental findings, there is also limited epidemiologic evidence showing an association between acute ambient O₃ exposure and airway inflammation in children acutely exposed to ambient O₃ concentrations (1-h max O₃ of approximately 0.1 ppm).
- The extensive human clinical and animal toxicological evidence, together with the limited available epidemiologic evidence, is clearly indicative of a causal role for O₃ in inflammatory responses in the airways.

Airway Responsiveness:

- Controlled human exposure studies have found that acute O₃ exposure causes an increase in nonspecific airway responsiveness, as indicated by reductions in concentrations of methacholine or histamine required to produce a given decrease in FEV₁ or increase in SR_{aw}.
- Acute (2- or 3-h) O₃ exposure at 0.25 or 0.4 ppm of allergic asthmatic subjects, who characteristically already have somewhat increased airway responsiveness at baseline, was found to cause further increases in airway responsiveness in response to allergen challenges. Also, repeated daily exposure to 0.125 ppm O₃ for 4 days exacerbated lung function decrements in response to bronchial allergen challenges among persons with preexisting allergic airway disease, with or without asthma.
- Ozone-induced exacerbation of airway responsiveness persists longer and attenuates more slowly than O₃-induced pulmonary function decrements and respiratory symptom responses. Heightened airway responsiveness (reactivity) has also been observed in several laboratory animal species with acute exposures (1 to 3 h) to 0.5 to 1.0 ppm O₃. Ozone increases airway hyperreactivity to bronchoconstrictive agents (e.g., ovalbumin), and there is a temporal relationship between inflammatory cell influx and O₃-induced increases in airway reactivity. Several studies of sensitized laboratory animals showing O₃-induced increases in airway hyperreactivity are consistent with O₃ exacerbation of airway hyperresponsiveness reported in atopic humans with asthma.
- Airway responsiveness has not been widely examined in epidemiologic studies. However, the evidence from human clinical and animal toxicological studies clearly indicate that acute exposure to O₃ can induce airway hyperreactivity, thus likely placing atopic asthmatics at greater risk for more prolonged bouts of breathing difficulties due to airway constriction in response to various airborne allergens or other triggering stimuli.

Respiratory Hospital Admissions and Emergency Department Visits:

- Aggregate population time-series studies observed that ambient O₃ concentrations are positively and robustly associated with respiratory-related hospitalizations and asthma ED

visits during the warm season. These observations are strongly supported by the human clinical, animal toxicologic, and epidemiologic evidence for lung function decrements, increased respiratory symptoms, airway inflammation, and airway hyperreactivity.

- Taken together, the overall evidence supports a causal relationship between acute ambient O₃ exposures and increased respiratory morbidity outcomes resulting in increased ED visits and hospitalizations during the warm season.

B. Cardiovascular Morbidity

At the time of the 1996 O₃ AQCD, the possibility of O₃-induced cardiovascular effects was a largely unrecognized issue. Newly-available evidence has emerged since then which provides considerable plausibility for how O₃ exposure could exert cardiovascular impacts.

- Direct O₃ effects such as O₃-induced release from lung epithelial cells of platelet activating factor (PAF) that may contribute to blood clot formation that would increase the risk of serious cardiovascular outcomes (e.g, heart attack, stroke, mortality). Also, interactions of O₃ with surfactant components in epithelial lining fluid of the lung results in production of oxysterols and reactive oxygen species that may exhibit PAF-like activity contributing to clotting and/or exert cytotoxic effects on lung and heart cells.
- Indirect effects of O₃ may involve O₃-induced secretions of vasoconstrictive substances and/or effects on neuronal reflexes that may result in increased arterial blood pressure and/or altered electrophysiologic control of heart rate or rhythm. Some animal toxicological studies have shown O₃-induced decreases in heart rate, mean arterial pressure, and core temperature.
- Some field/panel studies that examined associations between O₃ and various cardiac physiologic endpoints have yielded limited epidemiologic evidence suggestive of a potential association between acute O₃ exposure and altered HRV, ventricular arrhythmias, and incidence of MI.
- Highly suggestive evidence for O₃-induced cardiovascular effects is provided by a few population studies of cardiovascular hospital admissions which reported positive O₃

associations during the warm season between ambient O₃ concentrations and cardiovascular hospitalizations. Only one controlled human exposure study that evaluated effects of O₃ exposure on cardiovascular health outcomes found no significant O₃-induced differences in ECG, heart rate, or blood pressure in healthy or hypertensive subjects, but did observe an overall increase in myocardial work and impairment in pulmonary gas exchange.

- Overall, this generally limited body of evidence is highly suggestive that O₃ directly and/or indirectly contributes to cardiovascular-related morbidity, but much remains to be done to more fully substantiate links between ambient O₃ exposure and adverse cardiovascular outcomes.

C. Mortality

Numerous recent epidemiologic studies conducted in the United States and abroad have investigated the association between acute exposure to O₃ and mortality. Results from several large U.S. multicity studies as well as several single-city studies indicate a positive association between increases in ambient O₃ levels and excess risk of all-cause (nonaccidental) daily mortality.

- Consistent with observed O₃-related increases in respiratory- and cardiovascular-related morbidity, several newer multicity studies, single-city studies, and several meta-analyses of these studies have provided relatively strong epidemiologic evidence for associations between short-term O₃ exposure and all-cause mortality, even after adjustment for the influence of season and PM.
- Determining cause-specific mortality is more difficult due to reduced statistical power by which to examine cause-specific associations and the lack of clarifying information on contributing causes of death. That is, attribution to one or the other of the more specific cardiopulmonary causes may underplay contributions of chronic cardiovascular disease to “respiratory” deaths (e.g., a heart attack victim succumbing to acute pneumonia) or vice versa.

- Consistently positive associations have been reported for O₃-related cardiovascular mortality across approximately 30 studies, with two well-conducted multicity studies in the United States and Europe yielding small, but statistically significant positive associations.
- Both animal and human studies provide evidence suggestive of plausible pathways by which risk of respiratory or cardiovascular morbidity and mortality could be increased by ambient O₃ either acting alone or in combination with copollutants in ambient air mixes.
- This overall body of evidence is highly suggestive that O₃ directly or indirectly contributes to non-accidental and cardiopulmonary-related mortality, but additional research is needed to more fully establish underlying mechanisms by which such effects occur.

3. Health Effects of Long-term Exposures to Ozone

In the 1996 O₃ AQCD, the available epidemiologic data provided only suggestive evidence that respiratory health effects were associated with chronic O₃ exposure. Animal toxicologic studies indicated that chronic O₃ exposure caused structural changes in the respiratory tract, and simulated seasonal exposure studies in animals suggested that such exposures might have cumulative impacts. As summarized below, recent studies are generally consistent with the conclusions drawn in the previous 1996 AQCD.

A. Respiratory Morbidity

Lung Function:

- Recent epidemiologic studies observed that reduced lung function growth in children was associated with seasonal exposure to O₃; however, cohort studies investigating the effect of annual or multiyear O₃ exposure observed little clear evidence for impacts of longer-term, relatively low-level O₃ exposure on lung function development in children.
- The epidemiologic data, collectively, indicate that the current evidence is suggestive but inconclusive for respiratory health effects from long-term O₃ exposure.

Morphological Changes:

- Animal toxicologic studies continue to show chronic O₃-induced structural alterations in several regions of the respiratory tract including the centracinar region. Morphologic evidence from some recent studies using exposure regimens that mimic seasonal exposure patterns report increased lung injury compared to conventional chronic stable exposures.
- Infant rhesus monkeys repeatedly exposed to 0.5 ppm 8h/day O₃ for 11 episodes exhibited: (1) remodeling of the distal airways; (2) abnormalities in tracheal basement membrane; (3) eosinophil accumulation in conducting airways; and (4) decrements in airway innervation. Long-term O₃ exposure of rats to 0.5 or 1.0 ppm for 20 months resulted in upper respiratory tract mucus metaplasia and hyperplasia in the nasal epithelium (0.25 or 0.5 ppm, 8h/day, 7days/wk for 13 weeks).
- The persistent nature of these cytological changes raise the possibility of long-lasting alterations in human airways in response to chronic O₃ exposure, but it is highly uncertain as to what long-term patterns of exposure or O₃ concentrations in humans may be requisite to produce analogous morphological changes. Nor is it now possible to characterize the possible magnitude or severity of any such effects occurring in humans in response to ambient O₃ exposures at levels observed in the United States.

Incidence of Lung Cancer:

- The weight of evidence from recent animal toxicological studies and a very limited number of epidemiologic studies do not support ambient O₃ as a pulmonary carcinogen.

B. Mortality

- Results from the few available epidemiologic studies are inconsistent regarding the association between long-term exposure to O₃ and mortality. There is little evidence to suggest a causal relationship between chronic O₃ exposure and increased risk for mortality in humans.

4. Health Effects of Ozone-Containing Pollutant Mixtures

The potential interaction of pollutant mixtures with O₃ is poorly understood and the animal studies reviewed in the 1996 O₃ AQCD reported additive, synergistic or antagonistic effects depending on the exposure regimen and the endpoint studied. A few new controlled human exposure and animal toxicology studies reviewed in Chapters 4, 5, and 6 investigated health effects associated with O₃-containing pollutant mixtures of near ambient levels. As noted below, recent studies, although generally consistent with conclusions drawn in the 1996 O₃ AQCD, have added some new information, particularly with regard to interactions between O₃ and PM.

- Controlled human exposure studies indicate that continuous exposure of healthy human adults to SO₂ or NO₂ increases bolus dose O₃ absorption, suggesting that co-exposure to other gaseous pollutants in the ambient air may enhance O₃ absorption.
- Other controlled human exposure studies that evaluated response to allergens in asthmatics (allergic and dust-mite sensitive) suggest that O₃ enhances response to allergen challenge. Consistent with these findings, animal toxicology studies also reported enhanced response to allergen on exposure to O₃.
- A few other animal toxicology studies that exclusively investigated the co-exposure of PM and O₃ reported increased response (lung tissue injury, inflammatory and phagocytosis) to the mixture of PM + O₃ compared to either PM or O₃ alone.
- Recent investigations on the copollutant interactions using simulated urban photochemical oxidant mixes suggest the need for similar studies in understanding the biological basis for air pollutant mixture effects observed in epidemiologic studies.

5. Susceptibility or Vulnerability to Effects Associated with Exposure to Ozone

Various factors have been shown to influence individuals' responses to environmental air pollutants. Factors that increase susceptibility to O₃-related effects include innate factors, such as genetic predisposition or developmental effects, or disease status. Other factors can lead to enhanced vulnerability to O₃-related effects, such as heightened exposures or activity patterns. In the 1996 O₃ AQCD, available evidence suggested that children, asthmatics, and outdoor

workers were populations that may be more susceptible or vulnerable to effects of O₃ exposure. In addition, controlled human exposure studies also demonstrated a large variation in sensitivity and responsiveness to O₃ in studies of healthy subjects, but the specific factors that contributed to this intersubject variability were yet to be identified. Recent studies have built upon the evidence available in the previous review. Factors related to susceptibility or vulnerability to O₃ exposure-related effects are briefly summarized below:

People with Preexisting Pulmonary Diseases:

- Ozone-induced differential responses in lung function and AHR in people with allergic rhinitis suggest that asthmatics have potentially greater responses than healthy people with exposure to O₃. There is a tendency for slightly increased spirometric responses in mild asthmatics and allergic rhinitics relative to healthy young adults. Spirometric responses in asthmatics appear to be affected by baseline lung function, i.e., responses increase with disease severity.
- Repeated O₃ exposure over several days has been shown to increase responsiveness to bronchial allergen challenge in subjects with preexisting allergic airway disease, with or without asthma. Asthmatics also show a significantly greater neutrophil response (18 h postexposure) than similarly-exposed healthy individuals.
- Epidemiologic studies have reported associations with a range of respiratory health outcomes in asthmatics, from decreases in lung function to hospitalization or ED visits for asthma, thus supporting this population group as being likely to experience increased risk for O₃-induced health effects.
- Controlled human exposure studies have not found evidence of larger spirometric changes in people with COPD relative to healthy subjects, this may be due to the fact that most people with COPD are older adults who would not be expected to have such changes based on their age. However, new epidemiologic evidence indicates that people with COPD may be more likely to experience other effects, including emergency room visits, hospital admissions, or premature mortality.

Age-related:

- Controlled human exposure studies have shown that lung function responses to O₃ varies with age, with responsiveness generally diminishing after about 18 to 20 years of age. Children and older adults thus have lesser respiratory symptoms with O₃ exposure than young healthy adults. Potentially increased O₃ doses can be received by individuals experiencing less severe respiratory symptoms.
- Evidence from newer epidemiologic studies supports the 1996 O₃ AQCD conclusions that children are more likely at increased risk for O₃-induced health effects. Notably, epidemiologic studies have indicated adverse respiratory health outcomes associated with O₃ exposure in children. In addition, recently published epidemiologic studies also suggest that older adults (aged ≥65 years) appear to be at excess risk of O₃-related mortality or hospitalization.

Heightened vulnerability due to greater exposures:

- Epidemiologic studies have provided some evidence to indicate that outdoor workers are more vulnerable to O₃-related effects, which is likely related to their increased exposure to ambient air pollution.
- Controlled human exposure studies clearly established differential biological response to O₃ based on physical activity (exertion). Epidemiologic studies also suggest that exercising (moderate to high physical exertion) children and adolescents appear to demonstrate increased responsiveness to ambient concentrations of O₃ and may be more likely to experience O₃-induced health effects. Animal studies show a similar impact of exercise on responsiveness to O₃.

Genetic susceptibility:

- Animal toxicologic studies provide supportive evidence to the observations of innate susceptibility. Various strains of mice and rats have demonstrated the importance, in general, of genetic background in O₃ susceptibility. Moreover, genetic and molecular

characterization studies in laboratory animals identified genetic loci responsible for both sensitivity and resistance.

- New human clinical and epidemiologic studies also have shown that genetic polymorphisms for antioxidant enzymes and inflammatory genes (GSTM1, NQO1, and *Tnf- α*) may modulate the effect of O₃ exposure on pulmonary function and airway inflammation.

E.5 VEGETATION AND ECOLOGICAL EFFECTS

Data published since 1996, as assessed in Chapter 9 and associated annex materials, continues to support and strengthen the conclusions of previous O₃ AQCDs. The main findings/conclusions derived from the current Chapter 9 assessment of O₃ ecological effects are as follows.

General

- The ecological effects of O₃ appear to be widespread across the United States based on recent biomonitoring studies using clover and other species grown in plots across the United States, as well as regional forest health visible injury surveys.
- Some plant community compositions may be shifting based on recent studies of competition among plants in managed pasture lands, as well in natural unmanaged lands where increased O₃ effects on sensitive species in the community can reduce their presence in the community.
- Research to date has focused at the species level, with very few studies at the ecosystem level. The lack of data at this organizational level hampers the assessment of O₃ risk to ecosystem services, such as water quality and quantity, that contribute to human well-being.

Methodologies

- New methodologies coming into use since 1996 have not fundamentally altered our understanding of O₃ effects on plants or the conclusions of the 1996 O₃ AQCD. Since 1996, there has been a shift from chamber-based studies to the field-based approach, including plot

and regional visible injury surveys and the use of the non-chambered free air CO₂ exposure (FACE) systems. The FACE system results support earlier observations of foliar injury and reduced volume growth in aspen and indicate reduced yield in soybean cultivars similar to earlier studies in open-top chamber systems (OTC).

- The use of biomonitoring since 1996 has advanced identification and symptom verification of sensitive species and has been a useful tool for indicating the extent of O₃ effects across most of the eastern and southeastern United States and many parts of the West.
- The development and improvement of stomatal models for predicting O₃ uptake in Europe have fostered more universal measures of exposure response. These simulation tools may provide a better means to relate ambient exposure to plant response in the future but currently are insufficient for use across broad geographical areas of the United States.
- Since 1996, the use of passive samplers for monitoring O₃ in rural and remote areas has expanded, offering a potential for improved exposure data in areas not actively monitored. The testing and development of these samplers will ultimately provide a strategy to expand air quality monitoring into areas for which exposure characterization is currently done by geospatial extrapolation techniques such as Kriging.

Mode of Action

- There are several steps in the process of O₃ uptake and toxicity that are better understood now than in 1996, based on new information gained in part by use of improved molecular tools for following rapid changes that occur within the leaf. These advancements are important for refining hypotheses on O₃ uptake and improving understanding of exposure-response relationships.
- Ozone entrance into the leaf through the stomata remains the critical step in O₃ sensitivity. Although the initial reactions within the leaf are still unclear, the involvement of H₂O₂ is clearly indicated. The initial sites of membrane reactions seem to involve transport

properties. The primary set of metabolic reactions that O₃ triggers currently includes those typical of “wounding” responses generated by leaf cutting or by insect attack.

- The alteration of normal metabolism due to wounding spreads outside the cytoplasm. One of the secondary reactions is linked to a senescence response. The loss of photosynthetic capacity is linked to lower productivity (although not fully elucidated) and to problems with efficient translocation of carbon.
- Chronic O₃ effects are linked to the senescence process or some physiological process closely linked to senescence, e.g., translocation, re-absorption, allocation of nutrients and carbon.

Modification of Growth Response

- Many biotic and abiotic factors, including insects, pathogens, root microbes and fungi, temperature, water and nutrient availability, and other air pollutants, as well as elevated CO₂, influence or alter the plant’s response to O₃. A few studies published since 1996 have improved our understanding of the role of these interactions in modifying O₃-induced plant responses.
- Biotic Interactions: Recent studies have supported earlier conclusions that O₃ often increases the likelihood and success of insect attacks, at least by chewing insects. Less is known regarding sucking insects (e.g., aphids). It seems that some insect problems could be exacerbated by increased O₃ exposure, but predicting any particular O₃-insect interaction is not possible at this time.
- Biotic Interactions: More information is available regarding disease interactions. Ozone exposure generally increases plant diseases associated with facultative necrotrophic plant pathogens. Pathogens that benefit from damage to cells are enhanced by O₃ stress to their hosts, whereas pathogens that require healthy hosts are depressed by O₃ stress.

- Biotic Interactions: A few new studies have demonstrated O₃ impacts on intraspecific plant competition. In grass-legume pastures, the legume component is more O₃-sensitive and is reduced over time. Similarly, grass competition on pine seedlings can enhance O₃ effects on the seedlings, possibly through the grass's ability to outcompete the seedlings for water.
- Abiotic Interactions: New information on the role of abiotic or physical factors interacting with O₃ stress support 1996 O₃ AQCD conclusions. Some studies have shown an increasing effect of O₃ with increasing temperature, but others have shown little effect of temperature. Temperature is an important variable affecting plant O₃ response in the presence of elevated CO₂ levels associated with climate change. It also appears that low temperatures are important, in that O₃ exposure sensitizes plants to low temperature stress.
- Abiotic Interactions: New information on the role of drought and water availability published since 1996 confirms earlier conclusions regarding the increased effects of O₃ where readily available soil moisture results in increased needle/leaf conductivity and thus increased O₃ uptake. Additional studies demonstrated again the partial "protection" against adverse effects of O₃ by drought. There was also evidence that O₃ predisposes plants to drought stress. The net results of these interactions are negative, at least in the short term, although longer lived species like trees could benefit from increased water use efficiency.

Effects-Based Exposure Indices

- Exposure indices are metrics relating plant response (i.e., growth or yield) to monitored ambient O₃ concentrations over time to provide a consistent metric for reviewing and comparing exposure-response effects obtained from various studies. Such metrics may also provide a basis for developing air quality standards that are protective of ecological resources. The 1996 O₃ AQCD focused on research where a large number of indices were developed that included various functional and statistical summaries of ambient hourly concentrations over designated time periods. The development of those indices focused on considering and including some, but not all, the factors that affect O₃ uptake and expression of effects.

- Conclusions from the 1996 O₃ AQCD regarding an ambient-exposure based index are still valid. No information since 1996 significantly alters the basic conclusions, and most studies in this interim have further supported them. The key conclusions are as follows:
 - Ozone effects in plants are cumulative;
 - Higher O₃ concentrations appear to be more important than low concentrations in eliciting a response;
 - Plant sensitivity to O₃ varies with time of day and plant development stage; and
 - Exposure indices that accumulate the O₃ hourly concentrations and preferentially weight the higher concentrations have a better statistical fit to growth/yield response than do mean or peak indices.

- Based on the current state of knowledge, exposure indices that differentially weight the higher hourly average O₃ concentrations but include the mid-level values represent the best approach for relating vegetation effects to O₃ exposure in the United States. A large database for crops and tree seedlings exists and has been used for establishing exposure-response relationships and predicting effects for a range of exposure concentrations. In 1996, EPA considered three specific concentration-weighted indices for use as air quality indicators: the cutoff concentration-weighted SUM06, the AOT60, and the sigmoid-weighted W126. All three performed equally well based on goodness-of-fit tests. Since 1996, there have been no published experimental studies that would alter the consideration of these concentration-weighted cumulative indices.

- Studies available since 1996 strengthen earlier conclusions on the role of exposure components such as duration, concentration, and temporal patterns in determining plant growth response to O₃ exposure. New studies since 1996 have shown experimentally the disconnection of peak events and maximal stomatal conductance at a variety of sites. The identification of sensitivity linked to time of day (i.e., period of maximum conductance) was reported in the 1996 O₃ AQCD. The new studies will offer future avenues for building this temporal component into an exposure index. Similarly, recent reviews of the plant literature have reported a large number of species with nighttime conductance capable of O₃ uptake. The designated time interval for cumulating exposure (i.e., 12 to 24 h) needs reconsideration.

- Recent research in Europe has focused on a flux-based approach to improve upon the ambient air concentration-based (i.e., exposure indices) approach to assess risk from O₃ across different climate regions. However, such approaches need further development to incorporate the necessary complexity across space and time to be non-site and non-species specific. Also, at this time, the database is inadequate for linking O₃ flux to growth responses.

Ozone Exposure-Plant Response Relationships

- Data published since 1996 continues to support and strengthen the conclusions of previous O₃ AQCDs that there is strong evidence that current ambient O₃ concentrations cause (1) decreased growth and biomass accumulation in annual, perennial and woody plants, including agronomic crops, annuals, shrubs, grasses, and trees; (2) decreased yield and/or nutritive quality in a large number of agronomic and forage crops; and (3) impaired aesthetic quality of many native plants and trees by increased foliar injury.
- Since 1996, the published studies have supported earlier conclusions on reduced growth and yield in a number of crops and trees, and have used multiple approaches, including regional visible injury surveys, measured growth responses across ambient exposure gradients, empirical exposure studies in chambered and non-chambered systems, as well as process model simulations.
- Studies of growth response using open-top chambers have provided useful data for assessing O₃ impact on common and economically valuable species, and developing functional growth-response models that enable the prediction of O₃ impact over a wide range of ambient air exposures. The studies were designed to maximize statistical robustness by replicating a number of treatments and, at the same time, considering issues of extrapolation by conducting the studies across a wide range of crop-growing regions and forested sites in the United States. Such designs allowed the studies to account for climate and growing conditions, as well as regional crop growing practices.

- Recent exposure studies in Illinois using the non-chambered FACE system with soybean cultivars reported reductions in yield of soybeans in two year-long studies similar to the reductions found in multiple soybean studies conducted in the 1980's using open-top chamber systems. Multiple-year exposure studies using the FACE system (a) found foliar injury and reduced volume growth in aspen and maple similar to results reported from earlier open-top chamber studies and (b) highlight the importance of multiyear studies with longer-growing species.
- Since 1996, additional studies have supported 1996 O₃ AQCD conclusions that deciduous trees are generally less O₃ sensitive than most annual species or crop plants, with the exception of a few very sensitive genera (e.g., sensitive clones or genotypes of *Populus*) and sensitive species (e.g., black cherry). Coniferous species have a wide range of O₃ sensitivities but, in general, are less sensitive than deciduous species. Among conifers, the slower-growing species are less sensitive than faster-growing species. Data from a few European studies support these conclusions.
- For all types of perennial vegetation, cumulative effects over more than one growing season may be important; studies of one or a few seasons may under- or overestimate O₃ impacts on these species. Results from multiyear studies sometimes found a pattern of increased effects in subsequent years, whereas other studies reported growth decreases due to O₃ that become less significant or disappear over time. It is difficult to conduct empirical experiments with long-lived trees, because even multiyear exposures only account for a small fraction of the tree's lifetime. Model simulations of growth have been a tractable approach to account for time and changing climate in assessing the impact on long-lived trees.

Ecosystem Effects

- There is strong evidence that O₃, in locations where ambient levels are relatively high, is an important stressor of ecosystems, with documented impacts on the biotic condition, ecological processes, and chemical/physical nature of natural ecosystems. Experimentally documented effects on individual keystone species and their associated microflora and fauna may cascade through the ecosystem to the landscape level, but this has not been quantified.

- Systematic injury surveys (e.g., USDA Forest Health Network; Europe's ICP Forests) show that foliar injury and crown/canopy deformations occur in O₃-sensitive species in many regions of the United States and Europe. However, the lack of general correspondence between foliar symptoms and growth effects means that other methods must be used to estimate regional effects of O₃ on tree growth. Regional studies of radial growth in mature trees, combined with data from many controlled studies with seedlings and a few studies with mature trees, suggest that ambient O₃ may be reducing the growth of mature trees in some U.S. locations.
- The use of physiological-based process models to simulate tree growth, combined with stand-level models predicting forest composition and productivity, is an approach being used recently in assessing O₃ impacts on forests. These tools suggest that modest O₃ effects on growth may accumulate over time and interact with effects of other naturally occurring stresses (e.g., drought, nutrient availability). For mixed-species stands, the models predict that overall stand growth may not be affected, but competitive interactions among species may change the composition due to growth reductions in sensitive species.
- The knowledge base for examining the range of ecological effects of O₃ on natural ecosystems is growing, but significant uncertainties remain regarding O₃ effects at the ecosystem level. A number of significant areas for investigation that would improve our ability to assess O₃ effects on ecosystems and the services they provide for human well-being have been outlined and discussed (see Annex AX-9).

E.6 TROPOSPHERIC OZONE EFFECTS ON UV-B FLUX AND ITS ROLE IN CLIMATE CHANGE

Molecular properties specific to O₃ include a capacity for absorbing incoming ultraviolet (UV) and infrared (IR) radiation, and both incoming solar and outgoing terrestrial IR radiation. Consequently, O₃ plays an essential role in shielding the earth's surface from harmful levels of UV-B radiation, by way of the stratospheric O₃ layer. Its effectiveness as a screen for the residual UV-B flux that penetrates the stratosphere and passes into the troposphere and its role in

reducing UV-induced human health effects are addressed in Chapter 10. The radiation-absorbing properties of O₃ also make it a greenhouse gas (GHG) having global and regional consequences for climate, as also addressed in Chapter 10. Important conclusions from Chapter 10 are summarized below.

- The distribution of O₃ within the atmosphere. Ozone is distributed very unevenly within the atmosphere, with ~90% of the total atmospheric burden present in the stratosphere. The remaining ~10% is distributed within the troposphere, with higher relative concentrations near the source of its precursors at the surface. Concentrations of O₃ at the mid- and upper-troposphere vary, depending upon meteorological conditions.
- Multiple factors govern the flux of UV-B radiation at the Earth's surface. Latitude and altitude are the two most important factors that define the residual UV-B flux at the surface. Natural variation in the total column density of stratospheric O₃ is also an important factor. All of these factors are followed in importance by tropospheric clouds, particulate matter (PM) and O₃. The effect of natural stratospheric variation, clouds, PM and tropospheric O₃ on UV fluxes within the troposphere and at the surface are each very difficult to predict.
- A UV-B "climatology" is needed to predict human exposure levels. A UV-B climatology, representing patterns and trends in UV-B flux at the Earth's surface, must be based on extended in situ observations in order to adequately capture natural variability and the effects of human activities on atmospheric UV-B absorbers. At present, the body of UV-B measurements cannot support the development of a climatology.
- Human exposure to UV-B radiation. Quantitative evaluation of human exposure to UV-B radiation is necessary to perform health risk assessment for UV-B-related health effects. Individuals who participate in outdoor sports and activities, work outdoors, live in geographic areas with higher solar flux, and/or engage in high-risk behavior (e.g., extended sun bathing) can reasonably be projected to be at increased risk for higher UV radiation exposures. However, little is known about the impact of variability in these factors on individual exposure to UV radiation.

- Human health effects of UV-B radiation. Exposure to UV-B radiation is associated with increased risk of erythema, nonmelanoma and melanoma skin cancers, ocular damage, and immune system suppression. Some studies have attempted to estimate the potential effects of changes in surface-level UV flux resulting from stratospheric O₃ depletion on these health outcomes; however, the numerous simplifying assumptions made in the assessments limit the usefulness of the risk estimates. The effect of changes in surface-level O₃ concentrations on UV-induced health outcomes cannot yet be critically assessed within reasonable uncertainty.
- Vitamin D-related health benefits of UV-B radiation. A potential health benefit of increased UV-B exposure relates to the production of vitamin D in humans. Several studies have found that UV-B radiation, by increasing vitamin D production, is associated with reduced risks of various cancers. However, as with other impacts of UV-B on human health, this beneficial effect of UV-B has not been studied in sufficient detail to allow for a credible health benefits assessment. No study has been done of the decreased risk of cancer resulting from increased UV radiation attributable to decreased tropospheric O₃ levels, but the change in risk is expected to be unappreciable.
- Ozone is a potent GHG. Ozone traps incoming solar radiation at both ends of the spectrum, as well as shortwave radiation that is scattered from high-albedo portions of the Earth's surface. Outgoing terrestrial IR is absorbed by O₃ within the range where water vapor does not absorb, so that natural variability in humidity does not alter its radiative impact. These effects directly force climate. By participating in the oxidative chemistry of the atmosphere, O₃ can indirectly and negatively force climate by the removal of other greenhouse gases.
- Multiple factors influence the forcing effect of tropospheric O₃. Estimates of present-day forcing by O₃ depend upon currently available information on pre-industrial and current O₃ concentrations. Both are limited and, therefore, very uncertain. Other factors, including the albedo of underlying surface, altitude and co-occurrence of PM can also complicate the calculation of globally-averaged forcing.

- Globally-averaged direct forcing by O₃. On the basis of the best available information, a 2001 Intergovernmental Panel on Climate Change (IPCC) report offered an estimated value of $0.35 \pm 0.15 \text{ Wm}^{-2}$ for the annual, globally-averaged direct forcing by tropospheric O₃. Another recent estimate places this value at $0.5 \pm 0.2 \text{ Wm}^{-2}$.
- Projections of forcing by O₃ into the future. A CTM-climate modeling intercomparison study carried out as part of the third assessment by the IPCC yielded an estimated 0.4 to 0.78 Wm^{-2} forcing by O₃ by the year 2100. The authors of this study concluded that O₃ can be expected to be an important contributor to climate forcing into the future.
- Climate forcing by O₃ at the regional scale may be its most important impact on climate. Satellites have detected high O₃ concentrations localized at the regional scale that are associated with large urban centers and extensive biomass burning. Climate forcing by these high, regional-scale O₃ concentrations have been estimated to be on the order of 1 Wm^{-2} (a substantial fraction of the direct, globally-averaged forcing due to well-mixed GHGs, including CO₂). The impact of climate forcing at this level depends upon the particular characteristics of the region in which it occurs. At present, regional-scale modeling studies are not available that provide estimates of these effects.

E.7 MATERIALS DAMAGE

The Chapter 11 discussion of O₃ effects on man-made materials mainly summarizes key information from the 1996 O₃ AQCD, given that little new pertinent research information on O₃-related materials damage has been published since then. Key points include the following:

- Ozone and other photochemical oxidants react with many economically important man-made materials, decreasing their useful life and aesthetic appearance. Materials damaged by O₃ include elastomers; textiles and fibers; dyes, pigments, and inks; and paints and other surface coatings.

- Elastomeric compounds (natural rubber and synthetic polymers and copolymers of butadiene, isoprene, and styrene) are highly susceptible, to even low O₃ concentrations. These compounds are damaged by O₃ breaking molecular chains at the carbon-carbon double bond and by adding a chain of three oxygen atoms directly across the double bond. This structure change promotes characteristic cracking of stressed/stretched rubber called “weathering.” Tensile strain produces cracks on the surface of the rubber that increase in size and number with increased stress/stretching. The rate of crack growth is dependent on degree of stress, type of rubber compound, O₃ concentration, duration of exposure, O₃ velocity, and temperature. After initial cracking, further O₃ penetration results in additional cracking and, eventually, mechanical weakening.
- Ozone can damage textiles and fabrics by mechanisms similar to those associated with elastomers. Generally, synthetic fibers are less affected by O₃ than natural fibers. Overall, O₃ contribution to degradation of textiles and fabrics is not considered significant.
- Ozone fading of textile dyes is a diffusion-controlled process, with the rate of fading being controlled by diffusion of the dye to the fiber surface. Many textile dyes react with O₃. The rate and severity of the O₃ attack is influenced by the chemical nature of the textile fiber.
- Paints applied to exterior surfaces of buildings and other structures (e.g., bridges), as well as several artists’ pigments, are also sensitive to fading and oxidation by O₃ at concentrations found in U.S. urban areas.

1. INTRODUCTION

This is an update revision of the document, “*Air Quality Criteria for Ozone and Related Photochemical Oxidants*,” published by the U.S. Environmental Protection Agency (EPA) in 1996 (U.S. Environmental Protection Agency, 1996a). That 1996 Ozone Air Quality Criteria Document (O₃ AQCD) provided scientific bases for Congressionally-mandated periodic review by the EPA of the Ozone National Ambient Air Quality Standards (O₃ NAAQS), which led to promulgation of new O₃ NAAQS by EPA in 1997 (Federal Register, 1997).

The present document critically assesses the latest scientific information relative to characterizing health and welfare effects associated with the presence of various concentrations of O₃ and related oxidants in ambient air. It builds upon the previous 1996 EPA O₃ AQCD, by focusing on evaluation and integration of scientific information relevant to O₃ NAAQS criteria development that has become available since that covered by the 1996 criteria review; and it will provide scientific bases for the current periodic review of the O₃ NAAQS.

This introductory chapter of the revised O₃ AQCD presents: (a) background information on legislative requirements, the criteria and NAAQS review process, and the history of O₃ NAAQS reviews (including a chronology of changes in key elements of the O₃ standards); (b) an overview of the current O₃ criteria review process and associated key milestones; and (c) an orientation to the general organizational structure and content of the document.

1.1 LEGAL AND HISTORICAL BACKGROUND

1.1.1 Legislative Requirements

Two sections of the U.S. Clean Air Act (CAA) govern establishment, review, and revision of National Ambient Air Quality Standards (NAAQS). Section 108 of the CAA (42 U.S.C. 7408) directs the Administrator of the U.S. Environmental Protection Agency (EPA) to identify ambient air pollutants that may be reasonably anticipated to endanger public health or welfare and to issue air quality criteria for them. The air quality criteria are to reflect the latest scientific information useful in indicating the kind and extent of all identifiable effects on public health or welfare that may be expected from the presence of a given pollutant in ambient air.

Section 109(a) of the CAA (42 U.S.C. 7409) directs the Administrator of EPA to propose and promulgate primary and secondary NAAQS for pollutants identified under Section 108. Section 109(b)(1) defines a primary standard as one that, in the judgment of the Administrator, is requisite to protect the public health (see inset below) based on the criteria and allowing for an adequate margin of safety. The secondary standard, as defined in Section 109(b)(2), must specify a level of air quality that, in the judgment of the Administrator, is requisite to protect the public welfare (see inset below) from any known or anticipated adverse effects associated with the presence of the pollutant in ambient air, based on the criteria.

**EXAMPLES OF
PUBLIC HEALTH EFFECTS**

- □ Effects on the health of the general population, or identifiable groups within the population, who are exposed to pollutants in ambient air
- □ Effects on mortality
- □ Effects on morbidity
- □ Effects on other health conditions including indicators of:
 - pre-morbid processes,
 - risk factors, and
 - disease

**EXAMPLES OF
PUBLIC WELFARE EFFECTS**

- □ Effects on personal comfort and well-being
- □ Effects on economic values
- □ Deterioration of property
- □ Hazards to transportation
- □ Effects on the environment, including:

• animals	• vegetation
• climate	• visibility
• crops	• water
• materials	• weather
• soils	• wildlife

Section 109(d) of the CAA (42 U.S.C. 7409) requires periodic review and, if appropriate, revision of existing criteria and standards. If, in the Administrator's judgment, the Agency's review and revision of criteria make appropriate the proposal of new or revised standards, such standards are to be revised and promulgated in accordance with CAA Section 109(b). Or, the Administrator may find that revision of the standards is not appropriate and conclude the review by leaving existing standards unchanged. Section 109(d)(2) of the CAA also requires that an independent scientific review committee be established to advise the EPA Administrator on NAAQS matters, including the scientific soundness of criteria (scientific bases) supporting NAAQS decisions. This role is fulfilled by the Clean Air Scientific Advisory Committee (CASAC), which is administratively supported by EPA's Science Advisory Board (SAB).

1.1.2 Criteria and NAAQS Review Process

Periodic reviews by EPA of criteria and NAAQS for a given criteria air pollutant progress through a number of steps, beginning with preparation of an air quality criteria document (AQCD) by EPA's National Center for Environmental Assessment Division in Research Triangle Park, NC (NCEA-RTP). The AQCD provides a critical assessment of the latest available scientific information upon which the NAAQS are to be based. Drawing upon the AQCD, staff of EPA's Office of Air Quality Planning and Standards (OAQPS) prepare a Staff Paper that evaluates policy implications of the key studies and scientific information contained in the AQCD and presents EPA staff conclusions and recommendations for standard-setting options for the EPA Administrator to consider. The Staff Paper is intended to help "bridge the gap" between the scientific assessment contained in the AQCD and the judgments required of the Administrator in determining whether it is appropriate to retain or to revise the NAAQS. Iterative drafts of the AQCD and the Staff Paper (as well as other analyses, such as exposure and/or risk assessments, supporting the Staff Paper) are made available for public comment and CASAC review. Final versions of the AQCD and Staff Paper incorporate changes made in response to CASAC and public review. Based on the information in these documents, the EPA Administrator proposes decisions on whether to retain or revise the NAAQS, taking into account public comments and CASAC advice and recommendations. The Administrator's proposed decisions are published in the *Federal Register*, with a preamble that presents the rationale for the decisions and solicits public comment. After considering comments received on the proposed decisions, the Administrator then makes final decisions on retaining or revising the NAAQS, which are promulgated in a *Federal Register* notice that addresses significant comments received on the proposal.

NAAQS decisions involve consideration of the four basic elements of a standard: *indicator, averaging time, form, and level*. The indicator defines the pollutant to be measured in the ambient air for the purpose of determining compliance with the standard. The averaging time defines the time period over which air quality measurements are to be obtained and averaged, considering evidence of effects associated with various time periods of exposure. The form of a standard defines the air quality statistic that is to be compared to the level of the standard (i.e., an ambient concentration of the indicator pollutant) in determining whether an area attains the standard. The form of the standard specifies the air quality measurements that

are to be used for compliance purposes (e.g., the 98th percentile of an annual distribution of daily concentrations; the annual arithmetic average), the monitors from which the measurements are to be obtained (e.g., one or more population-oriented monitors in an area), and whether the statistic is to be averaged across multiple years. These basic elements of a standard are the primary focus of the staff conclusions and recommendations in the Staff Paper and in the subsequent rulemaking, building upon the policy-relevant scientific information assessed in the AQCD and on the policy analyses contained in the Staff Paper. These four elements taken together determine the degree of public health and welfare protection afforded by the NAAQS.

1.1.3 Regulatory Chronology¹

On April 30, 1971, primary and secondary NAAQS for photochemical oxidants were promulgated by EPA under Section 109 of the CAA (36 FR 8186). These NAAQS were set at an hourly average of 0.08 ppm total photochemical oxidants, not to be exceeded more than 1 h per year. On April 20, 1977, the EPA announced (42 FR 20493) the first review and updating of the 1970 Air Quality Criteria Document for Photochemical Oxidants in accordance with Section 109(d) of the CAA. In preparing that criteria document, EPA made two external review drafts of the document available for public comment, and these drafts were peer reviewed by the Subcommittee on Scientific Criteria for Photochemical Oxidants of EPA's Science Advisory Board (SAB). A final revised AQCD for Ozone and Other Photochemical Oxidants was then published on June 22, 1978.

Based on the revised 1978 AQCD and taking into account the advice and recommendations of the SAB Subcommittee and public comments, the EPA announced (44 FR 8202) a final decision to revise the NAAQS for photochemical oxidants on February 8, 1979. That final rulemaking revised the primary standard from 0.08 ppm to 0.12 ppm, set the secondary standard to be the same as the primary standard, changed the chemical designation of the standards from photochemical oxidants to O₃, and revised the definition of the point at which the standard is attained as indicated in Table 1-1.

¹This following text is excerpted and adapted from the "Proposed Decision on the National Ambient Air Quality Standards for Ozone," 57 FR 35542, 35542-35557 (August, 10, 1992) and the "National Ambient Air Quality Standards for Ozone; Final Rule," 62 FR 38856, 83356-38896 (July 18, 1997).

Table 1-1. National Ambient Air Quality Standards (NAAQS) for Ozone

Date of Promulgation	Primary and Secondary NAAQS	Averaging Time
February 8, 1979	0.12 ppm ^a (235 µg/m ³)	1 h ^b
July 18, 1997	0.08 ppm ^a (157 µg/m ³)	8 h ^c

^a1 ppm = 1962 µg/m³, 1 µg/m³ = 5.097 × 10⁻⁴ ppm @ 25 °C, 760 mm Hg.

^bThe standard is attained when the expected number of days per calendar year with a maximum hourly average concentration above 235 µg/m³ (0.12 ppm) is equal to or less than one.

^cBased on the 3-year average of the annual fourth-highest daily maximum 8-h average concentration measured at each monitor within an area.

Source: Federal Register (1979, 1997).

On March 17, 1982, in response to requirements of Section 109(d) of the CAA, the EPA announced (47 FR 11561) that it planned to revise the existing 1978 AQCD for Ozone and Other Photochemical Oxidants; and, on August 22, 1983, it announced (48 FR 38009) that review of the primary and secondary NAAQS for O₃ had been initiated. The EPA provided a number of opportunities for expert review and public comment on revised chapters of the AQCD, including two public peer-review workshops in December 1982 and November 1983. Comments made at both workshops were considered by EPA in preparing the First External Review Draft that was made available (49 FR 29845) on July 24, 1984, for public review. On February 13, 1985 (50 FR 6049) and then on April 2, 1986 (51 FR 11339), EPA announced two public CASAC meetings, which were held on March 4-6, 1985 and April 21-22, 1986, respectively. At these meetings, the CASAC reviewed external review drafts of the revised AQCD for O₃ and Other Photochemical Oxidants. After these two reviews, CASAC's consensus views were summarized by the CASAC Chair in an October 1986 letter to the EPA Administrator, which stated that the document "represents a scientifically balanced and defensible summary of the extensive scientific literature." Taking into account public and CASAC comments on the two external review drafts, revisions were made by EPA and the final document was released by EPA in August 1986.

The first draft of the Staff Paper "*Review of the National Ambient Air Quality Standards for Ozone: Assessment of Scientific and Technical Information*" drew upon key findings and conclusions from the AQCD and was reviewed by CASAC at the April 21-22, 1986 public

meeting. At that meeting, the CASAC recommended that new information on prolonged O₃ exposure effects be considered in a second draft of the Staff Paper. The CASAC reviewed the resulting second draft and also heard a presentation of new and emerging information on the health and welfare effects of O₃ at a December 14-15, 1987 public review meeting. The CASAC concluded that sufficient new information existed to recommend incorporation of relevant new data into a supplement to the 1986 AQCD (O₃ Supplement) and into a third draft of the Staff Paper.

A draft O₃ Supplement, entitled “*Summary of Selected New Information on Effects of Ozone on Health and Vegetation: Draft Supplement to Air Quality Criteria for Ozone and Other Photochemical Oxidants,*” and the revised Staff Paper were made available to CASAC and to the public in November 1988. The O₃ Supplement assessed selected literature concerning exposure- and concentration-response relationships observed for health effects in humans and experimental animals and for vegetation effects that appeared in papers published or in-press from 1986 through early 1989. On December 14-15, 1988, the CASAC held a public meeting to review these documents and then sent the EPA Administrator a letter (dated May 1, 1989), which stated that the draft O₃ Supplement, the 1986 AQCD, and the draft Staff Paper “provide an adequate scientific basis for the EPA to retain or revise the primary and secondary standards of ozone.” The CASAC concluded (a) that it would be some time before sufficient new information on the health effects of multihour and chronic exposure to O₃ would be published in scientific journals to receive full peer review and, thus, be suitable for inclusion in a criteria document and (b) that such information could be considered in the next review of the O₃ NAAQS. A final version of the O₃ Supplement was published in 1992 (U.S. Environmental Protection Agency, 1992).

On October 22, 1991, the American Lung Association and other plaintiffs filed suit to compel the Agency to complete the review of the criteria and standards for O₃ in accordance with the CAA. The U.S. District Court for the Eastern District of New York subsequently issued an order requiring the EPA to announce its proposed decision on whether to revise the standards for O₃ by August 1, 1992 and to announce its final decision by March 1, 1993.

The proposed decision on O₃, which appeared in the Federal Register on August 10, 1992 (57 FR 35542), indicated that revision of the existing 1-h O₃ NAAQS was not appropriate at that time. A public hearing on this decision was held in Washington, DC on September 1, 1992; and

public comments were received through October 9, 1992. The final decision not to revise the 1-h NAAQS was published in the Federal Register on March 9, 1993 (58 FR 13008). However, that decision did not take into consideration a number of more recent studies on the health and welfare effects of O₃ that had been published since the last of the literature assessed in the O₃ Supplement (i.e., studies available through 1985 and into early 1986).

The Agency initiated consideration of such studies as part of the next congressionally-mandated periodic review of O₃ criteria and NAAQS. The new studies were assessed in revised draft O₃ AQCD chapters that were peer reviewed in July and September 1993 workshops, followed by public release of the O₃ AQCD First External Review Draft in February 1994 and CASAC review on July 20-21, 1994. Further drafts of the O₃ AQCD, revised in response to public comments and CASAC review, were reviewed by CASAC on March 21-25, 1995, and at a final CASAC review meeting on September 19-20, 1995. The scientific soundness of the revised O₃ AQCD was recognized by CASAC in a November 28, 1995 letter to the EPA Administrator; and the final O₃ AQCD was published in July 1996.

The first draft of the associated Staff Paper, *“Review of the National Ambient Air Quality Standards for Ozone: Assessment of Scientific and Technical Information,”* was also reviewed by CASAC at the March 21-22, 1995 public meeting. CASAC also reviewed subsequent drafts of the Staff Paper at public meetings on September 19-20, 1995 and March 21, 1996, with completion of CASAC review of the primary and secondary standard portions of the draft Staff Paper being communicated in letters to the EPA Administrator dated November 30, 1995 and April 4, 1996, respectively. The final O₃ Staff Paper was published in June 1996 (U.S. Environmental Protection Agency, 1996b).

On December 13, 1996, EPA published its proposed decision to revise the O₃ NAAQS (61 FR 65716). Extensive opportunities for public comment on the proposed decision, including several public hearings and two national satellite telecasts, were then provided by EPA; and EPA’s final decision to promulgate a new 8-h O₃ NAAQS (see Table 1-1) was published on July 18, 1997 (62 FR 38856).

Following promulgation of the new standards, numerous petitions for review of the standards were filed in the U.S. Court of Appeals for the District of Columbia Circuit (D.C. Circuit)². On May 14, 1999, the Court remanded the O₃ NAAQS to EPA, finding that

²*American Trucking Associations v. EPA*, No. 97-1441.

Section 109 of the CAA, as interpreted by EPA, effected an unconstitutional delegation of legislative authority³. In addition, the Court directed that, in responding to the remand, EPA should consider the potential beneficial health effects of O₃ pollution in shielding the public from the effects of solar ultraviolet (UV) radiation. On January 27, 2000, EPA petitioned the U.S. Supreme Court for certiorari on the constitutional issue (and two other issues), but did not request review of the D.C. Circuit ruling regarding the potential beneficial health effects of O₃. On February 27, 2001, the U.S. Supreme Court unanimously reversed the judgment of the D.C. Circuit on the constitutional issue (holding that section 109 of the CAA does not delegate legislative power to the EPA in contravention of the Constitution) and remanded the case to the D.C. Circuit to consider challenges to the O₃ NAAQS that had not been addressed by that Court's earlier decisions⁴. On March 26, 2002, the D.C. Circuit issued its final decision, finding that the 1997 O₃ NAAQS were “neither arbitrary nor capricious,” and denied the remaining petitions for review⁵.

On November 14, 2001, EPA proposed to respond to the Court's remand to consider the potential beneficial health effects of O₃ pollution in shielding the public from the effects of solar UV radiation by leaving the 1997 8-h NAAQS unchanged. Following a review of information in the record and the substantive comments received on the proposed response, EPA issued a final response to the remand, reaffirming the 8 h O₃ NAAQS (68 FR 614, January 6, 2003).

1.2 CURRENT OZONE CRITERIA AND NAAQS REVIEW

1.2.1 Key Milestones and Procedures for Document Preparation

It is important to note at the outset that development of the present O₃ AQCD included substantial external expert review and opportunities for public input through (a) public workshops involving the general scientific community, (b) iterative reviews of successive AQCD drafts by CASAC, and (c) comments from the public on successive drafts. Extensive external inputs received through such reviews help to ensure that the current periodic review of

³ *American Trucking Associations v. EPA*, 175 F.3d 1027 (D.C. Cir., 1999).

⁴ *Whitman v. American Trucking Associations*, 531 U.S. 457 (2001).

⁵ *American Trucking Associations v. EPA*, 283 F.3d 355, (D.C. Cir. 2002).

the O₃ standards is based on critical assessment of the latest available pertinent science as presented in this document and drawn upon in the associated Ozone Staff Paper.

The procedures for developing this revised O₃ AQCD build on experience derived from the other recent criteria document preparation efforts, with key milestones for development of this O₃ AQCD being listed in Table 1-2. Briefly, respective responsibilities for production of the document and key milestones are as follows. An NCEA-RTP Ozone Team created and implemented a project plan for developing the O₃ AQCD, taking into account input from individuals in other EPA program and policy offices identified as part of the EPA Ozone Work Group. The resulting plan, i.e., the “*Project Work Plan for Revised Air Criteria for Ozone and Related Photochemical Oxidants*” (November 2002), was discussed with CASAC in January 2003. Under the processes established in Sections 108 and 109 of the CAA, the EPA officially initiated the current criteria and NAAQS review by announcing the commencement of the review in the Federal Register (65 FR 57810, September, 2000) with a call for information. That Federal Register notice included (1) a request asking for recently available research information on O₃ that may not yet have been published and (2) a request for individuals with the appropriate type and level of expertise to contribute to the writing of O₃ AQCD materials to identify themselves. The specific authors of chapters or sections of the proposed document included both EPA and non-EPA scientific experts, who were selected on the basis of their expertise on the subject areas and their familiarity with the relevant literature. The project team defined critical issues and topics to be addressed by the authors and provided direction in order to focus on evaluation of those studies most clearly identified as important for standard setting. An ongoing literature search that was underway prior to initiation of work on this document continued throughout its preparation to identify pertinent O₃ literature published since early 1996.

As with other NAAQS reviews, critical assessment of relevant scientific information is presented in this updated O₃ AQCD. The main focus of this document is the evaluation and interpretation of pertinent atmospheric science information, air quality data, human exposure information, and health and welfare effects information newly published since that assessed in the 1996 O₃ AQCD. Draft versions of AQCD chapter materials were evaluated via expert peer-consultation workshop discussions (see Table 1-2) that focused on the selection of pertinent studies to be included in the chapters, the potential need for additional information to be added to the chapters, and the quality of the characterization and interpretation of the literature. The

Table 1-2. Key Milestones for Development of Revised Ozone Air Quality Criteria Document (O₃ AQCD)

<u>Major Milestones</u>	<u>Dates</u>
1. Literature Search	Ongoing
2. Federal Register Call for Information	September 2000
3. Draft Project Plan Available for Public Comment	Dec 2001 - March 2002
4. Revised Draft Project Plan Released for CASAC Consultation	December 2002
5. CASAC Consultation on Draft Project Work Plan	January 2003
6. Peer-Consultation Workshop on Draft Ecological Effects Materials	April 2003
7. Peer-Consultation Workshops on Draft Atmospheric Science/Exposure and Dosimetry/Health Chapters	July 2004
8. First External Review Draft of O ₃ AQCD	January 2005
9. Public Comment Period (90 days)	Feb - April 2005
10. CASAC Public Review Meeting (<i>First External Review Draft</i>)	May 4-5, 2005
11. Second External Review Draft of O ₃ AQCD	August 2005
12. Public Comment Period (90 days)	Sept - Nov 2005
13. CASAC Public Review Meeting (<i>Second External Review Draft</i>)	December 6-8, 2005
14. Final O ₃ AQCD	February 28, 2006

authors of the draft chapters then revised them on the basis of the workshop and/or other expert review comments⁶. These and other integrative materials were then incorporated into the First External Review Draft (January 2005) of this O₃ AQCD, which was made available for public comment and CASAC review, as indicated in Table 1-2.

Following review of the First External Review Draft at a May 4-5, 2005 CASAC meeting, EPA incorporated revisions into the draft O₃ AQCD in response to comments from CASAC and the public and made a Second External Review Draft (August, 2005) available for further public comment and CASAC review as shown in Table 1-2. More specifically, the Second External Review Draft underwent public comment during September-November, 2005, and was reviewed by CASAC at a December 6-8, 2005 public meeting. This final O₃ AQCD, completed by

⁶It should be noted that materials contributed by non-EPA authors were, at times, modified by EPA Ozone Team staff in response to internal and/or external review comments and that EPA is responsible for the ultimate content of this O₃ AQCD.

February 28, 2006, incorporates revisions made in response to public comments and CASAC reviews of the earlier draft AQCD materials. An electronic version of this document can be accessed via an EPA website at: www.epa.gov/ncea.

The EPA's Office of Air Quality Planning and Standards (OAQPS) staff is also preparing further draft O₃ Staff Paper materials which draw upon key information contained in this final O₃ AQCD. After review of that draft O₃ Staff Paper by the public and by CASAC, EPA will take public and CASAC comments into account in producing a Final Ozone Staff Paper. That Staff Paper, in final form, will present options for consideration by the Administrator of EPA regarding whether to retain or, if appropriate, to revise the O₃ NAAQS.

1.3 ORGANIZATIONAL STRUCTURE OF THE DOCUMENT

1.3.1 General Document Format

The general format used in preparing this O₃ AQCD is to open each new section for the updated document with concise summarization of key findings and conclusions from the previous 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996a). After presentation of such background information, the remainder of each section typically provides an updated discussion of newer literature and resulting key conclusions. In some cases where no new information is available, the summary of key findings and conclusions from the previous criteria document must suffice as the basis for current key conclusions. Increased emphasis is placed in the main chapters of this revised O₃ AQCD on interpretative evaluation and integration of evidence pertaining to a given topic than has been typical of previous EPA air quality criteria documents, with more detailed descriptions of individual studies being provided in a series of accompanying annexes.

A list of references published since completion of the 1996 criteria document was made available to the authors. The references were selected from information data base searches conducted by EPA. Additional references were added to the list (e.g., missed or recently published papers or "in press" publications) as work proceeded in creating draft document materials. As an aid in selecting pertinent new literature, the authors were also provided with a summary of issues that needed to be addressed in this revised O₃ AQCD. These issues were identified by NCEA-RTP Ozone Team members, by the EPA Ozone Work Group, and by

authors and reviewers of draft O₃ AQCD materials, and they were further expanded, as appropriate, based on public discussions, workshops, or other comments received by EPA in the course of development of this document.

1.3.2 Organization and Content of the Document

This revised AQCD for Ozone and Related Photochemical Oxidants critically assesses scientific information on the health and welfare effects associated with exposure to the concentrations of these pollutants in ambient air. The document does not provide a detailed literature review; but, rather, discusses cited references that reflect the current state of knowledge on the most relevant issues pertinent to the derivation of NAAQS for O₃ and/or related photochemical oxidants. Although emphasis is placed on discussion of health and welfare effects information, other scientific data are presented and evaluated in order to provide a better understanding of the nature, sources, distribution, measurement, and concentrations of O₃ and related photochemical oxidants in ambient air, as well as the measurement of population exposure to these pollutants.

The main focus of the scientific information discussed in the text comes from literature published since completion of the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996a). Emphasis is placed on studies conducted at or near O₃ concentrations found in ambient air. Other studies are included if they contain unique data, such as the documentation of a previously unreported effect or of a mechanism for an observed effect; or if they were multiple-concentration studies designed to elucidate exposure-response relationships. Generally, this is not an issue for human clinical or epidemiology studies. However, for animal toxicology studies, consideration is given mainly to those studies conducted at less than 1 ppm O₃. Key information from studies assessed in the previous O₃ AQCD and whose data impacted the derivation of the current NAAQS are briefly summarized in the text, along with specific citations to the previous document. Prior studies are also discussed if they (1) are open to reinterpretation in light of newer data, or (2) are potentially useful in deriving revised standards for O₃. Generally, only information that has undergone scientific peer review and has been published (or accepted for publication) through December 2004 is included in this draft document. A few particularly pertinent and important new studies published or accepted for publication beyond the end of 2004 are also considered.

This revised O₃ AQCD consists of three volumes. The first volume includes an Executive Summary and Conclusions, as well as Chapters 1 through 11 of the document. This introductory chapter (Chapter 1) presents background information on the purpose of the document, legislative requirements, and the history of past O₃ NAAQS regulatory actions, as well as an overview of the organization and content of the document. Chapter 2 provides information on the physics and chemistry of O₃ and related photochemical oxidants in the atmosphere. Chapter 3 covers tropospheric O₃ environmental concentrations, patterns, and exposures. The accompanying annexes to each of these background chapters are found in Volume II.

Health information pertinent to derivation of the primary O₃ NAAQS is then mainly covered in the next several chapters (Chapters 4 through 8). Chapter 4 discusses O₃ dosimetry aspects; and Chapters 5, 6, and 7 discuss animal toxicological studies, controlled-exposure studies of human health effects, and epidemiologic studies of ambient air exposure effects on human populations, respectively. Chapter 8 then provides an integrative and interpretive evaluation of key information relevant to O₃ exposure and health risks of most pertinence to the review of primary O₃ NAAQS. The annexes to these health-related chapters are found in Volume II.

The remaining three chapters of the document assess welfare effects information pertinent to the review of secondary O₃ NAAQS. More specifically, Chapter 9 deals with ecological and other environmental effects of O₃ and related photochemical oxidants. Chapter 10 assesses tropospheric O₃ involvement in climate change processes, including impacts on solar UV flux in Earth's lower atmosphere. Lastly, Chapter 11 discusses O₃ effects on man-made materials as a third type of welfare effect of potential concern. Annex materials related to welfare effects (especially vegetation/ecological effects) are contained in Volume III.

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2. PHYSICS AND CHEMISTRY OF OZONE IN THE ATMOSPHERE

2.1 INTRODUCTION

Ozone (O₃) and other oxidants, such as peroxy nitrates and hydrogen peroxide (H₂O₂) form in polluted areas by atmospheric reactions involving two main classes of precursor pollutants, volatile organic compounds (VOCs) and nitrogen oxides (NO_x). Carbon monoxide (CO) is also important for O₃ formation in polluted areas. Ozone is thus a secondary pollutant. The formation of O₃, other oxidants and oxidation products from these precursors is a complex, nonlinear function of many factors: the intensity and spectral distribution of sunlight; atmospheric mixing and processing on cloud and aerosol particles; the concentrations of the precursors in ambient air; and the rates of chemical reactions of the precursors. Information contained in this chapter and in greater detail in Annex AX2 describes these processes, numerical models that incorporate these processes to calculate O₃ concentrations, and techniques for measuring concentrations of ambient oxidants.

The atmosphere can be divided into several distinct vertical layers, based primarily on the major mechanisms by which they are heated and cooled. The lowest major layer is the troposphere, which extends from the earth's surface to about 8 km above polar regions and to about 16 km above tropical regions. The planetary boundary layer (PBL) is the lower sublayer of the troposphere, extending from the surface to about 1 or 2 km and is most strongly affected by surface conditions. The stratosphere extends from the tropopause, or the top of the troposphere, to about 50 km in altitude (Annex AX2.2.1). The emphasis in this chapter is placed on chemical and physical processes occurring in the troposphere, in particular in the PBL. The processes responsible for producing summertime O₃ episodes are fairly well understood, as discussed in the previous Air Quality Criteria Document for Ozone and Related Photochemical Oxidants or 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996). This chapter mainly considers topics for which there is substantial new information and on topics that form the basis for discussions in later chapters.

2.2 CHEMICAL PROCESSES INVOLVED IN OZONE FORMATION AND DESTRUCTION

Ozone occurs not only in polluted urban atmospheres but throughout the troposphere, even in remote areas of the globe. The same basic processes, involving sunlight-driven reactions of NO_x and VOCs contribute to O_3 formation throughout the troposphere. These processes also lead to the formation of other photochemical products, such as peroxyacetyl nitrate (PAN), nitric acid (HNO_3), and sulfuric acid (H_2SO_4), and to other compounds, such as formaldehyde (HCHO) and other carbonyl compounds, such as aldehydes and ketones.

The photochemical formation of O_3 in the troposphere proceeds through the oxidation of nitric oxide (NO) to nitrogen dioxide (NO_2) by organic (RO_2) or hydro-peroxy (HO_2) radicals. The photolysis of NO_2 yields nitric oxide (NO) and a ground-state oxygen atom, $\text{O}(^3\text{P})$, which then reacts with molecular oxygen to form O_3 . Free radicals oxidizing NO to NO_2 are formed during the oxidation of VOCs (Annex AX2.2.2).

The term VOC refers to all carbon-containing gas-phase compounds in the atmosphere, both biogenic and anthropogenic in origin, excluding CO and CO_2 . Classes of organic compounds important for the photochemical formation of O_3 include alkanes, alkenes, aromatic hydrocarbons, carbonyl compounds (e.g., aldehydes and ketones), alcohols, organic peroxides, and halogenated organic compounds (e.g., alkyl halides). This array of compounds encompasses a wide range of chemical properties and lifetimes: isoprene has an atmospheric lifetime of approximately an hour, whereas methane has an atmospheric lifetime of about a decade.

In urban areas, compounds representing all classes of VOCs and CO are important for O_3 formation. In nonurban vegetated areas, biogenic VOCs emitted from vegetation tend to be the most important. In the remote troposphere, CH_4 and CO are the main carbon-containing precursors to O_3 formation. CO also can play an important role in O_3 formation in urban areas. The oxidation of VOCs is initiated mainly by reaction with hydroxyl (OH) radicals. The primary source of OH radicals in the atmosphere is the reaction of electronically excited O atoms, $\text{O}(^1\text{D})$, with water vapor. $\text{O}(^1\text{D})$ is produced by the photolysis of O_3 in the Hartley bands. In polluted areas, the photolysis of aldehydes (e.g., HCHO), nitrous acid (HONO) and hydrogen peroxide (H_2O_2) can also be significant sources of OH or HO_2 radicals that can rapidly be converted to OH (Eisele et al., 1997). Ozone can oxidize alkenes; and, at night, when they are most abundant, NO_3 radicals also oxidize alkenes. In coastal environments and other selected

environments, atomic Cl and Br radicals can also initiate the oxidation of VOCs (Annex AX2.2.3).

There are a large number of oxidized nitrogen containing compounds in the atmosphere including NO, NO₂, NO₃, HNO₂, HNO₃, N₂O₅, HNO₄, PAN and its homologues, other organic nitrates and particulate nitrate. Collectively these species are referred to as NO_y. Oxidized nitrogen compounds are emitted to the atmosphere mainly as NO which rapidly interconverts with NO₂ and so NO and NO₂ are often “lumped” together into their own group or family, or NO_x. NO_x can be oxidized to reservoir and termination species (PAN and its homologues, organic nitrates, HNO₃, HNO₄ and particulate nitrate). These reservoir and termination species are referred to as NO_z. The major reactions involving interconversions of oxidized nitrogen species are discussed in Annex AX2.2.4.

The photochemical cycles by which the oxidation of hydrocarbons leads to O₃ production are best understood by considering the oxidation of methane, structurally the simplest VOC. The CH₄ oxidation cycle serves as a model for the chemistry of the relatively clean or unpolluted troposphere (although this is a simplification because vegetation releases large quantities of complex VOCs, such as isoprene, into the atmosphere). In the polluted atmosphere, the underlying chemical principles are the same, as discussed in Annex AX2.2.5. The conversion of NO to NO₂ occurring with the oxidation of VOCs is accompanied by the production of O₃ and the efficient regeneration of the OH radical, which in turn can react with other VOCs. A schematic overview showing the major processes involved in O₃ production and loss in the troposphere and stratosphere is given in Figure 2-1.

The oxidation of alkanes and alkenes in the atmosphere has been treated in depth in 1996 O₃ AQCD and is updated in Annexes AX2.2.6 and AX2.2.7. In contrast to simple hydrocarbons containing one or two carbon atoms, detailed kinetic information about the gas phase oxidation pathways of many anthropogenic hydrocarbons (e.g., aromatic compounds such as benzene and toluene), biogenic hydrocarbons (e.g., isoprene, the monoterpenes), and their intermediate oxidation products (e.g., epoxides, nitrates, and carbonyl compounds) is lacking. Reaction with OH radicals represents the major loss process for alkanes. Reaction with chlorine atoms is an additional sink for alkanes. Stable products of alkane photooxidation are known to include carbonyl compounds, alkyl nitrates, and *d*-hydroxycarbonyls. Major uncertainties in the atmospheric chemistry of the alkanes concern the chemistry of alkyl nitrate formation; these

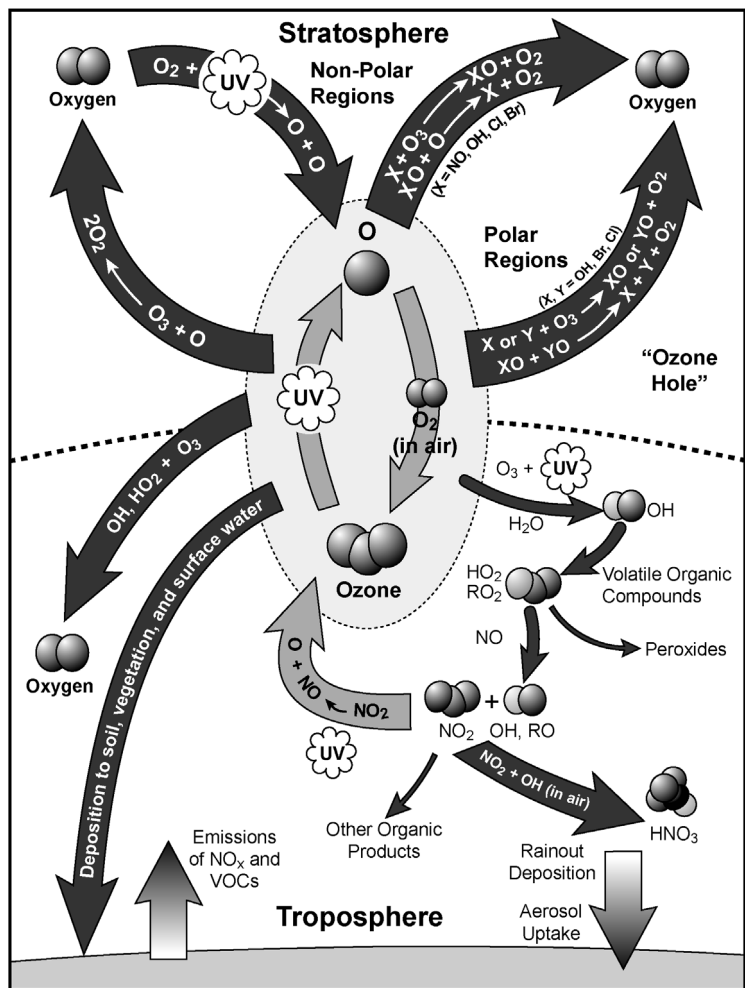


Figure 2-1. Schematic overview of O_3 photochemistry in the stratosphere and troposphere.

uncertainties affect the amount of NO -to- NO_2 conversion occurring and, hence, the amounts of O_3 formed during photochemical degradation of the alkanes.

The reaction of OH radicals with aldehydes produced during the oxidation of alkanes forms acyl ($R'CO$) radicals, and acyl peroxy radicals ($R'C(O)-O_2$) are formed by the further addition of O_2 . As an example, the oxidation of ethane (C_2H_5-H) yields acetaldehyde (CH_3-CHO). The reaction of CH_3-CHO with OH radicals yields acetyl radicals (CH_3-CO). The acetyl radicals will then participate with O_2 in a termolecular recombination reaction to form acetyl peroxy radicals, which can then react with NO to form $CH_3 + CO_2$ or they can react with

NO₂ to form PAN. PAN acts as a temporary reservoir for NO₂. Upon the thermal decomposition of PAN, either locally or elsewhere, NO₂ is released to participate in the O₃ formation process again.

Alkenes react in ambient air with OH, NO₃, and Cl radicals and with O₃. All of these reactions are important atmospheric transformation processes, and all proceed by initial addition to the >C=C< bonds. Products of alkene photooxidation include carbonyl compounds, hydroxynitrates and nitratocarbonyls, and decomposition products from the energy-rich biradicals formed in alkene-O₃ reactions. Major uncertainties in the atmospheric chemistry of the alkenes concern the products and mechanisms of their reactions with O₃, especially the yields of free radicals that participate in O₃ formation. Examples of oxidation mechanisms of complex alkanes and alkenes can be found in comprehensive texts such as Seinfeld and Pandis (1998).

The oxidation of aromatic hydrocarbons constitutes an important component of the chemistry of O₃ formation in urban atmospheres (Annex AX2.2.8). Virtually all of the important aromatic hydrocarbon precursors emitted in urban atmospheres are lost through reaction with the hydroxyl radical. Loss rates for these compounds vary from slow (i.e., benzene) to moderate (e.g., toluene), to very rapid (e.g., xylene and trimethylbenzene isomers). These loss rates are very well understood at room temperature and atmospheric pressure, and numerous experiments have been conducted that verify this. However, the mechanism for the oxidation of aromatic hydrocarbons following reaction with OH is poorly understood, as evident from the poor mass balance of the reaction products. The mechanism for the oxidation of toluene has been studied most thoroughly, and there is general agreement on the initial steps in the mechanism. However, at present there is no promising approach for resolving the remaining issues concerning the later steps. The oxidation of aromatic hydrocarbons also leads to particle formation which could remove gas-phase constituents that participate in O₃ formation. What is known of the chemistry of secondary organic aerosol formation from gaseous precursors was summarized in the latest PM AQCD (U.S. Environmental Protection Agency, 2004).

The reactions of oxygenated VOCs are also important components of O₃ formation (Annex AX2.2.9). They may be produced either by the oxidation of hydrocarbons or they may be present in ambient air as the result of direct emissions. For example, motor vehicles and some industrial processes emit formaldehyde and vegetation emits methanol.

As much as 30% of the carbon in hydrocarbons in many urban areas is in the form of aromatic compounds. Yet, mass balance analyses performed on irradiated smog chamber mixtures of aromatic hydrocarbons indicate that only about one-half of the carbon is in the form of compounds that can be identified. The situation is not much better for some smaller anthropogenic hydrocarbons. For example, only about 60% of the initial carbon can be accounted for in the OH initiated oxidation of 1,3-butadiene. About two-thirds of the initial carbon can be identified in product analyses of isoprene oxidation. Adequate analytical techniques needed to identify and quantify key intermediate species are not available for many compounds. In addition, methods to synthesize many of the suspected intermediate compounds are not available so that laboratory studies of their reaction kinetics cannot be performed. Similar considerations apply to the oxidation of biogenic hydrocarbons besides isoprene.

In addition to reactions occurring in the gas phase, reactions occurring on the surfaces of or within cloud droplets and airborne particles also occur. Their collective surface area is huge, implying that collisions with gas phase species occur on very short time scales. In addition to hydrometeors (e.g., cloud and fog droplets and snow and ice crystals) there are also potential reactions involving atmospheric particles of varying composition (e.g., wet [deliquesced] inorganic particles, mineral dust, carbon chain agglomerates and organic carbon particles) to consider. Most of the well-established multiphase reactions tend to reduce the rate of O₃ formation in the polluted troposphere. Removal of HO_x and NO_x onto hydrated particles will reduce the production of O₃. However, the photolysis of HONO formed in reactions such as these can increase the production of O₃. The reactions of Br and Cl containing radicals deplete O₃ in selected environments such as the Arctic during spring, the tropical marine boundary layer and inland salt flats and salt lakes. Direct reactions of O₃ and atmospheric particles appear to be too slow to reduce O₃ formation significantly at typical ambient PM levels. In addition, the oxidation of hydrocarbons by Cl radicals could lead to the rapid formation of peroxy radicals and higher rates of O₃ production in selected coastal environments. It should be stressed that knowledge of multiphase processes is still evolving and there are still many questions that remain to be answered as outlined in Annex AX2.2.10.

The oxidants, other than O₃, that are formed from the chemistry described above could exert effects on human health and perhaps also on vegetation. Gas phase oxidants include PAN, H₂O₂ and CH₃OOH and other organic hydroperoxides (Annex AX2.2). In addition to

transfer from the gas phase, oxidants can be formed by photochemical reactions occurring in particles (Annex 2.2.10.6). However, the pathways leading to the formation of oxidants in the particle phase are not as well understood as they are in the gas phase. However, it is to be expected that pathways leading to the formation of gas phase oxidants and secondary organic aerosols are linked to some degree. In addition, the reaction of O₃ with isoprene and other biogenic hydrocarbons may also form oxidants in particles.

Reactions of O₃ with monoterpenes have been shown to produce oxidants in the aerosol phase. Docherty et al. (2005) found evidence for the substantial production of organic hydroperoxides in secondary organic aerosol (SOA) resulting from the reaction of monoterpenes with O₃. Analysis of the SOA formed in their environmental chamber indicated that the SOA was mainly organic hydroperoxides. In particular, they obtained yields of 47% and 85% of organic peroxides from the oxidation of α- and β-pinene. The hydroperoxides then react with aldehydes in particles to form peroxyhemiacetals, which can either rearrange to form other compounds such as alcohols and acids or revert back to the hydroperoxides. The aldehydes are also produced in large measure during the ozonolysis of the monoterpenes. Monoterpenes also react with OH radicals resulting, however, in the production of more lower molecular weight products than in their reaction with O₃. Bonn et al. (2004) estimated that hydroperoxides lead to 63% of global SOA formation from the oxidation of terpenes. The oxidation of anthropogenic aromatic hydrocarbons by OH radicals may also produce organic hydroperoxides in SOA (Johnson et al., 2004). Although the results of chamber and modeling studies indicate substantial production of organic hydroperoxides, it should be noted that data for organic hydroperoxides in ambient aerosol samples are sparse.

Ozone chemical reactions that occur indoors are analogous to those occurring in ambient air. In the indoor environment, O₃ reacts with unsaturated VOCs, primarily terpenes or terpene-related compounds from cleaning products, air fresheners, and wood products. The reactions are dependent on the O₃ indoor concentration, the indoor temperature and, in most cases, the air exchange rate/ventilation rate. Some of the reaction products may more negatively impact human health and artifacts in the indoor environment than their precursors (Wolkoff et al., 1999; Wilkins et al., 2001; Weschler et al., 1992; Weschler and Shields, 1997; Rohr et al., 2002; Nøjgaard et al., 2005). Primary reaction products are Criegee biradicals, nitrate radicals, and peroxyacetyl radicals. Secondary reaction products are hydroxy, alkyl, alkylperoxy,

hydroperoxy, and alkoxy radicals. Reactions with alkenes can produce aldehydes, ketones, and organic acids (Weschler and Shields, 2000; Weschler et al., 1992).

2.3 METEOROLOGICAL PROCESSES AFFECTING OZONE

Since the 1996 O₃ AQCD, substantial new information about transport processes has become available from numerical models, field experiments and satellite-based observations. Ozone is produced naturally by photochemical reactions in the stratosphere, as shown in Figure 2-1. Some of this O₃ is transported downward into the troposphere throughout the year, with maximum contributions during late winter and early spring mainly in a process known as tropopause folding. Figure 2-2a shows a synoptic situation associated with a tropopause folding event. A vertical cross section taken through the atmosphere from a to a' is shown in Figure 2-2b. In this figure, the tropopause fold is shown folding downward above and slightly behind the surface cold front, bringing stratospheric air with it. Although the tropopause is drawn with a solid line, it should not be taken to mean that it is a material surface through which there is no exchange. Rather these folds should be thought of as regions in which mixing of tropospheric and stratospheric air is occurring (Shapiro, 1980). This imported stratospheric air contributes to the natural background of O₃ in the troposphere, especially in the free troposphere. It should be noted that there is considerable uncertainty in the magnitude and distribution of this potentially important source of tropospheric O₃. Stratospheric intrusions that reach the surface are rare. Much more common are intrusions which penetrate only to the middle and upper troposphere. However, O₃ transported to the upper and middle troposphere can still affect surface concentrations through various exchange mechanisms that mix air from the free troposphere with air in the planetary boundary layer. Substantial photochemical production of O₃ in the troposphere also begins in late winter and early spring; therefore, it cannot be assumed that O₃ present at these times is only stratospheric in origin. The basic atmospheric dynamics and thermodynamics of stratospheric-tropospheric exchange are outlined in Annex AX2.3.1.

Our understanding of the meteorological processes associated with summertime O₃ episodes remains basically the same as outlined in the 1996 O₃ AQCD. Major episodes of high O₃ concentrations in the eastern United States and in Europe are associated with slow moving, high pressure systems. High pressure systems during the warmer seasons are associated with the

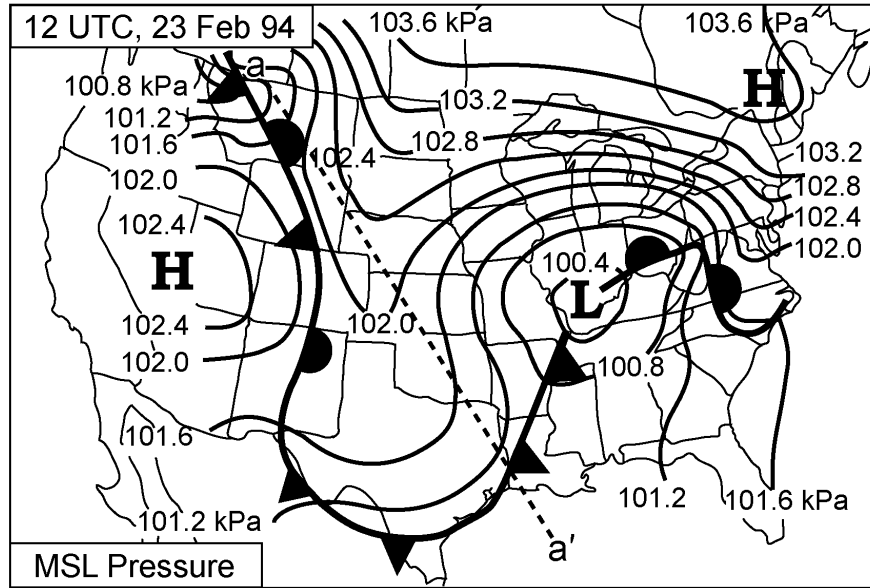


Figure 2-2a. Surface weather chart showing sea level (MSL) pressure (kPa), and surface fronts.

Source: Stull (2000).

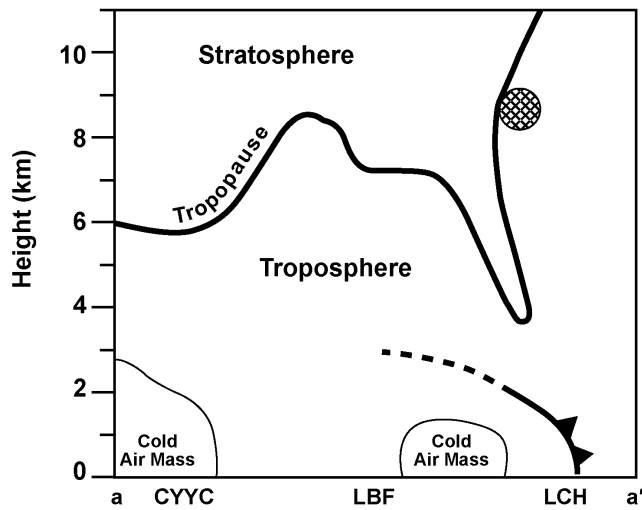


Figure 2-2b. Vertical cross section along dashed line (a-a') from northwest to the southeast (CYXC = Calgary, Alberta; LBF = North Platte, NB; LCH = Lake Charles, LA). The approximate location of the jet stream core is indicated by the hatched area. The position of the surface front is indicated by the cold-frontal symbols and the frontal inversion top by the dashed line. Note: This is 12 h later than the situations shown in Figure 2-2a.

Source: Adapted from Stull (2000).

sinking of air, resulting in warm, generally cloudless skies, with light winds. The sinking of air results in the development of stable conditions near the surface which inhibit or reduce the vertical mixing of O₃ precursors. The combination of inhibited vertical mixing and light winds minimizes the dispersal of pollutants emitted in urban areas, allowing their concentrations to build up. Photochemical activity involving these precursors is enhanced because of higher temperatures and the availability of sunlight. In the eastern United States, high O₃ concentrations during a large scale episode can extend over hundreds of thousands of square kilometers for several days. These conditions have been described in greater detail in the 1996 O₃ AQCD. The transport of pollutants downwind of major urban centers is characterized by the development of urban plumes. However, the presence of mountain barriers limits mixing (as in Los Angeles and Mexico City) and results in a higher frequency and duration of days with high O₃ concentrations. Ozone concentrations in southern urban areas (such as Houston, TX and Atlanta, GA) tend to decrease with increasing wind speed. In northern cities (such as Chicago, IL; New York, NY; Boston, MA; and Portland, ME), the average O₃ concentrations over the metropolitan areas increase with wind speed, indicating that transport of O₃ and its precursors from upwind areas is important (Husar and Renard, 1998; Schichtel and Husar, 2001).

Ozone and other secondary pollutants are determined by meteorological and chemical processes extending typically over spatial scales of several hundred kilometers (e.g., Civerolo et al., 2003; Rao et al., 2003). An analysis of the output of regional model studies conducted by Kasibhatla and Chameides (2000) suggests that O₃ can be transported over a few thousand kilometers in the upper boundary layer of the eastern half of the United States during some O₃ episodes. Convection is capable of transporting O₃ and its precursors vertically through the troposphere, as shown in Annex AX2.3.2. Nocturnal low level jets (LLJs) can also transport pollutants hundreds of kilometers (Annex AX2.3.3). Schematic diagrams showing the atmospheric conditions during the formation of LLJs and the regions in which they are most prevalent are given in Figures 2-3 and 2-4. Such LLJs have also been observed off the coast of California. Turbulence associated with LLJs can bring these pollutants to the surface and result in secondary O₃ maxima during the early morning in many locations (Corsmeier et al., 1997).

Aircraft observations indicate that there can be substantial differences in mixing ratios of key species between the surface and the atmosphere above (Fehsenfeld et al., 1996; Berkowitz and Shaw, 1997). In particular, mixing ratios of O₃ can be higher in the lower free troposphere

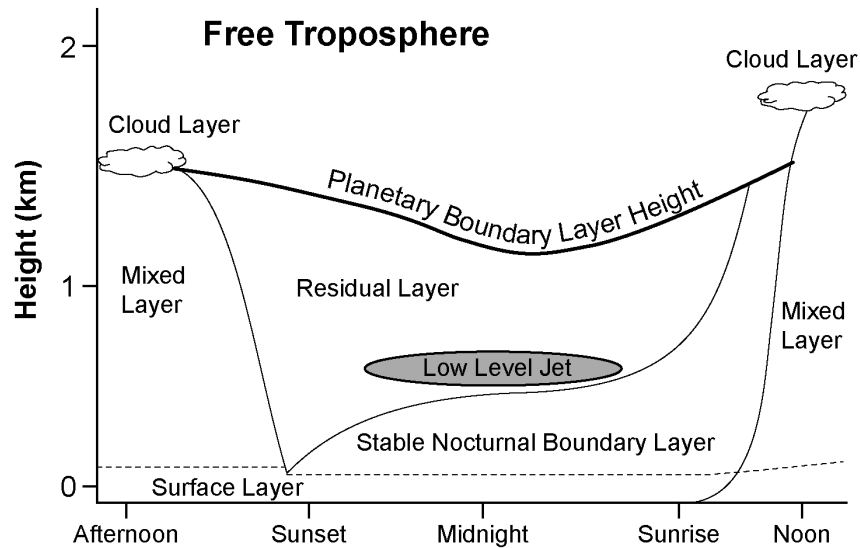


Figure 2-3. The diurnal evolution of the planetary boundary layer (PBL) while high pressure prevails over land. Three major layers exist (not including the surface layer): a turbulent mixed layer; a less turbulent residual layer which contains former mixed layer air; and a nocturnal, stable boundary layer that is characterized by periods of sporadic turbulence.

Source: Adapted from Figures 1.7 and 1.12 of Stull (1999).

(aloft) than in the planetary boundary layer (PBL) during multiday O_3 episodes (Taubmann et al., 2004, 2005). These conditions are illustrated schematically in Figure 2-5. Convective processes and small scale turbulence transport O_3 and other pollutants both upward and downward throughout the planetary boundary layer and the free troposphere. Ozone and its precursors can be transported vertically by convection into the upper part of the mixed layer on one day, then transported overnight as a layer of elevated mixing ratios, and then entrained into a growing convective boundary layer downwind and brought back down to the surface. High O_3 concentrations showing large diurnal variations at the surface in southern New England were associated with the presence of such layers (Berkowitz et al., 1998). Because of wind shear, winds several hundred meters above the ground can bring pollutants from the west, even though surface winds are from the southwest during periods of high O_3 in the eastern United States (Blumenthal et al., 1997). These considerations suggest that, in many areas of the United States, O_3 formation involves processes occurring over hundreds if not thousands of square kilometers.

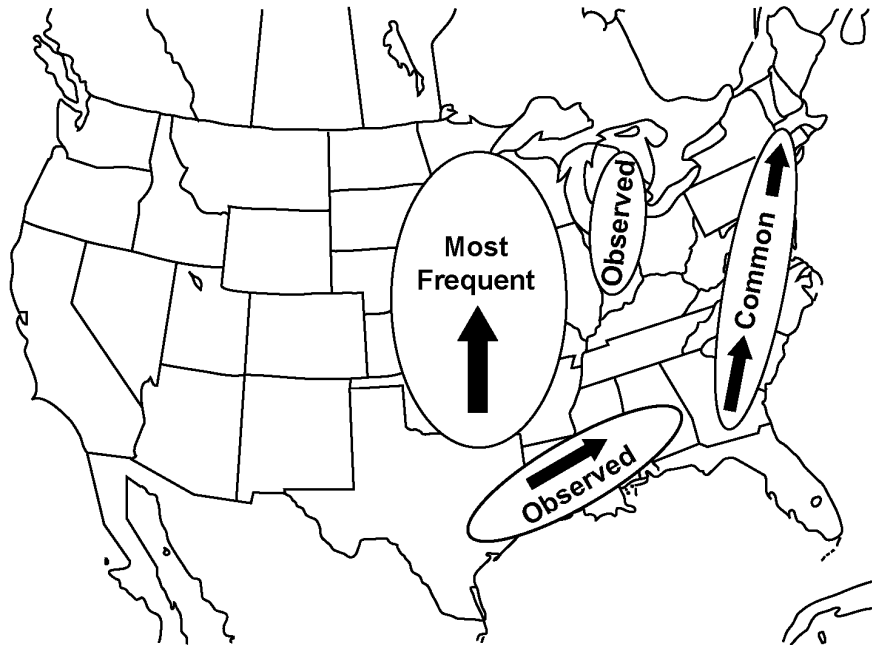


Figure 2-4. Locations of low level jet occurrences in decreasing order of prevalence (most frequent, common, observed). These locations are based on 2-years radiosonde data obtained over limited areas. With better data coverage, other low level jets may well be observed elsewhere in the United States.

Source: Bonner (1968).

Although the vast majority of measurements are made near the Earth's surface, there is substantial photochemistry and transport of O_3 occurring above the boundary layer in the free troposphere. In the free troposphere, pollutants are chemically more stable and can be transported over much longer distances and O_3 is produced more efficiently than in the planetary boundary layer. Results from the Atmosphere/Ocean Chemistry Experiment (AEROCE) indicated that springtime maxima for surface O_3 over the western North Atlantic Ocean result from tropopause folding in close proximity to convective clouds (Annex AX2.3.4). The convection lifts O_3 and its precursors to the free troposphere where they mix with O_3 from the stratosphere and the mixture is transported eastward. Results from the North Atlantic Regional Experiment (Annex AX2.3.4) indicate that summertime air is transported along the East Coast northeastward and upward ahead of cold fronts. New England and the Maritime Provinces of Canada receive substantial amounts of O_3 and other pollutants through this mechanism.

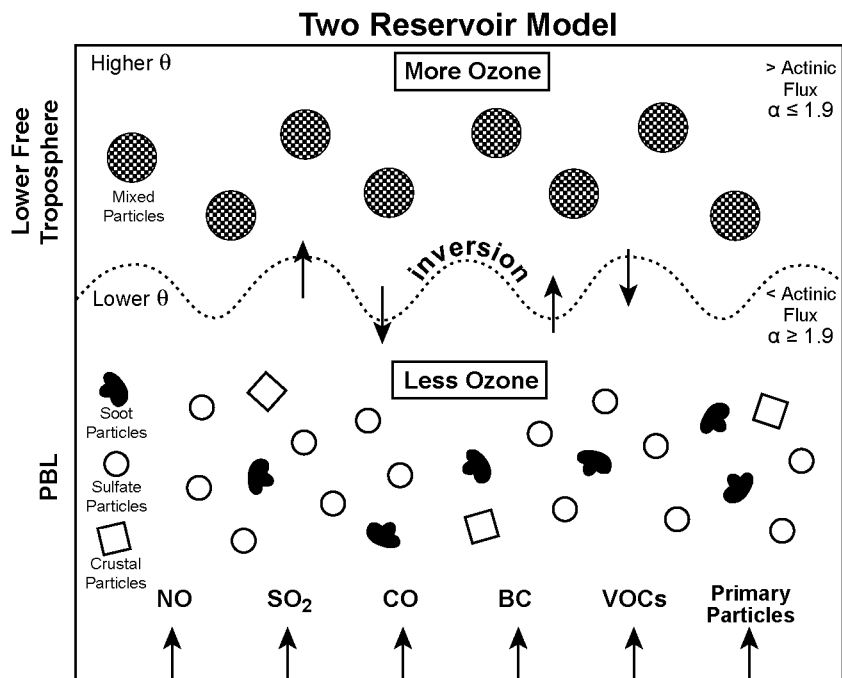


Figure 2-5. Conceptual two-reservoir model showing conditions in the planetary boundary layer (PBL) and in the lower free troposphere during a multiday O₃ episode. The dotted line represents the top of PBL. Emissions occur in the PBL, where small, unmixed black carbon, sulfate, and crustal particles in the PM_{2.5} size range are also shown. Ozone concentrations as well as potential temperature (θ) and actinic flux are lower in the PBL than in the lower free troposphere, while relative humidity and the Angstrom exponent for aerosol scattering (α) are higher. Larger, internally mixed sulfate and carbonaceous particles (still in the PM_{2.5} size range) and more O₃ exist in the lower free troposphere.

Source: Taubman et al. (2004, 2005).

Pollutants transported in this way can then be entrained in stronger and more stable westerly winds aloft and can travel across the North Atlantic Ocean. The pollutants can then be brought to the surface by subsidence in high pressure systems (typically behind the cold front in advance of the one mentioned above). Thus, pollutants from North America can be brought down either over the North Atlantic Ocean or in Europe. Pollutants can be transported across the North Pacific Ocean from Asia to North America in a similar way. Behind an advancing cold front, cold and dry stratospheric air is also being transported downward and southward. Stratospheric

constituents and tropospheric constituents can then mix by small-scale turbulent exchange processes. The results of these studies suggest that the mechanisms involved in the long-range transport of O₃ and its precursors are closely tied to the processes involved in stratospheric-tropospheric exchange. Land-sea breezes affect the concentration and dispersal of pollutants in coastal zone cities (Annex AX2.5).

The local rate of O₃ formation depends on atmospheric conditions such as the availability of solar ultraviolet radiation capable of initiating photolysis reactions, air temperatures and the concentrations of chemical precursors (Annex AX2.3.6). The dependence of daily maximum 8-h O₃ concentrations on daily maximum temperature is illustrated in Figure 2-6 for the Baltimore, MD area. As can be seen, O₃ concentrations tend to increase with temperature ($r = 0.74$). However, this trend is absent in data from Phoenix, AZ as can be seen in Figure 2-7 ($r = 0.14$). These figures show that relations of O₃ to precursor variables are location-specific and relations observed in one area cannot be readily extrapolated to another. Factors that may be responsible for the differences in O₃ behavior in the two areas are discussed in Section AX2.3.6.

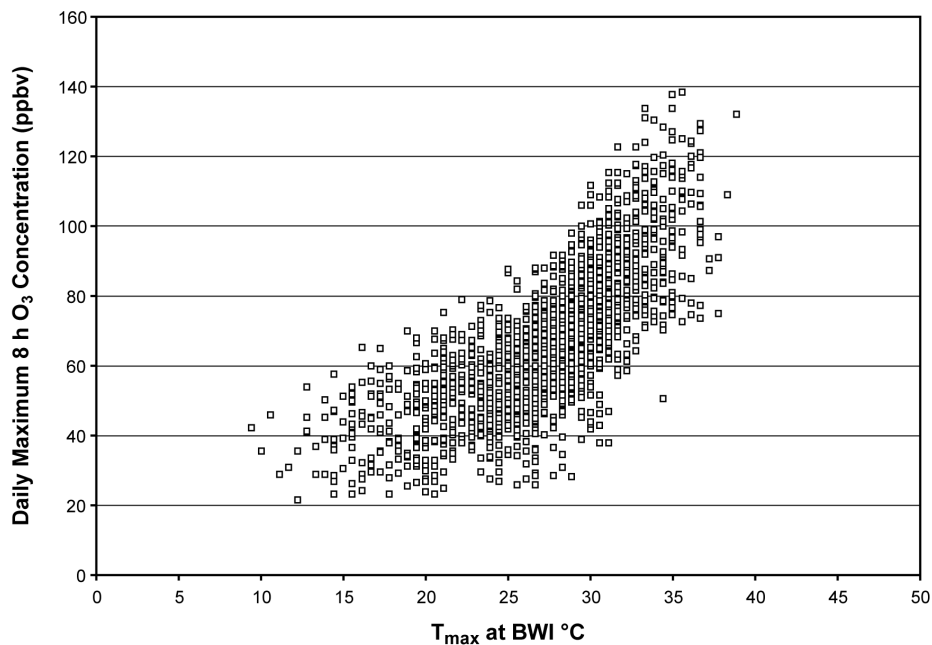


Figure 2-6. A scatter plot of daily maximum 8-h average O₃ concentrations versus daily maximum temperature for May through September 1994 to 2004 in the Baltimore, MD Air Quality Forecast Area.

Source: Piety (2005).

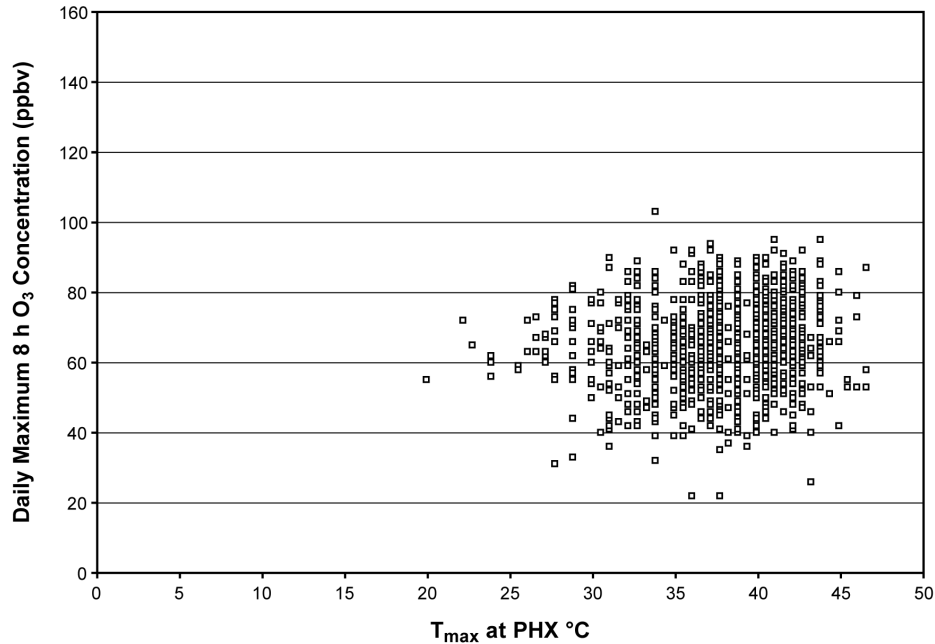


Figure 2-7. A scatter plot of daily maximum 8-h average O₃ concentrations versus daily maximum temperature for May through September 1996 to 2004 at sites downwind of Phoenix, AZ.

Source: Piety (2005).

2.4 RELATIONS OF OZONE TO ITS PRECURSORS

Rather than varying directly with emissions of its precursors, O₃ changes in a nonlinear fashion with the concentrations of its precursors (Annex AX2.4). At the low NO_x concentrations found in most environments, ranging from remote continental areas to rural and suburban areas downwind of urban centers (low - NO_x regime), the net production of O₃ increases with increasing NO_x. At the high NO_x concentrations found in downtown metropolitan areas, especially near busy streets and roadways and in power plant plumes, there is scavenging (titration) of O₃ by reaction with NO (high - NO_x regime). In between these two regimes, there is a transition stage in which O₃ shows only a weak dependence on NO_x concentrations. In the high - NO_x regime, NO₂ scavenges OH radicals which would otherwise oxidize VOCs to produce peroxy radicals, which in turn would oxidize NO to NO₂. In this regime, O₃ production is limited by the availability of free radicals. The production of free radicals is in turn limited by the availability of solar UV radiation capable of photolyzing O₃ (in the Hartley bands) or

aldehydes and/or by the abundance of VOCs whose oxidation produce more radicals than they consume. In the low- NO_x regime, the overall effect of the oxidation of VOCs is to generate (or at least not consume) free radicals, and O_3 production varies directly with NO_x . There are a number of ways to refer to the chemistry in these two chemical regimes. Sometimes the terms VOC-limited and NO_x -limited are used. However, there are difficulties with this usage because (1) VOC measurements are not as abundant as they are for nitrogen oxides, (2) rate coefficients for reaction of individual VOCs with free radicals vary over an extremely wide range, and (3) consideration is not given to CO nor to reactions that can produce free radicals without involving VOCs. The terms NO_x -limited and NO_x -saturated (e.g., Jaeglé et al., 2001) will be used wherever possible to more adequately describe these two regimes. However, the terminology used in original articles will also be used here.

The chemistry of OH radicals, which are responsible for initiating the oxidation of hydrocarbons, shows behavior similar to that for O_3 with respect to NO_x concentrations (Hameed et al., 1979; Pinto et al., 1993; Poppe et al., 1993; Zimmerman and Poppe, 1993). These considerations introduce a high degree of uncertainty into attempts to relate changes in O_3 concentrations to emissions of precursors. There are no definitive rules governing the levels of NO_x at which the transition from NO_x -limited to NO_x -saturated conditions occurs. The transition between these two regimes is highly spatially and temporally dependent and depends also on the nature and abundance of the hydrocarbons that are present.

Trainer et al. (1993) and Olszyna et al. (1994) have shown that O_3 and NO_y are highly correlated in rural areas in the eastern United States. Trainer et al. (1993) also showed that O_3 levels correlate even better with NO_z than with NO_y , as may be expected because NO_z represents the amount of NO_x that has been oxidized, forming O_3 in the process. NO_z is equal to the difference between measured total reactive nitrogen (NO_y) and NO_x and represents the summed products of the oxidation of NO_x . NO_z is composed mainly of HNO_3 , PAN and other organic nitrates, particulate nitrate, and HNO_4 .

Trainer et al. (1993) also suggested that the slope of the regression line between O_3 and NO_z can be used to estimate the rate of O_3 production per NO_x oxidized (also known as the O_3 production efficiency, or OPE). Ryerson et al. (1998, 2001) used measured correlations between O_3 and NO_z to identify different rates of O_3 production in plumes from large point sources. A number of studies in the planetary boundary layer over the continental United States

have found that the OPE ranges typically from one to nearly ten. However, it may be higher in the upper troposphere and in certain areas, such as the Houston-Galveston area. Observations indicate that the OPE depends mainly on the abundance of NO_x .

Various techniques have been proposed to use ambient NO_x and VOC measurements to derive information about the dependence of O_3 production on their concentrations. For example, it has been suggested that O_3 formation in individual urban areas could be understood in terms of measurements of ambient NO_x and VOC concentrations during the early morning (e.g., National Research Council, 1991). In this approach, the ratio of summed (unweighted) VOC to NO_x is used to determine whether conditions were NO_x -limited or VOC limited. This procedure is inadequate because it omits many factors that are important for O_3 production such as the impact of biogenic VOCs (which are typically not present in urban centers during early morning); important differences in the ability of individual VOCs to generate free radicals (rather than just total VOC) and other differences in O_3 forming potential for individual VOCs (Carter, 1995); and changes in the VOC to NO_x ratio due to photochemical reactions and deposition as air moves downwind from urban areas (Milford et al., 1994).

Photochemical production of O_3 generally occurs simultaneously with the production of various other species such as nitric acid (HNO_3), organic nitrates, and other oxidants such as hydrogen peroxide. The relative rate of production of O_3 and other species varies depending on photochemical conditions, and can be used to provide information about O_3 -precursor sensitivity. Sillman (1995) and Sillman and He (2002) identified several secondary reaction products that show different correlation patterns for NO_x -limited and NO_x -saturated conditions. The most important correlations are for O_3 versus NO_y , O_3 versus NO_z , O_3 versus HNO_3 , and H_2O_2 versus HNO_3 . The correlations between O_3 and NO_y , and O_3 and NO_z are especially important because measurements of NO_y and NO_x are more widely available than for VOCs. Measured O_3 versus NO_z (Figure 2-8) shows distinctly different patterns in different locations. In rural areas and in urban areas such as Nashville, TN, O_3 is highly correlated with NO_z . By contrast, in Los Angeles, CA, O_3 is not as highly correlated with NO_z , and the rate of increase of O_3 with NO_z is lower and the O_3 concentrations for a given NO_z value are generally lower. The different O_3 versus NO_z relations in Nashville, TN and Los Angeles, CA reflects the difference between NO_x -limited conditions in Nashville versus an approach to NO_x -saturated conditions in Los Angeles.

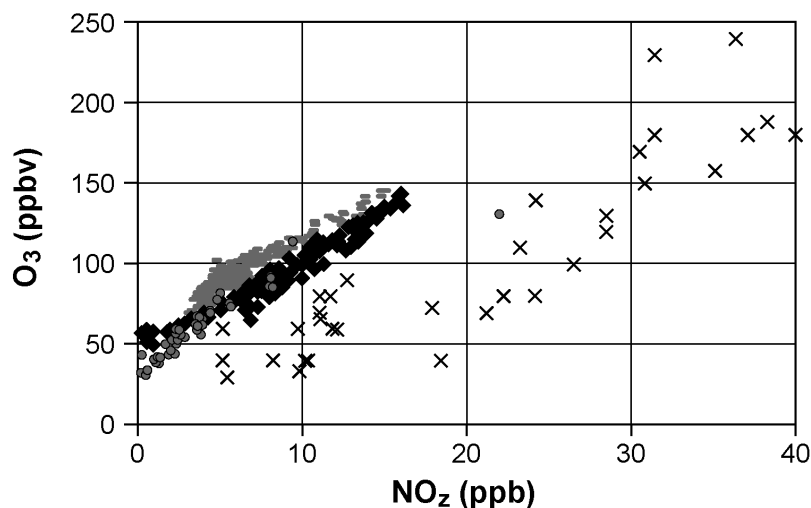


Figure 2-8. Measured values of O₃ and NO_z (NO_y – NO_x) during the afternoon at rural sites in the eastern United States (grey circles) and in urban areas and urban plumes associated with Nashville, TN (gray dashes); Paris, France (black diamonds); and Los Angeles CA (Xs).

Sources: Trainer et al. (1993), Sillman et al. (1997, 1998), Sillman and He (2002).

The difference between NO_x-limited and NO_x-saturated regimes is also reflected in measurements of hydrogen peroxide (H₂O₂). Hydrogen peroxide production is highly sensitive to the abundance of free radicals and is thus favored in the NO_x-limited regime. Measurements in the rural eastern United States (Jacob et al., 1995), Nashville, TN (Sillman et al., 1998), and Los Angeles, CA (Sakugawa and Kaplan, 1989), show large differences in H₂O₂ concentrations between likely NO_x-limited and NO_x-saturated locations.

2.5 THE ROLE OF CHEMISTRY-TRANSPORT MODELS IN UNDERSTANDING ATMOSPHERIC OZONE

Chemistry-transport models (CTMs) are used to improve understanding of atmospheric chemical processes and to develop control strategies (Annex AX2.5). The main components of a CTM are summarized in Figure 2-9. Models such as the CMAQ (Community Model for Air Quality) system incorporate numerical algorithms describing the processes shown in Figure 2-9. Also shown in Figure 2-9 is the meteorological model used to provide the inputs for calculating

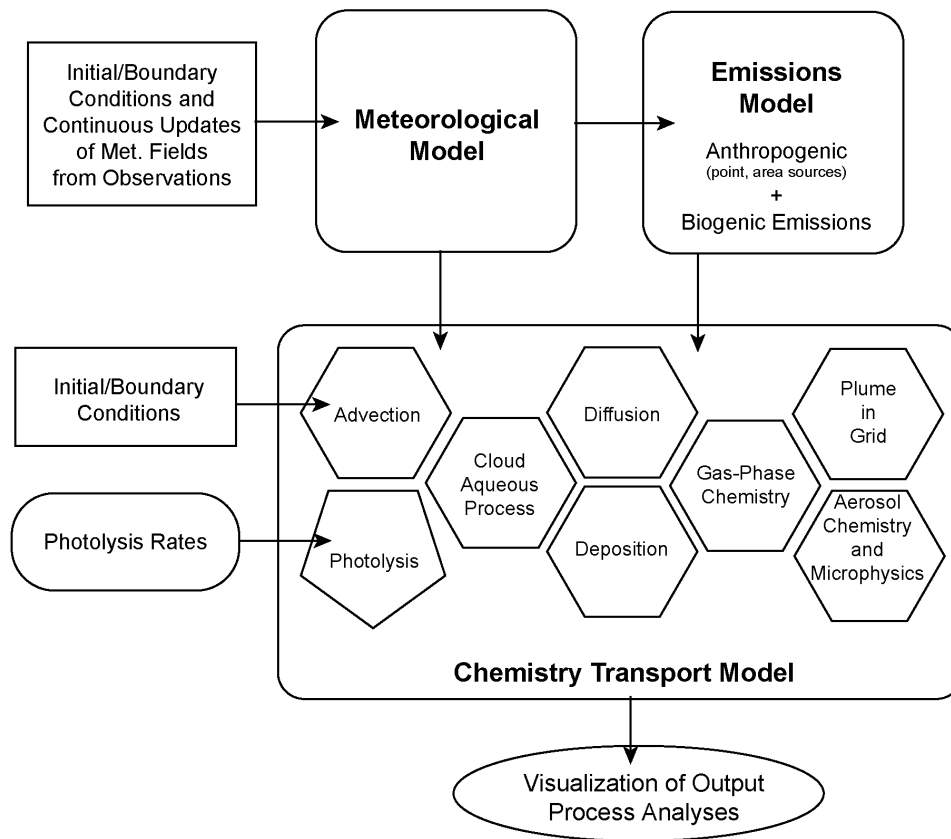


Figure 2-9. Main components of a comprehensive atmospheric chemistry modeling system, such as Models-3.

the transport of species in the CTM. Meteorological models, such as MM5, which supply these inputs to the CTMs mentioned above, also provide daily weather forecasts. The domains of these models extend typically over areas of millions of square kilometers.

Because these models are computationally intensive, it is often impractical to run them over larger domains without sacrificing some features. For these reasons, both the meteorological model and the CTM rely on boundary conditions that allow processes occurring outside the model domain to influence their predictions. The entire system, consisting of meteorological model, emissions processor, and output processors shown in Figure 2-9 constitutes the framework of EPA's Models-3.

Because of the large number of chemical species and reactions that are involved in the oxidation of realistic mixtures of anthropogenic and biogenic hydrocarbons, condensed mechanisms must be used in atmospheric models. These mechanisms are tested by comparison with smog chamber data. However, the existing chemical mechanisms often neglect many important processes such as the formation and subsequent reactions of long-lived carbonyl compounds, the incorporation of the most recent information about intermediate compounds, and heterogeneous reactions involving cloud droplets and aerosol particles.

Emissions inventories are compiled for O₃ precursors (NO_x, VOCs, and CO). Recent estimates and more detailed discussions of the estimates are given in Annex AX2.5.2. Anthropogenic NO_x emissions are associated with combustion processes. Most emissions are in the form of NO, which is formed at high combustion temperatures from atmospheric nitrogen and oxygen and from fuel nitrogen. The two largest sources of NO_x are electric power generation plants and motor vehicles. Emissions of NO_x therefore are highest in areas having a high density of power plants and in urban regions having high traffic density. Natural NO_x sources include stratospheric intrusions, lightning, soils, and wildfires. Lightning, fertilized soils, and wildfires are the major natural sources of NO_x in the United States. Both nitrifying and denitrifying organisms in the soil can produce NO_x, mainly in the form of NO. Emission rates depend mainly on fertilization levels and soil temperature and moisture. Spatial and temporal variability in soil NO_x emissions leads to considerable uncertainty in emissions estimates. Nationwide, about 60% of lightning generated NO_x occurs in the southern United States and about 60% the total NO_x emitted by soils occurs in the central corn belt of the United States. The oxidation of NH₃ emitted mainly by livestock and soils, leads to the formation of a small amount of NO. Uncertainties in natural NO_x inventories are much larger than for anthropogenic NO_x emissions.

Hundreds of VOCs, containing mainly two to about twelve carbon atoms, are emitted by evaporation and combustion processes from a large number of anthropogenic sources. The two largest anthropogenic source categories in the U.S. EPA's emissions inventories are industrial processes and transportation. Emissions of VOCs from highway vehicles account for roughly two-thirds of the transportation-related emissions.

The accuracy of VOC emission estimates is difficult to determine, both for stationary and mobile sources. Evaporative emissions, which depend on temperature and other environmental

factors, compound the difficulties of assigning accurate emission factors. In assigning VOC emission estimates to the mobile source category, models are used that incorporate numerous input parameters (e.g., type of fuel used, type of emission controls, age of vehicle), each of which has some degree of uncertainty. Data for the ratio of CO to NO_x and NMHC to NO_x in traffic tunnels (e.g., Pierson et al., 1990) indicated that emissions of NMHCs and CO from motor vehicles have been underestimated by as much as a factor of two (based on the assumption that emissions of NO_x were reasonably well represented in the inventories). However, the results of more recent studies have been mixed, with many studies showing agreement to within ±50%, as summarized in Air Quality Criteria for Carbon Monoxide (U.S. Environmental Protection Agency, 2000). Remote sensing data (Stedman et al., 1991) indicate that about 50% of NMHC and CO emissions are produced by about 10% of the vehicles. These “super-emitters” are typically poorly maintained. Vehicles of any age engaged in off-cycle operations (e.g., rapid accelerations) emit much more than if operated in normal driving modes.

Vegetation emits significant quantities of VOCs, such as terpenoid compounds (isoprene, 2-methyl-3-buten-2-ol, monoterpenes), compounds in the hexanal family, alkenes, aldehydes, organic acids, alcohols, ketones, and alkanes. The major chemicals emitted by plants are isoprene (35%), 19 other terpenoid compounds and 17 non-terpenoid compounds including oxygenated compounds (40%) (Guenther et al., 2000). Coniferous forests represent the largest source on a nationwide basis, because of their extensive land coverage. Most biogenic emissions occur during the summer, because of their dependence on temperature and incident sunlight. Biogenic emissions are also higher in southern states than in northern states for these reasons and because of species variations. The uncertainty in natural emissions is about 50% for isoprene under midday summer conditions and could be as much as a factor of ten higher for some compounds (Guenther et al., 2000). Uncertainties in both biogenic and anthropogenic VOC emission inventories prevent determination of the relative contributions of these two categories at least in many urban areas. On the regional and global scales, emissions of VOCs from vegetation are much larger than those from anthropogenic sources.

The performance of CTMs must be evaluated by comparison with field data as part of a cycle of model evaluations and subsequent improvements. Evaluations of the CMAQ are given in Arnold et al. (2003) and Fuentes and Raftery (2005). Discrepancies between model predictions and observations can be used to point out gaps in current understanding of

atmospheric chemistry and to spur improvements in parameterizations of atmospheric chemical and physical processes. Model evaluation does not merely involve a straightforward comparison between model predictions and the concentration field of the pollutant of interest. Such comparisons may not be meaningful because it is difficult to determine if agreement between model predictions and observations truly represents an accurate treatment of physical and chemical processes in the CTM or the effects of compensating errors in complex model routines. Ideally, each of the model components (emissions inventories, chemical mechanism, meteorological driver) should be evaluated individually. However, this is rarely done in practice. A comparison between free radical concentrations predicted by parameterized chemical mechanisms and observations suggests that radical concentrations were overestimated by current chemical mechanisms for NO_x concentrations $< \sim 5$ ppb (Volz-Thomas et al., 2003).

In addition to comparisons between concentrations of calculated and measured species, comparisons of correlations between measured primary VOCs and NO_x and modeled VOCs and NO_x are especially useful for evaluating results from chemistry-transport models. Likewise, comparisons of correlations between measured species and modeled species can be used to provide information about the chemical state of the atmosphere and to evaluate model representations (including O_3 production per NO_x , O_3 - NO_x -VOC sensitivity, and the general accuracy of photochemical representations). A CTM that demonstrates the accuracy of both its computed VOC and NO_x in comparison with ambient measurements and the spatial and temporal relations among the critical secondary species associated with O_3 has a higher probability of representing O_3 -precursor relations correctly than one that does not.

2.6 TECHNIQUES FOR MEASURING OZONE AND ITS PRECURSORS

Several techniques have been developed for sampling and measurement of O_3 in the ambient atmosphere at ground level. Although the chemiluminescence method (CLM) using ethylene is designated as the Federal Reference Method (FRM) for measuring O_3 , monitoring in the NAMS/SLAMS networks is conducted mainly with UV absorption spectrometry using commercial short path instruments. The primary reference standard instrument is a relatively long-path UV absorption spectrometer maintained under carefully controlled conditions at NIST (e.g., Fried and Hodgeson, 1982). Episodic measurements are made with a variety of other

techniques based on the principles of chemiluminescence, electrochemistry, differential optical absorption spectroscopy (DOAS), and LIDAR.

In principle, each of these methods is subject to interference. Kleindienst et al. (1993) found that water vapor could cause a positive interference in the CLM with an average positive deviation of 3% ozone/% water vapor at 25 °C. However, they also noted that water vapor could cause positive interferences of up to 9% at high humidities (dew point of 24 °C). The UV absorption spectrometers are subject to positive interference by atmospheric constituents, such as certain aromatic aldehydes that absorb at the 253.7 nm Hg resonance line and are at least partially removed by the MnO₂ scrubber. Parrish and Fehsenfeld (2000) did not find any evidence for significant interference (>1%) in flights through the Nashville urban plume. The same group tested the air of Houston, El Paso, Nashville, Los Angeles, San Francisco and the East Coast. They observed only one instance of substantive positive interference defined as the UV absorption technique showing more than a few ppb more than the CLM. This occurred in Laporte, TX under heavily polluted conditions and a low inversion, at night (Jobson et al., 2004). Leston et al. (2005) observed interference of from 20 to 40 ppb in Mexico City and in a separate smog chamber study. However, the concentrations of relevant compounds were many times higher than found in U.S. urban areas. Thus, it is not likely that such interference could be more than a few ppb under typical ambient conditions. However, Leston et al. (2005) suggested that the use of other materials in the scrubber could have eliminated the interference seen in their smog chamber study.

By far, most measurements of NO are made using the CLM, based on its reaction with O₃. Commercial instruments for measuring NO and NO₂ are constructed with an internal converter for reducing NO₂ to NO and then measuring NO by the CLM. In principle, this technique yields a measurement of NO_x with NO₂ found by difference between NO_x and NO; but, these converters also reduce NO_z compounds thereby introducing a positive interference in the measurement of NO₂. Other methods for measuring NO₂ are available, such as photolytic reduction followed by CLM, laser-induced fluorescence and DOAS. However, they require further development before they can be used for routine monitoring in the NAMS/SLAMS networks. More detailed descriptions of the issues and techniques discussed above and techniques for measuring HNO₃ and VOCs can be found in Annex AX2.6.

2.7 SUMMARY

Ozone is formed by atmospheric reactions involving two classes of precursor compounds, volatile organic compounds (VOCs) and nitrogen oxides (NO_x). Ozone is thus a secondary pollutant. Ozone is ubiquitous throughout the atmosphere; it is present even in remote areas of the globe. The photochemical oxidation of almost all anthropogenic and biogenic VOCs is initiated by reaction with hydroxyl (OH) radicals. At night, when they are most abundant, NO_3 radicals also oxidize alkenes. In coastal and other select environments, Cl and Br radicals can also initiate the oxidation of VOCs.

In urban areas, basically all classes of VOCs (alkanes, alkenes, aromatic hydrocarbons, carbonyl compounds, etc.) and CO are important for O_3 formation. Although knowledge of the oxidative mechanisms of VOCs has improved over the past several years, gaps in knowledge involving key classes, such as aromatic hydrocarbons, still remain. For example, only about half of the carbon initially present in aromatic hydrocarbons in smog chamber studies form compounds that have been identified.

In addition to gas phase reactions, reactions also occur on the surfaces of, or within cloud droplets and airborne particles. Most of the well-established multiphase reactions tend to reduce the rate of O_3 formation in polluted environments. Reactions of Cl and Br containing radicals deplete O_3 in selected environments such as the Arctic during spring, the tropical marine boundary layer and inland salt lakes. Direct reactions of O_3 with atmospheric particles appear to be too slow to reduce O_3 formation significantly at typical ambient PM levels.

Our basic understanding of the meteorological processes associated with summertime O_3 episodes has not changed over the past several years. However, the realization that long-range transport processes are important for determining O_3 concentrations at the surface is growing. In addition to synoptic scale flow fields, nocturnal low-level jets are capable of transporting pollutants hundreds of km from their sources in either the upper boundary layer or the lower free troposphere. Turbulence then brings O_3 and other pollutants to the surface. On larger scales, important progress has been made in identifying the mechanisms of intercontinental transport of O_3 and other pollutants.

Some O_3 would be found near the earth's surface as the result of its downward transport from the stratosphere, even in the absence of photochemical reactions in the troposphere. Intrusions of stratospheric O_3 that reach the surface are rare. Much more common are intrusions

that penetrate to the middle and upper troposphere. However, O₃ transported to the middle and upper troposphere can still affect surface concentrations through various mechanisms that mix air between the planetary boundary layer and the free troposphere above.

The formation of O₃ and associated compounds is a complex, nonlinear function of many factors, including the intensity and spectral distribution of sunlight; atmospheric mixing and other atmospheric processes; and the concentrations of the precursors in ambient air. At lower NO_x concentrations found in most environments, ranging from remote continental areas to rural and suburban areas downwind of urban centers, the net production of O₃ increases with increasing NO_x. At higher concentrations found in downtown metropolitan areas, especially near busy streets and highways and in power plant plumes, there is net destruction of O₃ by reaction with NO. In between these two regimes, there is a transition stage in which O₃ production shows only a weak dependence on NO_x concentrations. The efficiency of O₃ production per NO_x oxidized is generally highest in areas where NO_x concentrations are lowest and decrease with increasing NO_x concentration.

Chemistry transport models are used to improve understanding of atmospheric chemical and physical processes as well as to develop air pollution control strategies. The performance of these models must be evaluated by comparison with field data as part of a cycle of model evaluations and subsequent improvements. Discrepancies between model predictions and observations can be used to point out gaps in current understanding and thus to improve parameterizations of atmospheric chemical and physical processes. Model evaluation does not merely involve a straightforward comparison between model predictions and the concentration fields of a pollutant of interest (e.g., O₃). Such comparisons may not be meaningful because it is difficult to determine if agreement between measurements and model predictions truly represents an accurate treatment of physical and chemical processes in the model or the effects of compensating errors in model routines.

The main methods currently in use for routine monitoring of ambient O₃ are based on chemiluminescence or UV absorption. Measurements at most ambient monitoring sites are based on UV absorption. Both of these methods are subject to interference by other atmospheric components. One study found large positive interference in Mexico City and in a smog chamber, but a few studies conducted in urban plumes did not find significant positive interference in the UV absorption technique. Sufficient new information is not available to

amend estimates of the accuracy and precision of O₃ monitors. Such a reevaluation requires studies of the simultaneous effects of a number of potential interferants including water vapor, organic compounds, and temperature on the UV photometric and chemiluminescent methods.

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3. ENVIRONMENTAL CONCENTRATIONS, PATTERNS, AND EXPOSURE ESTIMATES

3.1 INTRODUCTION

Identification and Use of Existing Air Quality Data

Topics discussed in this chapter include the characterization of ambient air quality data for ozone (O₃), the uses of these data in assessing the exposure of vegetation to O₃, concentrations of O₃ in microenvironments, and a discussion of the currently available human exposure data and exposure model development. The information contained in this chapter pertaining to ambient concentrations is taken primarily from the U.S. Environmental Protection Agency (EPA) Air Quality System (AQS; formerly the AIRS database). The AQS contains readily accessible detailed, hourly data that has been subject to EPA quality control and assurance procedures. Data available in AQS were collected from 1979 to 2001. As discussed in the 1996 O₃ Air Quality Criteria Document or AQCD (U.S. Environmental Protection Agency, 1996), the data available prior to 1979 may be unreliable due to calibration problems and other uncertainties.

As noted in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996), O₃ is the only photochemical oxidant other than nitrogen dioxide (NO₂) that is routinely monitored and for which a comprehensive database exists. Data for peroxyacetyl nitrate (PAN), hydrogen peroxide (H₂O₂), and other oxidants either in the gas phase or particle phase typically have been obtained only as part of special field studies. Consequently, no data on nationwide patterns of occurrence are available for these non-O₃ oxidants; nor are extensive data available on the relationships of levels and patterns of these oxidants to those of O₃. However, available data for gas phase and particle phase oxidants are discussed in this chapter.

Characterizing Ambient Ozone Concentrations

The “concentration” of a specific air pollutant is typically defined as the amount (mass) of that material per unit volume of air. However, most of the data presented in this chapter are expressed as “mixing ratios” in terms of a volume-to-volume ratio (parts per million [ppm] or parts per billion [ppb]). Data expressed this way are often referred to as concentrations, both in

the literature and in the text, following common usage. Human exposures are expressed in units of mixing ratio times time.

Several different air quality metrics have been suggested for evaluating exposures of vegetation to O₃. The peak-weighted, cumulative exposure indicators used in this chapter for characterizing vegetation exposures are SUM06 and SUM08 (the sums of all hourly average concentrations ≥ 0.06 and 0.08 ppm, respectively) and W126 (the sum of the hourly average concentrations that have been weighted according to a sigmoid function that is based on a hypothetical vegetation response [see Lefohn and Runeckles, 1987]). Further discussion of these exposure indices is presented in Chapter 9.

The EPA has established “ozone seasons” during which measurement of ambient O₃ concentrations for different locations within the United States and the U.S. territories is required (CFR, 2000). Table AX3-1 shows the O₃ seasons during which continuous, hourly averaged O₃ concentrations must be monitored. Monitoring is optional outside of these O₃ seasons and indeed is conducted during the winter in a number of areas.

Data for O₃ in ambient air across the United States are summarized in Section 3.2. The data are summarized for urban, rural, and relatively remote sites. Relatively remote monitoring sites (RRMS) are sites that are not strongly influenced by nearby pollution sources and are located mainly in national parks in the West. However, this does not mean that they are free of the effects of regional or local pollution, especially during tourist seasons. Data for the spatial variability of O₃ within urban areas are summarized in Section 3.3. Data for the diurnal and seasonal variability of O₃ concentrations are given in Section 3.4. The long term temporal variability of O₃ concentrations is discussed in Section 3.5. Relationships among O₃ and other species are discussed in Section 3.6. Information about the occurrence of other oxidants and their relationship to O₃ is given in this section. A discussion of Policy Relevant Background (PRB) O₃ concentrations is presented in Section 3.7. PRB O₃ concentrations are background O₃ concentrations that would be observed in the U.S. in the absence of anthropogenic emissions of O₃ precursors in the U.S., Canada, and Mexico. Background levels so defined help facilitate the distinction between pollution levels that can be controlled by U.S. regulations (or through international agreements with neighboring countries) from levels that are generally uncontrollable by the U.S. Indoor sources and emissions of O₃ are discussed in Section 3.8.

Issues related to evaluating human exposure to O₃ are summarized in Section 3.9. Finally, a summary of key points in Chapter 3 is given in Section 3.10.

3.2 AMBIENT AIR QUALITY DATA FOR OZONE

Ozone Air Quality at Urban, Suburban, and Nonurban Sites

Figure 3-1 shows the mean daily maximum 8-h O₃ concentrations, and Figure 3-2 shows the 95th percentile values of the daily maximum 8-h O₃ concentrations, based on countywide site-wise averages across the United States from May to September 2000 to 2004. The period from May to September was chosen because, although O₃ is monitored for different lengths of time across the country, all O₃ monitors should be operational during these months. Data flagged because of quality control issues were removed with concurrence of the local monitoring agency. Only days with data for 18 of 24 hours were kept, and a minimum of 115 of 153 days were required in each year. Cut points for the tertile distributions on each map were chosen at the median and 95th percentile values. These cut points were chosen because they represent standard metrics for characterizing important aspects of human exposure used by the EPA. Any other percentiles or statistics that are believed to be helpful for characterizing human exposures could also be used. Blank areas on the maps indicate no data coverage. It should be noted that county areas can be much larger in the West than in the East, but monitors are not spread evenly within a county. As a result, the assigned concentration range might not represent conditions throughout a particular county and, so, large areas in western counties where there are no monitors were blanked out.

As shown in Figure 3-1, the median of the countywide, mean daily maximum 8-h O₃ concentration across the United States is 49 ppb, and 5% of these site-wise means exceeded 57 ppb. Though the median and 95th percentile values are fairly close, these results cannot be taken to imply that average O₃ concentrations lie in a relatively narrow range throughout the United States, because data coverage is not as complete in the West as it is in the East. High mean daily maximum 8-h O₃ concentrations are found in California and states in the Southwest as well as in several counties in the East. As shown in Figure 3-2, the nationwide median of the countywide, 95th percentile value of the daily maximum 8-h O₃ concentration is 73 ppb and 5% of these values are above 85 ppb. High values for the 95th percentiles are found in California,

Seasonal (May-September) Mean of Daily Maximum 8-Hour Values, 2002-2004

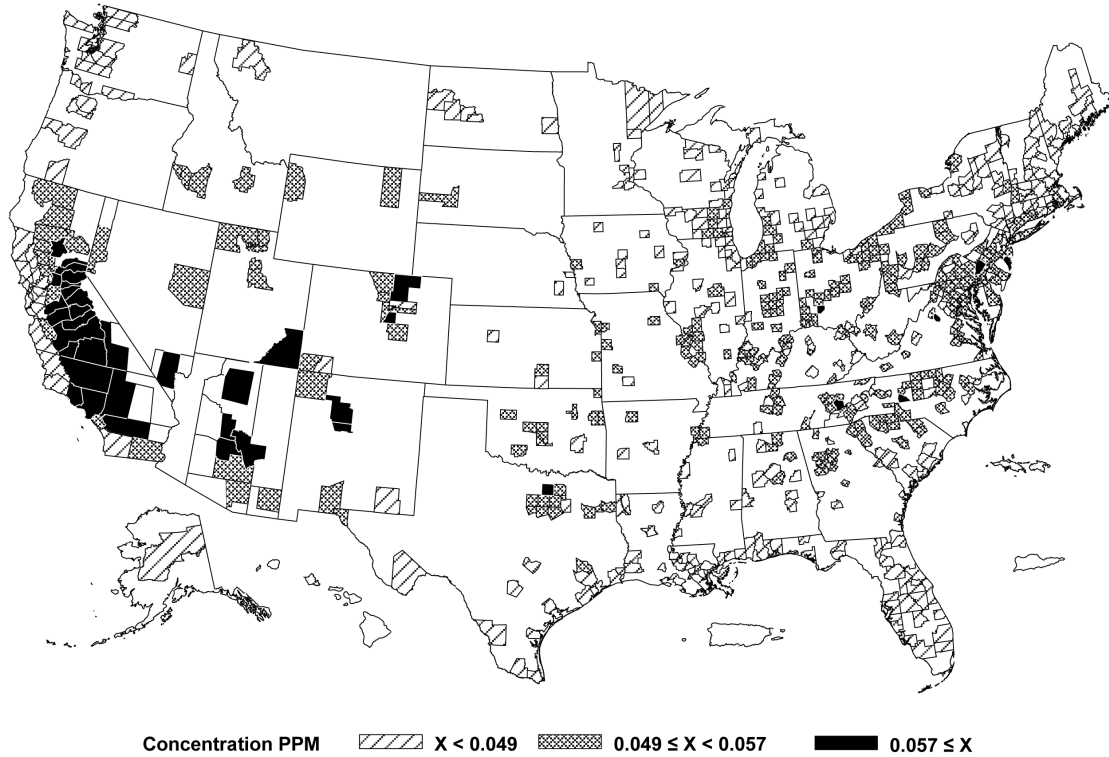


Figure 3-1. Countywide mean daily maximum 8-h O₃ concentrations, May to September 2000 to 2004.

Source: Fitz-Simons et al. (2005).

Texas, and some counties in the East, but not necessarily in the same counties in the East as shown for the mean daily maximum 8-h concentrations in Figure 3-1.

Although mean O₃ concentrations in Houston, TX were below the nationwide median, its 95th percentile value ranks in the highest 5% nationwide. Conversely, mean O₃ concentrations in southwestern states are among the highest in the United States, but values at the upper end of the distribution (e.g., the 95th percentile value) in these states are not among the highest peak values in the United States. In other areas where the highest mean O₃ concentrations occurred (e.g., California; Dallas-Fort Worth, TX; and the Northeast Corridor), the highest peak values were also observed.

Seasonal (May-September) 95th Percentile of Daily Maximum 8-Hour Values, 2002-2004

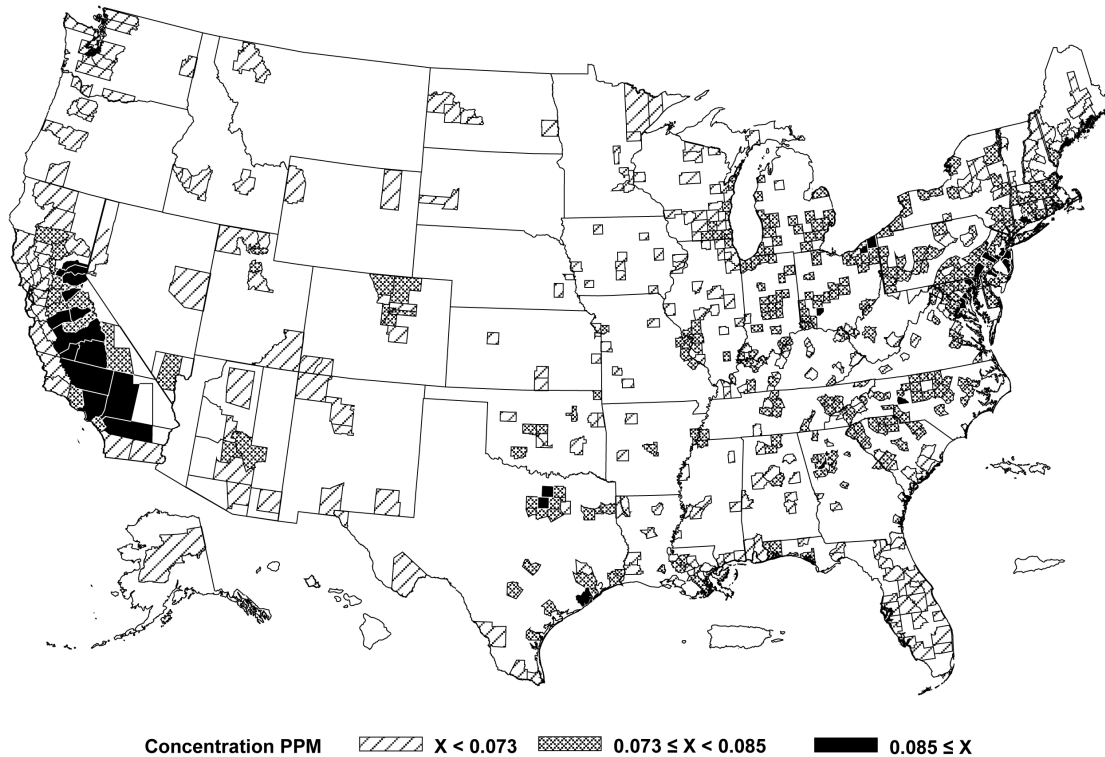


Figure 3-2. Countywide 95th percentile value of daily maximum 8-h O₃ concentrations, May to September 2000 to 2004.

Source: Fitz-Simons et al. (2005).

Although countywide averages are shown, it should be noted that considerable spatial variability can exist within a county, especially within urban areas as described in Section 3.3. In addition, there can also be differences in the diurnal profile of O₃ among monitors within counties.

Box plots showing the distribution of nationwide O₃ concentrations for different averaging periods (1-h daily maximum, 8-h daily maximum, and 24-h daily average) are given in Figures AX3-4 to AX3-6 and numerical values are given in Table AX3-2. The differences between the 50th and 95th percentile values indicate the range of O₃ levels between “typical” O₃ days and “high” O₃ days. These differences are approximately 40, 30, and 25 ppb for the daily 1-h and

8-h daily maxima and 24-h average O₃ concentrations, respectively. As might be expected, the daily maximum 1-h and 8-h O₃ concentrations are highly correlated.

Lehman et al. (2004) divided the eastern United States into five regions, each of which exhibit relatively distinct spatial and temporal patterns of O₃ concentrations at nonurban sites. Only sites classified as being rural or suburban and with land usage of forest, agriculture, or residential were included in the analyses. These criteria were chosen to avoid sites where O₃ is scavenged by NO, which can be found in high concentrations near major sources such as traffic in urban cores. The five regions, shown in Figure 3-3, are characterized by different patterns of O₃ properties such as temporal persistence and seasonal variability. Figure 3-3 shows nonurban, monthly average, daily maximum 8-h O₃ concentrations in the five regions in the eastern United States from April to October 1993 to 2002.

Regional differences are immediately apparent. Highest concentrations among all the regions are generally found in the Mid-Atlantic region (mean of 52 ppb), with highest values throughout the O₃ concentration distribution except for the overall maximum. Lowest mean concentrations (42 ppb) are found in Florida. In the northern regions (the Northeast, Great Lakes) and the Mid-Atlantic region, highest median and peak concentrations are found in July, whereas in the Southwest region, highest median concentrations are found in August, with highest peaks in June and September, i.e., outside the warmest summer months. In Florida, highest monthly averaged median and peak concentrations are found during the spring. High O₃ concentrations tend to be most persistent (3 to 4 days of persistence) in the southern regions, less persistent in the Mid-Atlantic region (2 to 3 days) and least persistent in the northern regions (1 or 2 days). Such analyses could not be made for the western United States, in part because of the difficulty in finding regions with relatively coherent O₃ properties as noted above for the eastern United States.

Box plots showing the distributions of hourly average O₃ concentrations for different types of rural sites for 2004 are given in Figures 3-4a (rural-agricultural), 3-4b (rural-forest), and 3-4c (rural-residential or commercial). Some associated metrics for vegetation exposures are given in Figures AX3-8 to AX3-10. Note that high O₃ concentrations are found at sites that are classified as rural, such as Anne Arundel Co., MD; Yosemite NP, CA; and Crestline, CA. Land use designations do not usually give an accurate picture of exposure regimes in rural areas, because the land use characterization of “rural” does not imply that a specific location is isolated from

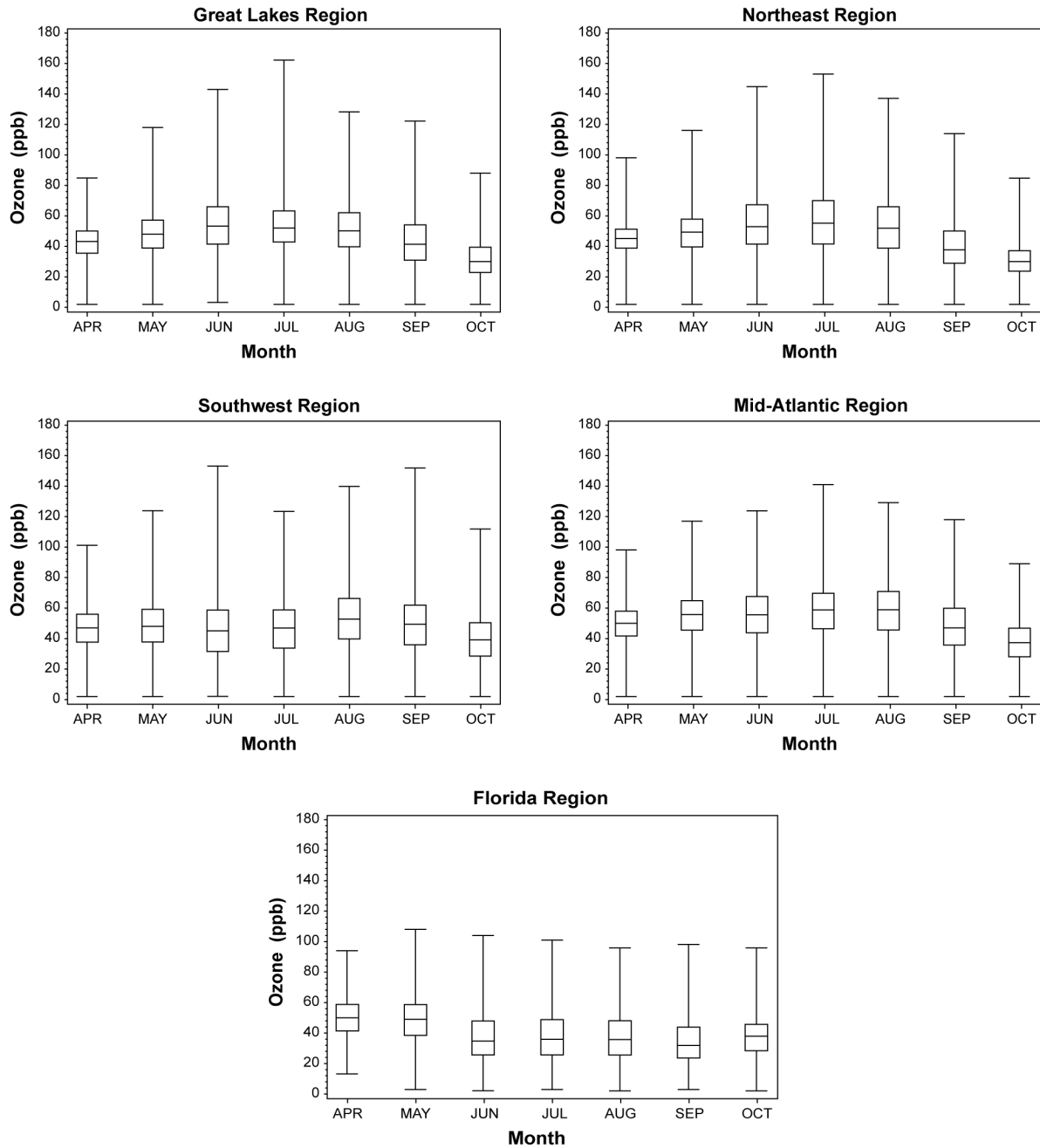


Figure 3-3. Box plots showing daily maximum 8-h O_3 averaged by month over 1993 to 2002 in the five regions in the eastern United States derived by Lehman et al.(2004). The boxes define the interquartile range and the whiskers, the extreme values.

Source: Lehman et al. (2004).

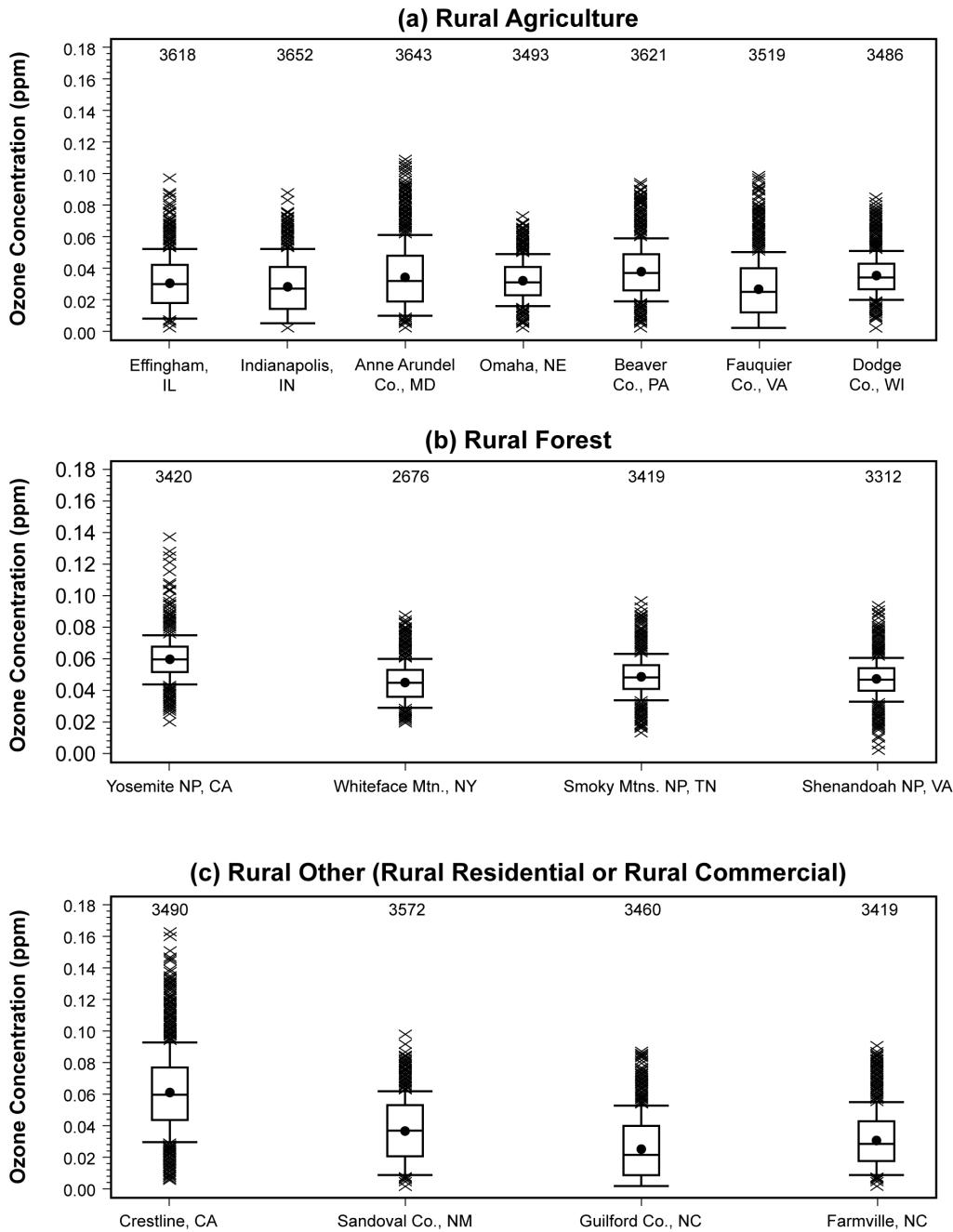


Figure 3-4a-c. Hourly average O₃ concentrations observed at selected (a) rural-agricultural (b) rural-forested, and (c) rural-residential or commercial sites for 2004. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean, for the statistic, at all sites. The number of observations is shown above each box plot.

Source: Fitz-Simons et al. (2005).

anthropogenic influences. Rather, the characterization refers only to the current use of the land, not to the presence of sources. Since O₃ produced from emissions in urban areas is transported to more rural downwind locations, elevated O₃ concentrations can occur at considerable distances from urban centers. In addition, major sources of O₃ precursors such as power plants and highways are located in nonurban areas and also produce O₃ in these areas. Due to lower chemical scavenging in nonurban areas, O₃ tends to persist longer in nonurban than in urban areas which tends to lead to higher exposures in nonurban areas influenced by anthropogenic precursor emissions.

Ozone Air Quality Data at Relatively Remote Monitoring Sites (RRMS)

Relatively Remote Monitoring Sites (RRMS) are sites located in national parks that tend to be less affected by obvious pollution sources than other sites. However, this does not mean that they are completely unaffected by local pollution, because of the large number of visitors to the national parks.

Box plots showing the distribution of annual hourly averaged O₃ concentrations at four RRMS are given in Figures 3-5a-d. It is important to characterize hourly average O₃ concentrations at RRMS so that assessments of the possible effects of O₃ on human health and vegetation use concentration ranges that span the range of O₃ concentrations found in the U.S. In many controlled exposure studies examining vegetation, O₃ is filtered out of ambient air before it is admitted into the exposure chambers. As a result, O₃ levels of only a few ppb are used as controls.

As can be seen from Figures 3-5a-d, annual mean values of the daily maximum 8-h O₃ concentration have not changed much over the past 10 years of available data. Mean values typically range from about 0.020 ppm to about 0.040 ppm. Concentrations only rarely exceed 0.080 ppm, in contrast to observations at other “rural” sites shown in Figures 3-4a-c.

It is unlikely that distributions found at sites with low maximum hourly average concentrations in the western United States could represent those at sites in the eastern and midwestern United States because of regional differences in sources of precursors and transport patterns. Given the high density of sources in the eastern and midwestern United States, it is unclear whether a site could be found in either of these regions that would not be influenced by the transport of O₃ from nearby urban areas. Thus, with the exception of the Voyageurs NP site

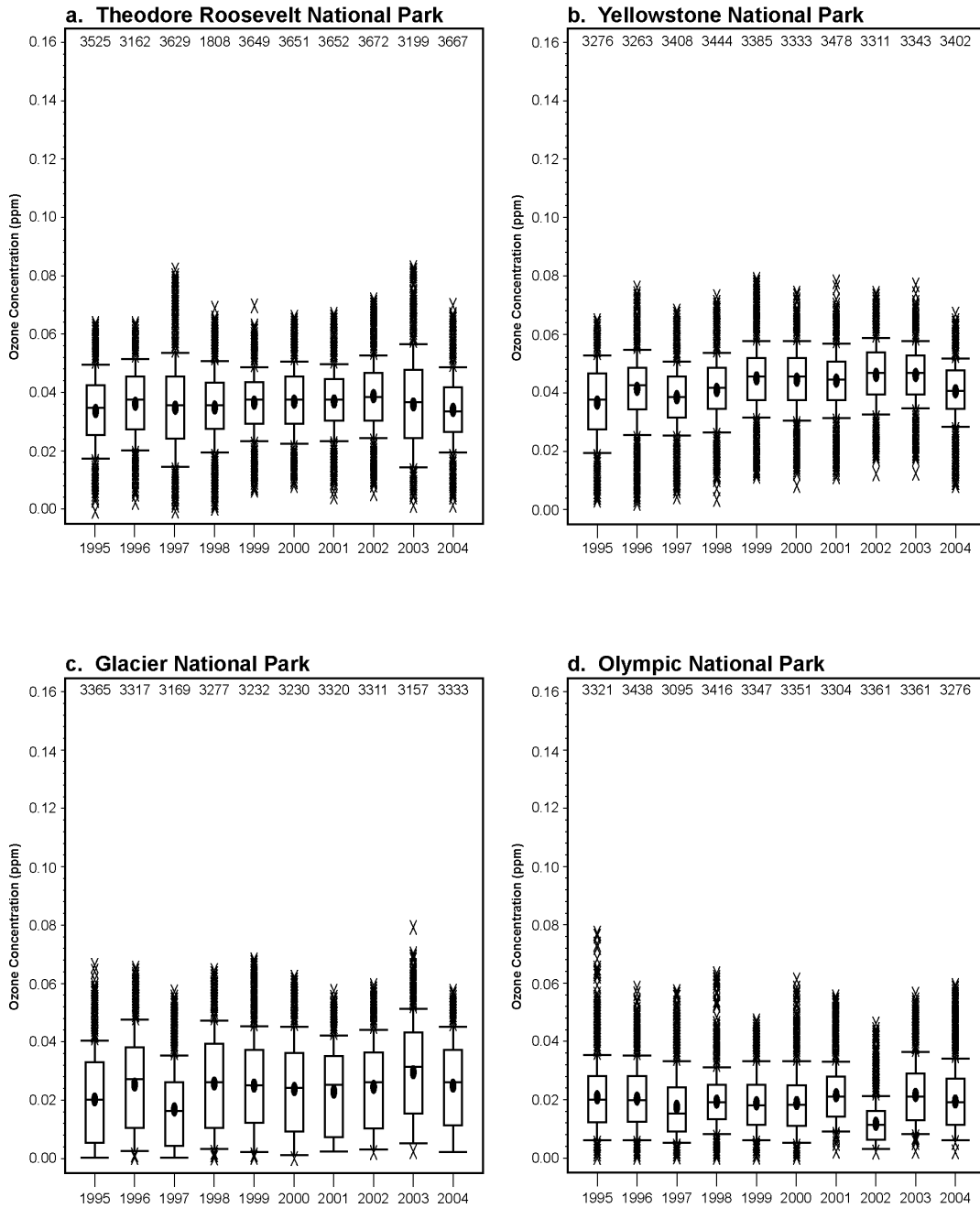


Figure 3-5a-d. Daily 8-h maximum O₃ concentrations observed at selected national park sites. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean. The number of observations is shown above each box plot.

Source: Fitz-Simons et al. (2005).

in Minnesota, observations at RRMS are limited to those obtained in the western United States. However, not all national park sites in the West can be considered to be free of strong regional pollution influences, e.g., Yosemite NP (CA) as shown in Figure 3-4b. Maps showing the nationwide distribution of various metrics for vegetation exposures are given in Section AX3.2, Figures AX3-13 to AX3-27.

3.3 SPATIAL VARIABILITY OF OZONE IN URBAN AREAS

The spatial variability in O₃ concentrations in 24 MSAs across the United States was examined. These MSAs were selected to provide (1) information helpful for risk assessments, (2) a general overview of the spatial variability of O₃ in different regions of the country, and (3) insight into the spatial distribution of O₃ in cities where health outcome studies have been conducted. Statistical analyses of the human health effects of airborne pollutants based on aggregate population time-series data have often relied on ambient concentrations of pollutants measured at one or more central sites in a given metropolitan area. In the particular case of ground-level O₃ pollution, central-site monitoring has been justified as a regional measure of exposure mainly on the grounds that correlations between concentrations at neighboring sites measured over time are usually high. In MSAs with multiple monitoring sites, averages over the monitors have often been used to characterize population exposures. However, substantial differences in concentrations between monitors can exist even though concentrations measured at the monitoring sites are highly correlated, thus leading to the potential for exposure misclassification error.

Metrics for characterizing spatial variability include the use of Pearson correlation coefficients (*r*), values of the 90th percentile (*P*₉₀) of the absolute difference in O₃ concentrations, and coefficients of divergence (COD)¹. These methods of analysis follow those used for characterizing PM_{2.5} and PM_{10-2.5} concentrations in Pinto et al. (2004) and in the latest edition of

¹The COD is defined as follows:

$$COD_{jk} = \sqrt{\frac{1}{p} \sum_{i=1}^p \left(\frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}} \right)^2}$$

where *x*_{*ij*} and *x*_{*ik*} represent the 24-h average PM_{2.5} concentration for day *i* at site *j* and site *k* and *p* is the number of observations.

the Particulate Matter Air Quality Criteria Document (PM AQCD) (U.S. Environmental Agency, 2004a). However, the calculations were performed on an hourly basis rather than on a 24-h basis. Data were aggregated over local O₃ seasons, the lengths of which vary from state to state. In several southwestern states, it lasts all year long. However, it typically last for 6 months in other areas, such as in New England, the mid-Atlantic states, the Midwest, and the Northwest (see Table AX3-1).

Table 3-1 shows the chosen urban areas, the range of 24-h average O₃ concentrations over the local O₃ season from 1999 to 2001, the range of intersite correlation coefficients, the range of P₉₀ differences in O₃ concentrations between site pairs, and the range in COD values. A COD of zero implies that values in both data sets are identical, and a COD of one indicates that the two data sets are uncorrelated, with no matching values from either data set. In general, statistics were calculated for partial MSAs. This was done so as to obtain reasonable lower estimates of the spatial variability that is present, as opposed to examining the consolidated MSAs. However, this could not be readily done for Boston, MA and New York, NY, so statistics were calculated for those consolidated MSAs. More detailed calculations for a subset of nine MSAs are given in Figures AX3-28 through AX3-36 in Section AX3.3.

As can be seen from Table 3-1, no clearly discernible regional differences were found in the ranges of parameters analyzed. Additional urban areas would need to be examined to discern broadscale patterns. The data indicate considerable variability in the concentration fields. Mean O₃ concentrations vary within individual urban areas by factors of 1.4 to 4 in Table 3-1. Intersite correlation coefficients show mixed patterns (i.e., in some urban areas all pairs of sites are moderately to highly correlated, while other areas show a larger range of correlations). As may be expected, those areas showing a smaller range of seasonal mean concentrations also show a smaller range of intersite correlation coefficients. However, there are a number of cases where sites in an urban area may be moderately to highly correlated, but show substantial differences in absolute concentrations. In many cases, P₉₀ values can equal or exceed seasonal mean O₃ concentrations.

It is instructive to compare the metrics for spatial variability shown in Table 3-1 to those calculated for PM_{2.5} and PM_{10-2.5} in the PM AQCD (U.S. Environmental Agency, 2004a). The values for concentrations and concentration differences are unique to the individual species, but the intersite correlation coefficients and the COD values can be directly compared. In general,

Table 3-1. Summary Statistics for the Spatial Variability of O₃ (in ppm) in Selected Urban Areas in the United States

Urban Area	Number of Sites	Minimum Mean Conc.	Maximum Mean Conc.	Minimum Corr. Coeff.	Maximum Corr. Coeff.	Minimum P ₉₀ ^a	Maximum P ₉₀	Minimum COD ^b	Maximum COD
Boston, MA	18	0.021	0.033	0.46	0.93	0.012	0.041	0.17	0.45
New York, NY	29	0.015	0.041	0.45	0.96	0.0080	0.044	0.17	0.55
Philadelphia, PA	12	0.020	0.041	0.79	0.95	0.011	0.036	0.23	0.46
Washington, DC	20	0.022	0.041	0.72	0.97	0.010	0.032	0.17	0.45
Charlotte, NC	8	0.031	0.043	0.48	0.95	0.012	0.038	0.17	0.32
Atlanta, GA	12	0.023	0.047	0.63	0.94	0.013	0.045	0.24	0.55
Tampa, FL	9	0.024	0.035	0.74	0.94	0.011	0.025	0.20	0.35
Detroit, MI	7	0.022	0.037	0.74	0.96	0.0090	0.027	0.19	0.36
Chicago, IL	24	0.015	0.039	0.38	0.96	0.0080	0.043	0.16	0.50
Milwaukee, WI	9	0.027	0.038	0.73	0.96	0.0090	0.025	0.18	0.33
St. Louis, MO	17	0.022	0.038	0.78	0.96	0.0090	0.031	0.15	0.41
Baton Rouge, LA	7	0.018	0.031	0.81	0.95	0.0090	0.029	0.23	0.41
Dallas, TX	10	0.028	0.043	0.67	0.95	0.011	0.033	0.16	0.36
Houston, TX	13	0.016	0.036	0.73	0.96	0.0090	0.027	0.20	0.38
Denver, CO	8	0.022	0.044	0.60	0.92	0.013	0.044	0.16	0.46
El Paso, TX	4	0.022	0.032	0.81	0.94	0.012	0.023	0.24	0.31
Salt Lake City, UT	8	0.029	0.048	0.52	0.92	0.012	0.043	0.13	0.51
Phoenix, AZ	15	0.021	0.058	0.29	0.95	0.011	0.057	0.15	0.61
Seattle, WA	5	0.015	0.038	0.63	0.94	0.0080	0.024	0.16	0.46
Portland, OR	5	0.015	0.036	0.73	0.91	0.011	0.025	0.20	0.50
Fresno, CA	6	0.030	0.047	0.90	0.97	0.0090	0.027	0.17	0.40
Bakersfield, CA	8	0.028	0.047	0.23	0.96	0.013	0.052	0.20	0.58
Los Angeles, CA	14	0.010	0.042	0.42	0.95	0.010	0.053	0.22	0.59
Riverside, CA	18	0.018	0.054	0.38	0.95	0.013	0.057	0.15	0.64

^a P₉₀ = 90th percentile absolute difference in concentrations.

^b COD = coefficient of divergence for different site pairs.

the variability in O₃ concentrations is larger than for PM_{2.5} concentrations and comparable to that obtained for PM_{10-2.5}. Intersite correlation coefficients in some areas (e.g., Philadelphia, PA; Atlanta, GA; Portland, OR) can be very similar for both PM_{2.5} and for O₃. However, there is much greater variability in the concentration fields of O₃ as evidenced by the much higher COD values. Indeed, COD values are higher for O₃ than for PM_{2.5} in each of the urban areas examined. In all of the urban areas examined for O₃, some site pairs are always very highly correlated with each other (i.e., $r > 0.9$) as seen for PM_{2.5}. These sites also show less variability in concentration and are probably influenced most strongly by regional production mechanisms.

The above considerations indicate that caution should be observed in using data from the network of ambient O₃ monitors to approximate community-scale human exposures. A similar conclusion was reached for PM using data from the PM_{2.5} FRM network, as indicated in Section 3.4 of the PM AQCD (U.S. Environmental Protection Agency, 2004a).

3.3.1 Small-Scale Horizontal and Spatial Variability in Ozone Concentrations

Ozone concentrations near roadways

Apart from the larger scale variability in surface O₃ concentrations, there is also significant variability on the micro-scale (< a few hundred meters), especially near roadways and other sources of emissions that react with O₃. These sources are not confined to urban areas. Sources of emissions that react with O₃ such as highways and power plants are also found in rural areas. Johnson (1995) described the results of studies examining O₃ upwind and downwind of roadways in Cincinnati, OH. In these studies, O₃ upwind of the roadway was about 50 ppb and values as high as this were not found again until distances of about 100 m downwind. The O₃ profile varied inversely with that of NO, as might be expected. For peak NO concentrations of 30 ppb immediately downwind of the road, the O₃ mixing ratio was about 36 ppb, or about 70% of the upwind value. The magnitude of the downwind depletion of O₃ depends on the emissions of NO, the rate of mixing of NO from the roadway and ambient temperatures. So depletions of O₃ downwind of roadways are expected, but with variable magnitude. In interpreting historical data, it should be noted that scavenging of O₃ by NO near roadways was more pronounced before the implementation of stringent NO_x emissions controls.

Guidance for the placement of O₃ monitors (U.S. Environmental Protection Agency, 1998) states a separation distance that depends on traffic counts. For example, a minimum separation distance of 100 m from a road with 70,000 vehicles per day (about 3,000 vehicles per hour) is recommended for siting an O₃ monitor to avoid interference that would mean a site is no longer representative of the surrounding area. An average rate of about 3,000 vehicles per hour passing by a monitoring site implies a road with rather heavy traffic. As noted in Section AX3.3.1 for the Lakewood, CA monitoring, O₃ levels are lower at sites located near traffic than those located some distance away, and the scavenging of O₃ by emissions of NO from roadways is a major source of spatial variability in O₃ concentrations.

Vertical Variations in Ozone Concentrations

In addition to horizontal variability in O₃ concentrations, consideration must also be given to variations in the vertical profile of O₃ in the lowest layers of the atmosphere. The planetary boundary layer (PBL) consists of an outer and an inner portion. The inner part extends from the surface to about one-tenth the height of the PBL. Winds and transported pollutants, such as O₃, are especially susceptible to interactions with obstacles, such as buildings and trees in the inner boundary layer (atmospheric surface layer) (e.g., Garratt, 1992). Inlets to ambient monitors (typically at heights of 3 to 5 meters) are located in, and human and vegetation exposures occur in this part of the boundary layer.

Photochemical production and destruction of O₃ occur throughout the PBL. However, O₃ is also destroyed on the surfaces of buildings, vegetation, etc. In addition, O₃ is scavenged by NO emitted by motor vehicles and soils. These losses imply that the vertical gradient of O₃ should always be directed downward. The magnitude of the gradient is determined by the intensity of turbulent mixing in the surface layer.

Most work characterizing the vertical profile of O₃ near the surface has been performed in nonurban areas with the aim of calculating fluxes of O₃ and other pollutants through forest canopies and to crops and short vegetation, etc. Corresponding data are sparse for urban areas. However, monitoring sites are often set up in open areas such as parks and playgrounds where surface characteristics may resemble those in rural areas more than those in the surrounding urban area. The vertical profiles of O₃ measured over low vegetation is shown in Figure 3-6. These measurements were obtained as part of a field campaign to measure the fluxes of several

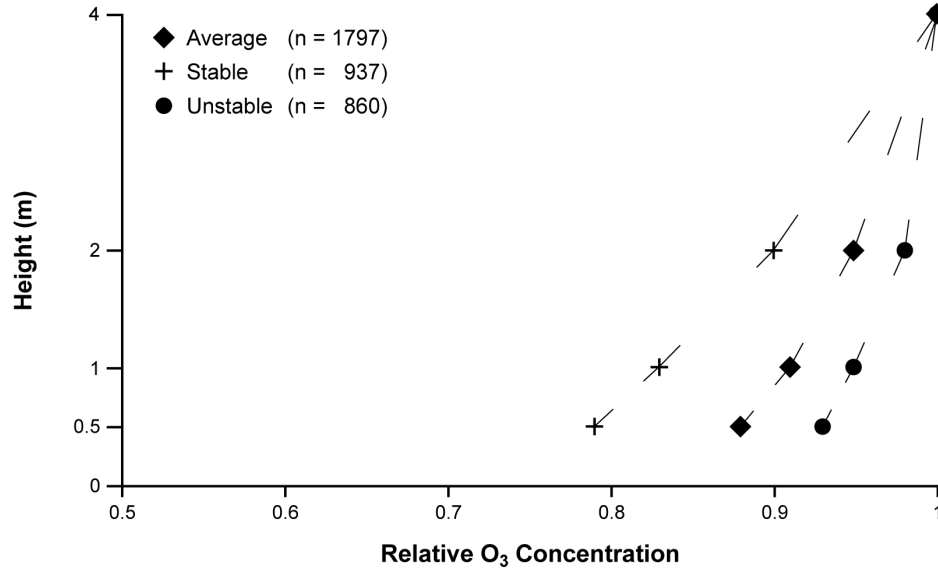


Figure 3-6. Vertical profile of O₃ obtained over low vegetation. Values shown are relative to concentrations at 4 m above the surface. Ozone concentrations for stable and unstable conditions were 41.3 and 24.1 ppb, and average O₃ concentration weighted by stability class was 33.1 ppb at 4 m.

Source. Horváth et al. (1995).

gas and aerosol phase pollutants using the gradient-flux technique (Horváth et al., 1995). The labels stable and unstable in the figure refer to atmospheric stability conditions and average represents the overall average. Ozone concentrations were normalized relative to their values at a height of 4 m. As can be seen from the figure, there was a decrease of about 20% in going from a height of 4 m down to 0.5 m above the surface during stable conditions, but O₃ decreased by only about 7% during unstable conditions. The average decrease was about 10% for all measurements. As might be expected, O₃ concentrations at all heights were very highly correlated with one another. Of course, these values represent averages and there is scatter about them. Under strongly stable conditions, they fall off toward the surface. However, these conditions tend to occur mainly during night and the stability regime during the day in urban areas tends more toward instability because of the urban heat island effect. Figure 3-7 shows the vertical profile of O₃ measured in a spruce forest (Horváth et al., 2003). The fall off in O₃ for

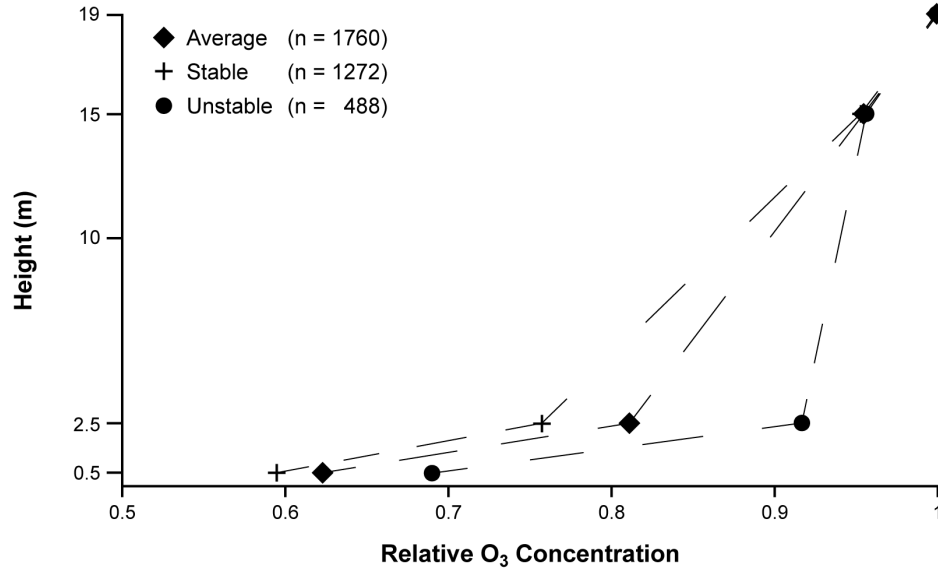


Figure 3-7. Vertical profile of O₃ obtained in a spruce forest. Values shown are relative to concentrations at 19 m above the surface. Mean tree height is 14.5 m. Ozone concentrations for stable and unstable conditions were 36.7 and 33.8 ppb, and the average O₃ concentration weighted by stability class was 34.6 ppb at 19 m.

Source: Horváth et al. (2003).

this case is due to uptake by trees, reaction with ambient NO and with NO emitted by the soil in the forest, and reaction with hydrocarbons emitted by the trees in addition to deposition on the surface.

3.4 DIURNAL AND SEASONAL VARIABILITY OF OZONE

Diurnal Variability

Diurnal variations in O₃ at a given location are controlled by a number of factors, such as the relative importance of transport versus local photochemical production and loss rates, the timing for entrainment of air from the nocturnal residual boundary layer and the diurnal variability in mixing layer height.

Diurnal Patterns in the Nationwide Data Set

Composite urban, diurnal variations in hourly averaged O₃ for April through October 2000 to 2004 are shown in Figure 3-8. As can be seen from Figure 3-8, daily 1-h maxima tend to occur in mid-afternoon and daily 1-h minima tend to occur during the early morning. However, there is also considerable spread in these times. Therefore, some caution must be exercised in extrapolating results from one city to another and when attempting to judge the time of day when the daily 1-h maximum occurs.

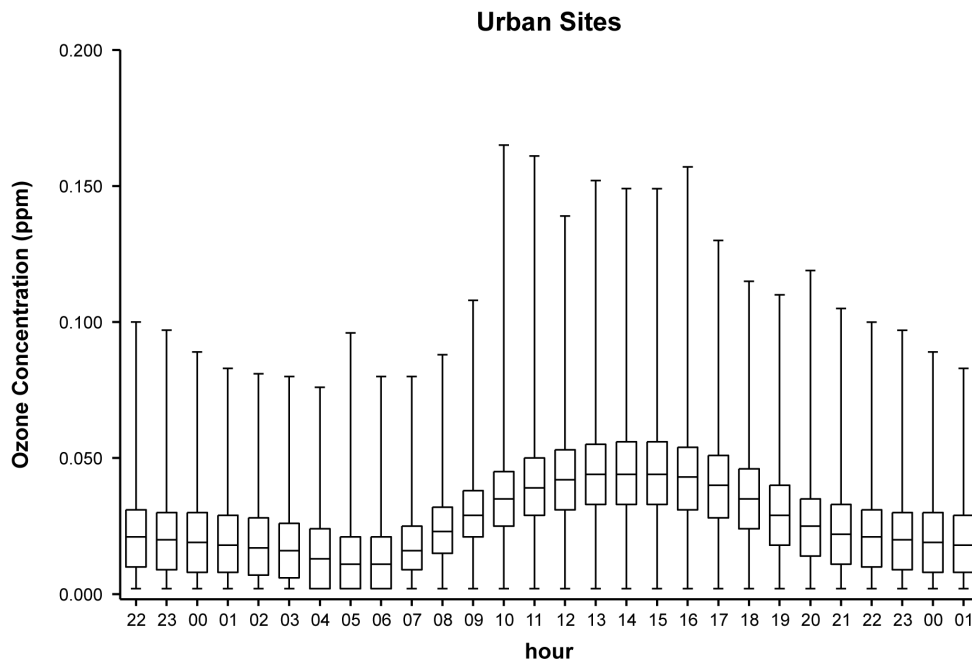


Figure 3-8. Composite, nationwide diurnal variability in hourly averaged O₃ in urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

Corresponding data for 8-h average O₃ variations are shown in Figure 3-9. As can be seen from Figure 3-9, daily maximum eight hour O₃ concentrations tend to occur from about 10 a.m. to about 6 p.m. As can be seen from Figure 3-9, they can also occur at slightly different times

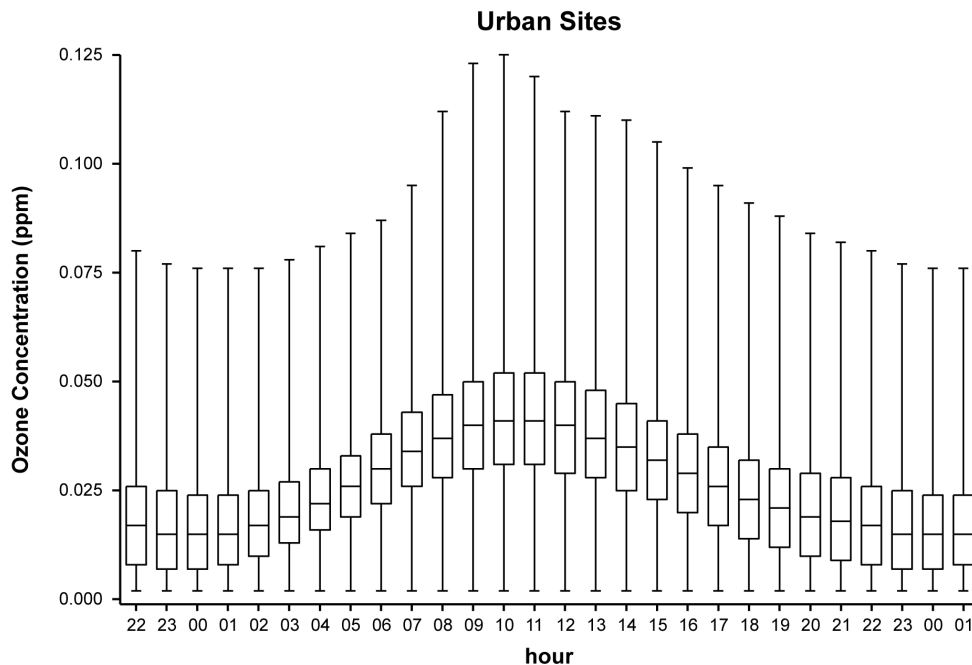


Figure 3-9. Composite, nationwide diurnal variability in 8-h average O₃ in urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

and the variation in the 8-h averages is smoother than for the 1-h averages. The minima in the 8-h averages tend to occur starting at about midnight.

Diurnal Patterns in EPA's 12 Cities

The diurnal variability of hourly averaged O₃ in the twelve urban areas considered for inclusion in EPA's human health exposure assessment risk assessment for the current review is illustrated in Figures 3-10a-l for April to October. Daily maximum 1-h concentrations tend to occur in mid-afternoon. However, as can be seen from the figures, the diurnal patterns vary from city to city, with high values (≥ 0.100 ppm) also occurring either late in the evening as in Boston, past midnight as in Los Angeles and Sacramento, or midmorning as in Houston. Typically, high values such as these are found during the daylight hours in mid to late afternoon. The reasons for the behavior of O₃ during the night at the above-mentioned locations are not

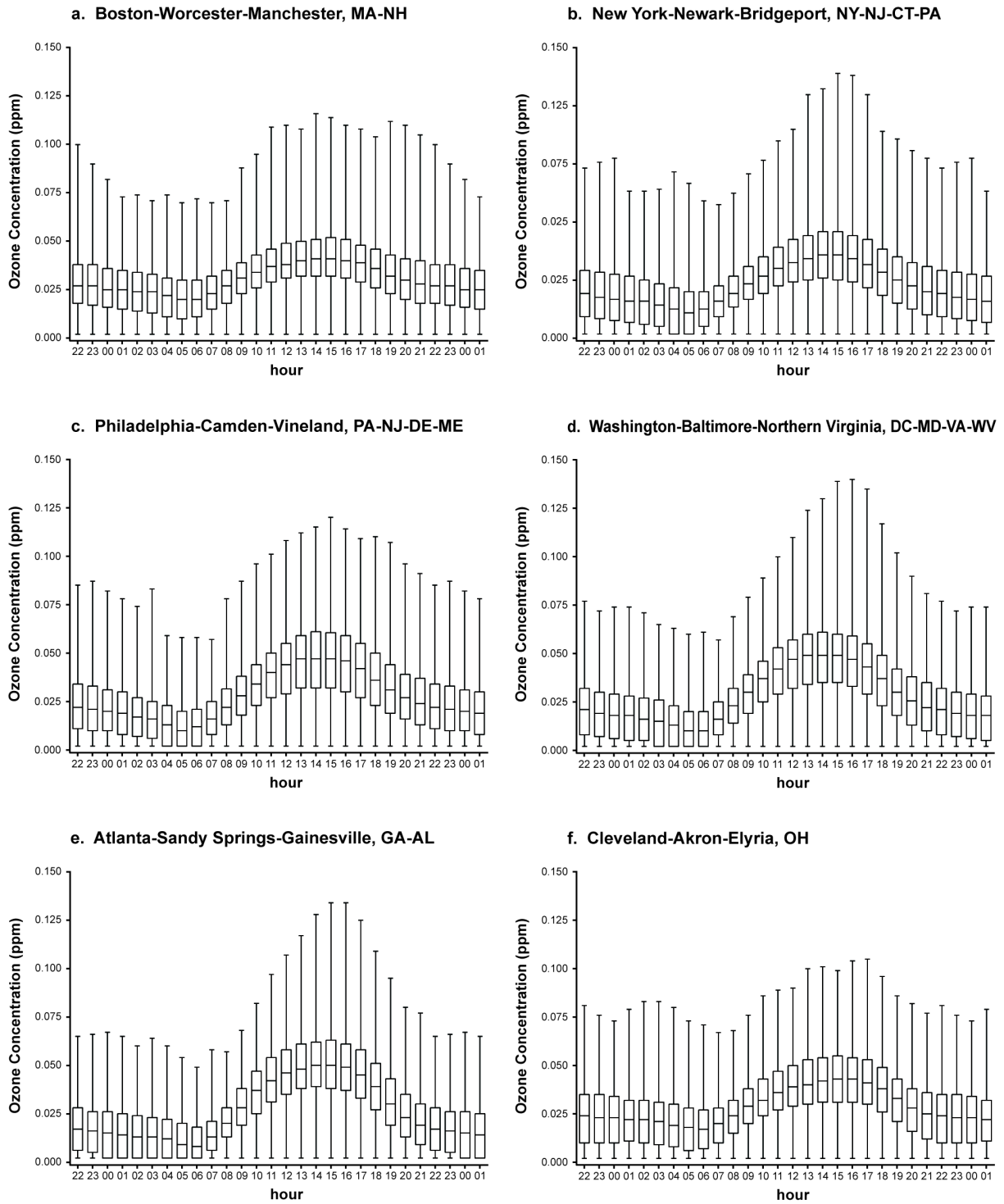


Figure 3-10a-f. Diurnal variability in hourly averaged O₃ in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

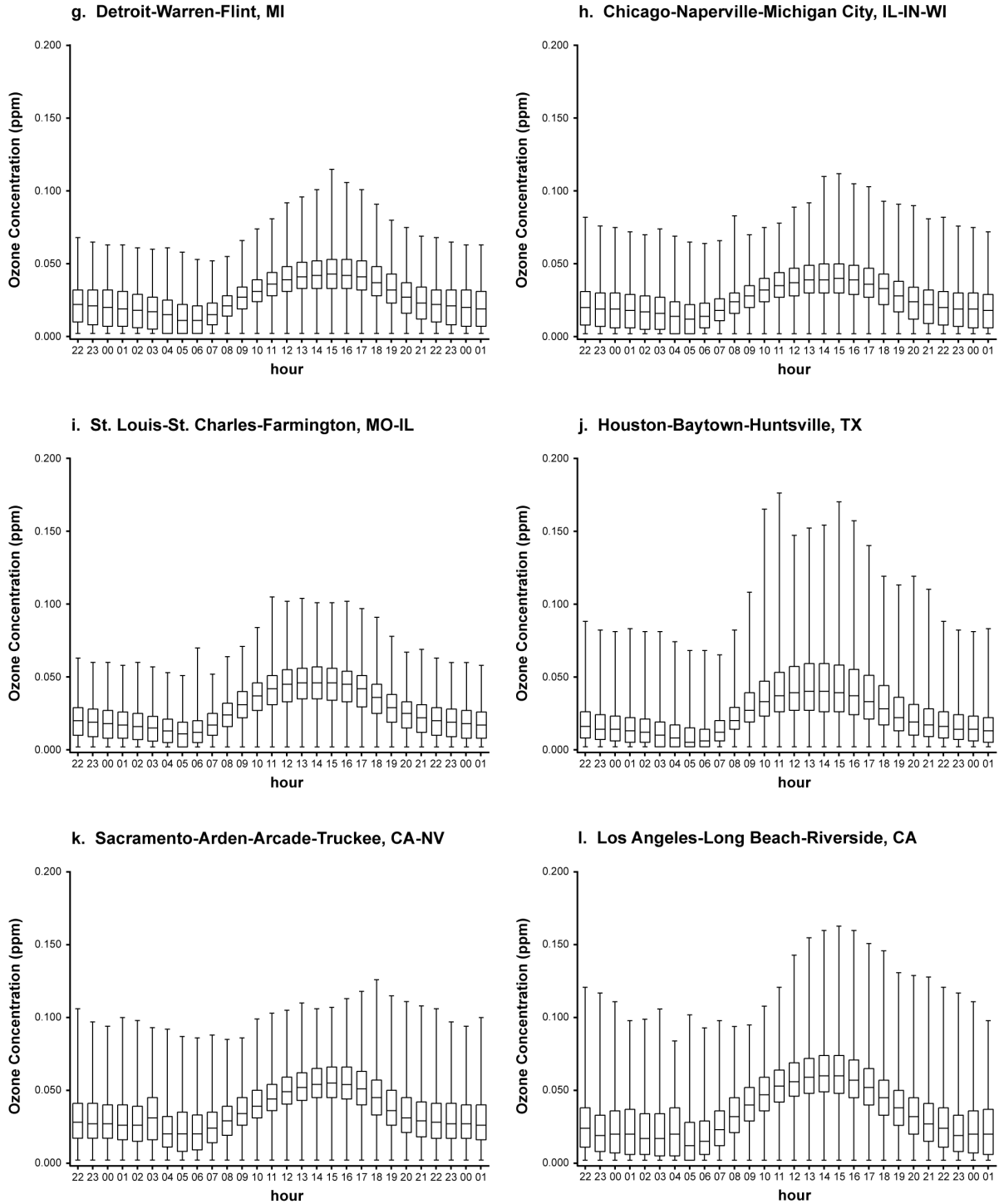


Figure 3-10g-l. Diurnal variability in hourly averaged O₃ in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

clear. Measurement issues may be involved, or there may be physical causes such as transport phenomena, as discussed in Chapter 2. As discussed in Chapter 2, and in greater detail in Section AX2.3.3, nocturnal low level jets are capable of producing secondary O₃ maxima at night.

The diurnal variability of O₃ averaged over 8 hours in the same twelve urban areas is shown in Figures 3-11a-l. The diurnal patterns of 1-h O₃ averages are broadly similar to 8-h averages. Typically, although the 8-h daily maximum occurs between 10 a.m. and 6 p.m., actual starting and ending times can differ from these characteristic times depending on location. For example, as shown in Figures 3-11a for Boston and 3-11k for Sacramento, the highest 8-h daily maximum values can start in mid-afternoon and extend into late evening. These results suggest that transport processes are playing the dominant role in determining the timing of the highest daily maxima in these areas.

On days with high 1-h daily maximum concentrations (e.g., ≥ 0.12 ppm), the maxima tend to occur in a smaller time window centered in the middle of the afternoon, compared to days on which the maximum is lower. For example, on high O₃ days the 1-h maximum occurs from about 11 a.m. to about 6 p.m. However, on days for which the 1-h daily maximum is ≤ 0.080 ppm, the daily maximum can occur at any time during the day or night, with only a 50% probability that it occurs between 1 and 3 p.m., in each of the 12 cities. (The time of day when the daily maximum 1-h O₃ concentration occurs is illustrated for four of the cities in Figures AX3-45a-d.). Photochemical reactions in combination with diurnal emissions patterns are expected to produce mid-afternoon peaks in urban areas. These results suggest that transport from outside the urban airshed plays a major role in determining the timing of the daily maxima for low peak O₃ levels. This pattern is typical for the Los Angeles-Long Beach-Riverside, CA area even on high O₃ days.

The same general timing patterns are found for 1-h daily maximum O₃ concentrations as for the daily maximum 8-h average O₃ concentration. As mentioned above, the daily maximum 8-h O₃ concentrations are generally found between the hours of 10 a.m. and 6 p.m. However, there are a significant number of days when this is not the case, e.g., for high values in Houston, TX and Los Angeles, CA, or in general for low values at any of the cities examined. (The time of day when the daily maximum 8-h average O₃ concentrations occurs is shown for four cities in Figures AX3-46a-d.). Although the 8-h average O₃ concentration is highly correlated with the

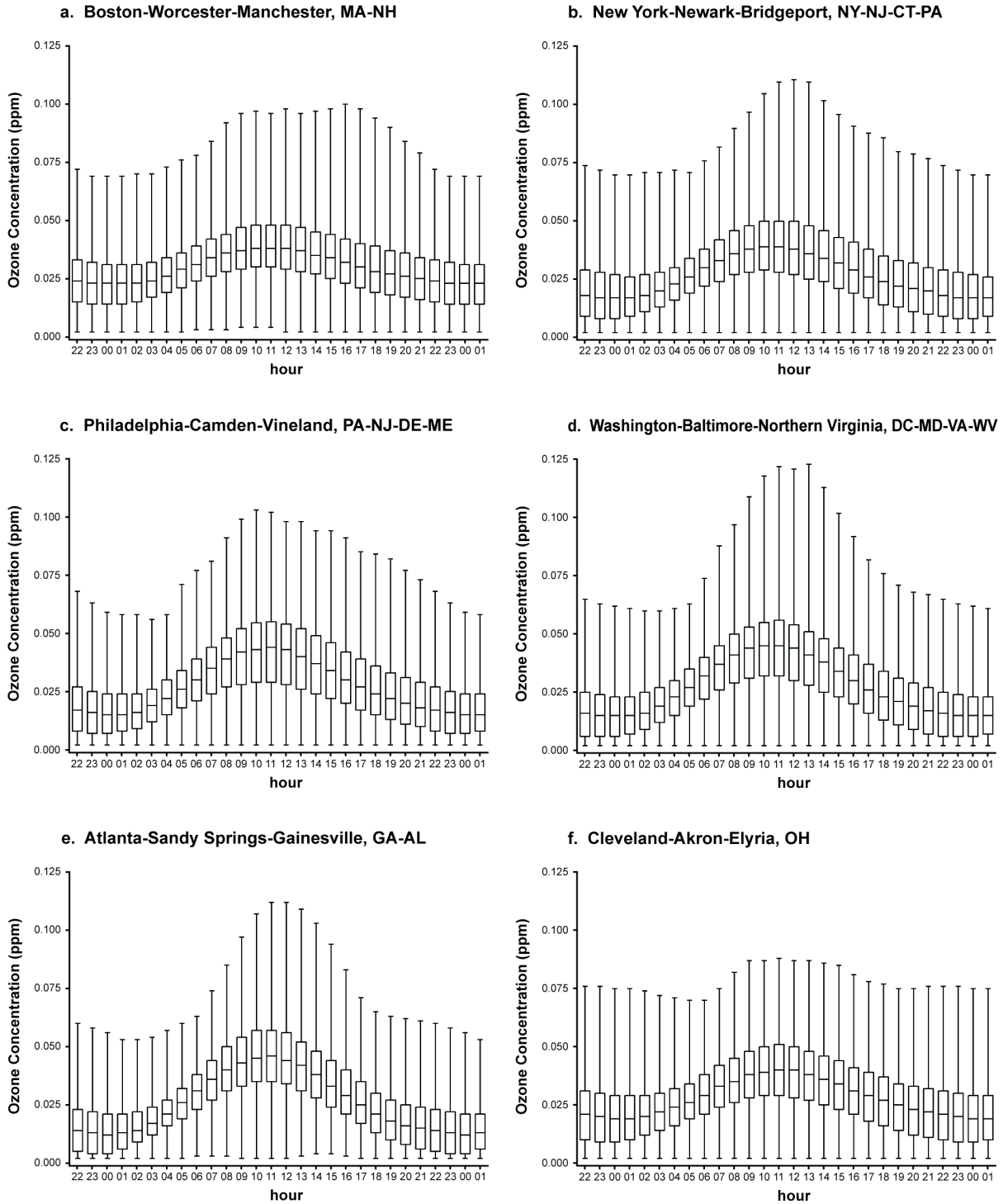


Figure 3-11a-f. Diurnal variability in 8-h O₃ in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

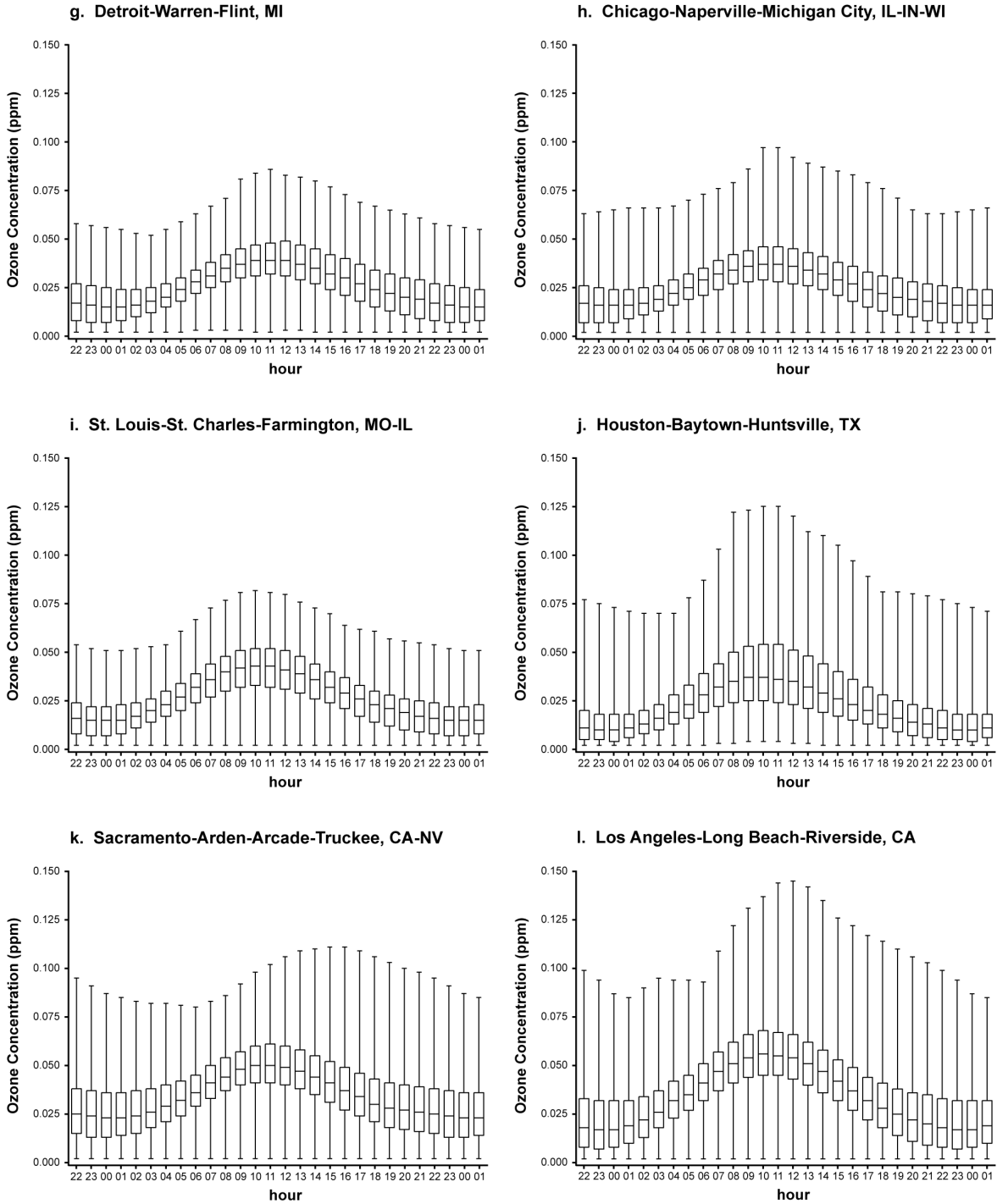


Figure 3-11g-l. Diurnal variability in 8-h O₃ in selected urban areas. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

daily maximum 1-h average O₃ concentration, there are situations where the daily maximum 8-h average O₃ concentration might be driven by very high values in the daily maximum 1-h average O₃ concentration as illustrated in Figure 3-10j for Houston, TX. In cases such as these, the predicted 8-h average may overestimate the short-term O₃ concentration later in the day.

The patterns of diurnal variability for both 1-h and 8-h averages have remained quite stable over the 15-year period from 1990 to 2004, with times of occurrence of the daily maxima varying by no more than an hour from year to year in each of the 12 cities.

Weekday/Weekend Differences

Differences in the diurnal behavior of O₃ have been observed in a number of cities (e.g., Heuss et al., 2003). Figures 3-12a-h show the contrast in the patterns of hourly averaged O₃ in the greater Philadelphia, Atlanta, Houston and Los Angeles areas from weekdays to weekends. Daily maximum concentrations occur basically at the same time on either weekdays or weekends. Differences are apparent in the hourly concentrations, especially in the extreme values. Weekday/weekend differences in 8-h average O₃ concentrations are shown in Figures 3-13a-h. As can be seen from a comparison of the weekend versus weekday patterns, there is a tendency for the lowest values in the distribution to be higher on weekends than on weekdays. Lower traffic volumes, in particular diesel truck traffic, lead to less NO emissions and titration of O₃ on weekends. The spike in values for Houston in midmorning, shown in Figure 3-12f, resulted from the release of highly reactive hydrocarbons from the petrochemical industry (which could occur on any day of the week). Otherwise, the maximum O₃ concentrations could be seen to occur on the weekdays as they do in Philadelphia and Atlanta, in contrast to Los Angeles. Indeed, the diurnal pattern in Houston is similar to that observed in Atlanta on weekdays, indicating some overall similarity in the sources of precursors of O₃.

Spatial Variability in Diurnal Patterns in Urban Areas

Daily maxima in either the 1-h or 8-h averages do not necessarily occur at the same time of day at each site in the 12 cities, and the diurnal pattern observed at individual sites can vary from the composites shown in Figures 3-8 and 3-9. Differences between sites are not only related to the distance between them; they also depend on nearby sources, such as highways, affecting one site to a greater extent than another. For example, in the Los Angeles basin, daily 1-h maxima

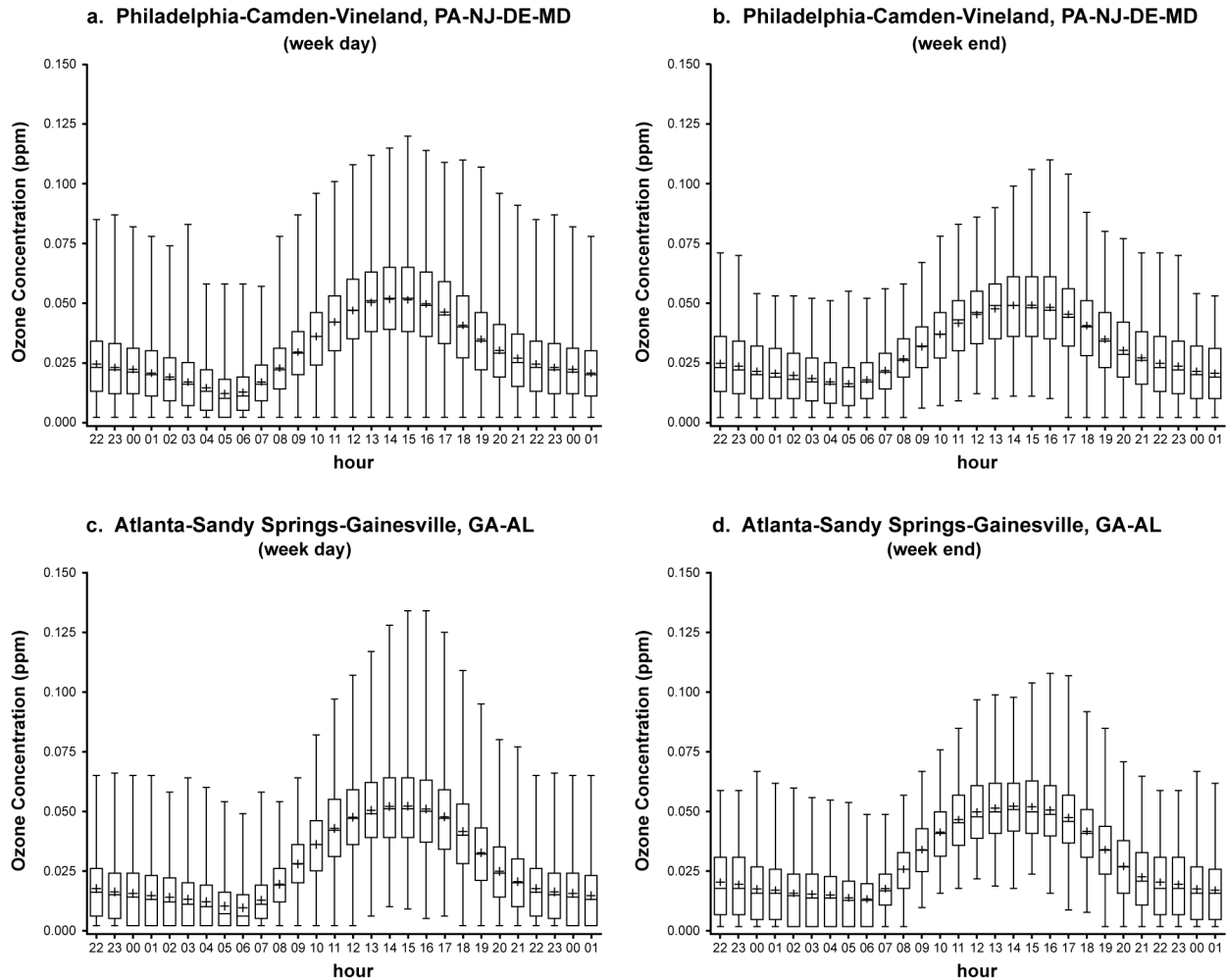


Figure 3-12a-d. Diurnal variations in hourly averaged O₃ on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

are reached in the late afternoon. However, in Riverside (which is typically located downwind of Los Angeles) there are sites in which the maximum is reached much earlier, because these sites are located close to highways.

The general pattern that emerges from the site-to-site variability within the urban areas examined is that peaks in 1-h average concentrations are higher and tend to occur later at downwind sites than in the urban cores. To the extent that monitoring sites are either near to or

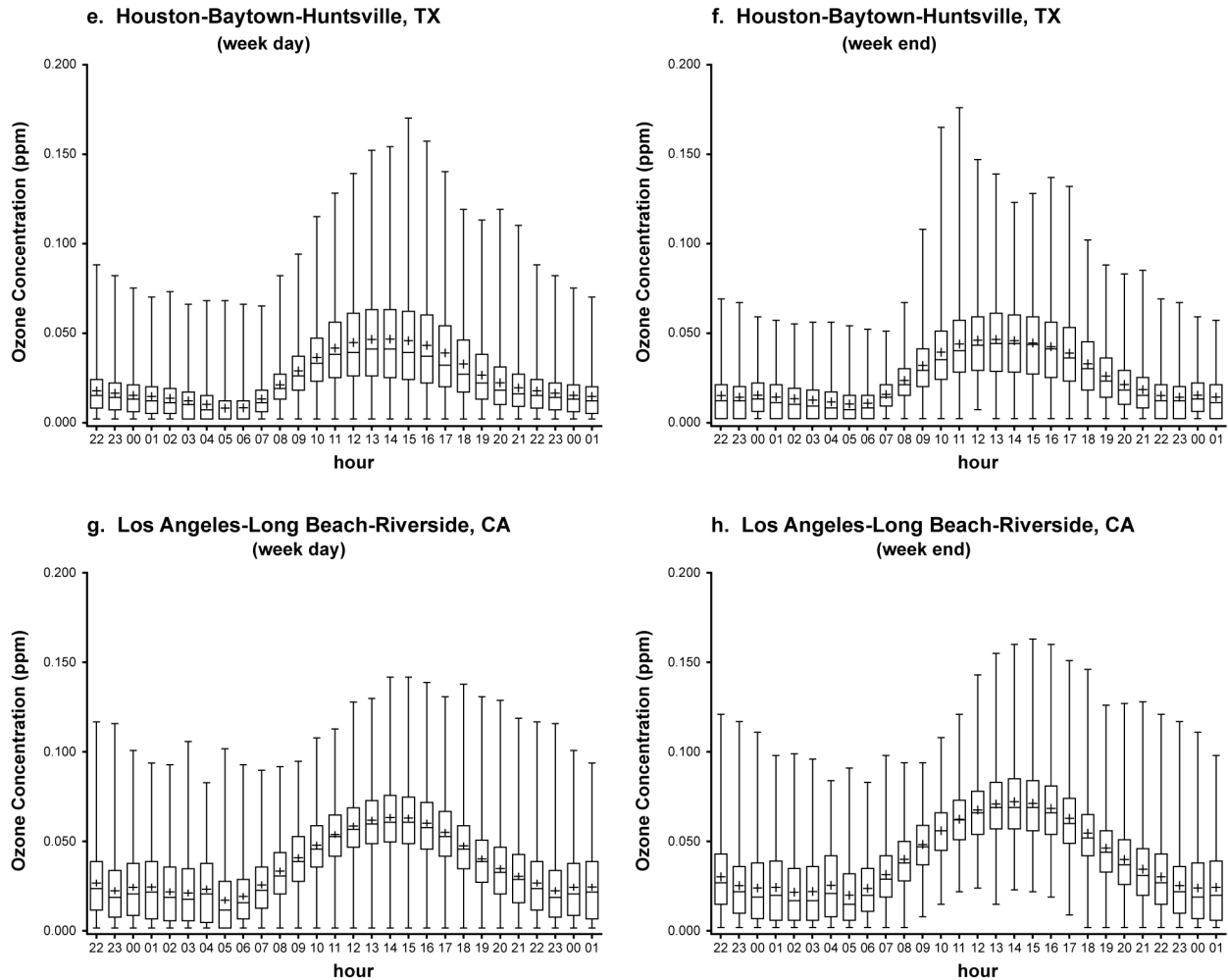


Figure 3-12e-h. Diurnal variations in hourly averaged O₃ on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

remote from sources of precursors in urban/suburban areas, the behavior of O₃ will follow these basic patterns. Similar relations are found for the 8-h average O₃ concentrations. Differences in diurnal patterns between sites in urban cores and sites downwind of urban cores are illustrated in Figures AX3-49a-b to AX3-51a-b for 1-h average O₃ and in Figures AX3-52a-b to AX3-54a-b for Detroit, MI, St. Louis, MO, and Riverside, CA areas.

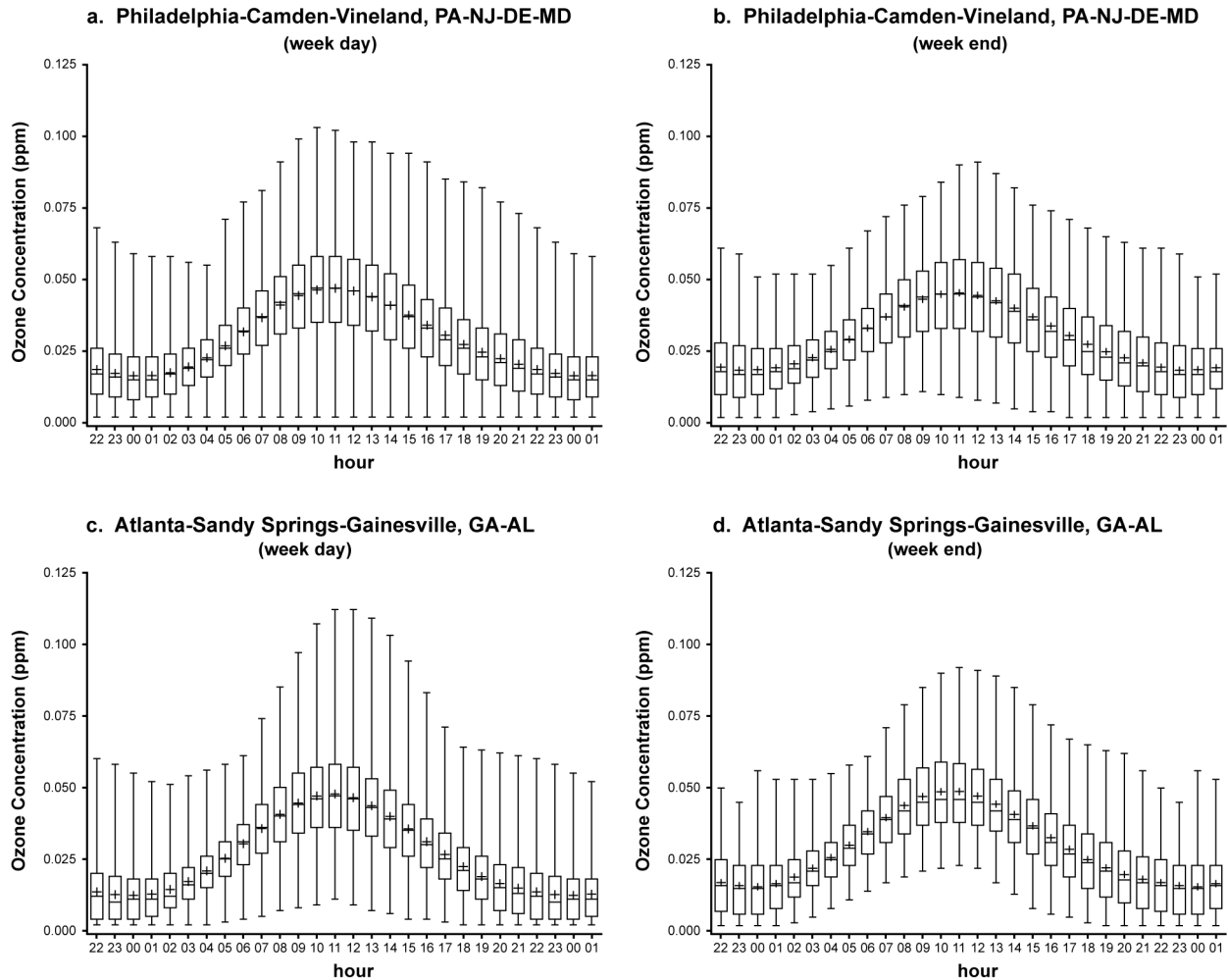


Figure 3-13a-d. Diurnal variations in 8-h average O_3 on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

Seasonal Variability

It should not be assumed that highest O_3 levels are confined to the summer. Highest average O_3 concentrations generally occur at background monitoring sites at midlatitudes in the Northern Hemisphere during late winter and spring versus summer, as for urban sites or for nonurban sites heavily affected by regional pollution sources.

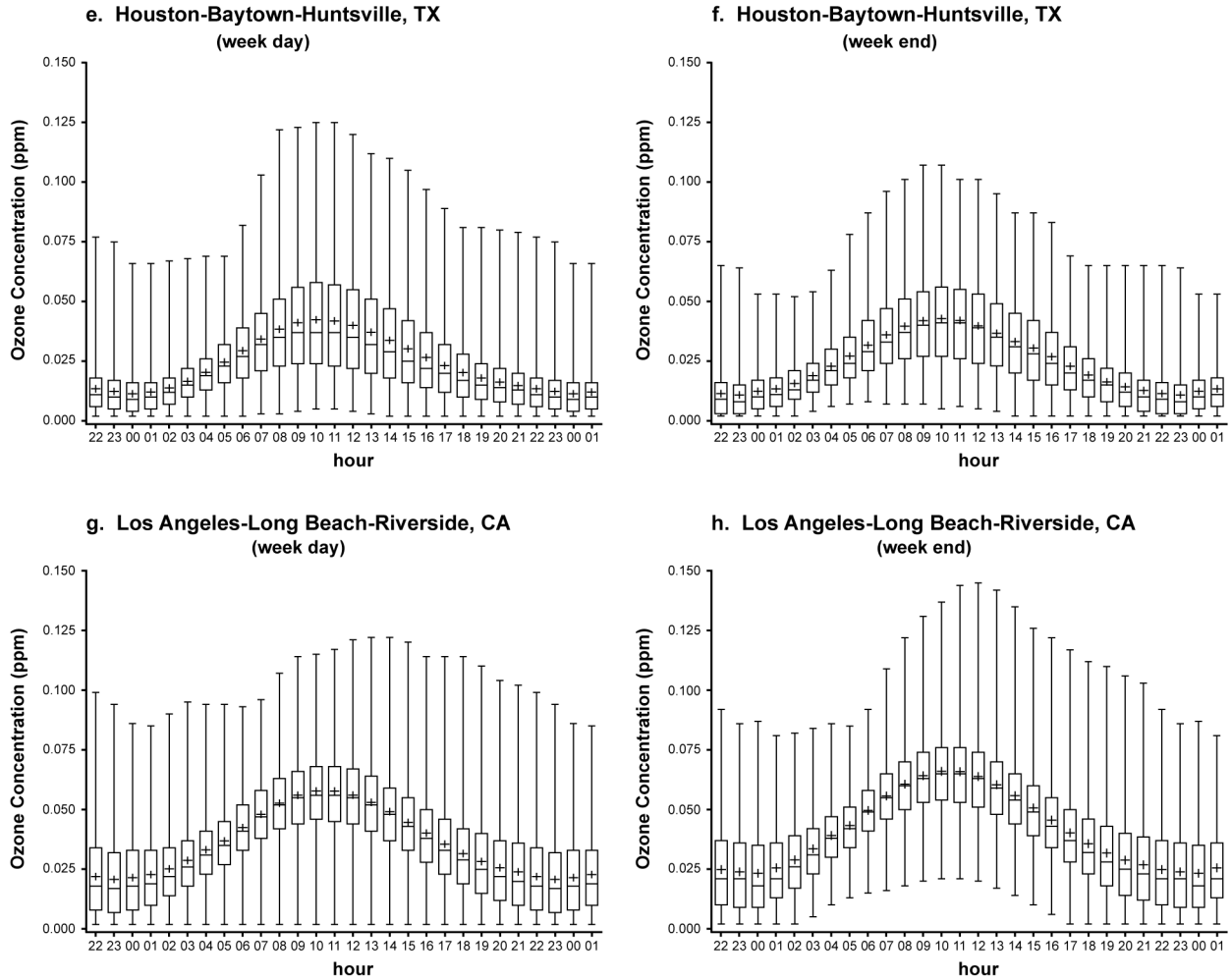


Figure 3-13e-h. Diurnal variations in 8-h average O₃ on weekdays and weekends in four cities. Values shown represent averages from May to September of 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

High O₃ values are also found at some of the 12 cities outside of summer. The seasonal behavior of O₃ varies across the 12 cities. In most northern cities, the extreme values of the daily maximum 8-h average O₃ concentration are a little more than half of those during the warm season, and the ratios of the medians are more similar as can be judged by comparison of Figures 3-11a-l with Figures 3-14a-l. Differences are even smaller for the southern cities. Indeed, some of the highest O₃ values are found in the Houston CSA outside of summer (Figure 3-14j).

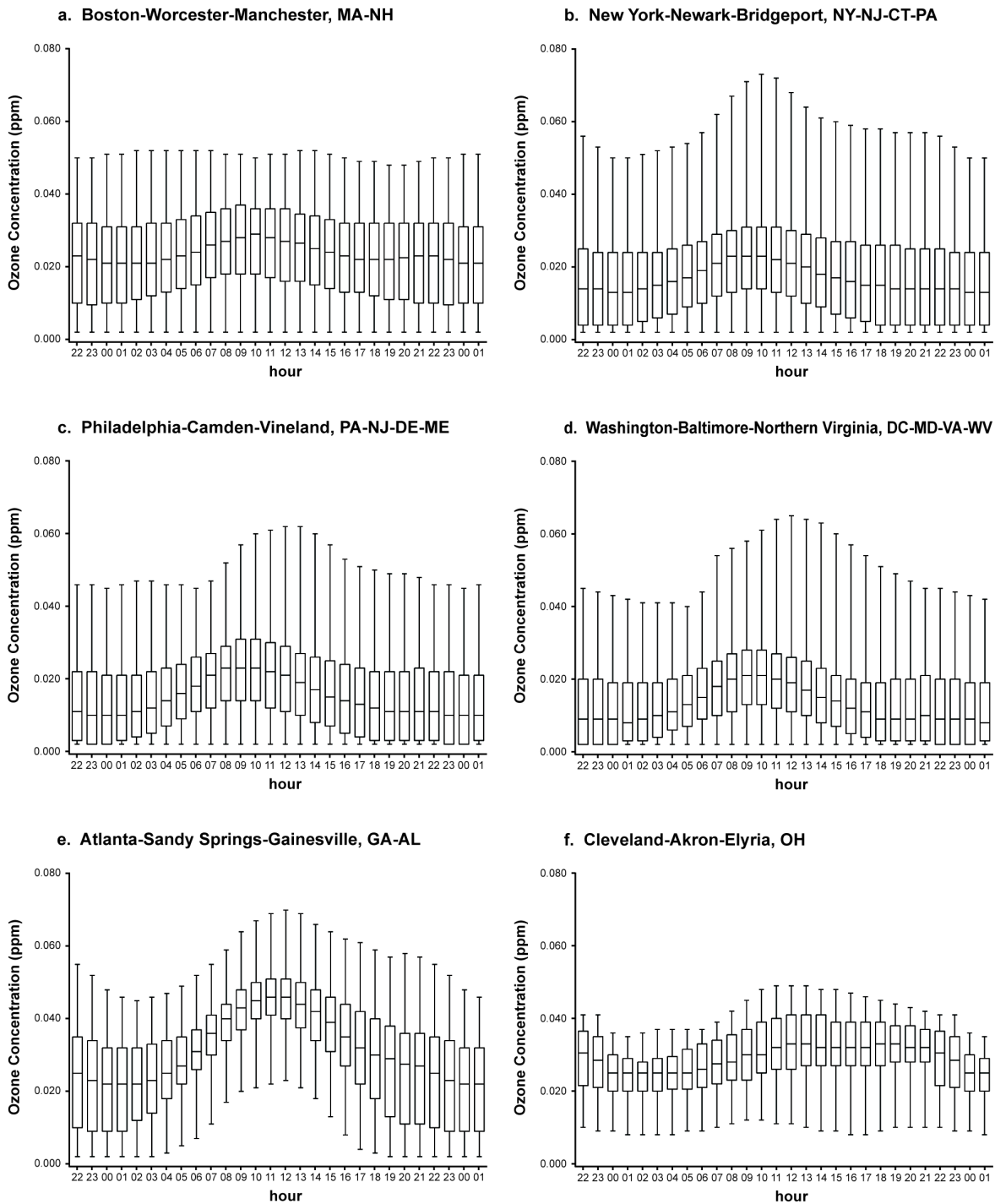


Figure 3-14a-f. Diurnal variability in 8-h average O₃ in selected urban areas. Values shown are averages from November to March 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

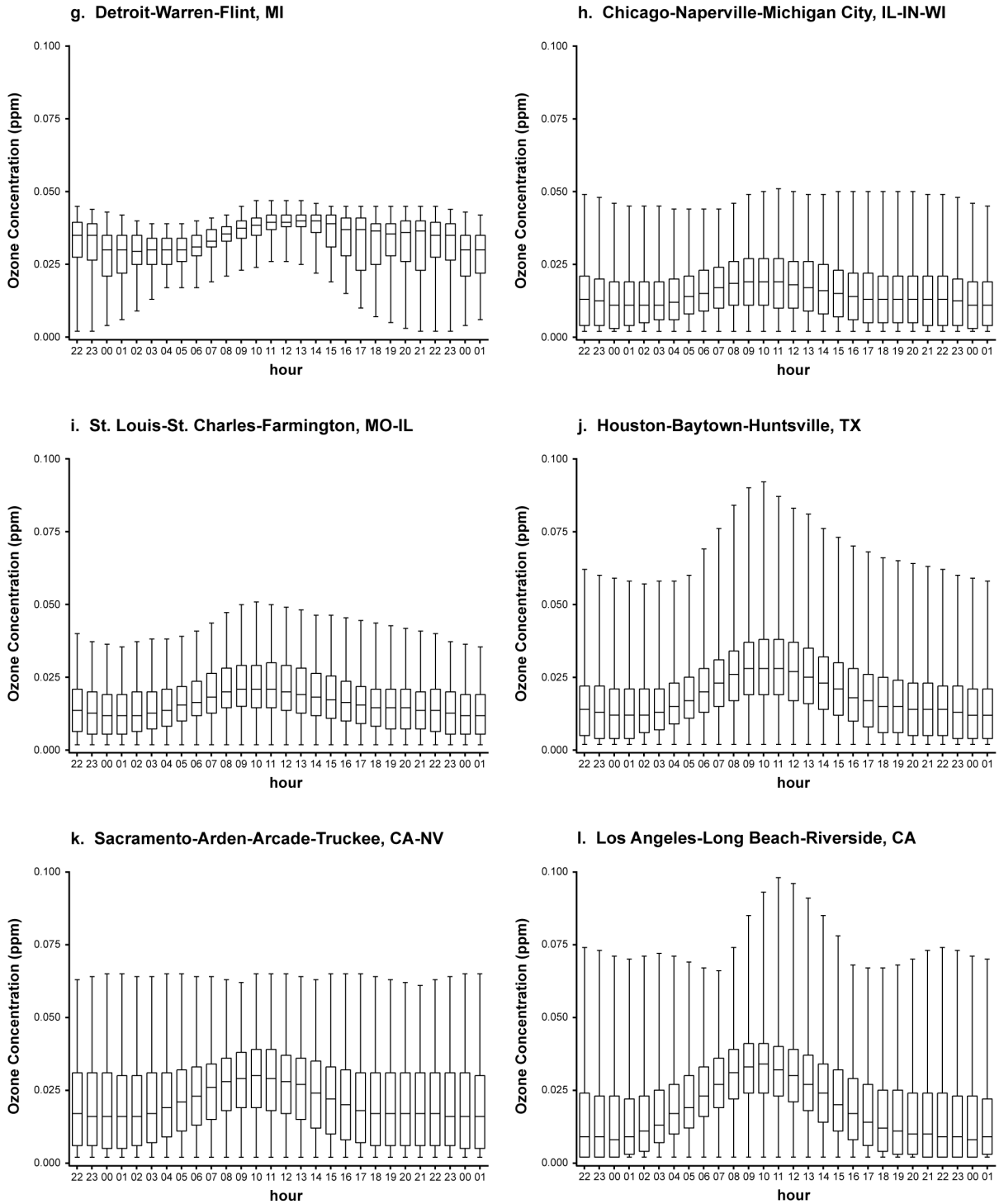


Figure 3-14g-l. Diurnal variability in 8-h average O₃ in selected urban areas. Values shown are averages from November to March 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

Diurnal Patterns in Nonurban Areas

Composite diurnal patterns of O₃ are shown in Figure 3-15 for hourly averaged O₃ and in Figure 3-16 for 8-hour average O₃ at rural (CASTNET) sites. As can be seen from a comparison of Figures 3-15 and 3-16 with Figures 3-8 and 3-9, diurnal patterns of O₃ are smoother and shallower at the rural sites than at the urban sites. Maxima in hourly average O₃ also tend to occur in afternoon. However, highest concentrations observed during any particular hour at night at the CASTNET sites (~0.130 ppm) are substantially higher than observed in urban areas (<0.100 ppm) and daily 1-h maxima at CASTNET sites have exceeded 0.150 ppm. The diurnal variations in 8-h average O₃ concentrations are also much smaller at the CASTNET sites than at the urban sites. Note also that the maxima in 8-h average O₃ concentrations are higher at the CASTNET sites than at the urban sites.

3.5 TRENDS IN OZONE CONCENTRATIONS

Year-to-year variability in the nationwide May to September, mean daily maximum 8-h O₃ concentrations are shown in Figure 3-17. The corresponding year-to-year variability in the 95th percentile concentrations is shown in Figure 3-18. Data flagged because of quality control issues were removed with concurrence of the local monitoring agency. Only days with data for 18 of 24 hours were kept, and a minimum of 115 of 153 days were required in each year. Missing years were filled in using simple linear interpolation, as done in EPA Trends reports. Year-to-year variability in the 95th percentile values of the daily maximum 8-h O₃ concentrations are shown in Figure 3-18. Sites considered in this analysis are shown in the map in Figure AX3-3. As was shown in Figures 3-1 and 3-2, most sites are located in the East. As can be seen from Figure 3-17, the highest O₃ concentrations have tended to decrease over the past 15 years, while there has been little change in O₃ concentrations near the center of the distribution. This is consistent with observations in Europe (Volz-Thomas et al., 2003). Mean O₃ concentrations were slightly lower in 2003 and 2004 than in earlier years. The summer of 2003 was slightly cooler than normal in the East (Levinson and Waple, 2004) and the summer of 2004 was much cooler than normal in the East (Levinson, 2005) accounting in part for the dip in O₃ during these two years. Observations of O₃ at a number of sites in the Northern Hemisphere likewise do not show convincing evidence of strong upward trends during the 1990s (Oltmans et al., 1998).

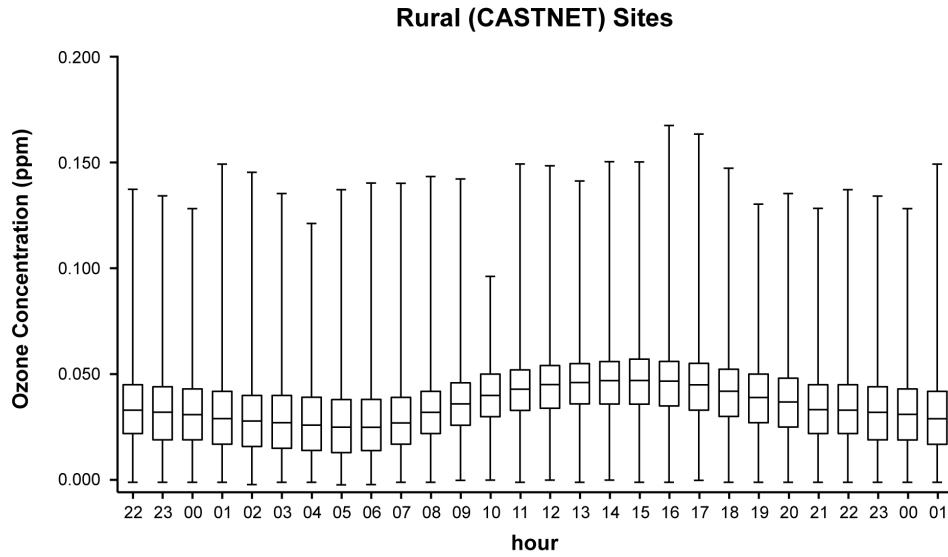


Figure 3-15. Composite diurnal variability in hourly O₃ concentrations observed at CASTNET sites. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima.

Source: Fitz-Simons et al. (2005).

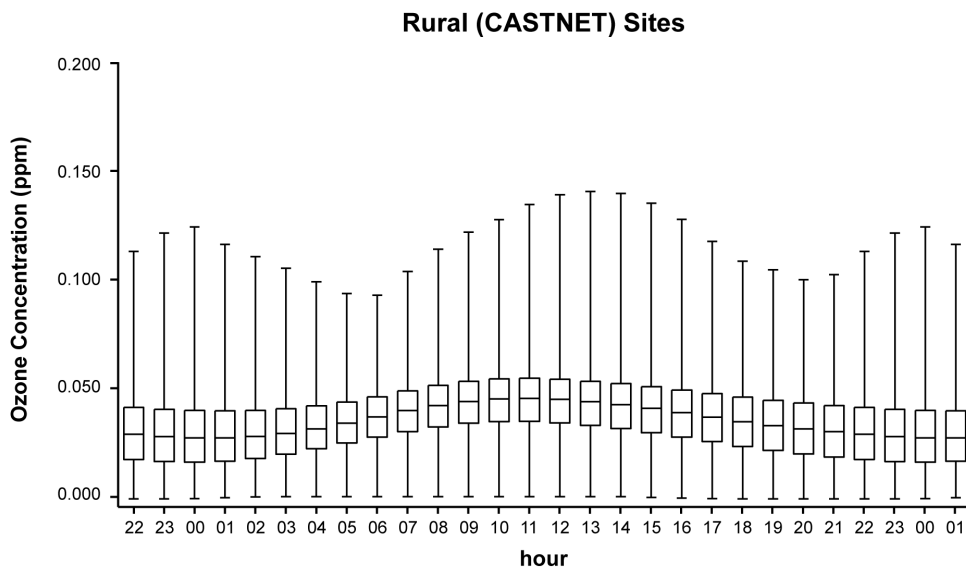


Figure 3-16. Composite diurnal variability in 8-h O₃ concentrations observed at CASTNET sites. Values shown are averages from April to October 2000 to 2004. Boxes define the interquartile range and the whiskers, the minima and maxima. The hour refers to the start of the 8-h averaging period.

Source: Fitz-Simons et al. (2005).

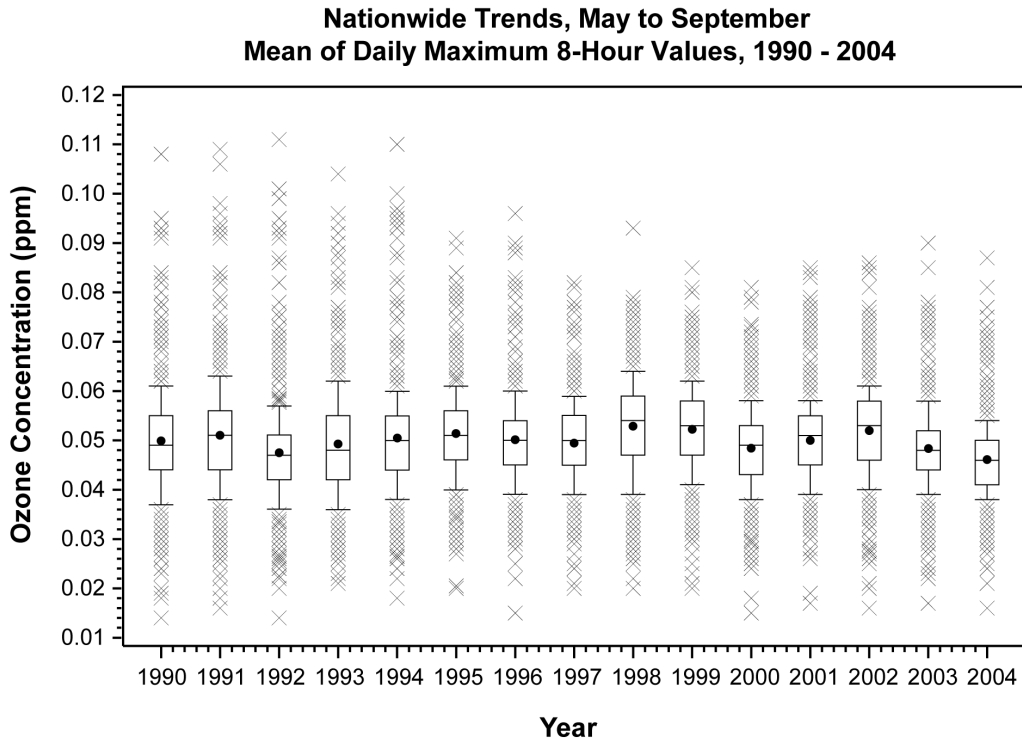


Figure 3-17. Year-to-year variability in nationwide mean daily maximum 8-h O₃ concentrations. The whiskers on the box plot represent the 10th and 90th percentile concentrations. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile concentrations. The dots inside the box represent the mean, for the statistic, at all sites.

Source: Fitz-Simons et al. (2005)

There may even have been a slight increase in O₃ concentrations near the bottom of the distribution throughout the monitoring period. This would be consistent with data obtained in Europe, showing that O₃ minima increased during the 1990s because of reduced titration of O₃ by reaction with NO in response to reductions in NO_x emissions. As a result, the concentration of O_x (NO₂ + O₃) shows little if any increase at all at the European sites (Volz-Thomas et al., 2003). Trends in compliance metrics such as the fourth highest daily maximum 8-h O₃ concentration can be found in the U.S. EPA Trends reports (<http://www.epa.gov/airtrends>) and so are not repeated here.

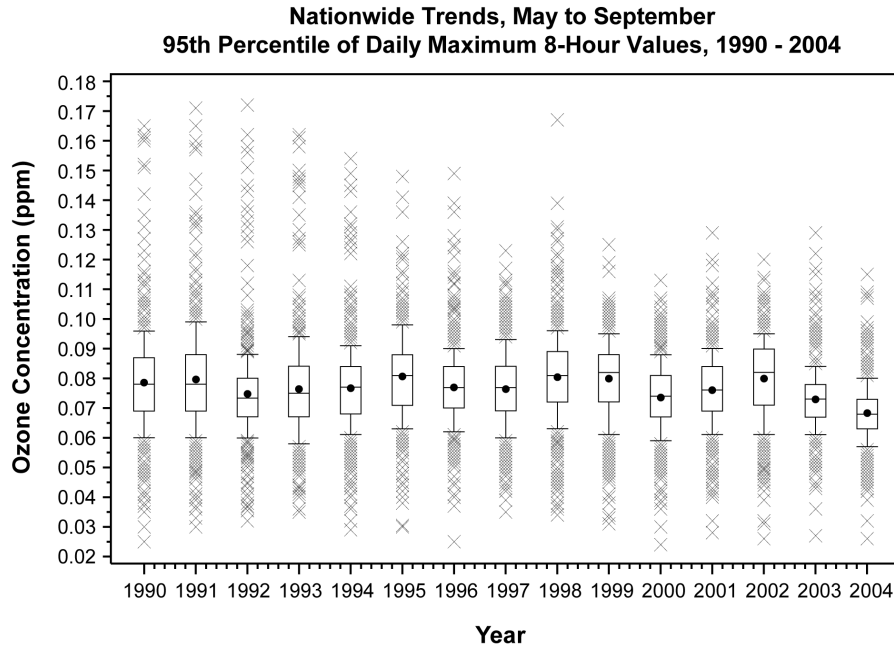


Figure 3-18. Year-to-year variability in nationwide 95th percentile value of the daily maximum 8-h O₃ concentrations. The whiskers on the box plot represent the 10th and 90th percentile values for the statistic. The “X”s above and below the whiskers are the values that fall below and above the 10th and 90th percentile values. The dots inside the box represent the mean, for the statistic, at all sites.

Source: Fitz-Simons et al. (2005).

Figures 3-19a-h show year-to-year variability in mean daily 8-h O₃ concentrations observed at selected national park sites across the United States. Figures 3-20a-h show year-to-year variability in the 95th percentile value of daily maximum 8-h O₃ concentrations at the same sites shown in Figures 3-19a-h. The same criteria used for calculating values in Figures 3-17 and 3-18 were used for calculating the May to September seasonal averages for the national parks shown in Figures 3-19a-h and 3-20a-h. Sites at 22 national parks met these criteria, and data for all 22 sites are given in Appendix AX3 in Figures AX3-66a-v and AX3-67a-v. However, several monitoring sites were moved during the period from 1990 to 2004. Sites were moved at Acadia NP in 1996, Joshua Tree NP in 1993, Mammoth Cave NP in 1996, Voyageurs NP in 1996, and Yellowstone NP in 1996. These moves often resulted in offsets in O₃ and so

May to September Mean of Daily Maximum 8-Hour Values, 1990 - 2004

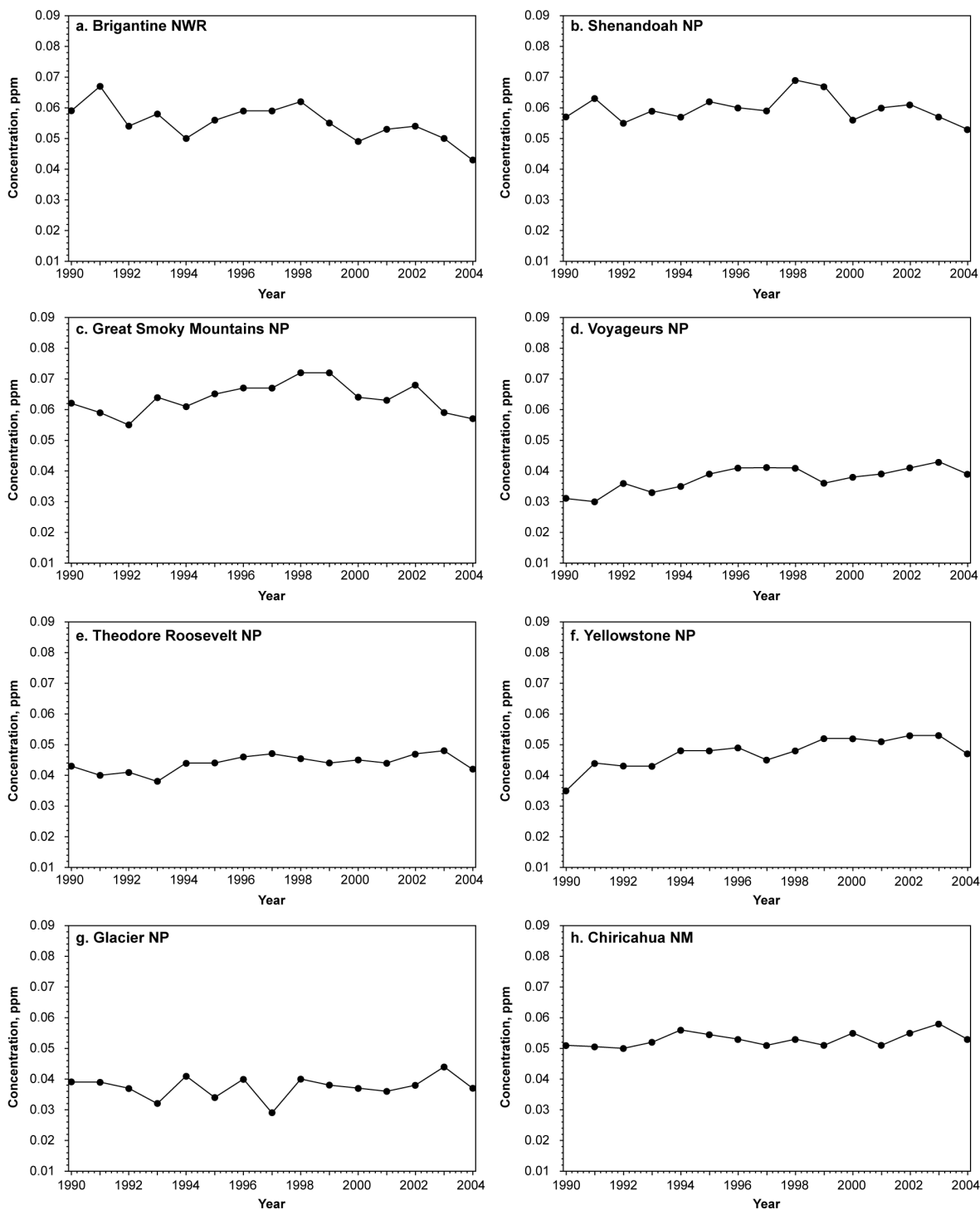


Figure 3-19a-h. Year-to-year variability in mean daily maximum 8-h O₃ concentrations at selected national park (NP), national wildlife refuge (NWR), and national monument (NM) sites.

Source: Fitz-Simons et al. (2005)

May to September 95th Percentile of Daily Maximum 8-Hour Values, 1990 - 2004

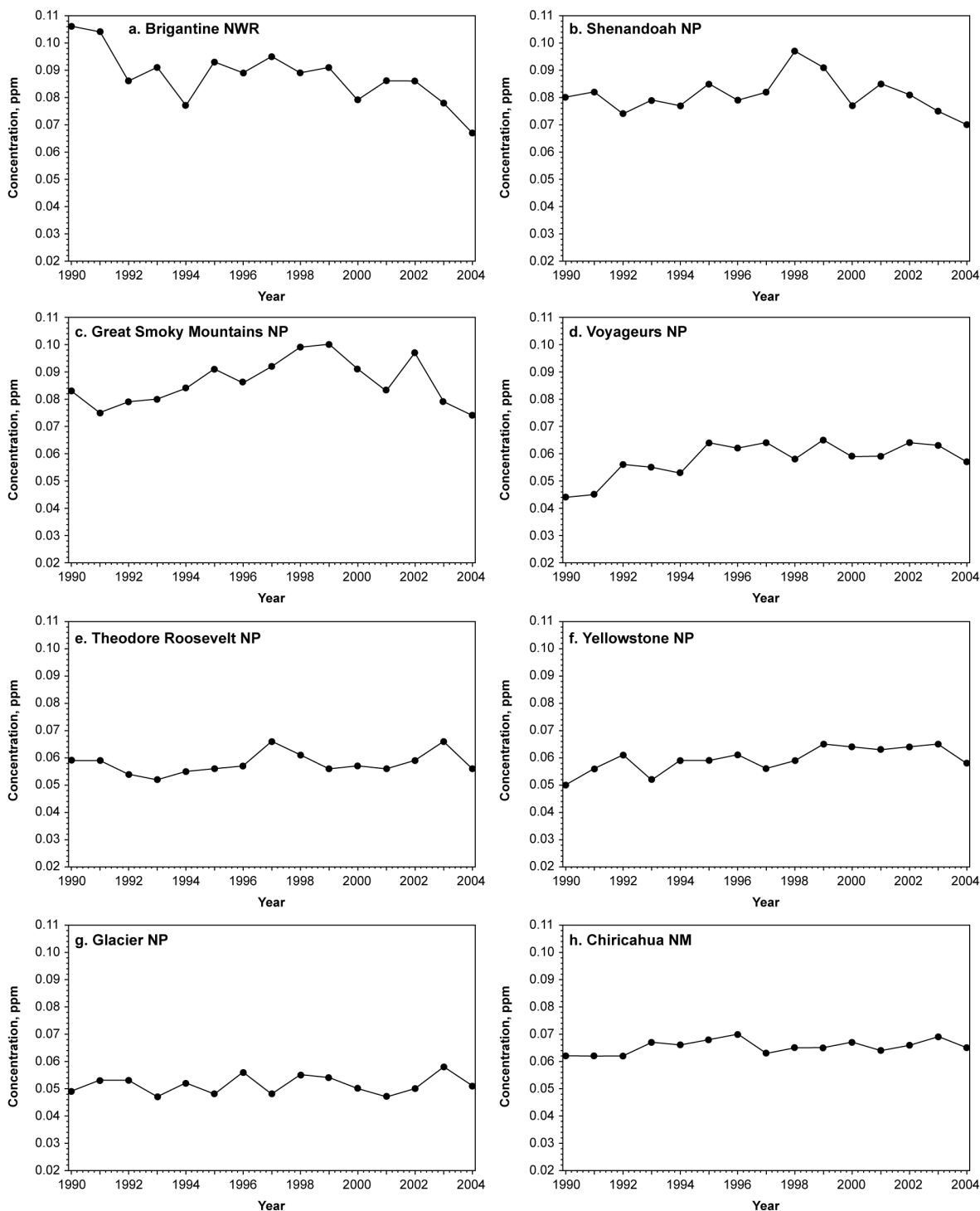


Figure 3-20a-h. Year-to-year variability in 95th percentile of daily maximum 8-h O₃ concentrations at selected national park (NP), national wildlife refuge (NWR), and national monument (NM) sites.

Source: Fitz-Simons et al. (2005).

trends for these locations have not been calculated (cf., Section AX3.6, Table AX3-9). As noted in *The Ozone Report—Measuring Progress through 2003* (U.S. Environmental Protection Agency, 2004b), O₃ trends in national parks in the South and the East are similar to nearby urban areas and reflect the regional nature of O₃ pollution. For example, O₃ trends in Charleston, SC and Charlotte, NC track those in nearby Cowpens NP and Cape Romaine NP in South Carolina; O₃ in Knoxville and Nashville, TN tracks O₃ in Great Smoky NP; O₃ in Philadelphia, PA and Baltimore, MD tracks Brigantine NP in New Jersey; and New York, NY and Hartford, CT track O₃ in Cape Cod NS. The situation is not as clear in the West, where some national parks are affected by local pollution sources (e.g., Lassen Volcanic National Park and Yosemite National Park, CA) more than others. However, data obtained at these sites still provide valuable information about the variability in regional background concentrations, especially since the West has not been broken down into regions as has been done by Lehman et al. (2004) for the East as shown in Figure 3-3. Comparison of Figures 3-19a-h and 3-20a-h shows that O₃ concentrations near the center of the distribution do not necessarily track those at the upper end, as pointed out earlier for the nationwide composite data set. Trends in the 98th and 95th percentiles, and mean O₃ concentrations for National Park sites presented in Table AX3-9 demonstrate this point. In addition, Table AX3-9 shows that trends reversed direction in going from the 98th to 95th percentile values.

Caution should be exercised in using trends calculated at national parks to infer contributions from distant sources either inside or outside of North America, because of the influence of regional pollution. For example, using a 15-year record of O₃ from Lassen Volcanic NP and data from two aircraft campaigns, and observations spanning 18 years from five U.S. west coast marine boundary layer sites, Jaffe et al. (2003) have estimated that the amount of O₃ in air arriving from the Eastern Pacific in spring has increased by approximately 10 ppb from the mid-1980s to the present. They suggested this positive trend might be due to increases of emissions of O₃ precursors in Asia. However, positive trends in O₃ were found during all seasons. Although the Lassen Volcanic NP site is not close to any major emission sources or urban centers, maximum hourly average O₃ concentrations of >0.080 ppm (during April-May) and >0.100 ppm (during the summer) occur at Lassen Volcanic NP, reflecting the influence of sources of O₃ precursors in the area. Thus, although there is evidence that O₃ levels may be

increasing at some rural locations, there is also evidence that O₃ levels at other locations have either not increased or have decreased over the same period.

3.6 RELATIONSHIPS BETWEEN OZONE AND OTHER SPECIES

Correlations between Ozone and Other Species

In order to understand relationships among atmospheric species, an important distinction must be made between primary (directly emitted) species and secondary (photochemically produced) species. In general, it is likely that primary species will be highly correlated with other primary species, and that secondary species will be highly correlated with other secondary species. By contrast, primary species are less likely to be correlated with secondary species. Secondary reaction products tend to correlate with each other, but there is considerable variation. Some species (e.g., O₃ and organic nitrates) are closely related photochemically and are highly correlated. Others (e.g., O₃ and H₂O₂) show a more complex correlation pattern. Further details are given in Annex AX3 in Section AX3.7.

Relationships between primary and secondary components are illustrated by considering data for O₃ and PM_{2.5}. Ozone and PM_{2.5} concentrations observed at a monitoring site in Fort Meade, MD are plotted as binned means for different intervals in Figure 3-21, based on data collected between July 1999 and July 2001. As can be seen from the figure, PM_{2.5} regarded as a function of O₃ increases to the left of the inflection point (at about 30 ppbv O₃) and also increases with O₃ to the right of the inflection point. Data to the left of the minimum in PM_{2.5} were collected mainly during the cooler months of the year, while data to the right of the minimum were collected during the warmer months. This situation arises because PM_{2.5} contains a large secondary component during the summer and has a larger primary component during winter. During the winter, O₃ comes mainly from the free troposphere, above the planetary boundary layer and, thus, may be considered a tracer for relatively clean air, and it is titrated by NO in the polluted boundary layer. Unfortunately, data for PM_{2.5} and O₃ are collected concurrently at relatively few U.S. sites throughout an entire year. So these results, while highly instructive, are not readily extrapolated to areas where appreciable photochemical activity occurs throughout the year. Ito et al. (2005) examined the relation between PM₁₀ and O₃ on a seasonal basis in several urban areas (cf., Figure 7-24). Although PM₁₀ contains proportionately more

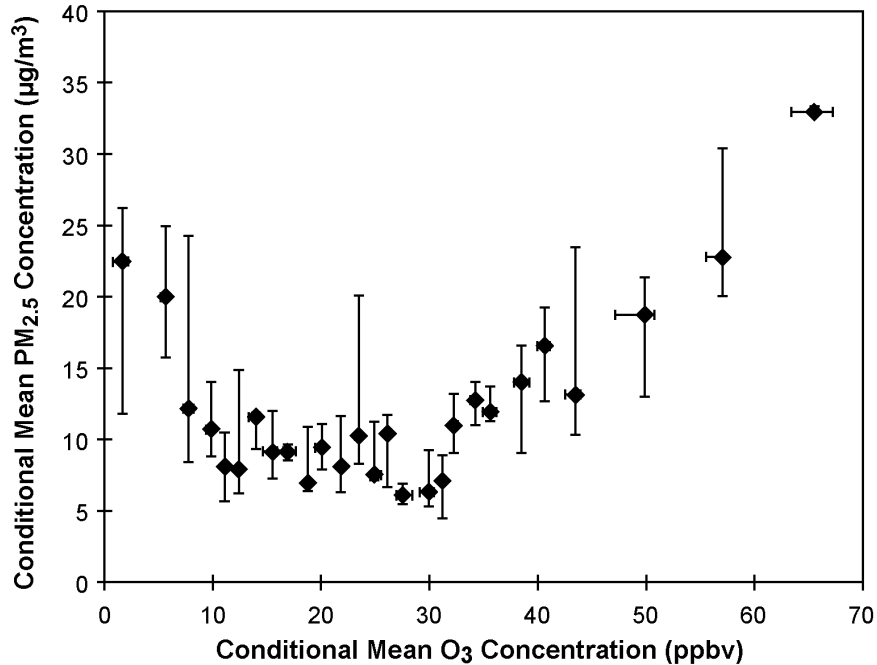


Figure 3-21. Binned mean PM_{2.5} concentrations versus binned mean O₃ concentrations observed at Fort Meade, MD from July 1999 to July 2001.

Source: Chen (2002).

primary material than does PM_{2.5}, relations similar to those shown in Figure 3-21 are found, reflecting the dominant contribution of PM_{2.5} to PM₁₀.

Other Oxidants

Measurements of gas phase peroxides in the atmosphere were reviewed by Lee et al. (2000). Ground level measurements of H₂O₂ taken during the 1970s indicated values of 180 ppb in Riverside, CA and 10 to 20 ppb during smog episodes in Claremont and Riverside, with values approaching 100 ppb in forest fire plumes. However, later surface measurements always found much lower values. For example, in measurements made in Los Angeles and nearby areas in the 1980s, peak values were always less than about 2 ppb and in a methods intercomparison study in Research Triangle Park, NC in June 1986, concentrations were <2.5 ppbv. Higher values ranging up to 5 ppb were found in a few other studies in Kinterbish, Alabama and Meadview, Arizona. Several of these studies found strong diurnal variations (typically about a

factor of three) with maximum values in the mid-afternoon and minimum values in the early morning. Mean concentrations of organic hydroperoxides at the surface at Niwot Ridge, CO in the summer of 1988 and State Park, GA during the summer of 1991 were all less than a few ppb.

Aircraft measurements of hydroperoxide (H_2O_2 , CH_3OOH and HOCH_2OOH) concentrations were made as part of the Southern Oxidants Study intensive campaign in Nashville, TN in July 1995 (Weinstein-Lloyd et al., 1998). The median concentration of total hydroperoxides in the boundary layer between 1100 and 1400 CDT was about 5 ppbv, with more than 50% contribution from organic hydroperoxides. Median O_3 was about 70 ppbv at the same time. The concentrations of the hydroperoxides depended strongly on wind direction with values about 40% lower when winds originated from the N/NW as opposed to the S/SW suggesting that local source areas were important.

Peroxyacetylnitrate (PAN) is produced during the photochemical oxidation of a wide range of VOCs in the presence of NO_x . It is removed by thermal decomposition and also by uptake to vegetation (Sparks et al., 2003; Teklemariam and Sparks, 2004). PAN is the dominant member of the broader family of peroxyacylnitrates (PANs) which includes as other significant atmospheric components peroxypropionyl nitrate (PPN) of anthropogenic origin, and peroxyacrylic nitrate (MPAN) produced from oxidation of isoprene. Measurements and models show that PAN in the United States includes major contributions from both anthropogenic and biogenic VOC precursors (Horowitz et al., 1998; Roberts et al., 1998). Measurements in Nashville during the 1999 summertime Southern Oxidants Study (SOS) showed PPN and MPAN amounting to 14% and 25% of PANs respectively (Roberts et al., 2002). Measurements during the TexAQS 2000 study in Houston indicated PAN concentrations of up to 6.5 ppbv (Roberts et al., 2003). PAN measurements in southern California during the SCOS97-NARSTO study indicated peak concentrations of 5-10 ppbv, which can be contrasted to values of 60 to 70 ppbv measured back in 1960 (Grosjean, 2003). Vertical profiles measured from aircraft over the U.S. and off the Pacific coasts show PAN concentrations above the boundary layer of only a few hundred pptv, although there are significant enhancements associated with long-range transport of pollution plumes from Asia (Kotchenruther et al., 2001a; Roberts et al., 2004). Decomposition of this anthropogenic PAN as it subsides over North America can lead to significant O_3 production, enhancing the O_3 background (Kotchenruther et al., 2001b; Hudman et al., 2004).

Oxidants are also present in airborne cloud droplets, rain drops and particulate matter. Measurements of hydroperoxides, summarized by Reeves (2003), are available mainly for hydrometeors, but are sparse for ambient particles. Venkatachari et al. (2005a) sampled the concentrations of total reactive oxygen species (ROS) in particles using a cascade impactor in Rubidoux, CA during July 2003. Although the species constituting ROS were not identified, the results were reported in terms of equivalent H_2O_2 concentrations. Unlike O_3 and gas phase H_2O_2 which show strong diurnal variability (i.e., about a factor of three variation between afternoon maximum and early morning minimum), the diurnal variation of particle phase ROS was found to be much weaker (i.e., less than about 20%) at least for the time between 8 a.m. and midnight. Because the ROS were measured in the fine aerosol size fraction, which has a lifetime with respect to deposition of much greater than a day, little loss is expected but their concentrations might also be expected to increase because of nighttime chemistry, perhaps involving NO_3 radicals. The ROS concentration, about $7 \times 10^{-9} \text{ M/m}^3$ (expressed as equivalent H_2O_2), was at most 1% that of O_3 (6.2 to $38 \times 10^{-7} \text{ M/m}^3$ or 15 to 90 ppb), with highest values at night. In a companion study conducted in Queens, NY during January and early February 2004, Venkatachari et al. (2005b) found much lower concentrations of ROS of about $1.2 \times 10^{-9} \text{ M/m}^3$. However, O_3 levels were also substantially lower, but ROS concentrations were still less than 1% those of O_3 . It is of interest to note that gas phase OH concentrations measured at the same time ranged from about $7.5 \times 10^4/\text{cm}^3$ to about $1.8 \times 10^6/\text{cm}^3$, implying the presence of significant photochemical activity even in New York City during winter.

Co-Occurrence of Ozone with Other Pollutants

The characterization of co-occurrence patterns under ambient conditions is important for relating human health and vegetation effects under ambient conditions to controlled research results as described in Annex AX3.8. Several attempts have been made to characterize gaseous air pollutant mixtures. The previous 1996 O_3 AQCD discussed various patterns of pollutant mixtures of SO_2 , NO_2 , and O_3 . Pollutant combinations can occur at or above a threshold concentration at either the same or different times.

The 1996 O_3 AQCD noted that studies of the joint occurrence of gaseous NO_2/O_3 and SO_2/O_3 reached two conclusions: (1) hourly simultaneous and daily simultaneous-only co-occurrences are fairly rare (when both pollutants were present at an hourly average

concentration ≥ 0.05 ppm) and (2) when co-occurrences are present, complex-sequential and sequential-only co-occurrence patterns predominate. Year-to-year variability was found to be insignificant.

Using 2001 hourly data for O_3 and NO_2 and for O_3 and SO_2 , the co-occurrence patterns for the data are similar to those of previous studies. As shown in Figure 3-22, fewer than 10 co-occurrences of O_3 and NO_2 were found for most of the collocated monitoring sites. Likewise, Figure 3-23 shows that fewer than 10 co-occurrences of O_3 and SO_2 were found for most of the collocated monitoring sites analyzed.

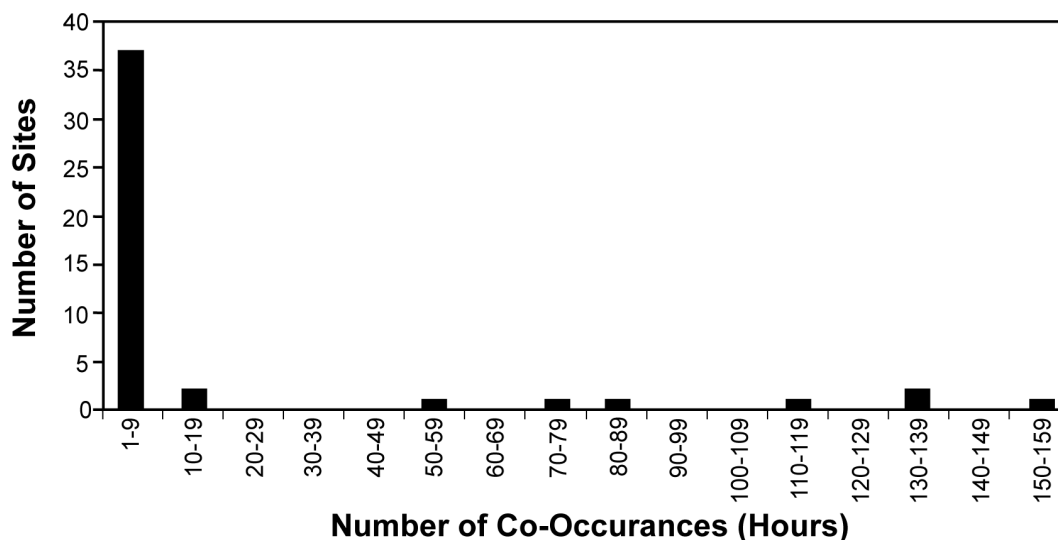


Figure 3-22. The co-occurrence pattern for O_3 and nitrogen dioxide using 2001 data from the AQS. There is co-occurrence when hourly average concentrations of O_3 and another pollutant are both ≥ 0.05 ppm.

Since 1999, monitoring stations across the United States have routinely measured 24-h average concentrations for $PM_{2.5}$. Daily co-occurrence of $PM_{2.5}$ and O_3 over a 24-h period was also characterized. Because $PM_{2.5}$ data are mostly summarized as 24-h average concentrations in the AQS database, a daily co-occurrence of O_3 and $PM_{2.5}$ was subjectively defined as an hourly average O_3 concentration ≥ 0.05 ppm and a $PM_{2.5}$ 24-h concentration $\geq 40 \mu\text{g}/\text{m}^3$ (corresponding to the EPA Air Quality Index, Level of Concern for $PM_{2.5}$) occurring during the same 24-h

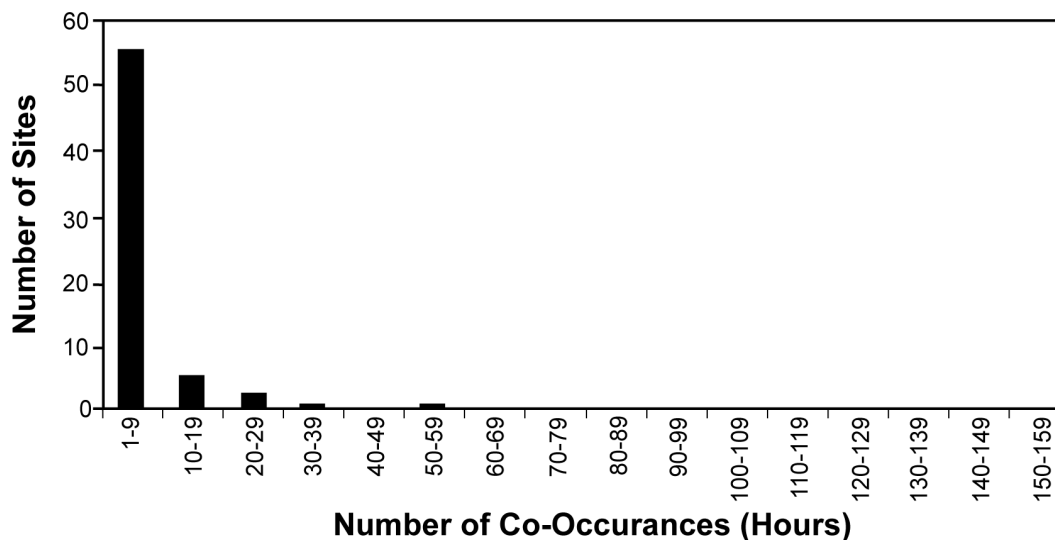


Figure 3-23. The co-occurrence pattern for O₃ and sulfur dioxide using 2001 data from AQS. There is co-occurrence when hourly average concentrations of O₃ and another pollutant are both ≥ 0.05 ppm.

period. Using 2001 data from the AQS database, the daily co-occurrence of PM_{2.5} and O₃ was infrequent (Figure 3-24). Only limited data are available on the co-occurrence of O₃ and other pollutants (e.g., acid precipitation and acidic cloudwater). In most cases, routine monitoring data are not available from which to draw general conclusions.

3.7 POLICY RELEVANT BACKGROUND OZONE CONCENTRATIONS

Background O₃ concentrations used for purposes of informing decisions about NAAQS are referred to as Policy Relevant Background (PRB) O₃ concentrations. Policy Relevant Background concentrations are those concentrations that would occur in the United States in the absence of anthropogenic emissions in continental North America (defined here as the United States, Canada, and Mexico). Policy Relevant Background concentrations include contributions from natural sources everywhere in the world and from anthropogenic sources outside these three countries. Background levels so defined facilitate separation of pollution levels that can be controlled by U.S. regulations (or through international agreements with neighboring countries)

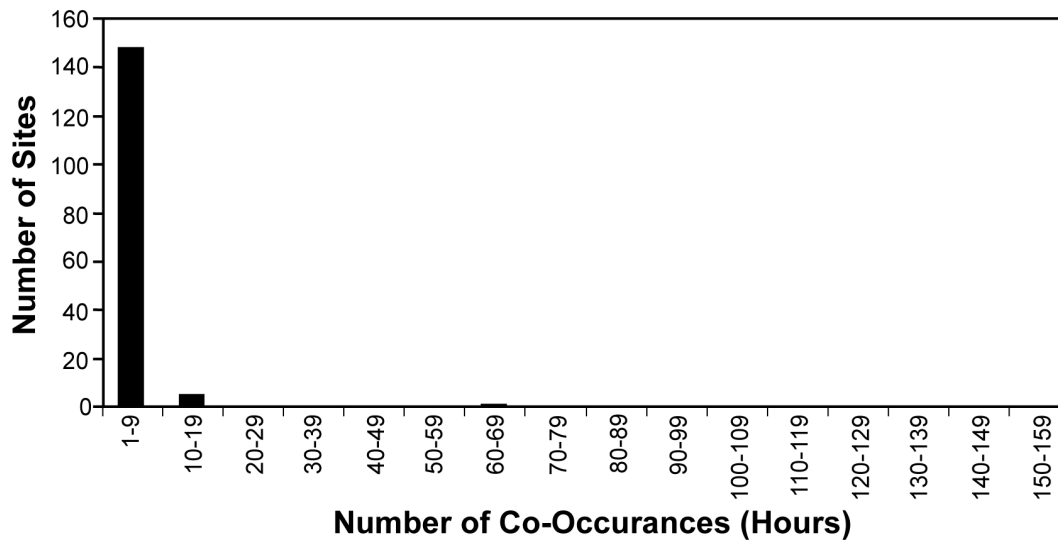


Figure 3-24. The co-occurrence pattern for O₃ and PM_{2.5} using 2001 data from AQS.

from levels that are generally uncontrollable by the United States. EPA assesses risks to human health and environmental effects from O₃ levels in excess of PRB concentrations. Issues concerning the methodology for estimating PRB O₃ concentrations are described in detail in Annex AX3, Section AX3.9.

Contributions to PRB O₃ include photochemical actions involving natural emissions of VOCs, NO_x, and CO as well as the long-range transport of O₃ and its precursors from outside North America and the stratospheric-tropospheric exchange (STE) of O₃. Processes involved in STE are described in detail in Annex AX2.3. Natural sources of O₃ precursors include biogenic emissions, wildfires, and lightning. Biogenic emissions from agricultural activities are not considered in the formation of PRB O₃.

Springtime maxima are observed at relatively remote (Annex AX3 and Figures 3-25a,b) national park sites, located mainly in the western United States and at a number of other relatively unpolluted monitoring sites throughout the Northern Hemisphere. The major issues concerning the calculation of PRB O₃ center on the capability of the current generation of global-scale, three-dimensional, CTMs to simulate the causes of high O₃ concentrations observed at monitoring sites in relatively unpolluted areas of the United States from late winter through spring (i.e., February through June). The issues raised do not affect interpretations of the causes

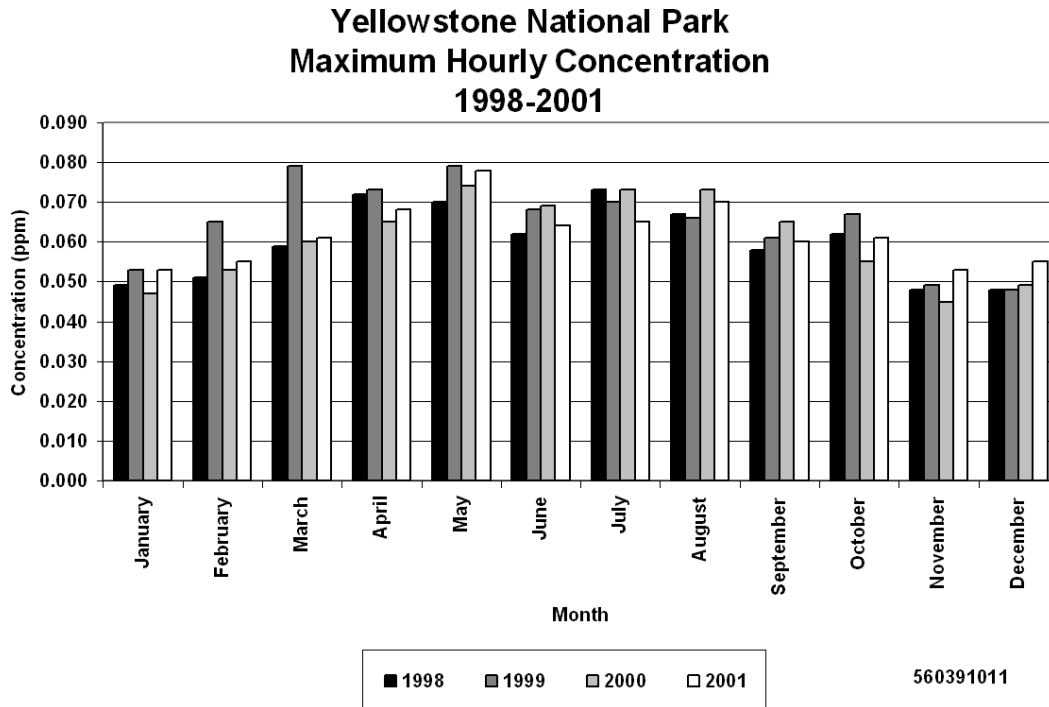


Figure 3-25a. Monthly maximum hourly average O₃ concentrations at Yellowstone National Park (WY) in 1998, 1999, 2000, and 2001.

Source: U.S. Environmental Protection Agency (2003).

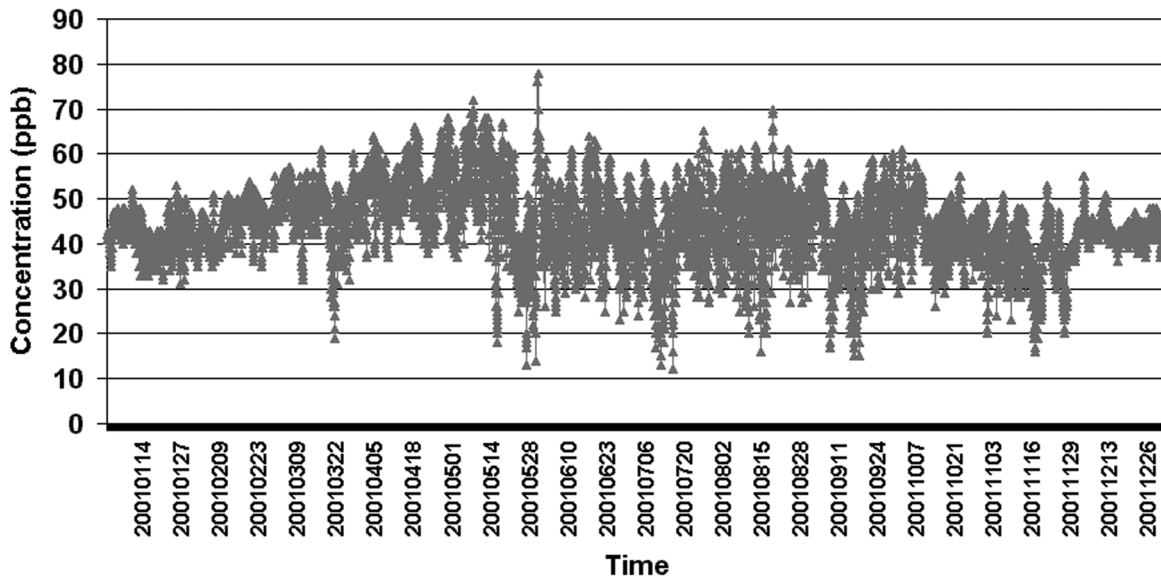


Figure 3-25b. Hourly average O₃ concentrations at Yellowstone National Park (WY) for the period January to December 2001.

Source: U.S. Environmental Protection Agency (2003).

of summertime O₃ episodes as strongly. Summertime O₃ episodes are mainly associated with slow- moving high-pressure systems characterized by limited mixing between the planetary boundary layer and the free troposphere, as noted in Annex AX2, Section AX2.3.

A large number of case studies document the occurrence of STE mainly during winter and spring in mid- and high-latitudes in Europe, Asia, and North America. These studies were based on aircraft, satellite, and ground-based measurements. Considerable uncertainty exists in the magnitude of the exchange; however, these studies have found that STE occurs throughout the year, but with a distinct preference for the transport of O₃ directly to the middle and lower troposphere during late winter and spring. Transport to the upper troposphere occurs throughout the year.

Springtime maxima in tropospheric O₃ observed at high latitudes are also associated with the winter buildup of O₃ precursors and thermally labile reservoir species, such as PAN and other reactive nitrogen species. These pollutants originate from all continents in the Northern Hemisphere. Ozone precursor concentrations reach a maximum in late March; and as sunlight returns to the Arctic, photochemical reactions generate tropospheric O₃ (Section AX3.9.1). The contribution of Asian sources to the U.S. levels is also largest during spring, reflecting the efficient lifting of Asian pollution ahead of cold fronts originating in Siberia and transport by strong westerly winds across the Pacific (e.g., Hudman et al., 2004). The longer lifetime of O₃ during spring also contributes to springtime maxima (Wang et al., 1998).

Estimates of PRB concentrations cannot be obtained solely by examining measurements of O₃ obtained at RRMS in the United States (Annex AX3, Section AX3.2.3) because of the long-range transport from anthropogenic source regions within North America. It should also be noted that it is impossible to determine sources of O₃ without ancillary data that could be used as tracers of sources or to calculate photochemical production and loss rates. The current definition of PRB implies that only CTMs can be used to estimate the range of PRB values. On the synoptic and larger spatial scales at least, all evidence indicates that global CTMs are adequate tools to investigate the factors controlling tropospheric O₃; and three-dimensional CTMs, as typified by Fiore et al. (2003) appear to offer the best methodology for estimating PRB concentrations that cannot be measured directly (Annex AX3, Section AX3.9.2), at least for averaging periods of longer than one hour.

Previous estimates of background O₃ concentrations, based on different concepts of background, are given in Table 3-2. Results from global three-dimensional CTMs, where the background is estimated by zeroing anthropogenic emissions in North America (Table 3-8) are on the low end of the 25 to 45 ppbv range. Lefohn et al. (2001) have argued that frequent occurrences of O₃ concentrations above 50 to 60 ppbv at remote northern U.S. sites in spring are mainly stratospheric in origin. Fiore et al. (2003) used a global CTM to determine the origin of the high-O₃ events reported by Lefohn et al. (2001), and to conduct a more general quantitative analysis of background O₃ as a function of season, altitude, and local O₃ concentration.

Table 3-2. Previous Estimates of Background O₃ in Surface Air Over the United States

Study	Method	Time Period	Region	Background Estimate (ppbv)
Trainer et al. (1993)	y-intercept of O ₃ vs. NO _y -NO _x regression line ^a	Summer 1988	Eastern United States	30-40 ^b
Hirsch et al. (1996)	y-intercept of O ₃ vs. NO _y -NO _x regression line	May-Sep 1990-1994	Harvard Forest ^c	25 (Sept) – 40 (May) ^d
Altshuller and Lefohn (1996)	y-intercept of O ₃ vs. NO _y regression line, and observations at remote/rural sites	Apr-Oct 1988-1993	Continental United States	25-45 (inland) ^e 25-35 (coastal)
Liang et al. (1998)	Sensitivity simulation in a 3-D model with anthropogenic NO _x emissions in the continental U.S. set to zero	Full year	Continental United States	20-30 (East) ^f 20-40 (West) (spring maximum)
Lin et al. (2000)	Median O ₃ values for the lowest 25th percentiles of CO and NO _y concentrations	1990-1998	Harvard Forest	35 (fall) – 45 (spring) ^g
Fiore et al. (2002)	O ₃ produced outside of the North American boundary layer in a global 3-D model	Summer 1995	Continental United States	15-30 (East) ^h 25-35 (West)

^a NO_y is the chemical family including NO_x and its oxidation products; NO_y-NO_x denotes the chemically processed component of NO_y.

^b 1300-1700 local time (LT) in flatland and valley sites; all daytime measurements at elevated sites.

^c rural site in central Massachusetts.

^d 1100-1700 EST hourly means.

^e seasonal 7-h (0900-1559) daylight average.

^f 1300-1600 LT monthly mean.

^g daily max 8-h averages.

^h 1300-1700 average.

Source: Fiore et al. (2003).

Figure 3-26 shows a comparison between observations obtained at CASTNet sites and model results of Fiore et al. (2003). They classified the CASTNet monitoring sites into low-lying sites (generally <1.5 km) and elevated sites (>1.5 km). All elevated sites are in the West. Results were then aggregated to construct the cumulative probability distributions shown in Figure 3-26 for the 58 low-altitude sites and the 12 high-altitude sites as well as for the three seasons. The calculated mean background at the surface sites in spring is 27 ppbv, compared to 23 ppbv in summer and fall. At these sites, the background is highest for O₃ concentrations near the center of the distribution, and it declines as total surface O₃ concentrations increase, for reasons summarized below and discussed by Fiore et al. (2002). The observed O₃ concentration thus serves a surrogate for meteorological variability (i.e., stagnant versus ventilated conditions), such that the background O₃ is smaller on days when total O₃ is highest. At the elevated sites, the calculated mean background is 36 ppbv in spring versus 30 ppbv in the summer and fall. Background concentrations in the fall resemble those in summer but show less variability and do not exceed 40 ppbv anywhere in this analysis.

Major conclusions from the Fiore et al. (2003) study (discussed in detail in Annex AX3, Sections AX3.9.3 and AX3.9.4) are:

- PRB O₃ concentrations in U.S. surface air from 1300 to 1700 local time are generally 15 to 35 ppbv. They decline from spring to summer and are generally <25 ppbv under the conditions conducive to high-O₃ episodes.
- PRB O₃ concentrations can be represented as a function of season, altitude, and total surface O₃ concentration, as illustrated in Figure 3-26.
- High PRB concentrations (40 to 50 ppbv) occur occasionally at high-elevation sites (>1.5 km) in spring due to the free-tropospheric influence, including a 4- to 12-ppbv contribution from hemispheric pollution (O₃ produced from anthropogenic emissions outside North America). These sites cannot be viewed as representative of low-elevation surface sites (Cooper and Moody, 2000), where the background is lower when O₃ >60 ppbv.
- The stratospheric contribution to surface O₃ is of minor importance, typically well <20 ppbv. While stratospheric intrusions might occasionally elevate surface O₃ at high-altitude sites, these events are rare.

Appropriate background concentrations should thus be allowed to vary as a function of season, altitude, and total O₃ level. The diamonds in Figure 3-26 can be applied for this purpose. In particular, the depletion of the background during high-O₃ events should be taken into account

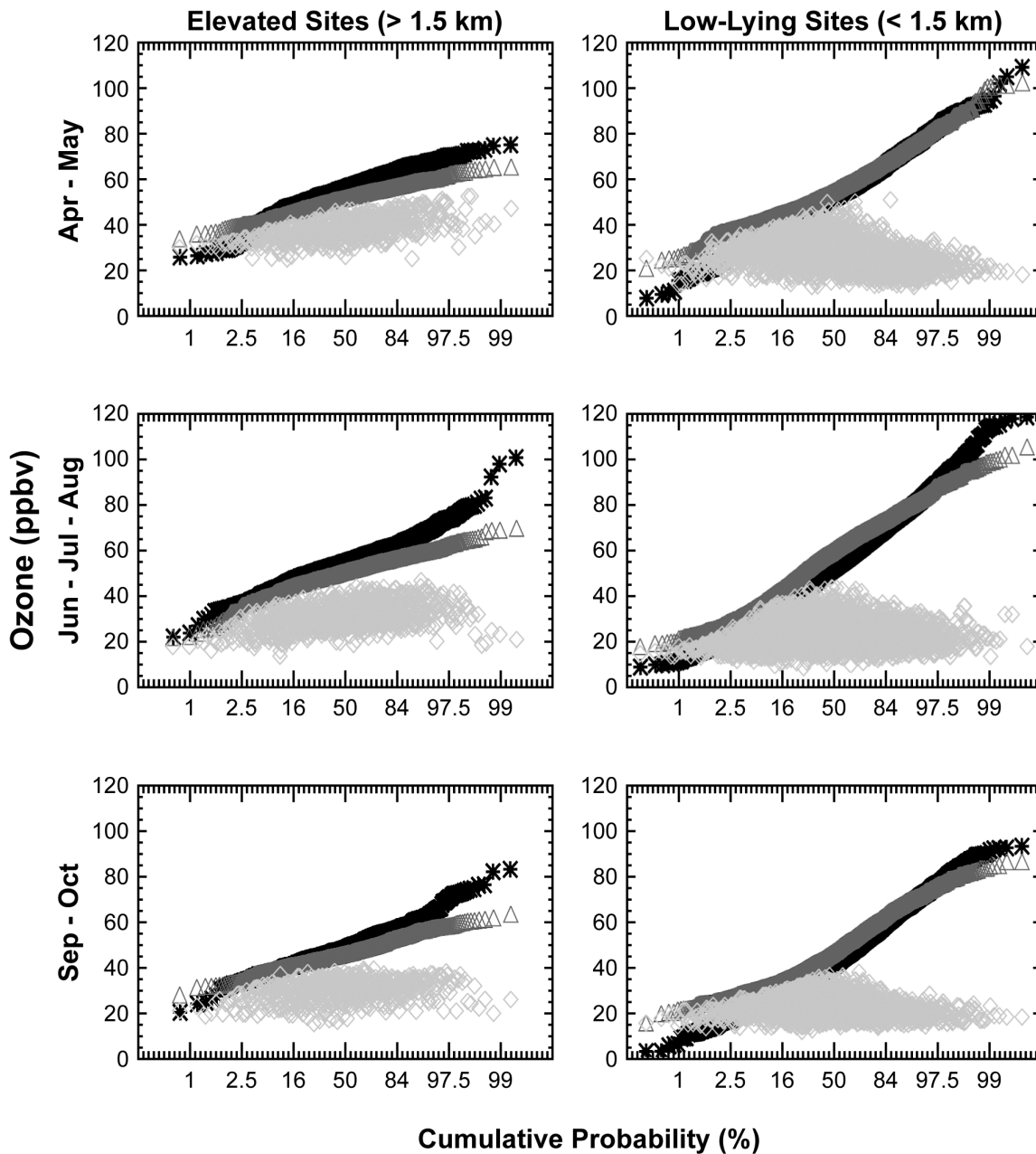


Figure 3-26. Estimates of background contribution to surface afternoon (13 to 17 LT) O₃ concentrations in the United States as a function of local O₃ concentration, site altitude, and season. The figure shows cumulative probability distributions of O₃ concentrations for the observations (asterisks) and the model (triangles). The corresponding distribution of background O₃ concentrations is shown as grey diamonds.

Source: Fiore et al. (2003).

(i.e., background O₃ is depleted by reactions in the atmosphere and by deposition to the surface but is not replenished at a significant rate in the stable, polluted boundary layer). This depletion is shown in the right-hand panels of Figure 3-26 for the highest O₃ values. Note that the model is generally able to reproduce the overall frequency distributions in Figure 3-26. Typically, models produce distributions flatter than are observed. Underpredictions, especially at the upper end of the frequency distribution during the warmer months, are likely related to sub-grid-scale processes that the model cannot resolve explicitly. The highest observed O₃ concentrations in all three seasons and at all altitudes are associated with regional pollution (i.e., North American anthropogenic emissions), rather than stratospheric influence.

PRB ozone is not a directly observable quantity and must therefore be estimated from models. Simple modeling approaches, such as the use of back-trajectories at remote U.S. sites to identify background conditions, are subject to errors involving the reliability of the trajectories, chemical production along the trajectories, and the hemispheric-scale contribution of North American sources to ozone in air masses originating outside the continent. They also cannot describe the geographical variability of the ozone background or the depletion of this background during pollution episodes. Global 3-D chemical transport models such as GEOS-Chem can provide physically-based estimates of the PRB and its variability through sensitivity simulations with North American anthropogenic sources shut off. These models are also subject to errors in the simulation of transport and chemistry, but the wealth of data that they provide on ozone and its precursors for the present-day atmosphere enables extensive testing with observations, and thus objective estimate of the errors on the PRB ozone values.

We present here such an error analysis for the Fiore et al. (2003a) PRB ozone estimates, focusing on evaluation of their GEOS-Chem simulation for present-day conditions with ozone observations of particular relevance to the PRB. Comparisons to ozonesonde observations of the vertical distribution of tropospheric ozone at northern mid-latitudes are of particular interest. These have been documented in a number of GEOS-Chem publications (Bey et al., 2001; Liu et al., 2002; Martin et al., 2002; Fusco and Logan, 2003), including specifically over North America (Li et al., 2002a, 2005). Results indicate no significant bias, and agreement to generally within 5 ppbv for monthly mean concentrations at different altitudes.

Fiore et al. (2002, 2003a,b) presented detailed evaluations of GEOS-Chem with ozone observations at U.S. surface sites. These evaluations focused on the afternoon hours (13-17 local

time), when surface measurements are representative of a deep mixed layer that can be resolved with the model. At night, surface ozone depletion often takes place by titration or deposition under locally stratified conditions, but such conditions cannot be simulated with confidence by a global model. The issue is not only one of vertical resolution (the lowest layers in GEOS-Chem extend to 20, 50, 100, 200, and 400 m above the local surface) but also of horizontal resolution ($2^\circ \times 2.5^\circ$). The PRB estimates presented by Fiore et al. (2003a) are also for 13-17 local time, and lower values would apply at night.

The GEOS-Chem evaluation of U.S. surface ozone presented by Fiore et al. (2003a) included monthly means, probability distributions, and time series at the sites previously used by Lefohn et al. (2001) to estimate background ozone. The simulated monthly mean concentrations in different seasons are typically within 5 ppbv of observations, with no significant bias, except in the southeast in summer when the model is 8-12 ppbv too high due to excessive background ozone advected in from the Gulf of Mexico. The time series comparisons for specific sites show that the model simulates the day-to-day variability of ozone and reveals no further bias. In particular, the model can reproduce the occurrences of relatively high ozone at remote sites previously reported by Lefohn et al. (2001), and shows that these can generally be explained by North American pollution.

Goldstein et al. (2004) presented comparisons of GEOS-Chem and MOZART global model results with observations at Trinidad Head, California, during April-May 2002. The observations, filtered to remove local influence, averaged 41 ± 5 ppbv, as compared to GEOS-Chem (39 ± 5 ppbv) and MOZART (37 ± 9 ppbv). Neither model was successful at reproducing the weak day-to-day structure in the observations, but they showed no bias in the simulation of occasional >50 ppbv days. Hudman et al. (2004) found that GEOS-Chem underestimated ozonesonde measurements in the free troposphere over Trinidad Head during the same period by 10 ppbv on average, apparently due to insufficient regional stratospheric influence. This bias does not extend to the surface, likely because stratospheric influence greatly diminishes as free tropospheric air is diluted in the boundary layer.

Several other papers have evaluated the GEOS-Chem simulation for surface ozone and its precursors over the United States. Fiore et al. (2003b) found a mean model bias of 6 ± 6 ppbv in surface air over the eastern United States in summer, reflecting the overestimate in the southeast; but they found that the model captured the large-scale spatial structure of the ozone observations

as revealed by an EOF analysis. The model also reproduces without significant bias the correlations of ozone with CO and NO_y over North America and downwind (Fiore et al., 2002; Li et al., 2002b, 2004). Comparisons to satellite observations of NO₂ and formaldehyde tropospheric columns over North America show excellent correspondence in the spatial patterns and no significant biases.

In conclusion, we estimate that the PRB ozone values reported by Fiore et al. (2003a) for afternoon surface air over the United States are likely 10 ppbv too high in the southeast in summer, and accurate within 5 ppbv in other regions and seasons. The PRB ozone is likely lower at night than in the afternoon though Fiore et al. (2003a) do not quantify the magnitude of this diurnal effect.

Chemistry transport models should be evaluated with observations given earlier in Chapter 3, in Annex AX3, and to simulate the processes causing the intra-day variability in O₃ concentrations shown in Figure 3-27 in addition to those summarized in Chapter 2. The diurnal patterns shown in Figure 3-27 do not fit the smooth pattern shown in Figure 3-15 and indicate processes capable of producing rapid rises in O₃ at times when substantial photochemical activity is not present and may indicate stratospheric effects. Higher resolution models capable of spatially and temporally resolving stratospheric intrusions and capable of resolving O₃ variability on hourly timescales have not been applied to this problem. Ebel et al. (1991) have demonstrated that regional-scale CTMs could be used to study individual stratospheric intrusions. As an example of the utility of different types of models, Zanis et al. (2003) were able to forecast, observe, and model a stratospheric intrusion (maximum penetration depth was to slightly >2 km altitude) that occurred from June 20 to 21, 2001, over a large swath of central Europe. Roelofs et al. (2003) compared results from six global tropospheric CTMs with lidar observations obtained during that event and concluded that the models qualitatively captured the features of this intrusion. It was also found that the coarser resolution models overestimated transport to lower altitudes. The use of higher resolution models, perhaps nested inside the coarser resolution models, may have helped solve this problem. They would also better address issues related to temporal (i.e., 1-h versus 8-h averages) and spatial (i.e., populated versus remote areas) scales needed by policymakers.

Although many of the features of the day-to-day variability of O₃ at RRMS in the United States are simulated reasonably well by Fiore et al. (2003), uncertainties in the calculation of the

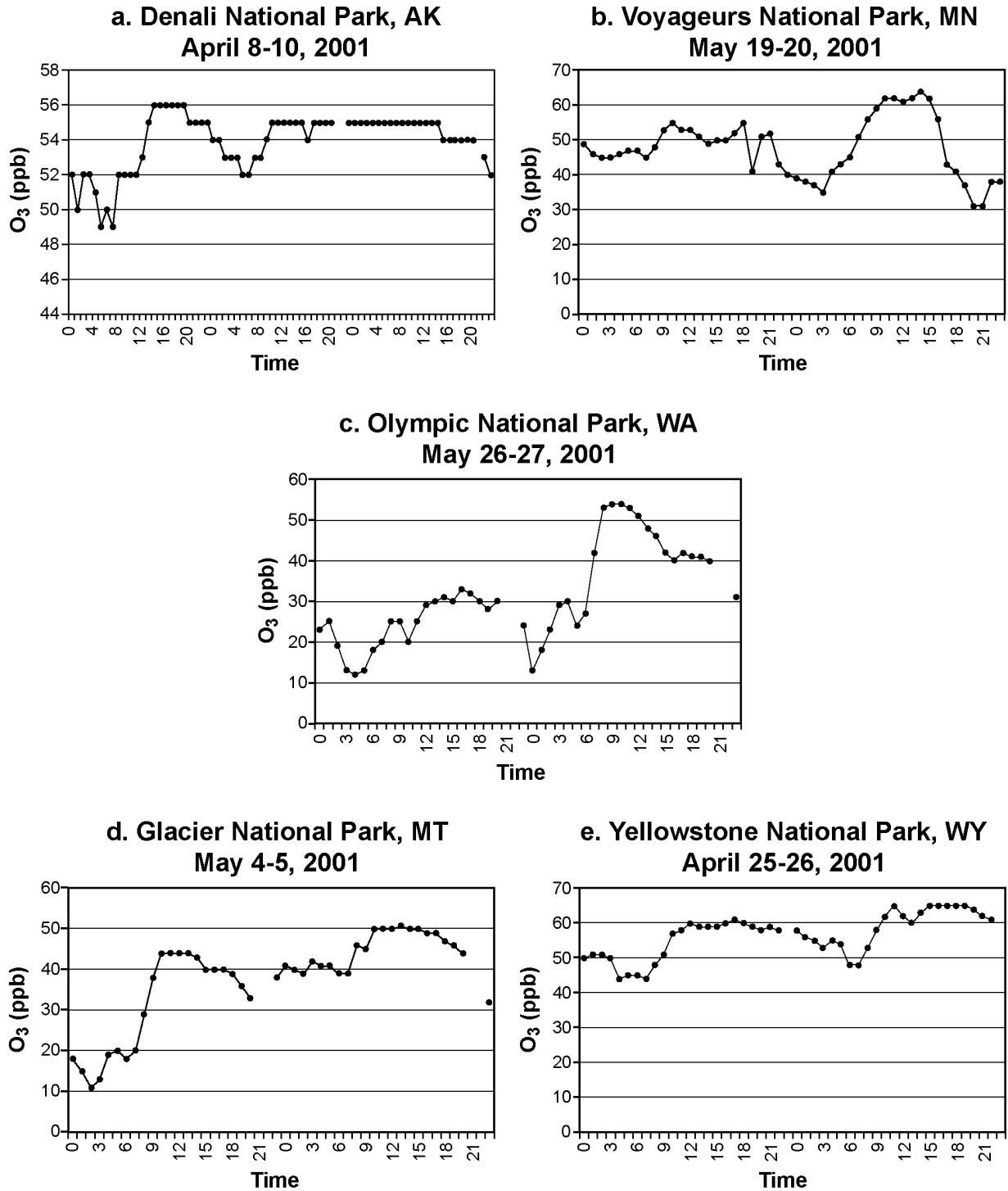


Figure 3-27. Time-series of hourly average O_3 concentrations observed at five national parks: Denali (AK), Voyager (MN), Olympic (WA), Glacier (MT), and Yellowstone (WY).

temporal variability of O₃ originating from different sources on shorter time scales must be recognized. The uncertainties stem in part from an underestimate in the seasonal variability in the STE of O₃ (Fusco and Logan, 2003), the geographical variability of this exchange, and the variability in the exchange between the free troposphere and the planetary boundary layer in the model.

Ideally, the predictions resulting from an ensemble of models should be compared with each other and with observations, so that the range of uncertainty inherent in the model predictions can be evaluated. In this regard, it should be noted that only one model (GEOS-Chem) is documented in the literature for estimating PRB concentrations.

3.8 OZONE EXPOSURE IN VARIOUS MICROENVIRONMENTS

Humans are exposed to O₃ and related photochemical oxidants through the exchange boundary, the skin and the openings into the body such as the mouth, the nostrils, and punctures and lesions in the skin (U.S. Environmental Protection Agency, 1992; Federal Register, 1986). Inhalation exposure to O₃ and related photochemical oxidants is determined by pollutant concentrations measured in the breathing zone that is not affected by exhaled air as the individual moves through time and space. A discussion of the basic terminology associated with exposure appears in AX3.

Quantification of Exposure

Ambient O₃ concentrations vary with time of day (peaking during the latter portion of the day) and season and among locations. Consequently, exposure to O₃ will change as a function of time of day and as an individual moves among locations. A hypothetical exposure is demonstrated in Figure 3-28. The actual dose received also changes during the day and is dependent on the O₃ concentration in the breathing zone and the individual's breathing rate, which is, in turn, dependent on the individual's level of exertion.

When measuring or modeling exposure to O₃ and related photochemical oxidants consideration should be given to the diurnal weekly (weekday-weekend) and seasonal variability. Peak concentrations lasting for several hours typically occur toward the latter portion of the day during the summer months. Regional O₃ episodes often co-occur with high

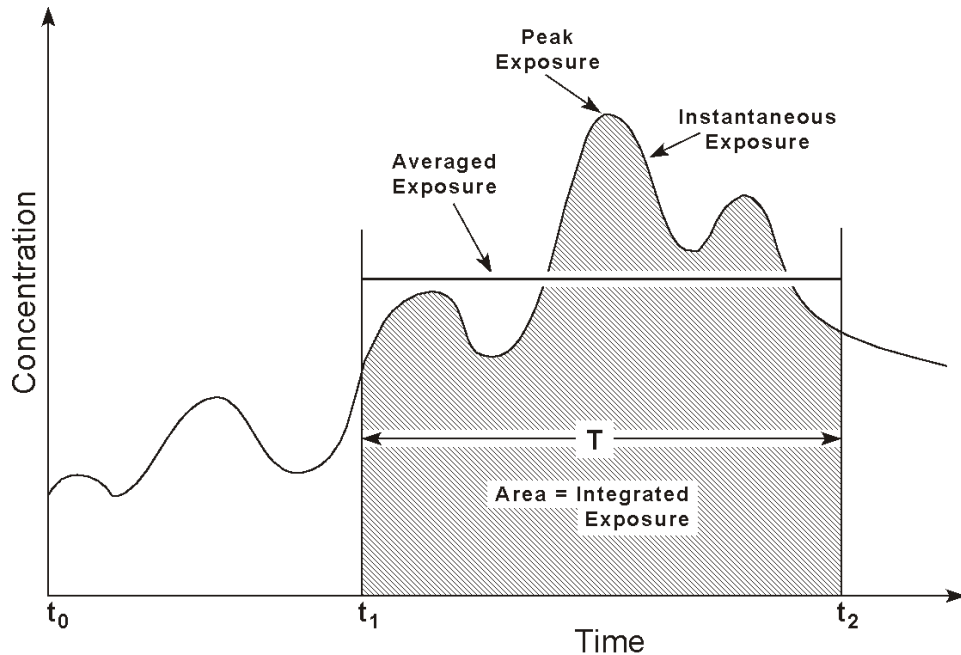


Figure 3-28. Hypothetical exposure time profile: pollutant exposure as a function of time showing how the average exposure, integrated exposure, and peak exposure relate to the instantaneous exposure. ($t_2 - t_1 = T$)

Source: U.S. Environmental Protection Agency (2004a).

concentrations of airborne fine particles, making it difficult to assess O_3 dynamics and exposure patterns. Also, while there are few indoor O_3 sources, O_3 will react with materials and other pollutants in the indoor environment in an analogous fashion to that occurring in the ambient atmosphere, potentially exposing subjects to other more toxic pollutants (Nazaroff and Weschler, 2004; Lee and Hogsett, 1999; Wainman et al., 2000; Weschler and Shields, 1997). (See discussion on O_3 chemistry and indoor sources and concentrations later in this chapter.).

Personal Exposure and Ambient Concentrations

The two approaches for measuring personal exposure are (a) the direct approach, using a personal exposure monitor (PEM) consisting of a passive sampler worn around the breathing zone, and (b) the indirect approach, which measures or estimates the O_3 concentrations through the use of models or biomarkers. Both approaches are associated with measurement error.

Although it is difficult to develop passive monitors for personal exposure measurements because of problems in identifying chemical or trapping reagents that can react with O₃, several modified passive samplers have been developed for use in personal O₃ exposure measurements (Bernard et al., 1999; Koutrakis et al., 1993; Avol et al., 1998b; Geyh et al., 1997, 1999). Personal exposure measurements using passive samplers show O₃ exposures below those O₃ concentrations measured at outdoor stationary sites (Delfino et al., 1996; Avol et al., 1998a; Sarnat et al., 2000; Geyh et al., 2000). However, there is a strong correlation between O₃ measured at stationary sites and personal monitored concentrations (Liard et al., 1999; Brauer and Brook, 1997; Linn et al., 1996; Lee et al., 2004; Avol et al., 1998a; O'Neill et al., 2003) when the time spent outdoors, age, gender, and occupation of the subjects were considered.

The indirect approach determines and measures the concentrations in all of the locations or “microenvironments.” The concept of microenvironments is important in the understanding of human exposure modeling. Often identified with a perfectly mixed compartment, microenvironments are more recently viewed as a controlled volume, indoors or outdoors, that can be characterized using a set of either mechanistic or phenomenological governing equations. This allows for a nonhomogeneous environment, including sources and sinks within the microenvironment. Microenvironments include indoor residences, other indoor locations, outdoors near roadways, other outdoor locations, and areas within vehicles.

Microenvironmental Concentration and Ozone Exposure Models

Outdoor concentrations of O₃ are estimated either through emissions-based mechanistic modeling, or through ambient-data-based modeling. Emissions-based models determine the spatiotemporal fields of the O₃ concentrations using precursor emissions and meteorological conditions as inputs. (They are described in Annex AX2.). The ambient-data-based models determine spatial or spatiotemporal distributions of O₃ through the use of interpolation schemes. The kriging approach provides standard procedures for generating an interpolated O₃ spatial distribution for a given period of time (Georgopoulos et al., 1997a,b). The Spatio-Temporal Random Field (STRF) approach has been used to interpolate monitoring data in both space and time (Christakos and Vyas, 1998a,b). The STRF approach can analyze information on temporal trends which cannot be directly incorporated by kriging.

Several approaches are available for modeling microenvironmental concentrations: empirical, mass balance, and detailed computational fluid dynamics (CFD) models. Empirical relationships provide the basis for future, “prognostic” population exposure models. Mass balance modeling is the most common approach used to model pollutant concentrations in enclosed microenvironments. Mass balance modeling ranges from very simple formulations, assuming ideal (homogeneous) mixing and only linear physicochemical transformations with sources and sinks, to models that account for complex multiphase chemical and physical interactions and nonidealities in mixing. Mass balance models take into account the effects of ventilation, filtration, heterogeneous removal, and direct emission as well as photolytic, thermal, and chemical reactions. The simplest form of the model is represented by the following differential equation:

$$\frac{dC_{IN}}{dt} = vC_{OUT} + \frac{S}{V} - vC_{IN}$$

where dC_{IN} is the indoor pollutant concentration (mass/volume), dt is time in hours, v is the air exchange rate, C_{OUT} is the outdoor pollutant concentration (mass/volume), V is the volume of the microenvironment, and S is the indoor source emission rate. When the model was used to estimate indoor O_3 concentrations, indoor concentrations were found to be 33% of outdoor O_3 concentrations (Freijer and Bloemen, 2000). A more in-depth discussion of the mass balance model has been reported in Nazaroff and Cass (1986). The pNEM/ O_3 model, discussed later in this chapter, includes a sophisticated mass balance model for indoor and vehicle microenvironments (Johnson, 2003). CFD models take into account the complex, multiphase processes that affect indoor concentrations of interacting gas phase pollutants, such as the interactions of O_3 with indoor sinks and sources (surfaces, gas releases) and with entrained gas (Sarwar et al., 2001, 2002; Sørensen and Weschler, 2002).

Exposure modeling is often used in evaluating exposure to large populations over time. The use of models is complicated by the fact that O_3 is a secondary pollutant with complex nonlinear and multiscale dynamics in space and time. Ozone is formed in the atmosphere through a series of chemical reactions involving precursor VOCs and NO_x . Therefore, O_3 exposures may be affected by: (1) emission levels and spatiotemporal patterns of VOCs and NO_x ; (2) ambient atmospheric as well as indoor microenvironmental transport, removal

and mixing processes; and (3) chemical transformations that take place over a multitude of spatial scales. The transformations are dependent on the presence of co-occurring pollutants and the nature of surfaces interacting with the pollutants.

Exposure models may be classified as (1) potential exposure models, typically the maximum outdoor concentrations versus “actual” exposure, including locally modified microenvironmental outdoor and indoor exposures; (2) population versus “specific individual”-based exposure models; (3) deterministic versus probabilistic models; and (4) observation versus mechanistic air quality model-driven estimates of spatially and temporally varying O₃ concentrations.

There are several steps involved in defining exposure models. The steps are based on frameworks described in the literature over the last 20 years and the structure of various existing inhalation exposure models (NEM/pNEM, MENTOR/SHEDS, REHEX, TRIM.Expo also known as APEX, AIRPEX, AIRQUIS). The steps include (1) estimation/ determination of the background or ambient levels of O₃; (2) estimation/determination of levels and temporal profiles of O₃ in various microenvironments; (3) characterization of relevant attributes of individuals or populations under study (age, gender, weight, occupation, other physiological characteristics); (4) development of activity event or exposure event sequences; (5) determination of appropriate inhalation rates during the exposure events; (6) determination of dose; (7) determination of event-specific exposure and intake dose distributions for selected time periods; and (8) extrapolation of population sample (or cohort) exposures and doses to the entire populations of interest. Figure 3-29 provides a conceptual overview of a current exposure model. A more detailed overview of an exposure model can be found in Annex AX3.

To estimate the actual O₃ dose delivered to the lung, information on the concentration, minute ventilation rate, activity level, and the morphology of the respiratory tract are needed. Limited data have been compiled for ventilation rates for different age groups, both healthy and compromised individuals, at various levels of activity (Klepeis et al., 1996, 2001; Avol et al., 1998a; Adams, 1993). Based on the available information, the highest level of outdoor activity occurs during the spring and summer months, during the mid- to late afternoon and early evening—the times when O₃ concentrations are highest. Children are likely more susceptible to the effects of O₃ than other groups. School-age children spend more time outdoors engaged in high-level activities than do other groups and breath more air in than adults relative to body

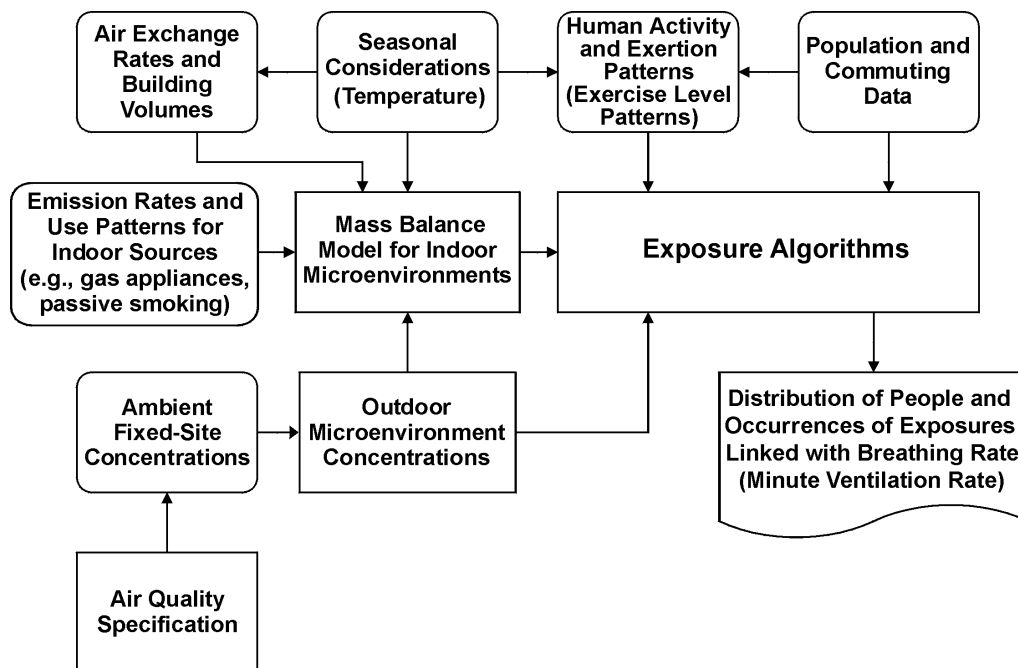


Figure 3-29. Conceptual overview of an exposure model. Model inputs (e.g., activity patterns, ambient monitoring data, air exchange rates) are in round-corner boxes and model calculations are shown in rectangles.

Source: Johnson et al. (1999).

surface area, breathing frequency, and heart rate. Asthmatic children spend the same amount of time outdoors as other more healthy children but the time spent engaged in high levels of activity are less.

Estimates of activity level have been compiled based on questionnaire data. The National Human Activity Pattern Survey (NHAPS), a probability-based telephone survey, was conducted in the early 1990s. The survey concluded that outdoor work-related activities were highest during the springtime and were more frequent during the morning and early afternoon. Exercise/sports-related activities were highest from noon to 3 p.m. during the summer months. During the spring months, exercise/sports-related activities were highest from mid- to late afternoon (Klepeis et al., 1996, 2001). A pilot study by Gonzales et al. (2003) evaluated the use of retrospective questionnaires for reconstructing past time-activity and location pattern information. Ozone concentration estimates using ambient stationary monitors and estimates

derived from diaries and questionnaires differed slightly. However, both estimates were greater than O₃ personal exposure measurements.

Existing comprehensive inhalation exposure models (NEM and pNEM) (Johnson, 2003; Johnson et al., 1996a,b,c, 1997), (MENTOR/SHEDS) Burke et al., 2001; McCurdy et al., 2000), and the APEX model treat human activity patterns as sequences of exposure events in which each event is defined by a geographic location and microenvironment and then assigned activity diary records from the CHAD (Consolidated Human Activities Database; www.epa.gov/chadnet1) (Glen et al., 1997; McCurdy, 2000; McCurdy et al., 2000). There are now about 22,600 person-days of sequential daily activity pattern data in CHAD representing all ages and both genders. The data for each subject consist of one or more days of sequential activities, in which each activity is defined by start time, duration, activity type (140 categories), and microenvironment classification (110 categories). Activities vary from 1 min to 1 h in duration. Activities longer than 1 h are subdivided into clock-hour durations to facilitate exposure modeling. A distribution of values for the ratio of oxygen uptake rate to body mass (referred to as metabolic equivalents or METs) is provided for each activity type listed. A table listing the activity patterns included in CHAD appears in AX3.

pNEM divides the population of interest into representative cohorts based on the combinations of demographic characteristics (age, gender, and employment), home/work district, and residential cooking fuel. APEX and MENTOR/SHEDS generate a population demographic file containing a user-defined number of person-records for each census tract of the population based on proportions of characteristic variables (age, gender, employment, and housing) obtained for the population of interest, and then assigns the matching activity information from CHAD to each individual record of the population based on the characteristic variables.

The APEX model is capable of simulating individual movement through time and space to provide estimates of exposure to a given pollutant in various microenvironments (e.g., indoor, outdoor, and in-vehicle microenvironments). One of the key strengths of the APEX model is its ability to estimate hourly exposures and doses for all simulated individuals in a sampled population. However, APEX is limited in that uncertainties in the predicted distributions (e.g., age, activity data, commuting patterns, personal activities) have not been addressed. The APEX model has not been evaluated, however, the pCNEM, a Canadian conceptual version of the

NEM model, has been evaluated for estimation of PM₁₀. Since pCNEM is similar to the pNEM/NEM, the APEX model should work as well as pCNEM. (See discussion on pCNEM later in this section.)

MENTOR/SHEDS is capable of simulating individuals exposures in various microenvironments (outdoors, residence, office, school, store, restaurant, bar, and vehicles) using spatial concentration data for each census tract. The indoor and in-vehicle pollutant concentrations are calculated using specific equations for the microenvironment and ambient pollutant concentration relationship. Randomly selected characteristics for a fixed number of individual are selected to match demographics within the census tract for age, gender, employment status, and housing type. Smoking prevalence statistics by gender and age is randomly selected for each individual in the simulation. Diaries for activity patterns are matched for the simulated individual by demographic characteristics (Burke et al., 2001).

Zidek et al. (2000, 2003, 2005) described a methodology for predicting human exposure to environmental pollutants. The methodology builds on earlier models such as SHEDS and pNEM/NEM and provides a WWW platform for developing a wide variety of models. pCNEM, a platform model developed from this methodology, is a Canadian PC version of NEM. pCNEM was used to estimate a conditional predictive exposure distribution for PM₁₀ in London. An important feature of pCNEM is its ability to estimate the effects of reductions in ambient levels of pollutants.

An important source of uncertainty in existing exposure modeling involves the creation of multiday, seasonal, or year-long exposure activity sequences based on 1- to 3-day activity data for any given individual from CHAD. Activity pattern data sets in CHAD vary for different studies and may not be representative of the population. The commuting data are for home-to-work and may not be representative of other commuting patterns. Correlations among human activities that can impact microenvironmental concentrations are not captured. See Annex AX3. Currently, appropriate longitudinal data are not available and the existing models use various rules to derive longer-term activity sequences utilizing 24-h activity data from CHAD.

Of the above models, only NEM/pNEM have been used in the prior NAAQS review. The pNEM probabilistic model builds on the earlier NEM deterministic exposure model. The model takes into consideration the temporal and spatial distribution of people and O₃ in the area of consideration, variations in O₃ concentrations in the microenvironment, and the effects of

exercise-increased ventilation on O₃ uptake. The pNEM models have been applied to nine urban areas and a summer camp. The models used activity data from the Cincinnati Activity Diary Study (CADS) along with time-activity data from several other studies. Data from stationary monitoring sites were used to estimate outdoor O₃ exposure. Indoor O₃ decay was assumed to be proportional to the indoor O₃ concentration. An algorithm assigned the EVR associated with each exposure event. The EVR for the outdoor children model was generated using a module based on heart rate data by Spier et al. (1992) and Linn et al. (1992).

Characterization of Exposure

The use of ambient air monitoring stations is the most common surrogate for assigning exposure in epidemiological studies. Since the primary source of O₃ exposure is the ambient air, monitoring concentration data would provide the exposure outdoors while exercising, a potential important exposure to evaluate in epidemiological studies. Monitored concentrations are useful for a relative assignment of exposure with time if the concentration were uniform across the region; the time-activity pattern were the same across the population; and the housing characteristics, such as ventilation rates and the O₃ sinks contributing to its indoor decay rates, were constant for the study area. Since these factors vary by population and location there will be errors in the magnitude of the total exposure and in the relative total exposure assignment based solely on ambient monitoring data.

Personal O₃ exposure measurements have been made for potentially susceptible populations (children, outdoor workers, the elderly, and individuals with chronic obstructive pulmonary disease). Children and outdoor workers have somewhat higher exposures than other individuals because they spend more time outdoors engaged in moderate and heavy exertion. Children are also more active outside and, therefore, have a higher minute ventilation rate than most adults (Klepeis et al., 1996, 2001). Available exposure studies suggest trends in exposure magnitude for some populations, however, additional exposure studies are needed to generalize differences in exposure between the general population and potentially susceptible populations. Table 3-3 summarizes the findings of available exposure studies.

Ozone concentrations in various microenvironments under a variety of environmental conditions have been reported in the literature. In the absence of an indoor O₃ source, concentrations of O₃ indoors are lower than that found in the ambient air. Ozone concentrations

Table 3-3. Personal Exposure Concentrations

Location, Population, Sample Duration	n	Personal Exposure Mean ^a (range) (ppb)	Reference
San Diego, CA, Asthmatics ages 9-18 years, 12 hour	12	12 ± 12 (0-84) 10 weekend 12 weekday	Delfino et al. (1996)
Vancouver, Canada, Adult Workers, Daily High indoor time Moderate indoor time Only outdoor	585	(ND-9) (ND-12) (2-44)	Brauer and Brook (1997)
Southern California, Subjects 10-38 years Spring Fall	24	13.6 ± 2.5 (- to 80) 10.5 ± 2.5 (- to 50)	Liu et al. (1997)
Montpellier, France, Adults, Hourly Winter Summer	16	34.3 ± 17.6 (6.5-88) 15.4 ± 7.7 (6.5-40) 44.1 ± 18.2(11-88)	Bernard et al. (1999)
Southern California, Children 6-12 years, ≥6 days Upland - winter - summer Mountain - winter - summer	169	6.2 ± 4.7 (0.5-41) 19 ± 18 (0.5-63) 5.7 ± 4.2 (0.5-31) 25 ± 24 (0.5-72)	Geyh et al. (2000)
Baltimore, MD, Technician, Hourly ^b Winter Summer	1	3.5 ± 7.5 (ND-49) 15 ± 18 (ND-76)	Chang et al. (2000)
Baltimore, MD, Adults 75 ± 7 years, Daily Winter Summer	20	3.5 ± 3.0 (ND-9.9) 0. ± 1.8 (ND-2.8)	Sarnat et al. (2000)

^aND = not detected.

^bMeasurements made following scripted activities for 15 days.

in microenvironments were found to be primarily controlled by ambient O₃ concentrations and the AER: they increase with increasing AER. To a lesser extent, O₃ concentrations in microenvironments are influenced by the ambient temperature, time of day, indoor characteristics (e.g., presence of carpeting), and the presence of other pollutants in the microenvironment. Table 3-4 describes the findings of the available studies.

Factors Affecting Ozone Concentrations

Ozone and other photochemical oxidants are formed in the ambient air from the reaction of sunlight with vehicle emissions, gasoline fumes, solvent vapors, and power plant and industrial

Table 3-4. Indoor/Outdoor Ozone Concentrations in Various Microenvironments

Location and Ventilation Conditions	Indoor/Outdoor Concentrations	Comments	Reference
New England States (9) Fall	20 ppb/40 ppb	Schools represented a variety of environmental conditions - varying ambient O ₃ concentrations, sources, geographic locations, population density, traffic patterns, building types. Average O ₃ concentrations were low in the morning and peaked during the early afternoon. O ₃ concentrations averaged for all schools monitored.	NESCAUM (2002)
Mexico City, School Windows/Doors Open (27) Windows/Doors Closed Cleaner Off (41) Windows/Doors Closed Cleaner On (47)	0 to 247 ppb/ 64 to 361 ppb	Study conducted over 4 d period during winter months. Two-minute averaged measurements were taken both inside and outside of the school every 30 min from 10 a.m. to 4 p.m. Estimated air exchange rates were 1.1, 2.1, and 2.5 h ⁻¹ for low, medium, and high flow rates. Ozone concentrations decreased with increasing relative humidity.	Gold et al. (1996)
Mexico City Homes	5 ppb/27ppb (7 d) 7 ppb/37 ppb (14 d)	Ozone monitoring occurred between September and July. Study included 3 schools and 145 homes. Most of the homes were large and did not have air conditioning. Ninety-two percent of the homes had carpeting, 13% used air filters, and 84% used humidifiers. Thirty-five percent opened windows frequently, 43% sometimes, and 22% never between 10 a.m. and 4 p.m. Ozone monitored at schools sites from 8 a.m. to 1 p.m. daily for 14 consecutive days. Homes monitored for continuous 24-h periods for 7 and 14 consecutive days.	Romieu et al. (1998)
Mexico City Schools	22 ppb/73 ppb		
Boston, MA, Homes (9) Winter - continuously	0 to 20.4 ppb/4.4 to 24.5 ppb	Study examined the potential for O ₃ to react with VOCs to form acid aerosols. Carbonyls were formed. No clear trend of O ₃ with AERs. The average AER was 0.9 h ⁻¹ during the winter and 2.6 h ⁻¹ during the summer. Four residences in winter and nine in summer with over 24 h samples collected.	Reiss et al. (1995)
Boston, MA, Homes (9) Summer - continuously	0 to 34.2 ppb/8.2 to 51.8 ppb		
Los Angeles, Homes (239)	13 ppb/37 ppb	Four hundred and eighty-one samples collected inside and immediately outside of home from February to December. Concentrations based on 24-h average O ₃ concentrations indoors and outdoors. Low outdoor concentrations resulted in low indoor concentrations. However, high outdoor concentrations resulted in a range of indoor concentrations.	Avol et al. (1998b)
Burbank, CA Telephone Switching Station	0.2 to 1.0/1.0–21.1 ppb	Major source of O ₃ was transport from outdoors. From early spring to late fall O ₃ concentrations peaked during the early afternoon and approach zero at sunset. AER ranged from 1.0 to 1.9 h ⁻¹ .	Weschler et al. (1994)

Table 3-4 (cont'd). Indoor/Outdoor Ozone Concentrations in Various Microenvironments

Location and Ventilation Conditions	Indoor/Outdoor Concentrations	Comments	Reference
Montpellier, France Homes (110)	15.5/32.0 ppb	Ozone measurements made over 5-d periods in and outside of 21 homes during summer and winter months. The winter I/O ratio was 0.31 compared to 0.46 during summer months.	Bernard et al. (1999)
Southern CA, Homes Upland Mountains	11.8/48.2 ppb 2.8/35.7 ppb	Ozone measurements were taken at 119 homes (57 in Upland and 62 in towns located in the mountains) during April and May. Concentrations were based on average monthly outdoor and average weekly indoor concentrations. Indoor based on the home location, number of bedrooms, and presence/absence of an air conditioner.	Geyh et al. (2000); Lee et al. (2002)
Krakow, Poland, Museums Cloth Hall Matejko Wawel Castle National	3.2/25.7-27.4 ppb 8.5/20.0 ppb 2.5/14.7 ppb 1.5/11.0 ppb	Ozone continuously monitored at five museums and cultural centers. Monitoring conducted during the summer months for 21 to 46 h or 28 to 33 days at each site. Indoor concentration found to be dependent on ventilation rate, i.e., when the ventilation rate was high, the indoor O ₃ concentrations approached that of ambient O ₃ . Rooms sequestered from outdoor air, or where air was predominantly recycled through charcoal filters, O ₃ levels indoors greatly reduced.	Salmon et al. (2000)
Buildings, Greece Thessalonki Athens	9.39/15.48 ppb 8.14/21.66 ppb	There was no heating/air conditioning system in the building at Thessaloniki. Windows were kept closed during the entire monitoring period. Complete air exchange took place every 3 h. The air conditioning system in continuous use at the Athens site recirculated the air. Complete air exchange estimated to be 1 h. Monitoring done for 30 days at each site, but only 7 most representative days used.	Drakou et al. (1998)
Patrol cars, NC	11.7/28.3 ppb	Patrol cars were monitored Mon. through Thurs. between the hours of 3 p.m. to midnight on 25 occasions during the months of Aug., Sept., and Oct. Outdoor O ₃ concentrations were taken from ambient monitoring station. Air inside the patrol car was recirculated cool air.	Riediker et al. (2003)
University of CA Photocopy room	<20 to 40 ppb/—	Room volume was 40 m ³ . Ozone concentrations increased proportionately with increasing use of photocopier.	Black et al. (2000)
Home/office O ₃ generators	BLD* to 290 ppb/—	Room volume was 27 m ³ . Doors and windows closed. Heating/air conditioning and mechanical ventilation systems off. Ozone generators operated for 90 min. High O ₃ concentrations noted when O ₃ generator used at high setting. AER was 0.3 h ⁻¹ . Ozone concentrations varied depending on unit tested.	Steiber et al. (1995)

*BLD = Below limit of detection

emissions (See Chapter 2 for a discussion of O₃ atmospheric chemistry). Ozone enters the indoor environment primarily through infiltration from outdoors through building components, such as windows, doors, and ventilation systems. There are also a few indoor sources of O₃ (photocopiers, facsimile machines, laser printers, and electrostatic air cleaners and precipitators) (Weschler, 2000). Generally O₃ emissions from office equipment and air cleaners are low except under improper maintenance conditions. Reported O₃ emissions from office equipment range from 1300 to 7900 µg/h (Leovic et al., 1996, 1998). Most air cleaners (particulate ionizers) emitted no or only a small amount (56 to 2757 µg/h) of O₃ during operation (Niu et al., 2001). Emissions from O₃ generators can range from tens to thousands of micrograms per hour (Weschler, 2000; U.S. Environmental Protection Agency, 1996).

Other photochemical oxidants (peroxyacyl nitrates; PAN and PPN) have no known direct emission sources indoors. Although not a significant source of indoor PAN, PAN can form in the indoor environment from the reaction of the OH· or NO₃ with acetaldehyde to form the acetyl radical, CH₃CO (Grosjean et al., 1996). The acetyl radical then reacts with oxygen to form an acetylperoxy radical which reacts with NO₂ to form PAN. Peroxyacyl nitrates primarily occur in the indoor environment from infiltration through the building envelope and through openings in the building envelope.

The concentration of O₃ in indoor environments is dependent on the outdoor O₃ concentration, the AER or outdoor infiltration, indoor circulation rate, and O₃ removal processes through contact with indoor surfaces and reactions with other indoor pollutants. Since O₃ concentrations are generally higher during the warmer months, indoor concentrations will likely be highest during that time period. (See earlier discussion on ambient concentrations of O₃.)

Air exchange rates vary depending on temperature differences, wind effects, geographical region, type of heating/mechanical ventilation system, and building type (Weschler and Shields, 2000; Colome et al., 1994). The balance of the flow of air in and out of a microenvironment is greatest in a residential building when a window or door is open (Johnson et al., 2004; Howard-Reed et al., 2002). The opening of windows or doors is dependent on the building occupancy, season, housing density, the presence of air conditioning, and wind speed (Johnson and Long, 2004). When windows and doors are closed, the dominant mechanism controlling AERs is infiltration through unintentional openings in the building envelope. Williams et al. (2003a, 2003b) reported AERs of 0.001 to 4.87 h⁻¹ in 37 homes in Research Triangle Park, NC. Chan

et al. (2005) compared air leakage measurements for 70,000 houses. Older and smaller houses had higher normalized leakage areas than newer and larger houses. Meng et al. (2004) also attributed higher AERs to the age of the housing stock. AERs for homes in Houston, TX and Elizabeth, NJ were averaged for all four seasons, the highest AER, 1.22 h^{-1} , was noted for homes in Elizabeth, NJ where the homes were older. Evaluations of AERs for residential structures was reported by Murray and Burmaster (1995) and includes AERs for 2,844 residential structures in four different climatic regions by season (winter, spring, summer, and fall). The AER for all seasons across all regions was 0.76 h^{-1} (arithmetic mean) (Region 1: IN, MN, MT, NH, NY1, VT, WI; Region 2: CO, CT, IL, NJ, NY2, OH, PA, WA; Region 3: CA3, MD, OR, WA; Region 4: AZ, CA4, FL, TX). The AERs were generally higher during the warm seasons, when ambient O_3 concentrations are highest. Data for the warmest region during the summer months may not be representative of all homes because measurements were made in southern California where windows are open and air conditioning is not used.

Average mean (median) AERs of 2.45 (2.24), 1.35 (1.09), and 2.22 (1.79) h^{-1} were reported by Lagus Applied Technology, Inc. (1995) for schools, offices, and retail establishments in California. Mean AERs for schools, offices, and retail establishments in Oregon and Washington were 0.32, 0.31, and 1.12 h^{-1} (Turk et al., 1989)—considerably less than that reported by Lagus Applied Technology. Park et al. (1998) reported mean AERs ranging from 1.0 to 47.5 h^{-1} for stationary vehicles under varying ventilating conditions. Where available, AERs for other studies are included in Table 3-10.

The most important removal process for O_3 in the indoor environment is deposition on, and reaction with, indoor surfaces. The rate of deposition is material-specific. The removal rate will depend on the indoor dimensions, surface coverings, and furnishings. Smaller rooms generally have larger surface-to-volume ratio (A/V) and remove O_3 faster than larger rooms. Fleecy materials, such as carpets, have larger surface-to-volume ratios and remove O_3 faster than smooth surfaces (Weschler, 2000). However, the rate of O_3 reaction with carpet diminishes with cumulative O_3 exposure (Morrison and Nazaroff, 2000, 2002). Weschler (2000) compiled the O_3 removal rates for a variety of microenvironments. Generally, the removal rates ranged between 3.0 and $4.3 k_d (A/V)/\text{h}^{-1}$. The highest removal rate, $7.6 k_d (A/V)/\text{h}^{-1}$, was noted for a clean room (Weschler et al, 1989).

Ozone chemical reactions in the indoor environment are analogous to those reactions occurring in the ambient air (See discussion on atmospheric chemistry in Chapter 2). Ozone reacts with unsaturated VOCs in the indoor environment, primarily terpenes or terpene-related compounds from cleaning products, air fresheners, and wood products. The reactions are dependent on the O₃ indoor concentration, the indoor temperature and, in most cases, the air exchange rate/ventilation rate. It is beyond the scope of this document to discuss actual concentrations of O₃ reaction products in the indoor environment; however, many of the reaction products may more negatively impact human health and artifacts in the indoor environment than their precursors (Wolkoff et al., 1999; Wilkins et al., 2001; Weschler et al., 1992; Weschler and Shields, 1997; Rohr et al., 2002; Nøjgaard et al., 2005). Primary reaction products are Criegee biradicals, nitrate radicals, and peroxyacetyl radicals. Secondary reaction products are hydroxy, alkyl, alkylperoxy, hydroperoxy, and alkoxy radicals. Reactions with alkenes can produce aldehydes, ketones, and organic acids (Weschler and Shields, 2000; Weschler et al., 1992).

Hydroxyl radicals formed from the reaction of O₃ with VOCs, nitric oxide and hydroperoxy, and other intermediate products can react with nitrogen compounds, sulfur dioxide, and carbon monoxide to produce significantly more toxic compounds (Sarwar et al., 2002; Orzechowska and Paulson, 2002; Fick et al., 2003, 2004; Van den Bergh et al., 2000; Fan et al., 2003; Wilkins et al., 2001; Clausen et al., 2001; Rohr et al., 2002, 2003; Poupard et al., 2005; Blondeau et al., 2005). The reaction between O₃ and terpenes also has been shown to increase the concentration of indoor particles (Weschler and Shields, 1999, 2003; Weschler, 2004; Clausen et al., 2001; Fan et al., 2003; Wainman et al., 2000), possibly from further reactions of the hydroxy radical with terpenes (Sarwar et al., 2002).

Decomposition and formation of PAN in the indoor environment are influenced by NO₂ and NO. Decomposition of PAN is expected to be a relatively fast process when indoor O₃ levels are low and when motor vehicle emissions are large or there is an indoor source of NO_x (Weschler and Shields, 1997).

Factors Affecting the Relationship between Ambient Concentrations and Personal Exposures to O₃

Ambient O₃ concentrations vary with the time of day, season of the year, and among locations. Personal exposure to O₃ is influenced by the microenvironmental concentration and the amount of time spent in each microenvironment. The majority of the population spends, on

average, nearly 90% of their time in an indoor microenvironment. Since there are few indoor sources of O₃, O₃ ambient concentration may be the most important factor that affects average population exposure in the indoor environment.

Indoor O₃ concentrations also are affected by several other factors and mechanisms. Studies have shown that in addition to the ambient O₃ concentrations, indoor O₃ concentrations are influenced by the air exchange rate or outdoor infiltration, increasing with increasing air exchange. Once indoors, the O₃ concentration is affected by the indoor circulation rate and O₃ removal through contact with indoor surfaces and reactions with other indoor pollutants.

Studies on the effect of elevation on O₃ concentrations found that concentrations increased with increasing elevation (Väkevä et al., 1999; Johnson, 1997). Since O₃ monitors are frequently located on rooftops in urban settings, the concentrations measured there may overestimate the exposure to individuals outdoors in streets and parks, locations where people exercise and their maximum O₃ exposure is more likely to occur.

In epidemiologic studies investigating acute and chronic health outcomes using ambient monitoring data from stationary monitoring sites, O₃ exposure assessment was affected by the distance between home and the monitoring site, gender, time-activity patterns (e.g., percentage of time spent outdoors, type of outdoor activity, time of day during outdoor activity), and indoor air exchange rates (e.g., ventilation conditions, home characteristics) (Geyh et al., 2000; Lee et al., 2002, 2004; Liu et al., 1995, 1997; Chang et al., 2000; Chan et al., 2005; O'Neill et al., 2003; Brauer and Brook, 1997). Geyh et al. (2000) observed higher indoor and personal O₃ concentrations in a southern California community with 2% air-conditioned homes compared to a community with 93% air-conditioned homes during the summer (high O₃) months, but showed no difference in O₃ levels during the winter (low O₃) months. People that work outdoors tend to be exposed to higher levels of O₃ (Brauer and Brook, 1997; O'Neill et al., 2003). Lee et al. (2004) observed that personal O₃ exposure was positively correlated with outdoor time ($r = 0.19$, $p < 0.01$) and negatively correlated with indoor time ($r = -0.17$, $p < 0.01$). Additional factors that affected indoor O₃ levels were air conditioning, window fans, and window opening. The O₃ exposure assessment study by Liu et al. (1995) found that after adjusting for time spent in various indoor and outdoor microenvironments (e.g., car with windows open, car with windows closed, school, work, home, outdoors near home, outdoors other than near home), mean 12-hour ambient O₃ concentrations explained 32% of the variance in personal exposure in the summer.

In a southern California study by Avol et al. (1998a), boys were found to spend more time outdoors and be more physically active than girls. Another southern California study found that boys were outdoors 30 minutes longer than girls and had higher personal O₃ exposure during both high and low O₃ months (Geyh et al., 2000).

The announcement of smog alerts or air quality indices may influence personal exposures to O₃ by causing individuals to alter behaviors (avoidance behavior). Neidell (2004), in his evaluation of the effect of pollution on childhood asthma, examined the relationship between the issuance of smog alerts or air quality indices for several counties in California and hospital admissions for asthma in children under age 18 years (not including newborns). Smog alerts are issued in California on days when O₃ concentrations exceed 200 ppb. There was a significant reduction in the number of asthma-related hospital admissions in children ages 1 to 12 years on smog alert days, indicating that avoidance behavior might be present on days of high O₃ concentrations. Changes in population behavior as a function of concentration complicate the estimation of health effects from population-based studies; thus, it may be desirable to include sensitivity analyses that eliminate high O₃ days, particularly in areas where avoidance behavior is expected.

Potential Sources of Error Resulting from the Use of Ambient Ozone Concentrations in Epidemiological Analyses

There is no clear consensus among exposure analysts as to how well stationary monitor measurements of ambient O₃ concentrations represent a surrogate for personal O₃ exposure. The microenvironmental (indirect) approach and the personal sampling (direct) approach (Navidi et al., 1999; Ott, 1982, 1985) have been used to assess personal exposure in air pollution epidemiologic studies; however, both approaches are associated with measurement error. To determine personal exposure using the microenvironmental approach, the concentrations of the various microenvironments are multiplied by the time spent in each microenvironment. Both the concentration and time component contribute to the measurement error. There is no time component to the measurement error in the personal sampling approach, however, the estimation of exposure using personal monitoring devices contributes to measurement error, especially in the case of O₃. Passive badges are commonly used for monitoring O₃ integrated personal exposure. Their sensitivity to wind velocity, badge placement, and interference with other copollutants may result in measurement error.

Results from the error analysis models developed by Navidi et al. (1999) using two different statistical designs (bidirectional case-crossover and multi-level analytic designs) indicated that neither the microenvironmental nor personal sampling approach gave reliable health effect estimates when measurement errors were uncorrected. The nondifferential measurement error biased the effect estimates toward zero under the model assumptions. However, if the measurement error was correlated with the health response, a bias away from the null could result. The use of central ambient monitors to estimate personal exposure has a greater potential to introduce bias since most people spend the majority of their time indoors, where O₃ levels tend to be much lower than outdoor ambient levels. If the error is of a fixed amount (i.e., absolute differences do not change with increasing concentrations), there is no bias. However, if the error is not a fixed difference, this error will likely attenuate the O₃ risk estimate as personal O₃ exposures are generally lower than ambient O₃ concentrations.

Several studies have examined relationships between measured ambient O₃ concentrations from fixed monitoring sites and personal O₃ exposure (Avol et al., 1998b; Brauer and Brook, 1995, 1997; Chang et al., 2000; Delfino et al., 1996; Lee et al., 2004; Liard et al., 1999; Linn et al., 1996; Liu et al., 1995, 1997; O'Neill et al., 2003; Sarnat et al., 2001). Two studies by Sarnat et al. (2001, 2005) examined relationships between individual variations in personal exposure and ambient O₃ concentrations. In the first study conducted in Baltimore, MD, the association between 24-h average ambient O₃ concentrations from a centrally-located monitoring site and 24-h average personal O₃ exposures was evaluated in a cohort of older adults (n = 20), individuals with COPD (n = 15), and children (n = 21) (Sarnat et al., 2001). Personal exposures were measured repeatedly for each subject for a total of 800 person-days; thus, analyses of personal exposure data were conducted using mixed models with subjects modeled as random variables to account for between-subject variation. The mixed regression effect estimates were $\beta = 0.01$ (95% CI: -0.01, 0.03) for the summer (196 paired samples) and $\beta = 0.00$ for the winter (449 paired samples). However, in the second study conducted in Boston, MA with a cohort of 20 healthy senior citizens and 23 school children, significant associations were observed between 24-h average ambient O₃ concentrations and 24-h average personal O₃ exposures (Sarnat et al., 2005). The mixed regression effect estimates were $\beta = 0.27$ (95% CI: 0.18, 0.37) and $\beta = 0.04$ (95% CI: 0.00, 0.07) during the summer (332 paired samples) and winter (288 paired samples), respectively. In the Boston study, the regression coefficients

indicated that ambient O₃ concentrations were predictive of personal O₃ exposures; however, ambient O₃ levels overestimated personal exposures 3- to 4-fold in the summer and 25-fold in the winter.

Important predictors for personal O₃ exposure were examined (Xue et al., 2005) in the Harvard Southern California Chronic Ozone Exposure Study. In addition to ambient O₃ concentrations from central monitors, indoor and outdoor O₃ concentrations at the participant's homes, and personal O₃ concentrations were measured using passive O₃ samplers. During a one-year period, 160 children from two southern California communities (Upland and the San Bernardino mountains) were monitored for six consecutive days for at least six of the 12 months. Outdoor O₃ concentrations at the participant's homes were very similar to central ambient O₃ concentrations, and personal O₃ concentrations were close to indoor O₃ concentrations. At different time points throughout the year, the average ratios of personal to central O₃ concentrations were relatively stable, being around 0.3 (SD 0.13). Central O₃ concentrations were approximately 3 times higher compared to personal concentrations during the O₃ season (May to September) and approximately 5 times higher during the non-O₃ season (October to April). Xue et al. (2005) found that ambient O₃ concentrations from central monitors, after adjusting for time-activity patterns and housing characteristics from questionnaire data, reasonably predicted personal O₃ concentrations; a 1 ppb increase in ambient O₃ concentration was associated with a 0.54 ppb (95% CI not provided) increase in personal O₃ exposure (Pearson $r = 0.76$). The regression coefficient for the relationship between ambient O₃ concentrations and personal O₃ exposures without adjustment for time-activity and housing characteristics, which is of most relevance to epidemiologic time-series studies, was not presented.

Chang et al. (2000) compared 1-h personal and ambient O₃ measurements in older adults in various microenvironments in Baltimore, MD, using activity data from the National Human Activity Pattern Survey study (Klepeis, 1999). Activities were scripted to simulate activities performed by older adults (65+ years of age). In total, 180 1-h personal samples were collected in each season (summer and winter). There was no correlation between personal and ambient O₃ concentrations in the indoor residence ($r = 0.09$ in summer and $r = 0.05$ in winter), although a moderate correlation was found in other indoor environments such as restaurants, hospitals, and shopping malls ($r = 0.34$ and $r = 0.46$ for summer and winter, respectively). In comparison, the

correlation in outdoor environments (near and away from roads) was moderate to high ($0.68 \leq r \leq 0.91$) and statistically significant. Regression coefficients for the relationship between personal and ambient O₃ concentrations were not reported in this study.

Brauer and Brook (1995, 1997) observed that the daily averaged personal O₃ measurements and ambient concentrations were well-correlated after stratifying groups by time spent outdoors. Clinic workers (n = 25; 24-hour samples), teenage camp counselors (n = 25; 24-hour samples), and farm workers (n = 15; 6-14 h work shift samples) spent 0 to 25%, 7.5 to 45%, and 100% of their monitored time outdoors, respectively. The personal to ambient O₃ concentration ratios were significantly different for the clinic workers (0.28) and farm workers (0.96). Ambient O₃ concentrations and time spent outdoors explained more of the variability in the personal O₃ measurements for outdoor farm workers compared to the clinical workers. However, the Spearman correlation coefficients were comparable, 0.60 and 0.64 for the clinic workers and farm workers, respectively, indicating that even though the clinic workers spent considerable amounts of time indoors or in transit there was still reasonable correlation between the day-to-day variations in mean personal O₃ exposures and mean O₃ concentrations.

A study by O'Neill et al. (2003) examined 107 pairs of ambient and personal O₃ measurements from 39 outdoor workers in Mexico City using a longitudinal analysis method. Two to seven personal measurements were collected on each of the 26 monitoring days, which were averaged and then compared with the ambient concentrations. They estimated that a 1 ppb increase in ambient O₃ concentration was associated with a 0.56 ppb (95% CI: 0.43, 0.69) increase in personal O₃ concentration. In a Paris, France study by Liard et al. (1999), adults (n = 55) and children (n = 39) wore passive O₃ monitors for 4 consecutive days during three periods. For each period, all adults wore the O₃ monitors over the same 4 days. Likewise, all children wore monitors over the same 4 days for each of the three periods, but on different days from the adults. The ambient O₃ concentrations from the stationary monitoring sites did not explain a high percentage of the variance of personal O₃ exposure (nonsignificant [value not stated] in adults and 21% in children). However, when personal measurements from all subjects were aggregated for each of the six periods, the 4-day mean personal O₃ exposure was found to be highly correlated with the corresponding mean ambient concentration ($r = 0.83$, $p < 0.05$). Similarly, a study of Los Angeles school children by Linn et al. (1996) found that daily 24-h

average ambient O₃ concentrations from a central site were well-correlated ($r = 0.61$) with daily averaged personal O₃ exposures.

The relationship between personal O₃ exposures and ambient O₃ concentrations from central monitoring sites differed by city in the two studies by Sarnat et al. (2001, 2005). In the Baltimore study (Sarnat et al., 2001) no association was observed, while in the Boston study (Sarnat et al., 2005) ambient O₃ concentrations were significantly associated with personal O₃ exposures in mixed effects regression models. These results suggest that O₃ concentrations measured at central ambient monitors may explain, at least partially, the variance of individual personal exposures; however, this relationship is influenced by factors such as air exchange rates in housing and time spent outdoors, which may vary by city. This was further supported by results from the southern California study by Xue et al. (2005). They found that incorporating information on time-activity and housing characteristics allowed reasonable prediction of personal O₃ exposures from ambient O₃ concentrations.

Other studies observed that the daily averaged personal O₃ exposures from the population were well correlated with monitored ambient O₃ concentrations, although substantial variability existed among the personal measurements. In other words, centrally-located ambient O₃ monitors are likely to be representative of day-to-day changes in O₃ exposure experienced by the population. Brauer and Brook (1997) indicated that these results have implications for large epidemiologic studies which often depend upon fixed-site outdoor ambient monitors to estimate exposures. It should be noted, however, that although there are correlations between aggregate personal and monitored ambient O₃ concentrations, the absolute personal concentrations are generally considerably lower than the monitored ambient O₃ concentrations.

In summary, results indicate that the relationship between ambient O₃ concentrations and personal exposure will vary depending on individual- or city-specific factors such as time-activity patterns, indoor air exchange rates, and housing conditions, creating potential measurement errors. The expectations based on statistical modeling considerations are that these exposure measurement errors or uncertainties will decrease estimates of effect in O₃ health effects analyses, making it difficult to detect a true underlying association between the exposure and the health outcome studied. Also of special concern in interpreting results from mortality and hospitalization time-series studies is to what extent the ambient O₃ concentrations are representative of personal O₃ exposures in a particularly susceptible group of individuals, the

debilitated elderly, as the correlation between the two measurements have not been examined in this population. However, until more data on O₃ exposure become available, the use of monitored ambient O₃ concentrations as a surrogate for exposures is not expected to change the principal conclusions from O₃ epidemiologic studies.

Exposure to Related Photochemical Oxidants

A variety of related photochemical oxidants produced outdoors, such as PAN and peroxypropionyl nitrate (PPN), can infiltrate into indoor environments. These compounds are thermally unstable and decompose to peroxyacyl radicals and NO₂. Exposure to related photochemical oxidants has not been measured, nor are these compounds routinely monitored at stationary monitoring sites. Available monitored concentrations of related photochemical oxidants may be found in Annex AX3.

3.9 SUMMARY OF KEY POINTS

The median of the daily maximum 8-h O₃ concentration averaged over May to September 2000 to 2004 was about 0.049 ppm. The daily maximum 1-h O₃ concentrations could have been much higher in large urban areas or in areas downwind of large urban areas. For example, in Houston, TX, the daily maximum 1-h O₃ concentrations have approached 0.20 ppm during this period.

Daily maximum 8-h average O₃ concentrations are lower than the maximum 1-h O₃ concentrations, but they are highly correlated. Within individual MSAs, O₃ concentrations tend to be well correlated across monitoring sites. However, there can be substantial variations in O₃ concentrations. Ozone in city centers tends to be lower than in regions either upwind or downwind because of titration by NO emitted by motor vehicles.

Ozone concentrations tend to peak in early- to mid-afternoon in areas where there is strong photochemical activity and later in the day in areas where transport is more important in determining the O₃ abundance. Summertime maxima in O₃ concentrations occur in areas in the United States where there is substantial photochemical activity involving O₃ precursors emitted from human activities. Monthly maxima can occur anytime from June through August. However, springtime maxima are observed in national parks, mainly in the western United States

and at a number of other relatively unpolluted monitoring sites throughout the Northern Hemisphere. For example, the highest O₃ concentrations at Yellowstone NP tend to occur during April and May. Generally, monthly minima O₃ concentrations tend to occur from November through February at polluted sites and during the fall at relatively remote sites.

Nationwide, daily maximum 8-h O₃ concentrations have decreased at the upper end of the distribution from 1990 to 2004. However, the daily maximum 8-h O₃ concentrations toward the center of the distribution have not reflected these changes. Trends have not been consistent at national park sites; with downward trends observed at some sites and upward or no trends observed at others. At some sites, trends reversed direction in going from the 98th to the 95th percentile values.

Sufficient data are not available for other atmospheric oxidants (e.g., H₂O₂, PAN) and oxidation products (e.g., HNO₃, H₂SO₄) to relate concentrations of O₃ to these species for use in time series studies. Data for these species are only obtained as part of specialized field studies. In general, secondary species, such as HNO₃, H₂SO₄, H₂O₂, and PAN, are expected to be at least moderately correlated with O₃. On the other hand, primary species are expected to be more highly correlated with each other than with secondary species, provided that the primary species originate from common sources. Concentrations of other oxidants are much lower than for O₃ and range from ≤1% for oxidants in particles to several percent for gas phase species. The relationship of O₃ to PM_{2.5} is complex, because PM is not a distinct chemical species but is a mix of primary and secondary species. PM_{2.5} concentrations were positively correlated with O₃ during summer, but negatively correlated with O₃ during winter at Ft. Meade, MD. PM₁₀ concentrations show similar relations with O₃.

Co-occurrences of O₃ (defined when both pollutants are present at an hourly average concentration of ≥0.05 ppm) with NO₂ and SO₂ are rare. For example, there were fewer than 10 co-occurrences with either NO₂ or SO₂ in 2001. The number of co-occurrences for O₃ and PM_{2.5} (defined as an hourly average O₃ concentration ≥0.05 ppm and a 24-h average PM_{2.5} concentration ≥40 µg/m³ occurring during the same 24-h period) also tended to be infrequent (<10 times) at most sites, but there were up to 20 such co-occurrences at a few sites.

Policy relevant background O₃ concentrations are used for assessing risks to human health associated with O₃ produced from anthropogenic sources in continental North America. Because of the nature of the definition of PRB concentrations, they cannot be directly derived from

monitored concentrations, instead they must be derived from modeled estimates. Current model estimates indicate that ambient air PRB concentrations in the United States are generally 0.015 ppm to 0.035 ppm. They decline from spring to summer and are generally <0.025 ppm under conditions conducive to high O₃ episodes. However, PRB concentrations can be higher, especially at elevated sites during spring, due to enhanced contributions from hemispheric pollution and stratospheric exchange.

Ozone exposure changes as a function of time of day, season, and microenvironment. Ambient O₃ concentrations are generally higher during warmer seasons and during the weekday, peaking during the later portion of the day. Ozone concentrations in indoor microenvironments are generally lower than those concentrations encountered in the ambient air. There are few indoor sources of O₃. Ozone occurs in indoor microenvironments primarily through infiltration through the building envelop and through windows, doors, and ventilation systems. The indoor O₃ concentration is dependent on the outdoor concentration, the AER, indoor circulation rate, and removal processes. Consequently, measured and modeled exposures should take into consideration O₃ diurnal weekly and seasonal variability and varying microenvironmental concentrations.

Once indoors, O₃ reacts with indoor surfaces, including surface coverings and furnishings. Ozone also will react with VOCs in indoor environments, primarily terpenes or terpene-related compounds. Ozone reactions with pollutants indoors are analogous to those reactions occurring in the ambient air, potentially exposing subjects to compounds significantly more toxic than O₃.

The available approaches for measuring personal O₃ exposure include the direct approach, using a PEM, and the indirect approach, which measures or models exposure in the microenvironments the individual encounters. Both approaches are associated with measurement errors.

There are difficulties in identifying chemical trapping agents for PEMs that can react with O₃, and PEMs are sensitive to wind velocity, badge placement, and interference with other copollutants. Studies using PEMs show personal O₃ exposures below those concentrations measured at stationary monitoring sites, when measurements are not adjusted for time spent outdoors, housing characteristics, age, gender, and occupation.

The use of measured O₃ concentrations from stationary ambient monitoring sites as surrogates for personal exposure may be affected by the O₃ ambient concentration, percentage of

time spent outdoors, and type of outdoor activity. Epidemiologic studies investigating health outcomes using data from stationary monitoring sites found O₃ exposure to be affected by the distance between the subjects' location and the stationary monitor, individual activity patterns, and the O₃ concentration in the microenvironment.

The use of exposure models to evaluate O₃ exposure to large populations over time is complicated by the fact that O₃ is a secondary pollutant with complex nonlinear and multiscale dynamics in space and time. The existing comprehensive inhalation exposure models (NEM, pNEM, MENTOR/SHEDS, APEX, pCNEM) treat human activity patterns as sequences of exposure events. Estimates of activity levels are assigned from CHAD, the Consolidated Human Activities Database.

Ambient O₃ concentrations are estimated using emissions-based mechanistic models or ambient-data-based models. Models for estimating microenvironmental concentrations include the empirical, mass balance, and detailed CFD models. Mass balance modeling is the most common modeling approach to estimating concentrations in enclosed microenvironments. The pNEM/O₃ population exposure model, the model used more extensively in O₃ exposure modeling, includes a sophisticated mass balance model for indoor and vehicle microenvironments. There are three versions of the pNEM/O₃ model: the general population, outdoor workers, and outdoor children.

Results from O₃ exposure studies indicate that the relationship between ambient O₃ concentrations and personal exposure/dose will vary depending on O₃ concentrations and time spent in the various microenvironments, particularly the time spent outdoors where O₃ concentrations tend to be higher, and the personal activity level. Consequently, the O₃ exposure/dose may differ from the concentrations measured at stationary monitoring sites. However, until more data on O₃ exposure become available, the use of monitored ambient O₃ concentrations as a surrogate for exposures is not expected to change the principal conclusions from O₃ epidemiologic studies using community average health and pollution data.

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4. DOSIMETRY, SPECIES HOMOLOGY, SENSITIVITY, AND ANIMAL-TO-HUMAN EXTRAPOLATION

4.1 INTRODUCTION

The dosimetry of ozone (O₃) in humans has been examined in a series of studies published in the past decade. These studies further characterize the dose of O₃ delivered to various sites in the respiratory tract (RT). Ozone, classified as a reactive gas, interacts with surfactant, antioxidants, and other compounds in the epithelial lining fluid (ELF). Researchers have attempted to obtain a greater understanding of how these complex interactions affect O₃ uptake and O₃-induced injury. New work has also been completed evaluating species differences in responses to O₃ exposures, which allow more accurate quantitative extrapolation from animals to humans.

This chapter is not intended to be a complete overview of O₃ dosimetry and animal-to-human comparisons, but rather, it is an update of the dosimetry/extrapolation chapter from the last O₃ criteria document (U.S. Environmental Protection Agency, 1996), or 1996 O₃ AQCD, and other reviews of the earlier published literature. The framework for presenting this chapter is first a discussion in Section 4.2 of general concepts of the dosimetry of O₃ in the RT. Bolus-response studies are then presented in Section 4.2.1 followed by general uptake studies in Section 4.2.2. Dosimetry modeling is presented in Section 4.2.3 followed by the summary and conclusions for the dosimetry material in Section 4.2.4. The chapter continues in Section 4.3 with a discussion of species comparisons and ends with a discussion of animal-to-human extrapolation. More detailed discussions of the studies are presented in the supporting material to this chapter (Annex AX4). The toxicological effects of O₃ in laboratory animals and in vitro test systems are discussed in Chapter 5 and direct effects of O₃ in humans are discussed in Chapter 6. The historical O₃ literature is very briefly summarized in this chapter, providing a very concise overview of previous work. The reader is referred to the 1996 O₃ AQCD for more detailed discussion of the literature prior to the early 1990s.

4.2 DOSIMETRY OF OZONE IN THE RESPIRATORY TRACT

Ozone dosimetry refers to the measurement or estimation of the amount of O₃ or its reaction products reaching and persisting at specific sites in the RT following an exposure. The compound most directly responsible for toxic effects may be the inhaled gas O₃ or one of its chemical reaction products. Complete identification of the actual toxic agents and their integration into dosimetry is a complex issue that has not been resolved. Dosimetric studies attempt to quantify the amount of O₃ retained in the lung (i.e., not exhaled) or the dose of O₃ or its active metabolites (e.g., aldehydes or peroxides) delivered to target cells or tissues (i.e., dose per cell or tissue surface area). For comparison, epidemiologic studies may simply consider exposure concentration while clinical studies may consider the total amount of O₃ inhaled (product of exposure concentration, duration, and minute ventilation). Hence, dosimetric studies seek to accurately quantify dose to target lung regions or tissues, whereas epidemiologic and clinical studies typically consider exposures.

Figure 4-1 illustrates the structure of the lower airways with progression from the large airways to the alveolus. Understanding dosimetry as it relates to O₃-induced injury is complex due to the fact that O₃ interacts primarily with the ELF, which contains surfactant and antioxidants. Reactive products created by O₃ can diffuse within the lung or be transported out of the lung to generate systemic effects. Antioxidant enzymes are the primary cellular defense against reactive species created by O₃. The level and type of antioxidants varies between species, regions of the RT itself, and can be altered by O₃ exposure.

A considerable number of dosimetric studies were summarized in the 1996 O₃ AQCD. These studies provided estimates of absorbed O₃ in the RT as a whole or in regions such as the upper airways (URT) or lower airways (LRT), defined as being proximal or distal to the tracheal entrance, respectively. Estimates were obtained for both humans and animals via direct measurement and mathematical modeling. The mathematical models also estimated O₃ doses to specific target sites such as the proximal alveolar region (PAR; first generation distal to the terminal bronchioles), which is also referred to as the centriacinar region (CAR; junction of conducting airways and gas exchange region) in some studies.

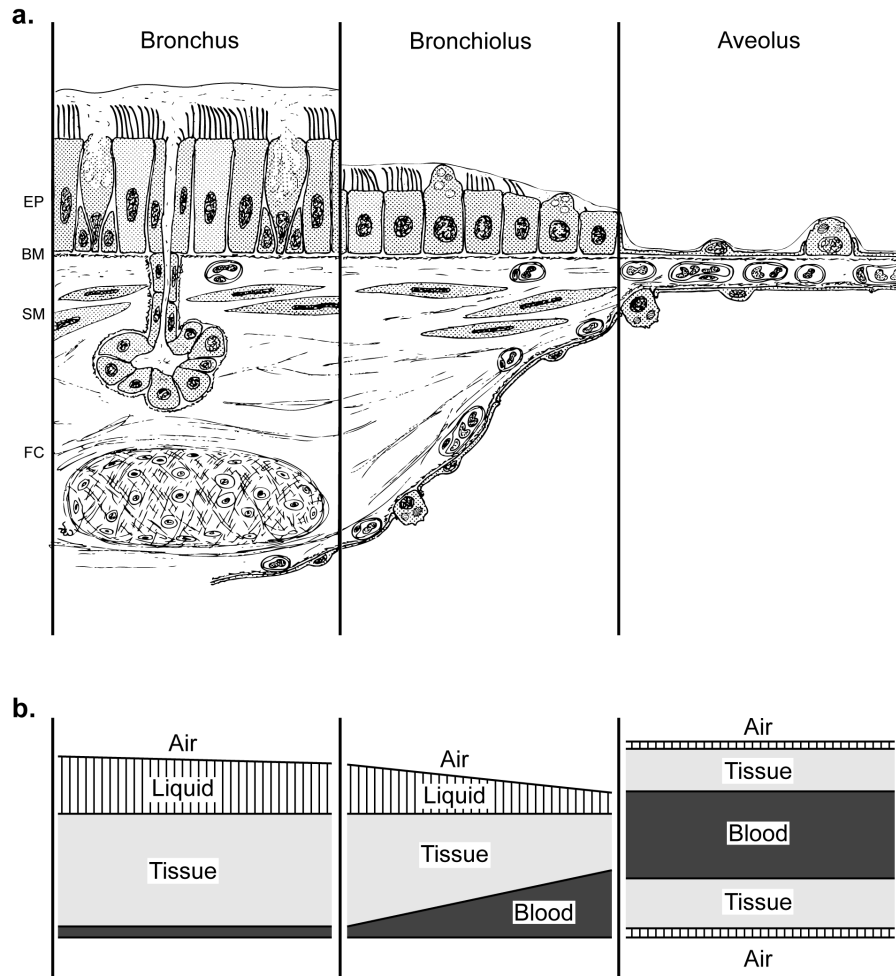


Figure 4-1. Structure of lower airways with progression from the large airways to the alveolus. Panel (a) illustrates basic airway anatomy. Structures are epithelial cells, EP; basement membrane, BM; smooth muscle cells, SM; and fibrocartilaginous coat, FC. Panel (b) illustrates the relative amounts of liquid, tissue, and blood with distal progression. In the bronchi there is a thick liquid lining over a relatively thick layer of tissues. Even highly soluble materials moving from the air into the liquid layer have minimal systemic access via the blood. With distal progress, the protective liquid lining diminishes allowing increased access of compounds crossing the air-liquid interface to the tissues and the blood.

Source: Panel (a) reproduced with permission (Weibel, E. R. [1980] Design and structure of the human lung. In: Fishman, A. P., ed. Pulmonary Diseases and Disorders. New York, NY: McGraw-Hill; p. 231).

In general, the consensus of experimental and modeling studies summarized in the 1996 O₃ AQCD supported the following conclusions: (1) for the URT, animal and human studies suggested that O₃ uptake is greater in the nose than the mouth but the effect of flow on uptake was equivocal; (2) for the LRT, predicted tissue doses (O₃ flux to liquid-tissue interface) were very low in the trachea, increased to a maximum in the terminal bronchioles or first airway generation in the pulmonary region, and rapidly decreased with distal progression; (3) increasing tidal volume (V_T) increases O₃ uptake, whereas, increasing flow or breathing frequency (f_B) decreases O₃ uptake; (3) increasing flow shifts O₃ uptake to the smaller peripheral airways, i.e., toward the CAR; and (4) similarly, the effect of exercise is to significantly increase the pulmonary region total dose (mass of O₃) and the CAR dose (mass per unit surface area).

Some cross-species in vivo comparisons were described in the 1996 O₃ AQCD. For instance, comparing bronchoalveolar lavage (BAL) cells from rats and humans, it was estimated that a 0.4 ppm O₃ exposure in exercising humans gave 4 to 5 times the O₃ dose (retained) relative to rats exposed at rest to the same concentration. In vitro dosimetry studies in the 1996 O₃ AQCD using isolated lung preparations showed that uptake efficiency is chemical-reaction dependent, indicating the importance of reaction product formation. These reaction products, created mainly by the ozonolysis of polyunsaturated fatty acids, included hydrogen peroxide, aldehydes, and hydroxyhydroperoxides, which are mediators of O₃ toxicity. Other products are created by the reaction of O₃ with other ELF constituents, all of which must be considered in understanding the dosimetry of O₃.

The next two sections (4.2.1 and 4.2.2) review the available new experimental studies on O₃ dosimetry. Table AX4-1 in Annex AX4 summarizes the new human studies.

4.2.1 Bolus-Response Studies in Humans

The bolus-response method has been used by the Ultman group as an approach to explore the distribution of O₃ absorption in the airways of humans. This noninvasive method consists of an injection of a known volume and concentration of O₃ at a predetermined point during inspiration. Ozone uptake is the amount of O₃ absorbed during a breath relative to the amount contained in the inhaled bolus. Figure 4-2 illustrates the uptake of a series of O₃ boli as a

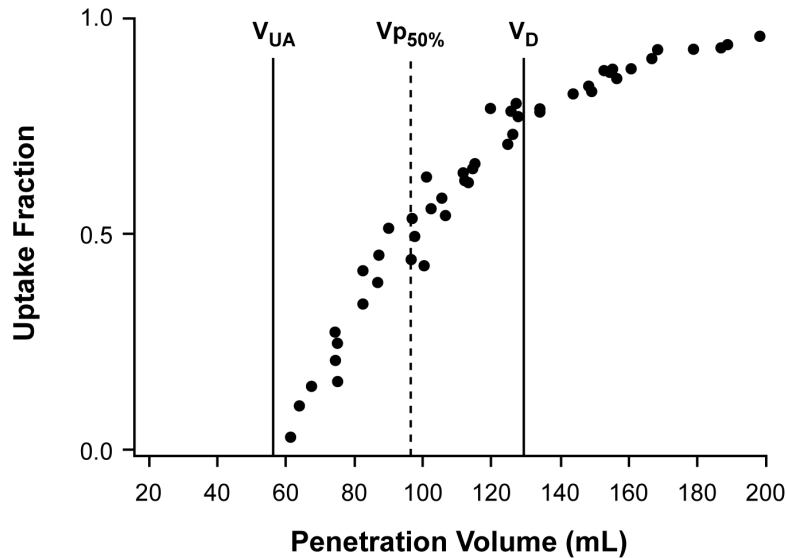


Figure 4-2. Ozone uptake fraction as a function of volumetric penetration (V_p) in a representative subject. Each point represents the O_3 uptake of a bolus inspired by the subject. The volumes, V_{UA} and V_D , are the volume of the upper airways and anatomical dead space, respectively, and $V_{P_{50\%}}$ is the V_p at which 50% of the inspired bolus was absorbed. In 47 healthy subjects (24 M, 23 F), Ultman et al. (2004) found that $V_{P_{50\%}}$ was well correlated with V_D ($r = 0.57$, $p < 0.001$) and better correlated with the volume of the conducting airways, i.e., V_D minus V_{UA} , ($r = 0.65$, $p = 0.001$).

Source: Adapted from Ultman et al. (2004).

function of volumetric penetration (V_p), i.e., the volume between the center of mass of an inhaled bolus and the end of inspiration. The inspired O_3 boli (for which the uptake fractions are illustrated in Figure 4-2) were 20 mL of 2 ppm O_3 . Kabel et al. (1994) have previously shown that varying the O_3 concentration of inspired boli between 0.4 and 4 ppm does not affect the distribution of uptake as a function of V_p .

The O_3 bolus-technique was used by Bush et al. (1996a) to ascertain differences in lung anatomy and gender that can alter the exposure-dose cascade. Forced vital capacity (FVC), total lung capacity (TLC) and anatomic dead space (V_D) were determined for ten male and ten female subjects. Differences between subjects in absorption of a 20 mL O_3 bolus injected into the inhaled breath could be explained by differences in V_D . In particular, they concluded that the

intrinsic mass transfer parameter (K_a) was proportional to the ratio of the respiratory flow to V_D . In a subsequent study, Ultman et al. (2004) showed that the volume at which 50% of an inspired O_3 bolus is absorbed was better associated with the volume of the lower conducting airways than V_D (see Figure 4-2). Bush et al. (1996a) pointed out that the applicability of their results may be limited because of their assumptions that K_a was independent of location in the RT and that there was no mucous resistance. They further suggested that the dependence of K_a on flowrate and V_D be restricted to flowrates ≤ 1000 mL/s until studies at higher rates have been performed.

Nodelman and Ultman (1999) demonstrated that the uptake distributions of O_3 boli were sensitive to the mode of breathing and to the airflow rate. As flowrates increased from 150 to 1000 mL/s, O_3 penetrated deeper into the lung and penetration was further increased by oral relative to nasal breathing. The authors suggest that the switch from nasal to oral breathing coupled with increases in respiratory flow as occurs during exercise causes a shift in the O_3 dose distribution, allowing O_3 to penetrate deeper into the lung, increasing the potential for damage to bronchiolar and alveolar tissues.

More recently, Ultman et al. (2004) measured O_3 uptake using the bolus technique in 60 young healthy nonsmoking adults (32 M, 28 F). Bolus were inspired at a rate of 1 L/s, equivalent to a moderate exercise rate with a minute ventilation of 30 L/min. Figure 4-2 illustrates the O_3 uptake fraction as a function of V_p in a representative subject. Anatomic dead space was measured in 47 of the subjects (24 M, 23 F). In these subjects, the volume at which 50% of an inhaled bolus was absorbed ($V_{P_{50\%}}$) was correlated with V_D ($r = 0.57$, $p < 0.001$) and the volume of the lower conducting airways, i.e., V_D minus the volume of the upper airways, ($r = 0.65$, $p = 0.001$). The better correlation found by subtracting off the upper airways from V_D can be explained by the fact that very little O_3 is absorbed in the upper airways during oral breathing at 1 L/s. Both $V_{P_{50\%}}$ and lower airways volume were greater in males than females. These findings suggest that in females the smaller airways, and associated larger surface-to-volume ratio, enhance local O_3 uptake and cause reduced penetration of O_3 into the distal lung. It is not clear from these findings, however, if the actual anatomical location of $V_{P_{50\%}}$ differed between males and females.

A few studies have measured the effect of a continuous pollutant exposure on O₃ bolus uptake. Asplund et al. (1996) randomly exposed young healthy adults (8 M, 3 F) for 2 h [presumably at rest] to 0.0 (air), 0.12, or 0.36 ppm O₃ on 3 separate occasions separated by at least 1-wk. Ozone bolus uptake was measured preexposure and subsequently at 30 min intervals during the exposure. Ozone uptake over the V_p range of 70 to 120 mL increased after the air exposure (0.045, absolute change in absorbed fraction), decreased slightly after the 0.12 ppm O₃ exposure (-0.005), and decreased more substantially following the 0.36 ppm O₃ exposure (-0.03). For clarification, these absolute changes in uptake due to filtered air or O₃ exposures are increases or decreases from an average uptake of ~0.70 over the VP range from 70 to 120 mL (Hu et al., 1994). Relative to uptake during the air exposure, Asplund et al. (1996) found O₃ bolus uptake was significantly decreased by 30 min of the 0.12 and 0.36 ppm O₃ exposures and remained significantly decreased for the duration of these exposures.

Using a similar protocol, Rigas et al. (1997) randomly exposed young healthy adults (6 M, 6 F) for 2 h at rest to filtered air, 0.36 ppm NO₂, 0.75 ppm NO₂, 0.36 ppm SO₂, or 0.36 ppm O₃. Ozone bolus uptake (V_p range of 70 to 120 mL) was measured preexposure and every 30 min during the exposures. The results of an *F* test indicated that exposure duration (30-, 60-, 90-, 120-min) was not a significant factor, but treatment (NO₂, SO₂, etc.) was (*p* < 0.01). Ozone bolus uptake was increased by 30 min during the NO₂ and SO₂ exposures and decreased during the O₃ exposure. The authors suggested that there may be increased production of an O₃-reactive substrate in the ELF due to airway inflammation. During NO₂ and SO₂ exposures the substrate was not depleted by these gases and so could react with the O₃ bolus. During O₃ exposure the substrate was depleted, causing the fractional absorption of the O₃ bolus to decrease.

4.2.2 General Uptake Studies

Ultman and colleagues have recently completed some general uptake studies to determine the ratio of O₃ uptake to the quantity of O₃ inhaled. Uptake efficiency was determined at exposures of 0.2 or 0.4 ppm O₃ while exercising at a minute volume of approximately 20 L/min for 60 min or 40 L/min for 30 min in both men and women (Rigas et al., 2000). Uptake efficiency ranged from 0.56 to 0.98 and had a statistically significant but weak dependence on

concentration, minute volume, and exposure time. Intersubject differences had the largest influence on uptake efficiency, resulting in a variation of approximately 10%. As the quantity of O₃ retained by the RT is equal to uptake efficiency times the quantity of O₃ inhaled, relatively large changes in concentration, minute volume, or exposure time may result in relatively large changes in the amount of O₃ retained by the RT or absorbed locally. The authors concluded that for exposure times <2 h, inhaled dose (product of O₃ concentration, exposure duration, and minute ventilation) is a reasonable predictor of actual uptake as long as there are fixed concentrations of O₃ and fixed levels of exercise. More importantly, similarly exposed individuals vary in the amount of actual dose received.

Santiago et al. (2001) studied the effects of airflow rate (3 to 15 L/min) and O₃ concentration (0.1, 0.2, or 0.4 ppm) on O₃ uptake in nasal cavities of males and females. As would be expected, uptake efficiency in the nose was inversely related to the flowrate and the concentration of O₃ in the inlet air. They computed a gas-phase diffusion resistance of <24% of overall diffusion resistance, which suggested to them that simultaneously occurring diffusion and chemical reactions in the mucous layer were the limiting factors in O₃ uptake. Difference in O₃ uptake ranged from 0.63 to 0.97 at flowrates of 3 L/min and 0.25 to 0.50 at 15 L/min. The small effects of flowrate and concentration on uptake efficiency were statistically significant, but intersubject differences accounted for approximately half of the total variation in uptake efficiency. Both these general uptake studies, done at environmentally relevant O₃ concentrations, indicate that interindividual differences in fractional uptake are extremely important in O₃ dose-response relationships.

In the research mentioned above, Ultman et al. (2004) also completed continuous exposure studies. The same 60 subjects were exposed continuously for 1 h to either clean air or 0.25 ppm O₃ while exercising at a target minute ventilation of 30 L/min. This is the first study to assess ventilatory and dosimetric parameters for an entire hour of exposure. In addition to measuring pre-to-post exposure changes in FEV₁, they used the peripheral bronchial cross-sectional area available for diffusion (inferred from the alveolar slope of CO₂ expirograms) as an alternative response variable. At a fixed minute ventilation of 30 L/min, the uptake fraction of O₃ decreased with increasing f_b (see Figure 4-3) and increased with increasing V_T . The uptake fraction was

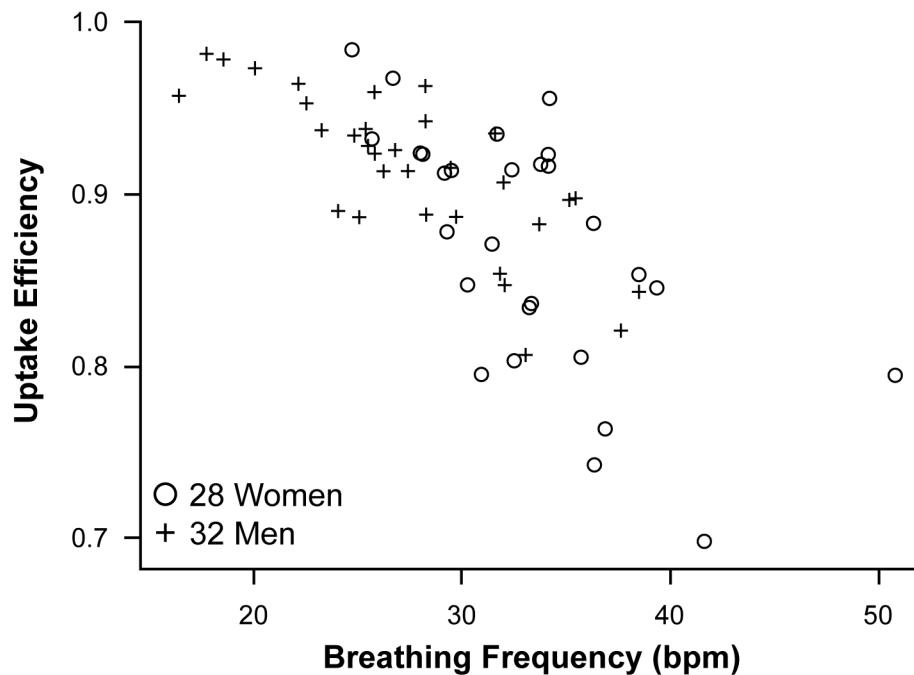


Figure 4-3. Ozone uptake efficiency as a function of breathing frequency at a minute ventilation of 30 L/min. The uptake efficiency was well correlated with breathing frequency ($r = -0.723$, $p < 0.001$) and tidal volume (*not illustrated*; $r = 0.490$, $p < 0.001$).

Source: From Ultman et al. (2004).

significantly greater in males (91.4%) than females (87.1%), which is consistent with the larger f_B and smaller V_T of the females than males. There was a small but statistically significant reduction in the breath-by-breath uptake of O_3 from 90.6% on average for the first 15 min to 87.3% on average for the last 15 min of exposure, although the biological significance of this small change is questionable. Ozone uptake rate correlated with percent changes in individual bronchial cross-sectional area but did not correlate with individual FEV_1 responses. Neither of these parameters correlated with the $V_{P_{50\%}}$ determined in the bolus studies mentioned above. The authors concluded that the intersubject differences in forced respiratory responses were not due to differences in O_3 uptake. However, these data did partially support the hypothesis that changes in cross-sectional area available for gas diffusion are related to overall O_3 retention.

Plopper et al. (1998) examined the relationship between O₃ dose and epithelial injury in Rhesus monkeys. Using ¹⁸O content in lung tissues, the respiratory bronchioles were confirmed as the site receiving the greatest O₃ dose (mass ¹⁸O per dry lung weight). Furthermore, the greatest cellular injury occurred in the vicinity of the respiratory bronchioles and was dependent on the delivered O₃ dose to these tissues. After a 2 h exposure, the antioxidant glutathione (GSH) was increased in the proximal intrapulmonary bronchus after 0.4 ppm O₃ and decreased in the respiratory bronchiole after 1.0 ppm O₃. Perhaps an adaptive response, chronic O₃ exposure leads to increased GSH levels in distal bronchioles of both rats and monkeys relative to GSH levels in filtered air-exposed animals (Duan et al., 1996).

4.2.3 Dosimetry Modeling

When all of the animal and human in vivo O₃ uptake efficiency data are compared, there is a good degree of consistency across data sets, which raises the level of confidence with which these data sets can be used to support dosimetric model formulations. Models predict that the net O₃ dose (O₃ flux to air-liquid interface) gradually decreases distally from the trachea toward the end of the TB and then rapidly decreases in the pulmonary region. However, the tissue dose (O₃ flux to liquid-tissue interface) is low in the trachea, increases to a maximum in the terminal bronchioles and the first generation of the pulmonary region, and then decreases rapidly distally into the pulmonary region. The increased V_T and flow, associated with exercise in humans or CO₂-stimulated ventilation increases in rats, shifts O₃ dose further into the periphery of the lung, causing a disproportionate increase in distal lung dose.

Table AX4-2 in the annex presents a summary of new theoretical studies of the uptake of O₃ by the RTs (or regions) of humans and laboratory animals that have become available since the 1996 review. These studies are briefly described below. Virtually all of these models have assumed that the reaction rate of O₃ in the liquid lining layer and in tissues is quasi first-order with respect to O₃ concentration. However, there is considerable discrepancy between rate constants used in models. For instance, Hu et al. (1994) and Bush et al. (2001) estimate a reaction rate constant that is more than 1000 times as large as that used by Cohen-Hubal et al. (1996). Both the uptake of O₃ at the gas-liquid interface and the fraction of O₃ (or its reaction

products) that reach epithelial cells are sensitive to the value of these reaction rate constants. A large disparity in rate constants between studies illustrates limitations in our current state of knowledge and affects the interpretation of O₃ model predictions.

Overton and Graham (1995) created a rat model combining multiple path anatomic models and one-dimensional convection-dispersion equations, which simulates transport and uptake of O₃ in airways and airspaces of the modeled TB region. Predictions from this model realistically detail O₃ transport and uptake of different but morphologically equivalent sites. Modeled lung tissue doses show variation of O₃ dose among anatomically equivalent ventilatory units as a function of path length from the trachea with shorter paths generally showing greater doses (Overton and Graham, 1995). This conflicts with Schelegle et al. (2001), who exposed rats to 1 ppm O₃ for 8 h and found that the terminal bronchioles supplied by short and long paths had similar epithelial injury. Interestingly, in rats with a C-fiber conduction block to prevent O₃-induced rapid shallow breathing, it was the long path terminal bronchioles that received the greatest epithelial injury. Overall, O₃-induced rapid shallow breathing appears to protect the large conducting airways while producing a more even distribution of injury to the terminal bronchioles (Joad et al., 2000; Schelegle et al., 2001). Postlethwait et al. (2000) have also identified the conducting airways as a primary site of acute O₃-induced cell injury. Such data must be considered when developing models that attempt to predict site-specific locations of O₃-induced injury. The early models computed relationships between delivered regional dose and response with the assumption that O₃ was the active agent responsible for injury. It is now known that reactive intermediates such as hydrohydroperoxides and aldehydes are important agents mediating the response to O₃ (further discussed in Section 5.3.1). Thus, models must consider O₃ reaction/diffusion in the ELF and ELF-derived reactions products.

Using computational fluid dynamics (CFD), Cohen-Hubal et al. (1996) modeled the effect of the mucus layer thickness in the nasal passage of a rat. Predictions of overall uptake were within the range of measured uptake. Predicted regional O₃ flux was correlated with measured cell proliferation for the CFD simulation that incorporated two regions, each with a different mucus thickness.

With a RT dosimetry model, Overton et al. (1996) investigated the sensitivity of uptake efficiency, proximal alveolar region (PAR) dose (O_3 mass per unit surface area per unit time), and PAR dose ratio to TB region volume (V_{TB}) and TB region expansion in humans and rats. The PAR was defined as the first generation distal to terminal bronchioles and the PAR dose ratio was defined as the ratio of a rat's predicted PAR dose to a human's predicted PAR dose. This ratio relates human and rat exposure concentrations so that both species receive the same PAR dose. In rats, the PAR is a region of major damage from O_3 . For each species, three values of V_{TB} were used: a mean value from the literature and the mean \pm twice the SD. For both the rat and human simulations, there were several general findings: (1) uptake efficiency and PAR dose both increased with decreasing V_{TB} , e.g., using the highest TB region mass transfer coefficient (k_{TB}), the PAR dose for $V_{TB} - 2SD$ was five times greater than the PAR dose for $V_{TB} + 2SD$, (2) uptake efficiency and PAR dose both decreased with TB expansion relative to no expansion, (3) PAR dose increased with tidal volume, (4) PAR dose increased with decreasing k_{TB} , and (5) uptake efficiency increased with k_{TB} .

Bush et al. (2001) modified their single-path model (Bush et al., 1996b) so that simulations would coincide with experimental uptake efficiency data for O_3 and Cl_2 during oral and nasal breathing. Relative to their original model, the Bush et al. (2001) model added lung expansion and modified the mass transfer coefficients for both the gas-phase (k_g) and the liquid-phase (k_l). Consistent with Overton et al. (1996), considering expansion of the TB airways reduced uptake efficiency versus no expansion. As very little inhaled O_3 reaches the peripheral lung, it was not surprising that alveolar expansion had minimal affect on uptake efficiency. Ignoring the O_3 reaction rate constant (k_r), the simulations for O_3 and Cl_2 were nearly the same since the gas-phase diffusion coefficients of O_3 and Cl_2 are similar. But for a given V_p the TB airways of the lung, experimental bolus uptake are always less for O_3 than for Cl_2 . The authors surmised that the difference between the uptake for these gases could be explained adequately based solely on the diffusive resistance of O_3 in airways surface fluid (modeled by k_r). Qualitatively, model simulations also agreed well with the experimental data of Gerrity et al. (1995).

Age- and gender-specific differences in both regional and systemic uptake in humans was modeled using a physiologically-based pharmacokinetic (PBPK) approach (Sarangapani et al., 2003). The model estimated that regional (URT, TB, pulmonary) extraction efficiency of O₃ is relatively insensitive to age and gender.

A recent attempt was made (Mudway and Kelly, 2004) to model O₃ dose-inflammatory response using a meta-analysis of 23 exposures in published human chamber studies. The O₃ concentrations ranged from 0.08 to 0.6 ppm and the exposure durations ranged from 60 to 396 min. The analysis showed linear relationships between O₃ dose and neutrophilia in bronchoalveolar lavage fluid (BALF). Linear relationships were also observed between O₃ dose and protein leakage into BALF, which suggested to the authors that a large-scale study could determine a possible O₃ threshold level for these inflammatory responses. These recent findings seem consistent with the linear relationship between O₃ dose to pulmonary tissues normalized for body weight and lavage fluid protein in rats, guinea pigs, and rabbits (Miller et al., 1988).

4.2.4 Summary and Conclusions - Dosimetry

Ozone is a highly reactive gas and powerful oxidant with a short half-life. Uptake occurs in mucous membranes of the RT where O₃ reacts with components of the ELF. Uptake efficiency is chemical-reaction dependent and the reaction products (hydrogen peroxide, aldehydes, and hydroxyhydroperoxides) created by ozonolysis of polyunsaturated fatty acids mediate O₃ toxicity. The 1996 O₃ AQCD reported that uptake of O₃ in rats is about 0.50 and in humans at rest is about 0.8 to 0.95. In humans, about 0.07 of the O₃ is removed in the larynx/trachea, about 0.50 in the head, and about 0.43 in the lungs, where the primary site of damage was believed to be the CAR. Increasing flow shifted O₃ uptake distally toward smaller airways of the lung. Studies in humans showed that increasing minute ventilation with exercise (by increasing both breathing frequency and tidal volume) causes only a small decrease in uptake efficiency by the total RT. The nasal passages appeared to absorb more O₃ than the oral passages. Comparing BAL cells, a 0.4 ppm exposure in exercising humans showed 4 to 5 times the retained dose of O₃ relative to rats exposed at rest to the same concentration.

Most new research on O₃ uptake has been performed in humans. Bolus-response studies demonstrated that a previous continuous exposure to O₃ decreases the absorption of a bolus of O₃, probably due to depletion of compounds able to absorb O₃. Continuous exposure to NO₂ and SO₂ increased absorption of a bolus of O₃. These data are of some relevance to environmental exposures where humans may receive differing concentrations of O₃ depending on time of day. Verifying prior work, the bolus-response method was used to demonstrate that O₃ bolus uptake is sensitive to the mode of breathing and to the airflow rate. As flow is increased from 150 to 1000 mL/s, O₃ boli penetrated deeper into the lung and penetration was further increased by oral versus nasal breathing. This suggests that the switch from nasal to oral breathing coupled with increases in respiratory flow as occurs during exercise causes a shift in regional O₃ dose deeper into the lung, increasing the potential of damage to bronchiolar and alveolar tissues. The finding that O₃ uptake is inversely related to airflow also agrees with earlier animal studies.

New general uptake study data demonstrate that exercising men and women receiving 0.2 or 0.4 ppm O₃ at 20 L/min for 60 min or 40 L/min for 30 min absorb 0.56 to 0.98. The absorbed fraction was affected only by large changes in concentration, minute volume, and exposure time. This suggests that for exposure times <2 h, inhaled dose (i.e., product of O₃ concentration, minute ventilation, and exposure duration) is a reasonable predictor of actual O₃ dose as long as the O₃ exposure concentration and level of exercise are relatively constant. However, individuals exposed to similar concentrations vary considerably in the amount of actual dose received. This intersubject variability has also been demonstrated in a study of O₃ uptake in nasal cavities of men and women. The absorbed fraction in the nose was inversely related to the flowrate and the concentration of O₃, suggesting that both gas phase diffusion and chemical reactions in the mucous layer were limiting O₃ uptake.

The consistency of uptake data generated in animal and human studies allow a high level of confidence in their use in dosimetry modeling. Early models predicted that net O₃ dose to ELF and tissue gradually decreases distally from the trachea toward the end of the TB and then rapidly decreases in the pulmonary region. Exercise-induced or CO₂-stimulated increases in V_T and flow, shift O₃ dose further into the periphery of the lung, causing a disproportionate increase

in distal lung dose. Localized damage to lung tissue has been modeled showing variation of O₃ dose among anatomically equivalent ventilatory units as a function of path length from the trachea with shorter paths showing greater damage.

New models have produced some refinements of earlier models such as: (1) the use of mucus resistance and thickness in describing O₃ dosimetry and determining the patterns of O₃-induced lesions; (2) the shape of the dose versus generation plot along any path from the trachea to alveoli is independent of path, with the tissue dose decreasing with increasing generation index; (3) simulations sensitive to conducting airway volume but relatively insensitive to characteristics of the respiratory airspace; (4) the importance of TB region expansion; (5) the importance of dose received in the PAR both inter-individual differences and extrapolations based on dose; and (6) reevaluation of mass transfer coefficients for conducting airways. Additionally, more recent data indicate that the primary site of acute cell injury occurs in the conducting airways and that reactive intermediates in the ELF, rather than O₃ itself, are responsible for pulmonary injury. These data must be considered when developing new models.

4.3 SPECIES HOMOLOGY, SENSITIVITY, AND ANIMAL-TO-HUMAN EXTRAPOLATION

Basic similarities exist across human and other animals species with regard to basic anatomy, physiology, biochemistry, cell biology, and disease processes. However, there are obviously some species differences that have the potential to affect both the patterns of O₃ uptake in the RT as well as responses. For instance, primates are oronasal breathers with a dichotomous branching lung structure, whereas, rodents are obligate nasal breathers with a monopodial branching lung structure (Miller et al., 1993). Even when comparing nasal breathing, differences in the nasal structure between primates and rodents can affect both the site and amount of gaseous uptake in this region (DeSesso, 1993; Morgan et al., 1989). Cellular profiles also differ between species as a function of location in the RT (Miller et al., 1993; Plopper et al., 1989; Stone et al., 1992).

The homology as it exists creates similarities in acute O₃-induced effects, especially in the RT and in lung defense mechanisms. Rodents appear to have a slightly higher tachypneic response to O₃, which is clearly concentration-dependent in most species and shows parallel exacerbation when hyperventilation (e.g., exercise or CO₂) is superimposed. What is not known is whether this is evidence of pulmonary irritant sensitivity, perhaps as a prelude to toxicity, or whether tachypnea is a defensive action taken by the respiratory system to minimize distal lung O₃ deposition. Airway or lung resistance in humans is not affected appreciably by acute exposure to O₃, except under conditions of heavy exercise; animals appear to need high-level exposures or special preparations that bypass nasal scrubbing. Dynamic lung compliance (C_{dyn}) has been shown to have small magnitude decreases in response to O₃ in some studies across species, but it is thought that these changes are of little biological significance for ambient exposures. Spirometric changes, the hallmark of O₃ response in humans, occur in rats, but to a lesser degree. It is unclear, however, the degree to which anesthesia (rat) and the comparability of hyperventilation induced by CO₂ (rat) or exercise (human) may influence this difference in responsiveness. Collectively, the acute functional response of laboratory animals to O₃ appears quite homologous to that of the human.

When humans are exposed to O₃ repeatedly for several consecutive days, lung function decrements subside, and normal spirometric parameters are regained (see Section 6.6). This phenomenon of functional attenuation also has been demonstrated in rats, not only in terms of spirometry, but also in terms of the classic tachypneic ventilatory response. Full or partial attenuation of some BAL parameters also appears to occur in both rats and humans, but exposure scenario appears to play a role; other cellular changes do not attenuate (see Section 6.9.4). Existing epidemiologic studies provide only suggestive evidence that persistent or progressive deterioration in lung function is associated with long-term oxidant-pollutant exposure (see Chapter 7). With chronic, repeated exposures to ≥ 0.12 ppm O₃, however, laboratory animals demonstrate changes in lung structure, function, and biochemistry that are indicative of airway irritation and inflammation with the possible development of chronic lung disease (U.S. Environmental Protection Agency, 1996). Based on the apparent homology of these responses

between humans and laboratory animals, animal studies appear to provide a means for assessing such chronic health concerns.

For similarly exposed animals (i.e., same O₃ concentration and exposure duration), intra- and inter-species differences in pulmonary responses are observed as a function of animal age, ventilation, and antioxidant status. Examination of BAL constituents show that the influx of inflammatory cells and protein from the serum is influenced by species, but perhaps to less extent than by ventilation and antioxidant status. Based on lavage protein levels (mg protein per mL lavage fluid), Hatch et al. (1986) reported that guinea pigs were the most responsive (to ≥0.2 ppm); rabbits were the least responsive (2.0 ppm only); and rats, hamsters, and mice were intermediate (effects at 0.5 to 1.0 ppm). Recognizing that there are differences in O₃ doses to tissues between species, Miller et al. (1988) examined the relationship protein levels reported by Hatch et al. (1986) and predicted pulmonary tissue dose (mass O₃ per body weight). Miller et al. (1988) found that protein levels in guinea pigs increased more rapidly with tissue dose than in rats and rabbits.

A species' susceptibility to the effects of O₃ exposure may be due, in part, to biochemical differences among species. Evidence for this is provided by differences in activity of SD rat and rhesus monkey CYP monooxygenases elicited by O₃ exposure (Lee et al., 1998). Additional characterization of species- and region-specific CYP enzymes will create a better understanding of the differences in response to O₃. This will allow more accurate extrapolation from animal exposures to human exposures and toxic effects.

Antioxidant metabolism varies widely among species, which can greatly influence the effects of O₃ (*discussed in greater detail in 5.2.1.3*). The guinea pig appears to be the species most susceptible to O₃. Early studies ranked mice > rats > guinea pigs in order of antioxidant responsiveness to O₃ challenge. Guinea pigs have been shown to have lower basal levels of GSH transferase activity, lower activity of GSH peroxidases, and lower levels of vitamin E compared to rats. However, differences in the levels of antioxidants between species and regions of the lung do not appear to be the primary factor determining susceptibility to O₃-induced tissue injury (Duan et al., 1993, 1996). Plopper et al. (1998) concluded that in monkeys there was a close association between site-specific O₃ dose, the degree of epithelial injury, and

reduced-glutathione depletion. Within a species, antioxidant defenses against O₃ can also vary with animal age (Servias et al., 2005) and exposure history (Duan et al., 1996).

Because cytokine and chemokine responses are so important in an animal's defense against O₃ exposure, comparisons of differences in species expression and activity of these inflammatory mediators is necessary. Aarsalane et al. (1995) compared guinea pig and human AM recovered in BALF and subsequently exposed in vitro to 0.1 to 1 ppm for 60 min. Measurement of inflammatory cytokines showed a peak at 0.4 ppm in both species. Guinea pig AM had an increase in IL-6 and TNF α while human AM had increases in TNF α , IL-1b, IL-6 and IL-8. This exposure also caused an increase in mRNA expression for TNF α , IL-1b, IL-6 and IL-8 in human cells. At 0.1 ppm exposures, only TNF α secretion was increased. These data suggest both qualitatively and quantitatively similar cytokine responses in AM from guinea pigs and humans. However, these in vitro AM responses can not be extended directly to an in vivo scenario since similar O₃ exposures (concentration and duration) do not give the same O₃ doses in different species (Hatch et al., 1994; Miller et al., 1978).

Species differences in morphological responses to O₃ exposure have been characterized by Dormans et al. (1999), as discussed in previous sections. Dormans et al. (1999) continuously exposed rats, mice, and male guinea pigs to filtered air, 0.2, or 0.4 ppm O₃ for 3, 7, 28, and 56 days. The animals exposed for 28 days were examined at 3, 7, or 28 days PE. Depending on the endpoint studied, the species varied in sensitivity. Greater sensitivity was shown in the mouse as determined by biochemical endpoints, persistence of bronchiolar epithelial hypertrophy, and recovery time. Guinea pigs were more sensitive in terms of the inflammatory response though all three species had increases in the inflammatory response after three days that did not decrease with exposure. These data on inflammation are in general agreement with Hatch et al., (1986), discussed above. In all species, the longest exposure to the highest O₃ concentration caused increased collagen in ductal septa and large lamellar bodies in Type II cells, but that response also occurred in rats and guinea pigs at 0.2 ppm. No fibrosis was seen at the shorter exposure times and the authors question whether fibrosis occurs in healthy humans after continuous exposure. The authors do not rule out the possibility that some of these

differences may be attributable to differences in total inhaled dose or dose actually reaching a target site. Overall, the authors rated mice as most susceptible, followed by guinea pigs and rats.

Comparisons of airway effects in rats, monkeys and ferrets resulting from exposures of 1.0 ppm O₃ for 8 h (Sterner-Kock et al. 2000) demonstrated that monkeys and ferrets had similar inflammatory responses and epithelial necrosis. The response of these two species was more severe than that seen in rats. These data suggest that ferrets are a good animal model for O₃-induced airway effects due to the similarities in pulmonary structure between primates and ferrets. However, the mechanisms of O₃ effects at these high concentrations may differ from those at more realistic levels.

A number of species, including nonhuman primates, dogs, cats, rabbits, and rodents, have been used to study the effects of O₃ exposure on airway bronchoconstriction. A commonly used model of bronchospasm utilizes guinea pigs acutely exposed to high O₃ concentrations (2 to 3 ppm) to induce airway hyperreactivity (AHR). As mentioned earlier, the model is helpful for determining mechanistic aspects of AHR, but is not really relevant for extrapolation to potential airway responses in humans exposed to ambient levels of O₃. Additionally, guinea pigs have been shown to have AHR in other studies that is very similar to asthmatic humans, but the utility of guinea pig data is somewhat limited by their disparity from other animal models.

The rat is a key species used in O₃ toxicological studies, but the rat has both behavioral and physiological mechanisms that can lower core temperature in response to acute exposures, thus limiting extrapolation of rat data to humans. Iwasaki et al. (1998) evaluated cardiovascular and thermoregulatory responses to O₃ at exposure of 0.1, 0.3, and 0.5 ppm O₃ 8 hrs/day for 4 consecutive days. A dose-dependent disruption of HR and T_{co} was seen on the first and second days of exposure, which then recovered to control values. Watkinson et al. (2003) exposed rats to 0.5 ppm O₃ and observed this hypothermic response, which included lowered HR, lowered T_{co}, and increased inflammatory components in BALF. The authors suggested that the response is an inherent reflexive pattern that can possibly attenuate O₃ toxicity in rodents. They discuss the cascade of effects created by decreases in T_{co}, which include: (1) lowered metabolic rate, (2) altered enzyme kinetics, (3) altered membrane function, (4) decreased oxygen consumption and demand, (5) reductions in minute ventilation, which would act to limit the dose

of O₃ delivered to the lungs. These effects are concurrent with changes in HR which lead to: (1) decreased CO, (2) lowered BP, and (3) decreased tissue perfusion, all of which may lead to functional deficits. The hypothermic response has not been observed in humans except at very high exposures, which complicates extrapolation of effects in rats to humans.

The importance of animal studies derives from their utilization in determining cause-effect relationships between exposure and health outcome, but the animal data must be integrated with epidemiological studies and controlled human clinical studies. Animal studies can corroborate both clinical and epidemiology studies and further provide important data that is impossible to collect in human studies. Toxic pulmonary and extrapulmonary effects following O₃ exposure have been well-studied in rodents, nonhuman primates, and a few other species; so, extrapolation, both qualitative and quantitative, to human exposures and consequent health effects is possible. Quantitative extrapolation, required to determine what specific exposure is likely to cause an effect in humans, is theoretically founded on the equivalency of mechanisms across species. At the molecular level, O₃ acts on the carbon-carbon double bond in polyunsaturated fatty acids and on sulfhydryl groups in proteins, both of which are found within cell membranes in animals and humans. At higher levels of cellular organization, cells affected in animals (e.g., AMs, Type 1 cells) have similar functions in humans, and organ systems (e.g., respiratory system) have major interspecies similarities. However, interspecies differences do occur and complicate extrapolation.

Quantitative extrapolation, which involves a combination of dosimetry and species sensitivity, still requires more research before it can be fully realized. Knowledge of dosimetric animal-to-human extrapolation is more advanced than that of species-sensitivity, but extrapolation models have not been completely validated, and therefore, significant uncertainties remain. Mathematical modeling of O₃ deposition in the lower RT (i.e., from the trachea to alveoli) of several animal species and humans shows that the pattern of regional dose is similar, but that absolute values differ. In spite of structural and ventilatory differences between species, the greatest predicted tissue dose is to the CAR. Even though the CAR of rats has very rudimentary respiratory bronchioles, compared to well-developed ones in primates, the CAR of both rats and nonhuman primates respond similarly to O₃.

Experimental measurement of delivered O₃ doses estimate that total respiratory uptake is ~47% in laboratory animals and ~87% in exercising humans, while nasopharyngeal removal is ~17% in rats and ~40% in humans. The previous O₃ AQCD (U.S. Environmental Protection Agency, 1996) provided the first quantitative animal-to-human extrapolation of morphological changes in the proximal alveolar region using rat and monkey studies. The extrapolation predicted that a 9-year-old child would have a 20% or 75% increase in PAR tissue thickness if their sensitivity to O₃ was equal to that of a rat or monkey, respectively. Adults would have 15 or 70% increase, suggesting the potential for chronic effects in humans. In spite of the significant uncertainties, this extrapolation raises concern about the potential for chronic effects in humans

Experiments using 2 h exposures to 0.4 ppm ¹⁸O₃ suggested that exercising (15 min intervals, rest and exercise at 60 L/min) humans received a 4- to 5-fold higher ¹⁸O₃ concentrations in BAL than resting rats (Hatch et al., 1994). That level of exposure increased BAL protein and PMNs in humans, while a concentration of 2.0 ppm in rats was necessary for similar effects. Caveats in the interpretation of ¹⁸O₃ studies include: (1) only a very small portion of the labeled compound is recoverable to assess incorporation; and (2) if species being compared differ in physiocochemical factors controlling mass transfer and downstream O₃ metabolism, it could cause significant differences in the amount of inhaled ¹⁸O₃ that is detected during subsequent tissue analysis. Further, species differences in pulmonary anatomy, ventilation, antioxidants, and susceptibility all influence dose, repair processes, and tolerance to subsequent O₃ exposure. Important differences between exercising humans and resting rats that can affect tissue O₃ dose include: (1) increased ventilation and O₃ delivery with exercise; (2) decreased pulmonary ventilation and body temperature during O₃ exposure in rats; (3) diminished dose received in rats due to their burying their noses in their fur during exposure; and (4) increased concentration of antioxidants in ELF in rats compared to humans. These antioxidants are important for converting O₃ to inactive products before toxicity occurs (Kari et al., 1997; Gunnison and Hatch, 1999; Plopper et al., 1998), though this quenching is not quantitative. These and possibly other differences between rats and humans suggest that a 2 ppm exposure in nonexercising rats approximates a 0.4 ppm exposure in exercising humans.

Further comparisons of exercising human exposure to 0.1 ppm for 6 hours (Devlin et al., 1991) and resting rat exposure to 0.3 ppm show inflammatory and permeability changes in humans but not rats.

4.3.1 Summary and Conclusions: Species Homology, Sensitivity, and Animal-to-Human Extrapolation

Comparisons of acute exposures in rats and humans suggest that, though both species have similar qualitative responses to O₃ exposure, there are interspecies mechanistic disparities that necessitate careful comparisons of dose-response relationships. There is no perfect nonhuman species with which to model O₃ toxicity. All have limitations that must be considered when attempting to extrapolate to human exposures. Awareness of these limitations, even at the level of subtle strain differences within a test species, is extremely important. The currently available data suggest that LOELs in resting rats are approximately 4- to 5-fold higher than for exercising humans for toxicological endpoints including BAL protein and BAL PMNs. Studies comparing species-specific differences in O₃-induced effects showed that guinea pigs were the most susceptible, rabbits the least susceptible, and rodents intermediate in susceptibility. The recent work being done utilizing various mouse strains with differing sensitivities to O₃ will help us to understand the extremely complex inter-individual differences in human sensitivity to O₃.

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5. TOXICOLOGICAL EFFECTS OF OZONE AND RELATED PHOTOCHEMICAL OXIDANTS IN LABORATORY ANIMALS AND IN VITRO TEST SYSTEMS

5.1 INTRODUCTION

A wide range of effects of ozone (O₃) has been demonstrated in laboratory animals. The major research findings are that environmentally relevant levels of O₃ cause lung inflammation; decreases in host defenses against infectious lung disease; acute changes in lung function, structure, and metabolism; chronic lung disease, some elements of which are irreversible; and systemic effects on target organs (e.g., brain, heart, liver, immune system) distant from the lung. The research also has served to expand the understanding of mechanisms of O₃ toxicity and the relationships between concentration and duration of exposure.

The framework for presenting the health effects of O₃ in animals begins with a presentation of respiratory tract effects, followed by systemic effects, and then interactions of O₃ with other common co-occurring pollutants. The information discussed in this chapter is founded on a very wide body of literature on studies in laboratory animals and on in vitro test systems of animal cell lines and organ systems that may mimic responses in intact animals. The direct effects of O₃ in humans are discussed in the following chapter (Chapter 6).

This chapter is not intended to be a compendium of all that is known about toxicologic effects of O₃; rather, it is an update of the toxicology chapter from the last previous O₃ criteria document (U.S. Environmental Protection Agency, 1996), or 1996 O₃ AQCD, and other reviews of the earlier published literature. The historical O₃ literature is very briefly summarized in an opening paragraph of each section or subsection. That paragraph is intended as a very concise overview of previous work, and the reader is referred to the 1996 O₃ AQCD for more detailed discussion of the literature prior to the early 1990's. Each section then continues with brief discussions of the key new studies (or somewhat older studies that were not included in the 1996 O₃ AQCD). Longer discussions of new studies are included where warranted. Sections are ended with comparisons of data from the previous AQCD with new data, and basic conclusions are drawn. Summaries of new studies and results are provided in tables in Annex AX5.

Except for nitrogen dioxide (NO₂), the subject of another criteria document (U.S. Environmental Protection Agency, 1993), there is very little relevant information on other photochemical oxidants in the published literature. What is known about the effects of these other oxidants is also summarized briefly in this chapter.

5.2 RESPIRATORY TRACT EFFECTS OF OZONE

5.2.1 Biochemical Effects

Biochemically detected effects of O₃ are integrally involved in effects on both structure and function (respiratory and nonrespiratory) of the respiratory tract. Changes in xenobiotic metabolism, antioxidant metabolism and oxygen consumption, lipids and arachidonic acid metabolism, and collagen metabolism are all observed with O₃ exposure, though the mechanisms and associations are not fully understood.

5.2.1.1 Cellular Targets of Ozone Interaction

Ozone has the potential to interact with a wide range of different cellular components that include polyunsaturated fatty acids (PUFAs); some protein amino acid residues; and some low-molecular-weight compounds that include glutathione (GSH), urate, vitamins C and E, and free amino acids. Early work demonstrated that O₃, being a highly reactive compound, does not penetrate much beyond the epithelial lining fluid (ELF); and reaction/diffusion analyses suggest that O₃, at environmentally-relevant concentrations, diffuses no more than 0.1 to 0.2 μm into the ELF. However, Miller (1995) points out that throughout the respiratory tract, the ELF varies in thickness and that the distal conducting airways may have only a patchy lining layer which allows O₃ to react directly with cell membranes. Various models utilizing differing reaction kinetics and rate constants do not agree as to whether or not inhaled O₃ can penetrate the ELF and reach the epithelial cells. Ozone-induced cell damage results, in part, from its reactions with PUFAs to form stable but less reactive ozonide, aldehyde, and hydroperoxide reaction products. These reaction products (Criegee ozonides and hydroxyhydroperoxides) may act as signal transduction molecules involved in signaling of cellular responses such as inflammation and thus mediate O₃ toxicity. These reactions are summarized in Figure 5-1. Studies published since the 1996 AQCD are listed in Table AX5-1.

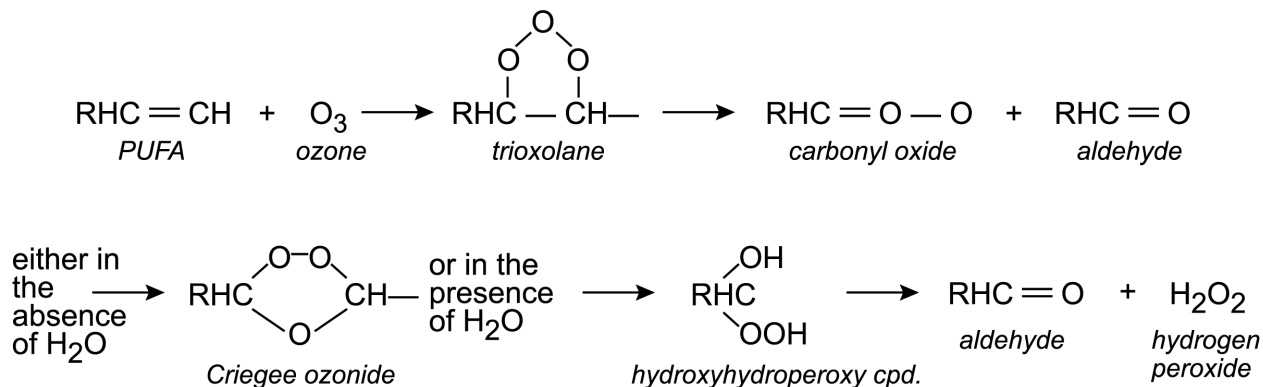


Figure 5-1. Schematic overview of ozone interaction with epithelial lining fluid and lung cells. It should be noted that not all secondary reaction products are shown.

Frampton et al. (1999) demonstrated the ozonation of PUFA to form nonanal and hexanal in rat bronchioalveolar lavage (BAL) after exposures to 0.22 ppm O₃ for 4 h with exercise. Increases in nonanal were not accompanied by significant changes in lung function, in epithelial permeability, or in airway inflammation. Hexanal levels did not increase significantly and levels of both aldehydes returned to baseline by 18 h postexposure (PE). Pryor et al. (1996) exposed rats to 0.5 to 10 ppm O₃ both with and without 5% CO₂ to measure the amount of aldehyde generated in BAL and also the rate of disappearance of aldehydes from the ELF following the O₃ exposure. Ozone exposure with CO₂ increased the tidal volume and the yield of aldehydes, with a maximal aldehyde yield at 2.5 ppm for 1 h. Absolute yields were impossible to ascertain in this system, because absorption of O₃ is unknown and aldehyde recovery is not complete due to loss of aldehyde by volatilization and by diffusion into underlying tissue. The data showed that at 0.5 ppm O₃ with 5% CO₂, levels of hexanal and nonanal increased at 30 min, decreased slightly from that levels at 60 min, was maximal at 90 min and then dropped to 60 min levels at 120 min. Heptanal levels did not change appreciably during this time course. Levels of these aldehydes were dependent on a dynamic relationship between their production and disappearance from the ELF. The authors stated that O₃ is the limiting reagent in this process because the amount of PUFA far exceeds the amount of O₃ on a molar basis. Because of the limitations of measuring aldehydes in this study protocol, the results are not useful for quantitative dosimetry; however,

the authors suggest that the study does serve to demonstrate the use of aldehydes as biomarkers of O₃ exposure, as nonanal is produced in an O₃-specific pathway.

Postlethwait et al. (1998) utilized three biologically relevant models (isolated epithelial lining fluid, intact lung, and liposome suspensions) to determine the O₃-induced production of heptanal, nonanal and hexanal in an attempt to estimate formation of lipid-derived bioactive compounds. Exposures used were 0.25 to 1.0 ppm for 30 to 60 min. The results suggest that PUFAs directly react with O₃ and the amount of bioactive lipids produced is inversely related to ascorbic acid (AA) availability. The authors caution that there are limitations to the use of measurements of these reaction products in determining O₃ dose-response relationships due to the heterogenous nature of O₃ reactions in the epithelial lining fluid. Connor et al. (2004) have recently examined the reactive absorption of O₃ (0.3 to 1.1 ppm for 1 to 2 h) within ELF using interfacial films composed of dipalmitoylglycero-3-phosphocholine (DPPC) and rat lung lavage fluid. The films reduced O₃-reactive absorption by antioxidants. Further experiments using a human lung fibroblast cell line exposed to O₃ demonstrated that AA produced cell injury, that high levels of O₃ and AA were needed to induce cell injury, and that DPPC films reduced the amount of cell injury. Based on these data, the authors suggest that O₃ reactions with ELF substrates cause cell injury, that films of active, saturated phospholipids reduce the local dose of O₃-derived reaction products, and that these interfacial phospholipids modulate the distribution of inhaled O₃ and the extent of site-specific cell injury.

Recent studies have examined the formation of ozonation products such as 4-hydroxynonenal (HNE), a toxic aldehyde that reacts with cysteine, histamine, and lysine amino acid residues and creates protein adducts. Hamilton et al. (1998) demonstrated (see Chapter 6) using human alveolar macrophages (AMs) exposed to 0.4 ppm O₃ for 1 h that exposure caused apoptosis, an increase in a 32-kDa protein adduct, and an increase in ferritin and a 72-kDa heat shock protein. By exposing AM to HNE in vitro, all of these effects are replicated, which the authors interpret to mean that creation of protein adducts and apoptotic cell death are cellular toxic effects of acute O₃ exposure that are mediated, at least in part, by HNE.

These recent reports, combined with observations reported in the previous O₃ AQCD (U.S. Environmental Protection Agency, 1996), suggest that interactions of O₃ with cellular components and ELF generate toxic ozonation products and mediate toxic effects through these products.

5.2.1.2 Monooxygenases

Both short- and long-term exposures to O₃ have been shown to enhance lung xenobiotic metabolism, possibly as a result of changes in the number and function of bronchiolar epithelial Clara cells and alveolar epithelial Type II cells. Studies of the effects of O₃ on lung monooxygenases are listed in Table AX5-2. Early studies showed that exposure to O₃ increased CYP 2B1 (the major CYP isoform in rat lung) content and activity in rat lung. Ozone exposures also caused hypertrophy and hyperplasia of CYP 2B1-immunoreactive Clara cells. Comparisons of rat and rhesus monkey CYP isoforms demonstrated species-specific and region-specific (e.g., trachea, parenchyma) differences in the activities of P450 isoforms (Lee et al., 1998).

Watt et al. (1998) found that, with both acute (8 h, 1 ppm) and chronic (90 days, 1 ppm) exposures in rat, 1 ppm O₃ increased CYP 2E1 in a region-specific manner. Paige et al. (2000a) showed that a long-term exposure (0.8 ppm, 8 h/day for 90 days) increased the activity of CYP 2B in distal lung but not in trachea or intrapulmonary airways. Studies have focused on P450 gene expression to examine possible genetic mechanisms that may explain differential O₃-sensitivity (Mango et al., 1998). Mice (129 strain) deficient in Clara cell secretory protein (CCSP^{-/-}), which are oxidant-sensitive, were exposed to 1 ppm O₃ for 2 hours. The CCSP null mice demonstrated increases in interleukin-6 (IL-6) and metallothionein (MT) mRNA that preceded decreases in Clara cell CYP 2F2 mRNA (normally expressed at high levels in mouse lung) levels. In 129 strain wild-type (WT) mice, RNA levels changed similarly, but to a lesser degree. These data suggest a protective role against oxidant damage for CCSP, and further, that genetic susceptibility to oxidant stress may be mediated, in part, by the gene coding for CCSP.

5.2.1.3 Antioxidants, Antioxidant Metabolism, and Mitochondrial Oxygen Consumption

Ozone also undergoes reactions with AA, reduced glutathione (GSH), and uric acid (UA), all antioxidants present in ELF (see Figure 5-2, A). In vivo experiments have shown that reactions with O₃ occur preferentially with antioxidants compared to proteins and lipids also present in ELF. This is a protective interaction, but even with environmentally relevant exposures to O₃, the reactivity of O₃ is not quantitatively quenched. Antioxidants offer some protection from O₃ exposure but often are not maintained at concentrations sufficient to fully protect the lung. Thus, O₃-induced cell injury occurs in both the lower and upper respiratory tract. Early work has shown that acute (1 week) exposures to <1 ppm O₃ increase antioxidant

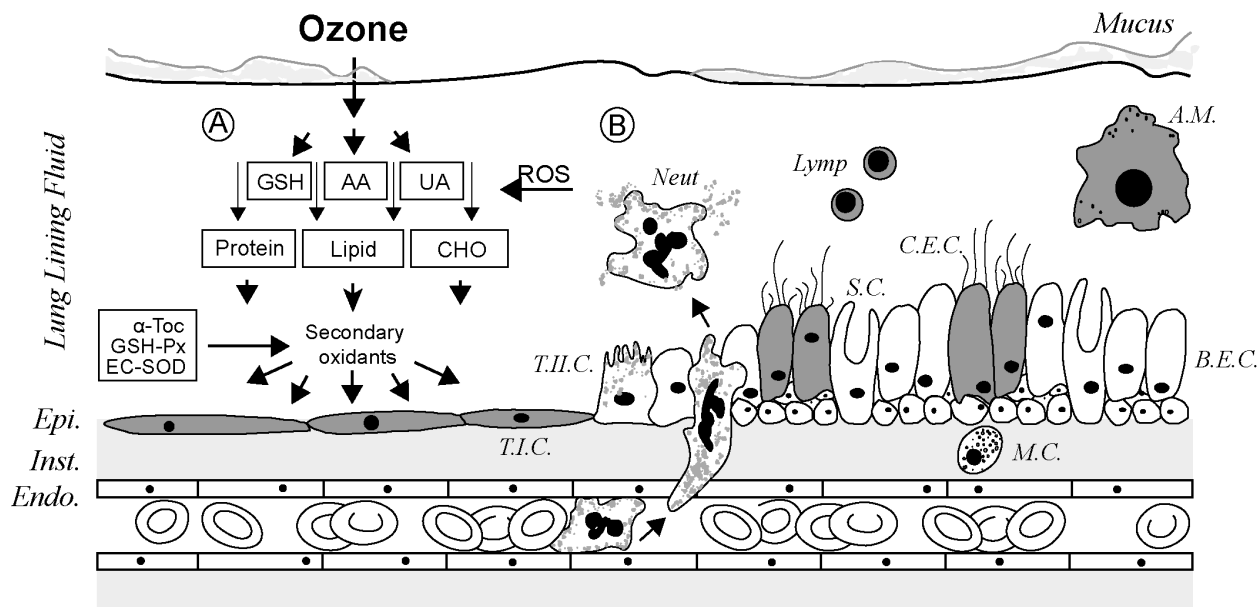


Figure 5-2. The major cellular targets and proposed mechanisms of ozone toxicity in the lung. Abbreviations: GSH - reduced glutathione; AA - ascorbate; UA - urate; α -Toc - alpha-tocopherol; GSH-Px - glutathione peroxidase; EC-SOD - extracellular superoxide dismutase; T.I.C. - Type I cells; T.II.C. - Type II cells; Neut - neutrophils; Lymph - lymphocytes; Epi. - epithelium; Inst. - interstitium; Endo. - endothelium; M.C. - monocyte; C.E.C. - ciliated epithelial cells; A.M. - alveolar macrophage; B.E.C. - bronchial epithelial cells; ROS - reactive oxygen species.

Source: Mudway and Kelly (2000), with permission.

metabolism, including levels of cytosolic enzymes glucose-6-phosphate dehydrogenase (G6PD), 6-phosphogluconate dehydrogenase (6PGD), glutathione reductase (GR), and glutathione peroxidase (GSHPx). Reexposure after a recovery period causes increases equivalent to first-time exposures; thus, previous exposure does not appear to be protective.

Increases in enzyme activity appear to increase as a function of age, suggesting that O_3 exposure can cause greater lung injury in the older animal. This has been attributed to differences in dose reaching lung target sites, differing base levels of antioxidants and antioxidant enzymes, and differences in cellular sensitivity. Species differences exist in antioxidant metabolism, with guinea pigs being very sensitive to O_3 due to their diminished increases in antioxidants and antioxidant enzymes. Chronic exposures of rats to urban patterns

of O₃ (daily peaks of 0.25 ppm) caused increases in GSHPx and GR, but not superoxide dismutase (SOD). The enzyme changes could be accounted for by differences in the steady-state cell population or in cellular antioxidant capacity. More recent studies examining antioxidants and O₃ exposure are listed in Table AX5-3.

Ozone induced both site- and cell-specific changes in copper-zinc (Cu-Zn) and manganese (Mn) SOD in rats exposed to 1.0 ppm O₃ for up to 3 months (Weller et al., 1997). CuZnSOD labeling was decreased in epithelial cells in airways and parenchyma. Mn-SOD labeling was increased in both AM and epithelial type II cells of the centriacinar region (CAR), which the authors suggest may allow these cells to tolerate further O₃ exposure. This work is in agreement with earlier work suggesting a role of SOD in protection of cells against oxidative stress.

Freed et al. (1999) evaluated the role of antioxidants in O₃-induced oxidant stress in dogs (exposed to 0.2 ppm for 6 h) by inhibiting the antioxidant transport using probenecid (an anion-transport inhibitor). Blocking antioxidant transport caused heterogeneously distributed increases in peripheral airway resistance (Raw) and reactivity, supporting the hypothesis that in the lung periphery, endogenous antioxidants moderate the effects of O₃ and that this exposure is a subthreshold stimulus for producing effects on peripheral Raw and reactivity in dogs. The authors further found that treatment with probenecid also inhibited O₃-induced neutrophilic inflammation, providing evidence for a dissociation between airway function and inflammation. This suggests that O₃-induced inflammation and airway hyperreactivity (AHR) are independent phenomena operating through multiple mechanistic pathways.

Mudway and Kelly (1998) modeled the interactions of O₃ with ELF antioxidants using a continually mixed, interfacial exposure set up with O₃ concentrations ranging from 0.1 to 1.5 ppm for durations ranging from 30 to 720 min. Uric acid was ranked the most O₃-reactive, AA the second most reactive, and GSH the least reactive. Thus, they concluded that GSH is not an important substrate for O₃, while UA appeared to be the most important reactive substrate, which confers protection from O₃ by removing it from inhaled air and limiting the amount that reaches the distal lung. By providing a substrate for O₃ reactions in the ELF, UA effectively reduces the diffusion resistance of O₃ (see Bush et al., 2001) in the tracheobronchial airways and thus may serve to limit the amount of O₃ reaching the distal lung. The authors acknowledge limitations in extrapolating these data to in vivo O₃ exposures due to the absence of surfactant lipids and airway mucus in this in vitro model.

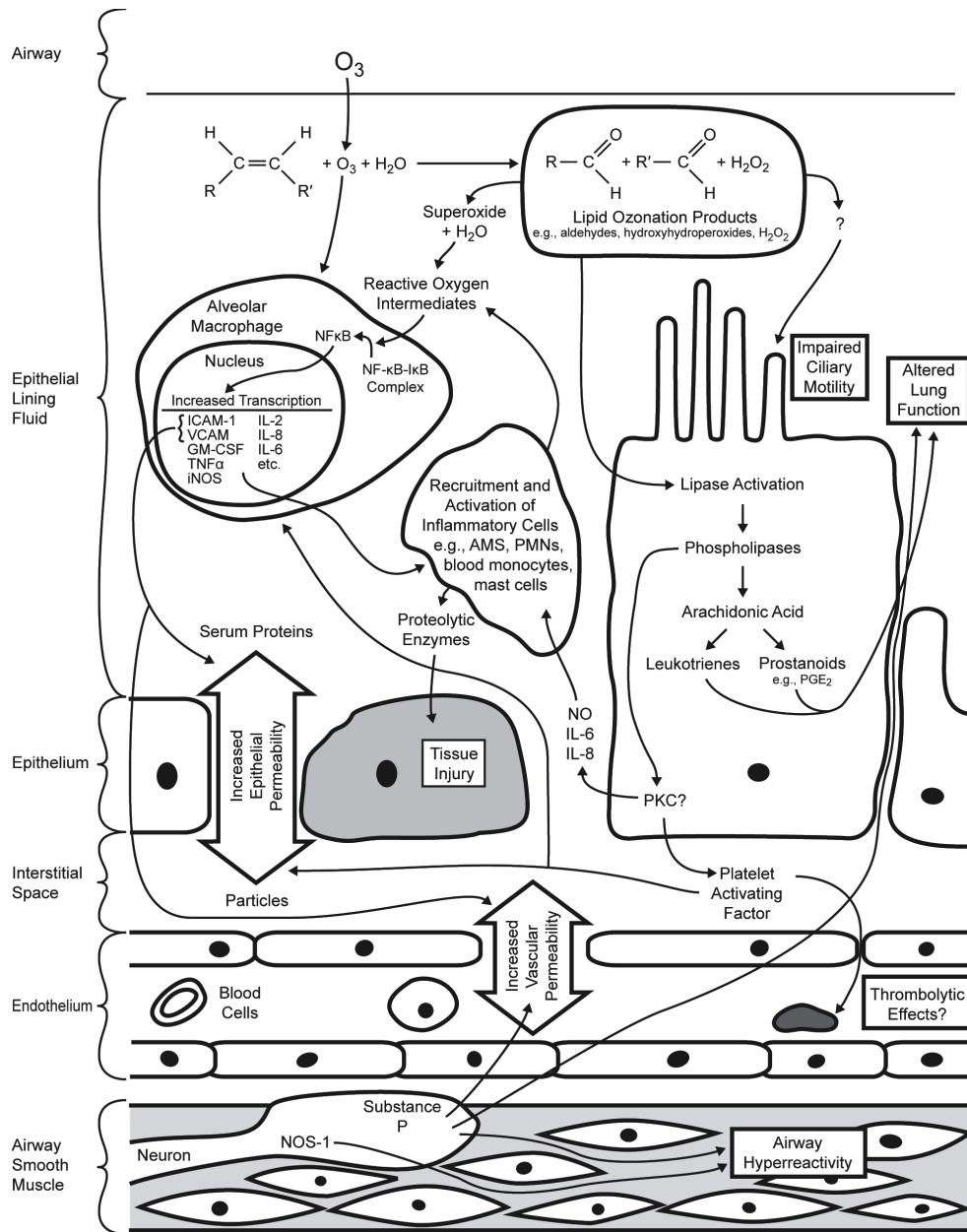
5.2.1.4 Lipid Metabolism and Content of the Lung

One of the major postulated molecular mechanisms of action of O₃ is peroxidation of mono- and polyunsaturated fatty acids and unsaturated neutral lipids in the lung. Because all of these lipids appear both in cell membranes and as secretions in the ELF, it is difficult to ascertain which lipid pool contributes to the formation of lipid ozonation products. As mentioned, O₃ can penetrate only a short distance into the ELF; and, therefore, it reacts with epithelial cell membranes only in regions of distal lung where ELF is very thin or absent. The inflammatory cascade (shown in Figure 5-3) initiated by O₃ generates a mix of secondary reactants (e.g., aldehydes), which then are likely to oxidize lipids and proteins in cell membranes.

In both acute and short-term studies, a variety of lung lipid changes occur, including an increase in arachidonic acid. Metabolism of arachidonic acid produces a variety of biologically active mediators that can, in turn, affect host defenses, lung function, the immune system, and other functions. The protein A component of surfactant is also a primary target of O₃ interaction. During the first few days of O₃ exposure, the changes in lung lipid biosynthesis can be accounted for by the alveolar epithelial proliferative repair. With longer exposures (e.g., 0.12 ppm for 90 days) an increase in PUFAs and a decrease in cholesterol-esters are seen, indicative of long-term alterations of surfactant lipid composition.

Several new studies listed in Table AX5-4 examined the effects of O₃ exposure on phospholipids in lung tissue. Ozonation of PUFAs has been shown to generate other aldehydes such as nonanal and hexanal in rats (Pryor et al., 1996; Frampton et al., 1999). These aldehydes are short-lived and found to not affect lung function (Frampton et al., 1999). These observations suggest that levels of these aldehydes are dependent on a dynamic relationship between their production and their disappearance from the ELF.

Pryor et al. (1995) proposed a cascade mechanism whereby ozonation products cause activation of specific lipases, which then trigger the activation of second messenger pathways (e.g., phospholipase A₂ or phospholipase C). This group (Kafoury et al., 1999) showed that exposure of cultured human bronchial epithelial cells to the lipid ozonation product 1-palmitoyl-2-(9-oxononanoyl)-sn-glycero-3-phosphocholine elicited release of platelet-activating factor (PAF) and PGE₂, but not IL-6. The lipid ozonation product 1-hydroxy-1-hydroperoxynonane caused release of PAF and IL-6 in these cells, but not PGE₂. These results suggest to the authors



Adapted from: Pryor et al. (1995); Krishna et al. (1998); Bhalla et al. (1999)

Figure 5-3. Mechanisms of ozone toxicity. The inflammatory cascade starts by interactions of O_3 with lipids in the epithelial lining fluid and in cell membranes. Lipid ozonation reaction products include aldehydes, hydroxyhydroperoxides, and H_2O_2 . These initiate lipase activation in epithelial cells resulting in the downstream production of phospholipases, arachidonic acid, leukotrienes, and prostanoids. Leukotrienes and prostanoids alter lung function. Phospholipases initiate the production of NO, inflammatory cytokines, and PAF. NO, IL-6, and IL-8 then participate in the recruitment and activation of inflammatory cells (AMs, PMNs, blood monocytes, and mast cells), which, in turn, can start a second cycle of inflammatory responses. PAF increases epithelial permeability and may have thrombolytic effects. Release of substance P by neurons causes altered lung function and increases vascular permeability. Production of NOS-1 by these neurons induces airway hyperreactivity.

that O₃-induced production of lipid ozonation products causes release of proinflammatory mediators that then generate an early inflammatory response.

Very new work (Ballinger et al., 2005) has shown that O₃-induced membrane oxidation is augmented by antioxidants present in ELF. They utilized a red cell membrane model exposed to 0.8 ppm O₃ for 30 min. The monolayer of cells was intermittently covered by an aqueous film consisting of rat bronchioalveolar lavage fluid (BALF) or BALF plus added antioxidants. Ascorbic acid and GSH induced dose-dependent oxidative damage to the cell membrane proteins and lipids via secondary oxidant formation. The authors concluded that early in O₃ exposure, ELF antioxidants are high enough to drive reactive absorption of O₃ into the ELF and to concurrently quench secondary reaction products, thus limiting cell injury. With continued exposure, levels of antioxidants decrease such that unreacted O₃ and cytotoxic products can diffuse to the cell membranes, causing injury. Limitations of this in vitro study include possible differences in chemical species and mechanisms compared to in vivo systems.

Uhlson et al. (2002) reacted O₃ with calf lung surfactant which resulted in the production of 1-palmitoyl-2-(9'-oxo-nonanoyl)-glycerophosphocholine (16:0a/9-al-GPCho). The biological activity of this oxidized phospholipid included: (1) decreased macrophage viability, (2) induction of apoptosis in pulmonary epithelial-like A549 cells, (3) and release of IL-8 from A549 cells. Exposures to 0.125 ppm O₃ for 2 to 4 h in this system were capable of generating biologically active phospholipids that were capable of mediating toxic effects of O₃.

In addition to PUFA, cholesterol, the most abundant neutral lipid present in ELF, is also a target of O₃. Pulfer and Murphy (2004) demonstrated the ozonolysis of cholesterol in an in vitro system using BALF isolated from rats that had been exposed to 2.0 ppm O₃ for 4 h. Production of 5-hydroperoxy-*B*-homo-6-oxa-cholestan-3,7a-diol, 5β,6β-epoxycholesterol, and 3β-hydroxy-5-oxo-5,6-seco-cholestan-6-al was shown. Additionally, both 5β,6β-epoxycholesterol and its most abundant metabolite, cholestan-6-oxo-3β,5α-diol, were demonstrated to be cytotoxic to 16-HBE cells and to inhibit cholesterol synthesis. Studies (Pulfer et al., 2005) in C57BL/6J mice exposed to 0.5, 1.0, 2.0, or 3.0 ppm O₃ for 3 h demonstrated that these oxysterols were produced in vivo also. The authors suggested that this may be an additional mechanism of O₃ toxicity. Though these oxysterol reaction products have not been fully characterized, they may be involved in O₃-induced inflammation by disrupting cellular membranes or altering signaling

between cells. Similar oxysterols have been implicated in the inflammatory cascade associated with atherosclerosis.

Thus, new work has attempted to elucidate the mechanisms by which reactions of O₃ with lipids create phospholipids that then mediate downstream toxic effects. It is uncertain whether these described changes in lipid content and/or metabolism lead to significant changes in surface tension or compliance properties of the lung.

5.2.1.5 Ozone Interactions with Proteins and Effects on Protein Synthesis

Epithelial lining fluid contains proteins arising from airway secretions and from blood. Ozone can react with four amino acid residues (cysteine, histidine, methionine, and tryptophan) and can cause oxidation of functional groups on proteins, including aldehydes, alcohol, amines and sulfhydryls. A number of enzymes have been shown to be inhibited by O₃, including cholinesterase, α 1-antiproteinase, and prostaglandin synthetase. Additionally, O₃ decreases the inhibitory activity of α 1-proteinase inhibitor, which is implicated in development of emphysema. Surfactant protein A (SP-A) is a target for O₃ toxicity by modulation of SP-A self association, vesicle aggregation, phospholipid secretion, and stimulation of AM superoxide anion generation (see Section 5.2.2.3). Further, O₃ is thought to interfere in SP-A's homeostatic role in surfactant release from alveolar Type II cell lamellar bodies and its subsequent uptake by Type II cells and AMs.

Lung collagen, collagen synthesis, and prolyl hydroxylase activity associated with fibrogenesis have been shown to increase in rodents with O₃ exposure of ≥ 0.45 ppm. Some studies have shown that this increase persists after exposure stops and that there is an influence of exposure pattern on the response. The increased collagen has been correlated with structural changes in the lung. Rats exposed to an urban pattern of O₃ with daily peaks of 0.25 ppm for 38 weeks displayed extracellular matrix thickening. Increased levels of collagen in CAR were demonstrated in female rats exposed to 0.5 to 1.0 ppm O₃ for 6 h/day for 20 months and in monkeys exposed to 0.61 ppm for 1 year. Both increased age and health status (e.g., emphysemic) were implicated in the increased collagen formation in response to O₃ exposure.

A time-course study (van Bree et al., 2001; Table AX5-5) evaluating the lung injury and changes in collagen content in rats exposed acutely or subchronically to 0.4 ppm O₃ demonstrated CAR thickening of septa, which progressed from 7 through 56 days of exposure.

Though collagen content decreased with PE recovery, the structural fibrotic changes in ductular septa and respiratory bronchioles persisted, suggesting that subchronic O₃ exposures in rats creates a progression of structural lung injury that can evolve to a more chronic form, which includes fibrosis. The biological relevance and adverse health effects of altered protein synthesis and collagen accumulation are uncertain.

5.2.1.6 Differential Gene Expression

Gohil et al. (2003) examined differential gene expression in C57BL/6 mice exposed to 1 ppm O₃ for three consecutive nights for 8 h (see Table AX5-6). Ozone exposure induced changes in expression of 260 genes (80% repressed and 20% induced). Differentially expressed genes included those involved in progression of the cell cycle, such as *S*-adenosyl methionine decarboxylase 3 (SAMDC3), ribonucleotide reductase (RR), and clusterin. Increased transcription of these genes suggests O₃-induced activation of the cell cycle, with subsequent cellular proliferation. This is in accord with the finding of increased epithelial proliferation with acute O₃ exposure, as discussed in studies in Sections 5.2.4.1 and 5.2.5.1. Several nuclear factor- κ B (NF- κ B)-induced genes were upregulated, including serum amyloid protein, topoisomerase II α , monocyte chemoattractant protein and platelet-derived growth factor, an inhibitor of apoptosis. Upregulation of these genes suggests to the authors that they may account for O₃-induced proliferation of nonciliated cells and Clara cells. Downregulation of transcripts for isoforms of myosins and actins were also observed, which may explain, in part, a mechanism of O₃-induced vascular permeability. Several members of the CYP family were downregulated, including 2a4, 2e1, and 2f2, as were aryl-hydrocarbon receptor and several glutathione transferases. Metallothionein 1 and 2 and lactotransferrin were upregulated, indicative of their function as antioxidants and anti-inflammatory agents. Ozone-induced suppression of immune function is suggested by downregulation of transcripts encoding major histocompatibility complex genes, lymphocyte-specific proteins, and immunoglobulins. Section 5.2.2.3 discusses the effects of O₃ exposure on the immune system.

Quinlan et al. (1994) have reviewed the regulation of antioxidant enzymes in lung after oxidant injury. A comparison of alterations in gene expression in rat following O₃ or hyperoxia exposure, both of which induce reactive oxygen species (ROS) and injury to vascular endothelial cells and cells of the alveoli, show that both ~1 ppm O₃ and 85 to 95% O₂ increase expression of

CuZnSOD, GSHPx, and catalase. Johnston et al. (1998) also found changes in gene expression indicative of inflammation and epithelial injury that occur with hyperoxia in mice (95% O₂) that are analogous to similar injury that occurs following O₃ exposure.

5.2.1.7 Summary and Conclusions—Biochemical Effects

Ozone has been shown to interact with a wide range of different cellular components including PUFAs, amino acid residues, and some low-molecular-weight compounds (GSH, urate, vitamins C and E). As O₃ does not penetrate much beyond the ELF, damage likely results from its PUFA ozonation products (mostly hydroxyhydroperoxides) involvement in signaling of cellular responses such as inflammation. New work has shown that ozonation of PUFA also forms the aldehydes nonanal, heptanal, and hexanal, the production of which are dependent on AA availability. Saturated phospholipids are thought to reduce the local dose and limit site-specific cell injury from O₃ exposure. Another ozonation product, HNE, creates protein adducts that have been linked to apoptosis and heat shock proteins in vitro.

Both short- and long-term exposures to O₃ have been shown to enhance lung xenobiotic metabolism, possibly as a result of changes in the number and function of bronchiolar epithelial Clara cells and alveolar epithelial Type II cells. This modulation is both species- and region-specific and includes the isoforms CYP 2B1, CYP 2E1. CCSP is also involved in inflammatory responses to O₃ exposure. Mouse strains with differing sensitivities to O₃ show that responses in protein, lactate dehydrogenase (LDH), and inflammatory cell influx are due to CCSP levels and changes in lung epithelial permeability.

Reactions of O₃ with AA, GSH, and UA (all antioxidants present in ELF) are a protective mechanism. But even with environmentally relevant exposures, the reactivity of O₃ is not quantitatively quenched, and cell injury occurs in both the lower and upper respiratory tract. Early work has shown that short-term exposures to <1 ppm O₃ increase antioxidant metabolism. Reexposure after a recovery period causes increases equivalent to first-time exposures, suggesting that previous exposure is not protective. Elevations in enzyme activity appear to increase as a function of age, suggesting that O₃ exposure can cause greater lung injury in the older animal. Long-term urban patterns of exposure to O₃ (daily peaks of 0.25 ppm) caused increases in GSHPx and GR, but not SOD. Recent work has suggested that endogenous antioxidants moderate the effects of O₃ and that this exposure is a subthreshold stimulus for

producing effects on peripheral Raw and reactivity, thus indicating a dissociation between airway function and inflammation.

In both acute and short-term studies, a variety of lung lipid changes occur with O₃ exposure, including an increase in AA. With longer exposures (e.g., 0.12 ppm for 90 days), an increase in PUFAs and a decrease in cholesterol-esters are seen, indicative of long-term alterations of surfactant lipid composition. Whether these changes in lipid content and/or metabolism lead to significant changes in surface tension or compliance properties of the lung remains unknown. New studies evaluating O₃-induced alterations in lipid metabolism have not been completed.

Collagen, a structural protein involved in fibrosis, increases with O₃ exposure, and some studies have shown that this increase persists after exposure stops. Urban patterns of exposure (daily peaks of 0.25 ppm O₃ for 38 weeks) created extracellular matrix thickening. Increases in centriacinar collagen were demonstrated in female rats exposed to 0.5 to 1.0 ppm O₃ for 6 h/day for 20 months and in monkeys exposed to 0.61 ppm for 1 year. New work examining the time course of lung injury and changes in collagen content in rats exposed acutely or subchronically to 0.4 ppm O₃ showed centriacinar thickening of septa. Collagen content decreased with PE recovery but not the structural fibrotic changes in ductular septa and respiratory bronchioles, which suggests that subchronic O₃ exposures in rats creates a progression of structural lung injury that can evolve to a more chronic form, including fibrosis.

5.2.2 Lung Host Defenses

Defense mechanisms, including the mucociliary clearance system, AMs, and humoral- and cell-mediated immune system, exist in the lung to protect it from infectious and neoplastic disease and inhaled particles. Summaries of key new animal studies examining the effects of O₃ on lung host defenses are presented in Table AX5-7 of Annex AX5. Acute human exposures to O₃ result in similar effects on AMs (see Chapter 6).

5.2.2.1 Clearance

Early studies of O₃ effects on the mucociliary escalator showed morphological damage to ciliated epithelial cells of the tracheobronchial tree at O₃ doses of <1 ppm. Functionally, O₃ slowed particle clearance in rats at doses of 0.8 ppm for 4 h and in rabbits at 0.6 ppm for 2 h

exposures. Acute exposures at 0.5 ppm O₃ in sheep caused increased basal secretion of glycoproteins, while longer exposures reduced tracheal glycoprotein secretions, both of which can alter the effectiveness of the mucociliary escalator. Early postnatal exposures of sheep to 1 ppm O₃ caused retardation of normal morphologic development of the tracheal epithelium, decreased epithelial mucosa density, decreased tracheal mucous velocity, and delayed development of carbohydrate composition. Conversely, alveolar clearance in rabbits after acute O₃ exposure (0.1 ppm, 2 h/day, for 1 to 4 days) is increased. Longer exposures showed no effect, and increased O₃ (1.2 ppm) slowed clearance. This pattern of altered clearance also occurs in rats. A study using rat tracheal explants exposed to O₃ for 10 min (Churg et al., 1996) showed that uptake of titanium dioxide (TiO₂) and asbestos was enhanced at 0.01 and 0.1 ppm O₃, respectively. The authors attribute the increased uptake as a direct effect of O₃, suggesting mediation by H₂O₂ or hydroxyl radical. Studies of the clearance of the radiolabeled chelate ^{99m}Tc diethylenetriamine pentaacetic acid (^{99m}Tc-DTPA) have shown that clearance is significantly increased following a 3 h exposure to 0.8 ppm O₃ in Sprague-Dawley (SD) rats (Pearson and Bhalla, 1997). Examination of regional clearance of ^{99m}Tc-DTPA in dogs following a 6 h isolated sublobar exposure to 0.4 ppm O₃ or air showed that O₃ decreased the clearance half-time by 50% at 1 day following exposure (Foster and Freed, 1999). Clearance was still elevated at 7 days PE but had recovered by 14 days. Thus, a single local exposure to O₃ increases transepithelial clearance but without any influence on contralateral segments, i.e., only for epithelia directly exposed to O₃.

Alveolar clearance is slower than tracheobronchial clearance and involves particle movement through interstitial pathways to the lymphatic system or movement of particle-laden AMs to the bottom of the mucociliary escalator. Exposures of rabbits to 0.1 ppm accelerated clearance, whereas 1.2 ppm O₃ slowed clearance. A chronic exposure has been shown to slow clearance. New evaluations of O₃ effects on alveolar clearance have not been performed.

5.2.2.2 Alveolar Macrophages

A primary function of AMs is to clear the lung of infectious and noninfectious particles by phagocytosis, detoxification, and removal. Further, AMs secrete cellular mediators that recruit and activate inflammatory cells in the lungs (see Figure 5-3). Ozone has been shown to inhibit phagocytosis at 0.1 ppm for 2 h in rabbits. This inhibition returns to control levels if exposures

are repeated for several days. The production of superoxide anion radicals and the activity of AM lysosomal enzymes (both involved in bactericidal activity) are inhibited by 3 h exposures to 0.4 and 0.25 ppm O₃ in rodents and rabbits, respectively. Production of interferon- γ (INF- γ) was decreased in rabbit AM by 1 ppm O₃ for 3 h.

New studies have shown that O₃ affects AM chemotaxis, cell adhesion, and surface expression of cell adhesion molecules (Bhalla, 1996). AM from SD rats exposed to 0.8 ppm O₃ for 3 h showed greater mobility and greater adhesion than air exposed controls. This increased mobility and adhesion were attenuated by CD16b and intercellular adhesion molecule-1 (ICAM-1) antibodies, suggesting that these adhesion molecules modulate O₃-induced inflammation. Antibodies to tumor necrosis factor- α (TNF- α) and IL-1 α also mitigated AM adherence, suggesting further that the inflammatory response to O₃ is mediated by these cytokines (Pearson and Bhalla, 1997). Cohen et al. (1996) showed that 1 ppm O₃ for 4 h reduces binding of INF- γ to AM in WEHI-3 cells and, additionally, reduces phagocytic activity, production of reactive oxygen intermediates, and elevation of intracellular Ca²⁺.

Cohen et al. (2001, 2002) exposed male Fisher 344 (F344) rats to either 0.1 or 0.3 ppm O₃ for 4 h/day, 5 days/week or either 1 or 3 weeks. In this study, superoxide anion production was increased at 1 week. Hydrogen peroxide (H₂O₂) production was reduced at both exposure concentrations and durations and was further reduced with INF- γ stimulation, suggesting that one effect of O₃ is compromised killing of bacteria by AM due to the reduction in H₂O₂ production.

Ozone treatment (2 ppm O₃, 3 h in female SD rats) caused a time-dependent increase in nitric oxide (NO) levels in both AM and type II epithelial cells that was correlated with increased expression of inducible nitric oxide synthase (iNOS) mRNA and protein (Laskin et al., 1998). Inhibition of NF- κ B, caused a dose-dependent inhibition of NO and iNOS production. Additionally, O₃ caused a time-dependent increase in NF- κ B binding activity in the nucleus of both cell types. The authors hypothesize that O₃ exposure causes the cytokines TNF- α and IL-1 β to bind to surface receptors and initiate intracellular signaling pathways in AM, leading to activation of NF- κ B, its entry into the nucleus, and its binding to the regulatory sequences of genes such as iNOS to allow their transcription. Additional studies (Laskin et al., 2002), using AM isolated from C57Bl6x129 mice with a targeted disruption of the gene for iNOS, showed no toxicity to 0.8 ppm O₃ for 3 h, as measured by BALF protein levels and nitrotyrosine staining of

the lung. Additionally, mice overexpressing human CuZnSOD and mice with a targeted disruption of p50 NF- κ B were resistant to O₃ toxicity. Wild-type mice exposed to O₃ showed an increase in expression of STAT-1, a protein that binds to the regulatory region of iNOS. Taken together, these results suggest to the authors that a number of proteins, including NF- κ B, phosphoinoside 3-kinase, and STAT-1, that bind to and regulate expression of iNOS are modulated by O₃ exposure. The same iNOS knockout mice strain exposed to 0.8 ppm O₃ for 3 h (Fakhrzadeh et al., 2002) showed no increase in AM superoxide anion and prostaglandin. These data provide further evidence that NO and its reactive oxidative product, peroxynitrite, are important in O₃-induced lung injury. Further discussions of the role of nitric oxide synthase/reactive nitrogen and cytokines/chemokines in O₃-induced inflammation are provided in Section 5.2.3.

5.2.2.3 Immune System

Other than by natural protection (e.g., opsonizing antibody, nonspecific phagocytosis by AM), the immune system defends the lung by mounting three major waves of response: natural killer (NK) cells (nonspecific lymphocytes that kill viruses, bacteria, and tumor cells), followed by cytotoxic T lymphocytes (T_{CTL}, lymphocytes that lyse specifically recognized microbial and tumor-cell targets), followed by antigen-specific antibodies. These T-cell types are involved with other immunologically active cells (e.g., B cells and AM), which in a complex manner, interact in immunological defense. To date, only a few of these mechanisms have been investigated in the context of their role in O₃ susceptibility. The effects of O₃ on the immune system are complex and depend on the exposure parameters and observation periods. T-cell-dependent functions appear to be more affected than B-cell-dependent functions. Generally, there is an early immunosuppressive effect that can, with continued exposure, either return to normal or actually enhance immunity. Changes in immune cell population occur with O₃ exposure, including T:B cell ratios in the mediastinal lymph node. Natural killer cell activity increases with 1 week exposures of 0.2 to 0.4 ppm O₃ but decreases with exposures to 0.82 ppm. Ozone exposure has also been shown to be responsible for enhancement of allergic sensitization at levels of 0.5 to 0.8 ppm for 3 days. Studies of the effects of O₃ on the immune system are summarized in Table AX5-7.

Garssen et al. (1997) have studied the effects of O₃ on non-IgE-mediated pulmonary hyper-immune reactions induced by picryl chloride (PCI). BALB/c mice sensitized with PCI, both actively and passively (by adoptive transfer of lymphoid cells from pre-sensitized mice), were then challenged with picryl sulfonic acid (PSA). The mice were exposed for 12 h to 0.2, 0.4, or 0.8 ppm O₃ during one night, at 4 days or 7 days after skin sensitization done either just before or just after PSA challenge (i.e., during the induction or effector phase). Nonsensitized mice showed no changes in tracheal reactivity to carbachol with O₃ exposure. Sensitized mice were hyperreactive to carbachol 48 h after PSA challenge, whereas sensitized mice exposed to all concentrations of O₃ showed no significant tracheal hyperreactivity to carbachol. The sensitized mice also showed a suppressed inflammatory reaction (polymorphonuclear leukocyte, PMN) with the 0.8 ppm O₃ exposure. Ozone exposure following PSA challenge also caused a suppression of tracheal hyperresponsiveness. In a separate experiment wherein mice were exposed to O₃ before sensitization and then lymphoid cells from these mice were injected into nonexposed mice, the recipients also manifested an inhibition of the induction of hyperreactivity. These results are opposite to the effect on type I (IgE-mediated) allergic reactions, which the authors suggest is due to activation of T-derived lymphocyte helper 2 (Th2) cell-dependent reactions that are possibly potentiated by O₃ or to a direct effect by O₃ on Th1 cells or other cells crucial for the tracheal hyperreactivity and inflammation seen in this mouse model.

Kleeberger et al. (2000, 2001a) have demonstrated a potential interaction between the innate and acquired immune system with O₃ exposure. Using O₃-susceptible (C57BL/6J) and O₃-resistant (C3H/HeJ) mice, they identified a candidate gene on chromosome 4, Toll-like receptor 4 (*Tlr4*). Ozone exposure (0.3 ppm for 24 to 72 h) of C3H/HeJ and C3H/HeOuJ mice, the latter differing from the O₃-resistant strain by a polymorphism in the coding region of *Tlr4*, demonstrated greater protein concentrations in the OuJ strain. The two strains exhibited differential expression of *Tlr4* mRNA with O₃ exposure. Thus, a quantitative trait locus on chromosome 4 appears to be responsible for a significant portion of the genetic variance in O₃-induced lung hyperpermeability. In these mouse strains, lavageable protein concentration was lowered by inhibition of iNOS and by targeted disruption of *Nos2*. Comparisons of C3H/HeJ and C3H/HeOuJ O₃ exposures demonstrated reduced *Nos2* and *Tlr4* mRNA levels in the O₃-resistant C3H/HeJ mice. These data are consistent with the hypothesis that O₃-induced lung hyperpermeability is mediated by iNOS. These studies suggest a role for the Toll-like

receptor 4 (TLR4) in the host response to O₃ similar to the role it has demonstrated in lipopolysaccharide (LPS) sensitivity (Schwartz 2002; Wells et al. 2003). TLR4 signaling is thought to be critical to linking the innate and acquired immune system through antigen presenting cells and Th1/Th2 differentiation.

Ozone exposure has been shown to affect antibody responses both in vitro and in mice. Becker et al. (1991) demonstrated changes in IgG production in cultured human lymphocytes with O₃ exposures of 1.0, 0.5, and 0.1 ppm for 2 h. Subsequent to O₃ exposure, cells were stimulated with pokeweed mitogen (PWM, a T-cell-dependent stimulus) or *Staphylococcus aureus* Cowan 1 strain (SAC, a T-cell-independent stimulus). Both B and T cells were affected by O₃. T cells also demonstrated an increase in IL-6 and a decrease in IL-2, which suggested to the authors that O₃ may have direct effects on IgG producing cells and concurrently an effect that is mediated by altered production of T cell immunoregulatory molecules. Responses to repeated O₃ (0.08 to 0.25 ppm) and ovalbumin (OVA, 1%) exposures were compared in “IgE-high responder” (BALB/c) and “IgE-low responder” (C57BL/6) mice (Neuhaus-Steinmetz et al., 2000). Ozone appeared to shift the immune response toward a Th2-like pattern in the two mouse strains, with differing potentials for developing allergic reactions.

Another study (Depuydt et al., 2002) demonstrated that O₃ (0.1 ppm for 2 h) increases allergen-induced airway inflammation in previously sensitized mice but has no effect on the sensitization process itself. This study used OVA-pulsed dendritic cells instead of systemic adjuvant, which the authors consider a more relevant model of sensitization, as it clearly separates the immune response from the challenge and does not obscure regulatory processes as does intraperitoneal (i.p.) injections of OVA. They further suggest that dendritic cells, the principal antigen-presenting cells in the airway, are an important component of O₃-induced eosinophilic airway inflammation.

Surfactant proteins A and D were shown to create an inflammatory feedback loop with perturbations in lung immune defenses (reviewed in Hawgood and Poulain, 2001). Earlier studies suggested that SP-A is a target for O₃ toxicity by causing inhibition of SP-A self-association and SP-A-mediated lipid vesicle aggregation. Further, O₃ reduced the ability of SP-A to inhibit phospholipid secretion by alveolar type II cells and reduced the capacity of SP-A to induce superoxide anion production and enhance phagocytosis of herpes simplex virus. Bridges et al. (2000) reported that both SP-A and SP-D directly protect surfactant phospholipids

and macrophages from oxidative damage by blocking accumulation of thiobarbituric acid reactive substances (TBARS) and conjugated dienes.

Eight human variants of SP-A in Chinese hamster ovary (CHO) cells exposed to O₃ (1 ppm for 4 h) showed decreased ability to stimulate cytokine (TNF- α and IL-8) production in THP-1 cells, a macrophage-like cell line (Wang et al., 2002). Each variant had a unique time- and dose-dependent pattern of stimulation of cytokine production with O₃ exposure, which the authors attribute to possible differences in susceptibility to O₃ oxidation. Targeted disruption of mouse SP-A and SP-D (Hawgood et al., 2002) caused increases in BAL phospholipid, macrophage, and protein through 24 weeks of age. Further, the deficient mice developed patchy lung inflammation and air space enlargement consistent with emphysema. Future experiments using these null mice will help to establish the role of SP-A and SP-D in pulmonary host defense to O₃ exposure.

5.2.2.4 Interactions with Infectious Microorganisms

Ozone-induced dysfunction of host defense systems results in enhanced susceptibility to bacterial lung infections. Acute exposures of 0.08 ppm (3 h) O₃ can overcome the ability of mice to resist infection (by decreasing lung bactericidal activity) with Streptococcal bacteria, resulting in mortality. Changes in antibacterial defenses are dependent on exposure regimens, species and strain of test animal, species of bacteria, and age of animal, with young mice being more susceptible to the effects of O₃. The effect of O₃ exposure on antibacterial host defenses appears to be concentration- and time-dependent. Early studies using the mouse “infectivity model,” consisting of exposure to clean air or O₃ followed by exposure to an aerosolized microorganism, showed that the difference in mortality between O₃-exposed groups and controls is concentration-related. Chronic exposures (weeks, months) of 0.1 ppm do not cause greater effects on infectivity than short exposures, due to defense parameters becoming reestablished with prolonged exposures.

More recent studies of O₃-induced modulation of cell-mediated immune responses showed effects on the onset and persistence of infection. Cohen et al. (2001, 2002) exposed male F344 rats subchronically to either 0.1 or 0.3 ppm O₃ for 4 h/day 15 days/week, for 1 or 3 weeks. Subsequent exposure with viable *Listeria monocytogenes* demonstrated no observed effect on cumulative mortality but did show a concentration-related effect on morbidity onset and

persistence. These data suggest that O₃ may cause a possible imbalance between Th1 and Th2 cells, which can subsequently lead to suppression of the resistance to intracellular pathogens.

Effects of O₃ on viral infections are dependent on the temporal relationship between O₃ exposure and viral infection. Only high concentrations (1.0 ppm O₃, 3 h/day, 5 days, mice) increased viral-induced mortality. No detrimental effects were seen with a 120-day exposure to 0.5 ppm O₃ on acute lung injury from influenza virus administered immediately before O₃ exposure started. But there were O₃-enhanced postinfluenzal alveolitis and lung parenchymal changes. As O₃ does not affect lung influenza viral titers, it apparently does not impact antiviral clearance mechanisms. In general, the evidence suggests that O₃ can enhance both bacterial and viral lung infections, but the key mechanisms have not yet been identified. New studies on the interactions of O₃ and viral infections have not been published.

5.2.2.5 Summary and Conclusions—Lung Host Defenses

New data on lung host defenses support earlier work which suggests that mucociliary clearance is affected in most test species at just under 1 ppm, with lower levels (~0.1 ppm) increasing clearance and somewhat higher levels decreasing clearance. These data also tend to suggest mechanisms whereby O₃ affects clearance, which include uptake being a direct effect of O₃, but modulated by ROS and hydroxyl radicals.

Alveolar macrophage function is disrupted by O₃, as shown by several studies showing inhibition of phagocytosis at concentrations ranging from 0.1 to 1.2 ppm. This inhibition returns to control levels if O₃ exposures are repeated for several days. Two new studies corroborate earlier findings of increases in AM number in that same exposure range. In this environmentally relevant exposure range, new studies support older findings of decreased resistance to microbial pathogens, as shown by those endpoints examining superoxide radical formation, altered chemotaxis/motility, decreased INF- γ levels, decreased lysosomal activity, increased prostaglandin E (PGE) levels, and increased NO mRNA and protein.

New research evaluating the effects of O₃ on immune function advances previous work that has shown that exposures can enhance or suppress immune responsiveness, depending on the species studied, concentration of O₃, route of exposure of allergen, and timing of exposure. Continuous exposure to O₃ impairs immune responses for the first several days of exposure, followed by an adaptation to O₃ that allows a return of normal immune responses. Most species

show little effect of O₃ exposures prior to immunization, but show a suppression of responses to antigen in O₃ exposures post-immunization. Evaluation of mouse strains with genetically-determined differential sensitivity or resistance to O₃ indicated a possible interaction between the innate and acquired immune systems and, further, that O₃ may shift the immune response toward a Th2-like pattern. Work has also focused on the deleterious effects of O₃ exposure on SP-A and SP-D and their immunomodulatory function in protecting against oxidative stress.

Findings from several new studies evaluating the effects of O₃ exposures on infectious microorganisms are in concurrence with those from previous studies which showed, in general, increased mortality and morbidity, decreased clearance, increased bacterial growth, and increased severity of infection at exposure levels of 0.1 to 1 ppm O₃ for 1 week.

5.2.3 Inflammation and Lung Permeability Changes

The normal lung has an effective barrier function that controls bidirectional flow of fluids and cells between the air and blood compartments. Ozone disrupts this function, resulting in two well-characterized effects of O₃ exposure, lung inflammation and increased permeability, which are distinct events controlled by independent mechanisms. Ozone initiates inflammation of lung tissue by reactions with antioxidants and lipids in ELF (discussed in Section 5.2.1, see Figure 5-2). Secondary reaction products generated in this process then cause changes in cell membranes, disruption of the lung barrier leading to leakage of serum proteins, influx of PMNs, release of bioactive mediators, and movement of compounds from the air spaces into the blood. This increased permeability allows accumulation of co-occurring pollutants into the lung tissue. The framework for presenting this stereotypical response to O₃ consists of discussions covering: (1) the time course of these changes, (2) concentration × time (C × T) relationships, (3) susceptibility factors, (4) mediators of inflammation, and (5) NO and reactive nitrogen.

Rats appear to be more resistant to O₃-induced inflammation than humans (see Chapter 4). With comparable exposure protocols, both species have similar observed inflammatory and permeability changes, i.e., controlled human exposure studies discussed in Chapter 6 indicate that the majority of acute responses in humans are similar to those observed in animals.

Ozone also increases the permeability from the air to the blood compartment. Ozone (0.8 ppm for 2 h) caused a 2-fold increase of the transport of labeled DTPA from the rat tracheal lumen to the blood. This coincided with a 2-fold increase in the number of endocytic vesicles in

epithelial cells that contained intraluminally instilled horseradish peroxidase (HRP) as a tracer. These studies also suggest an uneven disruption of tight junctions and alternative transport via endocytotic mechanisms. In studies aimed at detecting the effects of O₃ exposure on regional permeability, O₃ increased the transmucosal transport of diethylenetriamine pentaacetic acid (DTPA) and bovine serum albumin (BSA) more in the trachea and bronchoalveolar zone than in the nose. These changes in barrier integrity may allow increased entry of antigens and other bioactive compounds (e.g., bronchoconstrictors) into lung tissue. Data from analyses at regular intervals PE indicate that maximal increases in BALF protein, albumin and number of PMNs occur 8 to 18 h (depending on the study) after an acute exposure to 0.5 to 1.0 ppm O₃ ceases.

Increases in permeability and inflammation have been observed at levels as low as 0.1 ppm O₃ for 2 h/day for 6 days in rabbit and 0.12 ppm in mice (24 h exposure) and rats (6 h exposure). After acute exposures, the influence of the time of exposure increases as the concentration of O₃ increases. The exact role of inflammation in the etiology of lung disease is not known; nor is the relationship between inflammation and changes in lung function. Table AX5-8 in Annex AX5 summarizes new key studies describing the potential for O₃ exposure effects on lung permeability and inflammation.

5.2.3.1 Time Course of Inflammation and Lung Permeability Changes

A study of OVA-sensitized male Dunkin-Hartley guinea pigs exposed to 1.0 ppm O₃ for 3 h showed that levels of PMN significantly increased at 3 h PE, but BAL protein levels did not, suggesting a lack of correlation between the two endpoints (Sun et al., 1997). Increased PMN without a concordant increase in BAL protein levels were found when the guinea pigs were exposed to 1.0 ppm O₃ for 1 h and evaluated 24 h PE. The first group also had an increase in AHR, but not the second group, which suggests a dissociation between PMN levels and AHR.

Earlier work demonstrated that O₃ exposures of 0.8 to 1 ppm in rat and guinea pig transiently increase the permeability from the air to the blood compartment. This permeability is greatest in tracheal and bronchoalveolar zones, and may allow increased entry of antigens and other bioactive compounds (e.g., bronchoconstrictors) into lung tissues. The time course of the influx of PMNs into the lung and the BALF fluid levels of macrophage inflammatory protein-2 (MIP-2) were found to be roughly similar to that for proteins (Bhalla and Gupta, 2000). Adherence of neutrophils to pulmonary vascular endothelium is maximal within 2 h after

exposure and returns to control levels by 12 h PE (Lavnikova et al., 1998). Cheek et al. (1995) cultured monolayers of rat alveolar type II cells and exposed them to 0.1 or 0.5 ppm O₃ for 0.5 h to evaluate the effects of O₃ on permeability. Permeability increased dose-dependently and the higher exposures elicited greater numbers of injured epithelial cells. Exposure to 0.1 ppm O₃ was thought to expedite the restoration of epithelial barrier functions, while at higher exposures, neutrophils exacerbated the O₃-induced injury. Vesely et al. (1999a) have demonstrated that neutrophils contribute to the repair process in O₃-injured airway epithelium and they may play a role in removal of O₃-injured cells.

Exposures of 3 to 7 days have been found to cause increases in BALF protein and PMNs that typically peak after a few days (depending upon species tested and exposures) and return toward control even with continuing exposure. Van Bree et al. (2002) observed lower BALF levels of protein, fibronectin, IL-6, and inflammatory cells in rats exposed for 5 days to 0.4 ppm O₃ than in rats exposed for 1 day, suggesting adaptation to O₃ exposure. Postexposure challenge with single O₃ exposures at different time points showed recovery of susceptibility to O₃. McKinney et al. (1998) observed differences in IL-6 levels due to repetitive exposures and demonstrated a role of IL-6 in the adaptive response induced by repeated O₃ exposures of 0.5 ppm for 4 h.

5.2.3.2 Concentration and Time of Exposure

The relative influence of concentration and duration of exposure (i.e., C × T) has been investigated extensively in rats, using BALF protein as an endpoint. Earlier work utilizing concentrations of 0.1 to 2 ppm O₃ and durations of 1 to 8 h has shown that the interaction between C and T is complex. At these levels of exposure, concentration generally dominates the response. The smallest C × T product causing an increase in protein was 0.52 ppm • h (0.13 ppm × 4 h). C × T studies using the endpoints of changes in lung protein or cell type showed that acute damage is a function of cumulative dose. The impact of T is C-dependent (at higher Cs, the impact of T is greater); at the lowest C and T values, this dependence appears to be lost. The controlled human exposure data described in Chapter 6 concur with most animal data, showing that (a) concentration of O₃ is an important factor determining O₃ responses and (b) duration of exposure and ventilation rate also exert substantial influences.

New studies evaluating C × T relationships in animal models have not been completed. However, a full understanding of C × T relationships in ambient exposures must include the recognition that ‘real world’ exposures are cyclic in nature, due to the daily and seasonal variations in O₃ levels. The concentration of O₃, the duration of the exposure, and duration between exposures are all relevant to determining the type and level of O₃-induced injury.

5.2.3.3 Susceptibility Factors

Factors that have been studied for potential impact on the effects of O₃ exposure include age, gender, nutritional status, exposure to copollutants, exercise, and genetic variability. A full characterization of the effects of age on O₃ responses has not been completed. Data available indicate that effects of age on O₃ responses are endpoint-dependent, with young mice, rats, and rabbits having greater prostaglandin levels with exposure and senescent rats having greater IL-6 and N-acetyl-β-D-glucosaminidase levels with exposure.

A study (Johnston et al., 2000a) compared gene expression of chemokines and cytokine in newborn and 8-week-old C57BL/6J mice exposed to 1.0 or 2.5 ppm for 4, 20, or 24 h. The newborn mice displayed increased levels of MT mRNA only, while the 8-week-old mice had increases in MIP-1α, MIP-2, IL-6, and MT mRNA. Comparisons were made with mice of the same age groups with exposures to endotoxin (10 ng/mouse for 10 min). Both age groups displayed similar cytokine/chemokine profiles with endotoxin exposure. This suggested to the authors that the responses to endotoxin (which does not cause epithelial injury) and the responses to O₃ (which does) demonstrate that differences in inflammatory control between newborn and adult mice is secondary to epithelial injury.

Pregnancy and lactation increased the susceptibility of rats to acute O₃, but no clear effects of gender have been identified. The effects of vitamin C deficiency on O₃ responses are unclear. Ascorbate-deficient guinea pigs exposed to O₃ demonstrated only minimal effects on injury and inflammation (Kodavanti et al., 1995). Utilizing a diet-restricted (20% of the freely fed diet) rat model, Elsayed (2001) demonstrated higher survivability upon exposure to higher O₃ (0.8 ppm continuously for 3 days) compared to freely fed rats. Preexposure to sidestream cigarette smoke had been found to cause increased lung injury (Yu et al., 2002). In vitro studies on the macrophages from smoke + O₃-exposed animals responded by a greater release of TNF-α

following LPS stimulation when compared to macrophages exposed to air, smoke, or O₃ (0.5 ppm for 24 h) alone.

Lines of evidence illustrate that genetic background is an extremely important determinant of susceptibility to O₃. Earlier studies using inflammation-prone (susceptible) C57BL/6J (B6) and inflammation-resistant C3H/HeJ (C3) mouse strains and high doses of O₃ (2 ppm for 3 h) identified *Inf-2* as a locus controlling susceptibility. Further studies in these two strains of mice using more relevant exposures (0.3 ppm for 72 h) identified that the acute and subacute exposures are controlled by two distinct genes, referred to as *Inf-1* and *Inf-2*, respectively (Tankersley and Kleeberger, 1994). Exposures to 0.3 ppm O₃ for 48 or 72 h, when repeated 14 days after the initial exposures, caused a smaller increase in BALF protein and numbers of macrophages, lymphocytes, and epithelial cells in both strains, but PMN number was greater in both strains compared to initial exposure (Paquette et al., 1994). Kleeberger et al. (1997) also identified another potential susceptibility gene, tumor necrosis factor (*Tnf*, which codes for the pro-inflammatory cytokine TNF- α) on a qualitative trait locus on mouse chromosome 17. By neutralizing the function of TNF- α with a specific antibody, they were able to confer protection against O₃ (0.3 ppm, 48 h) injury in susceptible mice. They then demonstrated a role for TNF receptor 1 and 2 (TNFR1 and TNFR2, respectively) signaling in subacute (0.3 ppm for 48 h) O₃-induced pulmonary epithelial injury and inflammation (Cho et al., 2001). TNFR1 and TNFR2 knockouts were less sensitive to subacute O₃ exposure than were WT C57BL/6J mice.

An integrated and more comprehensive effort to identify the genetic basis for the susceptibility to O₃-induced lung injury was reported by Savov et al. (2004). In this report, acute lung injury in response to high O₃ dose (2 ppm for 3 h) was assessed and integrated with physiological, biochemical, and genetic observations using nine inbred mouse strains. This work indicated the presence of genetic loci on chromosomes 1, 7, and 15 associated with phenotypic characteristics for resistance to acute O₃-induced lung injury. They identified C3H/HeJ and A/J as consistently O₃-resistant; C57BL/6J and 129/SvIm as consistently O₃-vulnerable; and CAST/Ei, BTBR, DBA/2J, FVB/NJ, and BALB/cJ as intermediate in response to O₃.

Ozone-induced changes in CCSP (called CC16 by this group) expression were evaluated in five inbred mouse strains: C57BL/6J and CBA both considered sensitive to acute O₃-induced inflammation, C3H/HeJ and AKR/J both considered resistant, and SJL/J considered intermediate (Broeckeaert et al., 2003). Two exposure protocols, 1.8 ppm O₃ for 3 h or 0.11 ppm O₃, 24 h/day

for up to 3 days, were used; and BALF and serum were assayed immediately after exposure or at 6 h PE. Both exposure levels caused a transient increase in CC16 in serum that correlated with BALF changes in protein, LDH, and inflammatory cells. There was an inverse relationship between preexposure levels of CC16 in BALF and epithelial damage, based on serum CC16 levels and BALF markers of inflammation. There was also an inverse relationship between preexposure levels of albumin in BALF and lung epithelium damage. Based on these results, the authors concluded that a major determinant of susceptibility to O₃ is due to differences in the basal permeability of the airway epithelium. As all of the mouse strains had similar levels of preexposure CC16 mRNA, they explored the possible role of CC16 isozymes in differences among strains. The CC16 monomer, a 7kD protein, exists in two isoforms with differing *pI* values, CC16a (4.9) and CC16b (5.2). To evaluate the role of CC16 isoform profiles in permeability differences between C57BL/6J and C3H/HeJ, this group evaluated the CC16 protein profiles in BALF of the strains before and after O₃ exposure following two-dimensional protein electrophoresis analysis. C57BL/6J mice had lower levels of CC16a (the more acidic form) than C3H/HeJ. But both the strains had similar levels of CC16b. Based on these observations, Broeckeaert et al. (2003) concluded that greater epithelial permeability observed in C57BL/6J mice may be due to differences in the expression of CC16a and, possibly, other antioxidant/anti-inflammatory proteins.

Wattiez et al. (2003) examined BALF protein from C57BL/6J (O₃-sensitive) and C3H/HeJ (O₃-resistant) mice exposed to filtered air, using a two-dimensional polyacrylamide gel approach to analyze the protein profiles. C3H/HeJ mice expressed 1.3 times more Clara cell protein16 (CC16) than C57BL/6J mice and, further, expressed more of the acidic isoform of CC16. Strain-specific differential expression of isoforms of the antioxidant protein 2 (AOP2), the isoelectric point 5.7 isoform in C3H/HeJ and isoelectric point 6.0 isoform in C57BL/6J, were observed. These data suggest a protective role for CCSP against oxidative damage and, further, that genetic susceptibility to oxidant stress may be moderated, in part, by the gene coding for CCSP. Taken together, these mouse studies of genetic susceptibility are useful for helping to understand underlying mechanisms potentially leading to O₃-induced effects. However, at this point, corresponding human polymorphisms have not yet been identified as being associated with differing human sensitivities to O₃.

5.2.3.4 Mediators of Inflammatory Response and Injury

Ozone reacts with lipids in the ELF or epithelial cell membranes, creating ozonation products which then stimulate airway epithelial cells, AMs, and PMNs to release a host of pro-inflammatory mediators, including cytokines, chemokines, ROS, eicosanoids, and PAF (see Figure 5-3). While neutrophils in the lung characterize an inflammatory response to O₃, the release of chemotactic mediators by inflammatory cells indicates their state of activation and their role in continued inflammation and injury. At O₃ exposures of ≥1 ppm in rodents, these mediators recruit PMNs and increase expression of MIP-2 mRNA or BALF levels of MIP-2 (Driscoll et al., 1993; Haddad et al., 1995; Bhalla and Gupta, 2000). The increased mRNA expression was associated with an increased neutrophilia in the lung. Harkema et al. (1987) reported neutrophilic rhinitis in monkeys exposed to 0.15 ppm O₃ for 6 days. Zhao et al. (1998) showed that 0.6 ppm O₃ exposure for 2 h in mice and rats causes an increase in monocyte chemotactic protein-1 (MCP-1).

Fibronectin, an extracellular matrix glycoprotein, is thought to have a role in lung inflammation and inflammatory disorders, and has been shown to be increased with exposure to 1 ppm O₃ for 14 days. Gupta et al. (1998) observed an increase in both fibronectin protein and mRNA expression in the lung of rats exposed to 0.8 ppm O₃ for 3 h. A mechanistic role of fibronectin in O₃-induced inflammation and injury was suggested on the basis of comparability of temporal changes in BALF protein, fibronectin, and alkaline phosphatase activity with exposures of 1 ppm for 3 h (Bhalla et al., 1999). Studies have reported an effect of O₃ on other cytokines and inflammatory mediators. An increase occurred for cytokine-induced neutrophil chemoattractant (CINC) and NF-κB expression in vivo (Koto et al., 1997); for IL-8 in vivo and in vitro (Chang et al., 1998); TNF-α, fibronectin, IL-1 and CINC release by macrophages ex vivo (Pendino et al., 1994; Ishii et al., 1997); and NF-κB and TNF-α (Nichols et al., 2001; see Section 6.9.2 of Chapter 6). An increase in lung CINC mRNA occurred within 2 h after the end of a 3 h exposure of rats to 1 ppm O₃. The CINC mRNA expression was associated with neutrophilia at 24 h PE. Exposure of guinea pig AMs recovered in BALF and exposed in vitro to 0.4 ppm O₃ for 1 h produced a significant increase in IL-6 and TNF-α (Arsalane et al., 1995). An exposure of human AMs to an identical O₃ concentration increased TNF-α, IL-1β, IL-6, and IL-8 and their mRNAs. Ozone exposure (0.3 to 2.5 ppm, 1 to 48 h) of mice caused an increase in IL-6, MIP-1α, MIP-2, eotaxin, and MT abundance (Johnston et al., 1999a). The IL-6 and MT

increase was enhanced in mice deficient in CCSP, suggesting a protective role of Clara cells and their secretions (Mango et al., 1998). CCSP deficiency, also increased sensitivity of mice to O₃, as determined by an increase in abundance of MIP-1 α and MIP-2 following a 4 h exposure to 1.0 ppm O₃ (Johnston et al., 1999b).

Mast cells, which are located below the epithelium, release pro-inflammatory mediators and have been shown to contribute to O₃-induced epithelial damage. Greater increases in lavageable macrophages, epithelial cells and PMNs were observed in mast cell-sufficient mice than in mast cell-deficient mice exposed to 0.26 ppm O₃ for 8 h per day, 5 days per week (Kleeberger et al., 2001b). Increases in inflammatory cells were also observed in mast cell-deficient mice, however, O₃-induced permeability changes were similar between genotypic groups exposed to 0.26 ppm. When an RBL-2H3 mast cell line was exposed to 0.1 to 1.0 ppm O₃ for 1 h, spontaneous release of serotonin and modest generation of prostaglandin D₂ (PGD₂) occurred only under conditions that caused cytotoxicity (Peden and Dailey, 1995). Additionally, O₃ inhibited IgE- and A23187-induced degranulation. Mast cells recovered from O₃-exposed peripheral airways of ascaris-sensitive dogs released significantly less histamine and PGD₂ following in vitro challenge with ascaris antigen or calcium ionophore (Spannhake, 1996). Ozone (0.4 ppm, 5 weeks) exposure also promoted eosinophil recruitment in the nose and airways in response to instillation of OVA or OVA-pulsed dendritic cells and aggravated allergy like symptoms in guinea pigs (Iijima et al., 2001).

The role of PMNs and cellular mediators in lung injury and epithelial permeability has been investigated using antibodies and inhibitors of known specificity to block inflammatory cell functions and cytokine activity. Treatment of rats with cyclophosphamide prior to O₃ exposure (0.8 ppm, 48 h) resulted in a decreased recovery of PMNs in the BALF and attenuated permeability induced by O₃ (Bassett et al., 2001).

Pretreatment of animals with antiserum against rat neutrophils abrogated PMN accumulation in the lung, but did not alter permeability increase produced by O₃. Studies utilizing antibodies to selected pro- or anti-inflammatory cytokines suggest a role of TNF- α , IL-10, and IL-1 β in O₃-induced changes in permeability, inflammation and cytokine release (Ishii et al., 1997; Reinhart et al., 1999; Bhalla et al., 2002) in exposures of ~1 ppm for 3 to 6 h. An attenuation of O₃-induced increases in permeability and inflammation was also observed in mice treated either before or after exposure with UK-74505, a PAF receptor antagonist

(Longphre et al., 1999). These results were interpreted to indicate that O₃-induced epithelial and inflammatory changes are mediated in part by activation of PAF receptors.

Ozone exposure stimulates macrophage motility toward a chemotactic gradient, and macrophages isolated from rats exposed to 0.8 ppm O₃ for 3 h adhered to epithelial cells (ARL-14) in culture to a greater extent than macrophages from air-exposed controls (Bhalla, 1996). Both macrophage motility and chemotaxis were attenuated by antibodies to cell adhesion molecules CD-11b and ICAM-1, suggesting a role for cell adhesion molecules in O₃-induced cellular interactions. This may also explain the increased tissue localization and reduced recovery of macrophages in BALF (Pearson and Bhalla, 1997) following O₃ exposure (0.8 ppm, 3 h). Studies investigating the mechanisms of PMN recruitment in the lung have explored the role of cell adhesion molecules that mediate PMN-endothelial interactions. An exposure of female rats to O₃ (1 ppm, 2 h) had an attenuating effect on CD-18 expression on AMs and vascular PMNs, but the expression of CD62L, a member of selection family, on vascular PMNs was not affected (Hoffer et al., 1999). In monkeys, O₃-induced (0.8 ppm, 8 h) inflammation was blocked by treatment with a monoclonal antibody to CD18, suggesting dependence of PMN recruitment on this adhesion molecule (Hyde et al., 1999). Treatment of monkeys with CD18 antibody also reduced tracheal expression of the β6 integrin (Miller et al., 2001), suggesting that lung epithelial cell expression of this adhesion molecule is associated with sites of neutrophil recruitment. A single 3 h exposure of rats to 1 ppm O₃ caused an elevation in concentration of ICAM-1, but not CD-18, in the BALF (Bhalla and Gupta, 2000). Takahashi et al. (1995a) found an increase in tissue expression of ICAM-1 in mice exposed to 2 ppm O₃ for 3 h, noting a temporal correlation of inflammatory activity and ICAM-1 expression, which varied in different regions of the lung. A comparable pattern of time-related changes in total protein, fibronectin and alkaline phosphatase activity in the BALF of rats exposed to 0.8 ppm O₃ for 3 h was also noted by Bhalla et al. (1999). Together, these studies support the role of extracellular matrix protein and cell adhesion molecules in the induction of lung inflammation and injury.

5.2.3.5 The Role of Nitric Oxide Synthase and Reactive Nitrogen in Inflammation

Nitric oxide is a messenger molecule involved in many biological processes, including inflammation (see Figure 5-3). Cells in the respiratory tract (including mast cells, neutrophils, epithelial cells, neurons, and macrophages) produce three different forms of nitric oxide synthase

(NOS), the enzyme that catalyzes the formation of NO. NOS-1 (neuronal) and NOS-3 (endothelial) are constitutively expressed, whereas NOS-2 (also referred to as iNOS) is inducible, commonly by pro-inflammatory cytokines. Macrophages isolated from O₃-exposed (0.8 ppm for 3 h) mice produced increased amounts of NO, superoxide anion, and PGE₂, but production of these mediators by macrophages from NOS knockout mice was not elevated (Fakhrzadeh et al., 2002). Additionally, mice deficient in NOS or mice treated with N^G-monomethyl-L-arginine, an inhibitor of total NOS, were protected from O₃-induced permeability, inflammation, and injury, suggesting a role for NO in the production of O₃ effects (Kleeberger et al., 2001a; Fakhrzadeh et al., 2002). These results contrast with a study showing that O₃ exposure (of 1 ppm for 8 h/night for 3 nights) produced greater injury, as determined by measurement of MIP-2, matrix metalloproteinases, total protein, cell content, and tyrosine nitration of whole lung protein, in iNOS knockout mice than in wild-type mice (Kenyon et al., 2002). This group suggests that protein nitration is related to inflammation and is not dependent on iNOS-derived NO. They point out the possible experimental differences, such as O₃ concentration, for inconsistency between their results and those of Kleeberger et al. (2001a).

Rats that were pretreated with ebselen (a potent anti-inflammatory, immunomodulator, and NO/peroxynitrite scavenger) and then were exposed to 2 ppm O₃ for 4 h had decreased numbers of neutrophils, lowered albumin levels, and inhibited nitration of tyrosine residues in BALF 18 h PE, though macrophage iNOS expression was not changed (Ishii et al., 2000a). These results suggest an iNOS-independent mechanism for O₃-induced inflammation. Jang et al. (2002) showed dose-dependent increases in nitrate (indicative of in vivo NO generation) with O₃ exposure (0.12, 0.5, 1, or 2 ppm for 3 h). Functional studies of enhanced pause (P_{enh}) demonstrated increases with O₃ exposure which were also dose-dependent. Western blot analysis of lung tissue showed increases in NOS-1 but not in NOS -3 or iNOS isoforms. These results suggest that, in mice, NOS-1 may induce airway responsiveness by a neutrophilic airway inflammation. The literature regarding the effects of O₃ exposure on NOS activity is complex and conflicting. Similarly, the issue of protein nitration as it relates to cell injury due to O₃ exposure is somewhat controversial.

5.2.3.6 Summary and Conclusions—Inflammation and Permeability Changes

Figure 5-3 depicts many of the inflammatory and permeability changes that occur with O₃ exposure. The figure also demonstrates links between inflammatory/permeability responses and altered lung function (discussed in Section 5.2.5), ciliary motility (Section 5.2.2.1), AHR (Section 5.2.5.3), and possible thrombolytic effects (Section 5.3.3). Airway mucosa in the normal lung serves as an effective barrier that controls bidirectional flow of fluids and cells between the air and blood compartments. Ozone disrupts this function, resulting in a cascade of effects that includes an increase in serum proteins, bioactive mediators, and PMNs in the interstitium and air spaces of the lung. Damaged epithelial cells release cytokines, which function to recruit and activate AMs and PMNs. PMN recruitment into the lung is maximal at several hours PE. PMN recruitment is followed by recruitment of blood monocytes that enter the lung and enlarge to become AMs. The AMs persist for days to weeks, phagocytizing injured cells. Activated PMNs and AMs continue the cascade of effects by further releasing inflammatory mediators, which serve to amplify the initial effects of O₃. Generally, the initiation of inflammation is an important component of the defense process; however, its persistence and/or repeated occurrence can result in adverse health effects. Activation of this inflammatory cascade takes several hours. Chemical mediators released early in the cascade contribute to effects on pulmonary function. Events occurring later in the cascade, by which time O₃-induced alterations in pulmonary function have attenuated, are related to sustained inflammation. Further, mechanistic separation of inflammation, permeability, and AHR is suggested by the temporal disparities of their increases. The O₃-induced disruption of the tight junctions between epithelial cells also increases the permeability between the air and blood compartments. This disruption, occurring with exposures of 0.8 ppm for 2 h, is greater in the trachea and bronchoalveolar zone than in the nose and allows entry of particles, including bioactive compounds, into lung tissue.

For environmentally-relevant exposures to O₃, exposure concentration dominates the response. Studies evaluating C × T relationships have not been published recently. Other factors that have been studied for potential impact on the effects of O₃ include age, gender, nutritional status, genetic variability, exercise and exposure to copollutants. The effects of age on lung inflammation are not well known. After an acute exposure to 0.8 or 1 ppm, young mice, rats, and rabbits had greater changes in prostaglandins in BALF, but there were no

age-dependent effects on BALF protein or cell number. Comparisons of male and female animals, and ascorbate deficiency did not reveal significant differences in the effects of O₃, but exercise during exposure increased susceptibility.

Important new work has revealed that susceptibility to O₃ is, in part, genetically determined. Mouse strains with differing sensitivities to O₃ have identified genes on separate loci controlling various aspects of inflammation, providing additional evidence for the mechanistic separation of responses to O₃. Such research is summarized in Figure 5-4. Kleeberger's group has identified *Inf-1*, which modulates acute inflammatory responses; *Inf-2*, which modulates responses to subacute exposures; and TNF- α and TNF receptors, which are involved in inflammatory responses. Other research groups have identified loci linked to other endpoints. This line of research provides a groundwork for understanding the underlying mechanisms of O₃-induced injury and can shed light on genes responsible for human susceptibility to O₃.

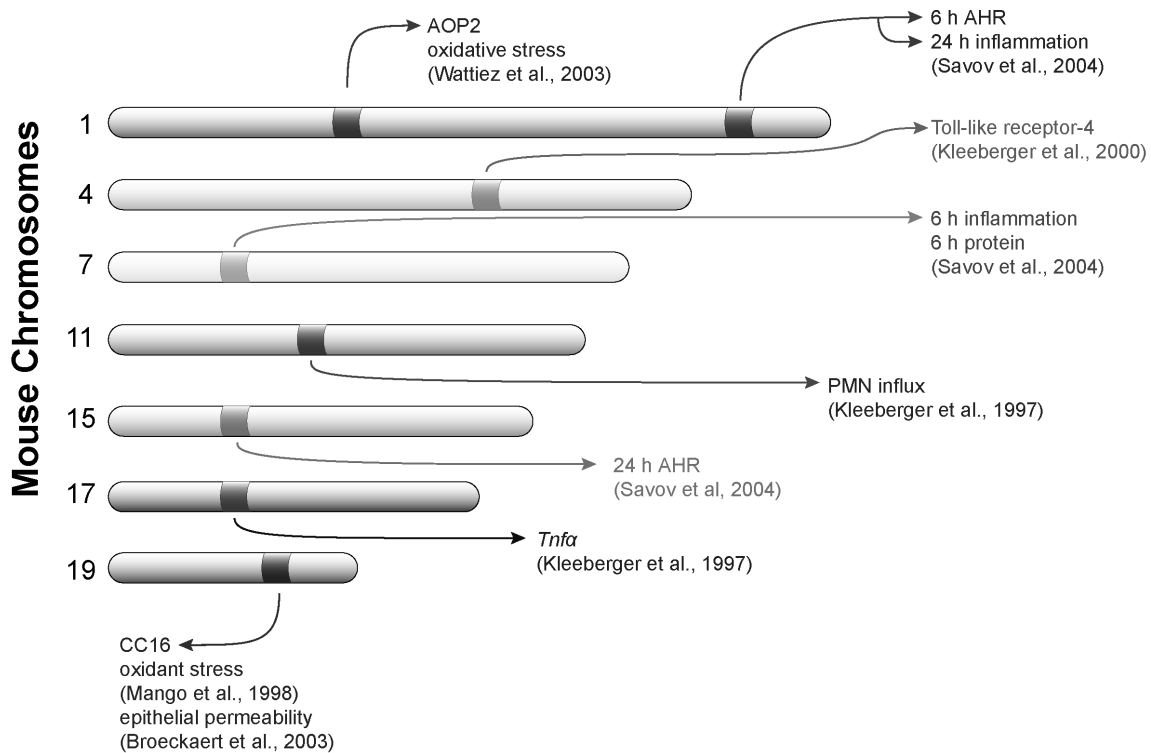


Figure 5-4. Mouse chromosomes on which genes or gene loci have been identified that modulate responses to ozone.

Recent studies have placed a major focus on mediators released from inflammatory cells to understand the mechanisms of O₃-induced inflammation and injury. Cytokines and chemokines have been shown to be released as a result of stimulation or injury of macrophages, epithelial cells, and PMNs. Exposure of guinea pig AMs recovered in BALF and exposed in vitro to 0.4 ppm O₃ produced a significant increase in IL-6 and TNF- α . An exposure of human AMs to an identical O₃ concentration increased TNF- α , IL-1 β , IL-6, and IL-8. The expression of MIP-2 mRNA or BALF levels of MIP-2 increased in mice and rats exposed to O₃ concentrations \geq 1 ppm. An increase after O₃ exposure has also been reported for other cytokines and inflammatory mediators, including CINC and fibronectin. The CINC mRNA expression was associated with neutrophilia at 24 h PE. Ozone exposure of mice also caused an increase in IL-6, MIP-1 α and eotaxin in mice. Further understanding of the role of mediators has come from studies utilizing antibodies and inhibitors of known specificity. In these studies, treatment of rats with an anti-IL-6-receptor antibody prior to a nighttime exposure to O₃ abolished O₃-induced cellular adaptive response following a subsequent exposure. Studies utilizing antibodies to selected pro- or anti-inflammatory cytokines suggest a role for TNF- α , IL-10 and IL-1 β in O₃-induced changes in permeability, inflammation, and cytokine release.

Studies investigating the mechanisms of PMN recruitment in the lung have explored the role of cell adhesion molecules that mediate PMN-endothelial cell interactions. An increase in tissue expression of ICAM-1 occurred in mice exposed to 0.8 ppm O₃. A comparable pattern of time-related changes in total protein, fibronectin, and alkaline phosphatase activity in the BALF was observed in rats exposed to 1 ppm O₃. In monkeys, the O₃-induced inflammation was blocked by treatment with a monoclonal antibody to CD18, suggesting dependence of PMN recruitment on this adhesion molecule. Together, these studies support the role of extracellular matrix protein and cell adhesion molecules in lung inflammation and injury.

Ozone exposure also affects macrophage functions and, consequently, their role in lung inflammation. Macrophages isolated from O₃-exposed mice produced increased amounts of NO, superoxide anion, and PGE₂, but production of these mediators by macrophages from NOS knockout mice was not elevated. Additionally, mice deficient in NOS or mice treated with N^G-monomethyl-L-arginine, an inhibitor of total NOS, were protected from O₃-induced permeability, inflammation, and injury. These findings suggest a role for NO in the production of O₃ effects.

5.2.4 Morphological Effects

Most mammalian species show generally similar morphological responses to <1 ppm O₃ that differ only by region, cell type, exposure parameters, and length of time between exposure and examination. Constant low exposures to O₃ create an early bronchoalveolar exudation that declines with continued exposure and drops sharply in the PE period. Epithelial hyperplasia also starts early, increases in magnitude for several weeks, plateaus with continuing exposure, and declines slowly during PE. Interstitial fibrosis has a later onset, continues to increase throughout the exposure, and can continue to increase after the exposure ends. A schematic comparison of these duration-response profiles is presented in Figure 5-5 (Dungworth, 1989). Nonhuman primates respond more than rats at this concentration, due to differences in antioxidants, the CAR (predicted to receive the highest dose of O₃), the presence of respiratory bronchioles, acinar volume, and differences in the nasal cavity's ability to "scrub" the O₃. Ciliated epithelial cells of the airway, Type I epithelial cells of the gas-exchange region, and ciliated cells in the nasal cavity are the cells most affected by O₃. Ozone-damaged ciliated cells are replaced by nonciliated cells, which are unable to provide clearance function, and Type I cells are replaced by Type II cells, which are thicker and produce more lipids. Inflammation also occurs, especially in the CAR, wherein the tissue is thickened as collagen accumulates. At exposures of 0.25 ppm O₃ (8 h/day, 18 months) in monkeys, the distal airway was found to be remodeled, as bronchiolar epithelium replaces the cells present in alveolar ducts. In both rodents and monkeys, it appears that the natural seasonal patterns of O₃ exposure alter morphology more than continuous exposures; thus, long-term animal studies with uninterrupted exposures may underestimate morphological effects.

5.2.4.1 Acute and Subchronic Exposure Effects

Morphological effects of key acute and subchronic exposure studies are summarized in Table AX5-9. Harkema et al. (1997a) reviewed toxicological studies of the nasal epithelial response to short-term O₃. New information regarding the effects of O₃ in this region include results indicating that the topical anti-inflammatory corticosteroid fluticasone propionate prevents inflammation and mucous cell metaplasia in rats after cumulative O₃ exposure (0.5 ppm O₃, 8 h/day, for 3 or 5 days) (Hotchkiss et al., 1998). Exposure to bacterial endotoxin, a common ambient air toxicant, can potentiate mucous cell metaplasia in the nasal transitional

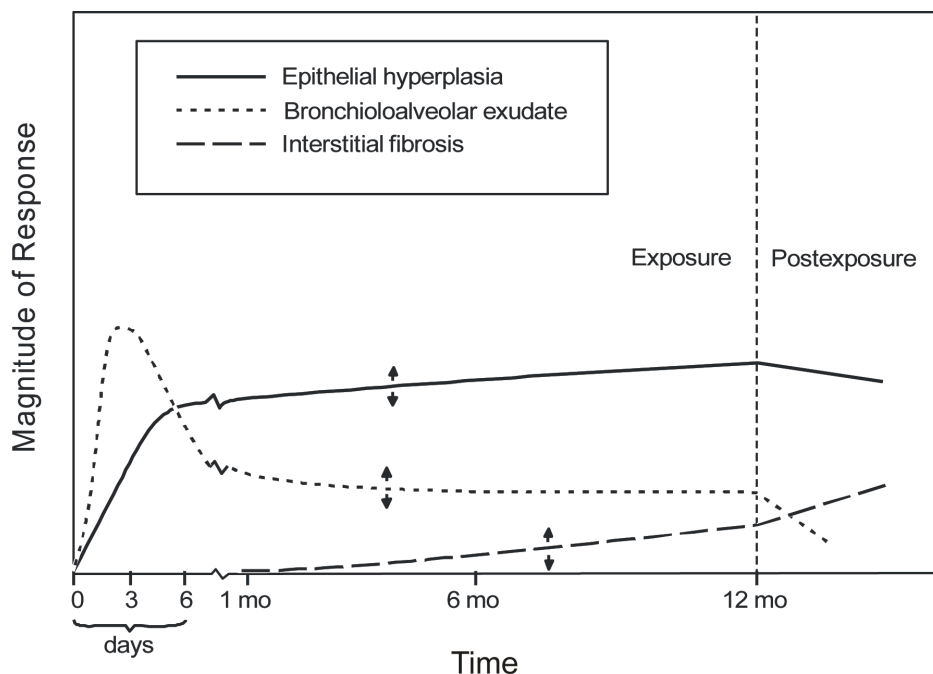


Figure 5-5. Schematic comparison of the duration-response profiles for epithelial hyperplasia, bronchioalveolar exudation, and interstitial fibrosis in the centriacinar region of lung exposed to a constant low concentration of ozone.

Source: Dungworth (1989).

epithelium of rats caused by a previous 3 day 0.5 ppm O₃ exposure (Fanucchi et al., 1998). Male F344/N Hsd rats were intranasally instilled with endotoxin after exposure to filtered air (FA) or 0.5 ppm O₃, (8 h/day for 3 days). Mucous cell metaplasia was not found in the air/endotoxin group but was found in the O₃/saline group and was most severe in the O₃/endotoxin group. A similar synergistic effect was demonstrated by Wagner et al. (2001a,b) with exposure of Fischer rats to O₃ and/or endotoxin for 8 h per day for 3 days. Ozone alone created epithelial lesions in the nasal transitional epithelium, while endotoxin alone caused lesions in the epithelium of the nose and conducting airways. The enhanced O₃-induced mucous cell metaplasia was related to neutrophilic inflammation.

Pre-metaplastic responses, such as mucin mRNA upregulation, neutrophilic inflammation, and epithelial proliferation, were shown to be responsible for O₃-induced mucous cell metaplasia in the transitional epithelium of rats (Cho et al., 1999a, 2000). Male F344/N rats exposed to O₃ (0.5 ppm, 8 h/day for 1, 2, or 3 days) showed a rapid increase in an airway-specific mucin gene

mRNA after exposure to O₃, both before and during the onset of mucous cell metaplasia. Neutrophilic inflammation coincided with epithelial DNA synthesis and upregulation of rMuc-5AC but was resolved before the development of epithelial metaplasia. The mucous cell metaplasia was neutrophil-dependent, whereas O₃-induced epithelial cell proliferation and mucin gene upregulation were neutrophil-independent.

Dormans et al. (1999) compared the extent and time course of fibrotic changes in mice, rats, and guinea pigs exposed to 0.2 or 0.4 ppm O₃ for 3, 7, 28, and 56 days. They found a concentration-related centriacinar inflammation in all three species, with a maximum after 3 days of exposure and total recovery within 3 days after exposure. Repair of O₃-induced damage by removal of injured epithelial cells is enhanced by the influx of neutrophils (Hyde et al., 1999; Veseley et al., 1999b; Miller et al., 2001; see Section 5.2.3). A study examining the kinetics of early cellular responses to O₃ utilized bromodeoxyuridine to label S-phase cells (Hotchkiss et al., 1997). Labeling indices for rat nasal transitional epithelial cell DNA were greatest 20 to 24 h after O₃ (0.5 ppm for 8 h) exposure, and remained greater than control at 36 h PE.

Very few published studies have explicitly explored susceptibility factors such as species, gender, age, antioxidant defense, acute and chronic airway disease, and exercise. Most typical laboratory species studied have qualitatively similar effects associated with O₃ exposure. Dormans et al. (1999) compared morphological, histological, and biochemical effects in the rat, mouse, and guinea pig following O₃ exposure and recovery in clean air. Wistar RIV:Tox male rats, specific-pathogen free (SPF)-bred National Institutes of Health (NIH) male mice, and Hartley Crl:(HA)BR male guinea pigs were continuously exposed to FA, 0.2 or 0.4 ppm for 3, 7, 28, and 56 days. Recovery from 28 days of exposure was studied at intervals of 3, 7, and 28 days PE. The mouse was the most sensitive as shown by a concentration and exposure time-dependent persistence of bronchiolar epithelial hypertrophy, elevated lung enzymes, and slow recovery from exposure. Exposure to the high dose for 56 days in both rats and guinea pigs caused increased amounts of collagen in ductal septa and large lamellar bodies in Type II cells. The inflammatory response was greater in the guinea pig. Overall, the authors rated mice as most susceptible, followed by guinea pigs and rats. However, this ranking was done without first adjusting for differences in species-specific delivered dose of O₃.

Ferrets, monkeys, and rats were exposed to O₃ (1.0 ppm, 8 h) to compare airway effects (Sternner-Kock et al., 2000). The ferrets and monkeys had similar epithelial necrosis and inflammation that was more severe than that found in rats. Because ferrets have a similar pulmonary structure as humans (e.g., well-developed respiratory bronchioles and submucosal glands), the authors concluded that the ferret would be a better model than rodents for O₃-induced airway effects. Age susceptibility is dependent on the endpoint examined (see Chapter 4 for discussions of age-related differences in O₃ dosimetry). One study (Dormans et al., 1996) demonstrated that O₃-induced centriacinar lesions are larger in younger rats than in older rats with exposures to 0.4 ppm O₃ for 1 to 7 days.

New studies have examined O₃-induced morphological effects in compromised laboratory animals. Rats with endotoxin-induced rhinitis were more susceptible to mucous cell metaplasia in the nasal transitional epithelium caused by a 3 day exposure to 0.5 ppm O₃ (Cho et al., 1999b). Wagner et al. (2002) reported a similar O₃-induced enhancement of inflammatory and epithelial responses associated with allergic rhinitis. Brown Norway rats were exposed to 0.5 ppm O₃, 8 h/day for 1 day or 3 consecutive days and then immediately challenged intranasally with either saline or OVA. Multiple exposures to O₃ caused greater increases in mucosubstances produced in the nose by allergen challenge.

Recent research has focused on the concept of O₃-susceptible and -nonsusceptible sites within the respiratory tract, including in situ antioxidant status and metabolic activity. Plopper et al. (1998) examined whether the variability of acute epithelial injury to short-term O₃ exposure within the tracheobronchial tree is related to local tissue doses of O₃ or to local concentrations of GSH. Adult male rhesus monkeys exposed to O₃ (0.4 or 1.0 ppm for 2 h) demonstrated significant cellular injury at all sites, but the most damage, along with increased inflammatory cells, occurred in the proximal respiratory bronchiole. A significant increase in GSH was found in the proximal bronchus at 0.4 ppm O₃ and a decrease in the respiratory bronchiole at 1.0 ppm O₃. A significant decrease in the percentage of macrophages, along with significant increases in the percentage of neutrophils and eosinophils, and a doubling of total lavage protein, were found after exposure to 1.0 ppm O₃ only. The authors concluded that the variability of local O₃ dose in the respiratory tract was related to inhaled O₃ concentration and was closely associated with local GSH depletion and with the degree of epithelial injury.

Plopper and colleagues (e.g., Watt et al., 1998; Paige et al., 2000a) explored the site-specific relationship between epithelial effects of O₃ exposure and the metabolism of bioactivated compounds within the respiratory tract of rats. The distribution of CYP 2E1-dependent activity, measured with a selective substrate (p-nitrocatechol), was found to be highest in the distal bronchioles and minor daughter airways and lower in the lobar bronchi and major daughter airways. Short-term O₃ exposure (1 ppm for 8 h) increased CYP 2E1 activity in the lobar bronchi/major daughter airways only; however, long-term O₃ exposure (1 ppm for 90 days) decreased CYP 2E1 activity in the major and minor airways, further complicating the interpretation of O₃ effects based on concentration and duration of exposure and recovery. Rats treated i.p. with 1-nitronaphthalene, a pulmonary toxicant requiring metabolic activation, and exposed to 0.8 ppm O₃, 8 h/day for 90 days showed greater histopathologic and morphometric effects in the CAR of the lung (Paige et al., 2000b). Despite reported tolerance to oxidant stress after long-term O₃ exposure, there was increased severity of ciliated cell toxicity.

5.2.4.2 Summary of Acute and Subchronic Morphological Effects

Short-term exposures to O₃ cause similar alterations in lung structure in a variety of laboratory animal species. Cells in the CAR are the primary targets of O₃, but ciliated epithelial cells in the nasal cavity and airways and Type I epithelial cells in the gas exchange region are also targets. New work has shown that a topical anti-inflammatory corticosteroid can prevent these effects in nasal epithelia, while exposure to bacterial endotoxin can potentiate the effects. Ozone-induced fibrotic changes in the CAR are maximal at 3 days of exposure and recover 3 days PE with exposures of 0.2 ppm in rodents. New studies of susceptibility factors demonstrated that ferrets and monkeys have similar inflammatory and necrotic responses to 1 ppm O₃, which differs from lesser injury seen in rats. Rats with induced allergic rhinitis are more susceptible to 0.5 ppm O₃ than are control animals. Important new work has demonstrated variability of local O₃ dose and subsequent injury in the respiratory tract due to depletion of GSH. The proximal respiratory bronchiole receives the most acute epithelial injury from exposures ≤ 1 ppm, while metabolic effects were greatest in the distal bronchioles and minor daughter airways. Table 5-1 summarizes new studies evaluating the morphological effects of a single acute O₃ exposure.

Table 5-1. Summary of Studies that Evaluated Morphological Effects of a Single Acute O₃ Exposure.

O ₃ Concentration (PPM)	Exposure Duration	Species	Nasal	Tracheal	Bronchi (Proximal)	Bronchi (Distal)	Bronchioles (Proximal)	Bronchioles (Distal)	Alveolar Ducts	Alveolar	References
0.25	20-90 min	Rat	nr	nr	↑	↑↑	nr	↑	nr	nr	Postlethwait et al. (2000)
0.5	"	"	nr	nr	↑↑	↑	nr	↑	nr	nr	Postlethwait et al. (2000)
1	"	"	nr	nr	↑↑↑	↑	nr	↑	nr	nr	Postlethwait et al. (2000)
1	90 min	Rat	nr	nr	↑↑↑	↑↑	↑	↑	nr	nr	Joad et al. (2000)
0.5	8 hr	Rat	↑↑↑	nr	nr	nr	nr	nr	nr	nr	Hotchkiss et al. (1997)
0.5	8 hr	Rat	↑↑↑	nr	nr	nr	nr	nr	nr	nr	Cho et al. (1999a, 2000)
0.5	8 hr	Rat	↑↑	nr	nr	nr	nr	nr	nr	nr	Wagner et al. (2002)
1	8 hr	Rat	nr	nr	↑	nr	nr	↑↑↑	nr	nr	Schelegle et al. (2001)
1	8 hr	Rat	↑↑	nr	↑	↑	nr	↑	nr	nr	Vesely et al. (1999a)
1	8 hr	Rat	nr	nr	nr	nr	nr	↑	↑	nr	Sterner-Kock et al. (2000)
0.4	2 hr	Monkey	nr	→	↑	↑	nr	↑↑	nr	nr	Plopper et al. (1998)
1	"	"	nr	↑	↑	↑	nr	↑↑	nr	nr	Plopper et al. (1998)
0.8	8 hr	Monkey	nr	nr	nr	nr	nr	↑↑	nr	nr	Hyde et al. (1999)
1	8 hr	Monkey	nr	nr	nr	nr	nr	↑↑	nr	nr	Sterner-Kock et al. (2000)
1	8 hr	Ferret	nr	nr	nr	nr	nr	↑↑	nr	nr	Sterner-Kock et al. (2000)

Symbols: → = no change; ↑ = small increase in indices reflecting O₃-induced morphological changes in specified upper or lower respiratory region; ↑↑ = relatively moderate morphological changes induced by O₃ exposure; ↑↑↑ = marked morphological changes or remodeling due to O₃ exposure; nr = changes not reported. The number of arrows in each study is based on the authors' description of injury or from graphs in the paper comparing O₃-exposed animals to air-exposed controls. As endpoints for the studies varied among papers and information about injury was largely descriptive, the table does not attempt a quantitative comparison, but only a subjective illustration of the current literature. Specific types of changes are discussed in text of Section 5.2.4.

5.2.4.3 Subchronic and Chronic Exposure Effects

Summaries of new studies of morphological effects of subchronic and chronic exposures are listed in Table AX5-10 in Annex AX5. In general, as the duration of exposure lengthens, there is no concomitant linear increase in the intensity of effect for any given endpoint. Rather, as exposure proceeds past 1 week to 1 year, Type I cell necrosis and inflammatory responses generally decrease to near control values, and hyperplastic and fibrotic changes remain elevated. After long-term exposure ended, some indicies of fibrosis persisted and, in some cases, became more severe during PE periods in clean air.

Effects of O₃ on the upper respiratory tract of F344 rats exposed to O₃ (0.12, 0.5, or 1.0 ppm for 20 months) included marked mucous cell metaplasia in the rats exposed to 0.5 and 1.0 ppm O₃, but not at 0.12 ppm O₃ (Harkema et al., 1997a). In a follow-up study, hyperplasia was found in the nasal epithelium of rats exposed to 0.25 and 0.5 ppm, 8 h/day, 7 days/week, for 13 weeks (Harkema et al., 1999). The mucous cell metaplasia, and associated intraepithelial mucosubstances, induced by 0.5 ppm O₃ persisted for 13 weeks after exposure. An acute (8 h) exposure to 0.5 ppm O₃ 13 weeks after the chronic exposure induced an additional increase of mucosubstances in the nasal epithelium of rats but not in rats chronically exposed to 0 or 0.25 ppm O₃. The persistent nature of the O₃-induced mucous cell metaplasia in rats reported in this study suggests that O₃ exposure may have the potential to induce similar long-lasting alterations in the airways of humans.

No significant changes in nasal tissue were seen in rats continuously exposed for 49 days to the ambient air of Mexico City, Mexico (Moss et al., 2001). A rat study using 6-month exposures to ambient air of Sao Paulo, with a different pollutant composition than that of Mexico City, demonstrated development of secretory hyperplasia in rats (Lemos et al., 1994). However, without information on specific differences in ambient pollution composition in the two cities, the studies cannot be compared; nor can the observed changes be confidently attributed to O₃ versus the overall oxidant-contaminated ambient air mix. Because of the persistent nature of these changes in the controlled studies with rats and the fact that the upper airways of humans are probably more sensitive (like in the monkey) the authors suggested that long-term exposure to ambient levels of O₃ could induce significant nasal epithelial lesions that may compromise the upper respiratory tract defense mechanisms of exposed human populations.

Rats exposed to 0.5 ppm O₃ for 1 month exhibited Bcl-2 in protein extracts of nasal epithelium (Tesfaigzi et al., 1998). Further, after 3 and 6 months of exposure, the number of metaplastic mucous cells in the transitional epithelium was indirectly related to the percentage of cells that were Bcl-2 positive. Cells from rats exposed to FA did not express any Bcl-2. This study suggests that apoptosis regulators like Bcl-2 may play a role in the development and resolution of mucous cell metaplasia in the nasal airway.

A spectrum of lesions was reported (Herbert et al., 1996) in the nasal cavity and centriacinar lung of male and female mice exposed to 0.5 or 1.0 ppm of O₃ for 2 years; the lesions persisted with continued exposure for 30 months. These lesions included bone loss in the maxilloturbinates, mucosal inflammation, mucous cell metaplasia in the nasal transitional epithelium and increased interstitial and epithelial thickening in the proximal alveolar region. In the CAR, there were increased numbers of nonciliated cells. However, changes in other endpoints including lung function and lung biochemistry were not evident. The investigators' interpretation of the entire study is that rodents exposed to the two higher O₃ concentrations had some of the structural hallmarks of chronic airway disease in humans.

A chronic study using a simulated, seasonal O₃-exposure pattern was reported by Plopper and colleagues (Evans et al., 2003; Schelegle et al., 2003a; Chen et al., 2003; Plopper and Fanucchi, 2000). Infant rhesus monkeys (30 days old) were exposed to FA, house dust mite allergen aerosol (HDMA), or O₃ + HDMA. The 0.5 ppm O₃ exposures were 8 h/day for 5 days, every 14 days for a total of 11 O₃ episodes. Half of the monkeys were sensitized to HDMA (*Dermatophagoides farinae*) at 14 and 28 days of age. The sensitized monkeys were exposed to HDMA for 2 h/day on Days 3 to 5 of the FA or O₃ exposures. The lungs were removed during the last FA exposure and the right and left cranial and right middle lobes were separately inflation fixed. Microdissection and morphometric analyses were performed on the conducting airways to the level of the most proximal respiratory bronchiole. Repeated exposures to O₃ or O₃ + HDMA over a 6-month period resulted in an atypical development of the basement membrane zone (BMZ) of airways in nonsensitized developing monkeys. Remodeling in the distal conducting airways was found in the sensitized monkeys as a result of the damage and repair processes occurring with repeated exposure (Evans et al., 2003; Schelegle et al., 2003a). Lung function changes in these monkeys (Schelegle et al., 2003b), and associated adaptation of the respiratory motor responses (Chen et al., 2003), are described in Section 5.2.5.2. Collectively,

these findings provide a plausible pathophysiologic basis for changes in airway function described in children growing up in polluted metropolitan areas (e.g., Tager, 1999) (see Chapter 7).

Necropsy of the left caudal lobe of these infant monkeys showed accumulation of mucous cells and eosinophils within the combined epithelial and interstitial compartments in the conducting airways and in the terminal/respiratory bronchioles (Schelegle et al., 2003a). House dust mite sensitization and HDMA challenge alone, or combined with O₃ exposure, resulted in significantly greater eosinophil accumulation in the conducting airways when compared to FA- and O₃-only exposures. A significant accumulation of eosinophils was found in the terminal/respiratory bronchioles of the sensitized monkeys challenged with HDMA when compared to monkeys exposed to FA, O₃, and HDMA + O₃. The mean mass of mucous cells increased in the fifth generation conducting airways of sensitized animals challenged with HDMA alone and when combined with O₃ exposure and in the terminal bronchioles of sensitized animals exposed to HDMA + O₃. The tracheal basement membrane of HDMA-sensitized monkeys exposed to HDMA or to HDMA + O₃ was significantly increased over controls; however, there were no significant changes in the airway diameter of proximal and mid-level airways. Exposures of sensitized young monkeys to HDMA alone, or to O₃ alone, resulted in eosinophilia of the mid-level conducting airways and the terminal/respiratory bronchioles, but without alterations in airway structure or function. The authors interpreted these findings to indicate that the combination of cyclic O₃ exposure and HDMA challenge in HDMA-sensitized infant monkeys acts synergistically to produce an allergic-reactive airway phenotype characterized by significant eosinophilia of mid-level conducting airways, transmigration of eosinophils into the lumen, and an altered structural development of conducting airways that is associated with increased Raw and nonspecific airway reactivity (see Section 5.2.5).

Examination of development of the tracheal BMZ in these monkeys (Evans et al., 2003) showed that with exposures to either O₃ or HDMA + O₃, BMZ development was affected. Abnormalities in the BMZ included: (1) irregular and thin collagen throughout the BMZ; (2) perlecan depleted or severely reduced; (3) fibroblast growth factor receptor-1 (FGFR-1) immunoreactivity reduced; (4) fibroblast growth factor-2 (FGF-2) immunoreactivity absent in perlecan-deficient BMZ, but present in the lateral intercellular space (LIS), in basal cells, and in attenuated fibroblasts; (5) syndecan-4 immunoreactivity increased in basal cells. The authors

interpret these data as suggesting that O₃ target cells are associated with synthesis of epithelial BMZ perlecan. The absence of FGF-2, normally stored in the BMZ, could affect downstream signaling in airway epithelium and could be responsible for the abnormal development of the airway seen in this study and thus be an important mechanism modulating O₃-induced injury. Mid-level bronchi and bronchioles from these monkeys (Larson et al., 2004) demonstrated decrements in the density of epithelial nerves in the axial path between the sixth and seventh airway generations in exposures to O₃. Combined O₃ + HDMA exposures exacerbated this reduction. They attribute this loss of nerve plexuses to neural regression or stunted nerve development, the latter corroborated by the Evans et al. (2003) finding of decreased growth factors following O₃ exposure. Additionally, they found streaks or clusters of cells that were immunoreactive for protein gene product 9.5 (PGP 9.5, a pan-neuronal marker) and negative for calcitonin gene-related peptide. The functional significance of this is unknown but suggests to the authors a possible injury-repair process induced by O₃.

Remodeling of the distal airways and CAR is one of the most disturbing aspects of the morphological changes occurring after subchronic and chronic exposure to O₃. Recently, bronchiolization was reported in rats exposed to 0.4 ppm O₃ for only 56 days (van Bree et al., 2001). They also found collagen formation progressively increased with increasing O₃ exposure and persisted into PE recovery. In addition to centriacinar remodeling, Pinkerton et al. (1998) reported thickening of tracheal, bronchial, and bronchiolar epithelium after 3 or 20 months exposure to 1 ppm O₃, but not to 0.12 ppm. Although some older literature had reported that chronic exposures to ≤1.0 ppm O₃ cause emphysema, none of the more recent literature supports this hypothesis.

5.2.4.4 Summary and Conclusions—Subchronic and Chronic Morphological Effects

The progression of effects during and after a chronic O₃ exposure across a range of 0.5 to 1.0 ppm is complex, with inflammation peaking over the first few days of exposure, then dropping, next plateauing, and then finally largely disappearing. Epithelial hyperplasia follows a somewhat similar pattern. Effects of 0.5 ppm O₃ for 20 months on the nasal mucosa include atrophy of nasal turbinates, epithelial hyperplasia, and mucous cell metaplasia, which persisted long after the exposure ceased. Fibrotic changes in lung tissue increase very slowly over months of exposure; and, after exposure ceases, the changes sometimes persist or increase. The pattern

of exposure in this same concentration range determines effects, with 18 months of daily exposure causing less morphologic damage than exposures on alternating months. This is important, given that environmental O₃ exposure is typically seasonal. Plopper and colleagues' long term study of infant rhesus monkeys exposed to simulated, seasonal O₃ (0.5 ppm 8 h/day for 5 days, every 14 days for 11 episodes) demonstrated: (1) remodeling in the distal airways; (2) abnormalities in tracheal basement membrane; (3) eosinophil accumulation in conducting airways; and (4) decrements in airway innervation. These findings advance earlier information regarding possible injury-repair processes occurring with seasonal O₃ exposures.

5.2.5 Effects on Pulmonary Function

5.2.5.1 Acute and Subchronic Exposure Effects on Pulmonary Function

Numerous pulmonary function studies of the effects of acute O₃ exposure (defined here as ≤ 1 week of exposure) in several animal species have been conducted and generally show responses similar to those of humans (e.g., increased breathing frequency, decreased tidal volume, increased resistance, decreased forced vital capacity [FVC] and changes in the expiratory flow-volume curve). These effects are seen at 0.25 to 0.4 ppm O₃ for several hours in a number of species. At concentrations of ≥ 1 ppm, breathing mechanics (compliance and resistance) are affected. The breathing pattern returns to normal after O₃ exposure. In rats exposed to 0.35 to 1 ppm O₃ for 2 h/day for 5 days, there was a pattern of attenuation of pulmonary function responses similar to that observed in humans. Concurrently, there was no attenuation of biochemical indicators of lung injury or of morphological changes.

Work demonstrating attenuation of pulmonary function effects (see Table AX5-11) was completed by Wiester et al. (1996), who exposed male Fischer 344 rats to 0.5 ppm O₃ for either 6 or 23 h/day over 5 days. Ozone-induced changes in lung volume were attenuated during the 5 exposure days and returned to control levels after 7 days recovery. The responses to repeated O₃ exposure in rats were exacerbated by reduced ambient temperature, presumably as a result of increased metabolic activity.

In an attempt to model susceptible populations, researchers have utilized inbred mouse strains with varying ventilatory responses to O₃. As differences were seen in inflammatory responses to acute O₃ exposures in C57BL/6J and C3H/HeJ mice, comparisons were also made of their ventilatory responses (Tankersley et al., 1993). Following an exposure of 2 ppm O₃ for

3 h, breathing frequency (f_B), tidal volume (V_T), and minute ventilation (\dot{V}_E) were measured 1 and 24 h in both normocapnia (or air at $\sim 0\%$ CO_2) and hypercapnia (5 or 8% CO_2). They demonstrated that acute O_3 exposures caused altered hypercapnic ventilatory control, which varied between strains. This suggested to the authors that O_3 -induced alterations in ventilation are determined, at least in part, by genetic factors. A caveat regarding studies such as this using high exposure concentrations is that events observed at high concentrations may differ from those observed at near-ambient O_3 levels.

Paquette et al. (1994) measured ventilatory responses in C57BL/6J and C3H/HeJ mice given repeated acute exposures of 0.3 ppm for 48 and 72 h. The two strains had differing responses to both normocapnia and hypercapnia. Normocapnic \dot{V}_E was greater following subacute O_3 exposure in C57BL/6J mice than in C3H/HeJ mice, due to increased f_B and reduced V_T , respectively. This suggests that the increased V_T in C57BL/6J mice may contribute to the increased susceptibility to lung injury due to a greater dose of O_3 reaching the lower lung. Hypercapnic ventilatory responses following subacute O_3 exposures demonstrated reduced \dot{V}_E (due to decreased V_T) in the C57BL/6J mice only. Evaluations of O_3 dosimetry were performed in these two strains using $^{18}\text{O}_3$ -labeled ozone (2 ppm for 2-3 h) (Slade et al., 1997). Immediately after exposures of 2 ppm $^{18}\text{O}_3$ for 2 to 3 h, C3H/HeJ mice had 46% less ^{18}O in lungs and 61% less in the trachea than C57BL/6J. Additionally, C3H/HeJ mice had a greater body temperature decrease following O_3 exposure than C57BL/6J mice, suggesting that the differences in susceptibility to O_3 are due to differences in the ability to decrease body temperature and, consequently, decrease the dose of O_3 to the lung.

Tracheal transepithelial potential has also been shown to differ in eight mouse strains 6 h after exposure to 2 ppm O_3 for 3 h (Takahashi et al., 1995b). AKR/J, C3H/HeJ, and CBA/J were identified as resistant strains; and 129/J, A/J, C57BL/6J, C3HeB/FeJ, and SJL/J were identified as susceptible strains. The authors noted that strains' responses in this parameter did not show concordance with inflammatory responses, suggesting to the authors that the two phenotypes are not controlled by the same genetic factors.

Savov et al. (2004) characterized ventilatory responses in nine mouse strains exposed to O_3 (2.0 ppm O_3 for 3 h). The C57BL/6J strain was hyporeactive to methacholine (MCh) prior to O_3 , but was very responsive to MCh following O_3 . Conversely, C3H/HeJ mice had an intermediate

baseline P_{enh} and a small response to MCh following O_3 exposure. This study corroborates the evidence of no consistent relationship between respiratory P_{enh} and inflammation.

5.2.5.2 Summary and Conclusions—Acute and Subchronic Effects on Pulmonary Function

Early work has demonstrated that during acute exposure of ~ 0.2 ppm O_3 in rats, the most commonly observed alterations are increased frequency of breathing and decreased tidal volume (i.e., rapid, shallow breathing). Exposures of ~ 1.0 ppm O_3 affect breathing mechanics (compliance and resistance). Additionally, decreased lung volumes are observed in rats with acute exposures at levels of 0.5 ppm. New work utilizing inbred mouse strains with varying ventilatory responses to O_3 has suggested that: (1) control of the ventilatory response is determined, at least in part, by genetic factors; (2) increased V_T in some strains may contribute to lung injury due to a greater dose of O_3 reaching the lower lung; (3) the ability to reduce body temperature in some strains may account for their decreased O_3 -induced lung injury; and (4) tracheal transepithelial potential is determined, in part, by genetic factors. Importantly, the genetic loci that appear to be modulating various aspects of pulmonary responses to O_3 differ from each other and from loci controlling inflammatory responses.

Exposures of 2 h/day for 5 days create a pattern of attenuation of pulmonary function decrements in both rats and humans without concurrent attenuation of lung injury and morphological changes, indicating that the attenuation did not result in protection against all the effects of O_3 . Chronic O_3 exposure studies evaluating pulmonary function are not available. Earlier work has demonstrated that repeated daily exposure of rats to an episodic profile of O_3 caused small, but significant, decrements in lung function that were consistent with early indicators of focal fibrogenesis in the proximal alveolar region without overt fibrosis.

5.2.5.3 Ozone Effects on Airway Responsiveness

Effects of O_3 on airway reactivity have been observed in a variety of species across an exposure range of 0.5 to 1 ppm. Many of the new studies on pulmonary function in laboratory animals allow a better prediction of the effects of O_3 exposure on the exacerbation of asthma symptoms and the risk of developing asthma in humans. However, it is necessary to understand the factors that determine airway responsiveness across different mammalian species, as is discussed in Chapter 4.

Traditional studies of airway responsiveness require sedation in both infants and laboratory animals. Laboratory animal studies employ intravenous agonist challenges as well as inhalation challenges, though inhaled agonist challenges are preferred in humans. Exercise testing is not possible with sedation unless exercise is “simulated” by increasing ventilation using elevated CO₂, and the need for artificial ventilation in laboratory animal studies may cause breathing patterns that affect O₃ deposition. Joad et al. (2000) reported that when 1 ppm O₃ for 90 min is administered to isolated rat lung at either 2.4 mL/40 breaths/minute (bpm) or 1.2 mL/80 bpm, the more rapid breathing pattern elicits less epithelial cell injury than the slower breathing pattern. Though this study design does not really model the rapid, shallow breathing elicited in the intact animal, it shows greater reduction in injury in the proximal axial airway compared to its adjacent airway branch and terminal bronchiole. The rapid, shallow breathing pattern protects the large conducting airways of rats but causes a more even distribution of epithelial cell injury to the terminal bronchioles (Schelegle et al., 2001). Postlethwait et al. (2000) demonstrated that the conducting airways are the primary site of acute cytotoxicity from O₃ exposure. Three-dimensional mapping of the airway tree in SD rat isolated lung exposed to 0, 0.25, 0.5, or 1.0 ppm O₃ showed a concentration-dependent increase in injured cells. Injury was evident in proximal and distal conduction airways, lowest in terminal bronchioles, and highest in the small side branches downstream of bifurcations. These exposure levels did not concurrently elicit changes in LDH activity or total protein in BALF, suggesting that the mapping technique is a more sensitive measure of injury and is useful in dosimetry studies.

Whole-body plethysmography of unanesthetized, unrestrained rodents has been used to indirectly measure pulmonary resistance (R_L) (Shore et al., 2002; Goldsmith et al., 2002; Jang et al., 2002). However, these indices of inspiratory/expiratory pressure differences, including P_{enh} , may be less sensitive than direct measurements of lung airflow resistance (Murphy, 2002). Changes in airway structure caused by viral infections also must be considered when evaluating laboratory animal studies. Animals with acute viral illness have morphological evidence of inflammatory cell infiltration, bronchiolar wall edema, epithelial hyperplasia, and increased airway mucous plugs that can cause airway narrowing, air trapping, and serious functional changes in the lung (Folkerts et al., 1998).

Exercise-induced bronchoconstriction in humans appears to be mediated by changes in the tonicity of the airway lining fluid (Anderson and Daviskas, 2000). Brannan et al. (1998) suggest

that a test in laboratory animals based on the inhalation of mannitol aerosol (hyperosmolar) may be feasible and provide information similar to that from exercise challenges in cooperative children and adults. Unfortunately, there have been few reports of mannitol or adenosine monophosphate challenges in laboratory animals; most studies have utilized histamine, MCh, acetylcholine (ACh), or carbachol to determine outcome. In active humans with asthma, adenosine monophosphate challenges appear to better reflect ongoing airway inflammation than histamine or MCh challenges (Polosa and Holgate, 1997; Avital et al, 1995a,b) and may be useful in identifying mechanisms of asthma in laboratory animals and their responsiveness to environmental pollutants.

The increased responsiveness to bronchoconstrictor challenge in asthma is thought to result from a combination of structural and physiological factors that include increased inner-wall thickness, smooth-muscle responsiveness, and mucus secretion. These factors also are likely to determine a level of innate airway responsiveness that is genetically influenced. Chapter 6 (Section 6.8) discusses cellular and biochemical changes that have been identified in human asthmatics. These studies suggest that the mechanisms involved in AHR are multifactorial, with general agreement that there is an inconsistent relationship between AHR and markers of inflammation.

A large database of laboratory animal research has been collected on the role of O₃ in producing an increase in AHR (see Table AX5-12). Exposure levels (≥ 1 ppm for ≥ 30 min) in many of these studies are not environmentally relevant, but information may be obtained regarding the mechanisms of action of O₃ concerning: O₃ concentration and peak response time, inhaled versus intravenous challenge with nonspecific bronchoconstrictors, neurogenic mediation, neutrophilic inflammation, and interactions with specific biological agents (e.g., antigens and viruses). However, as with other toxicants, high-dose and low-dose mechanisms may differ, so interpretation of results must take this into consideration.

Many species of laboratory animals have been used to study the effects of O₃ on airway bronchoconstriction. Ozone-induced AHR in guinea pigs has been used to model human bronchospasm (van Hoof et al., 1996; 1997a,b; Matsubara et al., 1997a,b; Sun and Chung, 1997; Aizawa et al., 1999a,b; Tsai et al., 1998; Nakano et al., 2000). Because these studies were done at 2 to 3 ppm O₃, these results are not directly relevant for extrapolation to potential airway responses in humans exposed to ambient levels of O₃. Humans with reactive airway disease

(e.g., asthma) appear to be sensitive to ambient levels of O₃ (see Chapters 6 and 7) and the current understanding is that O₃ exacerbates airway responsiveness to specific allergens, presumably by nonspecifically increasing AHR.

Shore et al. (2000, 2002) have shown that O₃-induced AHR is reduced in immature rats and mice. SD rats exposed to 2 ppm O₃ at ages 2, 4, 6, 8, or 12 weeks and A/J mice exposed to 0.3 to 3 ppm for 3 h at age 2, 4, 8, or 12 weeks had similar concentration-related decreases in \dot{V}_E except at the youngest ages. This smaller decrement in \dot{V}_E suggested a delivered dose that was much greater in the younger animals. This group (Shore et al., 2003) has also recently shown that obese mice have greater ventilatory responses to O₃. Exposures of 2.0 ppm O₃ for 3 h to lean, WT C57BL/6J and *ob/ob* mice (mice with a genetic defect in the coding for leptin, the satiety hormone) showed that the *ob/ob* mice had enhanced AHR and inflammation compared to the WT mice. These data correlate with epidemiological data showing increased incidence of asthma in overweight children.

Increased AHR to various nonspecific bronchoconstrictive agents (e.g., ACh, MCh, histamine, carbachol) given by inhalation or intravenous routes has been previously shown in laboratory animals exposed to O₃ concentrations ≤ 1.0 ppm. Dye et al. (1999) showed hyperresponsiveness to MCh in rats 2 h after exposure to 2 ppm O₃ for 2 h. AHR can be induced by specific antigens as well as O₃. The most commonly used laboratory animal model is the OVA-sensitized guinea pig. Animals sensitized with OVA have been shown to have similar responses to nonspecific bronchoconstrictors as control animals.

OVA-sensitized guinea pigs (Sun et al., 1997) and mice (Yamauchi et al., 2002) were used to determine the enhancement of antigen-induced bronchoconstriction by acute, high-level O₃ (1.0 ppm O₃ for 1 h). Male Dunkin-Hartley guinea pigs were sensitized by i.p. injection of OVA and exposed to O₃ alone, OVA aerosol, or O₃ + OVA. Ozone exposure alone increased bronchial responsiveness to ACh at 3 h, but not 24 h, whereas OVA alone had no effect. Combined exposure to O₃ and OVA (1 ppm for 1 h, then 3 min OVA) increased bronchial responsiveness to ACh 3 h after O₃ exposure. At 24 h following O₃ exposure, AHR increased when OVA challenge was performed at 21 h, suggesting that O₃ preexposure can potentiate OVA-induced AHR. Neutrophil counts in the BALF increased at 3 and 24 h after O₃ exposure alone but were not further increased when O₃ exposure was combined with OVA airway challenge; however, protein content of the BALF increased at 3 and 24 h in the O₃ and OVA

groups. Thus, this study also indicates that high-ambient O₃ exposure can augment antigen (OVA)-induced AHR in guinea pigs.

Yamauchi et al. (2002) sensitized male C57BL/6 mice by i.p. injection of OVA and then exposed them to O₃. The sensitized mice had AHR to MCh. Ozone exposure caused significant decreases in dynamic lung compliance (C_{dyn}), \dot{V}_E , and P_aO₂ in OVA-sensitized mice but not in controls. A marker of inflammation (soluble intercellular adhesion molecule-1 [sICAM-1]) was elevated in the BAL fluid of OVA-sensitized mice, but sICAM-1 levels were not significantly changed by O₃ exposure, indicating that the O₃-induced AHR to MCh was not caused by O₃-induced inflammation.

Ozone-induced AHR may be temporally associated with inflammatory cells stimulated by cytokines (Koto et al., 1997), mast cells (Igarashi et al., 1998; Noviski et al., 1999), or by oxygen radicals (Takahashi et al., 1993). One study, however, has shown that inflammation is not a prerequisite of AHR (Koto et al., 1997), and it has been suggested that O₃-induced AHR may be epithelium-dependent (McGraw et al., 2000). For example, neonatal rats pretreated with capsaicin, which will permanently destroy C-fibers and prevent O₃-induced (1 ppm, 8 h) release of neuropeptides (Vesely et al., 1999a), and then exposed to O₃ when adults, showed a marked increase in airway responsiveness to inhaled aerosolized MCh (Jimba et al., 1995). Takebayashi et al. (1998) has shown that depletion of tachykinins by capsaicin treatment, or by a specific tachykinin receptor antagonist, can block the induction of AHR by O₃. The seemingly disparate responses in laboratory animals may be due to species- or strain-specific differences in inherent reactivity to bronchoconstrictors or to inherent differences in susceptibility to O₃-induced inflammation (Zhang et al., 1995; Depuydt et al., 1999; Dye et al., 1999).

Studies that may be potentially relevant to ambient levels of O₃ were conducted in vivo, in an isolated perfused lung model, and in ex vivo lung segments using multihour and repeated multihour exposures with ambient levels of O₃. A study on the relationship between O₃-induced AHR and tracheal epithelial function was conducted in New Zealand White rabbits by Freed et al. (1996). Rabbits exposed to O₃ (0.2 ppm for 7 h) demonstrated significantly decreased tracheal transepithelial potential difference but no changes in lung resistance. Changes in the compartmentalized lung resistance, measured in response to ACh challenge before and after bilateral vagotomy, were not significantly different in air-exposed rabbits; however, bilateral vagotomy enhanced peripheral lung reactivity in O₃-exposed rabbits. The ACh-induced 140%

increase in lung resistance with O₃ exposure was two times higher than with air exposure, indicating that ambient-level O₃ exposure affects tracheal epithelial function in rabbits and increases central airway reactivity, possibly through vagally-mediated mechanisms.

Pulmonary mechanics and hemodynamics were studied in the New Zealand White rabbit isolated perfused lung model that allowed partitioning of the total pressure gradient into arterial, pre- and postcapillary, and venous components (Delaunois et al., 1998). Exposures to O₃ (0.4 ppm for 4 h) were followed by evaluation of airway responsiveness to ACh, substance P (SP), or histamine immediately or 48 h later. Ozone inhibited pulmonary mechanical reactivity to all three bronchoconstrictors that persisted for 48 h and modified vasoreactivity of the vascular bed, but only at 48 h PE. Arterial segmental pressure, normally insensitive to ACh and SP, was significantly elevated by O₃; precapillary segmental pressure decreased in response to Ach, suggesting that O₃ can induce direct vascular constriction, but the vascular responses are variable and depend on the agonist used and on the species studied.

Airway responsiveness to the same three compounds was evaluated by Segura et al. (1997) in guinea pigs exposed to O₃ (0.15, 0.3, 0.6, or 1.2 ppm for 4 h). Ozone did not cause AHR to ACh or histamine, except at the highest concentration (1.2 ppm O₃) for histamine. However, O₃ caused AHR to SP at ≥ 0.3 ppm, suggesting that O₃ destroys neutral endopeptidases (responsible for SP inactivation) in airway epithelial cells. Vargas et al. (1998), in a follow-up study, demonstrated that guinea pigs chronically exposed to 0.3 ppm O₃ for 4 h/day became adapted to SP-induced AHR. Ozone caused increased sensitivity to SP after 1, 3, 6, 12, and 24 days of exposure that was associated with airway inflammation; however, after 48 days of exposure, the increased sensitivity to SP was lost.

This study is in accordance with Szarek et al. (1995) who demonstrated that AHR associated with acute O₃ exposures does not persist during long-term exposure to near-ambient-levels of O₃ (≤ 1 ppm). Fischer 344 rats, exposed to 0.0, 0.12, 0.5, or 1.0 ppm O₃, 6 h/day, 5 days/week for 20 months, demonstrated significantly reduced responses to bethanechol, ACh, and electrical field stimulation in eighth generation airway segments. This suggests that some adaptation had taken place during long-term exposure, possibly due to increased inner wall thickness.

It is well known that the changes in breathing pattern and lung function caused by O₃ are attenuated with repeated daily exposures for at least 3 to 5 days. But guinea pigs exposed to

0.5 ppm O₃, 8 h/day for 7 days showed enhancement of responsiveness of rapidly adapting airway receptors (Joad et al., 1998). Repeated exposure increased receptor activity to SP, MCh, and hyperinflation; there were no significant effects on baseline or SP- and MCh-induced changes in lung compliance and resistance, suggesting that the responsiveness of rapidly adapting receptors was enhanced.

Male and female Hartley guinea pigs exposed to O₃ (0.1 and 0.3 ppm, 4 h/day, 4 days/week for 24 weeks) were evaluated for airway responsiveness following ACh or OVA inhalation challenges (Schlesinger et al., 2002a,b). Ozone exposure did not cause AHR in nonsensitized animals but did exacerbate AHR to both ACh and OVA in sensitized animals that persisted for 4 weeks after exposure. The effects of O₃ on airway responsiveness were gender independent and were concentration-related for the ACh challenges.

Schelegle et al. (2003a) evaluated airway responsiveness in infant rhesus monkeys exposed to a 5-day O₃ episode repeated every 14 days over a 6-month period. Half of the monkeys were sensitized to HDMA at 14 and 28 days of age before exposure to a total of 11 episodes of O₃ (0.5 ppm, 8 h/day for 5 days followed by 9 days of FA), HDMA, or O₃ + HDMA. Baseline Raw was significantly elevated after 10 exposure episodes in the HDMA + O₃ group compared to the FA, HDMA, and O₃ exposure groups. Aerosol challenge with HDMA at the end of the 10th episode did not significantly affect Raw, V_T, f_B, or S_aO₂. Aerosol challenge with histamine was not significantly different after 6 episodes; however, the EC150 Raw for the HDMA + O₃ group was significantly reduced after 10 episodes when compared to the FA, HDMA, and O₃ exposure groups, indicating the development of AHR in this group sometime between episodes 6 and 10. The results are consistent with altered structural development of the conducting airways.

During repeated episodic exposures to O₃, respiratory responses are first altered to a rapid, shallow breathing pattern, which has long been considered protective, especially to the deep lung. This dogma has been discounted recently as discussed above (Schelegle et al., 2001). Alfaro et al. (2004) examined the site-specific deposition of ¹⁸O (1 ppm 2 h) at breathing frequencies of 80, 120, 160, or 200 bpm. At all frequencies, parenchymal areas had a lower content of ¹⁸O than trachea and bronchi. As breathing frequency increased from 80 to 160 bpm, the deposition showed a reduction in mid-level trachea and an increase in both mainstream bronchi. At this frequency there was also an increase in deposition in parenchyma supplied by short (cranial) airway paths, consistent with the results of Schelegle et al., (2001).

At 200 bpm, ^{18}O deposition in trachea increased, concurrent with increases in right cranial and caudal bronchi regions. Right cranial parenchymal content decreased at 200 bpm, whereas right caudal parenchymal levels did not change at any breathing frequency. The authors list some limitations of this study, such as the possible effect on regional distribution of ventilation by use of the negative-pressure ventilator, the effect of paralysis on airway geometry, and possible translocation of ^{18}O during the 2 h exposure period. These two studies provide evidence that O_3 -induced rapid, shallow breathing creates a more evenly distributed injury pattern, with possibly greater protection from focal injury to the large conducting airways, including the trachea and the left mainstem bronchus.

Another study of the adaptive phenomena in SD rats used an exposure protocol consisting of 5 days of daily 8 h, 1 ppm O_3 exposures followed by 9 days of recovery in FA (Schelegle et al., 2003b). This O_3 /FA pattern was repeated for four cycles and demonstrated that the O_3 -induced rapid shallow breathing pattern was followed by adaptation that occurred with each cycle. However, the release of SP from the trachea, the neutrophil content, and cell proliferation became attenuated after the first cycle, suggesting a disconnect from the rapid, shallow breathing response. Hypercellularity of the CAR epithelium and thickening of the CAR interstitium, not linked to changes in cell proliferation, were also found. The authors suggest mechanism(s) of injury from repeated O_3 exposures, consisting of diminished neutrophilic inflammation/and or release of mitogenic neuropeptides, depressed cell proliferative response, and cumulative distal airway lesion.

Following the initial response of a rapid, shallow breathing pattern, animals eventually adapt with continued episodic exposure despite the continued presence of epithelial damage, altered structural development, and inflammation of the airways. Chen et al. (2003) used a subset of the monkeys from the Schlegel et al. (2003a) study to demonstrate that the attenuation of O_3 -induced rapid, shallow breathing and lung function changes typically seen with repeated O_3 exposure may be caused by the adaptation of the respiratory motor responses. This episodic O_3 exposure appeared to create neuroplasticity of the nucleus tractus solitarius (NTS, a region of the brainstem which controls respiration), including increased nonspecific excitability of the NTS neurons, an increased input resistance, and an increased spiking response to intracellular injections of depolarizing current.

5.2.5.4 Summary and Conclusions—Effects on Airway Responsiveness

Ozone-induced AHR has been reported in a number of laboratory species at an exposure range of 0.5 to 1.0 ppm and in human asthmatics at ambient levels. In asthmatics, O₃ is thought to exacerbate AHR to specific allergens by nonspecifically increasing AHR. New studies have demonstrated that AHR in asthmatics is due in part to chronic inflammation and airway remodeling. Animal studies have shown that O₃ exposure can augment OVA-induced AHR. Importantly, there is a temporal relationship between inflammatory cell influx and O₃-induced AHR, but inflammation is not a prerequisite of AHR. Repeated O₃ exposures enhance AHR, possibly by modulating rapidly adapting airway receptors or by altering the structure of conducting airways.

Currently reported investigations on AHR with repeated O₃ exposure to nonsensitized laboratory animals have shown equivocal results, especially at the most relevant ambient O₃ concentrations of ≤ 0.3 ppm. The few available studies in sensitized laboratory animals are consistent with the O₃-induced exacerbation of AHR reported in atopic humans with asthma (see Chapter 6), but the results are difficult to extrapolate because of interindividual and interspecies differences in responsiveness to bronchoprovocation and possible adaptation of airway responsiveness with long-term, repeated O₃ exposures. Therefore, further studies in laboratory animals are needed to investigate responses to the different challenges in relation to measurements of airway inflammation and the other physiological and structural factors known to contribute to airway responsiveness in human subjects.

Important new information indicates that rapid, shallow breathing in response to O₃ causes a more evenly distributed injury pattern rather than protection from injury. New insights into the mechanisms of O₃-induced AHR suggest that: (1) exercise-induced bronchoconstriction may be mediated by changes in tonicity of the bronchial smooth muscles; (2) vagally mediated mechanisms may affect tracheal epithelial function and increase central airway reactivity; (3) O₃ may induce direct vascular constriction; (4) O₃ may destroy neural endopeptidases in airway epithelial cells, thus preventing the inactivation of SP; and (5) repeated O₃ exposures may diminish neutrophilic inflammation, depress cell proliferation, and cause cumulative distal airway lesions.

5.2.6 Genotoxicity Potential of Ozone

There has been an historical interest in the ability of ground-level pollution to cause cancer, especially lung cancer. This interest has been amplified in recent years by results of an epidemiologic study that suggest association of increased risks of incident lung cancer with elevated long-term ambient concentrations of O₃, coarse particulate matter (PM₁₀), and sulfur dioxide (SO₂) in nonsmoking California males (Beeson et al., 1998; Abbey et al., 1999). However, another larger, nationwide American Cancer Society study (Pope et al., 2002) showed no significant effect of O₃ on mortality risk but showed positive associations between warm-season (July-September) O₃ concentrations and cardiopulmonary mortality. Studies of children and young adults of southwest metropolitan Mexico City, repeatedly exposed to high levels of O₃, particulate matter (PM), nitrogen oxides (NO_x), aldehydes, metals, and other components in a complex ambient mixture, also report DNA damage in blood leukocytes and nasal epithelial cells (Valverde et al., 1997; Calderón-Garcidueñas et al., 1999) and abnormal nasal biopsies (Calderón-Garcidueñas et al., 2001).

A number of experimental studies have been done to explore the mutagenic/carcinogenic potential of O₃. In vitro studies are difficult to interpret due to very high exposure levels and culture systems that allowed the potential formation of artifacts. Some recently published in vivo exposure studies (see Table AX5-13) found increased DNA strand breaks in respiratory cells from guinea pigs (Ferng et al., 1997) and mice (Bornholdt et al., 2002) but, again, only with exposure to high doses of O₃ (1 ppm for 72 h and 1 or 2 ppm for 90 min, respectively).

Exposing the A/J mouse strain (known to have a high incidence of spontaneous pulmonary adenomas) to 0.12, 0.50, and 1.0 ppm O₃ for 6 h/day, 5 days/week for up to 9 months, Witschi et al. (1999) did not find O₃ exposure-related differences in lung tumor multiplicity or incidence. Similarly, in a subchronic exposure study (B6C3F₁ mice to 0.5 ppm O₃ for 6 h/day, 5 days/week for 12 weeks), Kim et al. (2001) did not find statistically significant increases in the incidence of lung tumors. Significant differences in mean body weight as well as mean absolute and relative weights of several organs (e.g., liver, spleen, kidney, testes, and ovary) were observed between O₃-exposed and air-exposed mice. Histopathologic examination of major organs revealed oviductal carcinomas in 3/10 O₃-exposed female mice.

5.2.6.1 Summary and Conclusions—Genotoxicity Potential of Ozone

The weight of evidence from new experimental studies does not appear to support ambient O₃ as a pulmonary carcinogen in laboratory animal models. These new data are in agreement with a study of carcinogenicity of O₃ from the National Toxicology Program (NTP) study (National Toxicology Program, 1994; Boorman et al., 1994), which was negative in male and female rats, ambiguous in male mice, and positive only in female mice at high concentrations of O₃ (i.e., 1.0 ppm). As none of the new experimental studies of genotoxicity provided lifetime exposure durations such as those used in NTP cancer studies, the observation of no effects must be tempered by consideration of the limited duration of the exposure. Overall, then, the new animal studies are inconclusive, as are the epidemiologic studies discussed in Chapter 7, which may be due to significant species differences in this health endpoint. Also, O₃ could act as a co-carcinogen functioning to stimulate hyperplasia. In epidemiology studies, exposures typically consist of mixtures of copollutants, some of which are known carcinogens (see Section 5.4.3).

5.3 SYSTEMIC EFFECTS OF OZONE EXPOSURE

Ozone indirectly affects organs beyond the respiratory system due to O₃ reaction products entering the bloodstream and being transported to target sites. Extrapulmonary effects could also be due to the exposure-related production of mediators, metabolic products, and cell trafficking. Although systemic effects are of interest and indicate a very broad array of O₃ effects, they are of limited influence and difficult to interpret. By protecting from respiratory tract effects, these systemic effects will likely be protected against also. Systemic effects are only summarized briefly here and in Table AX5-14.

5.3.1 Neurobehavioral Effects

Animal behavior, both motor activity and operant behavior, has been shown to be suppressed by acute O₃ exposures (3 to 6 h) of 0.12 ppm. There is a dose dependent decrease in activity with increasing exposure levels. Additionally, these lowered activity levels tend to attenuate with longer exposure periods. New studies in adult laboratory animals confirm that environmentally relevant O₃ concentrations from 0.2 to 1.0 ppm can decrease motor activity and

affect short- and long-term memory, as tested by passive avoidance conditioning in 4 h exposures in rats (Rivas-Arancibia et al., 1998; Avila-Costa et al., 1999; Dorado-Martinez et al., 2001) or water-maze learning tasks in mice following a 30-day exposure (Sorace et al., 2001). The effects have been attributed to reactive oxygen/nitrogen species and/or ozonation products. The memory deficits could be blocked by administration of α -tocopherol (Guerrero et al., 1999) or taurine (Rivas-Arancibia et al., 2000). Increased freezing and decreased exploratory behaviors were accompanied by decreased serotonin levels and increased levels of NO, glutamate, dopamine, and striatal lipoperoxidation in rats exposed to 1 ppm of O₃ for 4 h (Rivas-Arancibia et al., 2003). The O₃-exposed animals also demonstrated neuronal cytoplasm and dendrite vacuolation and dilation of rough endoplasmic reticulum cisterns, which the authors interpret as a neurodegenerative process resulting from the oxidative stress of acute O₃ exposure. Niño-Cabrera et al. (2002) demonstrated that a 0.7 ppm O₃ exposure for 4 h can induce ultrastructural alterations in the hippocampus and prefrontal cortex in aged rats. These are areas of the brain where degenerative age-related changes in learning and memory functions have been reported (Bimonte et al., 2003).

Paz (1997) reviewed a series of studies that demonstrated significant alterations of electroencephalographic (EEG) patterns during sleep in animals acutely exposed to O₃ (0.35 to 1.0 ppm). Rats and cats both showed loss of paradoxical sleep time after 2 to 8 h of O₃ exposure (Paz and Bazan-Perkins, 1992; Paz and Huitrón-Reséndiz, 1996). Increased total wakefulness, alterations in circadian rhythm, and a permanent 50% loss of paradoxical sleep time were shown in rat pups born to dams exposed to 1.0 ppm O₃ during gestation (Haro and Paz, 1993). Effects on sleep patterns were associated with alterations in brain neurotransmitter levels (Huitrón-Reséndiz et al., 1994; González-Piña and Paz, 1997) thought to be caused by O₃ reaction products or prostaglandins (Koyama and Hayaishi, 1994). The permanent effects in pups caused by high O₃ exposure during gestation were attributed to the diminished antioxidant capability of fetal tissue (Günther et al., 1993).

High, nonambient levels of O₃ (e.g., >1.0 ppm) affect visual and olfactory neural pathways in the rat. For example, Custodio-Ramirez and Paz (1997) reported a significant delay in visual evoked potentials recorded in the visual cortex and the lateral geniculate nucleus of male Wistar rats acutely exposed to high levels of O₃ (1.5 and 3.0 ppm for 4 h). Colin-Barenque et al. (1999), using the same strain, reported cytological and ultrastructural changes in the granular layer of the

olfactory bulb after a 4 h exposure to 1 to 1.5 ppm O₃. Although these neural effects are thought to be caused by O₃ reaction products, especially free radicals, the studies do not add much to an understanding of the underlying mechanisms.

5.3.2 Neuroendocrine Effects

Early studies suggested an interaction of O₃ with the pituitary-thyroid-adrenal axis, because thyroidectomy, hypophysectomy, and adrenalectomy protected against the lethal effects of high concentrations of O₃. Concentrations of 0.7 to 1.0 ppm O₃ for a 1 day exposure in male rats caused changes in the parathyroid; thymic atrophy; decreased serum levels of thyroid stimulating hormone, triiodothyronine (T₃), thyroxine (T₄), free T₄, and protein binding; and increased prolactin. In more recent studies, increased toxicity to O₃ was reported in hyperthyroid rats by Huffman et al. (2001) and T₃ supplementation was shown to increase metabolic rate and pulmonary injury in the lungs of O₃-treated animals (Sen et al., 1993).

The mechanisms by which O₃ affects neuroendocrine function are not well understood. Cottet-Emard et al. (1997) examined catecholamine activity in rat sympathetic efferents and brain areas of prime importance to adaptation to environmental stressors. Exposures of 0.5 ppm O₃ for 5 days caused inhibition of norepinephrine turnover in heart (-48% of the control level) but not in lungs and failed to modify the tyrosine hydroxylase activity in superior cervical ganglia and the catecholamine content in the adrenal glands. In the CNS, O₃ inhibited tyrosine hydroxylase activity in noradrenergic brainstem cell groups and decreased catecholamine turnover in the cortex (-49%) and striatum (-18%) but not in the hypothalamus. This suggests that high ambient levels of O₃ can produce marked neural disturbances in structures involved in the integration of chemosensory inputs, arousal, and motor control, effects that may be responsible for some of the behavioral effects seen with O₃ exposure.

5.3.3 Cardiovascular Effects

Studies of the effects on hematological parameters and blood chemistry in rats have shown that erythrocytes are a target of O₃. Exposures to 1.0 ppm O₃ for 3 h have been found to decrease heart rate (HR), mean arterial pressure (MAP), and core temperature (T_{CO}) and to induce arrhythmias with some exposures in rats. These effects are more pronounced in adult and awake rats than in younger or sleeping animals. Exposures of 0.2 ppm for 48 h have been shown

to cause bradycardia, while exposures of 0.1 ppm O₃ for 3 days have been shown to cause bradyarrhythmia in these animals.

A more recent study of rats exposed to FA for 6 h, followed 2 days later by a 5 h exposure to 0.1 ppm O₃, 5 days later by a 5 h exposure to 0.3 ppm O₃, and 10 days later by a 5 h exposure to 0.5 ppm O₃, used the head-out plethysmograph for continuous measurements (Arito et al., 1997). Each of the O₃ exposures was preceded by a 1 h exposure to FA. Transient rapid, shallow breathing with slightly increased HR appeared 1 to 2 min after the start of O₃ exposures and was attributed to an olfactory response. Persistent rapid, shallow breathing with a progressive decrease in HR occurred with a latent period of 12 h. During the last 90-min of exposure, averaged values for relative \dot{V}_E tended to decrease with the increase in O₃ concentration for young (4 to 6 months) but not old (20 to 22 months) rats.

Studies by Watkinson et al. (1995, 2001) and Highfill and Watkinson (1996), that utilized radiotelemetry transmitters in unanesthetized unrestrained rats, demonstrated that when HR was reduced during a 5-day, 0.5 ppm O₃ exposure, T_{CO} and activity levels also decreased. The decreases in T_{CO} and BP reported in these studies and by Arito et al. (1997) suggest that the changes in ventilation and HR are mediated through physiological and behavioral defense mechanisms in an attempt to minimize the irritant effects of O₃ inhalation. Decreased activity was previously reported in laboratory animals during exposure to O₃ (see above).

Similar cardiovascular and thermoregulatory responses in rats to O₃ were reported by Iwasaki et al. (1998). Repeated exposure to 0.1, 0.3, and 0.5 ppm O₃ 8 h/day for 4 consecutive days caused disruption of circadian rhythms of HR and T_{CO} on the first and second exposure days that was concentration-dependent. The decreased HR and T_{CO} recovered to control values on the third and fourth days of O₃ exposure.

The thermoregulatory response to O₃ was further characterized by Watkinson et al. (2003). Male Fischer-344 rats were either exposed to 0.0 ppm for 24 h/day (air), 0.5 ppm for 6 h/day (intermittent), or to 0.5 ppm for 23 h/day (continuous) at 3 temperatures, 10 °C (cold), 22 °C (room), or 34 °C (warm). Another protocol examined the effects of O₃ exposure (0.5 ppm) and exercise (described as rest, moderate, or heavy) or CO₂-stimulated ventilation. Both intermittent and continuous O₃ exposure caused decreases in HR and T_{CO} and increases in BALF inflammatory markers. Exercise in FA caused increases in HR and T_{CO} while exercise in O₃ caused decreases in those parameters. Carbon dioxide and O₃ induced the greatest deficits in HR

and T_{CO} . Several factors were suggested that may modulate the hypothermic response, including dose, animal mass, and environmental stress.

Laboratory animals exposed to relatively high O_3 concentrations (≥ 0.5 ppm) demonstrate tissue edema in the heart and lungs. This may be due to increased circulating levels of atrial natriuretic factor (ANF), which is known to mediate capillary permeability, vasodilation, and BP (Daly et al., 2002). Increased levels of ANF were reported in the heart, lungs, and circulation of rats exposed to 0.5 ppm O_3 for 8 h (Vesely et al., 1994a,b,c).

Earlier work demonstrated O_3 -induced release of functionally active PAF from rodent epithelial cells and the presence of PAF receptors on AMs. New work examining lipid metabolism (Section 5.2.1.4) and mediators of inflammatory response and injury (Section 5.2.3.4) confirm earlier findings indicating that PAF (Kafoury et al., 1999) and PAF receptors (Longphre et al., 1999) are involved in responses to O_3 . In addition to the role of PAF in pulmonary inflammation and hyperpermeability, this potent inflammatory mediator may have clotting and thrombolytic effects, though this has not been demonstrated experimentally (see Figure 5-2). This cardiovascular effect may help explain, in part, some limited epidemiologic findings suggestive of possible association of heart attack and stroke with ambient O_3 exposure (see Chapter 7). The findings of Pulfer and Murphy (2004) and Pulfer et al. (2005), as discussed in Section 5.2.1.4, which characterize the *in vitro* and *in vivo* production of two biologically active oxysterols, are also suggestive of a mechanism whereby O_3 exposure might be implicated in the increased risk of cardiopulmonary disease.

5.3.4 Reproductive and Developmental Effects

Early studies of pre- and postnatal exposure to O_3 were performed at relatively high concentrations. Teratogenic effects were not observed with intermittent exposures of 0.44 to 1.97 ppm O_3 during any part of gestation. Continuous exposure during mid-gestation increased the resorption of embryos while exposures during late gestation delayed some behavioral developments (e.g., righting, eye opening). There were no effects on neonatal mortality up to 1.5 ppm O_3 , whereas some transient effects on weight gain were observed at exposures of 0.6 ppm O_3 .

More recent studies tend to confirm previous conclusions that prenatal exposures to O_3 concentrations < 1.0 ppm do not cause major or widespread somatic or neurobehavioral effects in

the offspring of laboratory animals. These studies generally add some weight toward a negative interpretation of the importance of contributions of low, ambient O₃ to lower birth weights and gross development defects reported in neonates born to women exposed to typical ambient pollution (e.g., Renner, 2002; Chen et al., 2002; Ritz and Yu, 1999). Some postnatal O₃ exposure studies continue to find a few subtle or borderline somatic and behavioral deficits that will require further research to better assess potential risk to developing humans.

Studies of somatic and neurobehavioral development in female CD-1 mice exposed during pregnancy (days 7 to 17) to O₃ (0, 0.4, 0.8, or 1.2 ppm) failed to show any O₃ effects on reproductive or behavioral performance (Bignami et al., 1994). The study did find significant decreases in body weight gain and delayed eye opening in pups in the 1.2 ppm exposure group. The lack of effect on behavioral performance contrasts with earlier findings, which may be due to the use of different species, differing exposure durations, cross-fostering used in the latter study, different species, and exposure durations during pregnancy. A second study using CD-1 mice exposed in utero from conception through day 17 of pregnancy to 0, 0.2, 0.4, and 0.6 ppm O₃ found no significant deficits in reproductive performance, postnatal somatic and neurobehavioral development, or adult motor activity (Petruzzi et al., 1995). A third study by the same group (Petruzzi et al., 1999), using O₃ exposures (0.3, 0.6, or 0.9 ppm) that continued postnatally until weaning, showed subtle changes in handedness and morphine reactivity. Exposures to 0.6 ppm O₃ caused a reduced preference for the right paw in adulthood. Exposures to 0.9 ppm O₃ altered hot plate avoidance after i.p. treatment with morphine in adulthood.

CD-1 mice exposed to 0.6 ppm O₃ from birth through weaning demonstrated no impairment of navigational performance during acquisition and only subtle changes during reversal (Dell'Omo et al., 1995a). Additionally, there were no O₃-induced effects on reproductive performance, but offspring showed a significant reduction in body weight. Effects on neurobehavioral development with this exposure were minor, with some attenuation of activity responses and impairment of passive avoidance acquisition (Dell'Omo et al. (1995b). The offspring of CD-1 mice continuously exposed from 30 days prior to the formation of breeding pairs until postnatal day 17 to 0.0, 0.3, or 0.6 ppm O₃ showed only small and selective effects on somatic and sensorimotor development (Sorace et al., 2001).

Morphological changes were found in the anterior cerebellar lobe of rat pups born to dams exposed during the entire gestation period to very high (1.0 ppm) O₃ concentrations for 12 h/day

(Rivas-Manzano and Paz, 1999). Additionally, the dams displayed significantly fewer implantations, increased rate of reabsorptions, a high incidence of spontaneous abortion, and offspring with low birth weight, as noted by previous investigators.

5.3.5 Effects on the Liver, Spleen, and Thymus

Early investigations of the effects of O₃ on liver centered on xenobiotic metabolism, and the prolongation of sleeping time, which was observed at 0.1 ppm O₃. In some species, only adults and especially females were affected. In rats, high (1.0 to 2.0 ppm for 3 h) acute O₃ exposures caused increased production of NO by hepatocytes and enhanced protein synthesis (Laskin et al., 1994; 1996). The O₃-associated effects shown in the liver are thought to be mediated by inflammatory cytokines or other cytotoxic mediators released by activated macrophages in the lungs (Vincent et al., 1996; Laskin et al., 1998; Laskin and Laskin, 2001). Except for the earlier work on xenobiotic metabolism, the responses occurred only after very high acute O₃ exposures.

Examinations of the effects of O₃ on spleen and thymus have shown that O₃ primarily affects T-cell mediated systemic immunity. As with the O₃-associated effects shown in the liver, most of the statistically significant changes occurred after acute exposures to very high O₃ concentrations and relate to systemic oxidative stress. Using more relevant ambient urban O₃ exposure patterns, effects were not found on systemic immune function of rats.

5.3.6 Effects on Cutaneous and Ocular Tissues

Ozone exposure not only affects various organ systems, when inhaled, but also has direct effects on the exposed skin and eyes. The outermost layer of the skin (SC, stratum corneum) may be oxidized, which can lead to compromise of the skin barrier and an epidermal pro-inflammatory response (Weber et al., 2001; Thiele, 2001). These effects are found only at very high concentrations (>1 to 5 ppm) and have not been shown at more relevant ambient levels of exposure. The skin possesses a well-developed defense system against oxidative stress, utilizing nonenzymatic (e.g., vitamin C, GSH, UA, α -tocopherol) and enzymatic (e.g., SOD, catalase, GR, and GSHPx) antioxidants (Cross et al., 1998). Ocular tissues have similar antioxidant protective functions as the skin but are not as well studied (Mucke, 1996; Rose et al., 1998). Effects of ground-level smog on the eyes have been reported but generally are attributed to related

photochemical oxidants like peroxyacetyl nitrate (PAN) (Vyskocil et al., 1998) or possibly to atmospheric O₃ precursors or reaction products like aldehydes. As in other tissues, O₃ may have disparate high-dose and low-dose mechanisms of effect on skin and eyes; so, results must be interpreted in this light.

Hairless mice (SKH-1) exposed to O₃ (0.8 to 10 ppm for 2 h) were used to demonstrate that O₃ depletes the low molecular weight antioxidants (e.g., α -tocopherol, vitamin C, GSH, UA) in the SC at ≥ 1.0 ppm and causes increased MDA at ≥ 5 ppm (Weber et al., 1999, 2000, 2001). Valacchi et al. (2000) demonstrated that preexposure to 0.5 O₃ for 2 h followed by low-dose ultraviolet (UV) radiation (0.33 MED) caused depletion of α -tocopherol. This suggests that combined low doses of UV radiation and near-ambient levels of O₃ may cause oxidative stress on the SC. Prolonged exposure to 0.8 ppm O₂ for 6 h also induces cellular stress responses that included the formation of HNE protein adducts, HSP27, and heme-oxygenase-1 in the deeper cellular layers of the skin that continued for up to 18 h after O₃ exposure, followed by repair processes (Valacchi et al., 2003).

The importance of O₃ and UV-induced cellular protein oxidation found in murine skin models to possibly similar environmentally-induced changes in human SC keratins was identified by Thiele et al. (1998, 1999) and Thiele (2001). Using the presence of carbonyl groups in proteins as a marker of reactive oxygen-mediated protein oxidation, they reported higher carbonyl levels in the upper SC from the tanned skin of humans and in the skin of healthy human volunteers exposed to model chemical oxidants (e.g., hypochlorite, benzoyl peroxide) that were inversely correlated with α -tocopherol levels. The environmentally-induced oxidative damage identified in human SC represents an early pathophysiological stage in the development of barrier disruption and inflammation, and possibly has implications for the process of desquamation. The relevance of potentiation of environmental oxidative stress by O₃ exposure of human skin needs further study.

5.3.7 Summary and Conclusions—Systemic Effects of Ozone

Neurobehavioral effects of O₃ at concentrations of 0.2 to 1.0 ppm include decreased motor activity, short- and long-term memory deficits, increased freezing behavior, and decreased exploratory behaviors. These effects have been associated with reactive oxygen/nitrogen species, ozonation products, altered neurotransmitter levels, morphological changes in several

brain regions, and altered EEG patterns during sleep. Neuroendocrine effects of O₃ include morphological and hormonal changes in the pituitary-thyroid-adrenal axis at concentrations of ~0.75 ppm and alterations of visual and olfactory neural pathways at concentrations >1 ppm. Mechanisms underlying these effects are not understood at this time. Cardiovascular effects of O₃ at concentrations of 0.3 to 0.5 ppm include decreased HR, T_{CO}, and BP, which have been termed a hypothermic response. Concentrations of O₃ ≥0.5 ppm cause tissue edema (possibly mediated by ANF).

Prenatal exposures to O₃ concentrations <1.0 ppm did not cause noticeable somatic or neurobehavioral effects in offspring, while concentrations of 1.0 to 1.5 ppm caused varying effects on neonatal mortality. Some studies have shown an effect of O₃ on liver xenobiotic enzymes at concentrations as low as 0.1 ppm, while other studies have shown no alterations in metabolic enzymes at even 1 ppm, with the effects appearing to be highly-species specific. Effects on spleen and thymus appear to only occur at high O₃ concentrations (>1.0 ppm), while relevant ambient urban exposures have no effect on systemic immune function in rats. Effects of O₃ on cutaneous and ocular tissue are only seen at high, nonrelevant concentrations.

5.4 INTERACTIONS OF OZONE WITH OTHER CO-OCCURRING POLLUTANTS

Ozone is part of a complex mixture of air pollutants with a composition and pattern that varies geographically and temporally (by hour of the day, day of the week, and season). Health effects caused by the complex mixture are undoubtedly different (either subtly or significantly) from the additive effects of a few of the hundreds of compounds present. The only disciplinary approach that can evaluate a “real-world” complex mixture is epidemiology (Chapter 7). Still, because of the difficulty in evaluating causative factors and quantitative relationships in epidemiology studies, it is useful to consider animal toxicological studies of mixtures. Such studies can be divided into three categories: (1) ambient air mixtures, (2) laboratory-generated complex mixtures (e.g., gasoline combustion mixtures having UV-irradiation, other reaction mixtures with O₃ and several other components), and (3) binary mixtures. In most cases, experimental designs in the first two classes did not have an O₃-only group, making it difficult or impossible to discern specifically the influence of O₃ per se. The more recent mixture studies

that are discussed here typically have been with NO₂, sulfuric acid (H₂SO₄), or ammonium sulfate ((NH₄)₂SO₄).

Interpreting the mixture studies in terms of real-world risk is difficult because laboratory exposure patterns do not always represent real-world exposure patterns. As shall be seen, all interaction possibilities have occurred, depending upon the composition of the mixture, the endpoint examined, and the exposure regimen. In some cases, no interaction was found. Most often, additivity (the effects of the mixture are equal to the sum of the effects of the individual components) or synergism (the effects of the mixture are greater than the sum of the effects of the individual components) was observed. Antagonism (the effects of the mixture are less than the sum of the individual components) was rarely found.

5.4.1 Ozone and Nitrogen Oxides

The most commonly studied copollutant in binary mixtures with O₃ is NO₂. Both earlier studies and more recent work indicate that, although interaction may occur between these two pollutants, in general, O₃ often masked the effects of the NO₂ or accounted for most of the response, due to the greater toxicity of O₃. Very generally, additivity occurred after acute exposure and synergism occurred with prolonged exposure. Interpreting the O₃/NO₂ mixture studies is challenging, because laboratory exposure patterns rarely simulate real-world exposure patterns. NO₂ typically peaks before O₃, with a mixture occurring between the individual gas peaks, but most laboratory exposures used mixtures only. Also, most studies of O₃ and NO₂ mixtures used ambient levels of O₃ and levels of NO₂ high above ambient. Table AX5-15 lists more recent studies evaluating coexposures to NO₂ and O₃.

Chronic exposures of rats to O₃ (0.8 ppm) and NO₂ (14.4 ppm) for 6 h/day caused development of respiratory insufficiency and severe weight loss. Half of these animals died after 55 to 78 days of exposure due to severe fibrosis (Farman et al., 1997). Increased total lung collagen and elastin were observed, with loss of mature collagen, suggesting breakdown and remodeling of the lung parenchyma. Morphological examination following these coexposures demonstrates a sequence of events starting with increasing inflammatory and mild fibrotic changes for the first 3 weeks of exposure, stabilized or even reduced changes after 4 to 6 weeks, and severe increases over 7 to 9 weeks of exposure (Farman et al., 1999). This suggests that repair processes occurring during the middle 4 to 6 weeks of exposure become overwhelmed,

leading to progressive fibrosis after 7 to 8 weeks of exposure. When the coexposure was extended for 90 days, lesions were noted far into the acinus, but the extent of tissue involvement was the same after 7, 78, and 90 days of exposure. At the end of exposure, high levels of procollagen types I and III mRNA were observed within central acini in the lungs from the combined exposure group but not in lungs from the rats exposed to O₃ or NO₂ alone.

Sprague-Dawley rats exposed to 0.3 ppm O₃ and the combined exposure of O₃ and 1.2 ppm NO₂ for 3 days demonstrated significant DNA single-strand breaks in AMs (Bermúdez et al., 1999). No changes were caused by NO₂-only exposure. The same exposures stimulated the activity of poly (adenosinediphosphate-ribose) (polyADPR) synthetase, suggesting a response to lung cellular DNA repair caused by oxidant-induced lung injury (Bermúdez, 2001). The laboratory animal model of progressive pulmonary fibrosis, utilizing long-term combined O₃ (0.4 to 0.8 ppm) and high-level NO₂ (7 to 14 ppm) exposure, causes an initial acute pulmonary inflammation, followed by adaptation and repair, and eventually causing pulmonary fibrosis after 6 to 13 weeks of exposure (Ishii et al., 2000a; Weller et al., 2000). Unfortunately, this model is not very useful for understanding potential interactive effects of ambient concentrations of O₃ and NO₂.

5.4.2 Ozone and Other Copollutants

Ozone and Formaldehyde

Early studies with combined exposures to O₃ and formaldehyde (HCHO) found evidence of both synergistic and non-interactive effects. Newer work listed in Table AX5-16 includes studies of biochemical and histopathological endpoints in rats exposed to 0.4 ppm O₃ and 3.6 ppm HCHO, alone and combined, for 8 h/day for 3 days (Cassee and Feron, 1994). They demonstrated no interactive effects in the nasal respiratory epithelium, despite the high levels of HCHO when compared to typical ambient levels of 1 to 10 ppb (e.g., Rehle et al., 2001). Mautz (2003) studied changes in breathing pattern and epithelial cell proliferation using exposures of 0.6 ppm O₃ and 10 ppm HCHO alone and in combination for 3 h with exercise at two times resting ventilation. Even with exercise, HCHO does not substantially penetrate to the lower respiratory tract to interact with O₃ and does not alter breathing patterns to modify local O₃ dose. Parenchymal injury was, therefore, due to O₃ alone. In the nasal transitional epithelium and in the trachea, however, combined exposure produced additive effects due to the increased volume

of toxicants during exercise. No other combined pollutant studies have been published in the peer-reviewed literature, although two studies compared the respiratory effects of O₃ to HCHO. Nielsen et al. (1999) compared upper airway sensory irritation caused by HCHO concentrations up to 4 ppm to the lower airway irritation caused by O₃. Using BALB/c mice, they continuously measured f_B , V_T , expiratory flow, time of inspiration, time of expiration, and respiratory patterns during acute, 30-min exposures. They reported a no-effect level of 0.3 ppm for HCHO and 1.0 ppm for O₃.

Thus, O₃ and HCHO do not appear to have additive effects except during exercise and that is due to an increased volume of gas reaching the tissue. Any possible synergism occurs in the nasal epithelium. HCHO exerts its effects primarily in the upper respiratory tract, whereas the primary site of acute cell injury from O₃ occurs in the conducting airways. EPA is currently completing a toxicological and epidemiological review and risk characterization for HCHO.

Ozone and Tobacco Smoke

Early studies of combined exposures of O₃ (1 ppm) and tobacco smoke demonstrated altered airway responsiveness to inhaled bronchoconstrictor challenge and tracheal vascular permeability in guinea pigs. Table AX5-17 lists studies completed since the 1996 AQCD which evaluated tobacco smoke/O₃ coexposures.

Wu et al. (1997) reported that inhalation of cigarette smoke evokes a transient bronchoconstrictive effect in anesthetized guinea pigs. Total pulmonary resistance and C_{dyn} were compared before and after acute exposure to 1.5 ppm O₃ for 1 h. Cigarette smoke alone (7 mL) at a low concentration (33%) induced a mild and reproducible bronchoconstriction that slowly developed and reached its peak after a delay of >1 min. After O₃ exposure, the same cigarette smoke inhalation challenge evoked an intense bronchoconstriction that occurred more rapidly, reaching its peak within 20 s, and being sustained for >2 min. Pretreatment with selective antagonists of neurokinin type 1 and 2 receptors completely blocked the enhanced airway responsiveness suggesting that O₃ exposure induced AHR to inhaled cigarette smoke, primarily from the bronchoconstrictive effect of endogenous tachykinins.

The above studies were conducted with undiluted tobacco smoke and high O₃ concentrations. To determine the effects of aged and diluted sidestream cigarette smoke (ADSS) as a surrogate of environmental tobacco smoke (ETS) on O₃-induced lung injury, Yu et al.

(2002) exposed male B6C3F1 mice to (1) FA, (2) ADSS, (3) O₃, or (4) ADSS followed by O₃ (ADSS/O₃). Exposure to 30 mg/m³ ADSS, 6 h/day for 3 days, followed by exposure to 0.5 ppm O₃ for 24 h was associated with a significant increase in the number of cells recovered by BAL compared with exposure to ADSS alone or O₃ alone. Neutrophils, lymphocytes, and total protein levels in BAL were increased following the combined exposure when compared with all other groups. Within the CAR, the percentage of proliferating cells was unchanged from control following exposure to ADSS alone but was significantly elevated following exposure to O₃ and further augmented in a statistically significant manner in mice exposed to ADSS/O₃. Following exposure to O₃ alone or ADSS/O₃, the ability of AMs to release IL-6 under LPS stimulation was significantly decreased, while exposure to ADSS alone or ADSS/O₃ caused a significantly increased release of TNF- α from AMs under LPS stimulation. These data suggest that ADSS exposure enhances the sensitivity of animals to O₃-induced lung injury.

Acute exposure to ETS also may make a healthy person more susceptible to sequential O₃ exposure by affecting lung barrier function or the underlying epithelium. Toxicological studies with components of ETS (e.g., nicotine receptor agonists, acrolein, and oxidants) have shown that the vagal bronchopulmonary C-fibers are stimulated by acute exposures that initiate both central and local responses (Bonham et al., 2001; Mutoh et al., 2000). The central responses (e.g., tachypnea, cough, bronchoconstriction, increased mucous secretion) are more protective of the lungs; however, local responses may include increased sensitization of the C-fibers to other irritants, including O₃. Active tobacco smokers should not be similarly affected, because they already have significant chronic airway inflammation and increased mucus production. In fact, chronic smokers appear to have diminished lung function responses to O₃ (see Chapter 6).

5.4.3 Complex (Multicomponent) Mixtures Containing Ozone

Ambient pollution in most areas is a complex mix of more than two chemicals. A number of new studies have examined the effects of exposure to multicomponent atmospheres containing O₃. Some of these studies attempted to simulate photochemical reaction products occurring under actual atmospheric conditions. However, the results of these studies are often difficult to interpret because of chemical interactions between the components, as well as the resultant production of variable amounts of numerous secondary reaction products, and a lack of precise control over the ultimate composition of the exposure environment. In addition, the role

of O₃ in the observed biological responses is often obscure. Prior studies using irradiated automobile exhaust mixtures containing total oxidant concentrations (expressed as O₃) in the range of 0.2 to 1.0 ppm have demonstrated pulmonary function changes in several species.

A more recent attempt has been made to examine multicomponent mixtures resulting from the reaction of O₃ with unsaturated hydrocarbons [e.g., isoprene (C₅H₈) and terpene (C₁₀H₁₆)], which produces HCHO, formic acid, acetone, acrolein, acetic acid, and other oxidation products (many of which are strong airway irritants). Wilkins et al. (2001) evaluated sensory irritation by measuring mean f_B in the mouse bioassay and found a 50% reduction after 30 min of exposure to reaction products of O₃ and isoprene. The mixture at this time period contained <0.2 ppm O₃, so the authors attributed the observed effects to the oxidation products. Clausen et al. (2001), using the same mouse model, evaluated the reaction products of O₃ and limonene. A 33% reduction in mean f_B was produced after 30 min of exposure to the mixture containing <0.3 ppm O₃, again implicating the effects of strong irritant products. Further work needs to be done with these complex reaction mixtures, because of their potential impact on the respiratory tract. The results would be particularly important, however, to the reaction of O₃ indoors (see Chapter 3).

Pollutant mixtures containing acid aerosols comprise another type of commonly examined exposure atmosphere (studies summarized in Table AX5-18). Earlier studies that employed simultaneous single, repeated, or continuous exposures of various animal species to mixtures of acid sulfates and O₃ found responses for several endpoints, including tracheobronchial mucociliary clearance, alveolar clearance, pulmonary mechanics, and lung morphology, to be due solely to O₃. Some synergism was noted for bacterial infectivity, response to antigen, and effects on lung protein content and the rate of collagen synthesis.

More recent studies found some differences in airway responses to inhaled acid particle-O₃ mixtures that may have been partly due to airway dosimetry. Various physical and chemical mechanisms may be responsible (see Schlesinger, 1995). For example, physical adsorption or absorption of O₃ or its reaction products on a particle could result in transport to more sensitive sites, or to sites where O₃, by itself, would not normally be reactive (Madden et al., 2000). Chemical reactions on the surface of particles can form secondary products that are more toxicologically active, or chemical characteristics of the particle may change the residence time or reactivity of oxidation products at the site of deposition. The hypothesis that synergism between O₃ and sulfates is due to decreased pH changing the residence time or reactivity of

reactants, such as free radicals, was tested by Chen et al. (1995) and El-Fawal et al. (1995). Male New Zealand White rabbits were exposed for 3 h to 125 $\mu\text{g}/\text{m}^3$ H_2SO_4 , 0.1, 0.3, or 0.6 ppm O_3 , and to combinations. Chen et al. (1995) demonstrated that decreased pH following exposure to acid aerosol was correlated with phagocytic activity and capacity of harvested macrophages and that exposure to $\text{O}_3/\text{H}_2\text{SO}_4$ removed this relationship. El-Fawal et al. (1995) showed that responsiveness of rabbit harvested bronchial rings to ACh was increased following a 3 h O_3 exposure, but that 0.1 to 0.6 ppm $\text{O}_3/0.5$ to $0.125 \text{ mg}/\text{m}^3$ H_2SO_4 combinations resulted in antagonism.

As discussed in Section 5.2.2.1, Churg et al. (1996) demonstrated increased uptake of asbestos or TiO_2 in response to 10 min O_3 (up to 1.0 ppm) preexposure, suggesting that low concentrations of O_3 may increase the penetration of some types of PM into epithelial cells. Using human epithelial cell cultures, Madden et al. (2000) demonstrated a greater potency for ozonized diesel PM to induce PGE_2 production. This suggests that 0.1 ppm O_3 for 24 h can modify the biological activity of PM derived from diesel exhaust.

Effects of combined exposures of O_3 and resuspended urban particles on cell proliferation in epithelial cells of the terminal bronchioles and the alveolar ducts were examined by Vincent et al. (1997) and Adamson et al. (1999). Rats exposed to 0.8 ppm O_3 in combination with 5 or 50 mg/m^3 particles for 4 h demonstrated greatly potentiated proliferative effects compared to O_3 exposure alone. These findings using resuspended dusts, although at high concentrations, are consistent with the studies demonstrating interaction between H_2SO_4 aerosols and O_3 . Effects of acute coexposure to 0.6 ppm O_3 and fine or ultrafine H_2SO_4 (0.5 to $0.3 \text{ mg}/\text{m}^3$) aerosols on lung morphology were examined by Kimmel et al. (1997). They demonstrated that alveolar septal volume was increased in animals coexposed to O_3 and ultrafine, but not fine, H_2SO_4 . Interestingly, cell proliferation was increased only in animals coexposed to fine H_2SO_4 and O_3 , as compared to animals exposed to O_3 alone. Subchronic exposure to acid aerosols (20 to 150 $\mu\text{g}/\text{m}^3$ H_2SO_4) had no interactive effect on the biochemical and morphometric changes produced by either intermittent or continuous exposure to 0.12 to 0.2 ppm O_3 for up to 90 days, which suggests that the interactive effects of O_3 and acid aerosol coexposure in the lung disappeared during the long-term exposure (Last and Pinkerton, 1997). Sindhu et al. (1998) observed an increase in rat lung putrescine levels after repeated, combined exposures to O_3 and a nitric acid (HNO_3) vapor for 40 weeks.

Other studies have examined interactions between carbon particles and O₃. The interactions of intratracheally instilled carbon particles followed by either a 7-day or 60-day exposure to 0.5 ppm O₃ in rats was evaluated by Creutzenberg et al. (1995). The carbon particles caused diminished phagocytotic capacity and chemotactic migration capability of AMs that was stimulated by the subsequent O₃ exposure. Inflammatory responses following exposures to low- and high-concentration mixtures of O₃ and acidic aerosols (0.2 ppm O₃ + 50 µg/m³ carbon + 100 µg/m³ H₂SO₄; 0.4 ppm O₃ + 250 µg/m³ carbon + 500 µg/m³ H₂SO₄, respectively) for 1 or 5 days was examined by Kleinman et al. (1999). The response with the O₃-particle mixture was greater after 5 days (4 h/day) than after day 1. This contrasted with O₃ exposure alone (0.4 ppm), which caused marked inflammation on acute exposure, but no inflammation after 5 consecutive days of exposure.

The effects of a mixture of elemental carbon particles, 0.2 ppm O₃, and 0.5 mg/m³ ammonium bisulfate on rat lung collagen content and macrophage activity was examined by Kleinman et al. (2000). Decreases in lung collagen, and increases in macrophage respiratory burst and phagocytosis were observed relative to other pollutant combinations. Mautz et al. (2001) used a similar mixture (i.e., elemental carbon particles, 0.16 to 0.59 ppm O₃, ammonium bisulfate 0.5 to 0.22 mg/m³, but with 0.11 to 0.39 ppm NO₂ also) and exposure regimen as Kleinman et al. (2000). Also observed were decreases in pulmonary macrophage Fc-receptor binding and phagocytosis and increases in acid phosphatase staining. Bronchoalveolar epithelial permeability and cell proliferation were increased. Altered breathing-patterns also were observed, with some adaptations occurring.

Bolarin et al. (1997) exposed rats to 50 or 100 µg/m³ carbon particles in combination with ammonium bisulfate and 0.2 ppm O₃. Despite 4 weeks of exposure, they observed no changes in protein concentration in BALF or in blood prolyl 4-hydroxylase, an enzyme involved in collagen metabolism. Slight decreases in plasma fibronectin were present in animals exposed to the combined pollutants versus O₃ alone. Thus, the potential for adverse effects in the lungs of animals challenged with a combined exposure to particles and gaseous pollutants is dependent on numerous factors, including the gaseous copollutant, concentration, and time.

In a complex series of studies, Oberdörster and colleagues examined the interaction of several pulmonary oxidative stress pollutants. Elder et al. (2000a,b) reported the results of combined exposure to ultrafine carbon particles (100 µg/m³) and O₃ (1 ppm for 6 h) in young

and old Fischer 344 rats that were pretreated with aerosolized endotoxin. In old rats, exposure to carbon and O₃ produced an interaction that resulted in a greater influx in neutrophils than that produced by either agent alone. This interaction was not seen in young rats. Oxidant release from lavage fluid cells also was assessed and the combination of endotoxin, carbon particles, and O₃ produced an increase in oxidant release in old rats. This mixture produced the opposite response in the cells recovered from the lungs of the young rats, indicating that the lungs of the aged animals underwent greater oxidative stress in response to a complex pollutant mix of particles, O₃, and a biogenic agent. Johnston et al. (2000a, 2002) reported the results of combined exposure to O₃ (1.0 and 2.5 ppm for 4, 20, or 24 h) and low-dose endotoxin, or to O₃ and endotoxin separately, in newborn and adult C57BL/6J mice. In the first study, adult (8 weeks old) mice showed greater sensitivity to O₃ than newborn (36 h old) mice on the basis of mRNAs encoding for various chemokines and cytokines. In contrast, adult and newborn mice responded similarly 2 h after endotoxin exposure (10 ng for 10 min), suggesting that age differences in O₃-generated inflammation is secondary to epithelial cell injury. In the second study, 8-week-old mice exposed to O₃ (1 ppm for 24 h) followed by endotoxin (37.5 EU for 10 min) showed increased responsiveness over either exposure alone, on the basis of increased expression of chemokine and cytokine messages and increased BALF levels of protein and PMNs.

Fanucchi et al. (1998) and Wagner et al. (2001a,b) examined the synergistic effect of coexposure to O₃ and endotoxin on the nasal transitional epithelium of rats that also was mediated, in part, by neutrophils. Fisher 344 rats intranasally instilled with endotoxin and exposed to 0.5 ppm O₃, 8 h per day for 3 days developed mucous cell metaplasia in the nasal transitional epithelium, an area normally devoid of mucous cells; whereas, intratracheal instillation of endotoxin (20 µg) caused mucous cell metaplasia rapidly in the respiratory epithelium of the conducting airways. A synergistic increase of intraepithelial mucosubstances and morphological evidence of mucous cell metaplasia were found in rat maxilloturbinates upon exposure to both O₃ and endotoxin, compared to each pollutant alone. A similar response was reported in OVA-sensitized Brown Norway rats exposed to 0.5 ppm O₃, 8 h/day for 3 days (Wagner et al., 2002), indicating that coexposure to O₃ and inflammatory biogenic substances like allergens (e.g., OVA) or bacterial endotoxin can augment epithelial and inflammatory responses in rat nasal passages.

In follow-up studies, Wagner et al. (2003) reported that coexposure of rats to O₃ and endotoxin also enhanced epithelial and neutrophilic inflammatory responses in the pulmonary airways. Fisher 344 rats were intranasally instilled with endotoxin and exposed to 1.0 ppm O₃ for 8 h, which was repeated 24 h later. Three days after the last exposure, BALF was analyzed for inflammatory cells and secreted mucosubstances (mucin 5AC), and lung tissue was processed for morphometric analysis. Endotoxin instillation alone caused a dose-dependent increase in BALF neutrophils that was further increased 2-fold in O₃-exposed rats given 20 µg endotoxin, the highest dose. Mucin glycoprotein 5AC also was increased in the BALF at this dose but not at lower endotoxin doses. Ozone exposure alone did not cause mucus hypersecretion, but it did potentiate mucus secretion in rats given both 2 and 20 µg endotoxin and increased intraepithelial mucosubstances 2-fold, which was further substantiated by significant increases in mucin gene (rMuc5AC) mRNA levels in the conducting airways.

The effect of O₃ modifying the biological potency of PM (diesel PM and carbon black) was examined by Madden et al. (2000) in rats. Reaction of National Institute of Standards and Technology (NIST) Standard Reference Material # 2975 diesel PM with 0.1 ppm O₃ for 48 h increased the potency (compared to unexposed or air-exposed diesel PM) to induce neutrophil influx, total protein, and LDH in lung lavage fluid in response to intratracheal instillation. Exposure of the diesel PM to high, nonambient O₃ concentration (1.0 ppm) attenuated the increased potency, suggesting destruction of the bioactive reaction products. Unlike the diesel particles, carbon black particles exposed to 0.1 ppm O₃ did not exhibit an increase in biological potency, which suggested that the reaction of organic components of the diesel PM with O₃ were responsible for the increased potency.

Ulrich et al. (2002) investigated the effect of ambient PM from Ottawa Canada (EHC-93) on O₃-induced inflammation. Male Wistar rats were exposed to 0.8 ppm O₃ for 8 h and allowed to recover before intratracheal instillation of 0.5, 1.5, and 5 mg of EHC-93 in 0.3 mL of saline. The high concentrations of PM used were sufficient to induce pulmonary inflammation, which was not exacerbated by preexposure to O₃. Rats from the combined exposure group had higher and more persistent lung lavage protein and albumin levels, as well as increased plasma fibrinogen levels, when compared to PM exposure alone.

The interaction of PM and O₃ was further examined in a murine model of OVA-induced asthma. Kobzik et al. (2001) investigated whether coexposure to inhaled, concentrated ambient

particles (CAPs) from Boston, MA and to O₃ could exacerbate asthma-like symptoms. On days 7 and 14 of life, half of the BALB/c mice used in this study were sensitized by i.p. injection of OVA and then exposed to OVA aerosol on three successive days to create the asthma phenotype. The other half received the i.p. OVA but were exposed to a phosphate-buffered saline aerosol (controls). The mice were further subdivided (n ≥ 61/group) and exposed for 5 h to CAPs, ranging from 63 to 1,569 μg/m³, 0.3 ppm O₃, CAPs + O₃, or to FA. Pulmonary resistance and airway responsiveness to an aerosolized MCh challenge were measured after exposures. A small, statistically significant increase in R_L and airway responsiveness, respectively, was found in both normal and asthmatic mice immediately after exposure to CAPs alone and to CAPs + O₃ but not to O₃ alone or to FA. By 24 h after exposure, the responses returned to baseline levels. There were no significant increases in airway inflammation after any of the pollutant exposures. In this well-designed study of a small-animal model of asthma, O₃ and CAPs did not appear to be synergistic. In further analysis of the data using specific elemental groupings of the CAPs, the acutely increased R_L was found to be associated with the AlSi fraction of PM. Thus, some components of concentrated fine particulate matter (PM_{2.5}) may affect airway caliber in sensitized animals, but the results are difficult to extrapolate to people with asthma.

Animal studies have examined the adverse cardiopulmonary effects of complex mixtures in urban and rural environments of Italy (Gulisano et al., 1997), Spain (Lorz and López, 1997), and Mexico (Vanda et al., 1998; Moss et al., 2001). Some of these studies have taken advantage of the differences in pollutant mixtures of urban and rural environments to report primarily morphological changes in the nasopharynx and lower respiratory tract (Gulisano et al., 1997; Lorz and López, 1997) of lambs and pigeons, respectively, after natural, continuous exposures to ambient pollution. Each study has provided evidence that animals living in urban air pollutants have greater pulmonary changes than those that would occur in a rural, and presumably cleaner, environment. However, these studies either did not report ambient O₃ levels or reported only annual means.

The study by Moss et al. (2001) examined the nasal and lung tissue of rats exposed (23 h/day) to Mexico City air for up to 7 weeks and compared them to controls similarly exposed to FA. No inflammatory or epithelial lesions were found using quantitative morphological techniques; however, the concentrations of pollutants were low. Extrapolation of these results to

humans is restricted, however, by uncontrolled exposure conditions, small sample sizes, and other unknown exposure and nutritional factors in the studies in mammals and birds, and the negative studies in rodents. They also bring up the issue of which species of “sentinel” animals is more useful for predicting urban pollutant effects in humans. Thus, in these field studies, it is difficult to assign a specific role to any specific component of the mixture for the significant cardiopulmonary effects reported.

Similar morphological changes (Calderón-Garcidueñas et al., 2000a, 2001) and chest X-ray evidence of mild lung hyperinflation (Calderón-Garcidueñas et al., 2000b) have been reported in children residing in urban and rural areas of Mexico City. The ambient air in urban areas, particularly in southwestern Mexico City, is a complex mixture of particles and gases, including high concentrations of O₃ and aldehydes that previously have been shown to cause airway inflammation and epithelial lesions in humans (e.g., Calderón-Garcidueñas et al., 1992, 1994, 1996) and laboratory animals (Morgan et al., 1986; Heck et al., 1990; Harkema et al., 1994, 1997a,b). The described effects demonstrate a persistent, ongoing upper and lower airway inflammatory process and chest X-ray abnormalities in children residing predominantly in highly polluted areas. Again, extrapolation of these results to urban populations of the United States is difficult, because of the unique complex mixture of urban air in Mexico City, uncontrolled exposure conditions, and other unknown exposure and nutritional factors.

5.4.4 Summary and Conclusions—Interactions of Ozone with Other Co-Occurring Pollutants

It is difficult to summarize the role that O₃ plays in exposure responses to binary mixtures, and even harder to determine its role in responses to multicomponent, complex atmospheres. Though the specific mechanisms of action of the individual pollutants within a mixture may be known, the exact bases for toxic interactions have not been elucidated clearly. Certain generic mechanisms that may underlie pollutant interactions: (1) physical, involving adsorption of one pollutant onto another and subsequent transport to more or less sensitive sites or to sites where one of the components of the mixture normally would not deposit in concentrated amounts (probably not involved in O₃-related interactions); (2) production of secondary products that may be more toxicologically active than the primary materials, demonstrated or suggested in a number of studies as a basis for interaction between O₃ and NO₂ and between O₃ and PM;

(3) biological or chemical alterations at target sites that affect response to O₃ or the copollutant, H₂SO₄, which has been suggested to underlie interactions with mixtures of O₃ and acid sulfates; (4) O₃- or copollutant-induced physiological change, such as alteration in ventilation pattern, resulting in changes in the penetration or deposition of one pollutant when another is present. This has been implicated in enhanced responses to various O₃-containing mixtures with exercise.

Evaluation of interactions between O₃ and copollutants is a complex task. Responses are dependent on a number of host and environmental factors, such that different studies using the same copollutants may show different types or magnitudes of interactions. The occurrence and nature of any interaction is dependent on the endpoint being examined and is also highly related to the specific conditions of each study, such as animal species, health status, exposure method, dose, exposure sequence, and the physicochemical characteristics of the copollutants. Because of this, it is difficult to compare studies, even those examining similar endpoints, that were performed under different exposure conditions. Thus, any description of interactions is really valid only for the specific conditions of the study in question and cannot be generalized to all conditions of exposure to a particular chemical mixture. Furthermore, it is generally not possible to extrapolate the effect of pollutant mixtures from studies of the effects of each component when given separately. In any case, what can be concluded from the database is that interactions of O₃-containing mixtures are generally synergistic (antagonism has been noted in a few studies), depending on the various factors noted above and that O₃ may produce more significant biological responses as a component of a mixture than when inhaled alone. Furthermore, although most studies have shown that interaction occurs only at higher than ambient concentrations with acute exposure, some have demonstrated interaction at more environmentally relevant levels (e.g., 0.05 to 0.1 ppm O₃ with NO₂) and with repeated exposures.

5.5 EFFECTS OF OTHER PHOTOCHEMICAL OXIDANTS

Peroxyacetyl nitrate and peroxypropionyl nitrate (PPN) are the most abundant non-O₃ oxidants in ambient air of industrialized areas, other than the inorganic nitrogenous oxidants such as NO₂, and possibly HNO₃. Ambient levels of PAN and PPN were reported to be decreasing over the 1990s, and available air quality data (Grosjean et al., 2001; Grosjean, 2003; Jakobi and Fabian, 1997) indicate that present peak concentrations of PAN and PPN in ambient

air from urban areas are in the low parts per billion range (e.g., <1 to 10 ppb). The levels found in nonurban areas are considerably lower (Gaffney et al., 1993).

Reactions occur in the troposphere between O₃ and hydrocarbons (e.g., D-limonene) to produce epoxides, hydroperoxides, and peroxides. The majority of the measured ambient hydroperoxides produced is H₂O₂, although a small amount of organic hydroperoxides (ROOH) also may be formed. Friedlander and Yeh (1998) have estimated that atmospheric aerosols can carry as high as 1 mM of H₂O₂ and ROOH (e.g., hydroxymethylhydroperoxide) in the associated water. In vitro cell and tissue damage are induced by high concentrations of liquid phase H₂O₂ (50 μM to 1 mM). Morio et al. (2001) (see Table AX5-19) demonstrated that a 2 h exposure of 10 and 20 ppb of inhaled H₂O₂ vapor can penetrate into lower lung regions where it causes inflammation. It is likely that hygroscopic components of PM transport ambient H₂O₂ into the lower lung and induce tissue injury as well. Exposure of rats to a H₂O₂-fine particle mixture (215 or 429 μg/m³ (NH₄)SO₄) resulted in increased neutrophil influx, and production of inflammatory mediators by AMs (Morio et al., 2001). Hygroscopic secondary organic aerosols generated by O₃/hydrocarbon reactions and their co-occurrence with H₂O₂ also provide another possible mechanism, yet to be validated, whereby H₂O₂ can be transported into the lower respiratory tract (e.g., Friedlander and Yeh, 1998). Interaction of inhaled O₃ with unsaturated fatty acids on cell membranes and mucus in the airways generates epoxides, hydroperoxides, and secondary ozonation products such as HNE (see Section 5.2.1)

Inhalation toxicological information on the effects of the non-O₃ oxidants has been limited to a few studies on PAN, but at concentrations much higher (approximately 100- to 1,000-fold) than levels typically found in ambient air. Such acute high levels can cause changes in lung morphology, behavioral modifications, weight loss, and susceptibility to pulmonary infections. Therefore, acute toxicity of PAN is much lower than O₃, and it is unlikely that present ambient PAN levels would affect pulmonary function responses to O₃ (reviewed in Vyskocil et al., 1998). Cytogenetic studies indicate that PAN is not a potent mutagen, clastogen, or DNA damaging agent in mammalian cells in vivo or in vitro at concentrations several orders of magnitude higher than those generally encountered in ambient air in most cities (Vyskocil et al., 1998; Kligerman et al., 1995; Heddle et al., 1993). Some studies suggest that PAN may be a weak bacterial mutagen at concentrations much higher than exist in present urban atmospheres (DeMarini et al., 2000; Kleindienst et al., 1990).

An additional level of complexity exists due to the possibility that other ambient oxidants may contribute to effects attributed to O₃. As discussed in Chapter 2, both short-lived radicals and secondary particles containing highly polar compounds are generated in the troposphere by the same photochemical mechanisms that produce O₃. It is plausible that, in addition to the direct effects of O₃, health effects are produced by ambient exposures to these gaseous and particulate secondary compounds. Little is known regarding the composition of these reaction products, and little research has been undertaken evaluating their toxicologic effects. Due to the many oxidizing species present in the atmosphere, interpretation of toxicology data based on O₃ exposures alone have the potential for underestimating health effects of ambient oxidant mixtures.

5.5.1 Summary and Conclusions—Effects of Other Photochemical Oxidants

Concentrations of PAN and PPN (<1 to 10 ppb) in ambient air are unlikely to affect pulmonary function or cause DNA damage. Levels of 10-20 ppm H₂O₂ can penetrate to the lower lung directly or be transported there by PM, where inflammation can result; however, ambient H₂O₂ levels are typically <~5 ppb. As toxicology studies of other photochemical oxidants are rare, quantitative scientific evaluations of possible health effects of environmental exposures cannot be completed at this time.

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6. CONTROLLED HUMAN EXPOSURE STUDIES OF OZONE AND RELATED PHOTOCHEMICAL OXIDANTS

6.1 INTRODUCTION

In the previous chapter, results of ozone (O₃) studies in laboratory animals and in vitro test systems were presented. The extrapolation of results from animal studies is one mechanism by which information on potential adverse human health effects from exposure to O₃ is obtained. More direct evidence of human health effects due to O₃ exposure can be obtained through controlled human exposure studies of volunteers or through field and epidemiologic studies of populations exposed to ambient O₃ (*see Chapter 7*). Controlled human exposure studies typically use fixed concentrations of O₃ under carefully regulated environmental conditions and subject activity levels. This chapter discusses studies in which volunteers were exposed for up to 8 h to O₃ concentrations ranging from 0.04 to 0.75 ppm O₃ while at rest or during varying intensities of exercise.

The majority of controlled human studies have investigated the effects of exposure to O₃ in young nonsmoking healthy adults (18 to 35 years of age) performing continuous exercise (CE) or intermittent exercise (IE). Various combinations of O₃ concentration, exercise routine, and exposure duration have been used in these studies. The responses to ambient O₃ concentrations include decreased inspiratory capacity; mild bronchoconstriction; rapid, shallow breathing patterns during exercise; and symptoms of cough and pain on deep inspiration. Reflex inhibition of inspiration results in a decrease in forced vital capacity (FVC) and total lung capacity (TLC) and, in combination with mild bronchoconstriction, contributes to a decrease in the forced expiratory volume in 1 s (FEV₁). In addition to physiological pulmonary responses and respiratory symptoms, O₃ has been shown to result in airway hyperresponsiveness, epithelial permeability, and inflammation.

The most salient observations from studies reviewed in the 1996 EPA Ozone Air Quality Criteria Document or O₃ AQCD (U.S. Environmental Protection Agency, 1996) were that: (1) young healthy adults exposed to O₃ concentrations ≥ 0.08 ppm develop significant reversible,

transient decrements in pulmonary function if minute ventilation (\dot{V}_E) or duration of exposure is increased sufficiently, (2) children experience similar spirometric responses but lesser symptoms from O₃ exposure relative to young adults, (3) O₃-induced spirometric responses are decreased in the elderly relative to young adults, (4) there is a large degree of intersubject variability in physiologic and symptomatic responses to O₃ but responses tend to be reproducible within a given individual over a period of several months, (5) subjects exposed repeatedly to O₃ for several days develop a tolerance to successive exposures, as demonstrated by an attenuation of responses that is lost after about a week without exposure, and (6) acute O₃ exposure initiates an inflammatory response which may persist for at least 18 to 24 h postexposure.

There are several important limitations associated with these clinical studies: (1) the ability to study only short-term, acute effects; (2) difficulties in trying to link short-term effects with long-term consequences; (3) the use of a small number of volunteers that may not be representative of the general population; and (4) statistical limitations associated with the small sample size. Sample size affects the power of a study, and having a small number of samples causes a risk of Type II error, i.e., the incorrect conclusion that no difference exists between treatments or groups when comparisons are not significantly different. This affects the confidence in estimates of a minimum O₃ concentration at which some degree of pulmonary impairment will occur in both the general population and susceptible subpopulations. As a result, the conclusions drawn from many of the studies cited in this chapter may underestimate the presence of responses at low O₃ concentrations and low activity levels.

Most of the scientific information summarized in this chapter comes from the literature published since the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996). In addition to further study of physiological pulmonary responses and symptoms of breathing discomfort, much of this literature has focused on mechanisms of inflammation and cellular responses to injury induced by O₃ inhalation. A more thorough discussion and review of this literature appears in Annex AX6 of this document. In summarizing the literature, effects are described if they are statistically significant at a probability (p-value) of less than 0.05; otherwise, trends are noted as such.

As spirometry typically *improves* in healthy young adults with exercise exposures to filtered air (FA), the term “O₃-induced” is used herein and in the annex to designate effects that have been corrected for responses during FA exposures. For healthy adults, an O₃-induced

change in lung function is the difference between the *decrement* experienced with O₃ exposure and the *improvement* observed with FA exposure. However, the distinction between an O₃-induced change and a post- versus preexposure change is particularly important in individuals with respiratory disease who may experience exercise-induced *decrements* in pulmonary function during both FA and O₃ exposures. Hence, in subjects with respiratory disease, exercise-induced responses could be mistaken for O₃-induced responses in the absence of a correction for FA responses.

6.2 PULMONARY FUNCTION EFFECTS OF OZONE EXPOSURE IN HEALTHY SUBJECTS

6.2.1 Introduction

As reviewed in the 1986 and 1996 O₃ AQCD's (U.S. Environmental Protection Agency, 1986, 1996), 0.5 ppm is the lowest O₃ concentration at which statistically significant reductions in FVC and FEV₁ have been reported in sedentary subjects. On average, young adults (n = 23; mean age, 22 yrs) exposed at rest for 2 h to 0.5 ppm O₃ had O₃-induced decrements of ~4% in FVC and ~7% in FEV₁ (Folinsbee et al., 1978; Horvath et al., 1979). During exercise, spirometric and symptoms responses are observed at lower O₃ concentrations. For acute exposures of 1 to 2 h to ≥ 0.12 ppm O₃, if \dot{V}_E is sufficiently increased by exercise, healthy human subjects generally experience (a) decreases in TLC, inspiratory capacity (IC), FVC, FEV₁, mean forced expiratory flow from 25% to 75% of FVC (FEF₂₅₋₇₅), and tidal volume (V_T) and (b) increases in specific airways resistance (sRaw), breathing frequency (f_B), and airway responsiveness. These exposures also cause symptoms of cough; pain on deep inspiration; rapid, shallow breathing patterns during exercise; throat irritation; and wheezing. With exposures of 4- to 8-h in duration, statistically significant pulmonary function and symptoms responses are observed at lower O₃ concentrations and lower \dot{V}_E than in shorter duration studies.

A large body of data regarding the interdependent effects of O₃ concentration (C), minute ventilation (\dot{V}_E), and duration of exposure (time, T) on pulmonary responses was assessed in the 1986 and 1996 O₃ AQCD's (U.S. Environmental Protection Agency, 1986, 1996). In an attempt to describe O₃ dose-response characteristics, acute responses were modeled as a function of total

inhaled O₃ dose ($C \times T \times \dot{V}_E$), which was found to be a better predictor of response than O₃ concentration, \dot{V}_E , or T of exposure, alone, or as a combination of any two of these factors. For example, in an analysis of 6 studies with 1- to 2-h exposures to between 0.12 and 0.18 ppm O₃ with exercise, Folinsbee et al. (1988) found a good correlation ($r = 0.81$) between total inhaled O₃ dose and FEV₁ decrements. Other studies, however, reported that O₃ concentration appears to be more important than either \dot{V}_E or T in determining pulmonary responses during acute exposures at constant concentration (Adams et al., 1981, 1987; Folinsbee et al., 1978; Hazucha, 1987; Larsen et al., 1991). When considering variable O₃ concentrations, where response is not a linear function of dose, the use of O₃ dose as a predictor of response becomes more complex, if appropriate at all (e.g., see Section 6.2.4).

6.2.2 Acute Exposure for Up to 2 h

With heavy CE ($\dot{V}_E = 89$ L/min), an O₃-induced decrement of 9.7% in FEV₁ has been reported for healthy young adults ($n = 17$; age, 24 ± 3 yrs) exposed for only 1 h to 0.12 ppm O₃ (Gong et al., 1986). With moderate-to-heavy IE (15 min intervals of rest and exercise [$\dot{V}_E = 68$ L/min]), McDonnell et al. (1983) reported a physiologically small, but significant, O₃-induced decrement of 3.4% in FEV₁ for young healthy adults ($n = 22$, age, 22 ± 3 yrs) exposed for 2 h to 0.12 ppm O₃. Using the same 2 h IE exposure protocol, Linn et al. (1986) found no statistically significant spirometric responses at O₃ concentrations of 0.16 ppm and lower. However, the subjects in the Linn et al. (1986) study were potentially exposed concurrently in Los Angeles to ambient O₃ levels of between 0.12 and 0.16 ppm and were on average 3 yrs older than the subjects in the McDonnell et al. (1983) study. (*The attenuating effects of increasing age and repeated O₃ exposures are discussed in Sections 6.5.1 and 6.6, respectively.*) The disparities between the Linn et al. (1986) and McDonnell et al. (1983) studies demonstrate the difficulty in determining a no-effect-level for O₃ based on relatively small study populations.

Studies analyzing large data sets (≥ 300 subjects) provide better predictive ability of acute changes in FEV₁ at low levels of O₃ and \dot{V}_E than is possible via comparisons between smaller studies. Such an analysis was performed by McDonnell et al. (1997), who examined FEV₁ responses in 485 healthy white males (18 to 36 years of age; subjects recruited from the area

around Chapel Hill, NC) exposed once for 2 h to O₃ concentrations of up to 0.40 ppm at rest or with IE. Decrements in FEV₁ were modeled by sigmoid-shaped curve as a function of subject age, O₃ concentration, \dot{V}_E , and duration of exposure. Regarding applicability to the general population, the McDonnell et al. (1997) model has an apparent limitation of considering only data for young adult white males. However, two other analyses (93 black females; 94 white females; 92 black males; 93 white males, a subset of subjects included in the McDonnell et al. study) found no significant gender or race effects on spirometric responses to O₃ exposure (Seal et al., 1993, 1996). It should be emphasized that the McDonnell et al. (1997) model only predicts average responses. For exposure conditions causing a predicted average FEV₁ decrement of 10%, individual decrements range from approximately 0 to 40% (see Figure 1 of McDonnell et al., 1997). (*The reader is referred to Sections 6.5.1 and 6.5.4 for further discussion of this model and its prediction of age and physical activity effects on FEV₁ responses.*)

In a more recent study, McDonnell et al. (1999) also reported a model predicting average symptom responses from O₃ exposure. Unfortunately, neither of these papers (McDonnell et al., 1997, 1999) provide predictions of intersubject variability in response. Ultman et al. (2004) recently reported pulmonary responses in 60 young healthy nonsmoking adults (32 M, 28 F) exposed to 0.25 ppm O₃ for 1 h with CE at a target \dot{V}_E of 30 L/min. Consistent with findings reported in the 1996 O₃ criteria document, considerable intersubject variability in FEV₁ decrements was reported by Ultman et al. (2004) with responses ranging from a 4% improvement to a 56% decrement. One-third of the subjects had FEV₁ decrements of >15% and 7% of the subjects had decrements of >40%. (*Section 6.4 of this Chapter discusses intersubject variability in response to O₃ exposure.*)

In addition to overt effects of O₃ exposure on the large airways indicated by spirometric responses, O₃ exposure also affects the function of the small airways and parenchymal lung. Foster et al. (1993, 1997) examined the effect of O₃ on ventilation distribution in healthy adult males. In healthy nonsmoking males (26.7 ± 7 years old) exposed to FA or 0.33 ppm O₃ for 2 h with IE, there was a significant reduction in ventilation to the lower lung (31% of lung volume) and significant increases in ventilation to the upper- and middle-lung regions relative to the FA values in 7 of the 9 subjects (Foster et al., 1993). In another study, 15 healthy nonsmoking

males (25.4 ± 2 years old) were exposed to FA or 0.35 ppm O₃ for 2.2 h with IE (Foster et al., 1997). Following O₃ exposure, an inert gas washout was delayed and resembled a two-compartment washout, whereas pre-O₃ exposure a log-linear gas clearance as a function of expired volume resembled a single-compartment washout. The pronounced slow phase of gas washout occurring post-O₃, represented a 24% decrease in the washout rate relative to pre-O₃. At 24-h post-O₃, 6 of the 12 subjects still had (or developed) a delayed washout relative to the pre-O₃ maneuver. This suggests a prolonged O₃ effect on the small airways and ventilation distribution in some individuals.

6.2.3 Prolonged Ozone Exposures

In the exposure range of 0.08 to 0.16 ppm O₃, a number of studies using moderate quasi continuous exercise (QCE; 50 min exercise and 10 min rest per h) for 4 to 8 h have shown significant responses under the following conditions: 0.16 ppm for 4 h with QCE at $\dot{V}_E \approx 40$ L/min (Folinsbee et al., 1994), 0.08 to 0.12 ppm for 6.6 h with QCE at $\dot{V}_E \approx 35$ to 40 L/min (Adams, 2002; Adams, 2003a; Folinsbee et al., 1988; Horstman et al., 1990), and 0.12 ppm for 8 h of IE (30 min per h) at $\dot{V}_E \approx 40$ L/min (Hazucha et al., 1992). Symptoms and spirometric responses increased with duration of exposure, O₃ concentration, and total \dot{V}_E . Airway resistance is only modestly affected with moderate or even heavy exercise combined with O₃ exposure (Folinsbee et al., 1978; McDonnell et al., 1983; Seal et al., 1993).

6.2.3.1 Effect of Exercise Ventilation Rate on FEV₁ Response to 6.6 h Ozone Exposure

It is well established that response to O₃ exposure is a function of \dot{V}_E in studies of 2 h or less in duration (*See Section AX6.2.2*). It is reasonable to expect that response to a prolonged 6.6-h O₃ exposure is also a function of \dot{V}_E , although quantitative analyses are lacking. Data from five similar prolonged exposure studies are available for evaluation of FEV₁ responses as a function of exercise \dot{V}_E (Adams, 2000; Adams and Ollison, 1997; Folinsbee et al., 1988, 1994; Horstman et al., 1990). Each of these studies exposed similarly aged subjects (mean ages 22 to 25 yrs) to 0.12 ppm O₃ for 6.6 h. In total, ten sets of mean FEV₁ decrements were available for exercise \dot{V}_E ranging from 20 to 43 L/min, although no data were available for \dot{V}_E between 20 and 30 L/min (*data illustrated in Figure AX6-2*). As in 2-h exposure studies, FEV₁

decrements are a function of \dot{V}_E in prolonged 6.6-h exposure studies as indicated by a significant correlation between these variables (Pearson, $r = 0.95$, $p < 0.001$; Spearman, $r = 0.84$, $p < 0.01$).

6.2.3.2 Exercise Ventilation Rate as a Function of Body/Lung Size on FEV₁ Response to 6.6 h Ozone Exposure

Based on the assumption that the total inhaled O₃ dose (product of O₃ concentration, exposure duration, and \dot{V}_E) is proportional to the lung size, exercise \dot{V}_E is typically selected to be a multiple of body surface area (BSA) or FVC. Data from several recent studies do not support the contention that \dot{V}_E should be normalized. In an analysis of data from 485 young adults, McDonnell et al. (1997) found that any effect of BSA, height, or baseline FVC on percent decrement in FEV₁ was small to nonexistent. This is consistent with Messineo and Adams (1990), who compared pulmonary function responses in young adult women having small ($n = 14$) or large ($n = 14$) lung sizes (mean FVC of 3.74 and 5.11 L, respectively) and found no significant group difference in FEV₁ decrements. For 30 subjects (15M, 15F) exposed to 0.12 ppm O₃ for 6.6 h, Adams (2000) also reported that FEV₁ responses were more closely related to \dot{V}_E than to \dot{V}_E normalized to BSA. The O₃ dosimetry study of Bush et al. (1996) suggested that normalization of the O₃ dose might more appropriately be a function of anatomic dead space. Ozone penetrates deeper into the lungs of individuals with larger conducting airway volumes, however, differences in FEV₁ responses between subjects exposed for 2 h to 0.25 ppm O₃ ($\dot{V}_E = 30$ L/min) do not appear to be explained by intersubject differences in the fraction of inhaled O₃ retained in the lung (Ultman et al., 2004).

6.2.3.3 Comparison of 2 h IE to 6.6 h O₃ Exposure Effects on Pulmonary Function

Adams (2003b) examined whether prolonged 6.6-h QCE exposure to a relatively low O₃ concentration (0.08 ppm) and a 2-h IE exposure at a relatively high O₃ concentration (0.30 ppm) elicited consistent individual subject FEV₁ responses. Individual subject O₃ exposure reproducibility was first examined via a regression plot of the postexposure FEV₁ response to the 6.6-h chamber exposure as a function of postexposure FEV₁ response to the 2-h IE chamber exposure. The resulting R² of 0.40, although statistically significant, was substantially less than that observed in a comparison of individual FEV₁ response to the two 2-h IE exposures by

chamber and face mask, respectively ($R^2 = 0.83$). The Spearman rank order correlation for the chamber 6.6-h and chamber 2-h exposure comparison was also substantially less (0.49) than that obtained for the two 2-h IE exposures (0.85). The primary reason for the greater variability in the chamber 6.6-h exposure FEV₁ response as a function of that observed for the two 2-h IE exposures is very likely related to the increased variability in response upon repeated exposure to O₃ concentrations lower than 0.18 ppm ($R = 0.57$, compared to a mean R of 0.82 at higher concentrations) reported by McDonnell et al. (1985). This rationale is supported by the lower r^2 (0.40) observed by Adams (2003b) for the FEV₁ responses found in 6.6 h chamber and face mask exposures to 0.08 ppm O₃, compared to an r^2 of 0.83 observed for responses found at 0.30 ppm O₃.

6.2.4 Triangular Ozone Exposures

To further explore the factors that determine responsiveness to O₃, Hazucha et al. (1992) designed a protocol to examine the effect of varying, rather than constant, O₃ concentrations. Subjects were exposed to an O₃ level that increased linearly from 0 to 0.24 ppm for the first 4 h and then decreased linearly from 0.24 to 0 ppm over the second 4 h of the 8 h exposure (triangular concentration profile) and to a constant level exposure of 0.12 ppm O₃ for 8 h. While total inhaled O₃ doses for the constant and the triangular concentration profile were almost identical, the FEV₁ response was dissimilar. For the constant 0.12 ppm O₃ exposure, FEV₁ declined ~5% by the fifth hour and then remained at that level. With the triangular O₃ concentration profile, there was minimal FEV₁ response over the first 3 h followed by a rapid decrease in FEV₁ (-10.3%) over the next 3 h. During the seventh and eighth hours, mean FEV₁ decrements improved to -6.3% as the O₃ concentration decreased from 0.12 to 0.00 ppm (mean = 0.06 ppm). (*The reader is referred to Figure AX6-3 for illustration of FEV₁ responses during square-wave and triangular O₃ exposures*).

Adams (2003a) used a less abrupt triangular O₃ exposure profile (ranging from 0.03 to 0.15 ppm) having an average exposure concentration of only 0.08 ppm, assumed to be typical of outdoor ambient conditions. Postexposure values for FEV₁ and symptoms were not significantly different between the 6.6 h triangular and a square-wave 0.08 ppm O₃ exposure. During the triangular exposure, however, FEV₁ responses became statistically significant after 4.6 h,

whereas they were not significant until 6.6 h during the square wave exposure (Adams, 2003a). Perhaps due to the lower O₃ concentrations, evidence of FEV₁ response recovery with the triangular exposure was less pronounced than as observed by Hazucha et al. (1992). Figure 6-1 illustrates the average O₃-induced FEV₁ responses and the O₃ exposure schemes for the Adams (2003a) and Hazucha et al. (1992) studies.

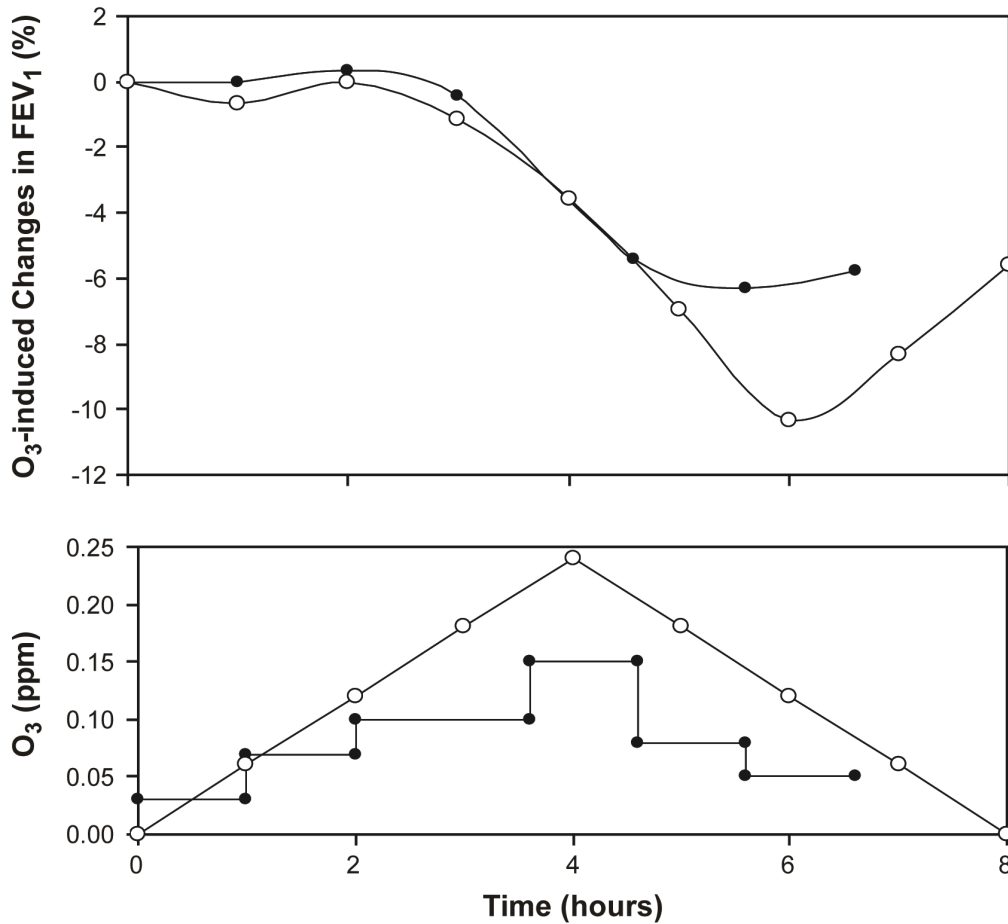


Figure 6-1. Ozone-induced changes in FEV₁ (top panel) and O₃ concentration profiles (bottom panel) as a function of exposure duration. Open (○) and closed (●) circles illustrate average data from Hazucha et al. (1992) and Adams (2003a), respectively. For clarification, the “O₃-induced changes in FEV₁” are the FEV₁ responses following O₃ exposure minus the FEV₁ responses following FA.

In a very recent chamber study, Adams (2006) employed the earlier design of square-wave and triangular exposure profiles. The O₃ concentrations for the 6.6-h exposures were 0.00, 0.06, and 0.08 ppm for square-wave profiles and averaged 0.04, 0.06, and 0.08 ppm for triangular profiles. The 6.6-h postexposure responses from the 0.08 ppm O₃ (average) triangular exposure did not differ significantly from those observed in the 0.08 ppm O₃ square-wave exposure. However, FEV₁ and symptoms were significantly different from preexposure at 4.6 h (when the O₃ concentration was 0.15 ppm) in the triangular exposure, but not until 6.6 h in the square-wave exposure. These observations have confirmed and expanded findings of the earlier study (Adams, 2003a) showing a clear divergence in the peak and hourly responses in FEV₁ between square-wave and triangular exposure profiles at the 0.08 ppm level. However, at the lower O₃ concentration of 0.06 ppm, no temporal pattern differences in FEV₁ responses between square-wave and triangular exposure profiles could be discerned. The author concluded that the results support previous evidence that O₃ concentration has a greater singular effect in the total inhaled O₃ dose than do \dot{V}_E and exposure duration. The author observed no significant differences in FEV₁ and symptom responses compared to FA for the 0.04 and 0.06 ppm exposures.

With square-wave O₃ exposures between 0.08 to 0.12 ppm, FEV₁ decrements may increase with time of exposure (and O₃ dose) or reach a plateau (Horstman et al., 1990; McDonnell et al., 1991). For the triangular exposures used by Hazucha et al. (1992) and Adams (2003a, 2006), maximal FEV₁ responses occurred 1 h to 2 h after peak O₃ concentration and 1 h to 2 h before the maximal O₃ dose occurred (at the end of the O₃ exposure). These three studies suggest that a triangular exposure profile can potentially lead to higher FEV₁ responses than square-wave profiles at overall equivalent O₃ doses.

6.2.5 Mechanisms of Pulmonary Function Responses

Inhalation of O₃ for several hours while physically active elicits both subjective respiratory tract symptoms and acute pathophysiologic changes. The typical symptomatic responses consistently reported in studies are that of airway irritation, cough, and pain on deep inspiration. Depending on the individual's responsiveness to O₃, this can be accompanied by several pathophysiologic changes, e.g., decrements in lung capacities and volumes, bronchoconstriction, airway hyperresponsiveness, airway inflammation, immune system activation, and epithelial injury. The severity of symptoms and the magnitude of response depend on inhaled dose, O₃

sensitivity of an individual, and the extent of tolerance resulting from the individual's previous exposures. The development of effects is time-dependent during both exposure and recovery periods, with considerable overlap of evolving and receding effects. The time sequence, magnitude and the type of responses of this complex series of events, both in terms of development and recovery, indicate that several mechanisms, activated at different times of exposure, must contribute to the overall lung function response (U.S. Environmental Protection Agency, 1996).

Available information on recovery from O₃ exposure indicates that an initial phase of recovery proceeds relatively rapidly, and some 40 to 65% of the acute spirometric and symptom response appears to occur within about 2 h (Folinsbee and Hazucha, 1989). Following a 2-h exposure to 0.4 ppm O₃ with IE, Nightingale et al. (2000) observed a 13.5% decrement in FEV₁. By 3 h postexposure; however, only a 2.7% FEV₁ decrement persisted, as illustrated in Figure 6-2. A similar postexposure recovery in FVC was also observed. Gerrity et al. (1993) suggested that, for healthy young adults, transient increases in mucus clearance (mediated by cholinergic receptors) due to O₃ exposure may be coincident to pulmonary function responses, i.e., the transient increases in clearance and decrements in lung function return to baseline values within 2 to 3 h postexposure. However, there is some indication that the spirometric responses, especially at higher O₃ concentrations, are not fully recovered within 24 h (Folinsbee and Horvath, 1986; Folinsbee et al., 1998). In hyperresponsive individuals, the recovery takes longer, as much as 48 h, to return to baseline values. Collectively, these observations suggest that there is a rapid recovery of O₃-induced spirometric responses and symptoms, which may occur during resting exposure to O₃ (Folinsbee et al., 1977) or as O₃ concentration is reduced during exposure (Hazucha et al., 1992), and a slower phase, which may take at least 24 h to complete (Folinsbee and Hazucha, 2000). Repeated exposure studies at higher concentrations typically show that FEV₁ response to O₃ is enhanced on the second of several days of exposure (Table AX6-8). This enhanced response suggests a residual effect of the previous exposure, about 22 h earlier, even though the preexposure spirometry may be the same as on the previous day. The absence of the enhanced response with repeated exposure at lower O₃ concentrations may be the result of a more complete recovery or less damage to pulmonary tissues (Folinsbee et al., 1994).

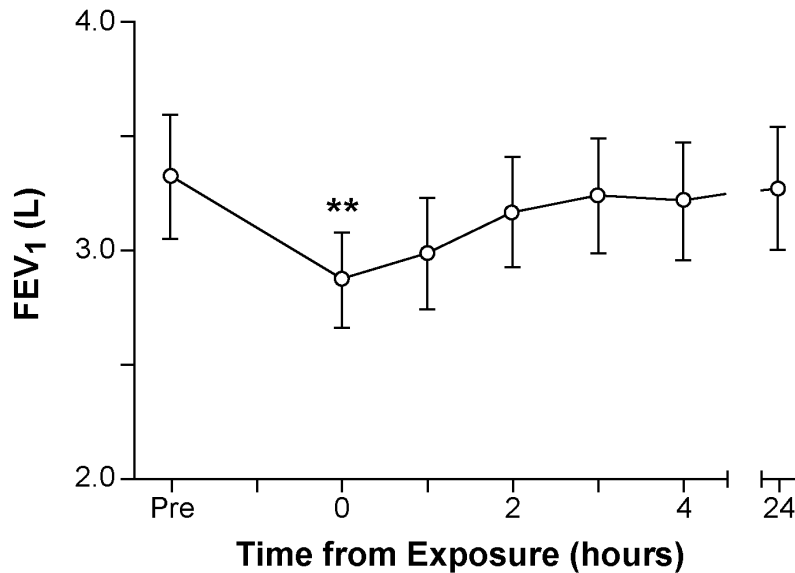


Figure 6-2. Recovery of FEV₁ responses following a 2 h exposure to 0.4 ppm O₃ with IE. Note that the 2 h exposure period is indicated by the Pre to 0 h interval. Immediately postexposure, FEV₁ was significantly (**p < 0.001) decreased. At 3 h postexposure, FEV₁ was at 97% of the preexposure value.

Adapted from Nightingale et al. (2000).

6.2.5.1 Pathophysiologic Mechanisms

Breathing pattern changes

Human studies consistently report that inhalation of O₃ alters the breathing pattern without significantly affecting minute ventilation. A progressive decrease in tidal volume and a “compensatory” increase in frequency of breathing to maintain steady minute ventilation during exposure suggests a direct modulation of ventilatory control. These changes parallel a response of many animal species exposed to O₃ and other lower airway irritants (Tepper et al., 1990). Bronchial C-fibers and rapidly adapting receptors appear to be the primary vagal afferents responsible for O₃-induced changes in ventilatory rate and depth in both humans (Folinsbee and Hazucha, 2000) and animals (Coleridge et al., 1993; Hazucha and Sant’Ambrogio, 1993; Schelegle et al., 1993).

The potential modulation of breathing pattern by activation of sensory afferents located in extrathoracic airways by O₃ has not yet been studied in humans. Nasal-only O₃ exposure of rats produces changes in breathing pattern that are similar to changes observed in humans (Kleinman et al., 1999).

Symptoms and lung function changes

As discussed, in addition to changes in ventilatory control, O₃ inhalation by humans can also induce a variety of symptoms, reduce vital capacity (VC) and related functional measures, and increase airway resistance.

Schelegle et al. (2001) demonstrated that the reduction in VC due to O₃ exposure is a reflex action and not a voluntary termination of inspiration as result of discomfort. They reported that O₃-induced symptom responses (mediated in part by bronchial C-fibers) are substantially reduced by inhaled topical anesthetic. However, the anesthetic had a minor and irregular effect on pulmonary function decrements and tachypnea. Since respiratory symptom responses were largely abolished, these findings support reflex inhibition of VC due to stimulation of both bronchial and pulmonary C-fibers.

The involvement of nociceptive bronchial C-fibers modulated by opioid receptors in limiting maximal inspiration and eliciting subjective symptoms in humans was studied by Passannante et al. (1998). Sufentanil (an opioid agonist and analgesic) rapidly reversed O₃-induced symptom responses and reduced spirometric decrements in “strong” responders. The incomplete recovery in FEV₁ following sufentanil administration, however, suggests involvement of non-opioid receptor modulated mechanisms as well. Interestingly, naloxone (an opioid receptor antagonist) had no significant effect on FEV₁ decrements in “weak” responders. Plasma levels of β -endorphin (a potent pain suppressor) were not related to O₃ responses.

Airway hyperreactivity

In addition to limitation of maximal inspiration and its effects on other spirometric endpoints, activation of airway sensory afferents also plays a role in receptor-mediated bronchoconstriction and an increase in airway resistance. Despite this common mechanism, post-O₃ pulmonary function changes and either early or late airway hyperresponsiveness (AHR)

to inhaled aerosolized methacholine or histamine are poorly correlated either in time or magnitude. Fentanyl and indomethacin, the drugs that have been shown to attenuate O₃-induced lung function decrements in humans, did not prevent induction of AHR when administered to guinea pigs prior to O₃ exposure (Yeadon et al., 1992). Neither does post-O₃ AHR seem to be related to baseline airway responsiveness. These findings imply that the mechanisms are either not related or are activated independently in time. Animal studies (with limited support from human studies) have suggested that an early post-O₃ AHR is, at least in part, vagally mediated (Freed, 1996) and that stimulation of C-fibers can lead to increased responsiveness of bronchial smooth muscle independently of systemic and inflammatory changes which may be even absent (Joad et al., 1996). An in vitro study of isolated human bronchi has shown that O₃-induced airway sensitization involves changes in smooth muscle excitation-contraction coupling (Marthan, 1996). Characteristic O₃-induced inflammatory airway neutrophilia, which at one time was considered a leading AHR mechanism, has been found in a murine model to be only coincidentally associated with AHR, i.e., there was no cause and effect relationship (Zhang et al., 1995). However, this observation does not rule out possible involvement of other cells (such as eosinophils or T-helper cells) in AHR modulation. There is some evidence that release of inflammatory mediators by these cells can sustain AHR and bronchoconstriction. In vitro and animal studies have also suggested that airway neutral endopeptidase activity can be a strong modulator of AHR (Marthan et al., 1996; Yeadon et al., 1992). Late AHR observed in some studies is plausibly due to sustained damage of the airway epithelium and continual release of inflammatory mediators (Foster et al., 2000). Thus, O₃-induced AHR appears to be a product of multiple mechanisms acting at different time periods and levels of the bronchial smooth muscle signaling pathways (*effects of O₃ on AHR are described in Section 6.8*).

6.2.5.2 Mechanisms at a Cellular and Molecular Level

Stimulation of vagal afferents by O₃ and reactive products, the primary mechanism of lung function impairment, is enhanced and sustained by what can be considered in this context to be secondary mechanisms activated at a cellular and molecular level. The complexity of these mechanisms is beyond the scope of this section, and the reader is directed to Section 6.9 of this chapter for more detail. A comprehensive review by Mudway and Kelly (2000) discusses the cellular and molecular mechanisms of O₃-induced pulmonary response in great detail.

Stimulation of bronchial C-fibers by O₃ not only inhibits maximal inspiration but, through local axon reflexes, induces neurogenic inflammation. This pathophysiologic process is characterized by release of tachykinins and other proinflammatory neuropeptides. Ozone exposure has been shown to elevate the C-fiber-associated tachykinin, substance P, in human bronchial lavage fluid (Hazbun et al. 1993) and to deplete neuropeptides synthesized and released from C-fibers in human airway epithelium rich in substance P-immunoreactive axons. Substance P and other transmitters are known to induce granulocyte adhesion and subsequent transposition into the airways, increase vascular permeability and plasma protein extravasation, cause bronchoconstriction, and promote mucus secretion (Solway and Leff, 1991). Although the initial pathways of neurogenic, antigen-induced, and innate immune-mediated inflammation are not the same, they eventually converge, leading to further amplification of airway inflammatory processes by subsequent release of cytokines, eicosanoids, and other mediators. Significantly negative correlations between O₃-induced leukotriene (LTC₄/D₄/E₄) production and spirometric decrements (Hazucha et al., 1996) and an increased level of postexposure PGE₂, a mediator known to stimulate bronchial C-fibers, show that these mediators play an important role in lung function impairment due to O₃ exposure (Mohammed et al., 1993; Hazucha et al., 1996). The reported post O₃ exposure dysfunction of small airways assessed by decrement in FEF₂₅₋₇₅ (Weinman et al., 1995; Frank et al., 2001) is likely induced by both neurogenic and inflammatory mediators, since the density of bronchial C-fibers is much lower in the small than large airways. Also, because of the relative slowness of inflammatory responses as compared to reflex effects, O₃-triggered inflammatory mechanisms are unlikely to initially contribute to progressive lung function reduction. It is plausible, however, that when fully activated, they sustain and possibly further aggravate already impaired lung function. Indeed, a prolonged recovery of residual spirometric decrements following the initial rapid improvement after exposure termination could be due to slowly resolving airway inflammation. Bronchial biopsies performed 6 h postexposure have shown that O₃ caused a significant decrease in immunoreactivity to substance P in the submucosa (Krishna et al., 1997). A strong negative correlation with FEV₁ also suggests that the release of substance P may be an important contributing mechanism to persistent post-O₃ bronchoconstriction (Krishna et al., 1997). Persistent spirometric changes observed for up to 48 h postexposure could plausibly be

sustained by the inflammatory mediators, many of which have bronchoconstrictive properties (Blomberg et al., 1999).

6.3 SUBJECTS WITH PREEXISTING DISEASE

Individuals with respiratory disease are of primary concern in evaluating the health effects of O₃ because even a small change in function is likely to have more impact on a person with reduced reserve, i.e., O₃-induced effects are superimposed on preexisting pulmonary impairment.

6.3.1 Subjects with Chronic Obstructive Pulmonary Disease

For patients with COPD performing light to moderate IE, no decrements in pulmonary function were observed after 1- and 2-h exposures to ≤ 0.30 ppm O₃ (Kehrl et al., 1985; Linn et al., 1982a, 1983a; Solic et al., 1982), and only small decreases in forced expiratory volume were observed for 3-h exposures of chronic bronchitics to 0.41 ppm O₃ (Kulle et al., 1984). More recently, Gong et al. (1997a) found no significant difference in response between age-matched controls and COPD patients to a 4 h exposure to 0.24 ppm O₃ with IE. Although the clinical significance is uncertain, small transient decreases in arterial blood oxygen saturation have also been observed in some of these studies.

6.3.2 Subjects with Asthma

Based on studies reviewed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996), asthmatic subjects appear to be at least as sensitive to acute effects of O₃ as healthy nonasthmatic subjects.

Several recent studies support a tendency for slightly increased spirometric responses in mild asthmatic versus healthy subjects. Alexis et al. (2000) reported reductions in FVC (12%, 10%) and FEV₁ (13%, 11%) for 13 mild asthmatic and 9 healthy subjects, respectively, exposed to 0.4 ppm O₃ for 2 h with IE ($\dot{V}_E = 30$ L/min). The FVC and FEV₁ responses were attenuated by indomethacin in the healthy subjects, but not in the asthmatic subjects. As assessed by the magnitude of reductions in mid-flows (viz. FEF₂₅, FEF₅₀, FEF_{60p}, FEF₇₅) following O₃ exposure, the small airways tended to be more affected in the asthmatic than in the healthy subjects. In a

larger study, Jörres et al. (1996) exposed 24 asthmatics, 12 allergic rhinitics, and 10 healthy subjects to 0.25 ppm O₃ for 3 h with IE ($\dot{V}_E = 30$ L/min). The O₃-induced FEV₁ decrements tended to be greater in the diseased populations (allergic rhinitis, 14.1%; asthmatic, 12.5%; healthy controls, 10.2%). Also, Scannell et al. (1996) exposed 18 asthmatics to 0.2 ppm O₃ for 4 h with IE ($\dot{V}_E \approx 25$ L/min/m² BSA). An O₃-induced increase in sRaw tended to be greater in the asthmatics compared to 81 healthy subjects who underwent similar experimental protocols (Aris et al., 1995; Balmes et al., 1996).

Similar O₃-induced spirometric responses of asthmatic and healthy subjects are suggested by some studies. The Scannell et al. (1996) study of 18 asthmatics reported FEV₁ and FVC decrements that were similar to 81 healthy subjects (Aris et al., 1995; Balmes et al., 1996). Similar group decrements in FEV₁ and FVC were also reported by Hiltermann et al. (1995) for 6 asthmatic and 6 healthy subjects exposed to 0.4 ppm O₃ for 2 h with light IE. Basha et al. (1994) also reported similar spirometric responses between 5 asthmatic and 5 healthy subjects exposed to 0.2 ppm O₃ for 6 h with IE. The lack of significant differences in the Hiltermann et al. (1995) and Basha et al. (1994) studies is not very compelling, however, given the extremely small sample sizes and corresponding lack of statistical power. The Basha et al. (1994) study was also confounded by the asthmatics having an average preexposure FEV₁ that was about 430 mL lower (a 12% difference) on the O₃-exposure day relative to the FA-exposure day.

One other study has reported results suggesting that some asthmatics may have smaller O₃-induced FEV₁ decrements relative to healthy subjects (3% versus 8%, respectively) when exposed to 0.2 ppm O₃ for 2 h with IE (Mudway et al., 2001). However, the asthmatics in the Mudway et al. (2001) study also tended to be older than the healthy subjects, which could partially explain their lesser response.

In a longer exposure duration (7.6 h) study, Horstman et al. (1995) exposed 17 mild-to-moderate asthmatics and 13 healthy controls to 0.16 ppm O₃ or FA with quasi continuous exercise ($\dot{V}_E \approx 30$ L/min). The FEV₁ decrement observed in the asthmatics was significantly larger than in the healthy subjects (19% versus 10%, respectively). There was also a notable tendency for a greater O₃-induced decrease in FEF₂₅₋₇₅ in asthmatics relative to the healthy subjects (24% versus 15%, respectively). A significant positive correlation in asthmatics was

also reported between O₃-induced spirometric responses and baseline lung function, i.e., responses increased with severity of disease.

With repeated O₃ exposures, asthmatic subjects, like healthy subjects (*see Section 6.6*), develop tolerance. Gong et al. (1997b) exposed 10 asthmatics to 0.4 ppm O₃, 3 h per day with IE ($\dot{V}_E \approx 32$ L/min), for 5 consecutive days. Symptom and spirometric responses were greatest on the first (-35 % FEV₁) and second (-34 % FEV₁) exposure days, and progressively diminished toward baseline levels (-6 % FEV₁) by the fifth exposure day. Similar to healthy subjects, asthmatics lose their tolerance to O₃ after 4 to 7 days without O₃ exposure.

Some, but not all, studies have reported that asthmatics have a somewhat exaggerated inflammatory response to acute O₃ exposure relative to healthy control subjects (e.g., McBride et al., 1994; Basha et al., 1994; Peden et al., 1995, 1997; Peden, 2001a; Scannell et al., 1996; Hiltermann et al., 1997, 1999; Michelson et al., 1999; Vagaggini et al., 1999; Newson et al., 2000; Holz et al., 2002) *see also Section 6.9 and Tables AX6-3 and AX6-12*). For example, at 18-h post-O₃ exposure (0.2 ppm, 4-h with IE, $\dot{V}_E = 25$ L/min/m² BSA) and corrected for FA responses, Scannell et al. (1996) found significantly increased neutrophils in 18 asthmatics (12%) compared to 20 healthy subjects (4.5%). This inflammatory response difference was observed despite no group differences in spirometric responses to O₃. Inflammatory responses do not appear to be correlated with lung function responses in either asthmatic or healthy subjects (Balmes et al., 1996, 1997; Holz et al., 1999). The lack of correlation between inflammatory and spirometric responses may be due to differences in the time kinetics of these different types of responses (Stenfors et al., 2002). In addition, airway responsiveness to inhaled allergens is increased by O₃ exposure in subjects with allergic asthma for up to 24 h (*see Section 6.8*).

6.3.3 Subjects with Allergic Rhinitis

Allergic rhinitis is a condition defined by inflammation of the nasal membranes. Nayak (2003) recently reviewed the commonalities between asthma and allergic rhinitis. Clinically, greater than 60% of asthmatics have allergic rhinitis and slightly less than 40% of allergic rhinitics have asthma. Leukotrienes and histamine are well-recognized mediators of responses (*viz.*, inflammation, hyperresponsiveness, and bronchoconstriction) in both asthma and allergic

rhinitis. Although, rhinitis and asthma are distinguished as affecting the upper and lower airways, respectively, it has been suggested that these diseases are manifestations of a similar disease process.

Given the prevalence of concomitant asthma and rhinitis and their common response mediators, it should be expected that allergic rhinitics might respond more similarly to asthmatics than healthy individuals. Regarding spirometric responses, Jörres et al. (1996) provide the only data demonstrating a trend in support of this supposition.

Studies demonstrating the interaction between air pollutants and allergic processes in the human nasal airways and rhinoconjunctival tissue have been reviewed by Peden (2001b) and Riediker et al. (2001), respectively. Ozone exposure of subjects with allergic rhinitis has been shown to induce nasal inflammation and increase airway responsiveness to nonspecific bronchoconstrictors.

Peden et al. (1995), who studied allergic asthmatics exposed to O₃, found that O₃ causes an increased response to nasal allergen challenge in addition to nasal inflammatory responses. Their data suggested that allergic subjects have an increased immediate response to allergen after O₃ exposure. In a follow-up study, Michelson et al. (1999) reported that 0.4 ppm O₃ did not promote early-phase-response mediator release or enhance the response to allergen challenge in the nasal airways of mild, asymptomatic dust mite-sensitive asthmatic subjects. Ozone did, however, promote an inflammatory cell influx, which helps induce a more significant late-phase response in this population.

Jörres et al. (1996) found that O₃ causes an increased response to bronchial allergen challenge in subjects with allergic rhinitis. This study also measured responses in healthy subjects and mildly allergic asthmatics (*see Sections AX6.3.2 and AX6.8*). All subjects were exposed to 0.25 ppm O₃ for 3 h with IE ($\dot{V}_E = 30$ L/min). Statistically significant O₃-induced decrements in FEV₁ occurred in rhinitics (14.1%), asthmatics (12.5%), and the healthy controls (10.2%), but these responses did not differ statistically between groups. Methacholine responsiveness was significantly increased in asthmatics, but not in subjects with allergic rhinitis. Airway responsiveness to an individual's historical allergen (grass and birch pollen, house dust mite, or animal dander) was significantly increased after O₃ exposure when compared

to FA exposure. The authors concluded that subjects with allergic rhinitis, but without asthma, could be at risk if a high O₃ exposure is followed by a high dose of allergen.

Holz et al. (2002) extended the results of Jörres et al. (1996) by demonstrating that repeated daily exposure to lower concentrations of O₃ (0.125 ppm for 4 days) causes an increased response to bronchial allergen challenge in subjects with preexisting allergic airway disease, with or without asthma. These investigators observed no major difference in the pattern of bronchial allergen response between asthmatics or rhinitics, except that the dose of allergen required to elicit a similar response ($\geq 20\%$ decrease in FEV₁) was significantly greater for rhinitic than asthmatic subjects (16 vs. 0.6 mg/mL). Early phase responses were more consistent in subjects with rhinitis, and late-phase responses were more pronounced in subjects with asthma. There also was a tendency towards a greater effect of O₃ in subjects with greater baseline response to specific allergens (chosen on the basis of skin prick test and history, viz., grass, rye, birch, or alder pollen, house dust mite, or animal dander). These data suggest that the presence of allergic bronchial sensitization, but not a history of asthma, may be a key determinant of increased airway allergen responsiveness following exposure to O₃ (*for a more complete discussion of airway responsiveness, see Section AX6.8*).

6.3.4 Subjects with Cardiovascular Disease

Possibly due to the age of subjects studied, O₃ exposure does not appear to result in significant pulmonary function impairment or evidence of cardiovascular strain in patients with cardiovascular disease relative to healthy controls. Gong et al. (1998) exposed 10 hypertensive and 6 healthy adult males, 41 to 78 years of age, to 0.3 ppm O₃ for 3 h with IE at 30 L/min. For all subjects combined (no significant group differences), there was an O₃-induced decrement of 7% in FEV₁ and an 70% increase in the alveolar-arterial oxygen tension gradient. The overall results did not indicate any major acute cardiovascular effects of O₃ in either the hypertensive or normal subjects. Gong et al. (1998) suggested that by impairing alveolar-arterial oxygen transfer, the O₃ exposure could potentially lead to adverse cardiac events by decreasing oxygen supply to the myocardium. However, the subjects in their study apparently had sufficient functional reserve so as to not experience significant ECG changes or myocardial ischemia and/or injury (*see Section 6.10 for additional discussion*).

6.4 INTERSUBJECT VARIABILITY AND REPRODUCIBILITY OF RESPONSE

Analysis of factors that contribute to intersubject variability is important for the understanding of individual responses, mechanisms of response, and health risks associated with acute O₃ exposures. Large intersubject variability in response to O₃ has been reported by numerous investigators (Adams et al., 1981; Aris et al., 1995; Folinsbee et al., 1978; Kulle et al., 1985; McDonnell et al., 1983). The magnitude of individual variability in FEV₁ response in 2 h IE exposures increases at higher O₃ concentrations (Kulle et al., 1985; McDonnell et al., 1983). McDonnell (1996) examined the FEV₁ response data from three 6.6-h exposure studies conducted at the EPA Health Effects Research Laboratory and showed that the FEV₁ responses in FA were small, with most tightly grouped around zero. With increasing O₃ concentrations between 0.08 and 0.12 ppm, the mean response became asymmetrical, with a few individuals experiencing quite large decrements in FEV₁ (*intersubject variability observed in O₃ dosimetry studies is discussed in Chapter 4, Section 4.2*).

As an example of the variation in spirometric responses to O₃ exposure, Hazucha et al. (2003) analyzed the distribution of O₃ responsiveness in 240 subjects (18 to 60 years of age) exposed to 0.42 ppm O₃ (on 3 occasions) for 1.5 h with IE at $\dot{V}_E = 20$ L/min/m² BSA. Across all ages, 18% of subjects were weak responders ($\leq 5\%$ FEV₁ decrement), 39% were moderate responders, and 43% were strong responders ($\geq 15\%$ FEV₁ decrement). Younger subjects (≤ 35 years of age) were predominately strong responders, whereas older subjects (> 35 years of age) were mainly weak responders. The influence of age on intersubject variability was also noted by Passannante et al. (1998), who found that subjects under 35 years of age were more like to be strong responders than older individuals.

For repeated exposures, Hazucha et al. (2003) reported that the reproducibility of FEV₁ responses was related to the length of time between exposures. A Spearman correlation coefficient of 0.54 was found between responses for exposures separated by 105 days (median), whereas a correlation coefficient of 0.85 was found between responses for exposures separated by only 7 days (median). The more reproducible the subject's response, the more precisely it indicates his/her intrinsic responsiveness. In 2 h IE O₃ exposures, McDonnell et al. (1985) found a relatively poor FEV₁ reproducibility (R = 0.58) at the lowest concentration, 0.12 ppm, due in

part to a lack of specific O₃ response or a uniformly small response in the majority of subjects. It was concluded that, for 2 h IE O₃ exposures equal to or greater than 0.18 ppm, the intersubject differences in magnitude of change in FVC and FEV₁ are quite reproducible over time (21 to 385 days; mean = 33 days) and are due primarily to differences in intrinsic responsiveness of individual subjects to O₃ exposure.

Intersubject variability, mechanisms of response, and health risks associated with acute O₃ exposures are complicated by weak associations between various O₃-induced responses. In a retrospective analysis of data for 485 male subjects (aged 18 to 36 yrs) exposed to one of six O₃ concentrations at one of three activity levels for 2 h, McDonnell et al. (1999) found significant, but low, Spearman rank order correlations between FEV₁ response and symptoms of cough (R = 0.39), shortness of breath (R = 0.41), and pain on deep inspiration (R = 0.30). These authors concluded that these different responses are mechanistically related to some degree, but indicated that there does not appear to be a single factor which is responsible for the observed individual differences in O₃ responsiveness across the spectrum of symptom and lung function responses.

The effect of large intersubject variability on the ability to predict individual responsiveness to O₃ was demonstrated by McDonnell et al. (1993). These investigators analyzed data for 290 male subjects (18 to 32 yrs of age) who underwent repeat 2 h IE exposures to one or more O₃ concentrations ranging from 0.12 to 0.40 ppm. They attempted to identify personal characteristics (i.e., age, height, baseline pulmonary function, presence of allergies, and past smoking history) that might predict individual differences in FEV₁ response. Only age contributed significantly to intersubject responsiveness (younger subjects were more responsive), accounting for just 4% of the observed variance. Interestingly, O₃ concentration accounted for only 31% of the variance, strongly suggesting the importance of as yet undefined individual characteristics that determine FEV₁ responsiveness to O₃. The authors concluded that much individual variability in FEV₁ response to O₃ remains unexplained.

6.5 FACTORS MODIFYING RESPONSIVENESS TO OZONE

6.5.1 Influence of Age

Children, adolescents, and young adults (<18 yrs of age) appear, on average, to have nearly equivalent spirometric responses to O₃, but have greater responses than middle-aged and older adults when exposed to comparable O₃ doses (U.S. Environmental Protection Agency, 1996). Symptomatic responses to O₃ exposure, however, appear to increase with age until early adulthood and then gradually decrease with increasing age (U.S. Environmental Protection Agency, 1996). In contrast to young adults, the diminished symptomatic responses in children and the elderly may put the latter groups at increased risk for continued O₃ exposure. Although no new laboratory studies investigating O₃ responses in children have been published since the 1996 O₃ AQCD, the epidemiological studies published during the last decade (*see Section 7.2.3.1 for details*) are generally in agreement with the earlier laboratory studies.

The ensuing discussion in this section provides information on average FEV₁ responses to O₃ exposure as a function of age in healthy adults ranging from 18 to 50 years of age. Beyond approximately 18 years of age, spirometric and symptom responses to O₃ exposure begin to decline with increasing age. In healthy individuals, the rate of decline in O₃ responsiveness appears to be greater in younger (18 to 35 yrs) versus middle aged (35 to 55 yrs) individuals (Passannante et al., 1998; Hazucha et al., 2003). Beyond this age (>55 yrs), acute O₃ exposure elicits minimal spirometric changes. An average FEV₁ decrement of ~3% has been reported by Gong et al. (1997a) for this older population under a “worst case” exposure scenario (0.24 ppm O₃ with 4 h IE). Although Gong et al. (1997a) and others have examined responses to O₃ exposure in subjects of various ages, the exposure conditions differ between most studies in such a manner so that age effects remain uncertain.

Three recent studies, which analyzed large data sets (≥240 subjects) of similarly exposed subjects, show clearly discernable changes in FEV₁ responses to O₃ as a function of age. In one, Seal et al. (1996) analyzed O₃-induced spirometric responses in 371 young nonsmokers (aged 18 to 35 yrs) exposed for 2.3 h during IE at a \dot{V}_E of 25 L/min/m² BSA. On average, for the same O₃ concentration (C), the response of 25-, 30-, and 35-yr old individuals are predicted to be 83, 65, and 48%, respectively, of the response in a 20-yr old. For example, a 5.4% decrement in FEV₁ is predicted for a 20-yr old exposed to 0.12 ppm O₃ for 2.3 h IE ($\dot{V}_E = 25$ L/min/m² BSA), whereas, a similarly exposed 35-yr old is predicted to have only a 2.6% decrement.

McDonnell et al. (1997) examined FEV₁ responses in 485 healthy white males (aged 18 to 36 yrs) exposed once for 2 h to an O₃ concentration of 0.0, 0.12, 0.18, 0.24, 0.30, or 0.40 ppm at rest or one of two levels of IE (\dot{V}_E of 25 and 35 L/min/m² BSA). For the same exposure conditions (C, \dot{V}_E , and duration), the average responses of 25-, 30-, and 35-yr old individuals are predicted to be 69, 48, and 33%, respectively, of the response in 20-yr olds. Hazucha et al. (2003) analyzed the distribution of O₃ responsiveness in 240 subjects (18 to 60 years of age) exposed to 0.42 ppm O₃ for 1.5 h with IE at $\dot{V}_E = 20$ L/min/m² BSA. In males, the FEV₁ responses of 25-, 35-, and 50-yr olds are predicted to be 94, 83, and 50%, respectively, of the average response in 20-yr old males. In females, the FEV₁ responses of 25-, 35-, and 50-yr olds are predicted to be 82, 46, and 18%, respectively, of the average response in 20-yr old females.

For subjects aged 18 to 36 yrs, McDonnell et al. (1999) recently reported that symptom responses from O₃ exposure also decrease with increasing age. Whether the same age-dependent pattern of O₃ sensitivity decline also holds for airway reactivity or inflammatory endpoints has not been determined.

6.5.2 Gender and Hormonal Influences

Several studies have suggested that physiological differences between the genders may predispose females to a greater susceptibility to O₃. Housley et al. (1996) reported that females have lower concentration of uric acid (the most prevalent antioxidant) in nasal lavage fluid than males. Since the source of uric acid (the plasma) is also known to be lower in females, the authors suggested that this could be a contributing factor leading to gender differences in response to oxidants. Plausibly, reduced absorption of O₃ in the upper airways may promote its deeper penetration into the lower respiratory system regions. Dosimetric measurements have shown that the absorption distribution of O₃ is independent of gender when absorption is normalized to anatomical dead space (Bush et al., 1996). More recently, Ultman et al. (2004) reported that the whole lung uptake fraction of O₃ was significantly greater in males (91.4%) than females (87.1%). But, this increase in O₃ uptake in the males was consistent with their larger tidal volume and slower breathing frequency relative to the females. Furthermore, O₃ uptake was not correlated with spirometric responses. Thus, a differential removal of O₃ by uric

acid seems to have minimal effect. In general, the spirometric responses of healthy young females to O₃ exposure appears comparable to the responses of young males (Hazucha et al., 2003). Although, during the follicular phase of the menstrual cycle, lung function response to O₃ is enhanced (Fox et al., 1993).

6.5.3 Racial, Ethnic, and Socioeconomic Status Factors

In the only laboratory study designed to compare spirometric responses of whites and blacks exposed to a range of O₃ concentrations (0 to 0.4 ppm), Seal et al. (1993) reported inconsistent and statistically insignificant FEV₁ differences between white and black males and females within various exposure levels. Perhaps with larger cohorts, the tendency for greater responses of black than white males may become significant. Thus, based on this study it is still unclear if race is a modifier of O₃ sensitivity, although the findings of epidemiologic studies reported in the previous, 1996 O₃ AQCD “can be considered suggestive of an ethnic difference” (U.S. Environmental Protection Agency, 1996).

Gwynn and Thurston (2001) pointed out, it appears that it is more the socioeconomic status (SES) and overall quality of healthcare that drives PM₁₀- and O₃-related hospital admissions than an innate or acquired sensitivity to pollutants. This assertion appears to be supported somewhat by the study of Seal et al. (1996), who employed a family history questionnaire to examine the influence of SES on the O₃ responsiveness. Seal et al. found that, of the three SES categories, individuals in the middle SES category showed greater concentration-dependent decline in % predicted FEV₁ (4 to 5% at 0.4 ppm O₃) than low and high SES groups. The authors did not have an “immediately clear” explanation for this finding. With such a paucity of data, it is not possible to discern the influence of racial or economic-related factors on O₃ sensitivity.

6.5.4 Influence of Physical Activity

Any physical activity will increase minute ventilation and therefore the dose of inhaled O₃. Consequently, the intensity of physiological response following an acute exposure will be strongly associated with minute ventilation (*see Figures 6-3 and AX6-2*).

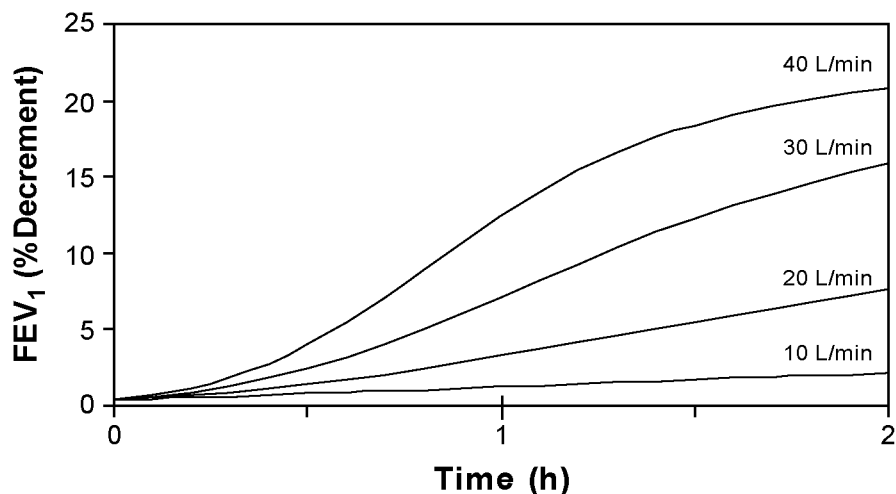


Figure 6-3. Predicted O₃-induced decrements in FEV₁ as a function of exposure duration and level of IE (line labels are \dot{V}_E levels) in healthy young adults (20 yrs of age) exposed to 0.3 ppm O₃. The illustrated activity levels range from rest ($\dot{V}_E = 10$ L/min) to moderate exercise ($\dot{V}_E = 40$ L/min). Predictions are for Model 1 coefficients in Table 3 of McDonnell et al. (1997).

Source: Based on McDonnell et al. (1997).

6.5.5 Environmental Factors

Since the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996) few human laboratory studies have examined the potential influence of environmental factors (e.g., rural versus urban environment, passive cigarette smoke exposure, and bioactive agents such as endotoxin) on O₃-induced pulmonary function changes in healthy individuals.

New controlled human exposure studies have confirmed that smokers are less responsive to O₃ than nonsmokers. Spirometric and plethysmographic pulmonary function decline, nonspecific airway hyperreactivity, and inflammatory response of smokers to O₃ were all weaker than those reported for nonsmokers. Although all these responses are intrinsically related, the functional association between them, as in nonsmokers, has been weak. Similarly, the time course of development and recovery of these effects, as well their reproducibility, was not different from nonsmokers. Chronic airway inflammation with desensitization of bronchial

nerve endings and an increased production of mucus may plausibly explain the smaller responses to O₃ observed in smokers (Frampton et al., 1997; Torres et al., 1997).

The effect of environmental tobacco smoke (ETS) on O₃ responses has received very little attention. In one study (Yu et al., 2002), preexposure of mice to sidestream cigarette smoke (ETS surrogate) elicited no immediate effects, but potentiated subsequent O₃-induced inflammatory responses (*see Chapter 5.4.2 for additional ETS details*). Endotoxin is a biologically active component of both mainstream and sidestream tobacco smoke (Hasday et al., 1999), which might contribute to the potentiation of O₃ effects.

The influence of ambient temperature on pulmonary effects induced by O₃ exposure in humans has been studied infrequently under controlled laboratory conditions. Several experimental human studies have reported additive effects of heat and O₃ exposure (see U.S. Environmental Protection Agency, 1986, 1996). Foster et al. (2000) exposed 9 young healthy subjects for 130 min (IE 10 min at 36 to 39 l/min) to filtered air and to ramp profile O₃ at 22 °C and 30 °C, 45-55% RH. The O₃ exposure started at 0.12 ppm, reached a peak of 0.24 ppm midway through, and subsequently declined to 0.12 ppm at the end of exposure. At the end of exposure, FEV₁ had decreased significantly ($p < 0.5$) by ~8% at 22 °C and ~6.5% at 30 °C. One day (19 h) later, the decline of 2.3% from baseline was still significant ($p < 0.05$) at both temperatures. FVC decrements were smaller and significant only for the 22 °C condition immediately postexposure. There was a decline in specific airway conductance (sGaw; $p < 0.05$) at 30 °C, but not at 22 °C. Nonspecific bronchial responsiveness to methacholine assessed as PC₅₀ sGaw was significantly ($p < 0.05$) higher one day following O₃ exposure at both temperatures, but more so at 30 °C. Thus, these findings suggest that elevated temperature may partially attenuate spirometric responses but enhance airway reactivity.

6.5.6 Oxidant-Antioxidant Balance

The first line of defense against oxidative stress involves antioxidants present in epithelial lining fluid (ELF), which scavenge free radicals and limit lipid peroxidation. Exposure to O₃ depletes the antioxidant level in nasal ELF, probably due to scrubbing of O₃ (Mudway et al., 1999); however, the concentration and the activity of antioxidant enzymes either in ELF or plasma do not appear to be related to O₃ responsiveness (Avissar et al., 2000; Blomberg et al., 1999; Samet et al., 2001). Carefully controlled studies of dietary antioxidant supplementation

have shown some protective effects of α -tocopherol and ascorbate on spirometric lung function from O₃ but not on the intensity of subjective symptoms and inflammatory response, including cell recruitment, activation and release of mediators (Samet et al., 2001; Trenga et al., 2001). Dietary antioxidants have also afforded partial protection to asthmatics by attenuating postexposure bronchial hyperresponsiveness (Trenga et al., 2001). Field studies performed in Mexico City (*described in Section 7.2.3.1*) and animal studies (*described in Section 5.2.1.3*) have also demonstrated protective effects of ELF antioxidants during O₃ exposures.

6.5.7 Genetic Factors

Several recent studies (Bergamaschi et al., 2001; Corradi et al., 2002; David et al., 2003; Romieu et al., 2004; Yang et al., 2005) have reported that genetic polymorphism of antioxidant enzymes and inflammatory genes may modulate pulmonary function and inflammatory response to O₃ challenge. It has been suggested that healthy carriers of NAD(P)H:quinone oxidoreductase wild type (NQO1wt) in combination with glutathione S-transferase μ -1 genetic deficiency (GSTM1null) are more responsive to O₃ (Bergamaschi et al., 2001). The authors reported that subjects having NQO1wt and GSTM1null genotypes had increased O₃ responsiveness (FEV₁ decrements and epithelial permeability), whereas subjects with other combinations of these genotypes were less affected. A subsequent study from the same laboratory reported a positive association between O₃ responsiveness, as characterized by the level of oxidative stress and inflammatory mediators (8-isoprostane, LTB₄ and TBARS) in exhaled breath condensate and the NQO1wt and GSTM1null genotypes (Corradi et al., 2002). However, none of the spirometric lung function endpoints were affected by O₃ exposure. It is of interest to note that human nasal mucosa biopsies of GSTM1null subjects showed higher antioxidant enzymes activity than biopsies of GSTM1 positive individuals when exposed to ozone (Otto-Knapp et al., 2003).

Asthmatic children with a genetic deficiency of GSTM1 were reported to be more responsive to ambient O₃ exposure than asthmatic children without this deficiency, as assessed by decrements in FEF₂₅₋₇₅, in this field study (Romieu et al., 2004). Antioxidant supplementation (vit. C and E) attenuated post-O₃ exposure lung function response in these children. More specific genotyping has shown that O₃ responsiveness of asthmatic children may be related to the presence of variant Ser allele for NQO1. The presence of at least one NQO1 Ser allele in

combination with GSTM1 null genotype lowered the risk of asthma in O₃-exposed asthmatic children relative to Pro/Pro genotype (David et al., 2003).

The influence of functional polymorphism in TNF- α , lymphotoxin- α (LTA), TLR4, SOD2 and GPX1 genes on ozone-induced lung function changes in healthy individuals, mild asthmatics and subjects with rhinitis was varied. Of the inflammatory genes studied, only TNF- α has appeared to show some promise as one of the genetic factors of susceptibility. However, as the authors stated, “the functional significance of individual TNF- α polymorphisms remains controversial” (Yang et al., 2005).

These recent studies have shown that an individual’s innate susceptibility to O₃ may be linked to the genetic background of an individual. Although a number of potential O₃ susceptibility genes have been identified, additional better designed and controlled studies are needed to ascertain the link between susceptibility and specific polymorphism.

6.6 REPEATED O₃ EXPOSURE EFFECTS

Based on studies reviewed here and in the previous O₃ criteria documents (U.S. Environmental Protection Agency, 1986, 1996), several conclusions can be drawn about repeated 1- to 2-h O₃ exposures. Repeated exposures to O₃ can cause an enhanced (i.e., greater) pulmonary function response on the second day of exposure (*see Tables AX6-8 and AX6-9 for added detail*). This enhancement appears to be dependent on the interval between the exposures (24 h is associated with the greatest increase) and is absent with intervals greater than 3 days (Bedi et al., 1985; Folinsbee and Horvath, 1986; Schonfeld et al., 1989). An enhanced response also appears to depend to some extent on the magnitude of the initial response (Horvath et al., 1981). Small responses to the first O₃ exposure are less likely to result in an enhanced response on the second day of O₃ exposure (Folinsbee et al., 1994). With continued daily exposures (i.e., beyond the second day) there is an attenuation of pulmonary function responses, typically after 3 to 5 days of repeated O₃ exposure. This attenuation of responses is lost in 1 week (Kulle et al., 1982; Linn et al., 1982b) or perhaps 2 weeks (Horvath et al., 1981) without O₃ exposure. In temporal conjunction with pulmonary function changes, symptoms induced by O₃ (e.g., cough, pain on deep inspiration, and chest discomfort), are increased on the second exposure day and attenuated with repeated O₃ exposure thereafter (Folinsbee et al., 1980, 1998; Foxcroft and

Adams, 1986; Linn et al., 1982b). O₃-induced changes in airway responsiveness persist longer and attenuate more slowly than pulmonary function and symptoms responses (Dimeo et al., 1981; Kulle et al., 1982), although this has been studied only on a limited basis (Folinsbee et al., 1994). In longer-duration (4 h to 6.6 h), lower-concentration studies that do not cause an enhanced second-day response, the attenuation of response to O₃ appears to proceed more rapidly (Folinsbee et al., 1994) (*Effects of repeated exposures on inflammatory responses are discussed in Section 6.9.4*).

6.7 EFFECTS ON EXERCISE PERFORMANCE

The effects of acute O₃ inhalation on endurance exercise performance have been examined in numerous controlled laboratory studies. These studies were discussed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996) and can be divided into two categories: (1) those that examined the effects of acute O₃ inhalation on maximal oxygen uptake ($\dot{V}O_{2\max}$) and (2) those that examined the effects of acute O₃ inhalation on the ability to complete strenuous continuous exercise protocols of up to 1 h in duration.

In brief, endurance exercise performance and $\dot{V}O_{2\max}$ may be limited by acute exposure to O₃ (Adams and Schelegle, 1983; Schelegle and Adams, 1986; Gong et al., 1986; Foxcroft and Adams, 1986; Folinsbee et al., 1977; Linder et al., 1988). Gong et al. (1986) and Schelegle and Adams (1986) found that significant reductions in maximal endurance exercise performance may occur in well-conditioned athletes while they perform CE ($\dot{V}_E > 80$ L/min) for 1 h at O₃ concentrations ≥ 0.18 ppm. Reports from studies of exposure to O₃ during high-intensity exercise indicate that breathing discomfort associated with maximal ventilation may be an important factor in limiting exercise performance in some, but not all, subjects.

6.8 EFFECTS ON AIRWAY RESPONSIVENESS

Airway hyperresponsiveness refers to a condition in which the propensity for the airways to bronchoconstrict due to a variety of stimuli becomes augmented. Airway responsiveness is typically quantified by measuring the decrement in pulmonary function (i.e., spirometry or

plethysmography) following inhalation of small amounts of an aerosolized specific (antigen, allergen) or nonspecific (methacholine, histamine) bronchoconstrictor agent or some other measured stimulus (e.g., exercise, cold air).

Ozone exposure causes an increase in nonspecific airway responsiveness as indicated by a reduction in the concentration of methacholine or histamine required to produce a given reduction in FEV₁ or increase in SRaw. Increased airway responsiveness is an important consequence of exposure to O₃, because its presence means that the airways are predisposed to narrowing upon inhalation of a variety of stimuli (e.g., specific allergens, SO₂, cold air).

Ozone exposure of asthmatic subjects, who characteristically have increased airway responsiveness at baseline, can cause further increases in responsiveness (Kreit et al., 1989). Similar relative changes in airway responsiveness are seen in asthmatics exposed to O₃ despite their markedly different baseline airway responsiveness. Several studies (Jörres et al., 1996; Kehrl et al., 1999; Molfino et al., 1991) have been published suggesting an increase in specific (i.e., allergen-induced) airway reactivity in response to O₃ exposure. An important aspect of increased airway responsiveness after O₃ exposure is that this represents a plausible link between ambient O₃ exposure and increased hospital admissions for asthma.

Changes in airway responsiveness after O₃ exposure appear to be resolved more slowly than changes in FEV₁ or respiratory symptoms (Folinsbee and Hazucha, 2000). Furthermore, in studies of repeated exposure to O₃, changes in airway responsiveness tend to be somewhat less susceptible to attenuation with consecutive exposures than changes in FEV₁ (Dimeo et al., 1981; Folinsbee et al., 1994; Gong et al., 1997b; Kulle et al., 1982). Ozone-induced increases in airway responsiveness do not appear to be strongly associated with decrements in lung function or increases in symptoms.

Mechanisms underlying O₃-induced increases in airway responsiveness are only partially understood, but such increases appear to be associated with a number of cellular and biochemical changes in airway tissue. Although inflammation could play a role in the increase in airway responsiveness, cyclooxygenase inhibitors have not been effective at blocking the O₃-induced influx of PMNs into bronchoalveolar lavage (BAL) fluid (Hazucha et al., 1996; Ying et al., 1990). Therefore, O₃-induced airway responsiveness may not be due to the presence of PMNs in the airway or to the release of arachidonic acid metabolites. Rather, it seems likely that the mechanism for this response is multifactorial, possibly involving the presence of cytokines,

prostanoids, or neuropeptides; activation of macrophages, eosinophils, or mast cells; and epithelial damage that increases direct access of mediators to the smooth muscle or receptors in the airways that are responsible for reflex bronchoconstriction.

6.9 EFFECTS ON INFLAMMATION AND HOST DEFENSE

6.9.1 Introduction

Short-term exposure of humans to O₃ can cause acute inflammation, and long-term exposure of laboratory animals results in a chronic inflammatory state (*see Chapter 5*). The relationship between repetitive bouts of acute inflammation in humans caused by O₃ and the development of chronic respiratory disease is unknown.

The presence of neutrophils (PMNs) in the lung has long been accepted as a hallmark of inflammation and is an important indicator that O₃ causes inflammation in the lungs. It is apparent, however, that inflammation within airway tissues may persist beyond the point that inflammatory cells are found in BAL fluid (BALF). Soluble mediators of inflammation such as the cytokines (IL-6, IL-8) and arachidonic acid metabolites (e.g., PGE₂, PGF_{2α}, thromboxane, and leukotrienes [LTs] such as LTB₄) have been measured in the BAL fluid of humans exposed to O₃. In addition to their role in inflammation, many of these compounds have bronchoconstrictive properties and may be involved in increased airway responsiveness following O₃ exposure.

Some recent evidence suggests that changes in small airways function may provide a sensitive indicator of O₃ exposure and effect, despite the fact that inherent variability in their measurement by standard spirometric approaches makes their assessment difficult (Frank et al., 2001). Observations of increased functional responsiveness of these areas relative to the more central airways, and of persistent effects following repeated exposure, may indicate that further investigation of inflammatory processes in these regions is warranted.

6.9.2 Inflammatory Responses in the Upper Respiratory Tract

The nasal passages constitute the primary portal for inspired air at rest and, therefore, the first region of the respiratory tract to come in contact with airborne pollutants. Nikasinovic et al. (2003) recently reviewed the literature of laboratory-based nasal inflammatory studies

published since 1985. Nasal lavage (NL) has provided a useful tool for assessing O₃-induced inflammation in the nasopharynx. Increased levels of PMNs in the NL fluid of humans exposed to 0.5 ppm O₃ at rest for 4 h has been reported (Graham et al., 1988; Bascom et al., 1990).

Graham and Koren (1990) compared inflammatory mediators present in both the NL and BAL fluids of humans exposed to 0.4 ppm O₃ for 2 h. Similar increases in PMN were observed in NL and BAL, suggesting a qualitative correlation between inflammatory changes in the lower airways (BAL) and the upper respiratory tract (NL). Torres et al. (1997) compared NL and BAL in smokers and nonsmokers exposed to 0.22 ppm O₃ for 4 h. In contrast to Graham and Koren (1990), they did not find a relationship between numbers or percentages of PMNs in the nose and the lung, perhaps in part due to the variability observed in their NL recoveries. Albumin, a marker of epithelial cell permeability, was increased 18 h later, but not immediately after exposure, as seen by Bascom et al. (1990).

McBride et al. (1994) reported that asthmatic subjects were more sensitive than nonasthmatics to upper airway inflammation at an O₃ concentration (0.24 ppm for 1.5 h with light IE) that did not affect pulmonary function. In the asthmatics, there was a significant increase in the number of PMNs in NL fluid both immediately and 24 h after exposure. Peden et al. (1995) also found that exposure to 0.4 ppm O₃ had a direct nasal inflammatory effect and a priming effect on response to nasal allergen challenge. A subsequent study in dust mite-sensitive asthmatic subjects indicated that O₃ at this concentration enhanced eosinophil influx in response to allergen but did not promote the early-phase nasal response to allergen (Michelson et al., 1999). Similar to observations made in the lower airways, the presence of O₃ molecular “targets” in nasal lining fluid is likely to provide some level of local protection against exposure. In a study of healthy subjects exposed to 0.2 ppm O₃ for 2 h, Mudway and colleagues (1999) observed a significant depletion of uric acid in NL fluid at 1.5 h following exposure.

6.9.3 Inflammatory Response in the Lower Respiratory Tract

As reviewed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996), acute exposure to O₃ results in an inflammatory reaction, increased epithelial cell permeability, and may stimulate fibrogenic processes. Inflammatory markers are observed in BALF of healthy subjects by 1 h post-O₃ exposure and may persist for at least 18 to 24 h. Not all inflammatory

markers, however, follow the same time course. Studies published since the 1996 O₃ AQCD support these earlier findings.

Inflammatory effects have been assessed *in vivo* by lavage (proximal airway and bronchoalveolar), bronchial biopsy, and more recently, induced sputum. *In vitro* studies of human alveolar macrophages (AM) and airway epithelial cells exposed to O₃ suggest that these cell types release mediators found in the BALF of O₃-exposed humans (U.S. Environmental Protection Agency, 1996). Recent evidence suggests that the release of mediators from AMs may be modulated by the products of O₃-induced oxidation of airway lining fluid components, such as human surfactant protein A (Wang et al., 2002).

Spirometric responses to O₃ are independent of inflammatory responses and markers of epithelial injury (Balmes et al., 1996; Blomberg et al., 1999; Hazucha et al., 1996; Torres et al., 1997). Significant inflammatory responses to O₃ exposures that did not elicit significant spirometric responses have been reported (Holz et al., 2005; McBride et al., 1994). A meta-analysis of 21 studies (Mudway and Kelly, 2004) showed that PMN influx in healthy subjects is associated with total O₃ dose (product of O₃ concentration, exposure duration, and \dot{V}_E).

The time course of the inflammatory response to O₃ in humans has not been fully characterized. From a review of the literature by Mudway and Kelly (2000), Figure 6-4 illustrates a plausible time course of acute O₃ responses. As the figure shows, different markers exhibit peak responses at different times. Studies in which lavages were performed 1 h after O₃ exposure (1 h at 0.4 ppm or 4 h at 0.2 ppm) have demonstrated that the inflammatory responses are quickly initiated (Devlin et al., 1996; Schelegle et al., 1991; Torres et al., 1997). Inflammatory mediators and cytokines such as IL-8, IL-6, and PGE₂ are greater at 1 h than at 18 h post-O₃ exposure (Devlin et al., 1996; Torres et al., 1997). However, IL-8 still remain elevated at 18 h post-O₃ (4 h at 0.2 ppm O₃ versus FA) in healthy subjects (Balmes et al., 1996). Schelegle et al. (1991) found increased PMNs in the “proximal airway” lavage at 1, 6, and 24 h after O₃ exposure (4 h at 0.2 ppm O₃), with a peak response at 6 h. Although, at 18 to 24 h after O₃ exposure, PMNs remain elevated relative to 1 h postexposure (Schelegle et al., 1991; Torres et al., 1997). In addition to the influx of PMNs, lymphocyte numbers in BALF are elevated significantly at 6 h following exposure (2 h at 0.2 ppm O₃) of healthy subjects (Blomberg et al., 1997). Flow cytometry also indicated the increased presence of CD3+, CD4+ and CD8+ T cell

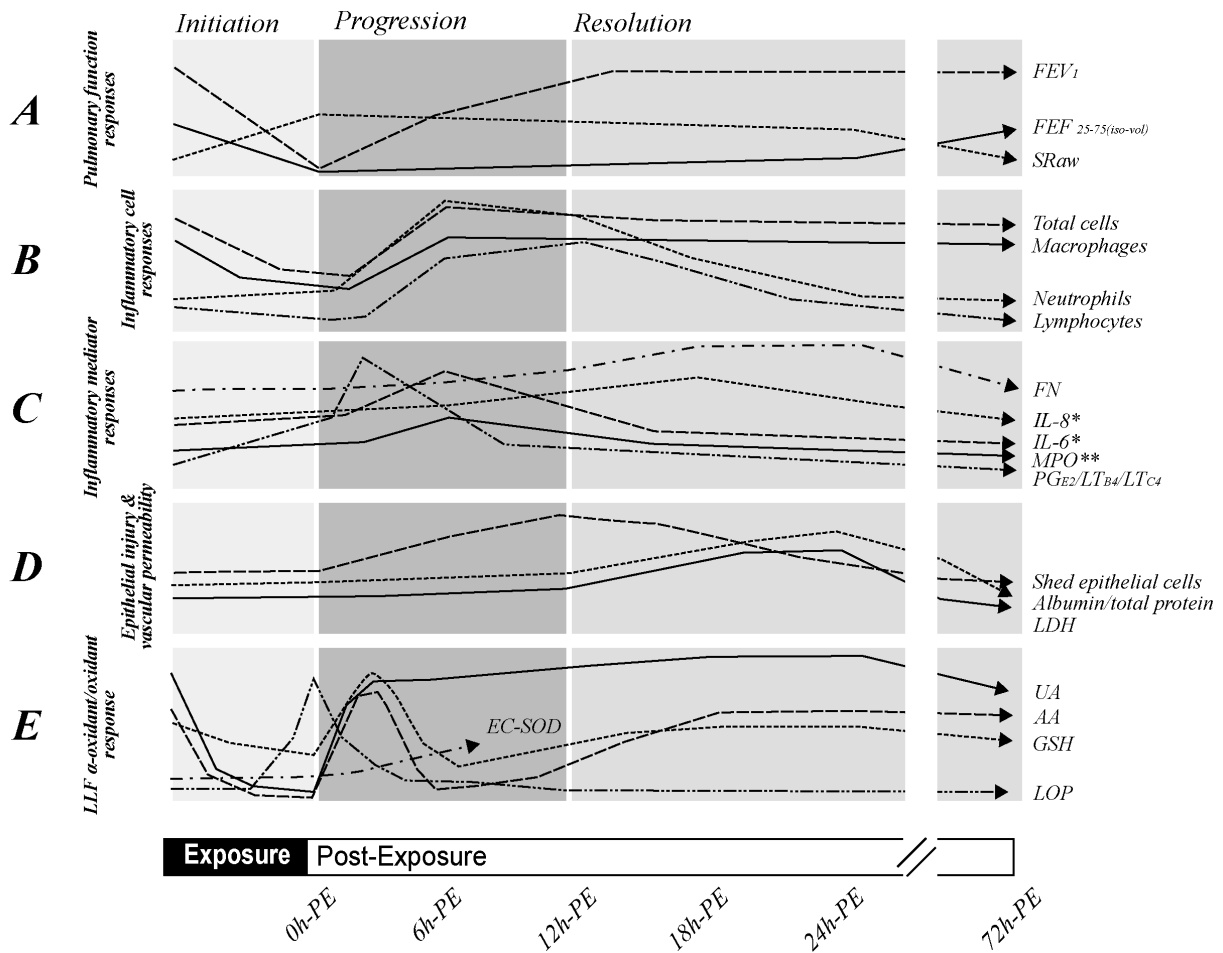


Figure 6-4. Time course of acute responses seen in humans exposed to O₃. Responses are divided into three phases: initiation, during O₃ exposure; progression, where responses develop post-O₃ exposure; and resolution, during which responses return to baseline levels. *The IL-8 response is shown, with a progressive increase peaking at 18 h postexposure (18h-PE). The IL-8 literature is inconclusive and neutrophils have been demonstrated in the absence of an IL-8 increase. **Few studies have measured MPO; however, a trend for an increase at 6h-PE has been reported.

Abbreviations: AA (ascorbic acid), FN (fibronectin), GSH (reduced glutathione), IL (interleukin), LDH (lactate dehydrogenase), LOP (lipid ozonation products), LT_{B4} (leukotriene B4), LT_{C4} (leukotriene C4), MPO (myeloperoxidase), PGE₂ (prostaglandin E2), UA (uric acid).

Source: Reprinted from Mudway and Kelly (2000) with permission from Elsevier.

subsets. This same laboratory later demonstrated that within 1.5 h following exposure of healthy subjects to the same O₃ regimen, expression of human leukocyte antigen (HLA)-DR on lavaged macrophages underwent a significant, 2.5-fold increase (Blomberg et al., 1999).

The inflammatory responses to O₃ exposure have also been studied in asthmatic subjects (Basha et al., 1994; Scannell et al., 1996; Peden et al., 1997). In these studies, asthmatics showed significantly more neutrophils in the BALF (18 h postexposure) than similarly exposed healthy individuals. In one of these studies (Peden et al., 1997), which included only allergic asthmatics who tested positive for *Dematophagoides farinae* antigen, there was an eosinophilic inflammation (2-fold increase), as well as neutrophilic inflammation (3-fold increase). In a study of subjects with intermittent asthma exposed to 0.4 ppm O₃ for 2 h, increases in eosinophil cationic protein, neutrophil elastase and IL-8 were found to be significantly increased 16 h post-exposure and comparable in induced sputum and BALF (Hiltermann et al., 1999). Scannell et al. (1996) also reported that IL-8 tends to be higher in the BALF of asthmatics compared to nonasthmatics following O₃ exposure, suggesting a possible mediator for the significantly increased neutrophilic inflammation in those subjects. Bosson et al. (2003) found significantly greater epithelial expression of IL-5, IL-8, granulocyte-macrophage colony-stimulating factor (GM-CSF) and epithelial cell-derived neutrophil-activating peptide 78 (ENA-78) in asthmatics compared to healthy subjects following exposure to 0.2 ppm O₃ for 2 h. In contrast, Stenfors et al. (2002) did not detect a difference in the O₃-induced increases in neutrophil numbers between 15 mild asthmatic and 15 healthy subjects by bronchial wash at the 6 h postexposure time point. However, the asthmatics were on average 5 years older than the healthy subjects in this study, and it is not yet known how age affects inflammatory responses. It is also possible that the time course of neutrophil influx differs between healthy and asthmatic individuals.

Vagaggini et al. (2002) investigated the effect of prior allergen challenge on responses in mild asthmatics exposed for 2 h to 0.27 ppm O₃ or filtered air. At 6 h postexposure, eosinophil numbers in induced sputum were found to be significantly greater after O₃ than after air exposures. Studies such as these suggest that the time course of eosinophil and neutrophil influx following O₃ exposure can occur at levels detectable within the airway lumen by as early as 6 h. They also suggest that the previous or concurrent activation of proinflammatory pathways within the airway epithelium may enhance the inflammatory effects of O₃. For example, in an *in vitro*

study of epithelial cells from the upper and lower respiratory tract, cytokine production induced by rhinovirus infection was enhanced synergistically by concurrent exposure to O₃ at 0.2 ppm for 3 h (Spannhake et al, 2002).

Although the release of mediators has been demonstrated to occur at exposure times and concentrations that are minimally cytotoxic to airway cells, potentially detrimental latent effects have been demonstrated in the absence of cytotoxicity. These include the generation of DNA single strand breaks (Kozumbo et al., 1996), the loss of cellular replicative activity (Gabrielson et al., 1994) in bronchial epithelial cells exposed *in vitro*, and the formation of protein and DNA adducts. A highly toxic aldehyde formed during O₃-induced lipid peroxidation is 4-hydroxynonenal (HNE). Healthy human subjects exposed to 0.4 ppm O₃ for 1 h underwent BAL 6 h later. Analysis of lavaged AMs by Western Blot indicated increased levels of a 32-kDa HNE-protein adduct, as well as 72-kDa heat shock protein and ferritin in O₃-versus air-exposed subjects (Hamilton et al., 1998). In a recent study of healthy subjects exposed to 0.1 ppm O₃ for 2 h (Corradi et al., 2002), formation of 8-hydroxy-2'-deoxyguanosine (8-OHdG), a biomarker of reactive oxidant species (ROS)-DNA interaction, was measured in peripheral blood lymphocytes. At 18 h post exposure, 8-OHdG was significantly increased in cells compared to preexposure levels, presumably linked to concurrent increases in chemical markers of ROS. Of interest, the increase in 8-OHdG was only significant in a subgroup of subjects with the wild genotype for NQO1 and the null genotype for GSTM1, suggesting that polymorphisms in antioxidant enzymes may confer "susceptibility" to O₃ in some individuals.

The generation of ROS following exposure to O₃ has been shown to be associated with a wide range of responses. In a recent study, ROS production by alveolar macrophages lavaged from subjects exposed to 0.22 ppm for 4 h was assessed by flow cytometry (Voter et al., 2001). Levels were found to be significantly elevated 18 h postexposure and to be associated with several markers of increased permeability. An *in vitro* study of human tracheal epithelial cells exposed to O₃ indicated that generation of ROS resulted in decreased synthesis of the bronchodilatory prostaglandin, PGE₂, as a result of inactivation of prostaglandin endoperoxide G/H synthase 2 (Alpert et al., 1997).

6.9.4 Adaptation of Inflammatory Responses

Physiologic and symptomatic responses in humans following repeated exposure to O₃ were discussed in Section 6.6. Inflammatory responses upon repeated O₃ exposures are discussed in this section. Animal studies suggest that while inflammation may be diminished with repeated exposure, underlying damage to lung epithelial cells continues (Tepper et al., 1989). Markers from BALF following both 2-h (Devlin et al., 1997) and 4-h (Christian et al., 1998; Jörres et al., 2000) repeated O₃ exposures (up to 5 days) indicate that there is ongoing cellular damage irrespective of the attenuation of some cellular inflammatory responses of the airways, pulmonary function, and symptom responses.

Devlin et al. (1997) examined the inflammatory responses of humans repeatedly exposed to 0.4 ppm O₃ for 5 consecutive days. Several indicators of inflammation (e.g., PMN influx, IL-6, PGE₂, fibronectin) were attenuated after 5 days of exposure (i.e., values were not different from FA). Several markers (LDH, IL-8, total protein, epithelial cells) did not show attenuation, indicating that tissue damage probably continues to occur during repeated exposure. The recovery of the inflammatory response occurred for some markers after 10 days, but some responses were not normalized even after 20 days. The continued presence of cellular injury markers indicates a persistent effect that may not necessarily be recognized due to the attenuation of spirometric and symptom responses.

Christian et al. (1998) randomly subjected healthy subjects to a single exposure and to 4 consecutive days of exposure to 0.2 ppm O₃ for 4 h. Both “bronchial” and “alveolar” fractions of the BAL showed decreased numbers of PMNs and fibronectin concentration at day 4 versus the single exposure, and a decrease in IL-6 levels in the alveolar fraction. Following a similar study design and exposure parameters, Jörres et al. (2000) found both functional and BALF cellular responses to O₃ were abolished at 24 h postexposure following the fourth exposure day. However, levels of total protein, IL-6, IL-8, reduced glutathione and ortho-tyrosine were still increased significantly. In addition, visual scores for bronchitis, erythema and the numbers of neutrophils in the mucosal biopsies were increased. Their results indicate that, despite reduction of some markers of inflammation in BALF and measures of large airway function, inflammation within the airways persists following repeated exposure to O₃.

Holz et al. (2002) made a comparison of early and late responses to allergen challenge following O₃ in subjects with allergic rhinitis or allergic asthma. With some variation, both early and late FEV₁ and cellular responses in the two subject groups were significantly enhanced by 4 consecutive days of exposure to 0.125 ppm O₃ for 3 h.

In another study, Frank and colleagues (2001) exposed healthy subjects to FA and to O₃ (0.25 ppm, 2 h) on 4 consecutive days each, with pulmonary function measurements being made prior to and following each exposure. BAL was performed on day 5, 24 h following the last exposure. On day 5, PMN numbers remained significantly higher following O₃ compared to FA. Of particular note in this study was the observation that small airway function, assessed by grouping values for isovolumetric FEF₂₅₋₇₅, Vmax50 and Vmax75 into a single value, showed persistent reduction from day 2 through day 5. These data suggest that techniques monitoring the function in the small peripheral airway regions (the primary sites of O₃ uptake in the lung), may provide important information regarding both acute and cumulative effects of O₃ exposure.

6.9.5 Effect of Anti-Inflammatory and Other Mitigating Agents

Pretreatment of healthy subjects with non-steroidal anti-inflammatory drugs (ibuprofen, etc.) has been found to partially suppress development of airway inflammation and pulmonary function changes (U.S. Environmental Protection Agency, 1996). Although atropine blocked the increase in Raw in response to O₃ exposure, it did not alter the spirometric or symptom responses (Beckett et al., 1985). Similarly, albuterol and salbutamol, which had no effect on O₃-induced changes in spirometry, also had no effect of symptom responses (McKenzie et al., 1987; Gong et al., 1988). The anti-inflammatory medications indomethacin and ibuprofen, which partially inhibit the spirometric responses to O₃ exposure, also cause a reduction in respiratory symptoms (Schelegle et al., 1987; Hazucha et al., 1994). Indomethacin attenuates decrements in FEV₁ and FVC in healthy subjects, but not asthmatics (Alexis et al., 2000). In contrast, inhalation of the corticosteroid budesonide does not prevent or even attenuate O₃-induced responses in healthy subjects as assessed by measurements of lung function, bronchial reactivity and airway inflammation (Nightingale et al., 2000). In asthmatic subjects, budesonide decreases airway neutrophil influx following O₃ exposure (Vagaggini et al., 2001).

Holz et al. (2005) studied inflammatory responses in healthy ozone-responders (>10% increase in sputum neutrophils from O₃) pretreated with single doses (the highest shown to be safe and well tolerated) of inhaled fluticasone and oral prednisolone. The O₃ exposure caused small changes in FEV₁ (-3.6% ± 6.8%) that were not significantly different from baseline or between treatment groups (i.e., prescreening, placebo, fluticasone, and prednisolone). Relative to placebo, the inhaled or oral corticosteroids significantly reduced O₃-induced neutrophil levels. These authors noted that their study design was intended to test the anti-inflammatory effects of the steroids and that such high-dose regimens should not be considered for potential long-term patient treatment.

6.9.6 Changes in Host Defense Capability Following Ozone Exposures

A number of studies clearly show that a single acute exposure (1 to 4 h) of humans to moderate concentrations of O₃ (0.2 to 0.6 ppm) while exercising at moderate to heavy levels results in a number of cellular and biochemical changes in the lung, including an inflammatory response characterized by increased numbers of PMNs, increased permeability of the epithelial cells lining the respiratory tract, cell damage, and production of proinflammatory cytokines and prostaglandins. These responses can be detected as early as 1 h after exposure (Koren et al., 1991; Schelegle et al., 1991) and can persist for at least 18 h (Aris et al., 1993; Koren et al., 1989). The response profile of these mediators is not defined adequately, although it is clear that the time course of response varies for different mediators and cells (Devlin et al., 1997; Schelegle et al., 1991). Following a 2 h exposure to 0.2 ppm O₃, Blomberg et al. (2003) observed increased Clara cell protein (biomarker of epithelial permeability) in blood at 2 h postexposure, which remained high at 6 h and had returned to baseline by 18 h. These changes also occur in humans exposed to 0.08 and 0.10 ppm O₃ for 6 to 8 h (Devlin et al., 1991; Peden et al., 1997). Ozone also causes inflammatory changes in the nose, as indicated by increased levels of PMNs and albumin, a marker for increased epithelial cell permeability. Nasal lavage analyses, however, are not necessarily parallel to BALF analyses.

There appears to be no strong correlation between any of the measured cellular and biochemical changes and changes in lung function measurements, suggesting that different mechanisms may be responsible for these processes (Balmes et al., 1996; Devlin et al., 1991). The idea of different mechanisms is supported by a study in which ibuprofen, a cyclooxygenase

inhibitor, blunted the O₃-induced decrements in lung function without altering the O₃-induced increase in PMNs or epithelial cell permeability (Hazucha et al., 1996). In vitro studies suggest that epithelial cells are the primary target of O₃ in the lung and that O₃ induces them to produce many of the mediators found in the BALF of humans exposed to O₃. Although O₃ does not induce AMs to produce these compounds in large quantities, it does directly impair the ability of AMs to phagocytize and kill microorganisms.

A number of studies have found that O₃ exposures increases epithelial cell permeability through direct (technetium-99m labeled diethylene triamine pentaacetic acid, ^{99m}Tc-DTPA, clearance) and indirect (e.g., increased BALF albumin, protein) techniques. Kehrl et al. (1987) showed increased ^{99m}Tc-DTPA clearance in healthy young adults at 75 minutes postexposure to 0.4 ppm O₃ for 2 h. More recently, Foster and Stetkiewicz (1996) have shown that increased ^{99m}Tc-DTPA clearance persists for at least 18-20 h post-O₃ exposure (130 min to average O₃ concentration of 0.24 ppm), and the effect is greater at the lung apices than at the base. Increased BALF protein, suggesting O₃-induced changes in epithelial permeability, have also been reported at 1 h and 18 h postexposure (Balmes et al., 1996; Devlin et al., 1997). A recent meta-analysis of results from 21 publications (Mudway and Kelly, 2004), showed that increased BALF protein is associated with total ozone dose (product of O₃ concentration, exposure duration, and \dot{V}_E). Changes in permeability associated with acute inflammation may provide increased access of inhaled antigens, particles, and other substances to the smooth muscle, interstitial cells, and the blood.

In addition to affecting epithelial permeability and AM-mediated clearance in the respiratory region of the lung, mucociliary clearance of the tracheobronchial airways is also affected by O₃ exposure. Only two studies (Foster et al., 1987; Gerrity et al., 1993) have investigated the effect of O₃ exposure on mucociliary particle clearance in humans. Foster et al. (1987) measured clearance during and after a 2 h exposure to 0.4 ppm O₃. Gerrity et al. (1993) measured clearance at 2 h postexposure (0.4 ppm O₃), by which time sRaw had returned to baseline and FVC was within 5% of baseline (versus an 11% decrement immediately postexposure). Foster et al. (1987) found a stimulatory effect of acute O₃ exposure on mucociliary clearance. Gerrity et al. (1993), who observed no effect on clearance, suggested that transient clearance increases are coincident to pulmonary function responses. Investigators in

both studies suggested that O₃-induced increases in mucociliary clearance could be mediated by cholinergic receptors.

6.10 EXTRAPULMONARY EFFECTS OF OZONE

Ozone reacts rapidly on contact with respiratory system tissue and is not absorbed or transported to extrapulmonary sites to any significant degree as such. Human exposure studies discussed in the previous O₃ AQCDs (U.S. Environmental Protection Agency, 1986, 1996) failed to demonstrate any consistent extrapulmonary effects. More recently, some human exposure studies have attempted to identify specific markers of exposure to O₃ in blood. Foster et al. (1996) found a reduction in the serum levels of the free radical scavenger α -tocopherol after O₃ exposure. Liu et al. (1997, 1999) used a salicylate metabolite, 2,3, dehydroxybenzoic acid (DHBA), to indicate increased levels of hydroxyl radical which hydroxylates salicylate to DHBA. Increased DHBA levels after exposure to 0.12 and 0.40 ppm suggest that O₃ increases production of hydroxyl radical. The levels of DHBA were correlated with spirometry changes. Relative to preexposure, Corradi et al. (2002) observed increased levels of 8-OHdG (a biomarker ROS-DNA interaction) in peripheral blood lymphocytes of healthy subjects at 18 h postexposure to 0.1 ppm O₃ for 2 h.

Gong et al. (1998) observed a statistically significant O₃-induced increase the alveolar-to-arterial PO₂ gradient in both healthy (n = 6) and hypertensive (n = 10) adult males (aged 41 to 78 yrs) exposed for 3 h with IE ($\dot{V}_E \approx 30$ L/min) to 0.3 ppm O₃. The mechanism for the decrease in arterial oxygen tension in the Gong et al. (1998) study could be due to an O₃-induced ventilation-perfusion mismatch. Foster et al. (1993) has demonstrated that even in relatively young healthy adults (26.7 ± 7 yrs old), O₃ exposure can cause ventilation to shift away from the well perfused basal lung. This effect of O₃ on ventilation distribution (and, by association, the small airways) may persist beyond 24-h postexposure (Foster et al., 1997). Gong et al. (1998) suggested that by impairing alveolar-arterial oxygen transfer, the O₃ exposure could potentially lead to adverse cardiac events by decreasing oxygen supply to the myocardium. The subjects in the Gong et al. (1998) study had sufficient functional reserve so as to not experience significant ECG changes or myocardial ischemia and/or injury.

Effects of O₃ exposure on alveolar-arterial oxygen gradients may be more pronounced in patients with preexisting obstructive lung diseases. Relative to healthy elderly subjects, COPD patients have reduced gas exchange and low SaO₂. Any inflammatory or edematous responses due to O₃ delivered to the well-ventilated regions of the COPD lung could further inhibit gas exchange and reduce oxygen saturation. In addition, O₃-induced vasoconstriction could also acutely induce pulmonary hypertension. Inducing pulmonary vasoconstriction and hypertension in these patients would perhaps worsen their condition, especially if their right ventricular function was already compromised.

6.11 EFFECTS OF OZONE MIXED WITH OTHER POLLUTANTS

Over the past 10 years only a handful of controlled-exposure human studies have examined the effects of pollutant mixtures containing O₃. The studies summarized in this section complement the studies reviewed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996). (*The complexities of O₃ and co-pollutant exposures in animal studies are discussed in Chapter 5, Section 5.4.4.*)

The results of a controlled study on children (Linn et al., 1997), designed to approximate exposure conditions of an epidemiologic study (Neas et al., 1995) by matching the population and exposure atmosphere (0.1 ppm O₃, 0.1 ppm SO₂ and 101 µg/m² h₂SO₄), did not support the findings of this epidemiologic study. The study points out the difficulties in attempting to link the outcomes of epidemiologic and controlled studies. Another vulnerable population, asthmatics, demonstrated enhanced airway reactivity to house dust mite following exposures to O₃, NO₂, and the combination of the two gases. Spirometric response, however, was impaired only by O₃, and O₃ + NO₂ at higher concentrations (Jenkins et al., 1999). Continuous exposure to SO₂ and NO₂ increases inhaled bolus O₃ absorption, while continuous exposure to O₃ decreases O₃ bolus absorption (Rigas et al., 1997). Inhalation of a mixture of PM_{2.5} and O₃ by healthy subjects increased brachial artery tone and reactivity (Brook et al., 2002). Since no other cardiovascular endpoints were affected by the exposure, the pathophysiological importance of this observation remains uncertain. However, acute pulmonary hypertension due to O₃-induced vasoconstriction could pose a risk to individuals with cardiovascular disease (*see Section 6-10*).

All in all, the contention that air pollutant mixtures elicit stronger pathophysiologic effects than individual pollutants of the mix is only weakly supported by human studies of either healthy or at-risk individuals.

6.12 CONTROLLED STUDIES OF AMBIENT AIR EXPOSURES

A large amount of informative O₃ exposure-effects data has been obtained in controlled exposure studies of humans or laboratory animals under a variety of different experimental conditions. However, laboratory simulation of the variable pollutant mixtures present in ambient air is not practical. Thus, the exposure effects of one or several artificially generated pollutants (i.e., a simple mixture) on pulmonary function and symptoms may not fully explain responses to ambient air exposures where complex pollutant mixtures exist.

6.12.1 Mobile Laboratory Studies

Quantitatively useful information on the effects of acute exposure to photochemical oxidants on pulmonary function responses and symptoms derived from field studies using a mobile laboratory were presented in prior criteria documents (U.S. Environmental Protection Agency, 1986, 1996). Relative to controlled exposure studies, mobile laboratory ambient air studies suffer an additional limitation in terms of a dependence on outdoor ambient conditions. Consistent with controlled exposure studies, mobile studies in California demonstrated that pulmonary effects from exposure to ambient air in Los Angeles are related to O₃ concentration and level of exercise. Healthy subjects with a history of allergy also appeared to be more responsive to O₃ than “nonallergic” subjects (Linn et al., 1980, 1983b), although a standardized evaluation of atopic status was not performed.

6.13 SUMMARY

Responses in humans exposed to ambient O₃ concentrations include: decreased inspiratory capacity; mild bronchoconstriction; rapid, shallow breathing pattern during exercise; and symptoms of cough and pain on deep inspiration. Reflex inhibition of inspiration results in a decrease in forced vital capacity (FVC) and, in combination with mild bronchoconstriction,

contributes to a decrease in the forced expiratory volume in 1 s (FEV_1). In addition to physiological pulmonary responses and symptoms of breathing discomfort, O_3 exposure also results in airway hyperresponsiveness, inflammation, immune system activation, and epithelial injury. With repeated O_3 exposures over several days, spirometric and symptom responses become attenuated, but this tolerance is lost after about a week without exposure. Airway responsiveness also appears to be attenuated with repeated O_3 exposures, but less so than FEV_1 . Unlike spirometric and symptom responses, airway inflammation and small airways dysfunction may not become attenuated by repeated O_3 exposures.

Healthy young adults exposed to O_3 concentrations ≥ 0.08 ppm develop significant reversible, transient decrements in pulmonary function if minute ventilation (\dot{V}_E) or duration of exposure are increased sufficiently. The pattern of FEV_1 response appears to depend on the O_3 exposure profile. Triangular exposure profiles can potentially lead to greater FEV_1 responses than square wave exposures at equivalent average O_3 doses. O_3 -induced decrements in FEV_1 do not appear to depend on gender, race, body surface area, height, lung size, or baseline FVC in healthy young adults. Healthy children experience similar spirometric responses but lesser symptoms from O_3 exposure relative to young adults. On average, spirometric and symptom responses to O_3 exposure appear to decline with increasing age beyond about 18 years of age. There is a large degree of intersubject variability in physiologic and symptomatic responses of healthy adults exposed to O_3 . However, responses tend to be reproducible within a given individual over a period of several months. With increasing O_3 concentration, the distribution of FEV_1 decrements becomes asymmetrical, with a few individuals experiencing large decrements.

There is a tendency for slightly increased spirometric responses in mild asthmatics and allergic rhinitics relative to healthy young adults. Spirometric responses in asthmatics appear to be affected by baseline lung function, i.e., responses increase with disease severity. With repeated daily O_3 exposures, spirometric responses of asthmatics become attenuated; however, airway responsiveness becomes increased in subjects with preexisting allergic airway disease (with or without asthma). Possibly due to patient age, O_3 exposure does not appear to cause significant pulmonary function impairment or evidence of cardiovascular strain in patients with cardiovascular disease or chronic obstructive pulmonary disease relative to healthy subjects.

Available information on recovery from O_3 exposure indicates that an initial phase of recovery in healthy individuals proceeds relatively rapidly, with acute spirometric and symptom

responses resolving within about 2 to 4 h. Small residual lung function effects are almost completely resolved within 24 h. Effects of O₃ on the small airways, assessed by persistent decrement in FEF₂₅₋₇₅ and altered ventilation distribution, may be due in part to inflammation. Indeed, a prolonged recovery of residual spirometric decrements following the initial rapid recovery could be due to slowly resolving airway inflammation. In hyperresponsive individuals, this recovery takes longer (as much as 48 hours) to return to baseline values. Persistent spirometry changes observed for up to 48 h postexposure could plausibly be sustained by the inflammatory mediators. Cellular responses (e.g., release of immunomodulatory cytokines) appear to still be active as late as 20 h postexposure. More slowly developing inflammatory and cellular changes may persist for up to 48 h, but the time course for these parameters in humans has not been explored fully.

Soluble mediators of inflammation such as the cytokines (IL-6, IL-8) and arachidonic acid metabolites (e.g., PGE₂, PGF_{2α}, thromboxane, and leukotrienes [LTs] such as LTB₄) have been measured in the BAL fluid of humans exposed to O₃. Many of these compounds have bronchoconstrictive properties and may be involved in increased airway responsiveness following O₃ exposure. Some indicators of inflammation (e.g., PMN influx, IL-6, PGE₂, fibronectin) are attenuated with repeated O₃ exposures. However, other markers (LDH, IL-8, total protein, epithelial cells) do not show attenuation, thus indicating that tissue damage probably continues to occur during repeated O₃ exposure. There appears to be no strong correlation between any of the measured cellular and biochemical changes and changes in lung function measurements. A limited number of studies suggest that inflammatory responses may be detected following O₃ exposures that are insufficient to cause decrements in pulmonary function. Whether airway reactivity or inflammatory responses to O₃ are dependent on the age of the exposed individual, such as spirometric responses, has not been determined.

Dietary antioxidant supplementation attenuates O₃-induced spirometric responses, but not the intensity of subjective symptoms nor inflammatory responses. Dietary antioxidants also appear to afford partial protection to asthmatics by attenuating postexposure bronchial hyperresponsiveness.

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7. EPIDEMIOLOGIC STUDIES OF HUMAN HEALTH EFFECTS ASSOCIATED WITH AMBIENT OZONE EXPOSURE

7.1 INTRODUCTION

This chapter evaluates current epidemiologic literature on health and physiological effects of ambient ozone (O₃) exposure. Epidemiologic studies linking community ambient O₃ concentrations to health effects were reported in the 1996 Ozone Air Quality Criteria Document (O₃ AQCD; U.S. Environmental Protection Agency, 1996a). Many of those studies reported that pulmonary function decrements, respiratory symptoms, and hospital and emergency department admissions in human populations were associated with ambient levels of O₃. Numerous more recent epidemiologic studies discussed in this chapter evaluate the relationship of ambient O₃ to morbidity and mortality, and thereby provide an expanded basis for assessment of health effects associated with exposures to O₃ at concentrations currently encountered in the United States.

As discussed elsewhere in this document (Chapters 5 and 6), a substantial amount of experimental evidence links O₃ exposure unequivocally with respiratory effects in laboratory animals and humans. These include structural changes in the bronchiolar-alveolar transition (centriacinar) region of the lung, biochemical evidence of acute cellular and tissue injury, inflammation, increased frequency and severity of experimental bacterial infection, and temporary reductions in mechanical lung function. These effects have been observed with exposure to O₃ at ambient or near-ambient concentrations. Thus, many of the reported epidemiologic associations of ambient O₃ with respiratory health effects have considerable biological credibility. Accordingly, the new epidemiologic studies of ambient O₃ assessed here are best considered in combination with information from the other chapters on ambient O₃ concentration and exposure (Chapter 3), and toxicological effects of O₃ in animals and humans (Chapters 5 and 6, respectively). The epidemiologic studies constitute important information on associations between health effects and exposures of human populations to “real-world” O₃ and also help to identify susceptible subgroups and associated risk factors.

A wide variety of oxidants in both the gaseous and particulate phases have not been examined in relation to health outcomes in the literature. Therefore, discussion in this chapter is limited to studies of human health effects associated with ambient O₃ exposure. Ambient concentrations of the most abundant non-O₃ oxidants (i.e., peroxyacetyl nitrate, peroxypropionyl nitrate, and H₂O₂) have not been shown to cause adverse health effects in toxicologic studies. However, as constituents of ambient air mixes, these oxidants may contribute to some of the effects attributed to O₃. Therefore, health effect associations observed in relation to ambient O₃ concentrations may represent O₃ effects, per se, or O₃ may be serving as a surrogate measure for the overall photochemical oxidant mix.

7.1.1 Approach to Identifying Ozone Epidemiologic Studies

Numerous O₃ epidemiologic papers have been published since completion of the 1996 O₃ AQCD. The U.S. Environmental Protection Agency (NCEA-RTP) has implemented a systematic approach to identify relevant epidemiologic studies for consideration in this chapter. In general, an ongoing search has been employed in conjunction with other strategies to identify O₃ epidemiologic literature pertinent to developing criteria for the O₃ National Ambient Air Quality Standards (NAAQS). A publication base was established using Medline, Pascal, BIOSIS, NTIS, and Embase, as well as a set of search terms proven by prior use to identify pertinent literature. The search strategy was later reexamined and modified to enhance identification of published papers. PubMed was added to the search regime. New studies accepted for publication through December 2004, as identified using the approaches above, have been included in this AQCD.

While the above search regime provided good coverage of the relevant literature, additional approaches augmented the traditional search methods. First, a Federal Register Notice was issued requesting information and published papers from the public at large. Next, non-EPA chapter authors, expert in this field, identified literature on their own. NCEA-RTP staff also identified publications as an element of their assessment and interpretation of the literature. Finally, additional potentially relevant publications were included following external review as a result of comments from both the public and the Clean Air Scientific Advisory Committee (CASAC). More recent studies accepted in 2005 and 2006 for publication also were included if they added significantly to the existing body of data on critically important topics,

such as those discussing methodological issues (e.g., O₃ exposure misclassification, potential confounding by copollutants and meteorological factors), the O₃-mortality relationship, and cardiovascular health outcomes. The combination of these approaches is believed to have produced a comprehensive ascertainment of studies appropriate for review and assessment here. The main criterion for selecting literature for the present assessment is to include those identified studies that evaluated relationships between measured ambient O₃ levels and human health outcomes, which are pertinent to the evaluation of scientific bases useful for derivation of O₃ NAAQS for the United States.

7.1.2 Approach to Assessing Epidemiologic Evidence

Definitions of the various types of epidemiologic studies assessed here were noted in an earlier particulate matter (PM) AQCD (U.S. Environmental Protection Agency, 1996b). Briefly, epidemiologic studies are generally divided into two groups, *morbidity* studies and *mortality* studies. *Morbidity* studies evaluate O₃ effects on a wide range of health endpoints, including: changes in pulmonary function, respiratory symptoms, self-medication in asthmatics, and airway inflammation; changes in cardiovascular physiology/functions; and cardiopulmonary emergency department visits and hospital admissions. *Mortality* studies investigate O₃ effects on total (nonaccidental) mortality and cause-specific mortality, providing evidence related to a clearly adverse endpoint. The epidemiologic strategies most commonly used in O₃ health studies are cross-sectional studies, prospective cohort studies, ecologic studies, time-series semi-ecologic studies, and case-crossover studies. All of these are observational rather than experimental studies.

The approach to assessing epidemiologic evidence has been stated most recently in the 2004 PM AQCD (U.S. Environmental Protection Agency, 2004) and is summarized here. The critical assessment of epidemiologic evidence presented in this chapter is conceptually based upon consideration of salient aspects of the evidence of associations so as to reach fundamental judgments as to the likely causal significance of the observed associations (see Hill, 1965). The general evaluation of the strength of the epidemiologic evidence reflects consideration not only of the magnitude and precision of reported O₃ effect estimates and their statistical significance, but also of the robustness of the effects associations. Statistical significance corresponds to the allowable rate of error (Type I error) in the decision problem constructed from assuming that a

simple null hypothesis of no association is true. It is a conditional probability; for statistical significance, typically there is a less than 0.05 chance of rejecting the null hypothesis given that it is true. Robustness of the associations is defined as stability in the effect estimates after considering a number of factors, including alternative models and model specifications, potential confounding by copollutants, as well as issues related to the consequences of measurement error.

Consideration of the consistency of the effects associations, as discussed in the following sections, involves looking across the results of multiple- and single-city studies conducted by different investigators in different places and times. Relevant factors are known to exhibit much variation across studies, including, for example, the presence and levels of copollutants, the relationships between central measures of O₃ and exposure-related factors, relevant demographic factors related to sensitive subpopulations, and climatic and meteorological conditions. Thus, in this case, consideration of consistency and the related heterogeneity of effects are appropriately understood as an evaluation of the similarity or general concordance of results, rather than an expectation of finding quantitative results within a very narrow range.

Looking beyond the epidemiologic evidence, evaluation of the biological plausibility of the O₃-health effects associations observed in epidemiologic studies reflects consideration of both exposure-related factors and dosimetric/toxicologic evidence relevant to identification of potential biological mechanisms. Similarly, coherence of health effects associations reported in the epidemiologic literature reflects consideration of information pertaining to the nature of the various respiratory- and cardiac-related mortality and morbidity effects and biological markers evaluated in toxicologic and human clinical studies. These broader aspects of the assessment are only touched upon here but are more fully integrated in the discussion presented in Chapter 8.

In assessing the relative scientific quality of epidemiologic studies reviewed here and to assist in interpreting their findings, the following considerations were taken into account:

- (1) To what extent are the aerometric data/exposure metrics used of adequate quality and sufficiently representative to serve as credible exposure indicators, well-reflecting geographic or temporal differences in study population pollutant exposures in the range(s) of ambient pollutant concentrations evaluated?
- (2) Were the study populations well defined and adequately selected so as to allow for meaningful comparisons between study groups or for meaningful temporal analyses of health effects results?

- (3) Were the health endpoint measurements meaningful and reliable, including clear definition of diagnostic criteria utilized and consistency in obtaining dependent variable measurements?
- (4) Were the statistical analyses used appropriate, as well as properly performed and interpreted?
- (5) Were likely important covariates (e.g., potential confounders or effect modifiers) adequately controlled for or taken into account in the study design and statistical analyses?
- (6) Were the reported findings internally consistent, biologically plausible, and coherent in terms of consistency with other known facts?

These guidelines provide benchmarks for judging the relative quality of various studies and in assessing the body of epidemiologic evidence. Detailed critical analysis of all epidemiologic studies on O₃ health effects, especially in relation to all of the above questions, is beyond the scope of this document. As discussed in the upcoming Sections 7.1.3 and 7.1.4, considerations in the interpretation and presentation of the epidemiologic evidence led to emphasis being placed on certain studies in the main chapter text, tables, and figures. Additional studies in the O₃ epidemiologic literature are presented in Chapter 7 Annex Tables (see Annex Section AX7.1). Of most importance are those studies which provide useful qualitative or quantitative information on concentration-response relationships for health effects associated with ambient air levels of O₃ likely to be encountered among healthy and susceptible populations in the United States.

7.1.3 Considerations in the Interpretation of Epidemiologic Studies of Ozone Health Effects

Prior to discussing results from recent O₃ epidemiologic studies, issues and questions arising from the study designs and analysis methods used in the assessment of O₃ effect estimates will be briefly presented. Study design can restrict the health effect parameters that can be estimated. Separate considerations need to be made for acute versus chronic effect studies, as well as individual- versus aggregate-level analyses. Time-series studies and panel studies are most frequently conducted in air pollution epidemiologic research. Aggregate-level exposure and/or outcome data are often used in these types of studies. Analyses using administrative health outcome data (e.g., numbers of deaths and emergency hospital admissions)

have inherent limitations as well as strengths (Virnig and McBean, 2001). The impact of study design or the loss of information due to aggregation depends on the source of exposure (Sheppard et al., 2005).

This section mainly focuses on the topics of exposure assessment and model specification in air pollution epidemiologic studies. Potential biases that may result from O₃ exposure measurement error and from the choice of exposure index and lag period are discussed first. Model specification issues and potential confounding by temporal factors, meteorological effects, seasonal trends, and copollutants are then discussed.

7.1.3.1 Exposure Assessment and Measurement Error in Epidemiologic Studies

In many air pollution epidemiologic studies, especially time-series studies with administrative data on mortality and hospitalization outcomes, data from central ambient monitoring sites generally are used as the estimate of exposure. Personal exposures of individual study participants generally are not directly measured in epidemiologic studies. The use of O₃ concentrations from ambient monitors as surrogate measures for personal O₃ exposures was discussed previously (Chapter 3, Section 3.9). Routinely collected ambient monitor data, though readily available and convenient, may not represent true personal exposure, which includes both ambient and non-ambient (i.e., indoor) source exposures.

In several studies focused on evaluating exposure to O₃, measurements were made in a variety of indoor environments, including homes (Lee et al., 2004), schools (Linn et al., 1996), and the workplace (Liu et al., 1995). Indoor O₃ concentrations were, in general, approximately one-tenth of the outdoor concentrations in these studies. Few indoor sources of O₃ exist, possible sources being office equipment (e.g., photocopiers, laser printers) and air cleaners. As described in Section 3.8 of this document, O₃ in the indoor environment is largely dependent on the outdoor ambient O₃ concentration. Other factors that influence indoor O₃ concentrations include air exchange rate, outdoor infiltration, indoor circulation rate, and O₃ removal processes.

Sheppard (2005) described the relationship between panel studies and time-series studies and discussed the use of personal exposure measurements versus ambient concentrations in these designs, with the main focus being on nonreactive pollutants (e.g., PM). She noted that non-ambient exposures typically varied across individuals but were not likely to have strong temporal correlations. In contrast, ambient concentrations across individuals should be highly correlated,

as they tend to vary over time similarly for everyone because of changes in source generation, weather, and season. The independence of ambient and non-ambient exposure sources has important implications for selection of study designs that are most effective for estimating health effects (Sheppard, 2005). A simulation study by Sheppard et al. (2005) examining nonreactive pollutants found that there was no noticeable difference between effect estimates using either total personal exposure or ambient concentration data when non-ambient source exposures were independent of ambient source exposures in time-series studies. In the case of O₃, there are limited non-ambient sources; thus, the non-ambient source exposures are likely to be independent from ambient source exposures. However, unlike PM, O₃ is a reactive pollutant. In applying these conclusions to O₃, an additional assumption needs to be made, i.e., that its chemical reactivity does not introduce strong temporal correlations.

Other complications for O₃ in the relationship between personal exposures and ambient concentrations include expected strong seasonal variation of personal behaviors and building ventilation practices that can modify exposure. In addition, the relationship may be affected by temperature (e.g., high temperature may increase air conditioning use, which may reduce O₃ penetration indoors), further complicating the role of temperature as a confounder of O₃ health effects. It should be noted that the pattern of exposure misclassification error and influence of confounders may differ across the outcomes of interest as well as in susceptible populations. For example, those who may be suffering from chronic cardiovascular or respiratory conditions may be in a more protective environment (i.e., with less exposure to both O₃ and its confounders, such as temperature and PM) than those who are healthy.

As discussed thoroughly in the 2004 PM AQCD (Section 8.4.5), the resulting exposure measurement error and its effect on the estimates of relative risk must be considered. In theory, there are three components to exposure measurement error in time-series studies as described by Zeger et al. (2000): (1) the use of average population rather than individual exposure data; (2) the difference between average personal ambient exposure and ambient concentrations at central monitoring sites; and (3) the difference between true and measured ambient concentrations. The first error component, having aggregate rather than individual exposure data, is a Berksonian measurement error, which in a simple linear model increases the standard error, but does not bias the risk estimate. The second error component resulting from the difference between average personal ambient exposure and outdoor ambient concentration level

has the greatest potential to introduce bias. If the error is of a fixed amount (i.e., absolute differences do not change with increasing concentrations), there is no bias. However, if the error is not a fixed difference, this error will likely attenuate the O₃ risk estimate as personal O₃ exposures are generally lower than ambient O₃ concentrations. The third error component, the instrument measurement error in the ambient levels, is referred to as nondifferential measurement error and is unlikely to cause substantial bias.

The impact of exposure measurement error on O₃ effect estimates was demonstrated in a study by Navidi et al. (1999). In this study, two statistical designs (bidirectional case-crossover design and multilevel analytic design) were considered. Simulations were conducted using data from the University of Southern California Children's Health Study of the long-term effects of air pollutants on children. The effect estimate from computed "true" O₃ exposure was compared to effect estimates from exposure determined by using several methods: (1) ambient stationary monitors; (2) the microenvironmental approach (multiply O₃ concentrations in various microenvironments by time present in each microenvironment); and (3) personal sampling. Effect estimates based on all three exposure measures were biased toward the null. The bias that results when using the microenvironmental and personal sampling approach is due to nondifferential measurement error. Use of ambient monitors to determine exposure will generally overestimate true personal O₃ exposure (because their use implies that subjects are outdoors 100% of their time and not in close proximity to sources that reduce O₃ levels such as NO emissions from mobile sources); thus, generally, their use can result in effect estimates that are biased toward the null if the error is not of a fixed amount.

Zidek (1997) noted that a statistical analysis must balance bias and imprecision (error variance). Ignoring measurement error in air pollution epidemiologic studies often results in underestimated risk estimates. For example, in a reanalysis of the study by Burnett et al. (1994) on the acute respiratory effects of ambient air pollution, Zidek et al. (1998) noted that accounting for measurement error, as well as making a few additional changes to the analysis, resulted in qualitatively similar conclusions. However, while the original analysis by Burnett et al. observed a 4.0% (95% CI: 1.9, 6.0) excess risk of daily respiratory admissions in the summer months attributable to a 40 ppb increase in 1-h max O₃ (combined effect of a 1- and 3-day lag of O₃), results from Zidek et al. indicated that a 9.4% (95% CI: 7.0, 11.8) increase was observed for a 40 ppb increase (a change from 10 to 50 ppb) in 1-h max O₃ (combined effect of a 2- and

3-day lag of O₃). Although the different lag periods examined in the two analyses make direct comparisons of the effect estimates difficult, it appears that correcting the measurement error resulted in a larger effect estimate. Available data and analysis limit our ability to weigh the importance of uncertainty due to measurement error relative to other sources in the various studies reviewed in this chapter.

In addition to overestimation of exposure and the resulting underestimation of effects, the use of ambient O₃ concentrations may obscure the presence of thresholds in epidemiologic studies at the population level. Using PM_{2.5} as an example, Brauer et al. (2002) examined the relationship between ambient concentrations and mortality risk in a simulated population with specified common individual threshold levels. They found that no population threshold was detectable when a low threshold level was specified. Even at high specified individual threshold levels, the apparent threshold at the population level was much lower than specified. Brauer et al. (2002) concluded that surrogate measures of exposure (i.e., those from centrally-located ambient monitors) that were not highly correlated with personal exposures obscured the presence of thresholds in epidemiologic studies at the population level, even if a common threshold exists for individuals within the population.

As discussed in Section 3.9, O₃ concentrations measured at central ambient monitors may explain, at least partially, the variance of individual personal exposures; however, this relationship is influenced by factors such as air exchange rates in housing and time spent outdoors, which may vary by city. Other studies conducted in various cities observed that the daily averaged personal O₃ exposures from the population were well correlated with monitored ambient O₃ concentrations, although substantial variability existed among the personal measurements. Thus, there is supportive evidence that ambient O₃ concentrations from central monitors may serve as valid surrogate measures for mean personal O₃ exposures experienced by the population, which is of most relevance to time-series studies. This is especially likely true for respiratory hospital admission studies for which much of the response is attributable to O₃ effects on asthmatics. In children, for whom asthma is more prevalent, ambient monitors are likely to correlate reasonably well with personal exposure to O₃ of ambient origin because children spend more time outdoors in the warm season. However, of some concern for mortality and hospitalization time-series studies is the extent to which ambient O₃ concentrations are representative of personal O₃ exposures in another particularly susceptible group of individuals,

the debilitated elderly, as the correlation between the two measurements has not been examined in this population. A better understanding of the relationship between ambient concentrations and personal exposures, as well as of the factors that affect the relationship will improve the interpretation of ambient concentration-population health response associations observed.

Existing epidemiologic models may not fully take into consideration all the biologically relevant exposure history or reflect the complexities of all the underlying biological processes. Using ambient concentrations to determine exposure generally overestimates true personal O₃ exposures (by approximately 2- to 4- fold in the various studies described in Section 3.9), resulting in biased descriptions of underlying concentration-response relationships (i.e., in attenuated risk estimates). The implication is that the effects being estimated occur at fairly low exposures and the potency of O₃ is greater than these effect estimates indicate. As very few studies evaluating O₃ health effects with personal O₃ exposure measurements exist in the literature, effect estimates determined from ambient O₃ concentrations must be evaluated and used with caution to assess the health risks of O₃.

The ultimate goal of the O₃ NAAQS is to set a standard for the *ambient level*, not personal exposure level, of O₃. Until more data on personal O₃ exposure become available, the use of routinely monitored ambient O₃ concentrations as a surrogate for personal exposures is not generally expected to change the principal conclusions from O₃ epidemiologic studies. Therefore, population health risk estimates derived using ambient O₃ levels from currently available observational studies (with appropriate caveats taking into account personal exposure considerations) remain useful.

7.1.3.2 Ozone Exposure Indices Used

The O₃-related effect estimates for mortality and morbidity health outcomes are usually presented in this document as a relative risk, i.e., the risk rate relative to a baseline mortality or morbidity rate. Relative risks are based on an incremental change in exposure. To enhance comparability between studies, presenting these relative risks by a uniform exposure increment is needed. However, determining a standard increment is complicated by the use of different O₃ exposure indices in the existing health studies. The three daily O₃ exposure indices that most often appear in the literature are the maximum 1-h average within a 24-h period (1-h max), the maximum 8-h average within a 24-h period (8-h max), and 24-h average (24-h avg)

concentrations. As levels are lower and less variable for the longer averaging times, relative risks of adverse health outcomes for a specific numeric concentration range are not directly comparable across metrics. Using the nationwide distributional data for O₃ monitors in U.S. Metropolitan Statistical Areas, increments representative of a low-to-high change in O₃ concentrations were approximated on the basis of annual mean to 95th percentile differences (Langstaff, 2003), as follows:

Daily Exposure Index	Exposure Increment (ppb)
1-h max O ₃	40
8-h max O ₃	30
24-h avg O ₃	20

In the following chapter sections, efforts were made to standardize the O₃ risk estimates using these increments, except as noted. The specified incremental change for each daily O₃ exposure index ensures that risk estimates are comparable across the different metrics. For example, a 2% excess risk for a 20 ppb change in mean 1-h max O₃ was standardized and presented as a 4% excess risk for a 40 ppb change in mean 1-h max O₃. This standardized risk estimate is approximately comparable to the excess risks observed for a 30 ppb change in mean 8-h max O₃ or a 20 ppb change in mean 24-h avg O₃. Thus, the different increments for each daily O₃ exposure index do not represent inconsistencies; rather, they are appropriately scaled to facilitate comparisons between the various studies that used different indices. Note that in the Chapter 7 Annex Tables (see Annex Section AX7.1), effect estimates are not standardized; there, the results are presented in the tables as reported in the published papers.

7.1.3.3 Lag Time: Period between Ozone Exposure and Observed Health Effect

Exposure lags may reflect the distribution of effects across time in a population and the potential mechanisms of effects. The choice of lag days for the relationship between exposure and health effects depends on the hypothesis being tested and the mechanism involved in the expression of the outcome. Effects can occur acutely with exposure on the same or previous day, cumulatively over several days, or after a delayed period of a few days. With knowledge of the mechanism of effect, the choice of lag days can be determined prior to analysis.

As one example, one could expect cough to occur acutely after exposure with a lag of 0 or 1 day, given that O₃ can act as a short-term irritant. However, an O₃-related inflammatory response may not lead to asthma exacerbation until several days later. An asthmatic may be impacted by O₃ on the first day of exposure, have further effects triggered on the second day, and then report to the emergency room for an asthmatic attack three days after exposure. Further, within a population of asthmatics, exacerbation of asthma symptoms may be observed over a period of several days, since each asthmatic may have varying individual aspects of the disease and may be affected by the exposure differently depending on his/her sensitivity and disease severity. The results from controlled human studies may be useful in assessing the adequacy of lags for some respiratory health outcomes.

Some studies attempted to examine the overall impact of O₃ through distributed lag models. The single-day lag model calculates a risk estimate that assumes dependence only on exposure from the specified day. In contrast, the distributed lag model provides an estimate that is a summary measure of the cumulative distributed lag effect from all included lag days. The standard error of the cumulative sum of the individual distributed lag coefficients takes into consideration the variance-covariance of the multiple lags, and it is generally larger than the standard error of the single-day lag coefficient due to positive auto-correlation. Thus, if the underlying O₃-health outcome relationship was a single-day effect, then modeling the relationship with a distributed lag model would make the estimate less significant. On the other hand, if the effect of O₃ on health outcomes persisted over several days, then applying a single-day lag model would result in an underestimation of the multiday effects, although the single-day estimate may still reflect a portion of the multiday effect, to the extent that daily O₃ levels are auto-correlated. Choosing a lag model requires balancing variance and bias, and the usefulness (i.e., ease of interpretation, comparability across studies) of the model, as well as knowledge about mechanisms underlying effects for a given health outcome. Sufficient information is not available to judge the adequacy of a given model for different outcomes.

As the parameters estimated from single-day lag versus multiday lag models are not the same, interpretation and comparison of these results will be difficult. When comparing the impacts of these different models, the nuance of increments used in calculating the estimates is different depending on the model. For example, an excess percent mortality risk “per 20 ppb increase in 24-h avg O₃” in a distributed lag model including lag 0- through 6-days tacitly means

a 20 ppb increase in each of the 7 days. The difference in the exposure scenarios in the single-day versus multiday lag model (i.e., 20 ppb increase in one day versus several consecutive days) complicates a simple comparison of risk estimates from two different models using “the same increment.” Thus, it should be recognized that these are distinct models that summarize the underlying process differently.

Only a limited number of studies have hypothesized a priori the lag structure to be examined. Most of the O₃ time-series studies examined relatively small numbers of single-day lag models, typically lags of 0 through 3 days. Sheppard et al. (1999) noted that when considering single-day lag estimates it is important to consider the effect estimate in the context of the pattern of adjacent lags as these estimates contain information from the adjacent days owing to serial correlation of the pollutant series. In many cases, a pattern of positive associations across several lag days was reported. For the respiratory and cardiovascular outcomes investigated, the “most significant” lags were generally 0- or 1-day lags, suggesting that the majority of the single-day associations are immediate, not a random pattern in which associations can be observed on any of the lags examined with equal probabilities. For example, two recent meta-analyses of O₃-mortality effects observed that the combined estimate from 0-day lag models was larger than the estimate from longer lag days (Bell et al., 2005; Levy et al., 2005).

Bias resulting from the selection of lags has not been examined specifically for O₃ effects. However, the issue of lags has been investigated for PM and the results of this analysis are most likely of relevance for O₃. Lumley and Sheppard (2000) performed a simulation study to examine model selection bias in air pollution epidemiology using PM_{2.5} as an example. Sheppard et al. (1999; reanalysis Sheppard, 2003) had investigated the association between asthma hospital admissions and ambient PM_{2.5} concentrations over an eight-year period in Seattle, WA. Note that the results from Lumley and Sheppard (2000) and Sheppard et al. (1999) were based on Generalized Additive Model (GAM) analyses using default convergence criteria (see Section 7.1.3.7). A negative control analysis, using simulated data with no association between PM exposure and the health outcome, and a positive control analysis, using a specified non-zero excess risk added to the simulation, were performed for comparison. The bias from selection of the best of seven lags (0 to 6 days) and residual seasonal confounding in the negative control analysis (median log relative risk of 0.0013) was approximately half the log relative risk

estimated from the observed data (0.0027), after adjusting for season and temperature. In the positive control model (true log relative risk of 0.0083), the bias was small (median log relative risk of 0.0080). Results from these simulations indicate that bias from selection of lags may be small, but of the same magnitude as the estimated health impacts.

Selection of lag periods should depend on the hypothesis of the study and the potential mechanism of the effect. When the mechanism of the health effect is unknown, investigating the association between outcome and exposure using cumulative distributed lag models may be informative. Analyzing a large number of lags and simply choosing the largest and most significant results may bias the air pollution risk estimates away from the null. Most studies have shown that O₃ has a fairly consistent, immediate effect on health outcomes, including respiratory hospitalizations and mortality. Several studies also observed significant O₃ effects over longer cumulative lag periods, suggesting that in addition to single-day lags, multiday lags should be investigated to fully capture a delayed O₃ effect on health outcomes. In this document, discussion largely focuses on effect estimates from 0- and 1-day lags, with some consideration of cumulative, multiday lag effects. It is not straightforward to compare and contrast results from single-day versus multiday lag models, because the parameters estimated from these models are not the same. These complications need to be taken into consideration when interpreting results from various lag models.

7.1.3.4 Model Specification to Adjust for Temporal Trends and Meteorologic Effects

Several challenges are encountered with respect to designing and interpreting time-series studies. The principal challenge facing the analyst in the daily time-series context is avoiding bias due to confounding by short-term temporal factors operating over time scales from days to seasons. In the current regression models used to estimate short-term effects of air pollution, two major potential confounders generally need to be considered: (1) seasonal trend and other “long-wave” temporal trends; and (2) weather effects. Both of these variables tend to predict a significant fraction of fluctuations in time-series. Unfortunately, as O₃ has strong seasonal cycles and is formed more at higher temperatures, both terms are also highly correlated with O₃. The correlation of O₃ with these confounding terms tends to be higher than that for PM or other gaseous pollutants. In the United States, the mass concentration of PM_{2.5} generally does not have strong seasonal cycles like O₃, because PM_{2.5} tends to reflect both primary emissions

(throughout the year, but often higher in winter in most U.S. cities) and secondary aerosols (higher in summer). Therefore, PM_{2.5} and O₃ effect estimates from studies primarily designed to examine PM_{2.5} health effects may not be comparable, because model specifications that may be appropriate for PM_{2.5} may not necessarily be adequate for O₃.

An examination of recent time-series studies indicates that several types of fitting approaches have been used to adjust for temporal trends and weather effects. The use of parametric and nonparametric smoothers with varying degrees of freedom per year has emerged as the prevailing approach. The use of larger degrees of freedom to adjust for potential confounding by time-varying factors may inadvertently result in ascribing more effects to these unmeasured potential confounders and mask the air pollution effect. Often smaller pollution effect estimates are observed when more degrees of freedom are used. Currently, the degrees of freedom used to adjust for temporal trends in time-series studies generally range from 4 to 12 degrees of freedom per year using either nonparametric or parametric smoothers. Statistical diagnostics such as Akaike's Information Criteria, residual autocorrelation, or dispersion of the regression model often are used to choose or to evaluate the adequacy of the degrees of freedom for temporal trend. However, these diagnostics do not guarantee "adequate" control for temporal confounding, as the appropriate extent of smoothing is not identifiable from the data and the proper selection of smoothing parameters requires prior knowledge of the nature of the confounding (e.g., shape and duration of influenza epidemics).

The issue of model specifications to adjust for temporal trends and weather variables in time-series studies was a consideration of several researchers that conducted sensitivity analyses of PM data (Health Effects Institute, 2003). The sensitivity of O₃ coefficients to model specifications for temporal trend adjustment has not been as well-studied. Recent multicity studies examined the sensitivity of O₃ coefficients to the extent of smoothing for adjustment of temporal trends and meteorologic factors (Bell et al., 2004; Huang et al., 2005; Ito et al., 2005). Most, if not all, O₃ studies used the same model specifications to estimate the excess risks for PM and other gaseous pollutants. The model specification designed to control confounding by meteorological and temporal factors for PM may not necessarily be adequate for O₃. As noted above, O₃ is expected to have the strongest correlation with both temporal (seasonal) trend and weather effects. The strong annual cycle in O₃ concentrations presents a unique problem in time-series analyses where time trends are fitted simultaneously with pollution and other model

terms (i.e., co-adjustment). In this setting, the annual O₃ cycle itself may compete with the smooth function of time to explain some of the annual, cyclic behavior in the health outcome, which can result in biased effect estimates for O₃ when data for all seasons are analyzed together.

Current weather models used in time-series analyses can be classified by their use of: (1) quantile (e.g., quartile, quintile) indicators; (2) parametric functional forms such as V- or U-shape functions; and (3) parametric (e.g., natural splines) or nonparametric (e.g., locally estimated smoothing splines [LOESS]) smoothing functions. More recent studies tend to use smoothing functions. While these methods provide flexible ways to fit health outcomes as a function of temperature and other weather variables, there are two major issues that need further examination to enable more meaningful interpretation of O₃ morbidity and mortality effects.

The first issue is the interpretation of weather or temperature effects. Most researchers agree about the morbidity and mortality effects of extreme temperatures (i.e., heat waves or cold spells). However, as extreme hot or cold temperatures, by definition, happen rarely, much of the health effects occur in the mild or moderate temperature range. Given the significant correlation between O₃ and temperature, ascribing the association between temperature and health outcomes solely to temperature effects may underestimate the effect of O₃. The second issue is that weather model specifications are fitted for year-round data in most studies. Such models will ignore the correlation structure that can change across seasons, resulting in inefficiency and model mis-specification. This is particularly important for O₃, which appears to change its relationship with temperature as well as with other pollutants across seasons.

This changing relationship between O₃ and temperature, as well as between O₃ and other pollutants across seasons, and its potential implications for health effects modeling have not been examined thoroughly in the time-series literature. Even the flexible smoother-based adjustments for seasonal and other time-varying variables cannot fully take into account these complex relationships. One obvious way to alleviate or avoid this complication is to analyze data by season. While this practice reduces sample size, its extent would not be as serious as for PM (which is collected only every sixth day in most locations) because O₃ is collected daily, though only in warm seasons in some states. An alternative approach is to include separate O₃ concentration variables for each season (by multiplying O₃ concentrations by a season indicator

variable). However, this approach assumes that all effects in the model that are not indicated to be season-specific do not vary seasonally.

In locations where seasonal variability may be a factor, O₃ effect estimates calculated using year-round data can be misleading, as the changing relationship between O₃, temperature, and other pollutants across seasons may have a significant influence on the estimates. Analyses have indicated that confounding from seasonal variability may be controlled effectively by stratifying the data by season.

7.1.3.5 Confounding Effects of Copollutants

Extensive discussion of issues related to confounding effects among air pollutants in time-series studies are provided in Section 8.4.3 of the 2004 PM AQCD (U.S. Environmental Protection Agency, 1996b). Since the general issues discussed there are applicable to all pollutants, such discussions are not repeated here. What was not discussed in the 2004 PM AQCD was the issue of changing relationships among air pollutants across seasons. Compared to other pollutants, O₃ has strong seasonal cycles. Ambient O₃ levels are typically higher in the summer or warm season, often referred to as the O₃ season. In the winter or colder months, O₃ levels tend to be much lower than in the summer months. During the winter in some urban locations, O₃ mainly comes from the free troposphere and can be considered a tracer for relatively clean air (i.e., cold, clear air coming down from the upper atmosphere), as discussed in Chapter 3 of this document. The clean air is associated with the passage of cold fronts and the onset of high-pressure conditions, which occur with colder temperatures. Thus, sunny clear winter days following a high-pressure system are the days when air pollution levels from primary emissions (e.g., NO₂, SO₂, and PM from local sources) tend to be lower and O₃ is relatively higher. This can lead to negative correlations between O₃ and the primary pollutants in the winter. As shown in Figure 3-21 in Chapter 3, the relationship between O₃ and PM_{2.5} was U-shaped for the year-round data in Fort Meade, MD. The negative PM_{2.5}/O₃ slope was in the range of O₃ concentrations less than 30 ppb, providing supporting evidence of the aforementioned winter phenomenon. Thus, the correlation between O₃ and PM for year-round data may be misleading. The high reactivity of O₃ with certain copollutants further complicates the analysis and interpretation. For example, the reaction between NO (emitted from motor vehicles) and O₃ results in reduced O₃ levels but increased NO₂ levels during high traffic periods.

Multipollutant regression models often are used to assess potential confounding by copollutants; however, there are limitations to these models. Zidek et al. (1996) examined, through simulation, the joint effects of multicollinearity and measurement error in a Poisson regression model. The results illustrated the transfer of effects from the “causal” variable to the confounder. However, in order for the confounder to have a larger effect size than the true predictor, the correlation between the two covariates had to be very high ($r \geq 0.9$), with moderate error ($\alpha > 0.5$, where α is the probability of rejecting the null hypothesis given that the null is true) for the true predictor and no error for the confounder in their scenarios. The transfer-of-causality effect was lessened when the confounder also became subject to error. Another interesting finding was the behavior of the standard errors of the coefficients. When the correlation between the covariates was high ($r = 0.9$) and both covariates had no error, the standard errors for both coefficients were inflated by a factor of two; however, this phenomenon disappeared when the confounder had error. The effect of multicollinearity is generally even more complex when analyzing real data. For further discussion, see the 2004 PM AQCD (Sections 8.4.3 and 8.4.5).

Uncertainty remains as to the use of multipollutant regression models in assessing the independent health effects of pollutants that are correlated. Particularly in the case of O_3 , concern remains as to whether multipollutant regression models for year-round data can adjust for potential confounding adequately because of the changing relationship between O_3 and other pollutants. Despite these limitations, use of multipollutant models is still the prevailing approach employed in most, if not all, studies of O_3 health effects; and it serves as an important tool in addressing the issue of confounding by copollutants, especially in season-stratified analyses.

7.1.3.6 Hypothesis Testing and Model Selection in Ozone Epidemiologic Studies

Epidemiologic studies investigated the association between various measures of O_3 (e.g., multiple lags, different metrics, etc.) and various health outcomes using different model specifications. Statistically testing a null hypothesis (i.e., there is no effect of O_3) requires one to calculate the value of a test statistic (i.e., the t-value). If the observed test statistic exceeds a critical value (oftentimes the 95th percentile) or is outside a range of values, the null hypothesis is rejected. However, when multiple testing is done using a critical value determined for a single

test, the chance that at least one of the hypotheses is significant may be greater than the specified error rate. Procedures are available to ensure that the rejection error rate does not exceed the expected error rate (usually 5%) when conducting multiple hypothesis testing. However, often the multiple hypotheses being tested are not statistically independent, thus some corrections, such as the Bonferroni adjustment, may be overly conservative.

Multiple hypothesis testing and model selection also contribute to publication bias. Publication bias is the tendency of investigators to submit and/or editors to accept manuscripts for publication based on the strength of the study findings. Although publication bias commonly exists for many topics of research, it may be present to a lesser degree in the air pollution literature. Several air pollutants often are examined in a single study, leading to the publication of significant, as well as nonsignificant, individual pollutant results. For example, many air pollution papers with a focus on PM health effects also published O₃ results. Ozone was largely considered a potentially confounding copollutant of PM; thus, O₃ effect estimates were often presented regardless of the statistical significance of the results. Another aspect of publication bias is only selecting the largest or statistically strongest effect estimate to report and not the array of models evaluated. Bell et al. (2005) conducted a meta-analysis of O₃ health effects studies and found that studies reporting a single estimate compared to multiple estimates generally showed larger effects for the same lag. They also examined the presence of this bias by comparing the results from the meta-analysis to the multicity National Morbidity, Mortality, and Air Pollution Study (NMMAPS). When comparing the marginal posterior distributions of the overall effect under the meta-analysis and in NMMAPS for the eight U.S. cities common to both the approaches, the meta-analysis effect estimate of same-day O₃ concentration was nearly 2-fold that of the NMMAPS pooled effect estimate.

Testing multiple hypotheses may, at times, be appropriate. For example, without an a priori biologic model that definitively specifies which of several lag periods or meteorologic-control models should take priority, multiple hypotheses may need to be developed for researchers to explore more thoroughly potential associations for an O₃-related health effect (Goodman, 2005). In this case, it would be useful to state which hypotheses are confirmatory versus exploratory. The key concern is the reporting of a single estimate or estimates from one model. Presentation of an array of models, exposure definitions, and outcome measures may be more appropriate in communicating the results. Sensitivity analyses, which are critical for

model validation, also involve multiple testing. When conducting sensitivity analyses, one should guard against multiple testing errors by restricting the inferences to consistency of the effect.

One method to address model uncertainty and multiple hypothesis testing is Bayesian model averaging. In Bayesian model averaging, predictions and inferences are based on a set of models rather than a single model, and each model contributes proportionally to the support it receives from the observed data (Clyde, 1999). In addition to the uncertainty of effect estimation, Bayesian model averaging can incorporate uncertainty regarding the choice of confounding variables, pollutants, and lags. Koop and Tole (2004) used Bayesian model averaging to analyze the effect of various air pollutants, including O₃, SO₂, CO, NO, NO₂, PM_{10-2.5}, and PM_{2.5}, on mortality in Toronto, Canada. The 50+ explanatory variables required the fitting of an enormous number of potential models. Although the point estimates for all pollutants were positive, very small effects were found. However, in the context of the many interaction terms, meteorological variables, smoothing surfaces, and the relatively loose posterior distribution, the analysis results cannot be interpreted meaningfully. Clyde (2000) and Clyde et al. (2000) also used Bayesian model averaging to analyze the relationship between PM and mortality. Clyde (2000) noted that Bayesian model averaging did not take into consideration factors that might bias the estimated effect toward the null. For example, measurement error in the exposure variables was not considered. In addition, the Poisson model (similar to many other regression models) assumed that all individuals in a population had equal risks, including potentially susceptible and vulnerable populations such as those with respiratory illnesses and outdoor workers. While Bayesian model averaging can theoretically be used to take into account uncertainty, claims of causality based on observational studies may be highly sensitive to the choice of prior distributions and class of models under consideration (Clyde et al., 2000). Another limitation of Bayesian model averaging is that the estimated posterior effects may be diluted (i.e., result in smaller coefficients) when variables are highly correlated, as may be the case for air pollution studies (George, 1999 in comments to Hoeting et al., 1999).

Additional methods to control for multiple hypothesis testing are by deciding a priori which hypotheses are confirmatory and exploratory, and then limiting the number of confirmatory tests. For example, Dominici et al. (2003) used a minimum number of tests in the U.S. 90 cities study, which reduced the uncertainty associated with multiple testing. In addition,

they performed sensitivity analyses to examine the consistency and robustness of the effects. Another approach is to partition the data into two sets, one for model identification and a second for model confirmation.

The summary of health effects in this chapter is vulnerable to the errors of publication bias and multiple testing. Efforts have been made to reduce the impact of multiple testing errors on the conclusions in this document. To address multiple hypothesis testing, emphasis will be placed in this chapter on a priori hypotheses. As identifying a priori hypotheses is difficult in the majority of the studies, the most common hypotheses will be considered. For example, although many studies examined multiple single-day lag models, priority would be given to effects observed at 0- or 1-day lags rather than at longer lags. Both single- and multiple-pollutant models that include O₃ will be considered and examined for robustness of results. Analyses of multiple model specifications for adjustment of temporal or meteorological trends will be considered sensitivity analyses. Sensitivity analyses shall not be granted the same inferential weight as the original hypothesis-driven analysis; however, these analyses will be discussed in this chapter as appropriate given their valuable insights that may lead scientific knowledge in new directions.

7.1.3.7 Impact of Generalized Additive Models Convergence Issue on Ozone Risk Estimates

Generalized Additive Models (GAM) have been widely utilized for epidemiologic analysis of the health effects attributable to air pollution. The impact of the GAM convergence issue was thoroughly discussed in Section 8.4.2 of the 2004 PM AQCD. Reports have indicated that using the default convergence criteria in the Splus software package for the GAM function can lead to biased regression estimates for PM and an underestimation of the standard error of the effect estimate (Dominici et al., 2002; Ramsay et al., 2003). GAM default convergence criteria has a convergence precision of 10^{-3} and a maximum number of 10 iterations. The more stringent convergence criteria refers to increased stringency of both the convergence precision and number of iterations. The use of default convergence criteria was found to be a problem when the estimated relative risks were small and two or more nonparametric smoothing curves were included in the GAM (Dominici et al., 2002). The magnitude and direction of the bias depend in part on the concavity of the independent variables in the GAM and the magnitude of the risk

estimate. Recent focus has been on the influence of the GAM function on effect estimates for PM. However, because O₃ covaries more strongly with both weather and time factors than does PM, the issue of GAM convergence criteria for O₃ also needs to be considered.

A meta-analysis by Stieb et al. (2003) found some difference in O₃-mortality risk estimates between the GAM and non-GAM studies. GAM studies were defined as studies that analyzed effect estimates using nonparametric smoothing functions of time or weather. Non-GAM studies were all other studies, including those using Generalized Linear Models (GLM) and Generalized Estimating Equations (GEE) in their analysis. In the single-pollutant models, the O₃-mortality risk estimates for the non-GAM studies (10 estimates) and GAM studies (15 estimates) were 1.8% (95% CI: 0.5, 3.1) and 2.2% (95% CI: 1.4, 2.8), respectively, per 40 ppb daily 1-h max O₃. In the multipollutant models, the pooled risk estimate was 1.0% (95% CI: -0.5, 2.6) for non-GAM studies (7 estimates) and 0.5% (95% CI: -1.0, 1.9) for GAM studies (4 estimates).

Results from recent meta-analyses of O₃-mortality effects suggest that there are no substantial differences between GAM-affected estimates and non-GAM-affected estimates (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005). GAM-affected studies included those that used default convergence criteria. Non-GAM-affected studies included GAM studies that used stringent convergence criteria or those that used other modeling techniques. Ito et al. (2005) found that the single-pollutant combined estimate for the GAM-affected studies (15 estimates) and non-GAM-affected studies (28 estimates) were 1.92% (95% CI: 1.02, 2.81) and 1.40% (95% CI: 0.78, 2.02), respectively, per 20 ppb increase in 24-h avg O₃. In the analysis by Levy et al. (2005), the single-pollutant combined estimate for the GAM-affected studies (29 estimates) and non-GAM-affected studies (17 estimates) were 1.56% (95% CI: 1.01, 2.11) and 1.80% (95% CI: 1.17, 2.43), respectively, per 40 ppb increase in 1-h max O₃. Bell et al. (2005) also reported that the pooled estimate was larger for the studies that were not GAM-affected.

A few GAM studies reanalyzed O₃ risk estimates using more stringent convergence criteria or GLM. Reanalysis of an asthma hospital admissions study in Seattle, WA (Sheppard et al., 1999; reanalysis Sheppard, 2003) indicated that there were only slight changes in the risk estimates when using more stringent convergence precision (10⁻⁸) in GAM. The original GAM analysis indicated an excess risk of 9% (95% CI: 3, 17), whereas the stringent GAM analysis found an excess risk of 11% (95% CI: 3, 19) per 30 ppb increase in 8-h max O₃ at a 2-day lag.

Similar results were found using GLM with natural splines: 11% (95% CI: 2, 20). In the reanalysis of Santa Clara County, CA data, Fairley (1999; reanalysis Fairley, 2003) used the same methods as the original analysis except the convergence precision (ϵ) was increased from 10^{-4} to 10^{-12} and the maximum number of iterations (M) were increased from 10 to 10^7 . The O₃-mortality risk estimate slightly increased from 2.8% (95% CI not provided) using default GAM parameters to 2.9% (95% CI: -0.3, 6.0) using stringent GAM parameters per 30 ppb increase in 8-h max O₃ at a 0-day lag. The O₃-mortality risk estimates further increased to 3.0% (95% CI: -0.3, 6.3) using GLM with natural cubic splines. In the reanalysis of the Netherlands data by Hoek et al. (2000; reanalysis Hoek, 2003), the O₃ nonaccidental mortality risk estimates increased from 1.3% (95% CI: 0.8, 1.9) using default GAM to 1.5% (95% CI: 1.0, 2.1) using stringent GAM ($\epsilon = 10^{-8}$, M = 10^3) and 1.6% (95% CI: 0.9, 2.4) using GLM with natural splines per 30 ppb increase in 8-h avg O₃ (12 p.m.-8 p.m.) at a 1-day lag.

In the limited number of studies that have reanalyzed O₃ risk estimates, there is little evidence that default GAM analyses resulted in positively biased estimates, as was observed for PM. Generally, it appears that the use of default convergence criteria in GAM tends to bias risk estimates toward the null, in addition to underestimating the standard errors. However, one study by Cifuentes et al. (2000) in Santiago, Chile observed a large difference in the O₃-mortality excess risks calculated using default GAM (0.9% [95% CI: 0.2, 1.6] per 40 ppb increase in 1-h max O₃) and GLM (0.1% [95% CI: -0.6, 0.8]). The GAM convergence problem appears to vary depending on data sets, and likely depends upon the intercorrelation among covariates and the magnitude of the risk estimate; thus, its impact on the results of individual studies cannot be known without a reanalysis. Consistent with the approach used in the 2004 PM AQCD, the results from studies that analyzed data using GAM with default convergence criteria and at least two nonparametric smoothing terms are generally not considered in this chapter, with some exceptions as noted.

7.1.3.8 Summary of Considerations in the Interpretation of Ozone Epidemiologic Studies

The previous sections discussed the topics of exposure assessment and model specification in O₃ epidemiologic studies. Also examined were the issues of hypothesis testing and model selection, as well as the impact of the GAM convergence issue on O₃ risk estimates.

- Exposure measurement error may result from the use of stationary ambient monitors as an indicator of personal exposure in population studies. The relationship between individual personal O₃ exposures and ambient O₃ concentrations was found to be affected by factors such as air exchange rates in housing and time spent outdoors, which varied by individual or by city. However, ambient O₃ concentrations, though they do tend to overestimate personal exposures, were found in general to be well correlated to daily averaged personal O₃ exposures for the populations studied. Therefore population risk estimates derived from ambient O₃ concentrations remain useful and appropriate, but should be evaluated and used with caution in O₃ health risk assessments.
- The three daily O₃ exposure indices used most often in O₃ epidemiologic studies are 1-h max O₃, 8-h max O₃, and 24-h avg O₃. To facilitate comparison of health effects calculated using the different O₃ metrics, risk estimates are standardized or scaled by specified exposure increments: 40 ppb for 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-h avg O₃.
- The lag period between O₃ exposure and the observed health effect may reflect the distribution of effects across time in a population and the potential mechanisms of effects. Among the single-day lags examined, the strongest health effects were observed with a 0- or 1-day lag in O₃ exposure. Multiday lags of O₃ exposure also were investigated. As the parameters estimated from single-day versus multiday models are not the same, interpreting results from these distinct lag models requires caution.
- In time-series studies estimating short-term effects of air pollution, temporal trends and weather effects are two major potential confounders that require consideration. Of all the air pollutants, O₃ is expected to have the strongest correlation with both temporal and meteorological factors. The use of parametric and nonparametric smoothers with varying degrees of freedom per year has been the prevailing approach. Confounding from seasonal variability may be controlled effectively by stratifying analyses by season.
- A major methodological issue affecting O₃ epidemiologic studies concerns the evaluation of the extent to which other air pollutants may confound or modify O₃-related effect estimates. The changing relationship between O₃ and copollutants across seasons further complicates the issue. The use of multipollutant regression models is the prevailing approach for controlling potential confounding by copollutants in O₃ health effects studies.
- In air pollution epidemiologic studies, multiple hypotheses are often tested and certain models are selected for presentation. In many cases, the multiple hypotheses being tested are not independent and, thus, some correlations (e.g., the Bonferroni adjustment) may be overly conservative. To reduce the potential impact of multiple testing errors, emphasis should be placed on a limited number of a priori hypotheses. Additional analyses of varying model specifications considered along with sensitivity analyses.

- GAM is widely used in epidemiologic analyses of air pollution health effects. Recent reports have indicated that the commonly used default convergence criteria lead to biased regression estimates for PM. The limited number of studies that have evaluated this issue for O₃ observed little evidence that default GAM analyses resulted in positively biased estimates.

7.1.4 Approach to Presenting Ozone Epidemiologic Evidence

To produce a thorough appraisal of newly available evidence, key information (including study design, analysis, mean O₃ concentrations, and health outcome results) from important new studies is presented in summary tables in the Chapter 7 Annex (Annex Section AX7.1). Each section of this chapter starts by concisely highlighting important points derived from the 1996 O₃ AQCD assessment. In the main body of the chapter, particular emphasis is focused on studies and analyses that provide pertinent information for the critical assessment of health risks from O₃ exposure. Not all studies are accorded equal weight in the overall interpretive assessment of evidence regarding O₃-associated health effects. Among well-conducted studies with adequate control for confounding, increasing scientific weight is accorded in proportion to the precision of their effect estimates. Small-scale studies without a wide range of exposures generally produce less precise estimates compared to larger studies with a broad exposure gradient. The size of the study, as indicated by the length of the study period and total number of events, and the variability of O₃ exposures are important components that help to determine the precision of the health effect estimates. In evaluating the epidemiologic evidence in this chapter, more weight is accorded to estimates from studies with narrow confidence bands.

Emphasis is placed in the text on the discussion of (1) new multicity studies that employ standardized methodological analyses for evaluating O₃ effects across several or numerous cities and often provide overall effect estimates based on combined analyses of information pooled across multiple cities; (2) studies that consider O₃ as a component of a complex mixture of air pollutants including PM and other gaseous criteria pollutants (CO, NO₂, SO₂); and (3) North American studies conducted in the United States or Canada. Multicity studies are of particular interest and value due to their evaluation of a wider range of O₃ exposures and large numbers of observations. They generally provide more precise effect estimates than most smaller scale studies of single cities. Compared to meta-analyses of multiple “independent” studies, a potential advantage of multicity studies is their consistency in data handling and model

specifications, which eliminates variation due to analysis approach. Also, unlike meta-analyses, they do not suffer from potential omission of nonsignificant results due to “publication bias.” Furthermore, geographic patterns of air pollution effects have the potential to provide especially valuable evidence regarding relative homogeneity and/or heterogeneity of O₃ health effects relationships across geographic locations. Due to the potential for confounding by copollutants, preference is given to studies with effect estimates from multipollutant models, i.e., models with both O₃ and PM rather than O₃-only models. The potential impacts of different health care systems and the underlying health status of populations also need to be accounted for in the assessment (Hubbell et al., 2005; Levy et al., 2001); thus, U.S. studies are emphasized over non-U.S. studies. In accordance to the emphasis placed on the O₃ epidemiologic studies in this chapter, the tables in the Chapter 7 Annex Section AX7.1 were organized by region, with multicity studies in each region presented first.

In the coming sections, field/panel studies and studies of emergency department visits and hospital admissions, which contributed to the establishment of the revised 1997 NAAQS for O₃, are presented first. This is followed by a discussion of O₃-related mortality and effects of chronic exposures to O₃. The chapter ends with an integrative discussion providing a summary and conclusions.

7.2 FIELD STUDIES ADDRESSING ACUTE EFFECTS OF OZONE

7.2.1 Summary of Key Findings on Field Studies of Acute Ozone Effects from the 1996 Ozone AQCD

In the 1996 O₃ AQCD, individual-level camp and exercise studies provided useful quantitative information on the concentration-response relationships linking human lung function declines with ambient O₃ concentrations. The available body of evidence supported a dominant role of O₃ in the observed lung function decrements. Extensive epidemiologic evidence of pulmonary function responses to ambient O₃ has been derived from camp studies. Six studies from three separate research groups provided a combined database on individual concentration-response relationships for 616 children (mostly healthy, nonasthmatic) ranging in age from 7 to 17 years, each with at least six sequential measurements of FEV₁ (forced expiratory volume in 1 second) while attending summer camps (Avol et al., 1990; Higgins et al.,

1990; Raizenne et al., 1987, 1989; Spektor et al., 1988a, 1991). In the combined reanalysis by Kinney et al. (1996a) using consistent analytical methods, these data yielded an average relationship between afternoon FEV₁ and concurrent-hour O₃ concentration of -0.50 mL/ppb (95% CI: -0.63, -0.36), with study-specific slopes ranging from -1.29 to -0.19 mL/ppb. At the time of the afternoon lung function measurement, mean 1-h avg O₃ concentrations ranged from 53 to 123 ppb (maximum range 95 to 245 ppb). Exposure in camp studies usually extended for multiple hours. Although the regression results noted above were based on 1-h O₃ levels, single- and multiple-hour averages were observed to be highly correlated; thus, these results might represent, to some extent, the influence of multihour exposures. In addition to the camp study results, two studies involving lung function measurements before and after well-defined exercise events in adults yielded concentration-response slopes of -0.4 mL/ppb (95% CI: -0.7, -0.1) (Selwyn et al., 1985) and -1.35 mL/ppb (95% CI: -2.04, -0.66) (Spektor et al., 1988b). Ozone concentrations during exercise events of approximately ½-hour duration ranged from 4 to 135 ppb in these studies.

Results from other field/panel studies also supported a consistent relationship between ambient O₃ exposure and acute respiratory morbidity in the population. Respiratory symptoms (or exacerbation of asthma) and decrements in peak expiratory flow (PEF) occurred with increased ambient O₃ concentrations, especially in asthmatic children (Lebowitz et al., 1991; Krzyzanowski et al., 1992). The results showed greater responses in asthmatic than in nonasthmatic individuals (Lebowitz et al., 1991; Krzyzanowski et al., 1992), indicating that asthmatics might constitute a sensitive group in epidemiologic studies of oxidant air pollution. Since the 1996 O₃ AQCD, new research has examined a broad scope of field studies as discussed next.

7.2.2 Introduction to Recent Field Studies of Acute Ozone Effects

Numerous field studies carried out over the past decade have tested for and, in many cases, observed acute associations between O₃ concentrations and measures of respiratory ill-health and O₃ concentrations in groups of subjects (Table AX7-1 in Chapter 7 Annex Section AX7.1). Acute field studies are distinguished from time-series studies in that they are designed to recruit and collect data from individual human subjects instead of utilizing administrative data on aggregate health outcomes such as daily mortality, hospital admissions, or emergency

department visits. Although individual-level health outcome data are collected in field studies, ambient O₃ concentrations from centrally located monitoring stations are generally used to assess exposure. Because of the logistical burden associated with direct data collection from individual subjects, field/panel studies tend to be small in both numbers of subjects and in duration of follow-up. While this may limit the statistical power of field studies as compared with time-series studies, the ability to determine individual-level information on health outcomes and potentially confounding factors adds scientific value.

The most common outcomes measured in acute field studies on the effects of air pollution exposure are lung function and various respiratory symptoms. Other respiratory outcomes examined on a limited basis include inflammation and generation of hydroxyl radicals in the upper airways, as well as numbers of school absences. Also, several studies examined cardiovascular outcomes, including heart rate variability (HRV) and risk of myocardial infarctions (MI). The first group of studies provided varying degrees of evidence supporting the conclusion that elevated O₃ levels have negative impacts on lung function and symptoms, confirming and adding to the body of knowledge that was presented in the 1996 O₃ AQCD. Some emphasis has been placed in examining the independent role of O₃ in the presence of PM and other pollutants. The other new studies contribute information regarding possible O₃-related cardiopulmonary outcomes that have not previously been as well documented.

7.2.3 Effects of Acute Ozone Exposure on Lung Function

As discussed in the 1996 O₃ AQCD and in the earlier chapter of this document on controlled human exposure studies (Chapter 6), a large body of literature from clinical and field studies has clearly and consistently demonstrated reversible decrements in pulmonary function following acute O₃ exposure. Significant O₃-induced spirometric and symptom responses have been observed in clinical studies of exercising healthy young adults (see Section 6.2) and in some potentially susceptible subpopulations, namely asthmatics and children (see Sections 6.3.2 and 6.5.1). Field studies of acute O₃ exposure that examine pulmonary function fall into two distinct groupings, those that conduct spirometry (measuring FEV₁, FVC [forced vital capacity], and other spirometric indices) and those that measure PEF using peak flow meters. Results from the previous O₃ AQCD and Chapter 6 of this document support the conclusion that the spirometric parameter, FEV₁, is a strong and consistent measure of lung function and may be

used in the assessment of asthma (Fuhlbrigge et al., 2001). PEF is a closely related but different metric of lung function. PEF measurements have been shown to be more variable than FEV₁ in some studies (Vaughan et al., 1989; Cross and Nelson, 1991) and can have an element of uncertain reliability when self-administered by study subjects. However, Lippmann and Spektor (1998) state that PEF measurements from small inexpensive flow meters, which are more feasible to use in field studies, can produce similar results to PEF measured spirometrically.

Studies of FEV₁ are assessed here first, followed by a discussion of PEF studies. Other dividing aspects within these two major types of lung function studies include health status of subjects (e.g., healthy, mildly asthmatic, severely asthmatic), age group, time spent outdoors, and exertion levels. Several studies brought these factors together to produce informative data. Some FEV₁ studies involved both increased outdoor O₃ exposure and higher exertion levels. The results from this group of subjects are relatively comparable to those seen with exercising subjects in the clinical studies discussed in Chapter 6.

7.2.3.1 Spirometry (FEV₁) Studies in Outdoor Worker, Exercise, Children, Elderly, and Asthmatic Panels

Studies published over the past decade have provided some new insights on the acute effects of O₃ on FEV₁. The results of all studies that investigated quantitative O₃-related effects on FEV₁ are summarized in the following tables. Tables 7-1a,b,c present changes in FEV₁ associated with O₃ exposure in adults, whereas Tables 7-2a,b,c present effects in children. Borsboom et al. (1999) examined the circadian variation in spirometric parameters and observed that FEV₁, FVC, and PEF, in general, increased from early morning until around noon, then decreased afterwards. Average variations in FEV₁, FVC, and PEF were 2.8%, 4.8%, and 3.1%, respectively. To take into account the circadian variation, several studies stratified their analyses by time of day, examining the effect of O₃ on lung function separately for morning and afternoon measurements. Results from these studies are shown in Tables 7-1b and 7-2b. In other studies, O₃ exposure was related to the cross-day change in spirometric measurements (i.e., the difference between same day morning and afternoon measurements). Results from these studies are presented in Tables 7-1c and 7-2c. Studies that did not provide quantitative O₃ data (Cuijpers et al., 1994; Delfino et al., 2004; Frischer et al., 1997) are not included in the tables. The data presented in Höppe et al. (1995a) were further analyzed in a subsequent paper (Höppe et al., 2003); results from the latter paper are included in the tables. In general, the O₃ effect estimates

Table 7-1a. Field Studies that Investigated the Association Between Acute Ambient O₃ Exposure and Changes in FEV₁ in Adults

Reference	Study Location	Study Period	Mean O ₃ (SD) Level, ppb	O ₃ Index
Korrick et al. (1998)	Mount Washington, NH	Summers 1991, 1992	40 (12)	8-h avg ^a
Brauer et al. (1996)	Fraser Valley, British Columbia, Canada	Jun-Aug 1993	40.3 (15.2)	1-h max
Schindler et al. (2001)	Eight communities in Switzerland	May-Sep 1991	46.6 (1.5-127.6) ^b	8-h avg
Höppe et al. (2003)	Munich, Germany	Apr-Sep 1992-1995	65.9 - 70.4 ^c	½-h max
Romieu et al. (1998)	Mexico City	Mar-May 1996; Jun-Aug 1996	123 (40)	1-h max

^a Average of the hourly O₃ concentrations during each hike. Hikes averaged 8 hours in duration.

^b Range of 8-h avg concentrations is presented by Schindler et al. (2001).

^c Range of mean ½-h max O₃ concentrations on high O₃ days is presented for Höppe et al. (2003).

showed decrements for FEV₁ across studies, especially in children. The studies presented in the tables are discussed in further detail below, starting with O₃ effects on individuals with elevated exertion levels and increased exposure due to time spent outdoors, followed by O₃ effects on persons in other potential risk groups.

Outdoor worker and exercise panels

A very important part of the basis for the current 8-h NAAQS for O₃ was the results from controlled human exposure studies, as discussed in Chapter 6. These field studies with subjects at elevated exertion levels are of particular interest due to their similarities to the human chamber studies. The majority of human chamber studies have examined the effects of O₃ exposure in subjects performing continuous or intermittent exercise for variable periods of time (see Chapter 6 of this document).

A study by Brauer et al. (1996) reported unusually large O₃ effects on lung function among outdoor workers. This study presented O₃ effects observed during an extended outdoor exposure period combined with elevated levels of exertion. The investigators repeatedly measured spirometric lung function before and after outdoor summer work shifts over 59 days among a

Table 7-1b. Percent Changes in FEV₁ (95% CI) Associated with Acute Ambient O₃ Exposures in Adults, Ordered by Size of the Estimate ^a

Reference	Study Population/Analysis	N	% Change in FEV ₁
1 Brauer et al. (1996) ^b	Berry pickers, next morning	58	-6.36 (-8.02, -4.70)
2 Brauer et al. (1996) ^b	Berry pickers, afternoon	58	-5.40 (-6.51, -4.28)
3 Romieu et al. (1998) ^c	Street workers on placebo, 1st phase (lag 0-1)	19	-3.55 (-6.28, -0.82)
4 Schindler et al. (2001)	Adults who never smoked (lag 0)	3,912	-2.96 (-5.11, -0.76)
5 Romieu et al. (1998) ^c	Street workers on placebo, 1st phase (lag 0)	19	-2.17 (-3.45, -0.89)
6 Höppe et al. (2003) ^b	Athletes, afternoon (lag 0)	43	-1.26 (-2.63, 0.10)
7 Romieu et al. (1998) ^c	Street workers on supplement, 1st phase (lag 0-1)	22	-1.25 (-4.36, 1.86)
8 Romieu et al. (1998) ^c	Street workers on supplement, 1st phase (lag 0)	22	-0.53 (-2.08, 1.01)
9 Romieu et al. (1998) ^c	Street workers on placebo, 2nd phase (lag 0)	23	-0.40 (-1.94, 1.14)
10 Romieu et al. (1998) ^c	Street workers on placebo, 2nd phase (lag 0-1)	23	-0.36 (-2.93, 2.20)
11 Höppe et al. (2003)	Elderly, morning (lag 2)	41	-0.22 (-3.86, 3.42)
12 Höppe et al. (2003)	Athletes, morning (lag 0)	43	-0.01 (-0.10, 0.09)
13 Romieu et al. (1998) ^c	Street workers on supplement, 2nd phase (lag 0)	19	0.18 (-0.72, 1.08)
14 Höppe et al. (2003) ^b	Athletes, afternoon (lag 2)	43	0.24 (-0.64, 1.12)
15 Höppe et al. (2003) ^b	Athletes, afternoon (lag 1)	43	0.48 (-0.97, 1.94)
16 Höppe et al. (2003) ^b	Athletes, morning (lag 2)	43	0.62 (-0.45, 1.68)
17 Höppe et al. (2003) ^b	Athletes, morning (lag 1)	43	0.71 (-0.65, 2.07)
18 Höppe et al. (2003)	Elderly, afternoon (lag 0)	41	0.75 (-2.08, 3.58)
19 Romieu et al. (1998) ^c	Street workers on supplement, 2nd phase (lag 0-1)	19	0.82 (-0.77, 2.42)
20 Höppe et al. (2003)	Elderly, afternoon (lag 1)	41	1.16 (-1.26, 3.58)
21 Höppe et al. (2003)	Elderly, morning (lag 0)	41	1.68 (-3.72, 7.07)
22 Höppe et al. (2003)	Elderly, morning (lag 1)	41	1.82 (-2.19, 5.84)
23 Höppe et al. (2003)	Elderly, afternoon (lag 2)	41	2.88 (-0.24, 6.00)

^aChange in FEV₁ is per standard unit ppb O₃ (40 ppb for ½-h max O₃ and 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-hr avg O₃).

^bBrauer et al. (1996) and Höppe et al. (2003) studies also included children. The study population for Brauer et al. ranged in age from 10 to 69 years (mean age 44 years). For Höppe et al. (2003), the athletes ranged in age from 13 to 38 years (mean age 18 years).

^cRomieu et al. (1998) present change in FEV₁ (mL). The data from Romieu et al. (1998) were transformed to percent change by dividing the estimates by 3,300 mL (average FEV₁ for 40 year old Mexican-American males by Hankinson et al., 1999).

Table 7-1c. Cross-day Percent Changes in FEV₁ (95% CI) Associated with Acute Ambient O₃ Exposures in Adults, Ordered by Size of the Estimate ^a

Reference	Study Population/Analysis	N	Cross-day % Change in FEV ₁	
1	Korrick et al. (1998)	Hikers with wheeze or asthma (post-pre-hike)	40	-4.47 (-7.65, -1.29)
2	Korrick et al. (1998)	Hikers who hiked 8-12 hours (post-pre-hike)	265	-2.07 (-3.78, -0.36)
3	Korrick et al. (1998)	Hikers age 28-37 years (post-pre-hike)	185	-2.01 (-3.42, -0.60)
4	Korrick et al. (1998)	Hikers who never smoked (post-pre-hike)	405	-1.77 (-3.24, -0.30)
5	Korrick et al. (1998)	Hikers male (post-pre-hike)	375	-1.65 (-3.12, -0.18)
6	Korrick et al. (1998)	Hikers age 38-47 years (post-pre-hike)	142	-1.59 (-3.12, -0.06)
7	Korrick et al. (1998)	All hikers (post-pre-hike)	530	-1.53 (-2.82, -0.24)
8	Korrick et al. (1998)	All hikers, with PM _{2.5} and acidity in model (post-pre-hike)	530	-1.44 (-3.32, 0.44)
9	Korrick et al. (1998)	Hikers age 18-27 years (post-pre-hike)	135	-1.29 (-2.88, 0.30)
10	Korrick et al. (1998)	Hikers female (post-pre-hike)	155	-1.17 (-3.46, 1.12)
11	Korrick et al. (1998)	Hikers age 48-64 years (post-pre-hike)	68	-1.14 (-3.08, 0.80)
12	Korrick et al. (1998)	Hikers without wheeze or asthma (post-pre-hike)	490	-1.08 (-2.49, 0.33)
13	Korrick et al. (1998)	Hikers who hiked 2-8 hours (post-pre-hike)	265	-0.99 (-2.70, 0.72)
14	Korrick et al. (1998)	Hikers who formerly smoked (post-pre-hike)	125	-0.72 (-3.07, 1.63)
15	Brauer et al. (1996) ^b	Berry pickers (post-pre-work shift)	58	0.00 (-1.66, 1.66)

^aChange in FEV₁ is per standard unit ppb O₃ (40 ppb for ½-h max O₃ and 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-h avg O₃).

^bBrauer et al. (1996) study also included children. The study population ranged in age from 10 to 69 years (mean age 44 years).

group of 58 berry pickers in Fraser Valley, British Columbia, Canada. The subjects, both male and female native Punjabi-speakers, ranged in age from 10 to 69 years old, with a mean age of 44 years. Outdoor work shifts averaged 11 hours in duration. The mean 1-h max O₃ concentration was 40.3 ppb (SD 15.2). Exertion levels were estimated using portable heart rate monitors carried over a period of four or more hours by a representative subset of subjects during 16 work shifts. Heart rates over the work shift averaged 36% higher than resting levels. Post-shift FEV₁ and FVC decreased as a function of O₃ concentration, and the effects of O₃

Table 7-2a. Field Studies that Investigated the Association Between Acute Ambient O₃ Exposure and Changes in FEV₁ in Children

Reference	Study Location	Study Period	Mean O ₃ (SD) Level, ppb	O ₃ Index
Linn et al. (1996)	Rubidoux, Upland, and Torrance, CA	Fall-spring 1992-1993, 1993-1994	23 (12)	24-h avg
Scarlett et al. (1996)	Surrey, England	Jun-Jul 1994	50.7 (24.48)	8-h max
Höppe et al. (2003)	Munich, Germany	Apr-Sep 1992-1995	65.9 - 70.4 ^a	½-h max
Ulmer et al. (1997)	Freudenstadt and Villingen, Germany	Mar-Oct 1994	Freudenstadt: 50.6 (22.5-89.7) ^b Villingen: 32.1 (0.5-70.1) ^b	½-h max ½-h max
Castillejos et al. (1995)	SW Mexico City	Aug 1990-Oct 1991	112.3 (0-365) ^c	1-h max
Romieu et al. (2002)	Mexico City	Oct 1998-Apr 2000	102 (47)	1-h max
Chen et al. (1999)	Sanchun, Taihsi, and Linyuan, Taiwan	May 1995-Jan 1996	19.7 - 110.3 ^c	1-h max

^aRange of mean ½-h max O₃ concentrations on high O₃ days is presented for Höppe et al. (2003).

^bMedian and 90th percentile interval are presented for Ulmer et al. (1997).

^cRange of 1-h max O₃ concentrations are presented by Castillejos et al. (1995) and Chen et al. (1999).

remained significant after adjusting for PM_{2.5} in the analysis (see Table 7-1b). Declines in lung function also were observed on the morning following high O₃ exposure. Ozone was not significantly associated with a cross-day change in lung function (i.e., difference between afternoon and morning spirometry measurements) (see Table 7-1c). One explanation for the lack of an association may be that each spirometric measurement incorporated the impact of the previous day's O₃ in addition to the O₃ effect on that particular day. The effects seen in this study are larger than have been reported previously in studies with briefer exposure durations. For example, a change of -3.8 mL (95% CI: -4.6, -3.0) in afternoon FEV₁ was shown per 1 ppb increase in O₃ concentrations, compared to the decline of 0.4 mL/ppb and 1.35 mL/ppb observed in the earlier adult exercise studies (Spektor et al., 1988b; Selwyn et al., 1985). These results are consistent with the interpretation that extended exposures to O₃ produce more marked effects on lung function. Further, when data were restricted to days with 1-h max O₃ concentrations under 40 ppb, the O₃ effects on afternoon FEV₁ did not change in magnitude and remained significant. However, a possible role of copollutants cannot be completely excluded.

Table 7-2b. Percent Changes in FEV₁ (95% CI) Associated with Acute Ambient O₃ Exposures in Children, Ordered by Size of the Estimate ^a

Reference	Study Population/Analysis	N	% Change in FEV ₁	
1	Ulmer et al. (1997) ^b	School children in Freudenstadt (lag 1)	57	-4.60 (-7.54, -1.67)
2	Ulmer et al. (1997) ^b	School boys in Freudenstadt and Villingen (lag 1)	67	-3.23 (-6.47, 0.00)
3	Ulmer et al. (1997) ^b	School children in Freudenstadt and Villingen (lag 1)	135	-2.98 (-5.33, -0.63)
4	Ulmer et al. (1997) ^b	School girls in Freudenstadt and Villingen (lag 1)	68	-2.32 (-5.53, 0.88)
5	Höppe et al. (2003) ^c	Asthmatics, afternoon (lag 2)	43	-2.08 (-6.24, 2.08)
6	Chen et al. (1999)	Children, with NO ₂ in model (lag 1)	895	-1.97 (-3.51, -0.43)
7	Chen et al. (1999)	Children (lag 1)	895	-1.48 (-2.84, -0.12)
8	Höppe et al. (2003)	Children, morning (lag 0)	44	-1.45 (-4.27, 1.38)
9	Romieu et al. (2002) ^b	Moderate to severe asthmatic children on placebo (lag 1)	35	-0.99 (-1.80, -0.18)
10	Romieu et al. (2002) ^b	Moderate to severe asthmatic children on placebo, with NO ₂ and PM ₁₀ in model (lag 1)	35	-0.97 (-1.87, -0.07)
11	Chen et al. (1999)	Children (lag 2)	895	-0.93 (-2.56, 0.71)
12	Ulmer et al. (1997) ^b	School children in Villingen (lag 1)	78	-0.79 (-3.93, 2.34)
13	Chen et al. (1999)	Children (lag 7)	895	-0.72 (-1.81, 0.37)
14	Höppe et al. (2003) ^c	Asthmatics, afternoon (lag 1)	43	-0.56 (-4.61, 3.50)
15	Linn et al. (1996) ^b	School children, next morning	269	-0.27 (-0.79, 0.24)
16	Linn et al. (1996) ^b	School children, afternoon	269	-0.19 (-0.73, 0.35)
17	Romieu et al. (2002) ^b	All asthmatic children on placebo (lag 1)	78	-0.19 (-0.71, 0.33)
18	Höppe et al. (2003)	Children, afternoon (lag 0)	44	-0.14 (-2.71, 2.42)
19	Höppe et al. (2003) ^c	Asthmatics, afternoon (lag 0)	43	-0.10 (-6.59, 6.39)
20	Romieu et al. (2002) ^b	Moderate to severe asthmatic on supplement (lag 1)	47	-0.04 (-0.80, 0.72)
21	Romieu et al. (2002) ^b	Moderate to severe asthmatic on supplement, with NO ₂ and PM ₁₀ in model (lag 1)	47	-0.01 (-0.82, 0.80)
22	Scarlett et al. (1996) ^d	School children (lag 1)	154	0.01 (-0.20, 0.22)
23	Romieu et al. (2002) ^b	All asthmatic children on supplement (lag 1)	80	0.04 (-0.52, 0.60)
24	Höppe et al. (2003) ^c	Asthmatics, morning (lag 1)	43	0.30 (-3.93, 4.53)

Table 7-2b (cont'd). Percent Changes in FEV₁ (95% CI) Associated with Acute Ambient O₃ Exposures in Children, Ordered by Size of the Estimate ^a

Reference	Study Population/Analysis	N	% Change in FEV ₁
25 Höppe et al. (2003)	Children, morning (lag 1)	44	0.83 (-0.53, 2.20)
26 Höppe et al. (2003)	Children, afternoon (lag 1)	44	0.93 (-0.80, 2.66)
27 Höppe et al. (2003)	Children, morning (lag 2)	44	1.17 (-0.36, 2.70)
28 Höppe et al. (2003)	Children, afternoon (lag 2)	44	1.20 (-0.12, 2.52)
29 Höppe et al. (2003) ^c	Asthmatics, morning (lag 2)	43	1.40 (-3.69, 6.49)
30 Höppe et al. (2003) ^c	Asthmatics, morning (lag 0)	43	3.41 (-2.50, 9.33)

^aChange in FEV₁ is per standard unit ppb O₃ (40 ppb for ½-h max O₃ and 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-h avg O₃).

^bLinn et al. (1996), Romieu et al. (2002), and Ulmer et al. (1997) present change in FEV₁ (mL). The data were transformed to percent change by dividing the estimates by 1,900 mL (average FEV₁ among 8 to 10 year olds by Hankinson et al., 1999).

^cHöppe et al. (2003) study also included young adults. The study population age for the asthmatics ranged from 12 to 23 years (mean age 15 years).

^dFEV_{0.75} results are presented in Scarlett et al. (1996).

Table 7-2c. Cross-day Percent Changes in FEV₁ (95% CI) Associated with Acute Ambient O₃ Exposures in Children, Ordered by Size of the Estimate ^a

Reference	Study Population/Analysis	N	Cross-day % Change in FEV ₁
1 Linn et al. (1996) ^b	School children (p.m. - a.m.)	269	-0.61 (-1.09, -0.14)
2 Castillejos et al. (1995)	Private primary school (post-pre-exercise)	40	-0.48 (-0.72, -0.24)

^aChange in FEV₁ is per standard unit ppb O₃ (40 ppb for ½-h max O₃ and 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-h avg O₃).

^bLinn et al. (1996) present change in FEV₁ (mL). The data were transformed to percent change by dividing the estimates by 1,900 mL (average FEV₁ among 8 to 10 year olds by Hankinson et al., 1999).

In a Mexico City study of 47 outdoor street workers (Romieu et al., 1998), spirometry was performed repeatedly at the end of the work shift over a 2-month period. Subjects were exposed to outdoor ambient O₃ for a mean of 7.4 h during the workday. The mean 1-h max O₃ concentration during the study period was 123 ppb (SD 40). Among those who had never taken an antioxidant supplement (subjects who received a placebo during the first phase of the study),

same day O₃ concentrations were associated with decreases in FEV₁. A mean change of -71.6 mL (95% CI: -113.9, -29.3) (approximately a 4% decline) was observed per 40 ppb increase in 1-h max O₃. The results from this study, in addition to those from the Canadian study of berry pickers (Brauer et al., 1996), indicate that outdoor workers are a potentially vulnerable population that may need protection from O₃ exposures.

Höppe et al. (1995a) examined forestry workers (n = 41) for changes in pulmonary function attributable to O₃ exposure in Munich, Germany. In addition, athletes (n = 43) were monitored in the afternoon following a 2-h outdoor training period. Pulmonary function tests were conducted on days of both “high” (mean ½-h max O₃ of 64 to 74 ppb) and “low” (mean ½-h max O₃ of 32 to 34 ppb) ambient O₃ concentrations. Ventilation rates were estimated from the average activity levels. Athletes, who had a fairly high ventilation rate of 80 L/min, experienced a significant decrease of 60.8 mL (95% CI: 6.4, 115.2) in FEV₁ per 40 ppb increase in mean ½-h max O₃. Among the forestry workers, a similar O₃-related decline in FEV₁ also was observed (-56.0 mL [95% CI: -118.4, 6.4]). In a subsequent study, Höppe et al. (2003) reanalyzed the results of the athletes after stratifying the spirometric data by time of day (morning versus afternoon) and at different lag periods (lags of 0 to 2 days). The reanalysis indicated that O₃-related decrements were observed only with the afternoon FEV₁ at a 0-day lag, -1.26% (95% CI: -2.63, 0.10) change in FEV₁ per 40 ppb increase in mean ½-h max O₃. The mean ½-h max O₃ in the afternoon (1 p.m. to 4 p.m.) was 65.9 ppb (range 51 to 86) on O₃ days.

One FEV₁ study clearly demonstrated small but measurable effects of multihour O₃ exposures on adults exercising outdoors. In Korrnick et al. (1998), adult hikers (n = 530) of Mount Washington, NH performed spirometry before and after hiking for a mean of 8 hours (range 2 to 12). The mean hourly O₃ concentration during each hike ranged from 21 to 74 ppb. After the hike, all subjects combined experienced a small mean decline of 1.5% (95% CI: 0.2, 2.8) in FEV₁ and 1.3% (95% CI: 0.5, 2.1) in FVC per 30 ppb increase in the mean of the hourly O₃ concentration during the hike. In addition, Korrnick et al. (1998) compared hikers who hiked 8 to 12 hours to those who hiked 2 to 8 hours. Among those who hiked longer, the percent change in FEV₁ was more than 2-fold greater per ppb exposure compared to those who hiked only for 2 to 8 hours. Each hour hiked, which may reflect dose, was associated with a decline of 0.3% (p = 0.05) in FEV₁, after adjusting for O₃.

In a Mexico City study, the O₃ effect attributable to exercise was determined using a group of school children (n = 40) who were chronically exposed to moderate to high levels of O₃ (Castillejos et al., 1995). Children were tested up to 8 times between August 1990 and October 1991. Spirometry was performed by the children before and after a 1-h intermittent exercise session outdoors. Outdoor O₃ levels ranged up to 365 ppb, with a mean of 112.3 ppb. Linear trend analyses indicated a relationship between quintiles of O₃ and percent change in lung function. However, stratified analyses indicated that significant changes were observed only with higher quintiles of O₃ exposure (72 to 125 ppb and 183 to 365 ppb). Therefore, children exercising at higher O₃ levels experienced declines in pulmonary function despite the repeated daily exposure to moderate and high levels of O₃ in Mexico City.

Collectively, the above studies confirm and extend clinical observations that prolonged exposure periods, combined with elevated levels of exertion or exercise, may magnify the effect of O₃ on lung function. The most representative data come from the Korrick et al. (1998) hiker study. This U.S. study provided outcome measures stratified by several factors (e.g., gender, age, smoking status, presence of asthma) within a population capable of more than normal exertion.

Panel studies of children, elderly, and asthmatics

Höppe et al. (1995a,b) examined several potentially susceptible populations for changes in pulmonary function attributable to O₃ exposure in Munich, Germany. The forestry workers and athletes were discussed in the previous section. Senior citizens (n = 41) and juvenile asthmatics (n = 43) were also monitored on “low” O₃ (mean ½-h max O₃ of 32 to 34 ppb) and “high” O₃ (mean ½-h max O₃ of 64 to 74 ppb) days. Subjects were requested to stay outdoors for at least 2 hours just before the afternoon pulmonary function test. Clerks (n = 40) were considered the nonrisk control group. Although clerks spent the majority of their time indoors, their outdoor exposures on high O₃ days were similar to that of the four other risk groups. The results showed no significant O₃ effects on the senior citizens. Clinical studies also have consistently shown that seniors are less responsive to O₃ (Bedi et al., 1989; Drechsler-Parks, 1995). Asthmatics and clerks experienced slight reductions in FEV₁ on high O₃ days. Among all risk groups, juvenile asthmatics experienced the largest O₃-related decline in FEV₁, -84.0 mL (95% CI: -196.4, 28.4) per 40 ppb increase in mean ½-h max O₃. To further examine their hypotheses on

characteristics of O₃ risk groups, Höppe et al. (2003) conducted a different analysis on a more expanded data base than utilized in the earlier study. Children were examined as an additional risk group. Mean ½-h max O₃ ranged from 65.9 ppb to 70.4 ppb on O₃ days for each risk group. Höppe et al. (2003) presented both group mean values and analyses on an individual basis. For the group mean values, consistent O₃ effects were not detectable. On an individual basis, a potential pattern of O₃ sensitivity was observed (see Table AX7-1 in Annex Section AX7.1 for details). About 20% of the children and asthmatics were regarded as O₃ responders (i.e., individuals that had greater than 10% change in FEV₁) compared to only 5% of the elderly and athletes. These results indicated that while the majority of the population did not react to O₃ exposure, a small group of susceptible individuals experienced health effects from O₃. The sample size limits quantitative extrapolation to larger populations, but may allow cautious first estimates.

Several other panel studies performed spirometry in children, another potentially susceptible group (Avol et al., 1998; Chen et al., 1999; Cuijpers et al., 1994; Frischer et al., 1997; Linn et al., 1996; Romieu et al., 2002; Scarlett et al., 1996; Ulmer et al., 1997). All studies, with the exception of Avol et al. (1998) and Scarlett et al. (1996), observed a decrease in FEV₁ associated with O₃ exposure. In a cohort of 154 children in Surrey, England, Scarlett et al. (1996) observed no association between ambient O₃ concentrations and FEV_{0.75} (0.2 mL [95% CI: -3.6, 3.9] increase per 30 ppb increase in 8-h max O₃ at a 1-day lag), but noted a small effect of PM₁₀ on lung function. The mean 8-h max O₃ concentration was 50.7 ppb (range 6.8 to 128 ppb). The study by Avol et al. (1998) examined three groups of children, asthmatic (n = 53), wheezy (n = 54), and healthy (n = 103). Ozone levels were reported as being very low (values not provided). The authors advised that noncompliance by the subjects might have been a problem, and further noted limitations in the analysis methods and other aspects of the study design.

One large study measured spirometric lung function in 895 school children in three towns in Taiwan (Chen et al., 1999). The 1-h max O₃ concentrations ranged from 19.7 to 110.3 ppb. Lung function was measured only once for each subject. The authors reported significant associations between diminished FEV₁ and FVC with a 1-day lag of O₃ concentrations. Effect sizes were typical of those observed in past studies, i.e., 0.5 to 1.0 mL decline in FEV₁ per ppb

increase in O₃ concentration. Ozone was the only air pollutant associated with changes in lung function in multipollutant models including SO₂, CO, PM₁₀, and NO₂.

Linn et al. (1996) repeatedly measured spirometric lung function among 269 school children in three southern California communities (Rubidoux, Upland, and Torrance). Lung function was measured over 5 consecutive days, once in each of 3 seasons over 2 school years. Between-week variability was controlled in the analysis by seasonal terms in the model. Statistical power was limited by the relatively narrow range of exposures that were experienced within each week. In addition, the study was restricted to the school year, eliminating most of the “high” O₃ season from consideration. During the study period, 24-h avg O₃ levels at the central monitoring site ranged up to 53 ppb (mean 23 ppb), whereas personal measurements ranged up to 16 ppb (mean 5 ppb). A mean change of -11.6 mL (95% CI: -20.6, -2.6) (approximately a 1% decline) in FEV₁ was observed from morning to afternoon per 20 ppb increase in 24-h avg O₃. Other associations (involving individual morning or afternoon FVC and FEV₁ measurements) went in the plausible direction, but the O₃ effect estimates were much smaller.

Ulmer et al. (1997) examined 135 children aged 8 to 11 years in two towns in Germany from March to October 1994 for O₃ effects on pulmonary function at four time periods. The cross-sectional results at each of the four time points showed limited FVC and no FEV₁ associations. However, the longitudinal analysis, which combined data from all four periods yielded a mean change of -87.5 mL (95% CI: -143.2, -31.7) (approximately a 5% decline) in FEV₁ per 40 ppb increase in ½-h max O₃ for the town with the higher O₃ levels (median ½-h max of 50.6 ppb versus 32.1 ppb). In the cross-sectional analysis, only between-person variability was analyzed. The longitudinal analysis, in which the subjects provided multiple days of measurements, provided information on both between- and within-subject responses.

There are a limited number of new epidemiologic studies examining the effects of O₃ on FEV₁; however, results from these studies indicate that acute exposure to O₃ is associated with declines in FEV₁ in children. These results further support the negative effects of O₃ on lung function observed in the meta-analysis on children attending summer camp (Kinney et al., 1996a) and in the clinical literature.

7.2.3.2 Peak Flow Meter (PEF) Studies in Asthmatics and Healthy Individuals

Many studies of the acute effect of O₃ on PEF examined self-administered PEF levels daily, both in the morning and afternoon. PEF follows a circadian rhythm, with the highest values found during the afternoon and lowest values during the night and early morning (Borsboom et al., 1999). Due to the diurnal variation in PEF, most studies analyzed their data after stratifying by time of day. The peak flow studies examined both asthmatic panels and healthy individuals. The asthma panels are discussed first.

Asthma panels

The effects of acute O₃ exposure on PEF in asthmatics were examined in several panel studies. Figures 7-1 and 7-2 present percent changes in morning and afternoon PEF outcomes from seven panel studies of children, mostly asthmatic, ranging in age from 5 to 13 years. The effect estimates from all single-day and multiday lag models are presented. Only single-city results with analyses stratified by morning and afternoon are included in the figure. Studies that examined cross-day changes and daily variability in PEF (e.g., Just et al., 2002; Thurston et al., 1997) are not included in the figure since such outcomes are not directly comparable. Collectively, nearly all of the studies indicated decrements of peak flow but most of the individual estimates were not statistically significant. The results from the individual studies are further discussed below.

In Mexico City, two studies of asthmatic school children were carried out simultaneously in the northern (Romieu et al., 1996) and southwestern sections of the city (Romieu et al., 1997). In the northern study, 71 mildly asthmatic school children aged 5 to 13 years old, were followed over time for daily morning (before breakfast) and afternoon (bedtime) PEF. The mean 1-h max O₃ was 190 ppb (SD 80). In single-pollutant models, O₃ concentrations at 0-, 1-, and 2-day lags were associated with diminished morning and afternoon PEF, but only the 0-day lag morning effect was significant. The O₃ effect became nonsignificant when PM_{2.5} was added to the model. In the southwestern study, 65 mildly asthmatic children aged 5 to 13 years old were followed during the summer and winter for daily morning and afternoon PEF. The mean 1-h max O₃ was 196 ppb (SD 78). Ozone concentrations at a 0- and 1-day lag were associated with afternoon PEF, with larger effects at a 1-day lag. Associations involving O₃ were stronger than

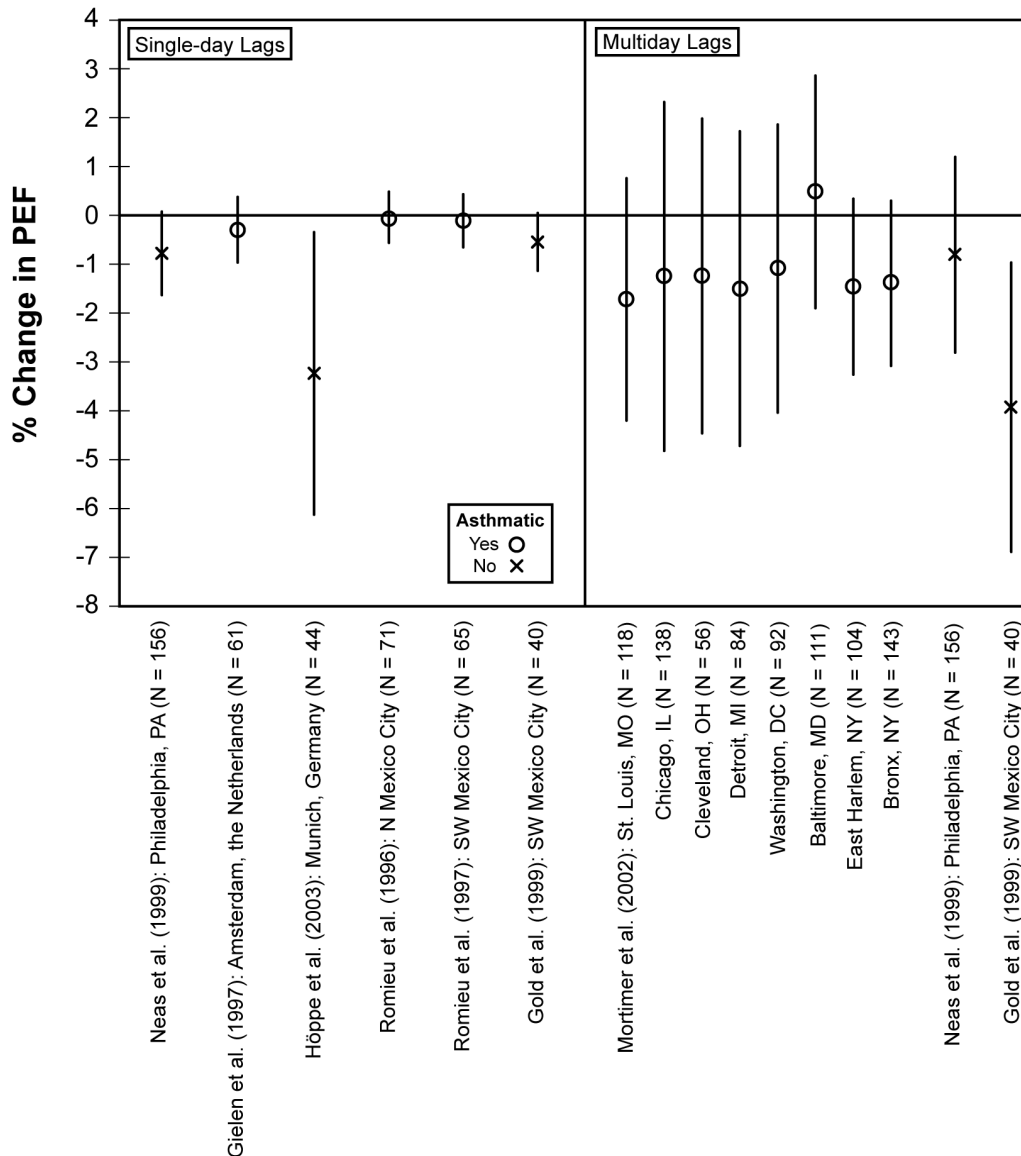


Figure 7-1. Percent change (95% CI) in morning PEF in children per standardized increment (see Section 7.1.3.2). For single-day lag models, previous day O₃ effects are shown. For multiday lag models, the cumulative effects of a 1- to 5-day lag are shown for Mortimer et al. (2002) and Neas et al. (1999), and the effect of a 1- to 10-day lag is shown for Gold et al. (1999).

those involving PM₁₀. Several additional studies, both in the United States and in other countries, reported significant associations between O₃ exposure and decrements in PEF among asthmatics (Gielen et al., 1997; Jalaludin et al., 2000; Just et al., 2002; Ross et al., 2002; Thurston et al., 1997).

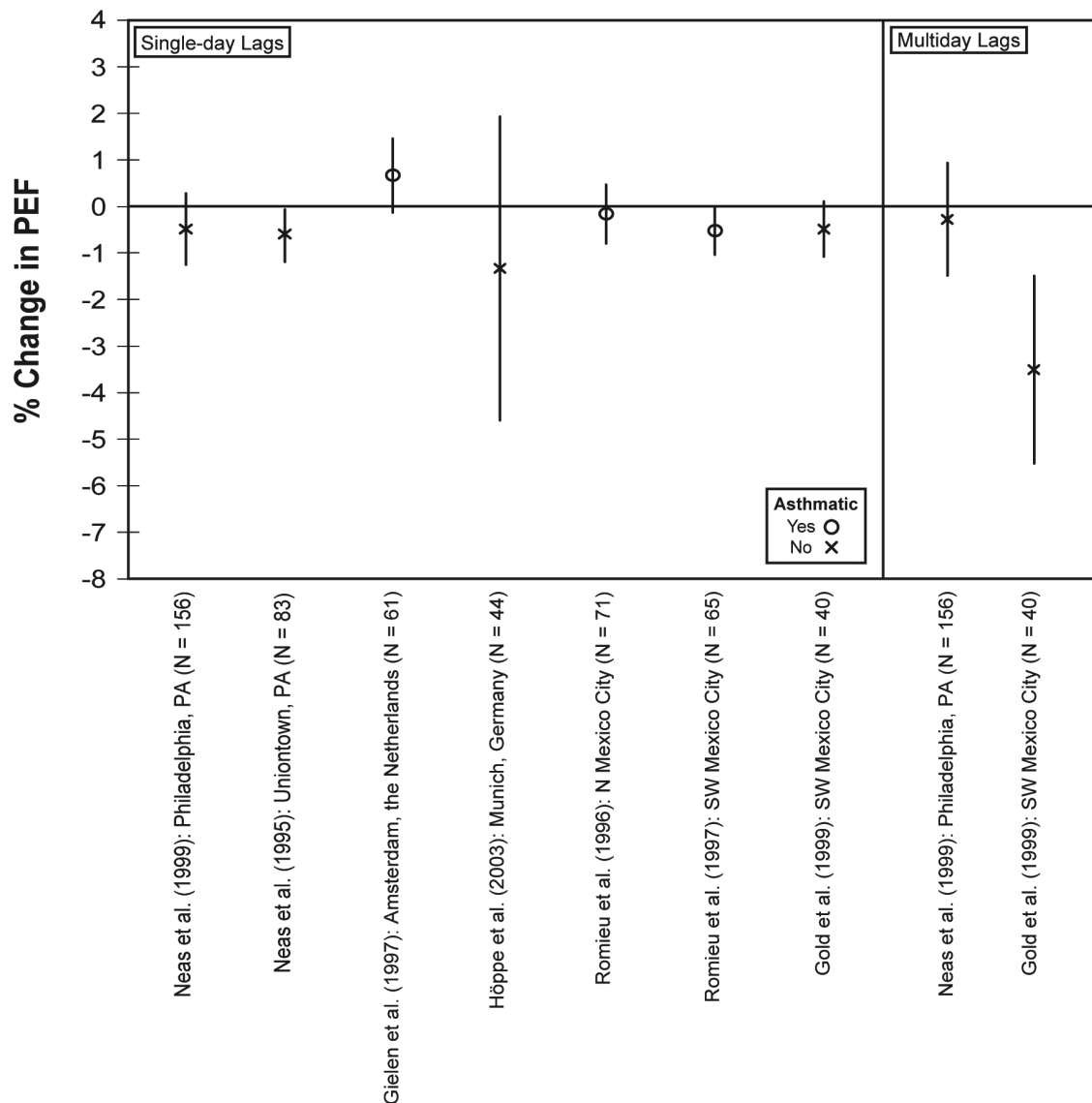


Figure 7-2. Percent change (95% CI) in afternoon PEF in children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O₃ effects are shown. For multiday lag models, the cumulative effect of a 1- to 5-day lag is shown for Neas et al. (1999) and a 1- to 9-day lag is shown for Gold et al. (1999).

Other epidemiologic studies did not find a significant O₃ effect on the lung function of asthmatics. Delfino et al. (1997a) examined morning and evening PEF among 22 asthmatics ranging in age from 9 to 46 years, living in Alpine, CA. Daily ambient 12-h avg O₃ (8 a.m. to 8 p.m.) concentrations ranged from 34 to 103 ppb, with a mean value of 64 ppb. Unique to this study, personal O₃ exposures were measured using 12-h passive O₃ samplers that were worn by

the subjects. The personal 12-h avg O₃ (8 a.m.-8 p.m.) concentrations, which had a mean value of 18 ppb, were much lower than the fixed-site ambient levels. Quantitative O₃ results were not reported, but the researchers stated that no O₃ effects were observed on morning and evening PEF. In Hiltermann et al. (1998), 60 nonsmoking adults aged 18 to 55 years in Bilthoven, the Netherlands, were followed between July and October 1995 with morning and afternoon PEF measurements. Although negative associations were observed between O₃ and cross-day changes in PEF, the results were not significant. The mean ambient 8-h max O₃ was 41.3 ppb (range 3 to 49).

Mortimer et al. (2002) examined 846 asthmatic children from the National Cooperative Inner-City Asthma Study (NCICAS) for O₃-related changes in PEF. Children from eight U. S. urban areas (St. Louis, MO; Chicago, IL; Detroit, MI; Cleveland, OH; Washington, DC; Baltimore, MD; East Harlem, NY; and Bronx NY) were monitored from June through August 1993. Median 8-h avg O₃ (10 a.m.-6 p.m.) concentrations ranged from 34 ppb in Chicago to 58 ppb in Washington, DC. The mean 8-h avg O₃ level across the eight cities was 48 ppb. This study provides representative data for the United States, in so much as children from multiple cities throughout the East and Midwest were examined. Asthmatic children from urban areas are an important subgroup of potentially at-risk populations. Study children either had physician-diagnosed asthma and symptoms in the past 12 months or respiratory symptoms consistent with asthma that lasted more than 6 weeks during the previous year.

Mortimer et al. (2002) examined O₃-related changes in PEF for single-day lags from 1 to 6 days and a multiday lag period of 5 days. Of all the pollutants examined, including O₃, PM₁₀, NO₂, and SO₂, none were associated with evening PEF. Only O₃ was found to be associated with morning PEF. The effect estimates of the association between O₃ and morning PEF for the single-day and multiday lags are depicted as error density curves in Figure 7-3 (for description of error density curves, see Annex Section AX7.2). Small morning effects were observed at 1- and 2-day lags. The effect of O₃ on morning outcomes increased over several days. A strong association between O₃ and PEF also was found with a multiday lag period (cumulative lag of 1 to 5 days). Unrestricted lag models suggested that the O₃ exposure from 3 to 5 days prior had a greater impact on morning % PEF than more immediate exposures. Mortimer et al. discussed biological mechanisms for delayed effects on pulmonary function, which included increased bronchial reactivity secondary to airway inflammation associated with irritant exposure. Animal

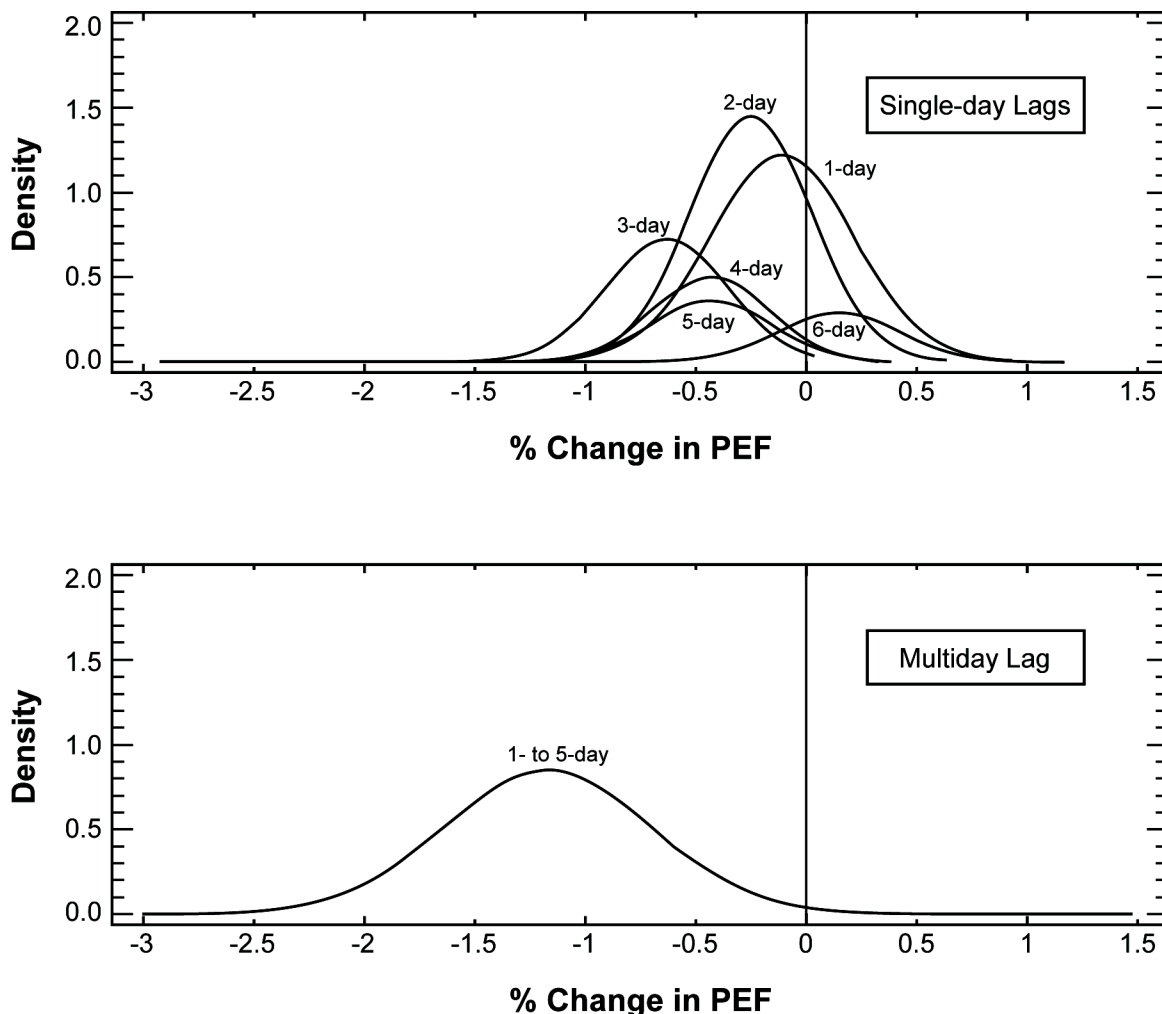


Figure 7-3. Percent changes in PEF per 30 ppb increase in 8-h avg O₃ in urban children. Single-day lags (1-, 2-, 3-, 4-, 5-, and 6-day) are shown in the upper panel. The cumulative multiday lag (1- to 5-day) is shown in the lower panel.

Source: Derived from Mortimer et al. (2002).

toxicology and human chamber studies (see Chapters 5 and 6) provide further evidence that exposure to O₃ may augment cellular infiltration and cellular activation, enhance release of cytotoxic inflammatory mediators, and alter membrane permeability.

Figure 7-4 illustrates the probability density curves of the results from the individual-cities analysis and that from the pooled analysis of all eight cities. The error density curve for the all-cities analysis is a graphical presentation of the all-cities regression analysis presented by Mortimer et al. (2002), a change in morning PEF of -1.18% (95% CI: -2.10, -0.26) per 30 ppb

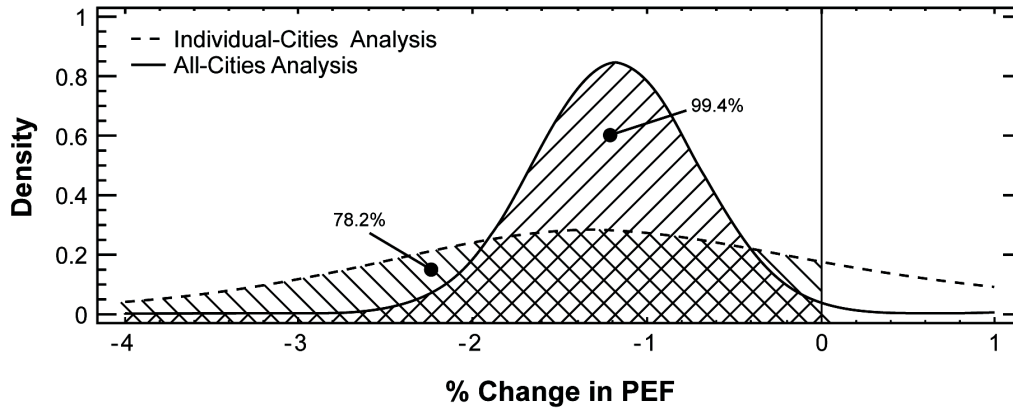


Figure 7-4. Percent change in PEF per 30 ppb increase in 8-h avg O₃ with a cumulative lag of 1 to 5 days. The error density curve is shown for the pooled analysis of the eight NCICAS cities and the summary density curve is presented for the analyses from the individual cities. Note that 99% and 78% of the areas under the curves are less than zero for the pooled-cities analysis and individual-cities analysis, respectively.

Source: Derived from Mortimer et al. (2002).

increase in 8-h avg O₃ with a cumulative lag of 1 to 5 days. The summary density curve for the individual-cities analysis was calculated by summing together eight normal distribution functions, one for each of the study cities, then taking the derivative of the summed function (see Annex Section AX7.2 for further explanation of summary density curves). The area under the density curve and to the left of a value on the x-axis is an estimate of the probability that the effect estimate will be less than or equal to that value. For example, the area under the density curve to the left of 0% change in PEF is 99% in the all-cities analysis. A wider distribution was observed in the individual-cities analysis, with only 78% of the area less than zero. The summary density curve of the individual-cities analysis has a larger standard error than the error density curve of the all-cities analysis, because the summary density does not average the city effects. The regression analysis by Mortimer et al. (2002) suggested a lack of heterogeneity by city, as indicated by the nonsignificant interaction term between O₃ effect and city. As shown in Figure 7-4, the summary density curve of the individual-cities analysis has a peak at about the same value as the curve of the all-cities analysis, suggesting a common O₃ effect for all eight

cities and small variation among them. The unimodal shape of the density curve of the individual-cities analysis also indicates the absence of outlying cities.

Mortimer et al. (2002) further noted that small declines in morning PEF may be of uncertain clinical significance; thus they calculated the incidence of $\geq 10\%$ declines in PEF. A 5 to 15% change in FEV₁ has been expressed as having clinical importance to asthma morbidity (American Thoracic Society, 1991; Lebowitz et al., 1987; Lippmann, 1988). Although greater variability is expected in PEF measurements, a $\geq 10\%$ change in PEF also may have clinical significance. In Mortimer et al. (2002), O₃ was associated with an increased incidence of $\geq 10\%$ declines in morning PEF (odds ratio of 1.30 [95% CI: 1.04, 1.61] per 30 ppb increase in 8-h avg O₃ for a 5-day cumulative lag). This finding suggests that exposure to O₃ might be related to clinically important changes in PEF in asthmatic children. This study also observed that excluding days when 8-h avg O₃ levels were greater than 80 ppb provided effect estimates that were similar to those when all days were included in the analysis, indicating that the negative effect of O₃ on morning PEF persisted at levels below 80 ppb. There is some concern, however, regarding the lack of an association between O₃ and afternoon PEF.

Results from the multicities study by Mortimer et al. (2002), as well as those from several regional studies, provide evidence of a significant relationship between O₃ concentrations and PEF among asthmatics. Collectively, these studies indicate that O₃ may be associated with declines in lung function in this potentially susceptible population.

Panels of healthy subjects

The effect of O₃ on PEF in healthy subjects also was investigated in several studies. A study of 162 children (9 years of age) in England examined the relationship between O₃ and PEF in the winter and summer seasons (Ward et al., 2002). The median 24-h avg O₃ concentrations were 13.0 ppb in the winter and 22.0 ppb in the summer. The O₃ effect estimates were generally positive in the winter and negative in the summer. Single-day lags of 0- to 3-days were examined; however, the strongest association was found with a multiday lag period. During the summer, a decline of 11.10 L/min (95% CI: 0.18, 21.98) was observed in morning PEF per 20 ppb increase in 24-h avg O₃ with a 7-day cumulative lag. Smaller O₃ effects were observed on afternoon PEF.

During the summer of 1990, Neas et al. (1995) examined 83 children in Uniontown, PA and reported twice daily PEF measurements. Researchers found that evening PEF was associated with O₃ levels weighted by hours spent outdoors. Using a similar repeated measures design, Neas et al. (1999) saw evidence for effects due to ambient O₃ exposure among 156 children attending two summer day camps in the Philadelphia, PA area. The mean daytime 12-h avg O₃ (9 a.m. to 9 p.m.) levels were 57.5 ppb at the southwestern camp and 55.9 ppb at the northeastern camp. Negative associations were found between the preceding 12-h avg O₃ and afternoon PEF (recorded before leaving camp), as well as morning PEF (recorded upon arrival at camp). However, the only significant relationship was between O₃ and both morning and afternoon PEF considered jointly when a multiday lag period was used. Naeher et al. (1999), in a sample of 473 nonsmoking women (age 19 to 43 years) living in Vinton, VA, also showed the strongest association between O₃ and evening PEF with a 5-day cumulative lag exposure.

Another study in southwestern Mexico City analyzed morning and afternoon PEF data collected from 40 school children aged 8 to 11 years (Gold et al., 1999). Subjects provided measurements upon arriving and before departing from school each day. The mean 24-h avg O₃ was 52.0 ppb (IQR 25). A negative effect of O₃ on PEF was observed, -1.60 mL/s (95% CI: -3.56, 0.36) and -1.80 mL/s (95% CI: -3.76, 0.16) per 20 ppb increase in 24-h avg O₃ on the same day afternoon and next day morning PEF, respectively. A greater effect was observed for PEF regressed on O₃ concentrations with a cumulative 10-day lag period (-3.50 mL/s [95% CI: -5.52, -1.49] on same day afternoon). These results suggest a longer, cumulative effect of O₃ on PEF. Alternatively, the associations observed at the 10-day lag period may reflect confounding by other time-varying factors or be a chance finding from an exploratory analysis.

In a recent study of 43 mail carriers in Taichung City, Taiwan, PEF was monitored twice daily during a six-week period (Chan and Wu, 2005). The mean 8-h avg O₃ (9 a.m.-5 p.m.) concentration during their work shift was 35.6 ppb (SD 12.1). Associations were observed between evening PEF and O₃ concentrations at lags of 0, 1 and 2 days. The greatest effect was observed at a lag of 1 day, a 2.07% decline in PEF per 30 ppb increase in 8-h avg O₃ (quantitative results for 95% CI not provided). Similar O₃ effects on morning PEF were observed. The effect of O₃ on PEF was robust to adjustment for copollutants; no association with PEF was observed for PM₁₀ and NO₂ in multipollutant models.

7.2.4 Respiratory Symptoms

Studies published over the past decade represent an improved new body of data on the symptom effects of O₃. Respiratory symptoms in acute air pollution field studies are usually measured using questionnaire forms (or “daily diaries”) that are filled out by study subjects, usually without the direct supervision of research staff. Questions address the daily experience of coughing, wheezing, shortness of breath (or difficulty breathing), production of phlegm, and others. While convenient and potentially useful in identifying acute episodes of morbidity, measurements of daily symptoms are prone to a variety of errors. These include, for example, misunderstanding of the meaning of symptoms, variability in individual interpretation of symptoms, inability to remember symptoms if not recorded soon after their occurrence, reporting bias if days of high air pollution levels are identifiable by subjects, and the possibility of falsified data. In spite of these potential problems, the ease of data collection has made daily symptom assessment a common feature of field studies. Many of the studies reviewed above for lung function results also included measurements of daily symptoms. Pearce et al. (1998) reported that one advantage associated with the study of asthma panels is that the population is usually already familiar with symptom terms, e.g., wheezing and cough. Delfino et al. (1998a) further states that the use of repeated daily symptom diaries has additional advantages of reducing recall bias, given the proximity of events, and allowing for health effects to be modeled with each subject serving as their own control over time. Also, study design can blind the participants from the air pollution aspect of the study. Careful efforts by study staff can help ensure that the symptom diaries provide information that is less affected by the potential problems noted above.

Similar to studies of lung function, respiratory symptom studies can be divided into two groups, asthma panels or healthy subjects. Asthma panel studies are discussed first.

Asthma panels

Most studies examining respiratory symptoms related to O₃ exposure focused on asthmatic children. Among the health outcomes, of particular interest were those associated with asthma, including cough, wheeze, shortness of breath, and increased medication use. Figures 7-5 and 7-6 present the odds ratios for O₃-related cough and medication use among asthmatic children from six studies (Gielen et al., 1997; Jalaludin et al., 2004; Just et al., 2002; Ostro et al., 2001; Romieu et al., 1996, 1997). For consistency, only single-city or single-region studies that presented odds

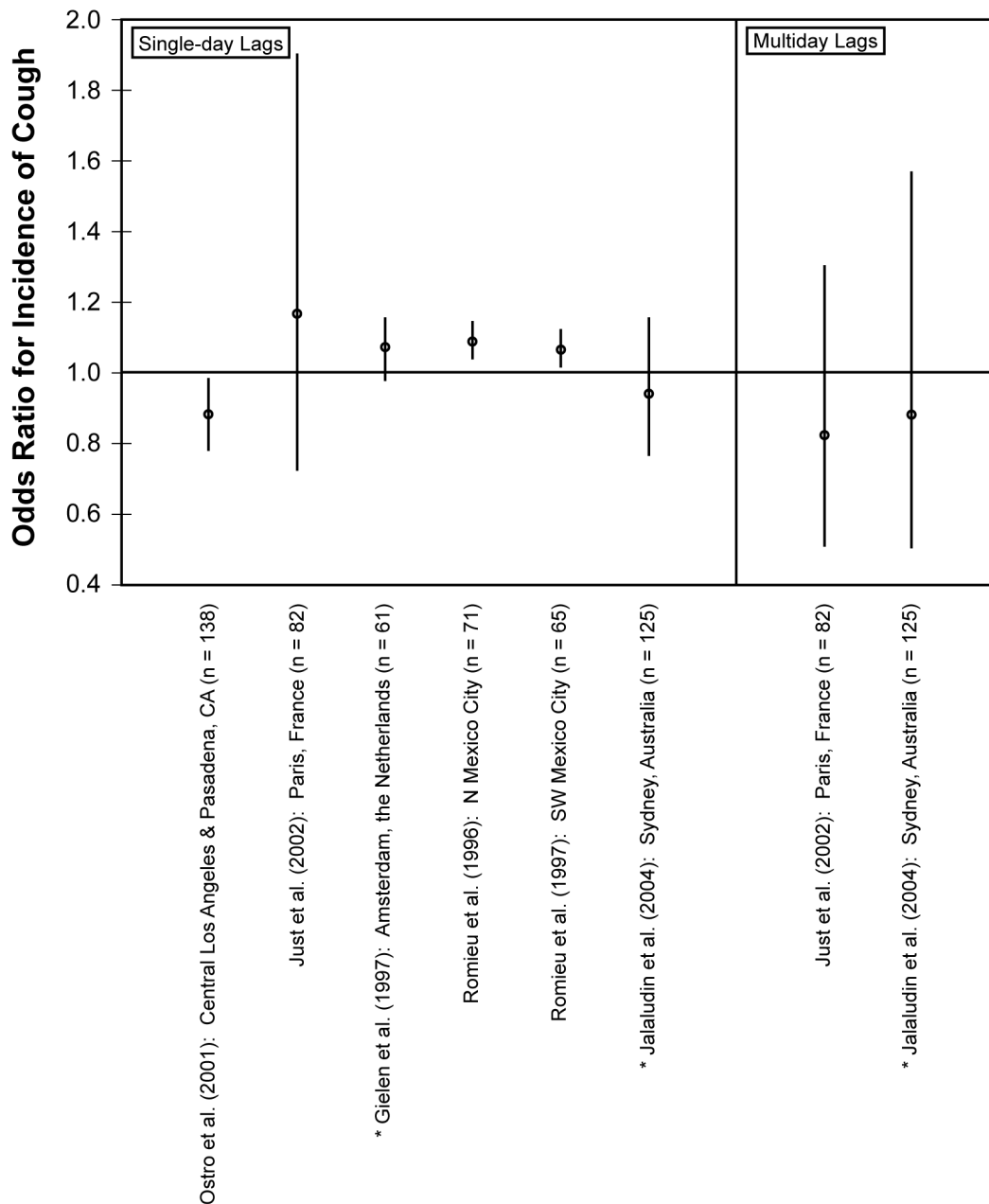


Figure 7-5. Odds ratios for the incidence of cough among asthmatic children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O₃ effects are shown with the exception of Ostro et al. (2001), which only presented results from a 3-day lag. For multiday lag models, the cumulative effects of a 0- to 4-day lag are shown.

*Note that Gielen et al. (1997) and Jalaludin et al. (2004) presented results for prevalence of cough.

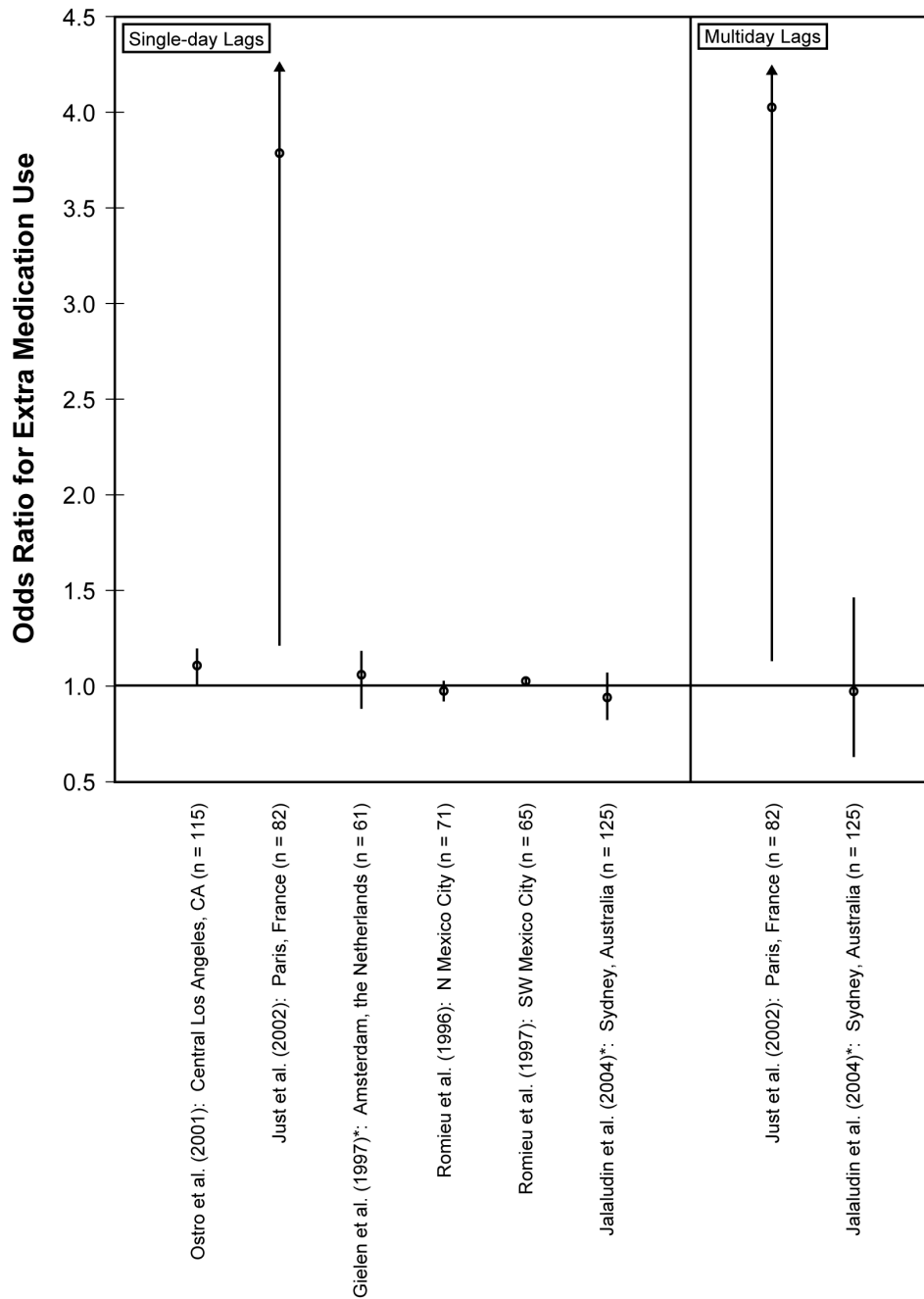


Figure 7-6. Odds ratios for extra medication use among asthmatic children per standardized increment (see Section 7.1.3.2). For single-day lag models, current day O₃ effects are shown. For multiday lag models, the cumulative effects of a 0- to 4-day lag are shown.

ratios are included in the figure. Studies that presented change in severity of symptoms or some other informative health outcome are not included in the figure, since such symptom outcomes differ from indicating simple presence or absence of symptoms. The study by Gent et al. (2003) also is not included in this figure, as odds ratios for cough and medication use were analyzed for quintiles of O₃ concentrations using the lowest quintile as the reference. These studies are discussed separately.

The various effect estimates for the association between O₃ concentrations and cough are depicted in Figure 7-5. Despite the variability in the individual effect estimates, there is some consistency in the O₃ effects. In general, the majority of the odds ratios appear to be greater than one among the single-day lag models, suggesting an association between acute exposure to O₃ and increased cough among asthmatic children. Figure 7-6 presents the odds ratios for O₃-associated bronchodilator use. The results for medication use are less consistent than those for cough; one study by Just et al. (2002) observed strong positive associations, but had wide confidence intervals.

Among the studies reporting results for daily symptoms and asthma medication use, several observed associations with O₃ concentrations that appeared fairly robust (Delfino et al., 2003; Desqueyroux et al., 2002a,b; Gent et al., 2003; Hiltermann et al., 1998; Just et al., 2002; Mortimer et al., 2000, 2002; Newhouse et al., 2004; Romieu et al., 1996, 1997; Ross et al., 2002; Thurston et al., 1997). Mortimer et al. (2002) reported morning symptoms in 846 asthmatic children from 8 U.S. urban areas to be most strongly associated with a cumulative 1- to 4-day lag of O₃ concentrations in the NCICAS. The NCICAS used standard protocols that included instructing caretakers of the subjects to record symptoms in the daily diary by observing or asking the child (Mitchell et al., 1997). Symptoms reported included cough, chest tightness, and wheeze. In the analysis pooling individual subject data from all eight cities, the odds ratio for the incidence of symptoms was 1.35 (95% CI: 1.04, 1.69) per 30 ppb increase in 8-h avg O₃ (10 a.m.-6 p.m.). The mean 8-h avg O₃ was 48 ppb across the 8 cities. Excluding days when 8-h avg O₃ was greater than 80 ppb (less than 5% of days), the odds ratio was 1.37 (95% CI: 1.02, 1.82) for incidence of morning symptoms. Figure 7-7 presents the probability density curves of the odds ratios for the incidence of symptoms with a 1- to 4-day cumulative lag from the individual-cities analysis and the all-cities analysis. This figure confirms the regression results indicating that there is a significant increase in odds for incidence of symptoms, as the area

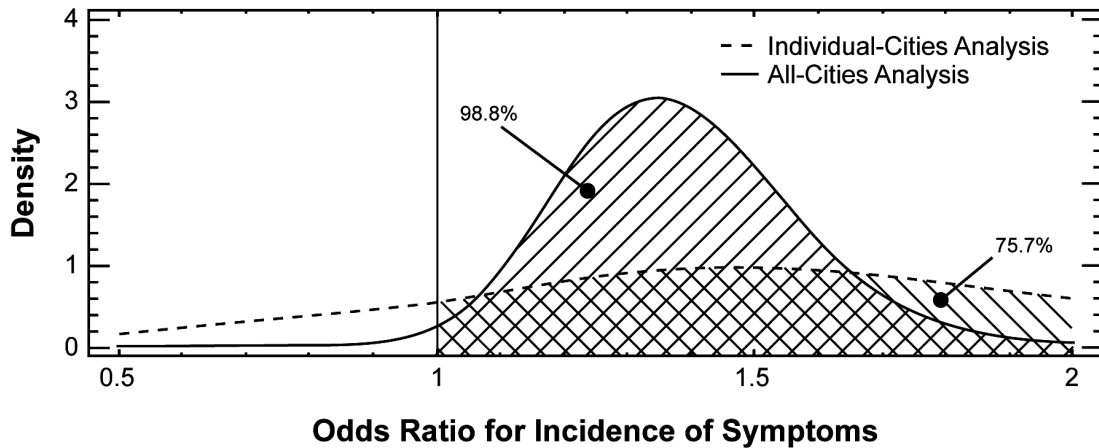


Figure 7-7. Odds ratio for the incidence of symptoms per 30 ppb increase in 8-h avg O_3 with a cumulative lag of 1 to 4 days. The error density curve is shown for the pooled analysis of the eight NCICAS cities and the summary density curve is presented for the analyses from the individual cities. Note that 99% and 76% of the areas under the curves are less than zero for the pooled-cities analysis and individual-cities analysis, respectively.

Source: Derived from Mortimer et al. (2002).

under the density curve with an odds ratio greater than one is 99%. Mortimer et al. (2002) did not observe significant interactions among the eight cities, indicating that there was no heterogeneity among the city-specific estimates. The unimodal distribution of the city-stratified summary density curve also suggests a lack of significant heterogeneity in O_3 effects among the cities. It should be noted that other pollutants, including PM_{10} (monitored in 3 cities), NO_2 (in 7 cities), and SO_2 (in all 8 cities), also were associated with increased incidence of morning symptoms. In multipollutant models, the O_3 effect was shown to be slightly diminished. The odds ratios for the incidence of symptoms per 30 ppb increase in 8-h avg O_3 were 1.23 (95% CI: 0.94, 1.61) with SO_2 and 1.14 (95% CI: 0.85, 1.59) with NO_2 . In the three urban areas with PM_{10} data, the odds ratios were 1.21 (95% CI: 0.61, 2.40) in the O_3 -only model and 1.08 (95% CI: 0.41, 2.40) when PM_{10} also was included in the model.

Another one of the larger studies was that of Gent and colleagues (2003), where 271 asthmatic children under age 12 and living in southern New England were followed over 6 months (April through September) for daily symptoms in relation to O_3 and $PM_{2.5}$. Mean 1-h max O_3 and 8-h max O_3 concentrations were 58.6 ppb (SD 19.0) and 51.3 ppb (SD 15.5),

respectively. The data were analyzed for two separate groups of subjects, 130 who used maintenance asthma medications during the follow-up period and 141 who did not. The need for regular medication was considered to be a proxy for more severe asthma. Not taking any medication on a regular basis and not needing to use a bronchodilator would suggest the presence of very mild asthma. Effects of 1-day lag O₃ were observed on a variety of respiratory symptoms only in the medication user group. Both daily 1-h max and 8-h max O₃ concentrations were similarly related to symptoms such as chest tightness and shortness of breath. Effects of O₃, but not PM_{2.5}, remained significant and even increased in magnitude in two-pollutant models. Some of the associations were noted at 1-h max O₃ levels below 60 ppb. In contrast, no effects were observed among asthmatics not using maintenance medication. In terms of person-days of follow-up, this is one of the larger studies currently available that address symptom outcomes in relation to O₃, and provides supportive evidence for effects of O₃ independent of PM_{2.5}. Study limitations include limited control for meteorological factors and the post-hoc nature of the population stratification by medication use.

Some international studies have reported significant associations of respiratory symptoms with O₃. The incidence of asthma attacks was associated with O₃ concentrations in a group of 60 severe asthmatics (mean age 55 years) followed over a 13-month period in Paris, France (Desqueyroux et al., 2002a). In a similar study, Desqueyroux et al. (2002b) observed O₃-associated exacerbation of symptoms in 39 adult patients (mean age 67 years) with chronic obstructive pulmonary disease (COPD). Interestingly, in contrast to results from controlled human exposure studies (see Section 6.3.1, Subjects with COPD), the O₃ effect appeared larger in this study among subjects who smoked and among those with more severe COPD. However, the low O₃ concentrations experienced during this study (summer mean 8-h max O₃ of 21 ppb [SD 9]) raise plausibility questions. In a study of 60 nonsmoking asthmatic adults (aged 18 to 55 years) in Bilthoven, the Netherlands, Hiltermann and colleagues (1998) reported associations between O₃ and daily symptoms of shortness of breath and pain upon deep inspiration. The mean 8-h max O₃ was 41 ppb (range 3 to 49). The O₃ associations were stronger than those for PM₁₀, NO₂, SO₂, and black smoke (BS). No differences in response were evident between subgroups of subjects defined on the basis of steroid use or airway hyperresponsiveness. Daily use of bronchodilators or steroid inhalers was not found to be associated with O₃ in this study.

Other studies showed only limited or a lack of evidence for symptom increases being associated with O₃ exposure (Avol et al., 1998; Chen et al., 1998; Delfino et al., 1996, 1997a, 1998a; Gielen et al., 1997; Jalaludin et al., 2004; Ostro et al., 2001; Taggart et al., 1996). Avol et al. (1998) studied symptoms in asthmatic, wheezy, and healthy children aged 10 to 12 years in southern California. Some symptom associations were noted but they were inconsistent. For example, children with wheeze were at increased risk for difficulty of breathing and for wheezing at low O₃ concentrations (1-h max O₃ of 30 ppb), but not at higher O₃ concentrations (1-h max O₃ of 120 ppb). The authors noted that O₃ concentrations were relatively low and that children studied did not spend substantial time engaged in physical activities outdoors. Ostro et al. (2001) reported no associations between daily symptoms and ambient O₃ concentrations in a cohort of 138 African-American children with asthma followed over 3 months (August to October) in Central Los Angeles and Pasadena, CA. However, the use of extra asthma medication was associated with 1-h max O₃ concentrations at a 1-day lag. Mean 1-h max O₃ concentrations were 59.5 ppb in Los Angeles and 95.8 ppb in Pasadena. Delfino and colleagues (1996) followed 12 asthmatic teens living in San Diego, CA for respiratory symptoms over a 2-month period and saw no relationship with central site ambient O₃ (mean 12-h avg O₃ of 43 ppb [SD 17]). Personal O₃ exposures (mean 12-h avg O₃ of 11.6 ppb [SD 11.2]) were associated with the composite symptom score and β_2 -agonist inhaler use, but the relationship with symptom score disappeared when weekday/weekend differences were controlled in the statistical analysis. Study power was likely compromised by the small sample size. This observation of stronger associations with O₃ levels from personal monitors implies that gains in power may be achieved if exposure misclassification is reduced through the use of personal exposure measurements rather than central site ambient O₃ concentrations. A similar study of 22 asthmatics in Alpine, CA observed no effects of O₃ on symptoms when personal O₃ exposure (mean 12-h avg O₃ of 18 ppb) was used as the exposure metric (Delfino et al., 1997a). However, a later study in the same location involving 24 subjects (Delfino et al., 1998a) did find an association between respiratory symptoms and ambient O₃ concentrations (mean 1-h max O₃ of 90 ppb), with stronger O₃ effects experienced by asthmatics not on anti-inflammatory medication. In this study, a binary symptom score was used, whereas the earlier study used a linear symptom score of 0 through 6.

The multicities study by Mortimer et al. (2002), which provides an asthmatic population most representative of the United States, and several single-city studies indicate a robust association of O₃ concentrations with respiratory symptoms and increased medication use in asthmatics. However, there are a number of well-conducted, albeit relatively smaller, studies that have not found such effects.

Panels of healthy subjects

Fewer studies examined the effect of O₃ on respiratory symptoms in healthy individuals. Neas et al. (1995) reported that, in school children, evening cough was associated with O₃ levels weighted by hours spent outdoors. The mean daytime 12-h avg O₃ (8 a.m.-8p.m.) concentration was 50.0 ppb. The study by Linn and colleagues (1996) of 269 school children in southern California found no associations between respiratory symptoms and O₃, but subjects were exposed to fairly low O₃ concentrations (mean 24-h avg O₃ of 5 ppb [SD 3]) as determined by personal monitors. Gold et al. (1999) examined symptoms in 40 healthy children in southwest Mexico City. The mean ambient 24-h avg O₃ concentration was 52.0 ppb (IQR 25). Pollutant exposures were associated with increased production of phlegm in the morning, although the effects of the air pollutants (PM_{2.5}, PM₁₀, and O₃) could not be separated in multipollutant models. Hoek and Brunekreef (1995) did not find a consistent association between ambient O₃ levels and the prevalence or incidence of respiratory symptoms in children living in two rural towns in the Netherlands. Mean 1-h max O₃ concentrations were 57 ppb (range 22 to 107) in Deurne and 59 ppb (range 14 to 114) in Enkhuizen. Collectively, these studies indicate that there is no consistent evidence of an association between O₃ and respiratory symptoms among healthy children.

7.2.5 Acute Airway Inflammation

Acute airway inflammation has been shown to occur among adults exposed to 80 ppb O₃ over 6.6 hours with exercise in controlled chamber studies (Devlin et al., 1991). Kopp and colleagues (1999) attempted to document inflammation of the upper airways in response to summer season O₃ exposures by following a group of 170 school children in two towns in the German Black Forest from March to October of 1994. To assess inflammation, nasal lavage samples were collected at 11 time points spanning the follow-up period. The mean ½-h max O₃

concentrations were 33 ppb (5th % to 95th %: 1 to 72) in Villingen and 54 ppb (5th % to 95th %: 23 to 92) in Freudenstadt. The nasal lavage samples were analyzed for markers of inflammation, including eosinophil cationic protein, albumin, and leukocyte counts. Subjects who were sensitized to inhaled allergens were excluded. When analyzed across the entire follow-up period, no association was detected between upper airway inflammation and O₃ concentrations. More detailed analysis showed that the first significant O₃ episode of the summer was followed by a rise in eosinophil cationic protein levels; however, subsequent and even higher O₃ episodes had no effect. These findings suggest an adaptive response of inflammation in the nasal airways that is consistent with controlled human exposure studies (see Section 6.9, Effects of Inflammation and Host Defense).

Frischer and colleagues (1993) collected nasal lavage samples from 44 school children in Umkirch, Germany the morning after “low” O₃ days (<140 µg/m³ or approximately 72 ppb) and “high” O₃ days (>180 µg/m³ or approximately 93 ppb) to measure levels of biochemical markers of inflammation. The researchers found that higher O₃ levels were associated with increased polymorphonuclear leukocyte counts in all children, and increases in myeloperoxidases and eosinophilic cation proteins among children without symptoms of rhinitis (n = 30). These results indicated that O₃ was associated with inflammation in the upper airways. Frischer et al. (1997) further investigated whether hydroxyl radical attacks played a role in mediating the O₃-associated inflammatory response of the airways. *Ortho*- and *para*-tyrosine levels were measured in the nasal lavage samples and the *ortho/para* radical ratio was used to determine the generation of hydroxyl radicals. Significant increases in the *ortho/para* ratio were observed on days following high ambient O₃ levels. However, the *ortho/para* ratio was not related to polymorphonuclear leukocyte counts, suggesting that there was no detectable relationship between hydroxyl radical attacks and the inflammatory response seen in these children. Similar to the study by Kopp et al. (1999), the *ortho/para* ratio decreased at the end of the summer, although O₃ concentrations were still high—thus providing additional evidence for a possible adaptive response. These findings, however, do not preclude the possibility that other unmeasured effects, including cell damage or lower airway responses, may have occurred with ongoing summer season exposures. In fact, a study of joggers repeatedly exposed to O₃ while exercising over the summer in New York City suggested that cell damage may occur in the absence of ongoing inflammation (Kinney et al., 1996b).

In two Mexico City studies by Romieu et al. (1998, 2002), the effect of antioxidant supplements on the association between O₃ and lung function in outdoor workers and asthmatic children was investigated. Mean 1-h max O₃ concentrations were 123 ppb for the outdoor workers and 102 ppb for the children. Romieu and colleagues (1998) observed significant inverse associations between O₃ and lung function parameters, including FVC, FEV₁, and FEF₂₅₋₇₅ (forced expiratory flow at 25 to 75% of FVC), among outdoor workers who received the placebo but not among those taking the antioxidant supplement during the first phase of testing. Likewise, O₃ concentrations were associated with declines in lung function among children with moderate-to-severe asthma who were on the placebo, but no associations were found among those who were taking the vitamin C and E supplement (Romieu et al., 2002). These results indicate that supplementation with antioxidants may modulate the impact of O₃ exposure on the small airways of two potentially at-risk populations, i.e., outdoor workers and children with moderate-to-severe asthma. In a further analysis of the study conducted in asthmatic children, genetic factors were found to contribute to the variability between individuals in the effects of O₃ on lung function (Romieu et al., 2004). Individuals with polymorphism of the glutathione S-transferase gene (GSTM1 null genotype) lack glutathione transferase enzyme activity, which plays an important role in protecting cells against oxidative damage. Results from this analysis indicate that asthmatic children with GSTM1 null genotype were found to be more susceptible to the impact of O₃ exposure on small airways function. Romieu et al. (2004) noted that supplementation with the antioxidant vitamins C and E above the minimum daily requirement might compensate for the genetic susceptibility.

7.2.6 Acute Ozone Exposure and School Absences

The association between school absenteeism and ambient air pollution was assessed in a limited number of studies (Chen et al., 2000; Gilliland et al., 2001). In the study by Chen and colleagues (2000), daily school absenteeism was examined in 27,793 students (kindergarten to sixth grade) from 57 elementary school students in Washoe County, NV over a 2-year period. The mean 1-h max O₃ was 37.5 ppb (SD 13.4) during the study period. One major limitation of this study was that the percent of total daily absences was the outcome of interest, not illness-related absences, because reasons for absences were not noted in all schools. In models adjusting for PM₁₀ and CO, ambient O₃ levels were associated with school absenteeism. With a

distributed lag of 1 to 14 days, O₃ concentrations were associated with a 10.4% (95% CI: 2.7, 18.1) excess rate of school absences per 40 ppb increase in 1-h max O₃. PM₁₀ and CO concentrations also were associated with school absenteeism; but, the effect estimate for PM₁₀ was negative. The inverse relationship between O₃ and PM₁₀ may have partially attributed to the negative association observed between PM₁₀ and school absenteeism.

Ozone-related school absences also were examined in a study of 1,933 fourth grade students from 12 southern California communities participating in the Children's Health Study (Gilliland et al., 2001). Due to its comprehensive characterization of health outcomes, this study is valuable in assessing the effect of O₃ on illness-related school absenteeism in children. The study spanned a period, January through June 1996, that captured a wide range of exposures while staying mostly below the highest levels observed in the summer season. Mean 8-h avg O₃ (10 a.m.-6 p.m.) concentrations ranged from 35 to 55 ppb across the 12 communities. All school absences that occurred during this period were followed up with phone calls to parents to determine whether they were illness-related. For illness-related absences, further questions assessed whether the illness was respiratory or gastrointestinal, with respiratory symptoms including runny nose/sneeze, sore throat, cough, earache, wheezing, or asthma attacks. Multiple pollutants were measured at a central site in each of the 12 communities. A two-stage GAM was used to examine the effects of O₃ on school absences. The analysis controlled for temporal cycles, day of week, and temperature, and expressed exposure as a distributed lag out to 30 days. The 30-day distributed lag was found to best fit the data. Associations were found between the 30-day distributed lag of 8-h avg O₃ (10 a.m.-6 p.m.) and all illness-related absence categories. A 108% (95% CI: 29, 235) increase in illness-related absences was observed with a 30 ppb increase in 8-h avg O₃. Larger O₃ effects were seen for respiratory causes (147% [95% CI: 6, 478]) than for nonrespiratory causes (61% [95% CI: 9, 138]). Among the respiratory absences, larger effects were seen for lower respiratory diseases than for upper respiratory diseases. Multipollutant analyses were not performed; however, in single-pollutant models neither PM₁₀ or NO₂ were associated with any respiratory or nonrespiratory illness-related absences. Some concern exists regarding the possibility of residual seasonal confounding given the 6-month time span of the monitoring period and the long lag periods of exposure, which are likely to capture seasonally changing factors such as pollen episodes. Further, the biological relevance of O₃ concentrations lagged 30 days present an interpretive challenge.

The school absenteeism data from the Children's Health Study was used to illustrate alternative new modeling approaches in two subsequent papers. Berhane and Thomas (2002) noted that systematic comparison of modeling approaches could potentially provide more insight into the modeling process. They used a two-stage iteratively weighted filtered least squares model which allowed explicit modeling of the autoregressive structure and overdispersion; this model provided information pertaining to how air pollution operates in various communities. Results indicated that the overall model might not be adequate for some communities, but the community-specific correction for overdispersion and autocorrelation compensated for this shortcoming. For a 30-day distributed lag, the effect estimate for illness-related absences was similar, 105% (95% CI: -10, 368) increase per 30 ppb increase in 8-h avg O₃. The largest O₃ effect was observed in the communities with low long-term O₃, PM₁₀, and NO₂ concentrations. Once again, PM₁₀ was not associated with illness-related absences. Analyses for specific illness categories were not conducted. Berhane and Thomas (2002) also examined effects from long-term exposure to air pollutants. Long-term average concentrations of PM₁₀, but not O₃, were associated with increases in illness-related absenteeism.

A further analysis by these authors (Rondeau et al., 2005) used another new approach, a three-stage logistic transition model. Unlike the previous two analyses, this analysis allowed the simultaneous examination of the effects of daily exposure to air pollution and individual risk factors using binary time-series data structures, without aggregating over subjects or time. In contrast to the results from Berhane and Thomas (2002), the Rondeau et al. results suggested a chronic effect of O₃ on school absenteeism (41% [95% CI: 2, 95] increase per 30 ppb increase in 8-h avg O₃ over a 5-day lag period) but no acute effect (-0.2% [95% CI: -24, 31] with a 30-day distributed lag) after adjustment for individual factors. The acute O₃ effect on respiratory absences was positive but nonsignificant (12% [95% CI: -22, 62]). Rondeau et al. did not compare their results to those from Gilliland et al. Statistical tests conducted to compare the effect estimates indicated that the acute O₃ estimates for total illness-related absences were significantly different (p = 0.03) in the two studies; however, the estimates for respiratory absences were not found to be different (p = 0.15). Both acute and chronic exposure to PM₁₀ was not associated with illness-related school absences. The authors noted that the different results compared to the analyses by Gilliland et al. (2001) and Berhane and Thomas (2002) might be due to different modeling approaches or the use of binary instead of Poisson time-series

data structures. The two methods papers by Berhane and Thomas (2002) and by Rondeau et al. (2005) focused on introducing new methods and used the Children's Health Study data for illustrative purposes, as opposed to comprehensive analysis of the substantive findings. Hence, greatest emphasis should be placed on the results from Gilliland et al (2001).

Limited evidence from Chen et al. (2000) and Gilliland et al. (2001) suggest that ambient O₃ concentrations, accumulated over 2 to 4 weeks, may be associated with school absenteeism, particularly illness-related absences. Further replication is needed before firm conclusions can be reached regarding the effect of O₃ on school absences.

7.2.7 Cardiovascular Endpoints

Several air pollution studies have examined various cardiovascular endpoints (see Table AX7-2 in Annex 7, Section AX7.1). The earlier studies focused on PM effects. For a more thorough discussion of these PM studies and their health endpoints, refer to Section 8.3.1 of the 2004 PM AQCD (U.S. Environmental Protection Agency, 2004). More recent studies have examined associations of O₃ and other pollutants with various measures of heart beat rhythms in panels of elderly subjects, as discussed below. Other studies examined the increased risk of MI related to air pollutant exposures.

7.2.7.1 Cardiac Autonomic Control

Alterations in heart rate and/or rhythm are thought to reflect pathophysiologic changes that may represent possible mechanisms by which ambient air pollutants such as O₃ may exert acute effects on human health. Decreased HRV has been identified as a predictor of increased cardiovascular morbidity and mortality. Brook et al. (2004) state that HRV, resting heart rate, and blood pressure are modulated by a balance between the two determinants of autonomic tone (the sympathetic and parasympathetic nervous systems). They noted that decreased HRV predicts an increased risk of cardiovascular morbidity and mortality in the elderly and those with significant heart disease, which is generally determined by analyses of time (e.g., standard deviation of normal R-R intervals) and frequency domains (e.g., low frequency/high frequency ratio by power spectral analysis, reflecting autonomic balance) measured during 24 hours of electrocardiography. Decreased parasympathetic input to the heart may provide an important

mechanistic link between air pollution and cardiovascular mortality by promoting fatal tachyarrhythmias.

The potentially adverse effects of air pollutants on cardiac autonomic control were examined in a large population-based study, among the first in this field. Liao et al. (2004) investigated short-term associations between ambient pollutants and cardiac autonomic control from the fourth cohort examination (1996-1998) of the population-based Atherosclerosis Risk in Communities Study (ARIC). PM_{10} , O_3 , and other gaseous air pollutants were examined in this study. PM_{10} (24-h avg) and O_3 exposures (8-h avg, 10 a.m. to 6 p.m.) one day prior to the randomly allocated examination date were used. The mean 8-h avg O_3 level was 41 ppb (SD 16). They calculated 5-minute HRV indices between 8:30 a.m. and 12:30 p.m. and used logarithmically-transformed data on high-frequency (0.15 to 0.40 Hz) and low-frequency (0.04 to 0.15 Hz) power, standard deviation of normal R-R intervals, and mean heart rate. The effective sample sizes for O_3 and PM_{10} were 5,431 and 4,899, respectively, from three U.S. study centers in North Carolina, Minnesota, and Mississippi. PM_{10} concentrations measured one day prior to the HRV measurements were inversely associated with both frequency- and time-domain HRV indices. Ambient O_3 concentrations were inversely associated with high-frequency power among whites. Consistently more pronounced associations were suggested between PM_{10} and HRV among persons with a history of hypertension. Liao et al. (2004) noted that these findings may represent potentially important arrhythmogenic mechanisms of ambient air pollution. The acute adverse effect of air pollution on cardiac autonomic control is based on the hypothesis that increased air pollution levels may stimulate the autonomic nervous system and lead to an imbalance of cardiac autonomic control characterized by sympathetic activation unopposed by parasympathetic control. Such an imbalance of cardiac autonomic control may predispose susceptible people to greater risk of life-threatening arrhythmias and acute cardiac events. The Liao et al. (2004) findings were cross-sectionally derived from a population-based sample and reflect the short-term effects of air pollution on HRV. When the regression coefficients for each individual pollutant model were compared, the effects for PM_{10} were considerably larger than the effects for gaseous pollutants such as O_3 . Because of the population-based sample, this study does have better generalizability than other smaller panel studies. The findings are suggestive of short-term effects of air pollutants, including O_3 , on HRV at the population level.

Another population-based study of air pollutants and HRV was conducted in Boston, MA on 497 men from the VA Normative Aging Study (NAS) (Park et al., 2005). The mean 24-h avg O₃ concentration was 23.0 ppb (SD 13.0). Several associations with HRV outcomes were observed with a 4-h moving average of O₃ concentrations; effects generally diminished with longer averaging times of 24 and 48 hours. Stronger associations were reported with PM_{2.5}. In two-pollutant models, the magnitude of the percent changes for both PM_{2.5} and O₃ diminished slightly. In analyses by ischemic heart disease, hypertension, and diabetes status, stronger associations of HRV with O₃ and PM_{2.5} were observed for individuals with ischemic heart disease and hypertension. These results are consistent with a Mexico City study (n = 34) by Holguín et al. (2003) which reported an HRV effect for O₃ in subjects with hypertension. The association of O₃ exposure with reduced low-frequency power in the full cohort seemed to be driven by subjects not taking calcium-channel blockers (Park et al., 2005). This suggests that this drug is blocking effects of O₃ on the sympathetic pathway. This study cohort consists of all males and almost all whites. This population-based study suggests that short-term exposure to O₃ is a predictor of alterations in cardiac autonomic function as measured by HRV among older male adults. A potential limitation of this study is that electrocardiograms were only taken once for each subject, so subject-specific variation of HRV measures may not be ruled out as a potential confounder.

Two separate analyses of the same cohort of patients examined the association between air pollution and the incidence of ventricular arrhythmias (Dockery et al., 2005; Rich et al., 2005). A total of 203 patients with implanted cardioverter defibrillators who lived within 25 miles of the ambient monitoring site at the Harvard School of Public Health, Boston, MA, were monitored. They had a total of 635 person-years of follow-up or an average of 3.1 years per subject. The median 48-h avg O₃ concentration was 22.9 ppb (IQR 15.4). In the analysis by Dockery et al. (2005), positive associations were observed between ventricular arrhythmias within 3 days of a prior event and a 2-day mean of several air pollutants, including PM_{2.5}, black carbon, NO₂, CO, and SO₂. No associations were observed with O₃. There was, however, a suggestion of increasing risk with increasing quintiles of O₃ (p <0.05). The analysis by Rich et al. (2005) observed stronger O₃ effects on ventricular arrhythmias using a case-crossover study design. Case periods were defined by the time each arrhythmic event began; for each case, three to four control periods were selected by matching on weekday and hour of the day within

the same calendar month. The median 24-h avg O₃ level was 22.6 ppb (IQR 15.7). For a 20 ppb increase in 24-h moving average O₃, a 27% (95% CI: 0, 60) increased risk of ventricular arrhythmias was estimated. Significant effects also were found for PM_{2.5}, NO₂, and SO₂. In two-pollutant models, the O₃ effect was found to be generally robust. Stratified analysis by the presence of a recent ventricular arrhythmia within the previous three days indicated that O₃ was associated with increased risk among subjects without a recent event (37% [95% CI: 6, 79]), but not among those with recent events (5% [95% CI: -27, 49]). Rich et al. (2005) explained that the use of the case-crossover study design and conditional analysis might have contributed to the stronger associations observed in their study compared to Dockery et al. (2005). The case-crossover design and conditional analysis controlled for season, time trends, weekday, as well as other non-time-varying confounders such as underlying medical conditions and smoking status by design, thereby eliminating any residual confounding. In the analysis by Dockery et al. (2005), these factors were controlled through modeling. In addition, the use of a 24-h moving average instead of a calendar-day air pollution concentration might have reduced exposure misclassification, resulting in larger effect estimates.

Other studies do not provide evidence for an O₃ effect on HRV and cardiac arrhythmias (Peters et al., 2000a; Rich et al., 2004; Vedal et al., 2004). These studies, however, may have had limited power to examine subtle effects. Gold et al. (2000; reanalysis Gold et al., 2003) reported results that suggest that O₃ exposure may decrease vagal tone, leading to reduced HRV. In this Boston, MA study, the mean 1-h max O₃ level was 25.7 ppb (IQR 23.0). In another Boston study, Schwartz et al. (2005) reported a weak association of O₃ with the root mean squared differences between adjacent R-R intervals in a study of 28 elderly subjects. The median 1-h max O₃ level was 34 ppb (IQR 26). The authors noted that lack of personal exposure measurements might render such studies less able to assess autonomic functions. This study by Schwartz et al. (2005) reported the strongest effects for black carbon.

7.2.7.2 Acute Myocardial Infarction

The effect of O₃ on the incidence of MI was examined in a limited number of studies. Acute MI was studied in relation to air pollution in Toulouse, France based on the existence of an acute MI registry (Monitoring Trends and Determinants in Cardiovascular Disease [MONICA]) and an air quality network covering the same population (Ruidavets et al., 2005).

The mean 8-h max O₃ level was 38.6 ppb (range 2.0-82.6). After adjustment for temperature, relative humidity, and influenza epidemics, the relative risk of acute MI occurrence was 1.76 (95% CI: 1.12, 2.45) for current-day O₃ concentrations. The increased risk of MI was more evident in the oldest group, 55 to 64 years of age. Further, the oldest subjects without a personal history of ischemic heart disease were more susceptible to an acute event when O₃ levels increased. No PM data was reported in this study.

In a case-crossover study (n = 772) in Boston, MA, Peters et al. (2001) reported an odds ratio of 1.27 (95% CI: 0.87, 1.88) per 40 ppb increase in 2-h avg O₃ (1 hour before onset of event). The mean 24-h avg O₃ level was 19.9 ppb (5th % to 95th %: 6 to 36). Stronger effects on the incidence of MI were observed for PM_{2.5} and PM₁₀. It should be noted that Peters et al. (2001) used unidirectional sampling (i.e., control periods were selected only before the case period) in their case-crossover design. Unidirectional sampling can lead to time trend bias and overlap bias (i.e., biased conditional logistic regression estimating equations), which can result in overestimated effects of exposure (Janes et al., 2005).

7.2.7.3 Cardiovascular Endpoints in Human Clinical Studies

In a controlled human exposure study discussed in Chapter 6, Sections 6.3.4 and 6.10, Gong et al. (1998a) studied 10 nonmedicated hypertensive and 6 healthy male adults exposed to 0.3 ppm O₃ with intermittent exercise in relation to various cardiovascular effects. The overall results did not indicate acute cardiovascular effects of O₃ in either the hypertensive or control subjects. The authors observed an increase in rate-pressure product and heart rate, a decrement for FEV₁, and a >10 mm Hg increase in the alveolar/arterial pressure difference for O₂ following O₃ exposure. These findings suggest that O₃ may exert cardiovascular effects indirectly by impairing alveolar-arterial O₂ transfer and potentially reducing O₂ supply to the myocardium. Ozone exposure may increase myocardial work and impair pulmonary gas exchange to a degree that could perhaps be clinically important in persons with significant pre-existing cardiovascular impairment.

7.2.7.4 Summary of Field Studies with Cardiovascular Outcomes

A limited epidemiologic database examining cardiovascular outcomes in relation to O₃ exposures is available. Among these studies, three were population-based and involved cohorts

such as the ARIC (Liao et al., 2004), MONICA (Ruidavets et al., 2005), and NAS (Park et al., 2005). Such studies may offer more informative results based on their large subject-pool and design. Results from these three studies were suggestive of an association between O₃ exposure and the cardiovascular endpoints studied. As in the case of respiratory disease outcomes, Brook et al. (2004) stated that the increase in relative risk for cardiovascular disease due to air pollution is small compared with the impact of the established cardiovascular risk factors. However, because of the enormous number of people affected, even conservative risk estimates can translate into a substantial increase in mortality due to cardiovascular disease within the population and, therefore, could potentially imply a notable public health problem.

7.2.8 Summary of Field Studies Assessing Acute Ozone Effects

- Results from recent field/panel studies support the evidence from controlled human exposure studies that acute O₃ exposure is associated with a significant effect on lung function, as indicated by decrements in FEV₁, FVC, and PEF. The declines in lung function were noted particularly in children and asthmatics.
- Limited evidence suggests that more time spent outdoors, higher levels of exertion, and the related increase in O₃ exposure may potentiate the risk of respiratory effects. In addition to children and asthmatics, adults who work or exercise outdoors may be particularly vulnerable to O₃-associated health effects.
- Many new studies have examined the association between O₃ concentrations and a wide variety of respiratory symptoms (e.g., cough, wheeze, production of phlegm, and shortness of breath). Collectively, the results suggest that acute exposure to O₃ is associated with increased respiratory symptoms and increased as-needed medication use in asthmatic children.
- Other panel studies evaluated O₃ effects on other health outcomes, including school absences and markers of inflammation and oxidative damage. Ozone exposure was associated with increased inflammation and generation of hydroxyl radicals in the upper airways. Use of antioxidant supplements was found to diminish the O₃ effect on lung function. Some studies suggest that O₃ exposure, accumulated over 2-4 weeks, is associated with increases in respiratory-related school absences, but further replication is needed before firm conclusions can be drawn about O₃ effects on school absences.
- Some field studies have examined the association between O₃ and cardiac physiologic outcomes. The current evidence is rather limited but suggestive of a potential effect on HRV, ventricular arrhythmias, and MI incidence.

7.3 ACUTE EFFECTS OF OZONE ON DAILY EMERGENCY DEPARTMENT VISITS AND HOSPITAL ADMISSIONS

7.3.1 Summary of Key Findings on Studies of Emergency Department Visits and Hospital Admissions from the 1996 Ozone AQCD

In the 1996 O₃ AQCD, aggregate population time-series studies of O₃-related health effects provided relevant evidence of acute responses, even below a 1-h max O₃ of 0.12 ppm. Emergency room visits and hospital admissions were examined as possible outcomes following exposure to O₃. In the case of emergency room visits, the evidence was limited (Bates et al., 1990; Cody et al., 1992; Weisel et al., 1995; White et al., 1994), but results generally indicated an O₃ effect on morbidity. The strongest and most consistent evidence of O₃ effects, at levels both above and below 1-h max O₃ levels of 0.12 ppm, was provided by the multiple studies that had been conducted on summertime daily hospital admissions for respiratory causes in various locales in eastern North America (Bates and Sizto, 1983, 1987, 1989; Burnett et al., 1994; Lipfert and Hammerstrom, 1992; Thurston et al., 1992, 1994). These studies consistently demonstrated that O₃ air pollution was associated with increased hospital admissions, accounting for roughly one to three excess respiratory hospital admissions per million persons with each 100 ppb increase in 1-h max O₃. This association had been shown to remain even after statistically controlling for the possible confounding effects of temperature and copollutants (e.g., H⁺, SO₄⁻², PM₁₀), as well as when considering only days with 1-h max O₃ concentrations below 0.12 ppm. Overall, the aggregate population time-series studies considered in the 1996 O₃ AQCD provided strong evidence that ambient exposures to O₃ can cause significant exacerbations of preexisting respiratory disease in the general public.

7.3.2 Review of Recent Studies of Emergency Department Visits for Respiratory Diseases

Emergency department visits represent an important acute outcome that may be affected by O₃ exposures. Morbidities that result in emergency department visits are closely related to, but are generally less severe than, those that result in unscheduled hospital admissions. In many cases, acute health problems are successfully treated in the emergency department; however, a subset of more severe cases that present initially to the emergency department may require hospital admission.

Several studies have been published in the past decade that examined temporal associations between O₃ exposures and emergency department visits for respiratory diseases (Table AX7-3 in Annex 7, Section AX7.1). Total respiratory causes for emergency room visits typically include asthma, pneumonia, bronchitis, emphysema, upper and lower respiratory infections such as influenza, and a few other minor categories. Asthma visits typically dominate the daily incidence counts. Chronic bronchitis and emphysema often are combined to define COPD, which is a prominent diagnosis among older adults with lung disease. Figure 7-8 presents percent changes in emergency department visits for asthma from single-pollutant models, with results expressed in standardized increments. The lags presented in the figure vary depending on reported results. Most studies reported effect estimates from a short lag period (0 to 2 days). Results from Weisel et al. (2002) are not included as comparable risks estimates for O₃ were not presented. Several of the studies conducted in the United States and Canada examined the effects of O₃ during the warm season only. In general, ambient O₃ concentrations were associated with emergency department visits for asthma in warm-season only analyses. The warm-season effect estimates tended to be positive and larger than results for cool-season or all-year analyses.

Among studies with adequate controls for seasonal patterns, many reported at least one positive association with O₃. These studies examined emergency department visits for total respiratory complaints (e.g., Delfino et al., 1997b, 1998b; Hernández-Gardûno et al., 1997; Ilabaca et al., 1999; Jones et al., 1995; Lin et al., 1999), asthma (e.g., Friedman et al., 2001; Jaffe et al., 2003; Stieb et al., 1996; Tenías et al., 1998; Tobías et al., 1999; Tolbert et al., 2000; Weisel et al., 2002), and COPD (Tenías et al., 2002).

One recent study examined emergency department visits for total and cause-specific respiratory diseases in Atlanta, GA over an 8-year period (Peel et al., 2005). A distributed lag of 0 to 2 days was specified a priori. The mean 8-h max O₃ concentration was 55.6 ppb (SD 23.8). Ozone concentrations were associated with emergency department visits for total respiratory diseases and upper respiratory infections in all ages. A marginally significant association was observed with asthma visits (2.6% [95% CI: -0.5, 5.9] excess risk per 30 ppb increase in 8-h max O₃), which became stronger when the analysis was restricted to the warm months (3.1% [95% CI: 0.2, 6.2] excess risk). In multipollutant models adjusting for PM₁₀, NO₂ and CO, O₃ was the only pollutant that remained significantly associated with upper respiratory infections.

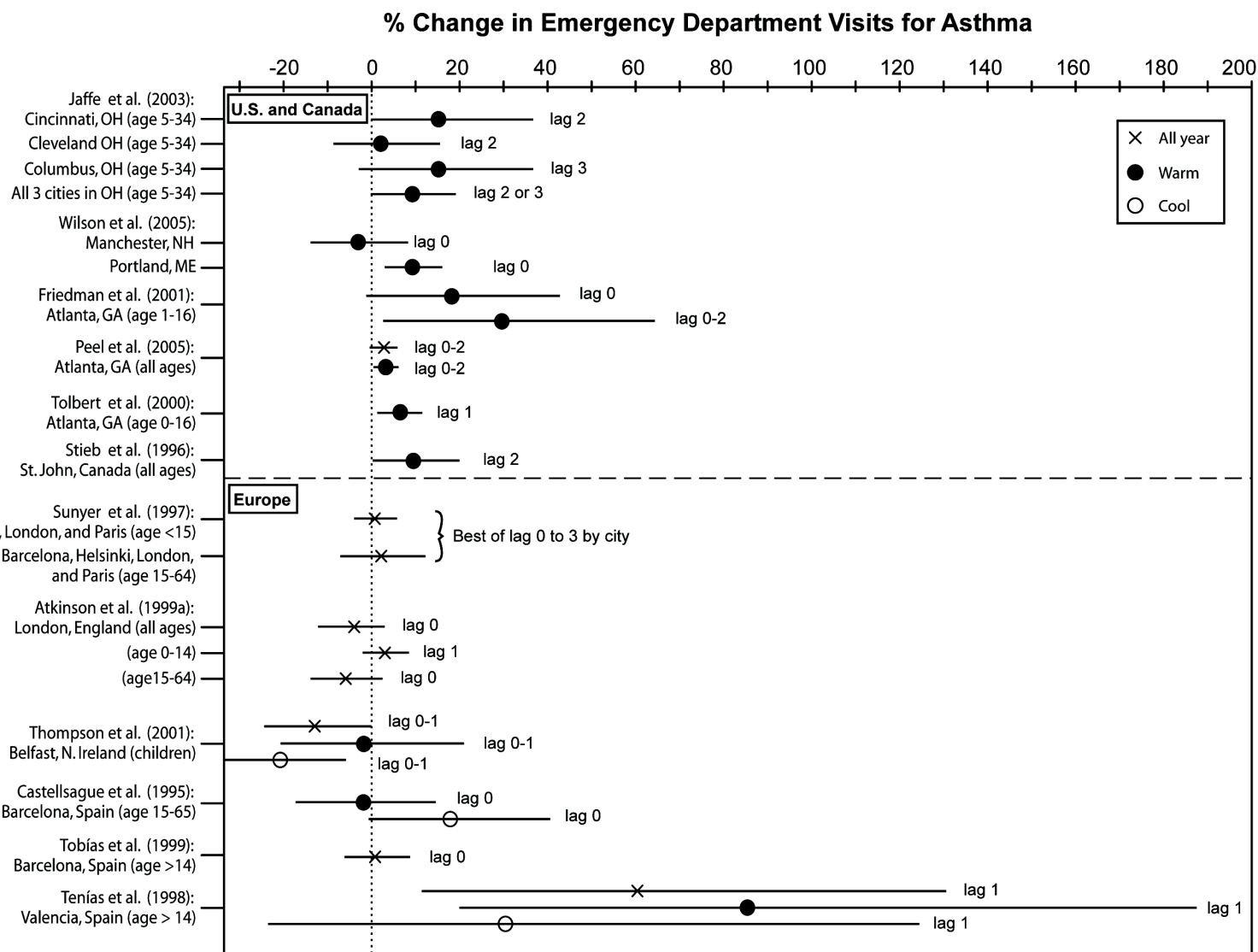


Figure 7-8. Ozone-associated percent change (95% CI) in emergency department visits for asthma per standardized increment (see Section 7.1.3.2).

Another large asthma emergency department study was carried out during the months of May through September from 1984 to 1992 in St. John, New Brunswick, Canada (Stieb et al., 1996). The mean 1-h max O₃ level was 41.6 ppb (range 0-160). Effects were examined separately among children aged less than 15 years and in persons aged 15 years and older. A significant effect of O₃ on emergency department visits was reported among persons 15 years and older. There was suggestion of a threshold somewhere in the range below a 1-h max O₃ of 75 ppb. Another study in Valencia, Spain from 1994 to 1995 found that emergency room visits for asthma among persons over 14 years old were robustly associated with relatively low O₃ levels (mean 1-h max O₃ of 32.4 ppb [range 6.9-81.2]) (Tenías et al., 1998). The excess risk of asthma emergency room visits was larger in the warm season (May to October; mean 1-h max O₃ of 38.2 ppb), with an 85% (95% CI: 20, 188) excess risk per 40 ppb increase in 1-h max O₃. During the cool season (November-April; mean 1-h max O₃ of 26.5 ppb), the excess risk was 31% (95% CI: -24, 125).

Among the studies that reported a positive association between O₃ and emergency department visits for respiratory outcomes, O₃ effects were found to be robust to adjustment for PM₁₀, NO₂, SO₂, and BS (Lin et al., 1999; Peel et al., 2005; Tenías et al., 1998). One study by Tolbert and colleagues (2000) observed that the significant univariate effects of both O₃ and PM₁₀ on pediatric asthma emergency department visits in Atlanta, GA became nonsignificant in two-pollutant regressions, reflecting the high correlation between the two pollutants ($r = 0.75$).

For several other studies with total respiratory and asthma outcomes, inconsistencies confound the interpretation of likely effects. For example, in a Montreal, Canada study by Delfino et al. (1997b), O₃ effects on total respiratory emergency department visits were seen in a short data series from the summer of 1993 (mean 8-h max O₃ of 30.7 ppb [SD 11.5]) but not in a similar data series from the summer of 1992 (mean 8-h max O₃ of 28.8 ppb [SD 11.3]). The significant 1993 results were seen only for persons older than 64 years. A very similar analysis of two additional summers (1989 [mean 8-h max O₃ of 44.1 ppb] and 1990 [mean 8-h max O₃ of 35.4 ppb]) revealed an O₃ association only for 1989, but again only in persons over 64 years old (Delfino et al., 1998b). An analysis of data on respiratory emergency department visits from June to August of 1990 (mean 24-h avg O₃ of 28.2 ppb [SD 11.7]) in Baton Rouge, LA reported O₃ effects in adults, but not in children or among the elderly (Jones et al., 1995).

Tobías and colleagues (1999) showed that regression results for asthma emergency department visits could be quite sensitive to methods used to control for asthma epidemics. Ozone was associated with the outcome variable in only one of eight models tested. An Atlanta, GA study by Zhu et al. (2003) examined asthma emergency department visits of children during three summers using Bayesian hierarchical modeling to address model variability. Data were analyzed at the ZIP code level to account for spatially misaligned longitudinal data. Results indicated a positive, but nonsignificant relationship between O₃ and emergency room visits for asthma. Ozone levels were not reported in this study.

Other studies also reported no association between O₃ and emergency department visits for respiratory causes (Atkinson et al., 1999a; Castellsague et al., 1995; Chew et al., 1999; Hwang and Chan, 2002; Sunyer et al., 1997). Using Bayesian hierarchical modeling, Hwang and Chan (2002) examined the effect of air pollutants on daily clinic visits for lower respiratory illnesses across 50 communities in Taiwan. The mean 1-h max O₃ for all 50 communities was 54.2 ppb, with individual-community means ranging between 38.9 to 78.3 ppb. All pollutants except O₃ were associated with daily clinic visits. In a pooled analysis of emergency admissions for asthma in four European cities as part of the Air Pollution on Health: European Approach (APHEA) study, no overall effect of O₃ was observed (Sunyer et al., 1997). Median 1-h max O₃ levels were relatively low, ranging from 14 to 37 ppb across the four cities. Atkinson et al. (1999a) also did not find an association in London, England between O₃ and emergency department visits at a mean 8-h max O₃ of 17.5 ppb (SD 11.5). One study by Thompson et al. (2001) in Belfast, Northern Ireland found no O₃ effect in the warm season, but a decreased risk of childhood asthma admissions (-21% [95% CI: -33, -6] per 20 ppb increase in 24-h avg O₃) in the cold season. The O₃ levels were similar in both seasons, with mean 24-h avg O₃ concentrations of 18.7 ppb in the warm season and 17.1 ppb in the cold season. After adjusting for benzene levels, O₃ was no longer associated with asthma emergency department visits during the cold season. The inverse relationship of O₃ with benzene concentrations ($r = -0.65$), and perhaps with other pollutants, might have contributed to the apparent protective effect of O₃. A study by Hajat et al. (1999, 2002) of physician consultations for asthma, lower respiratory diseases, and upper respiratory diseases in London reported negative associations with O₃, which was also suggestive of residual confounding by copollutants or weather factors (note that data were analyzed using Poisson GAM with default convergence criteria). Several other emergency

department studies looking at O₃ are more difficult to interpret because of inadequate control for seasonal patterns, very low O₃ levels, or because no quantitative results were shown for O₃ (Buchdahl et al., 1996, 2000; Garty et al., 1998; Holmén et al., 1997; Lierl and Hornung, 2003; Lipsett et al., 1997; Nutman et al., 1998).

Although several studies found a significant association between O₃ concentrations and emergency department visits for respiratory causes, some inconsistencies were observed. The inconsistencies may be attributable, at least partially, to differences in model specifications and analysis approach among the various studies. For example, ambient O₃ concentrations, length of the study period, and statistical methods used to control confounding by seasonal patterns and copollutants appear to affect the observed O₃ effect on emergency department visits. Studies that stratified analyses by season generally reported a positive association between O₃ concentrations and emergency department visits for asthma in the warm season.

7.3.3 Studies of Hospital Admissions for Respiratory Diseases

Hospital admissions represent a medical response to a serious degree of morbidity for a particular disease. Scheduled hospitalizations are planned in advance when a particular clinical treatment is needed. However, unscheduled admissions are ones that occur in response to unanticipated disease exacerbations and are more likely to be affected by environmental factors, such as air pollution. As such, the hospital admissions studies reviewed here focused specifically on unscheduled admissions. Study details and results from hospital admissions studies published during the past decade are summarized in Table AX7-4 (in Annex 7, Section AX7.1). As a group, these hospitalization studies tend to be larger, in terms of geographic and temporal coverage, and indicate results that are generally more consistent than those reviewed above for emergency department visits. As in the case for all studies that examine changes in aggregate measures of acute disease outcomes over time, the following should be considered in comparing results: (1) difference in types of respiratory diseases for hospital admission; (2) age of study population; (3) mean level of O₃ during study; (4) single-city versus multicity studies; (5) length of study (e.g., <5 years versus >5 years); (6) analysis by season versus all year; (7) O₃-only versus multipollutant models; (8) number of exposure lag days; and (9) type of study (e.g., case-crossover versus time-series). These factors are

considered in the sections below with further discussion on potential confounding of the O₃ effect estimate by seasonal factors and copollutants.

7.3.3.1 All-year and Seasonal Effects of Ozone on Respiratory Hospitalizations

The effect of O₃ on respiratory hospitalizations was examined in various studies conducted in the United States and abroad. Figures 7-9 and 7-10 present risk estimates from all total respiratory hospital admission studies. Burnett et al. (1995), which did not present quantitative results for O₃, and Yang et al. (2003), which only presented odds ratios, were not included in the figures. In cases where multiple lags were presented, the multiday lag was selected to represent the cumulative effect from all days examined. If only single-day lags were analyzed, the effect estimate of the shortest lag time, usually a lag of 0 or 1 day, was presented. Figure 7-9 plots the effect estimates and 95% CIs from 15 studies that analyzed all-year data. The risk estimates are arranged by age groups. The preponderance of positive risk estimates, with some that are statistically significant, is readily apparent. Figure 7-10 presents the season-stratified effect estimates by region. For studies that reported risk estimates from all four seasons, only the summer and winter estimates are presented. It appears that the warm-season estimates, collectively, tend to be larger, positive values compared to all-year and cool-season estimates. All of the negative estimates were from analyses using cool-season data only, which might reflect the inverse correlation between O₃ and one or another copollutant, especially PM, during that season. None of the studies presented results for the O₃ effect during the cool season after adjusting for potential confounding by PM.

Among the respiratory hospitalization studies, the most robust and informative results were observed when a broad geographic area was examined using a consistent analytical methodology (Anderson et al., 1997; Burnett et al., 1995, 1997a). These studies have all reported an O₃ effect on respiratory hospital admissions. The largest such study to date was carried out using data on all-age respiratory hospital admissions from 16 Canadian cities with populations exceeding 100,000 during the period 1981 to 1991 (Burnett et al., 1997a). In addition to O₃, the authors evaluated health effects of SO₂, NO₂, CO, and coefficient of haze (a surrogate for black carbon particle concentrations). Pooling the 16 cities, a positive association was observed between respiratory hospital admissions and the 1-day lag O₃ concentration in the spring (5.6% [95% CI: 1.6, 9.9] excess risk per 40 ppb increase in 1-h max O₃) and summer (6.7% [95% CI: 3.5, 10.0]).

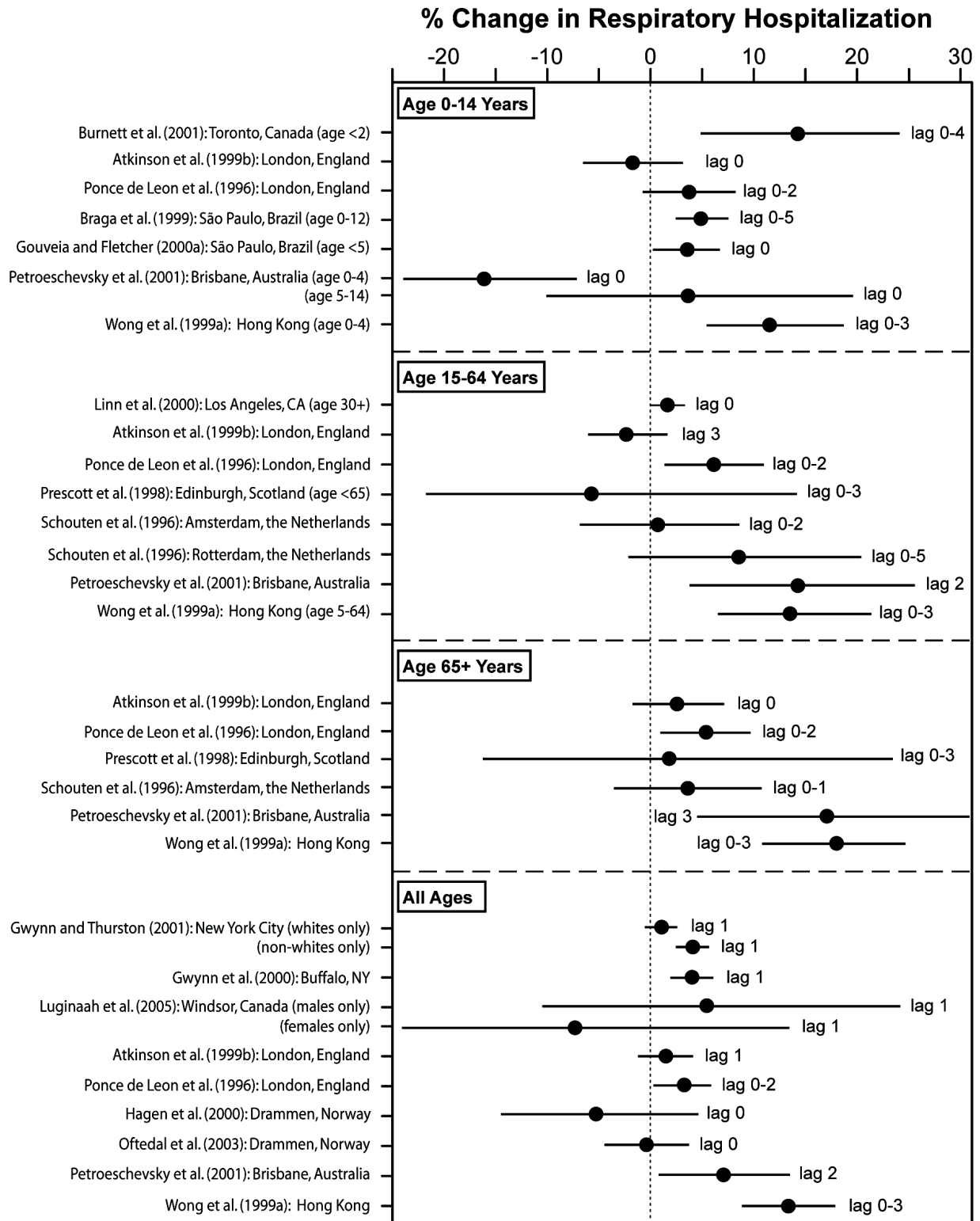


Figure 7-9. Ozone-associated percent change (95% CI) in total respiratory hospitalizations for all-year analyses per standardized increment (see Section 7.1.3.2). Effect estimates are arranged by age groups.

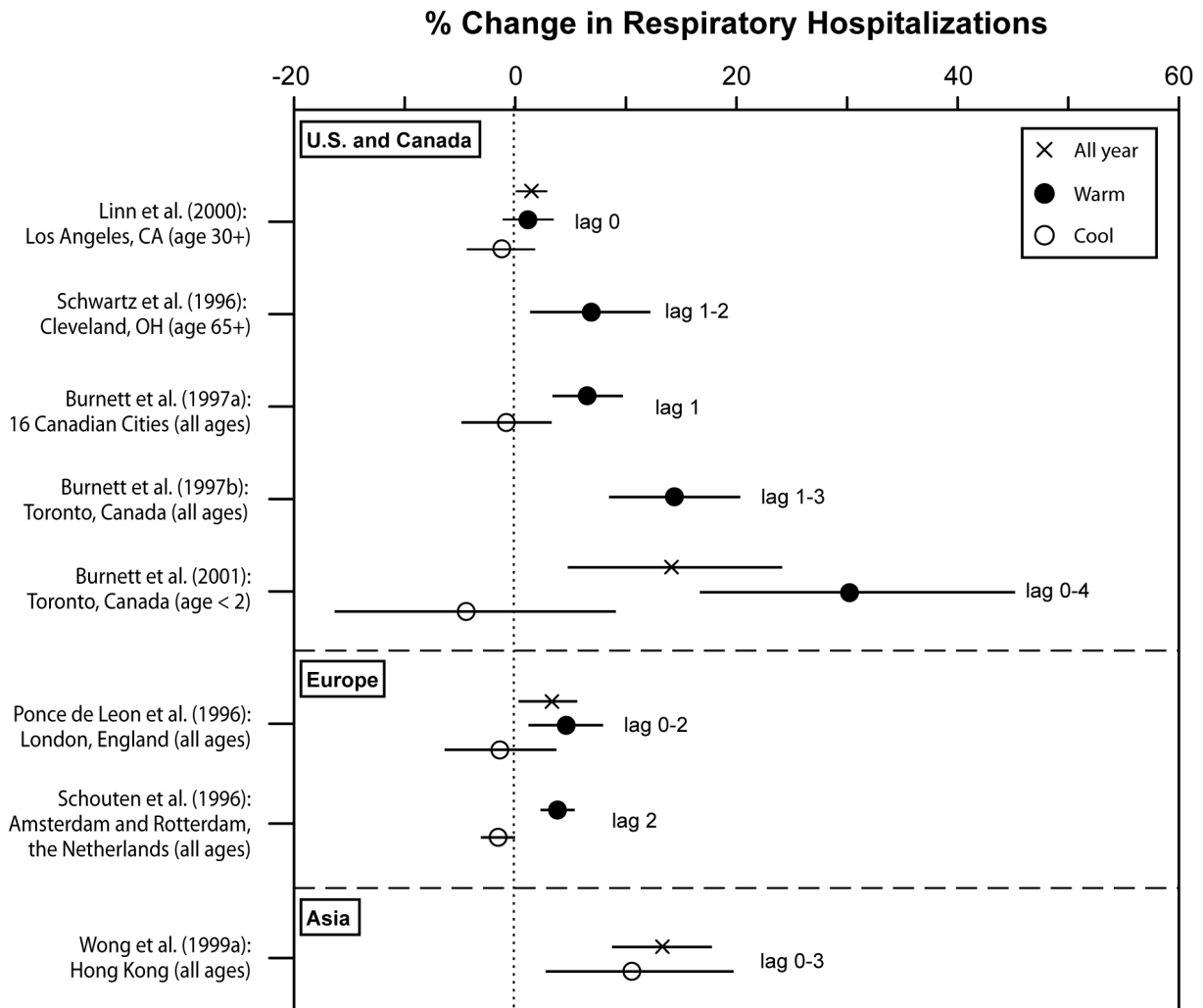


Figure 7-10. Ozone-associated percent change (95% CI) in total respiratory hospitalizations by season per standardized increment (see Section 7.1.3.2).

The results for fall were also positive, though of smaller magnitude (3.8% [95% CI: -0.2, 7.9]). There was no evidence for an O₃ effect in the winter season (-0.8% [95% CI: -4.8, 3.3]). The mean 1-h max O₃ concentrations for the spring, summer, fall, and winter were 40 ppb, 38 ppb, 21 ppb, and 26 ppb, respectively. Control outcomes related to blood, nervous system, digestive system, and genitourinary system disorders were not associated with O₃.

In a previous study focused mainly on evaluating health impacts of sulfate particles, Burnett et al. (1995) reported results from a time-series analysis of all-age respiratory hospital admissions to 168 hospitals in Ontario, Canada over a 6-year period (1983 to 1988). The mean 1-h max O₃ was 36.3 ppb for the entire year, ranging from 21.9 ppb in December to 52.9 ppb in July. The outcome data were prefiltered to remove seasonal variations using a weighted 19-day moving average. The authors reported that O₃ was associated with respiratory hospital admissions; however, no quantitative results for O₃ were presented.

Results from an analysis of five European cities indicated strong and consistent O₃ effects on unscheduled hospital admissions for COPD (Anderson et al., 1997). The five cities examined (London, Paris, Amsterdam, Rotterdam, and Barcelona) were among those included in the multicity APHEA study. The number of years of available data varied from 5 to 13 years among the cities. City-specific effect estimates were pooled across cities using weighted means. An association with O₃ was observed in full-year analyses. All-year median 1-h max O₃ levels ranged from 19 to 40 ppb across the five cities. Season-stratified analyses indicated that the O₃ effect was larger in the warm season (median 1-h max O₃ range 25-47 ppb), 4.7% (95% CI: 1.6, 7.9) excess risk per 40 ppb increase in 1-h max O₃, compared to the cool season (median 1-h max O₃ range 10-33 ppb), 1.6% (95% CI: -3.1, 7.9) excess risk. There was no significant heterogeneity in O₃ effects among the cities.

Several additional studies carried out in one or two cities over a span of five or more years provided substantial additional evidence regarding O₃ effects on respiratory hospital admissions (Anderson et al., 1998; Burnett et al., 1999, 2001; Moolgavkar et al., 1997; Petroeschevsky et al., 2001; Ponce de Leon et al., 1996; Sheppard et al., 1999 [reanalysis Sheppard, 2003]; Yang et al., 2003). Moolgavkar and colleagues (1997) reported significant and robust O₃ effects on respiratory hospital admissions in adults 65 years and older in Minneapolis and St. Paul, MN (mean 24-h avg O₃ of 26.2 ppb), but not in Birmingham, AL (mean 24-h avg O₃ of 25.1 ppb). The absence of effects in the southern city may reflect less penetration of O₃ into the indoor environment due to greater use of air conditioning, and thus less correlation between central site O₃ monitoring and actual exposures of the urban populace. Ozone effects on all-age and age-stratified asthma and total respiratory hospital admissions were observed in Brisbane, Australia (Petroeschevsky et al., 2001). Effect sizes were found to be consistent in the warm and cool seasons (data not provided). Petroeschevsky et al. commented that the year-round effect

of O₃ might reflect the relatively small degree of seasonal variation in O₃ levels observed in Brisbane. Mean 8-h avg O₃ (10 a.m.-6 p.m.) levels for the winter, spring, summer, and fall were 16.1 ppb, 23.3 ppb, 19.9 ppb, and 16.7 ppb, respectively. The authors also noted that given the subtropical climate in Brisbane, characterized by warm, dry winters, perhaps the proportion of the population exposed to winter O₃ concentrations was higher than in cities where inclement winter weather might force populations indoors.

Another set of studies examined associations between O₃ and respiratory-related hospitalizations in single cities over shorter (<5 year) time spans. Positive and significant O₃ effects were reported in Cleveland, OH (Schwartz et al., 1996); New York City (Gwynn and Thurston, 2001); Northern New Jersey (Weisel et al., 2002); Toronto, Canada (Burnett et al., 1997b); Helsinki, Finland (Pönkä and Virtanen, 1996); São Paulo, Brazil (Braga et al., 1999; Gouveia and Fletcher, 2000a); and Hong Kong (Wong et al., 1999a). The Helsinki study by Pönkä and Virtanen (1996) reported significant effects of O₃ on both asthma and on digestive disorders in a setting of very low O₃ concentrations (mean 8-h max O₃ of 11 ppb), which raises questions of plausibility.

Less consistent effects of O₃ were seen in other respiratory hospitalization studies (Schouten et al., 1996; Lin et al., 2003, 2004; Linn et al., 2000; Morgan et al., 1998a; Oftedal et al., 2003). In a study conducted in Amsterdam and Rotterdam, the Netherlands, associations between O₃ and respiratory admissions were observed; however, results were difficult to interpret due to the large number of statistical tests performed (Schouten et al., 1996). In a California study by Neidell (2004), a negative association was observed between hospitalizations for asthma and naturally occurring seasonal variations in O₃ within ZIP codes for children aged 0 to 18 years. However, the O₃ effect was found to be influenced by socioeconomic status. Among children of low socioeconomic status, O₃ generally was associated with increased hospitalizations, with statistical significance reached in certain age groups. Neidell further stated that avoidance behavior on high O₃ days (i.e., 1-h max O₃ >200 ppb) may have attributed to the negative relationship observed in children of higher socioeconomic status.

No associations between respiratory hospital admissions and O₃ were seen in studies from Los Angeles, CA (Nauenberg and Basu, 1999); Detroit, MI (Lippman et al., 2000 [reanalysis Ito, 2004]); Vancouver, Canada (Lin et al., 2004); London, England (Atkinson et al., 1999b); Edinburgh, Scotland (Prescott et al., 1998); and Drammen, Norway (Hagen et al., 2000).

Several of these studies were carried out in locations with low O₃ levels, suggestive of a nonlinear concentration-response relationship. For example, the mean 1-h max O₃ was 28.02 ppb in the Vancouver, Canada study by Lin et al. (2004). The Edinburgh, Scotland study by Prescott et al. (1998) had a mean 24-h avg O₃ of 14.5 ppb. Inadequate control of seasonal confounding may underlie some of the nonsignificant and negative findings. An additional factor likely contributing to lack of associations observed is the relatively small sample sizes included in some of these studies.

For respiratory hospitalization outcomes, the largest, most significant associations with O₃ concentrations were observed when using short lag periods, in particular a 0-day lag (exposure on same day) and a 1-day lag (exposure on previous day). In the study of 16 Canadian cities by Burnett et al. (1997a), the strongest association between O₃ and respiratory hospitalizations was found at a 1-day lag. A decline in the magnitude and significance of the effect was seen with increasing days lagged for O₃. Anderson et al. (1997) investigated the association between O₃ and daily hospital admissions for COPD in five European cities. Lags up to 5 days were examined, and the largest risk estimates were found using 0- and 1-day lags. These results suggest that O₃ has a short-term effect on respiratory hospitalizations.

Burnett et al. (2001) investigated the association between respiratory hospitalizations and O₃ in children less than 2 years of age (note that analyses were performed using default convergence criteria for Poisson GAM with a nonparametric LOESS prefilter applied to pollution and hospitalization data). Lags up to 5 days were examined after stratifying by season (Figure 7-11). The mean 1-h max O₃ during the summer season was 45.2 ppb (IQR 25). In the summer, significant associations between O₃ and daily admissions were found in several of the single-day lags, with the largest risk estimate of 12.5% (95% CI: 5.7, 19.7) excess risk per 40 ppb increase in 1-h max O₃ at a 1-day lag. The positive effect estimates observed at multiple single-day lags indicated that O₃ exposure likely had an immediate effect that persisted over several days. Using a cumulative lag period of 0- to 4-days, the O₃-related risk estimate was 30.2% (95% CI: 18.0, 42.4).

Weisel et al. (2002) stated that a lag period of 1 to 3 days between exposure to O₃ and hospital admissions or emergency department visits for asthma was plausible, because it might take time for the disease to progress to the most serious responses following exposure. Also, taking medication could further delay the progression of the adverse effect. Thus, although

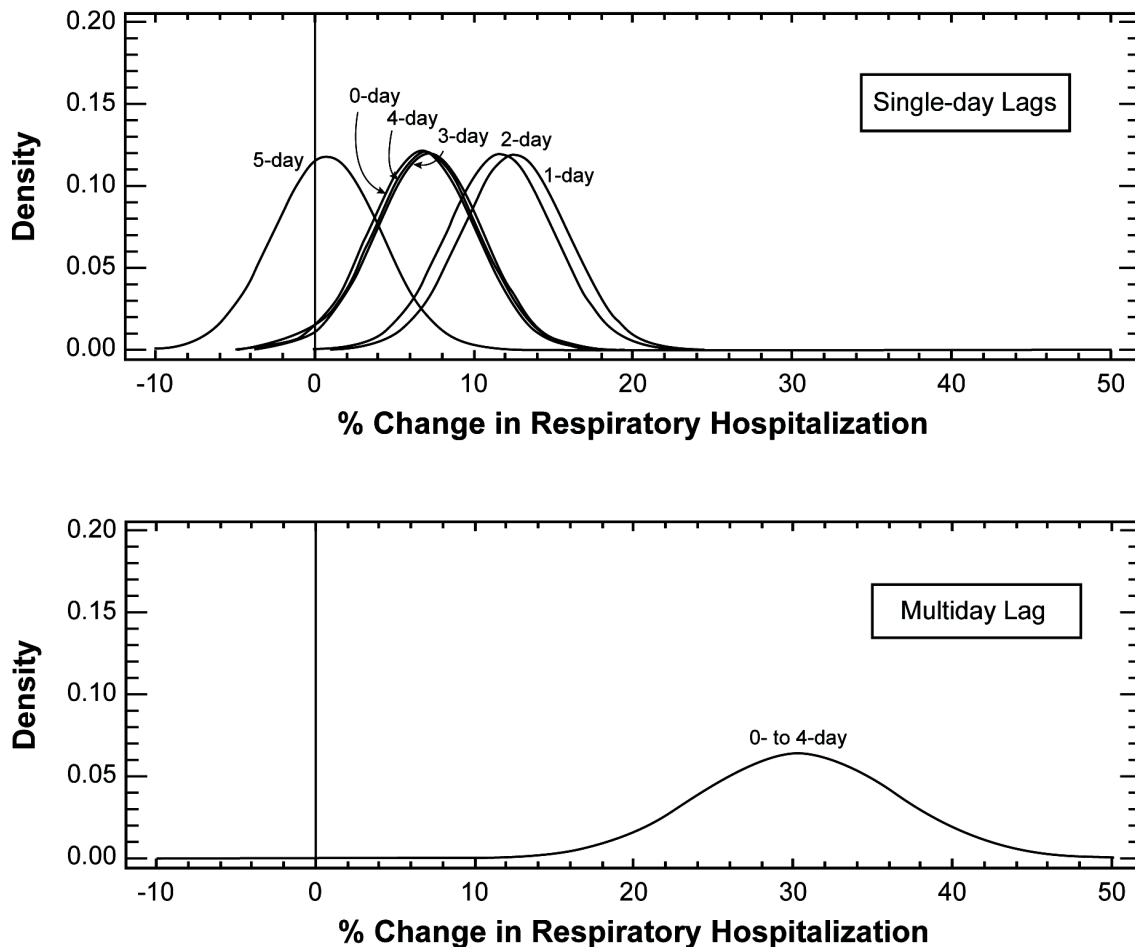


Figure 7-11. Percent changes in total respiratory hospitalizations per 40 ppb increase in 1-h max O₃ in children less than two years of age during the summer (May to August). Single-day lags (0-, 1-, 2-, 3-, 4-, and 5-day) are shown in the upper panel. The cumulative multiday lag (0- to 4-day) is shown in the lower panel.

Source: Derived from Burnett et al. (2001).

strongest associations are found at lags of 0 and 1 days, examining longer single-day lag periods or multiday lag periods may enhance the understanding of O₃ effects on hospitalizations.

In conclusion, the evidence generally supports the findings of significant and robust effects of O₃ on various respiratory disease hospitalization outcomes. Large multicity studies, as well as many studies from individual cities have reported positive O₃ associations with total respiratory hospitalizations, asthma, and COPD, especially in studies analyzing the O₃ effect during the summer or warm season.

7.3.3.2 Potential Confounding of the Ozone Effect on Respiratory Hospitalizations by Copollutants

As in the case for most air pollution studies, potential confounding of the association between O₃ and respiratory hospitalizations by copollutants generally was examined using multipollutant regression models. The changing relationship between O₃ and copollutants by season complicates assessment of potential confounding of the O₃ effect by copollutants. Figure 7-12 compares the risk estimates from models with and without adjustment for PM indices. This figure indicates that O₃ risk estimates are fairly robust to PM adjustment in all-year and warm-season only data. Given the inverse relationship typically observed between O₃ and PM during the cool season, the influence of PM adjustment on O₃ risk estimates during that season is of particular interest. However, none of the hospitalization studies examined O₃ risk estimates after adjusting for PM in cool-season only data.

Several analyses of a large data set from Toronto, Canada spanning the years 1980 to 1994 reported O₃ effects on respiratory hospitalizations for all ages (Burnett et al., 1997b, 1999) and for persons less than 2 years old (Burnett et al., 2001). In the 1999 and 2001 studies, analyses were performed using Poisson GAM (default convergence criteria) with a nonparametric LOESS prefilter applied to the pollution and hospitalization data. All studies found that O₃ effects were robust when adjusting for PM indices, whereas PM effects from single-pollutant models were markedly attenuated when O₃ was added to the regression. These results imply more robust associations with respiratory hospitalizations for O₃ than PM.

Results from the APHEA study indicated strong and consistent O₃ effects on unscheduled hospital admissions for COPD (Anderson et al., 1997). Significant effects also were seen for BS, TSP, and NO₂. The authors reported that among all pollutants examined, the most consistent and significant findings were for O₃. No two-pollutant model results were reported. Several additional studies also observed that there was no substantial difference in the O₃ effect after adjusting for PM in the regression model (Gouveia and Fletcher, 2000a; Petroschevsky et al., 2001; Ponce de Leon et al., 1996).

Collectively, these results suggest that copollutants generally do not confound the association between O₃ and respiratory hospitalizations. Ozone risk estimates were robust to PM adjustment in all-year and warm-season only data.

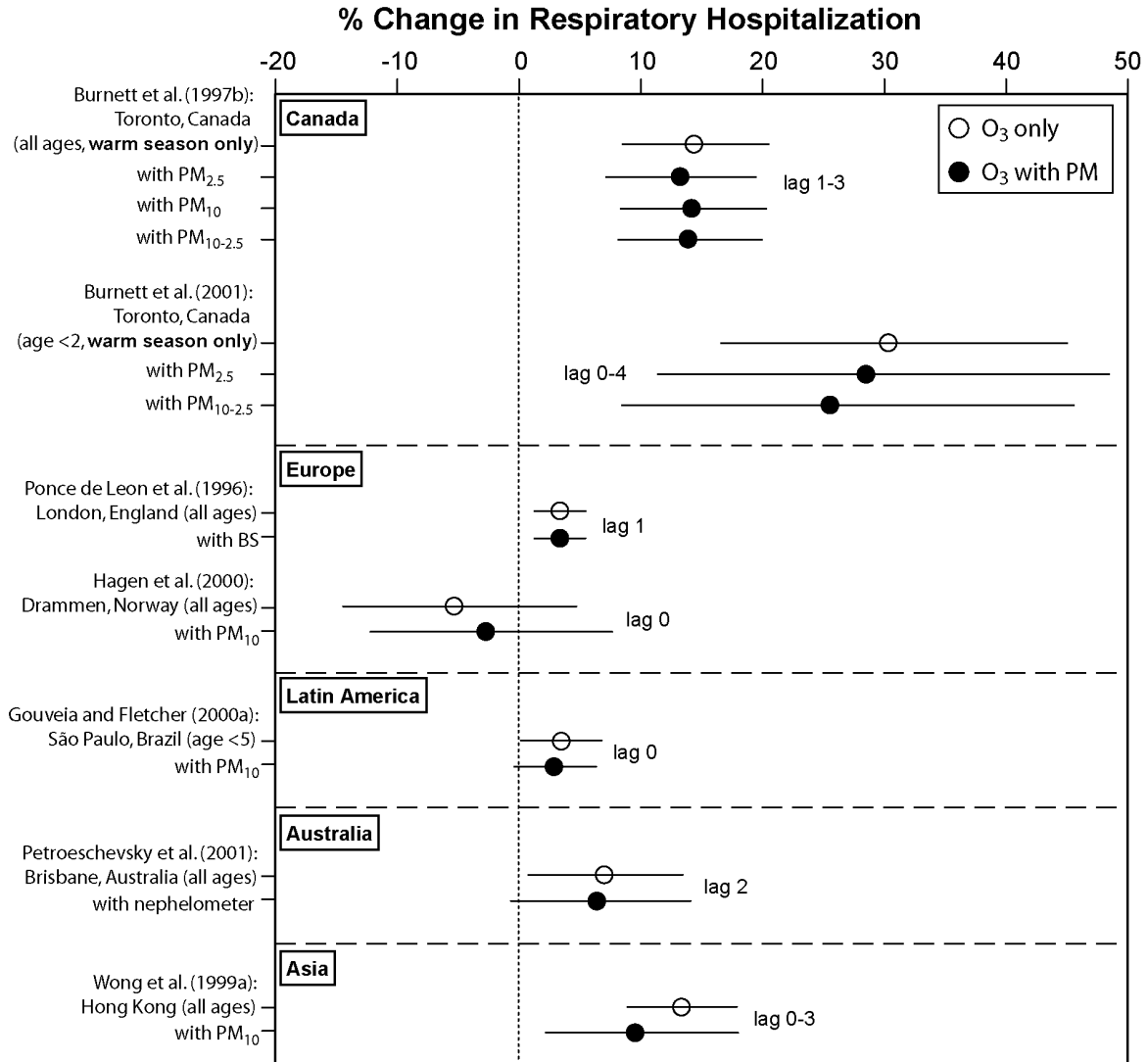


Figure 7-12. Ozone-associated percent change (95% CI) in total respiratory hospitalizations with adjustment for PM indices per standardized increment (see Section 7.1.3.2). Analyses performed using all-year data unless noted otherwise.

7.3.4 Association of Ozone with Hospital Admissions for Cardiovascular Disease

Some hospital admissions studies have examined the association of O₃ with cardiovascular outcomes (see Figure 7-13). Many reported negative or inconsistent associations (Ballester et al., 2001; Burnett et al., 1999; Fung et al., 2005; Koken et al., 2003; Linn et al., 2000;

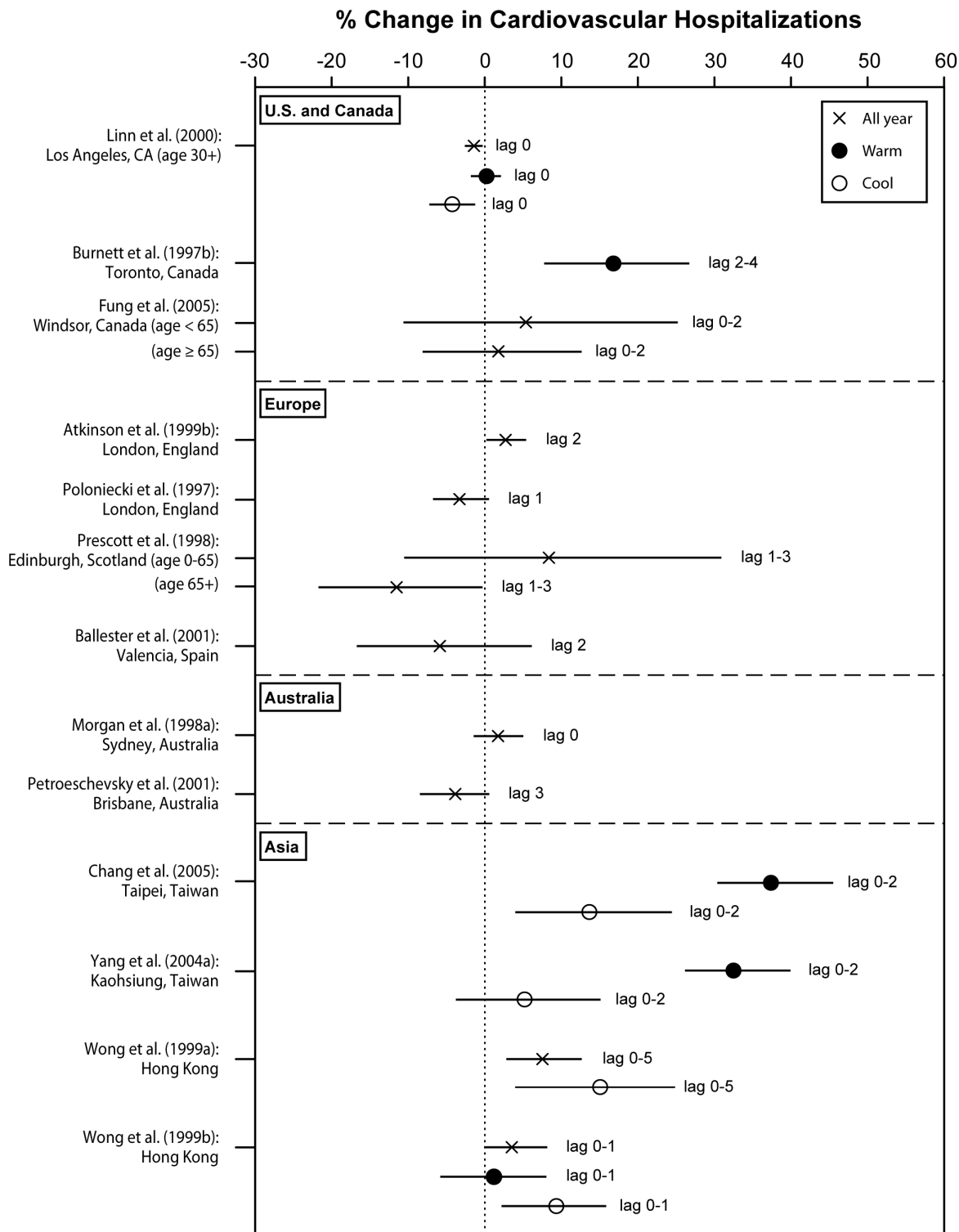


Figure 7-13. Ozone-associated percent change (95% CI) in total cardiovascular hospitalizations per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.

Lippmann et al., 2000 [reanalysis Ito, 2004]; Mann et al., 2002; Morgan et al., 1998a; Petroschevsky et al., 2001; Poloniecki et al., 1997; Prescott et al., 1998). In the Ito (2004) reanalysis of the Lippmann et al. (2000) study in Detroit, MI, no associations were observed between cause-specific cardiovascular admissions (i.e., ischemic heart disease, dysrhythmias, heart failure, and stroke) and O₃ concentrations (mean 24-h avg O₃ of 25 ppb) using all-year data. The study by Linn et al. (2000) examined associations between O₃ concentrations and cardiovascular admissions during all four seasons as well as using all-year data in Los Angeles, CA. Mean 24-h avg O₃ levels for the spring, summer, fall, and winter were 32 ppb, 36 ppb, 13 ppb, and 14 ppb, respectively. Positive, but nonsignificant effects were observed during the spring and summer (0.6% [95% CI: -1.4, 2.6] and 0.2% [95% CI: -1.7, 2.2] respectively, per 20 ppb increase in 24-h avg O₃), while negative associations were observed during the fall and winter (-0.6% [95% CI: -3.3, 2.2] and -4.1% [95% CI: -7.1, -1.1] respectively).

Some other studies, especially those that examined the relationship when O₃ exposures were higher, have found robust positive associations between O₃ and cardiovascular hospital admissions (Atkinson et al., 1999b; Burnett et al., 1997b; Chang et al., 2005; Tsai et al., 2003a; Wong et al., 1999a,b; Yang et al., 2004a). For example, Burnett et al. (1997b) reported a positive association between O₃ and cardiovascular hospital admissions in Toronto, Canada in a summer-only analysis (mean 1-h max O₃ of 41.2 ppb). The results were robust to adjustment for various PM indices, whereas the PM effects diminished when adjusting for gaseous pollutants. Other studies stratified their analysis by temperature, i.e., by warm days (≥ 20 °C) versus cool days (< 20 °C). Several analyses using warm days consistently produced positive associations (Chang et al., 2005; Tsai et al., 2003a; Yang et al., 2004a). On the other hand, in two studies conducted in Hong Kong, total cardiovascular (as well as circulatory, ischemic heart disease, and heart failure) were all reported to be significantly associated with O₃ in the cool but not the warm season (Wong et al., 1999a,b). In Wong et al. (1999b), O₃ concentrations were similar in both seasons, with warm-season O₃ levels being slightly lower (mean 8-h avg O₃ = 16.1 ppb) than the cool season levels (mean O₃ = 18.0 ppb). The authors speculated that differing activity patterns and home ventilation factors may have contributed to the seasonal differences in O₃ effects. Weather in Hong Kong is mild throughout the year, but less humid and cloudy in the cool season. Thus, during the cool season people are more likely to open windows

or stay outdoors, resulting in higher personal exposures even with similar ambient O₃ concentrations across seasons.

An increasing number of hospitalization studies have examined the association between O₃ and cardiovascular admissions. Some studies, especially those that performed stratified analyses by seasonal or meteorological factors, have observed positive associations. However, the overall evidence remains inconclusive regarding the effect of O₃ on cardiovascular hospitalizations.

7.3.5 Summary of Acute Ozone Effects on Daily Emergency Department Visits and Hospital Admissions

- The vast majority of emergency room visits and hospitalization studies conducted over the past decade have looked at effects of O₃ on either total respiratory diseases and/or asthma. Many of these studies analyzed O₃ risk estimates using year-round data. Given the strong seasonal variations in O₃ concentrations and the changing relationship between O₃ and other copollutants by season, inadequate adjustment for seasonal effects might have masked or underestimated the association between O₃ and the respiratory disease outcomes. Ozone was generally found to be associated with respiratory hospitalizations and asthma emergency department visits during the warm season but not during the cool season.
- Several studies have examined the association between O₃ and respiratory hospitalizations while controlling for other pollutants in the analytical model. In most cases, O₃ effects have been reported to be robust to adjustment for copollutants, particularly PM. Therefore, the evidence is supportive of independent O₃ effects on respiratory hospital admissions.
- Some hospital admission studies examined the effect of O₃ on cardiovascular outcomes. A few studies observed positive O₃ associations, largely in the warm season. Overall, however, the currently available evidence is inconclusive regarding any association between ambient O₃ exposure and cardiovascular hospitalizations.

7.4 ACUTE EFFECTS OF OZONE ON MORTALITY

7.4.1 Summary of Key Findings on Acute Effects of Ozone on Mortality from the 1996 Ozone AQCD

A limited number of studies, most of which were from the 1950s and 1960s, had examined O₃-mortality associations at the time of the 1996 O₃ AQCD. The 1996 O₃ AQCD considered these historical studies to be flawed because of inadequate adjustment for seasonal trends or temperature and the use of questionable exposure indices. There were only a few other

time-series studies that had examined O₃-mortality associations between the 1980s and mid-1990s. These latter studies used more sophisticated approaches in addressing seasonal confounding and weather models. One of them (Shumway et al., 1988) focused on possible associations with long-term O₃ fluctuations in Los Angeles, CA but did not examine short-term associations. A study that reanalyzed Los Angeles, CA data with a focus on short-term associations (Kinney and Özkaynak, 1991) did find that, of the PM and gaseous criteria pollutants, O₃ (reported as total oxidants) was most strongly associated with total nonaccidental mortality. Then two more studies, one conducted in Detroit, MI (O₃ concentrations not provided) (Schwartz, 1991) and the other in St. Louis, MO (mean 24-h avg O₃ of 22.5 ppb) and Kingston-Harriman, TN (mean 24-h avg O₃ of 23.0 ppb) (Dockery et al., 1992), reported that PM but not O₃ was significantly associated with mortality. However, the 1996 O₃ AQCD noted that, without sufficient presentation of model specifications, it was difficult to evaluate whether the lack of O₃-mortality associations was possibly due to mis-specification of the weather model. In summary, because of the insufficient number of studies that examined O₃-mortality associations and the uncertainties regarding weather model specifications, the 1996 O₃ AQCD was unable to quantitatively assess O₃-mortality excess risk estimates or, even, to provide qualitative assessment of the likelihood of O₃-mortality associations.

7.4.2 Introduction to Assessment of Current Ozone-Mortality Studies

Introductory discussions of PM-mortality effects often cite historical air pollution incidents such as the 1952 London, England smog episode, in which thousands of deaths were attributed to the air pollution from coal burning. There is no counterpart “historical episode” for O₃-mortality effects. Instead, the early recognition of the adverse health effects of summer oxidant air pollution, mainly from Los Angeles and other major cities with a high density of automobiles, were based on symptoms such as eye and throat irritations. Thus, the focus of PM epidemiology and that of O₃ epidemiology have been historically different.

As shown in Table AX7-5 in Annex 7, Section AX7.1, the number of short-term mortality studies that analyzed O₃ has increased markedly since the last publication of the 1996 O₃ AQCD. The increased attention to PM-mortality associations in the early 1990s led to an increase in studies that also examined O₃, most often as a potential confounder for PM. Although many of these PM studies also reported O₃ estimates, they often lacked specific hypotheses regarding

mortality effects of O₃, given that the main focus of these studies was to examine PM-mortality associations. This is in contrast to the O₃-morbidity studies, most of which were specifically designed to examine effects of “summer haze” and O₃ (or oxidants) on respiratory and other symptoms, lung function, emergency department visits, etc. However, new studies with hypotheses developed specifically for O₃ effects on mortality have become available, such as the large U.S. 95 communities study by Bell et al. (2004), the U.S. 14 cities study by Schwartz (2005), and the 23 European cities study by Gryparis et al. (2004) discussed in the next section.

7.4.3 Single-Pollutant Model Ozone-Mortality Risk Estimates

To facilitate a quantitative overview of the O₃-mortality effect estimates and their corresponding uncertainties, the percent excess risks of total nonaccidental mortality calculated using all-year data are plotted in Figures 7-14 and 7-15. Studies that only conducted seasonal analyses are presented in the next section. These figures do not include studies that only examined cause-specific mortality. Figure 7-14 presents only those results derived from single-day lag models. Results from multiday lag models are shown in Figure 7-15. All effect estimates are from single-pollutant models and include all age groups unless otherwise noted. The majority of the estimates are positive, with a few exceptions. Of particular note, five multicity studies, three from the United States (Bell et al., 2004; Samet et al., 2000 [reanalysis Dominici et al., 2003]; Schwartz, 2005) and two from Europe (Gryparis et al., 2004; Touloumi et al., 1997), showed generally positive associations.

The initial primary objective of the original NMMAPS (Samet et al., 2000; reanalysis Dominici et al., 2003) was to investigate the effects of PM, but the study also examined mortality risk estimates from gaseous pollutants in 90 U.S. cities over the period of 1987 to 1994. Among the 90 cities, 80 monitored O₃ either year-round or during the warm season. The study illustrated that the mortality risk estimates for O₃ varied by season. The estimate using all available data was about half of that for summer-only data at a lag of 1-day (see Section 7.6.3.2 for further discussion). Bell et al. (2004) extended the original NMMAPS by adding six more years (from 1987 to 2000) and 15 more communities (a total of 95 communities), and examined the effects of O₃ on mortality. Due to its extensive coverage and its specific focus on O₃-mortality effects, Bell et al. (2004) offer a more comprehensive analysis than the original Samet et al. (2000) analysis.

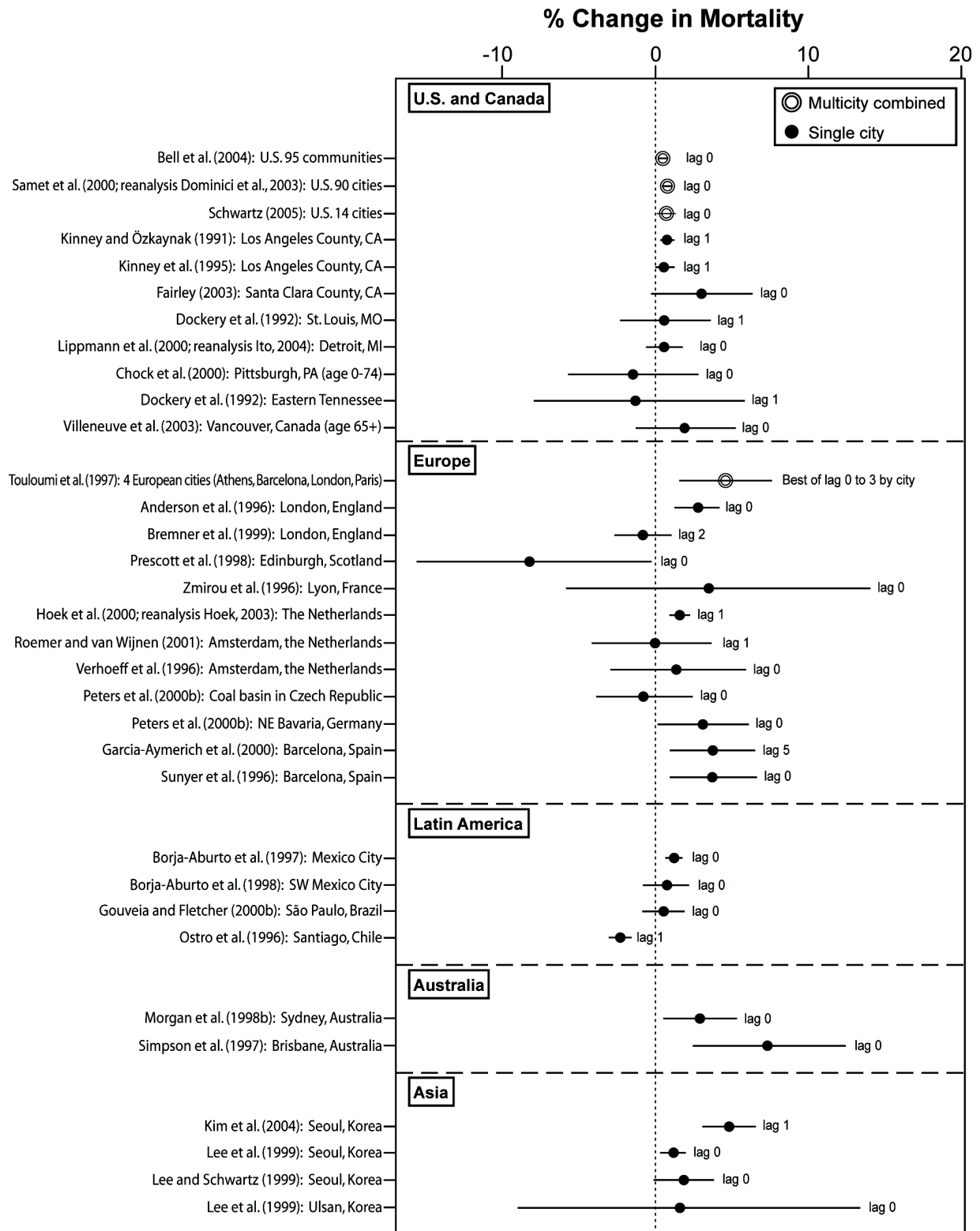


Figure 7-14. All cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. Only results from single-day lag models are presented.

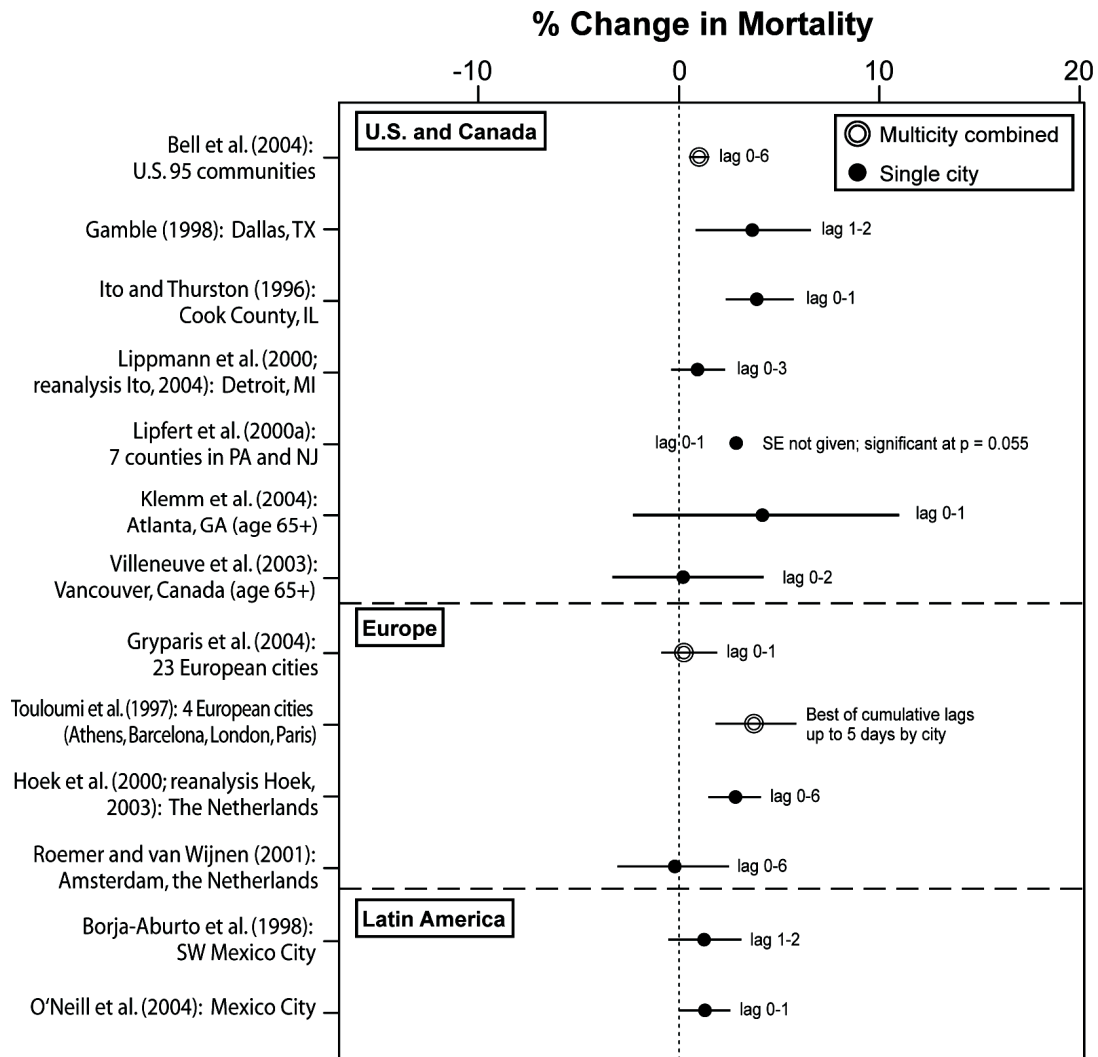


Figure 7-15. All cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) for all year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted. Only results from multiday lag models are presented.

The results of the study by Bell et al. (2004) are discussed in detail in this document because of the study's emphasis on U.S. data and the inclusion of 95 large communities across the country, making this mortality study the most representative one of the U.S. population. In addition, this study is one of the few that have focused specifically on O₃ hypotheses testing and investigated several important issues. Among the 95 communities examined in this study, 55 monitored O₃ throughout the year and 32 monitored O₃ only during the warm season (generally April to October). Eight additional cities switched from warm-season only to year-

round monitoring or year-round to warm-season only monitoring at some point during the study period. Figure 7-16 presents the median 24-h avg O₃ concentrations for the 95 communities from 1987 to 2000. In the 55 communities that had all-year data available, the median 24-h avg O₃ concentrations ranged from 14.4 ppb in Newark, NJ to 37.3 ppb in Bakersfield, CA. In the 40 communities with warm-season only data, the median 24-h avg O₃ levels ranged from 20.4 ppb in Portland, OR to 36.2 ppb in Memphis, TN. The range of median concentrations from communities monitored all year overlap considerably with the range from communities that only monitored during the warm season. This is expected, given that communities that have higher O₃ concentrations year-round are generally monitored throughout the year. The mean 24-h avg O₃ concentration from all available data was approximately 26 ppb for the 95 communities.

Within-community results were first calculated, using single-day lags of 0, 1, 2, and 3 days, and a 7-day distributed lag in O₃ exposure. The individual-community maximum likelihood effect estimates for all cause mortality per 20 ppb increase in 24-h avg O₃ from a constrained 7-day distributed lag model are presented in Figure 7-17. The heterogeneity of the effect estimates from the individual communities is partially attributable to differences in pollution characteristics, the use of air conditioning, time-activity patterns, and socioeconomic factors. A two-stage Bayesian hierarchical model was also used to determine a national average effect estimate, taking into consideration community-to-community variation. Figure 7-18 presents the Bayesian community-specific and national average O₃ risk estimates for all cause mortality per 20 ppb increase in 24-h avg O₃ with a constrained 7-day distributed lag. The Bayesian community-specific estimates were shrunk to the national average estimate by a factor that was inversely proportional to the heterogeneity of the community-specific relative rates. Because of the random variation, as well as the smaller sample sizes within each city, particular consideration is given to the Bayesian national average effect estimate.

In the U.S. 95 communities study, the largest risk estimate for O₃-mortality was obtained with a 0-day lag, followed by diminishing risk estimates with 1-, 2-, and 3-day lags (Figure 7-19, upper panel). Ozone exposure at a 0-day lag was associated with a 0.50% (95% PI: 0.24, 0.78) excess risk in mortality per 20 ppb increase in 24-h avg O₃. The 7-day distributed lag model, which examined the cumulative effect from the same day and six previous days, also is shown in Figure 7-19 (lower panel). A cumulative excess mortality risk of 1.04% (95% PI: 0.54, 1.55)

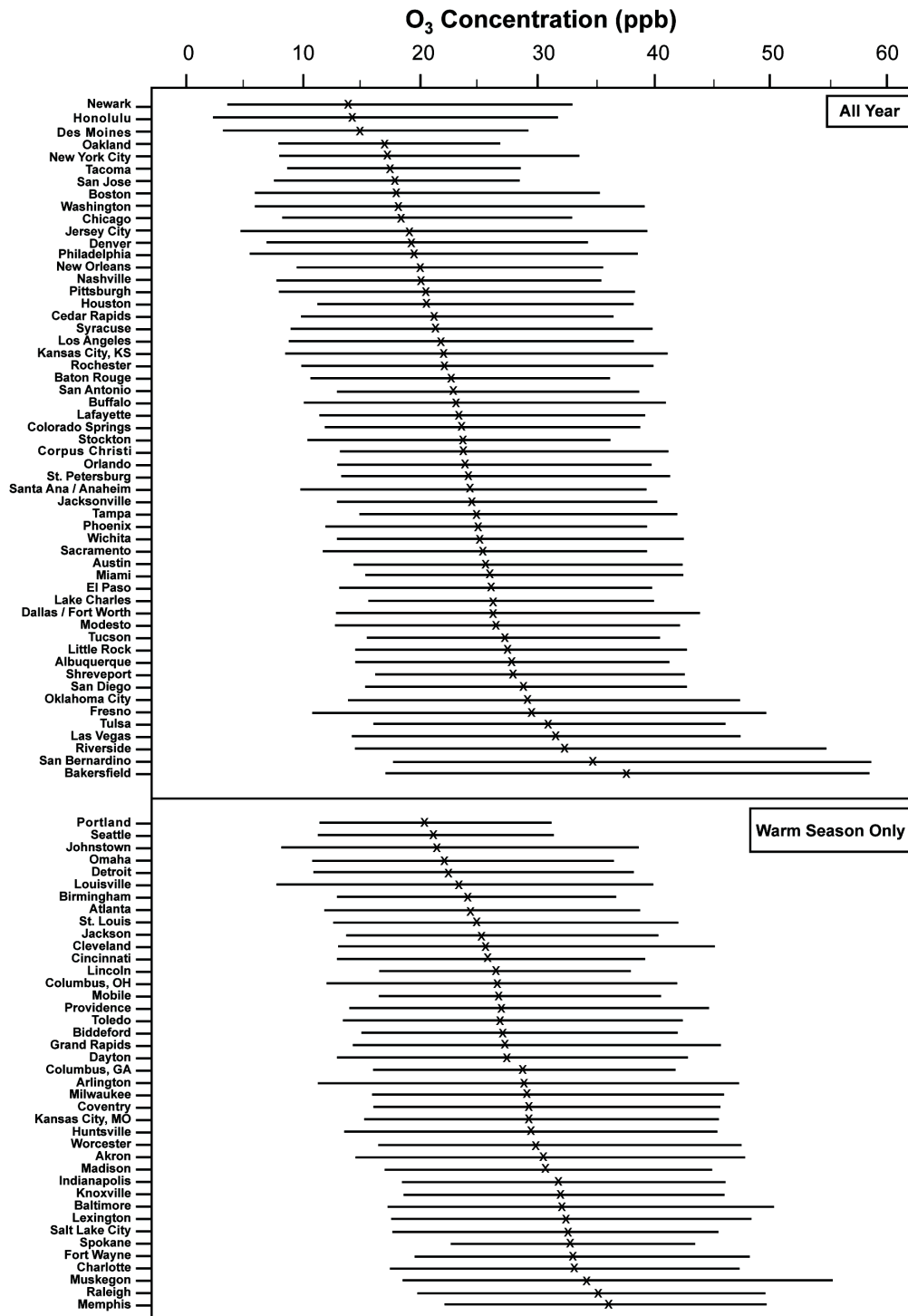


Figure 7-16. Median 24-h avg O₃ concentrations (10th percentile to 90th percentile range) for 95 U.S. communities (NMMAPS) from 1987 to 2000, arranged by O₃ concentration. Results from all available data are presented. Fifty-five of the 95 communities had year-round data. The remaining communities mostly had data during the warm season only.

Source: Derived from iHAPSS (2005).

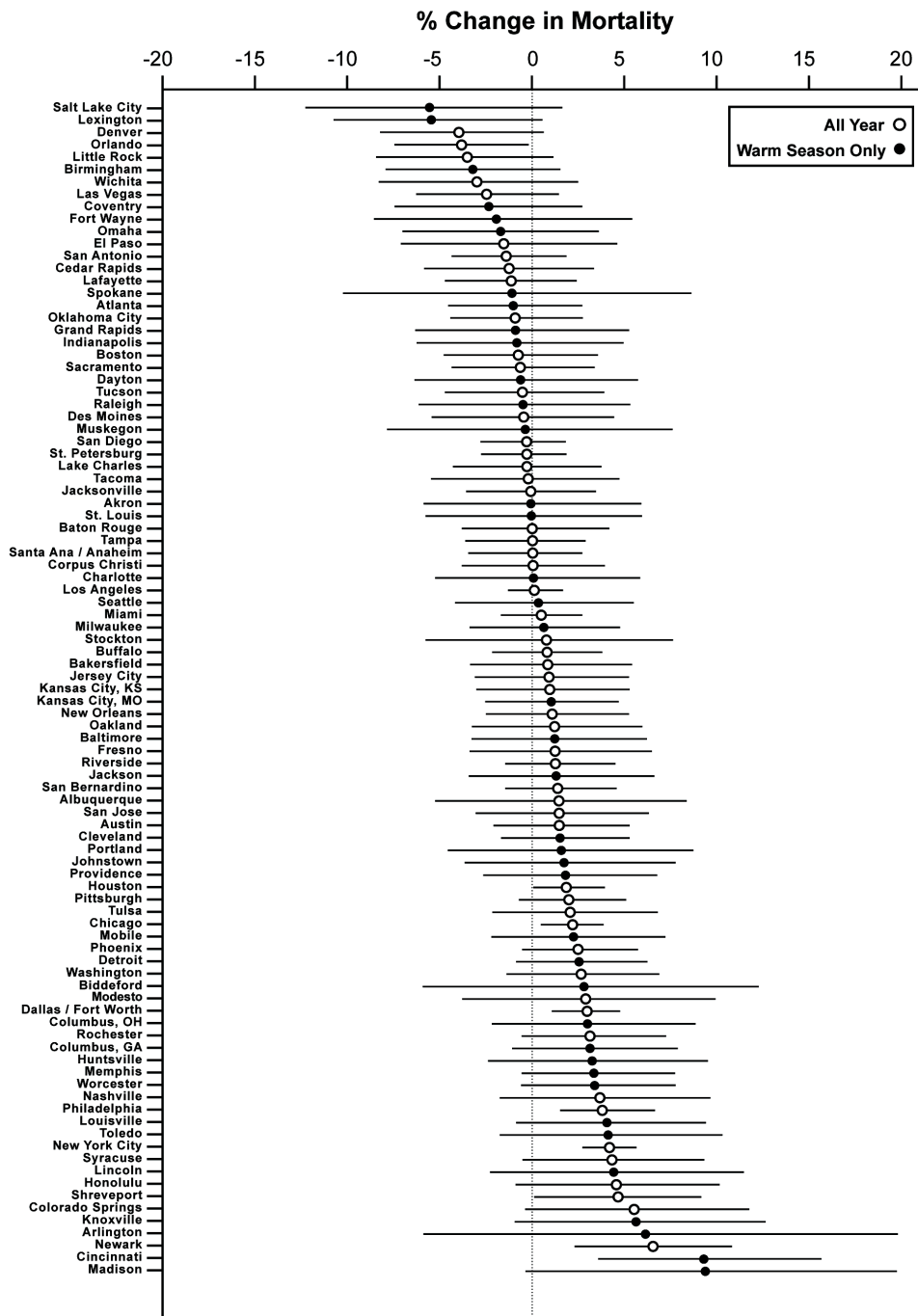


Figure 7-17. Community-specific maximum likelihood estimates for the percent change (95% CI) in daily mortality per 20 ppb increase in 24-h avg O₃ in the previous week using a constrained distributed lag model for 95 U.S. communities (NMMAPS), arranged by size of the effect estimate. Results from all available data are presented. Fifty-five of the 95 communities had year-round data, marked by open circles (○). The cities with mostly warm-season data are marked by closed circles (●).

Source: Derived from Bell (2006).

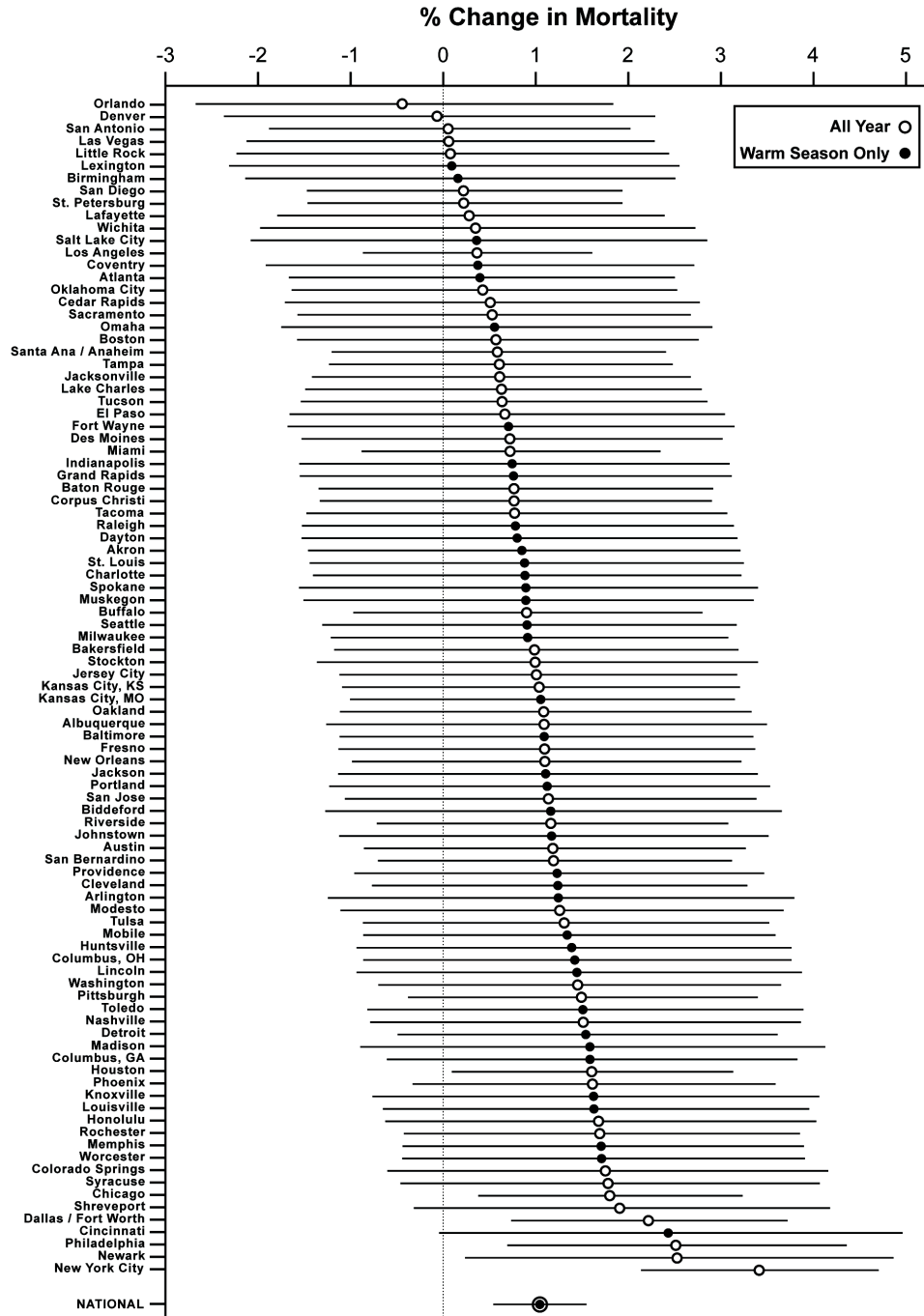


Figure 7-18. Community-specific Bayesian estimates and national average for the percent change (95% PI) in daily mortality per 20 ppb increase in 24-h avg O₃ in the previous week using a constrained distributed lag model for 95 U.S. communities (NMMAPS), arranged by size of the effect estimate. Results from all available data are presented. Fifty-five of the 95 communities had year-round data, marked by open circles (○). The cities with mostly warm-season data are marked by closed circles (●).

Source: Derived from Bell et al. (2004).

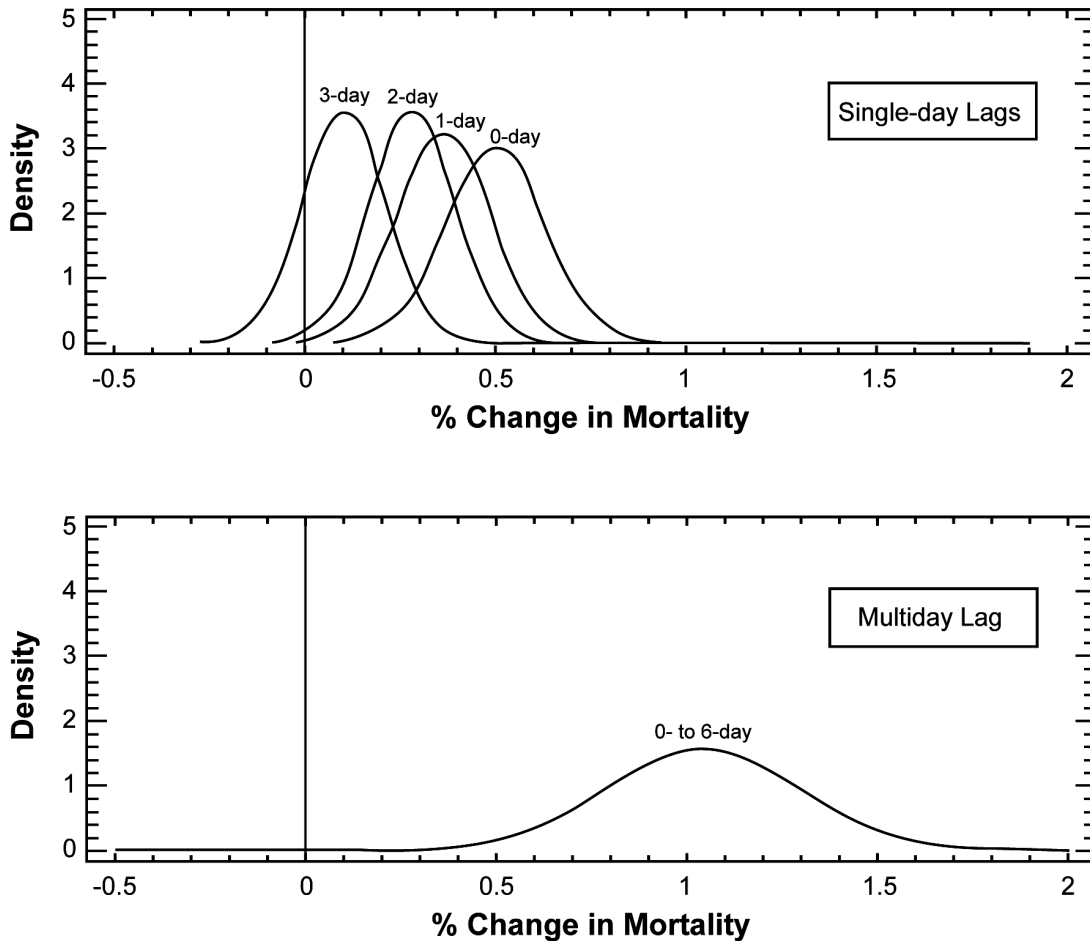


Figure 7-19. Percent changes in all cause mortality per 20 ppb increase in 24-h avg O₃ in all ages. Single-day lags (0-, 1-, 2-, and 3-day) are shown in the upper panel. The cumulative multiday lag (0- to 6-day) is shown in the lower panel.

Source: Derived from Bell et al. (2004).

per 20 ppb increase in 24-h avg O₃ during the previous week was observed. In a related U.S. study of the 19 largest cities by Huang et al. (2005), the O₃ estimate for the summer season was 1.47% (95% PI: 0.54, 2.39) excess risk of cardiopulmonary mortality with current-day exposure. Mean 24-h avg O₃ levels ranged from 18 ppb in Oakland, CA to 56 ppb in San Bernadino, CA. Smaller effects also were observed with 1- and 2-day lags of exposure. The effect estimate for the 7-day distributed lag was 2.52% (95% PI: 0.94, 4.10) excess risk of cardiopulmonary mortality. These findings suggest that the effect of O₃ on mortality is immediate, but also may persist over multiple days.

The influence of higher O₃ levels on the risk estimate also was evaluated in the U.S. 95 communities study. When the data were restricted to days with 24-h avg O₃ levels less than 60 ppb for the 1-day lag analysis, the national estimate did not substantially change (0.30% [95% PI: 0.06, 0.54] per 20 ppb increase for days with levels below 60 ppb versus 0.36% [95% PI: 0.12, 0.61] for all days). These results suggest that the O₃-mortality associations may occur at 24-h avg O₃ levels below 60 ppb.

Schwartz (2005) examined O₃-mortality associations using data from 14 U.S. cities. Median 1-h max O₃ levels ranged from 35.1 ppb in Chicago, IL to 60.0 ppb in Provo, UT. A case-crossover study design was used to compare the influence of adjustment methods for temperature (regression splines of temperature versus matching case and control periods by temperature). The risk estimate obtained by matching (0.92% [95% CI: 0.06, 1.80] per 40 ppb increase in 1-h max O₃ at a 0-day lag) was similar to that obtained with regression splines (0.76% [95% CI: 0.13, 1.40]), suggesting that the O₃-mortality risk estimates were not sensitive to these adjustment methods for temperature.

The APHEA 1 project (Touloumi et al., 1997) reported a pooled random effects estimate of 4.5% (95% CI: 1.6, 7.7) per 40 ppb increase in 1-h max O₃ using the best single-day lag model results from four European cities (Athens, Barcelona, London, and Paris). Mean 1-h max O₃ levels ranged from 21.2 ppb in London to 48.4 ppb in Athens. As an extension of the four European cities study, researchers of the APHEA 2 project investigated the effect of O₃ on total, cardiovascular, and respiratory mortality in 23 cities throughout Europe (Gryparis et al., 2004). Ozone data were available year-round in all 23 cities. A cumulative lag of 0 to 1 days was hypothesized a priori. A two-stage hierarchical model, which accounted for statistical variance and heterogeneity among cities, was used to estimate the pooled regression coefficients. Because of substantial heterogeneity among cities, random effects regression models were applied. The pooled effect estimate for the 23 European cities (0.23% [95% CI: -0.85, 1.95] per 40 ppb increase in 1-h max O₃ for all seasons) was positive but considerably smaller compared to that obtained in the APHEA 1 study. Mean O₃ concentrations for all-year data were not presented. Gryparis et al. (2004) noted that there was a considerable seasonal difference in the O₃ effect on mortality; thus, the small effect for the all-year data might be attributable to inadequate adjustment for confounding by season. This seasonal effect is discussed further in the next section.

Collectively, the single-pollutant model estimates from the single- and multiple-city studies shown in Figures 7-14 and 7-15 suggest that an excess risk of total nonaccidental mortality is associated with acute O₃ concentrations. Despite the different analytical approaches and alternative model specifications used in the various studies, overall, most of the positive estimates range from 0.5 to 5% excess risk in mortality per standardized increment for all single-pollutant, all-year analyses.

7.4.4 Meta-Analyses of Ozone-Mortality Risk Estimates

Several studies in recent years conducted meta-analyses of O₃-mortality associations (Levy et al., 2001; Stieb et al., 2002, 2003; Thurston and Ito, 2001; World Health Organization, 2004). Figure 7-20 presents the combined O₃ risk estimates from the various meta-analyses. Most of these analyses included GAM studies using default convergence criteria except for the one by Stieb et al. (2003), which compared effect estimates from GAM-affected studies to non-GAM studies. All of these meta-analyses reported fairly consistent and positive combined estimates, approximately 2% excess total nonaccidental mortality per standardized increment (see Section 7.1.3.2). However, most of these studies were not analytical in design, in that they did not attempt to examine the source of heterogeneity, although one suggested an influence of weather model specification (Thurston and Ito, 2001) and another reported evidence of publication bias (World Health Organization, 2004) in the past literature. None of these studies address the issue of season-specific estimates, therefore, interpreting these combined estimates requires caution.

Most recently, three research groups conducted independent meta-analyses of O₃-mortality associations (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005) which attempted to evaluate the source of heterogeneity using the most up-to-date literature database. These analyses have also been systematically compared and discussed (Bates, 2005; Goodman, 2005). The all-season combined point estimates per standardized increment from these three meta-analyses were remarkably consistent: 1.75% (95% PI: 1.10, 2.37), 1.6% (95% CI: 1.1, 2.0), and 1.64% (95% CI: 1.25, 2.03), for the Bell et al., Ito et al., and Levy et al. studies, respectively. All three studies also indicated that the estimates were higher in warm seasons. Each of these studies is briefly summarized below. Their findings related to specific issues are discussed later in the corresponding sections.

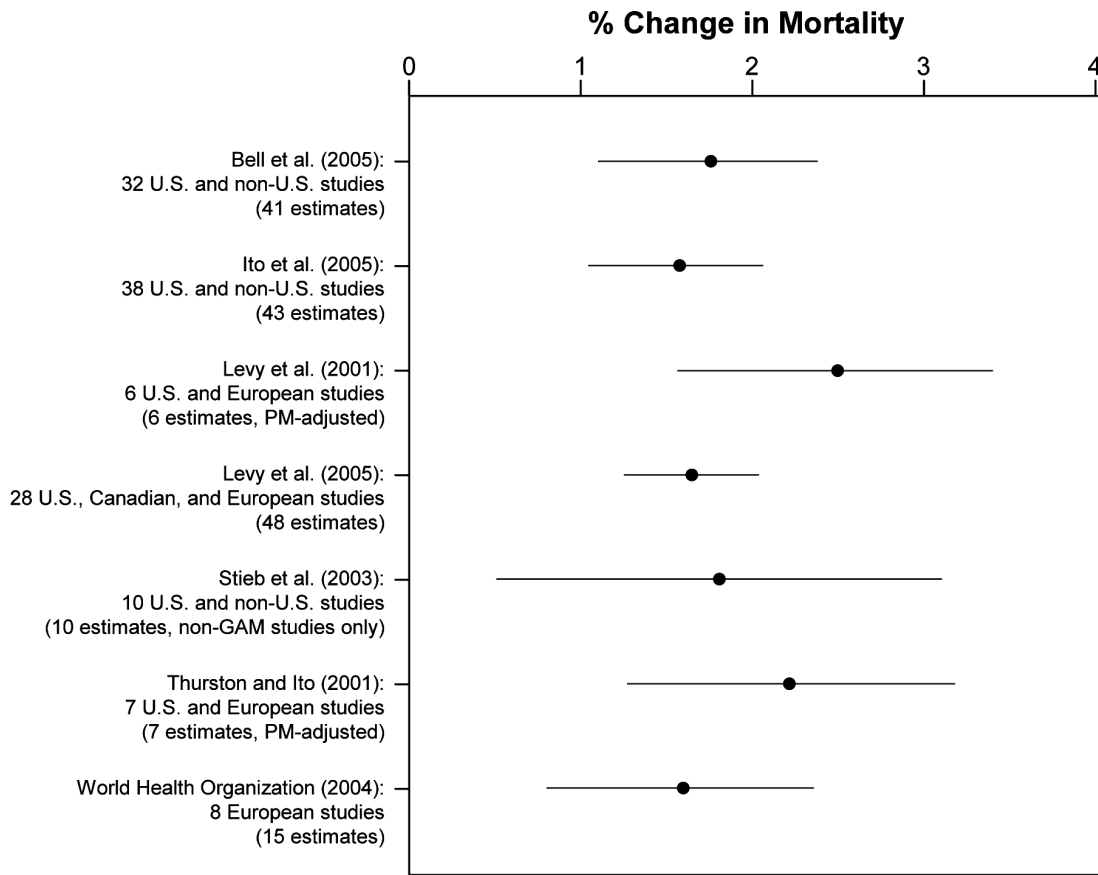


Figure 7-20. Combined all cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) from recent meta-analyses per standardized increment (see Section 7.1.3.2). Note that all meta-analyses, except Stieb et al. (2003), included studies which used Poisson GAM with default convergence criteria.

Bell et al. (2005) conducted a meta-analysis of 144 effect estimates from 39 U.S. and non-U.S. studies and estimated pooled effects by lags, age groups, specific causes, and exposure metrics. The results were also compared with their NMMAPS results (Bell et al., 2004). A two-stage Bayesian hierarchical model was used to estimate the combined estimate by taking into account the within-city variance (the statistical uncertainty) and between-study variance (the heterogeneity across cities). Bell et al. (2005) concluded that the results provided strong evidence of a short-term association between O₃ and mortality that was not sensitive to adjustment for PM or for model specifications (discussed in Section 7.4.6). However, they

suggested that, based on comparisons between the meta-analysis results and NMMAPS results, there was evidence of publication bias (1.75% [95% CI: 1.10, 2.37] per 20 ppb increase in 24-h avg O₃ for meta-analysis versus 0.50% [95% CI: 0.24, 0.78] for NMMAPS 0-day lag results).

Ito et al. (2005) both conducted a meta-analysis of 43 U.S. and non-U.S. studies and also analyzed data from 7 U.S. cities to further examine issues identified by their meta-analysis. Adjusting for PM did not substantially influence the O₃-mortality effect estimates in either the meta-analysis or 7 U.S. cities analysis. The multicity analysis further indicated that the difference in the weather adjustment model could result in a 2-fold difference in risk estimates (e.g., 1.96% versus 0.96% in multicity combined estimates across alternative weather models for the O₃-only, all-year case). In the meta-analysis, they found suggestive evidence of publication bias (a significant asymmetry in the funnel plot), but adjusting for the asymmetry reduced the combined estimate only slightly (from 1.6% [95% CI: 1.1, 2.0] to 1.4% [95% CI: 0.9, 1.9] per 20 ppb increase in 24-h avg O₃). The extent of potential bias implicated in this study differed compared to that reported by Bell et al. (2005). The source of this difference is not clear, but Ito et al. stated that sensitivity analyses comparing estimates from commonly used weather model specifications suggest that the stringent weather model used in NMMAPS may tend to yield smaller risk estimates than those used in other studies.

Levy et al. (2005) analyzed 48 estimates from 28 studies from the United States, Canada, and Europe using an empiric Bayesian meta-regression with covariates including the relationship between O₃ and other pollutants, proxies for the relationship between personal exposure and ambient concentration such as air conditioning prevalence, and statistical methods used. They found that the air conditioning prevalence (a greater effect in cities with less air conditioning) and lag time (same-day effects larger than lagged effects) were the strongest predictors of between-study variability. The warm-season estimates were larger than the cool-season estimates. The influences of copollutants were inconsistent, but a potential influence of summertime PM_{2.5} was found.

As stated earlier, the combined O₃ excess mortality risk estimates from the meta-analyses by Bell et al. (2005), Ito et al. (2005), and Levy et al. (2005) were all very consistent. Although the analyses were conducted independently, there was considerable overlap among the risk estimates used in the three meta-analyses; thus, the agreement in the combined risk estimates was not unexpected. The common findings among these three meta-analyses, aside from the

consistency in their combined estimates, include: (1) no difference in estimates between GAM studies using default versus stringent convergence criteria; (2) larger estimates in warm seasons; and (3) no strong indication of PM confounding. Both the Bell et al. and Levy et al. analyses found that the estimates at lag 0-day were larger than for longer lags. Both the Bell et al. and Ito et al. studies also suggested evidence of publication bias. The positive O₃ effect estimates, along with the sensitivity analyses in these three meta-analyses provide evidence of a robust association between ambient O₃ and mortality. The combined effect estimates from the various meta-analyses ranged from 1.5 to 2.5% excess risk in all-cause mortality.

7.4.5 Seasonal Variation in Ozone-Mortality Risk Estimates

Since the O₃ seasonal cycle follows the seasonal cycle of temperature (which is inversely related to the mortality seasonal cycle), inadequate adjustment of temporal trends in the regression model could lead to negative O₃-mortality risk estimates. In addition (as discussed in Section 7.1.3.5) low-level O₃ during the winter in some cities may be negatively correlated with PM and other primary pollutants, resulting in negative correlations between O₃ and mortality even in short-term relationships. Such a confounding effect by season could be substantially reduced by conducting season-stratified analyses.

A fewer number of O₃-mortality studies performed seasonal analyses. Figure 7-21 presents the studies that reported O₃ risk estimates for all-cause mortality by season. For those studies that obtained O₃ risk estimates for each of the four seasons, only summer and winter results are shown. The estimates for year-round data analyses, when available, also are shown for comparisons. In all the studies, the O₃ risk estimates are larger during the warm season than the cool season, with the all-year estimates generally falling between the two seasonal estimates.

In three U.S. or European multicity studies (Gryparis et al., 2004; Samet et al., 2000 [reanalysis Dominici et al., 2003]; Schwartz, 2005), season-stratified analyses indicated that the O₃-mortality effect estimates were significant and positive in the warm season, with larger effects observed compared to the year-round analyses. The effect estimates from the cool season were notably smaller and less significant. In 14 U.S. cities, Schwartz (2005) observed an excess mortality risk of 1.0% (95% CI: 0.3, 1.8) per 40 ppb increase in 1-h max O₃ during the warm season and 0% (95% CI: -1.1, 1.1) during the cold season with a 0-day lag of exposure. Ozone

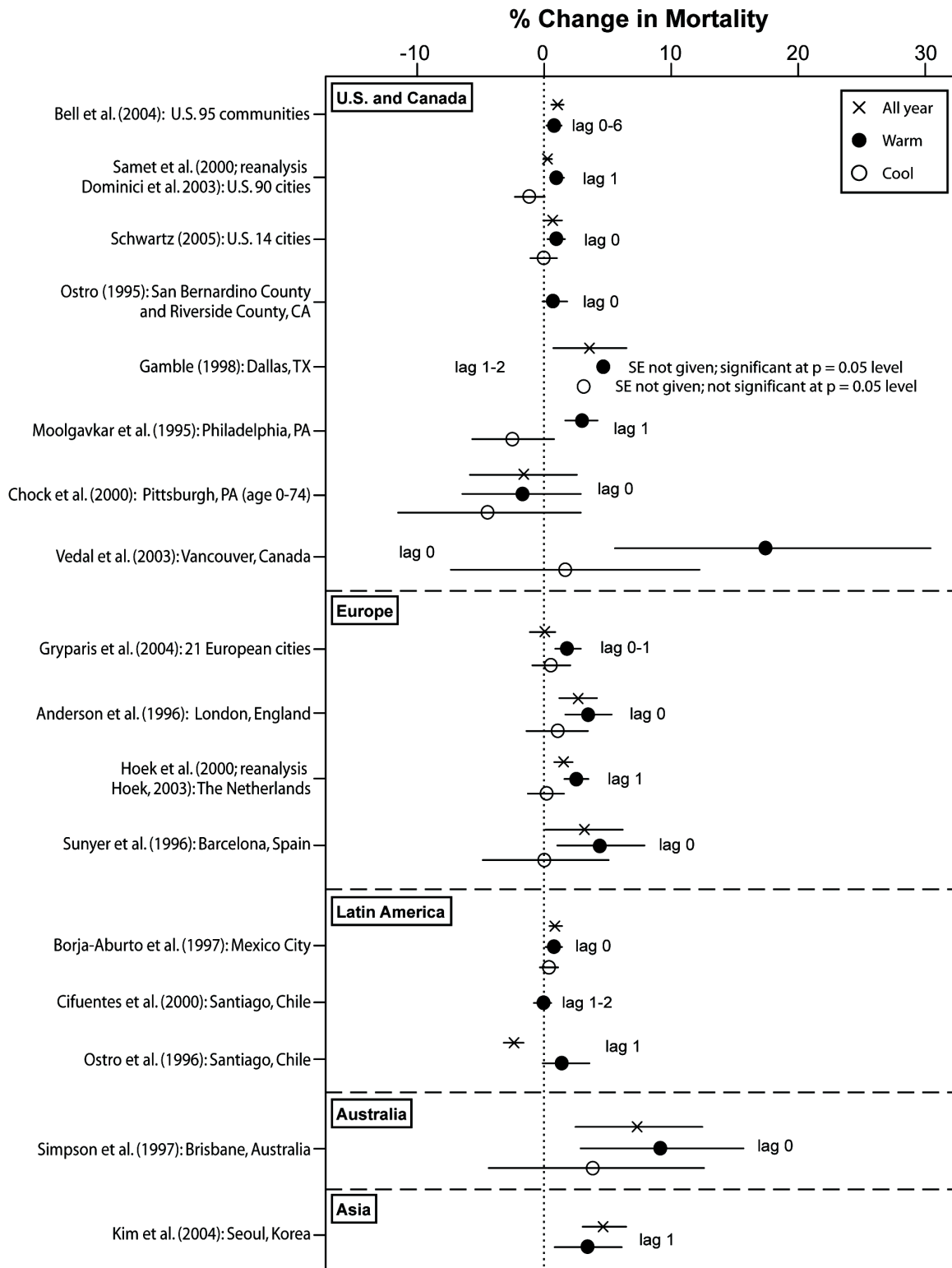


Figure 7-21. All-cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) by season per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.

concentrations by season were not presented. Gryparis et al. (2004) estimated an excess risk of 1.8% (95% CI: 0.99, 3.06) and 0.70% (95% CI: -0.70, 2.17) per 30 ppb increase in 8-h max O₃ at a 0- and 1-day lag during the warm season and cold season, respectively, in 21 European cities. The median 8-h max O₃ concentrations ranged from 30 to 99 ppb during the warm season and 8 to 49 ppb during the cold season. In the U.S. 90 cities study (of which 80 cities had O₃ data available), the winter (December, January, and February) mortality estimate was negative (Samet et al., 2000 [reanalysis Dominici et al., 2003]) and was most likely attributable to the inverse relationship between O₃ and PM in the winter.

In the U.S. 95 communities study by Bell et al. (2004), no significant difference was found between the estimates from all available data and warm-season-only data (for April- October); however, no cool-season-only analyses were performed. The warm-season effect estimate using the 7-day constrained distributed lag model was 0.78% (95% PI: 0.26, 1.30) excess risk per 20 ppb increase in 24-h avg O₃, compared to 1.04% (95% PI: 0.54, 1.55) calculated using all available data. In the 55 communities with year-round O₃ data, the all-year effect estimate was 0.96% (95% PI: 0.32, 1.57). As stated in the previous section, the range of median concentrations from the 55 communities monitored all year overlapped considerably with the range from the 40 communities with warm-season-only data (see Figure 7-16).

All three recent meta-analyses (Bell et al., 2005; Ito et al., 2005; Levy et al. 2005), found that the warm-season estimates were larger than all-year estimates. In Bell et al. (2005), the warm-season estimate was 3.02% (95% PI: 1.45, 4.63), compared to the all-year estimate of 1.75% (95% PI: 1.10, 2.37). In the subset of 10 cities examined by Ito et al. (2005), the warm-season and all-year estimates were 3.5% (95% CI: 2.1, 4.9) and 2.2% (95% CI: 0.8, 3.6), respectively. Similarly, Levy et al. (2005) observed a 3.38% (95% CI: 2.27, 4.42) excess risk in the warm season compared to a 1.64% (95% CI: 1.25, 2.03) excess risk using all-year data. All results presented are percent excess risk in mortality per standardized increment.

Studies that conducted analysis by season indicate that O₃-mortality risk estimates are often larger in the warm season compared to the colder season. The seasonal dependence of O₃-mortality effects complicates interpretation of O₃ risk estimates calculated from year-round data without adequate adjustment of temporal trends.

7.4.6 Ozone-Mortality Risk Estimates Adjusting for PM Exposure

The confounding between “winter type” pollution (e.g., CO, SO₂, and NO₂) and O₃ is not of great concern, because the peaks of these pollutants do not strongly coincide. The main confounders of interest for O₃, especially for the northeast United States, are “summer haze-type” pollutants such as acid aerosols and sulfates. Since very few studies included these chemical measurements, PM (especially PM_{2.5}) data, may serve as surrogates. However, due to the expected high correlation among the constituents of the “summer haze mix,” multipollutant models including these pollutants may result in unstable coefficients; and, therefore, interpretation of such results requires some caution.

Figure 7-22 shows the O₃ risk estimates with and without adjustment for PM indices using all-year data in studies that conducted two-pollutant analyses. Approximately half of the O₃ risk estimates increased slightly, whereas the other half decreased slightly with the inclusion of PM in the models. In general, the O₃-mortality risk estimates were robust to adjustment for PM in the models, with the exception of Los Angeles, CA data with PM₁₀ (Kinney et al., 1995) and Mexico City data with TSP (Borja-Aburto et al., 1997).

The U.S. 95 communities study by Bell et al. (2004) examined the sensitivity of acute O₃-mortality effects to potential confounding by PM₁₀. Restricting analysis to days when both O₃ and PM₁₀ data were available, the community-specific O₃-mortality effect estimates as well as the national average results indicated that O₃ was robust to adjustment for PM₁₀ (Bell et al., 2004). There were insufficient data available to examine potential confounding by PM_{2.5}. One study (Lipfert et al., 2000a) reported O₃ risk estimates with and without sulfate adjustment. Lipfert et al. (2000a) calculated O₃ risk estimates based on mean (45 ppb) less background (not stated) levels of 1-h max O₃ in seven counties in Pennsylvania and New Jersey. The O₃ risk estimate was not substantially affected by the addition of sulfate in the model (3.2% versus 3.0% with sulfate) and remained statistically significant.

Several O₃-mortality studies examined the effect of confounding by PM indices in different seasons (Figure 7-23). In analyses using all-year data and warm-season only data, O₃ risk estimates were once again fairly robust to adjustment for PM indices, with values showing both slight increases and decreases with the inclusion of PM in the model. In the analyses using cool-season data only, the O₃ risk estimates all increased slightly with the adjustment of PM indices, although none reached statistical significance.

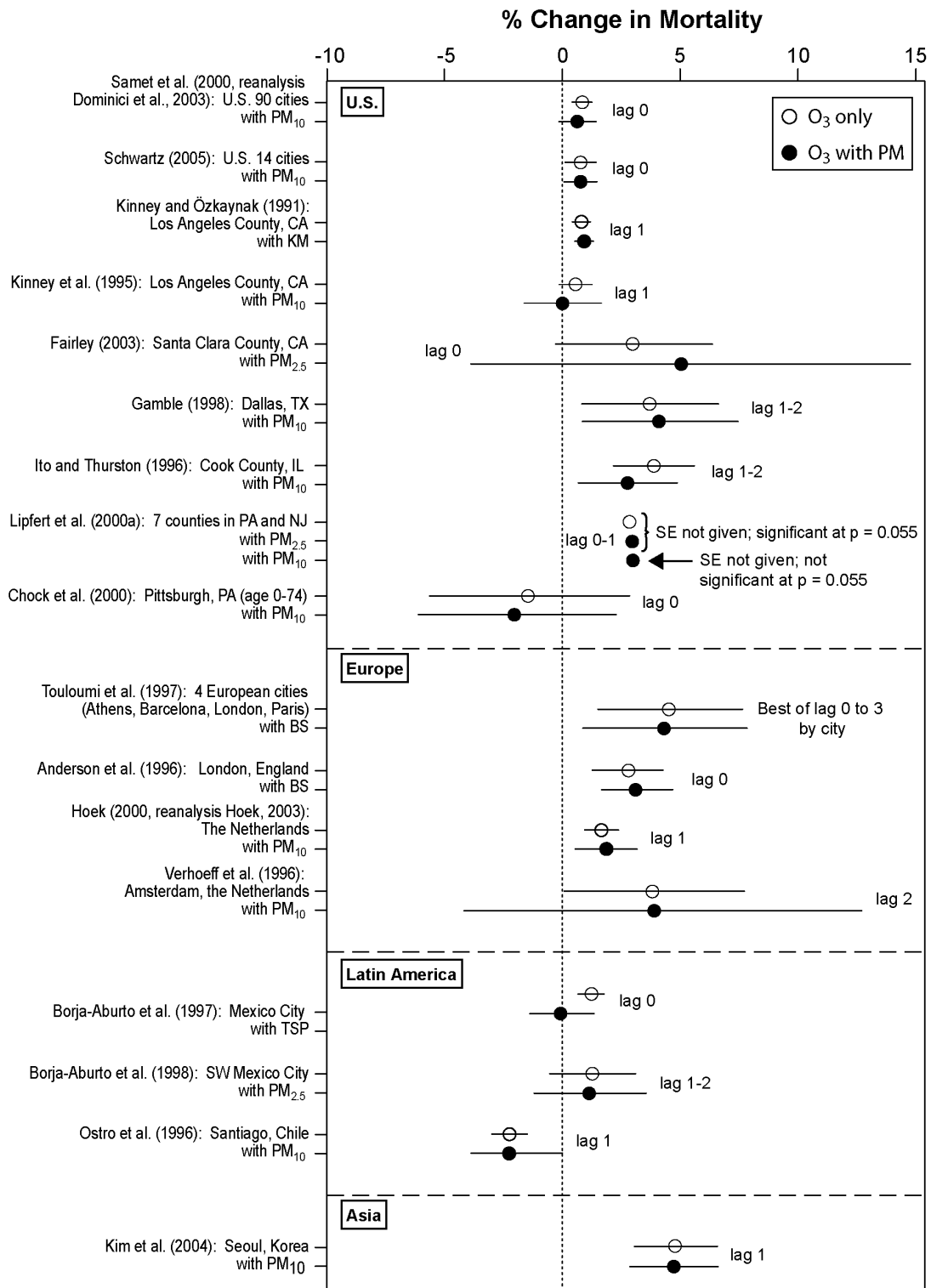


Figure 7-22. All-cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) with adjustment for PM indices for all-year analyses per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.

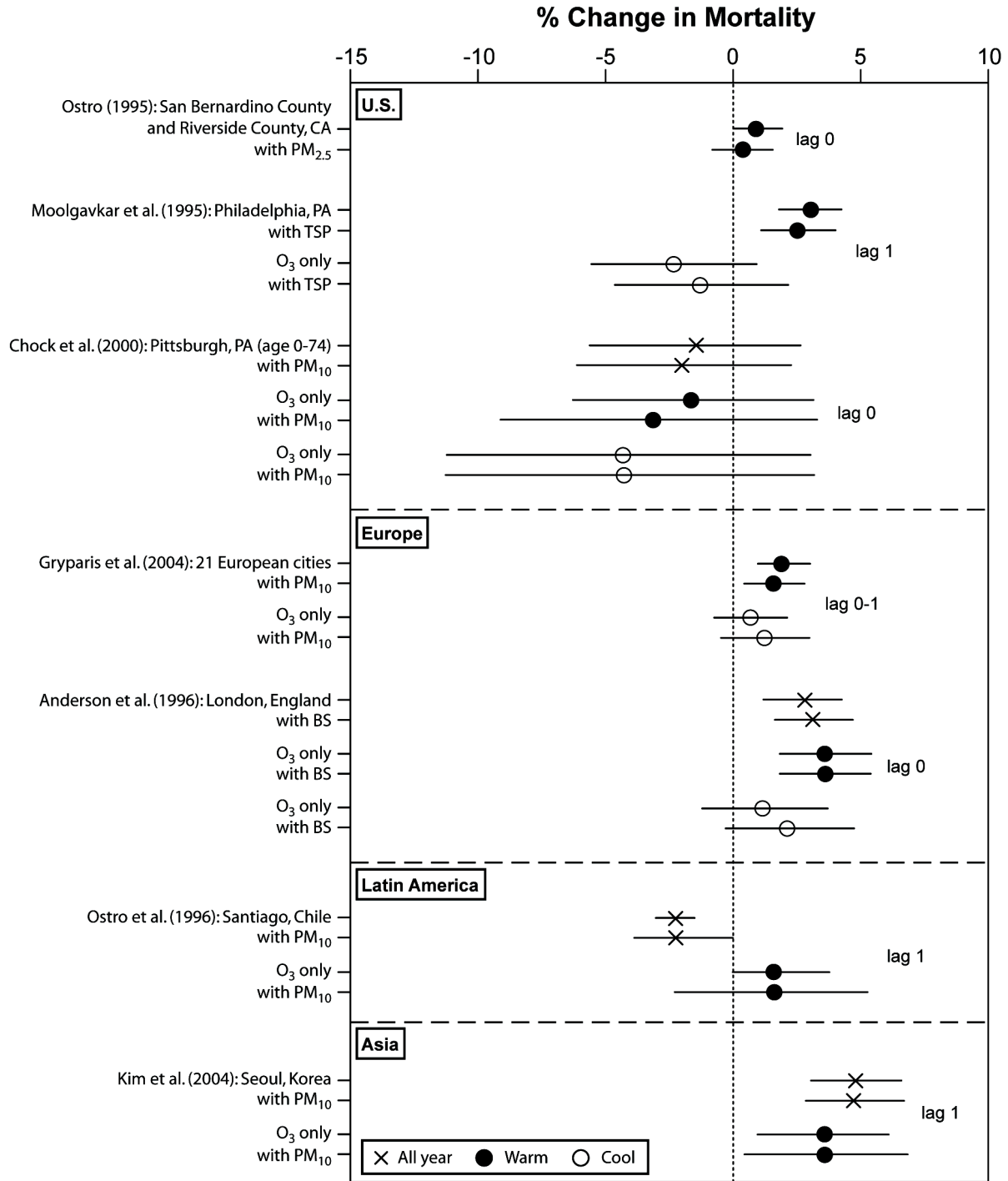


Figure 7-23. All-cause (nonaccidental) O₃ excess mortality risk estimates (95% CI) with adjustment for PM indices by season per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.

The three recent meta-analyses (Bell et al., 2005; Ito et al., 2005; Levy et al. 2005) all examined the influence of PM on O₃ risk estimates. No substantial influence was observed in any of these studies. In the analysis by Bell et al. (2005), the combined estimate without PM adjustment was 1.75% (95% PI: 1.10, 2.37) from 41 estimates, and the combined estimate with PM adjustment was 1.95% (95% PI: -0.06, 4.00) from 11 estimates per 20 ppb increase in 24-h avg O₃. In the meta-analysis of 15 cities by Ito et al. (2005), the combined estimate was 1.6% (95% CI: 1.1, 2.2) and 1.5% (95% CI: 0.8, 2.2) per 20 ppb in 24-h avg O₃ without and with PM adjustment, respectively. The additional time-series analysis of six cities by Ito et al. found that the influence of PM by season varied across alternative weather models but was never substantial. Levy et al. (2005) examined the regression relationships between O₃ and PM indices (PM₁₀ and PM_{2.5}) with O₃-mortality effect estimates for all year and by season. Positive slopes, which might indicate potential confounding, were observed for PM_{2.5} on O₃ risk estimates in the summer and all-year periods, but the relationships were weak. The effect of one causal variable (i.e., O₃) is expected to be overestimated when a second causal variable (e.g., PM) is excluded from the analysis, if the two variables are positively correlated and act in the same direction. However, the results from these meta-analyses, as well as several single- and multiple-city studies, indicate that copollutants generally do not appear to substantially confound the association between O₃ and mortality.

7.4.7 Ozone Risk Estimates for Specific Causes of Mortality

In addition to all-cause mortality, several studies examined broad underlying causes of mortality, such as cardiovascular and respiratory causes. The U.S. 95 communities study (1987-2000) analyzed O₃ effect estimates from cardiovascular and respiratory mortality (Bell et al., 2004). The analysis by Bell et al. (2005) used all available data, which included all-year data from 55 communities and warm-season only data from 40 communities. The national average estimate from the constrained distributed lag model was slightly greater for cardiopulmonary deaths than deaths from all causes, with an excess risk of 1.28% (95% PI: 0.62, 1.97) compared to 1.04% (95% PI: 0.54, 1.55) per 20 ppb increase in 24-h avg O₃ in the preceding week.

In a related study, Huang et al. (2005) examined O₃ effects on cardiopulmonary mortality during the summers (June to September) of 1987 to 1994 in 19 large U.S. cities from the NMMAPS database. Figure 7-24 presents the Bayesian city-specific and overall average O₃ risk

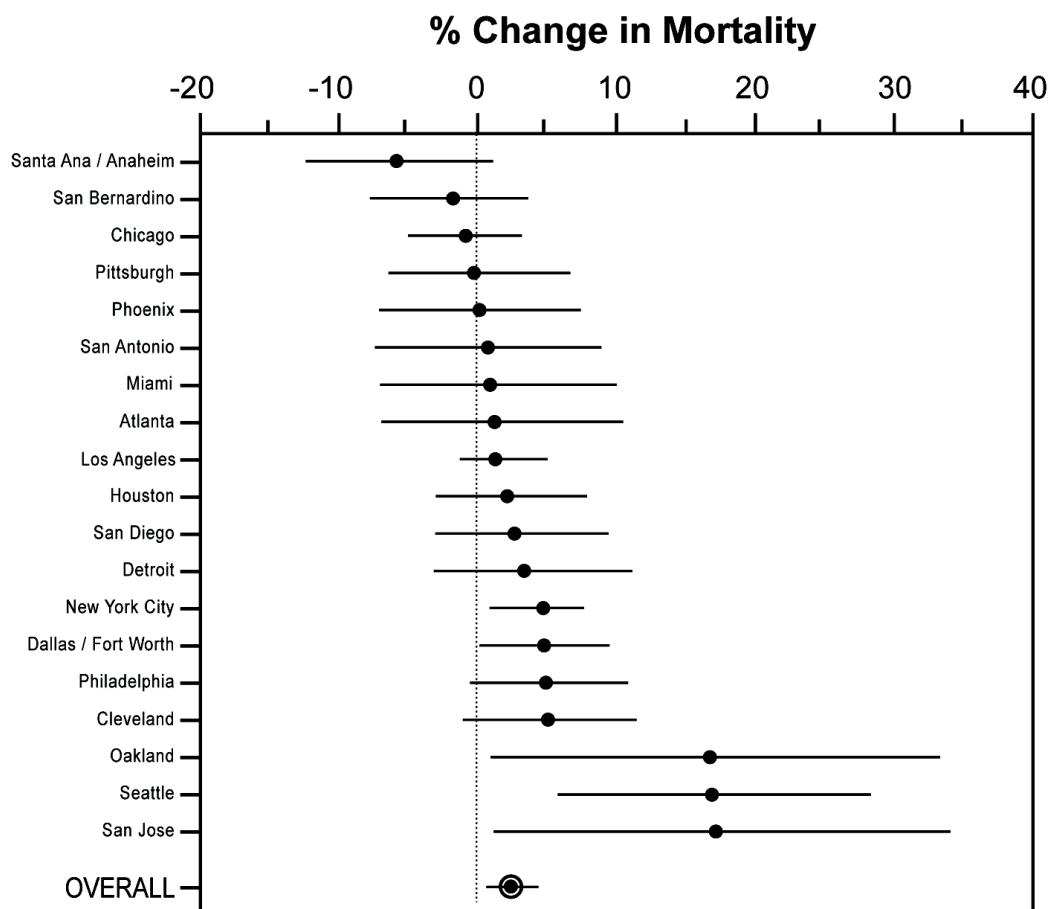


Figure 7-24. City-specific Bayesian estimates and overall average for the percent change (95% PI) in cardiopulmonary mortality per 20 ppb increase in 24-h avg O₃ in the previous week using a constrained distributed lag model for 19 U.S. cities (NMMAPS), arranged by size of the effect estimate. Analyses were conducted using summer (June to September) data only.

Source: Derived from Huang et al. (2005).

estimates for cardiopulmonary mortality per 20 ppb increase in 24-h avg O₃ from a constrained 7-day distributed lag model. The O₃ effect estimate was 2.52% (95% PI: 0.94, 4.10) excess risk in cardiopulmonary mortality per 20 ppb increase in 24-h avg O₃ in the preceding week for the combined analysis of all cities. For analyses of summer data, confounding of the O₃ effect by PM is of concern as daily variations in O₃ may be correlated to PM during the summer months. Huang et al. observed that when PM₁₀ was included in the model, the O₃ effect estimate, on average, remained positive and significant. As PM₁₀ measurements were available only every

1 to 6 days, only single-day lags were examined. At a 0-day lag, O₃ was associated with a 1.47% (95% PI: 0.54, 2.39) excess risk versus a 1.49% (95% PI: -0.66, 3.47) excess risk in cardiopulmonary mortality in the O₃-only model and after adjustment for PM₁₀, respectively. The slight sensitivity of the O₃ health effects to the inclusion of PM₁₀ in the model may indicate a true confounding effect. However, as only the days with PM₁₀ data available were included in the analysis, the lack of significance is likely attributable to higher statistical uncertainty due to the lack of daily PM₁₀ measurements.

Figure 7-25 presents effect estimates for associations between O₃ and cardiovascular mortality for all-year and warm-season analyses. All studies, with the exception of Pönkä et al. (1998), showed positive associations between O₃ and cardiovascular mortality. The median 24-h avg O₃ was 9 ppb (5th % to 95th %: 2 to 26) in the Helsinki study by Pönkä et al. (1998). In addition to Cook County, IL (median 24-h avg O₃ of 18 ppb), Moolgavkar (2000) also examined the effect of O₃ on cardiovascular mortality in Los Angeles County, CA (median 24-h avg O₃ of 24 ppb) and Maricopa County, AZ (median 24-h avg O₃ of 25 ppb). Ozone was not found to be associated with cardiovascular or cerebrovascular mortality in these two counties, even when data was restricted to the warm season (quantitative risk estimates not presented). In a follow-up study, Moolgavkar (2003) analyzed all circulatory system deaths in one broad group in order to increase the power of the season-specific analyses using the Cook County and Los Angeles County data. In this analysis, O₃ concentrations were marginally associated with increased circulatory mortality during the summer in Los Angeles County (1.6% [95% CI: -0.2, 3.5] excess risk per 20 ppb increase in 24-h avg O₃). Note that both analyses were performed using Poisson GAM with default convergence criteria.

As with all-cause mortality, there appears to be heterogeneity in the effect estimates across studies. The cardiovascular mortality estimate from the meta-analysis by Bell et al. (2005) appears to be close to the mode of the effect estimates from the various studies, as shown in Figure 7-25. This is expected, given that many of these studies were also included in the meta-analysis. Bell et al. (2005) observed that the posterior mean estimate for cardiovascular causes (2.23% [95% PI: 1.36, 3.08] excess risk per 20 ppb increase in 24-h avg O₃ from 25 estimates) was slightly larger than that for total mortality (1.75% [95% PI: 1.10, 2.37] excess risk from 41 estimates). However, since cardiovascular deaths account for the largest fraction (over 40%) of total deaths, it is not surprising that the risk estimates for cardiovascular mortality

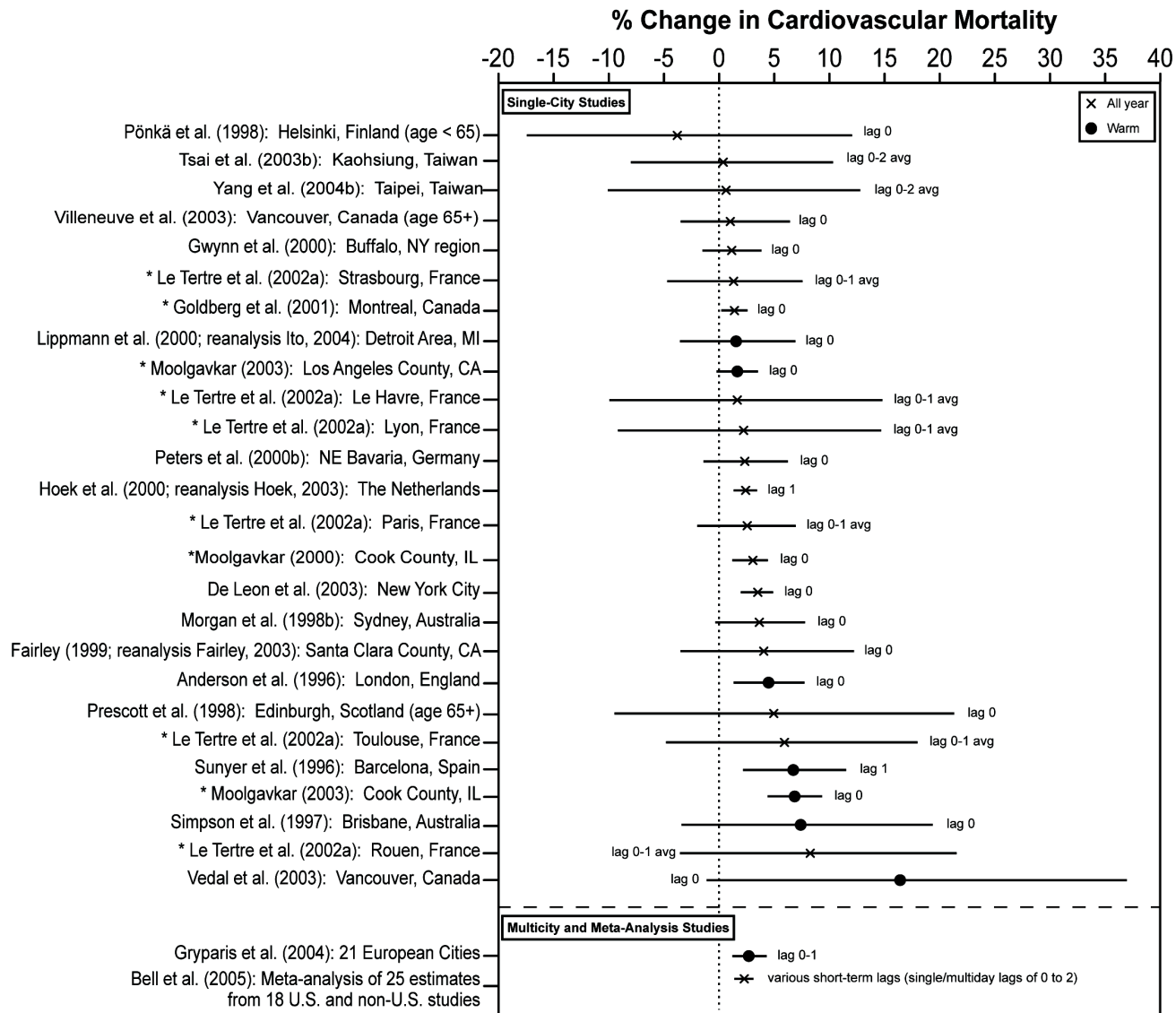


Figure 7-25. Ozone-associated cardiovascular mortality risk estimates (95% CI) per standardized increment (see Section 7.1.3.2). Analyses include all ages unless otherwise noted.

*Note that Goldberg et al. (2001), Le Tertre et al. (2002a), and Moolgavkar (2001, 2003) performed analyses using Poisson GAM with default convergence criteria.

are somewhat similar to those from all-cause mortality. Overall, the cardiovascular mortality risk estimates in the current literature show consistently positive associations with some heterogeneity (most estimates fall within the range of 1 to 8% per 40 ppb increase in 1-h avg O₃).

Several studies observed that the risk estimates for the respiratory category were larger than the cardiovascular and total nonaccidental categories (e.g., Anderson et al., 1996; Gouveia and Fletcher, 2000b; Gryparis et al., 2004; Zmirou et al., 1998). In the European 21 multicity study (Gryparis et al., 2004), the warm-season effect estimate for respiratory mortality was 6.75% (95% CI: 4.38, 9.10) excess risk per 30 ppb increase in 8-h max O₃, compared to 2.70% (95% CI: 1.29, 4.32) for cardiovascular mortality and 1.82% (95% CI: 0.99, 3.06) for total mortality. In contrast, other studies have found that the risk estimates for the respiratory category were smaller or even negative, whereas the risk estimates for total or cardiovascular categories were positive (e.g., Borja-Aburto et al., 1998; Bremner et al., 1999; Lipfert et al., 2000a; Morgan et al., 1998b). The apparent inconsistencies across studies may be due in part to the differences in model specifications, but they may also reflect the lower statistical power associated with the smaller daily counts of the respiratory category (usually accounting for less than 10% of total deaths) compared to the larger daily counts for the cardiovascular category (approximately 40 to 50% of total deaths). Thus, an examination of the differences in risk estimates across specific causes requires a large population and/or a long period of data collection. In the meta-analysis by Bell et al. (2005), which combined 23 estimates from 17 studies for respiratory mortality, the effect estimate for respiratory causes was smaller (0.94% [95% PI: -1.02, 2.96] excess risk per 20 ppb increase in 24-h avg O₃) compared to the estimates for total mortality (1.75% excess risk) and cardiovascular mortality (2.23% excess risk).

The analyses of a 9-year data set for the whole population of the Netherlands (population = 14.8 million) provided risk estimates for more specific causes of mortality, including COPD, pneumonia, and subcategories of cardiovascular causes (Hoek et al., 2000, 2001; reanalysis Hoek, 2003). The median 8-h avg O₃ (12 p.m.-8 p.m.) was 24 ppb (range 1 to 117). The effect estimate for total nonaccidental mortality was 1.6% (95% CI: 0.9, 2.4) excess risk per 30 ppb increase in 8-h avg O₃. In comparison, the excess risk estimates for pneumonia and COPD were 5.6% (95% CI: 1.8, 9.5) and 0.8% (95% CI: -2.4, 4.2), respectively. The effect estimates for some of the cardiovascular subcategories, including heart failure (3.8% [95%

CI: 0.5, 7.3]) and thrombosis-related disease (6.0% [95% CI: 1.1, 10.8]), showed greater risk estimates than that for total mortality. However, these associations were not specific to O₃. For example, most of the pollutants examined, including PM₁₀, BS, SO₂, NO₂, CO and NO₃⁻, were associated with pneumonia. Therefore, it is difficult to make a causal inference specific to O₃ on the basis of these results.

De Leon et al. (2003) examined the role of contributing respiratory causes in the association between air pollution and nonrespiratory mortality (circulatory and cancer) in New York City during the period of 1985 to 1994. The mean 24-h avg O₃ concentration was 21.6 ppb (5th % to 95th %: 7.0 to 45.0). The main finding of this study was that the estimated excess mortality risks for PM₁₀ were higher for nonrespiratory deaths that had contributing respiratory causes compared to deaths without contributing respiratory causes in the older age (75+ years) group. This pattern was also seen for CO and SO₂, but not for O₃. Therefore, this study did not suggest a role of contributing respiratory causes in the association between O₃ and nonrespiratory causes of deaths.

In summary, several single-city studies observed positive associations between ambient O₃ concentrations and cardiovascular mortality. In addition, a meta-analysis that examined specific causes of mortality found that the cardiovascular mortality risk estimates were higher than those for total mortality. The findings regarding the effect size for respiratory mortality have been less consistent, possibly because of lower statistical power in this subcategory of mortality.

7.4.8 Ozone-Mortality Risk Estimates for Specific Subpopulations

Some studies have examined O₃-mortality risk estimates in potentially susceptible subpopulations, such as those persons with underlying cardiopulmonary disease. Sunyer et al. (2002) examined the association between air pollution and mortality in a cohort of patients (467 men and 611 women) with severe asthma in Barcelona, Spain during the period of 1986 to 1995. A case-crossover study design was used to estimate excess odds of mortality adjusting for weather and epidemics in three groups: (1) those who had only one asthma emergency department visit; (2) those who had more than one asthma emergency department visit; and (3) those who had more than one asthma and COPD emergency department visit. Those with more than one asthma emergency department visit showed the strongest associations with the examined air pollutants, with NO₂ being the most significant predictor, followed by O₃. Sunyer

et al. (2002) reported a significant association between O₃ and all-cause deaths for this group during the warm season, with an odds ratio of 2.82 (95% CI: 1.15, 6.87) per 40 ppb increase in 1-h max O₃, compared to an odds ratio of 1.03 (95% CI: 0.60, 1.78) for those with only one asthma emergency department visit and 1.08 (95% CI: 0.60, 1.92) for the group with a concurrent diagnosis of COPD. The median 1-h max O₃ concentration for all-year data was 36 ppb, but seasonal O₃ concentrations were not presented. In another Barcelona study, Saez et al. (1999) examined asthma mortality death among persons aged 2 to 45 years. Once again, O₃ and NO₂ were the only air pollutants found to be significantly associated with asthma mortality death. While the similarity of the patterns of associations between O₃ and NO₂ makes it difficult to speculate on the specific causal role of O₃, the results of these studies suggest that individuals with severe asthma may make up a subpopulation that is sensitive to these pollutants.

Sunyer and Basagna (2001) also performed an analysis of emergency department visits for individuals with COPD. The results from this study suggested that PM₁₀, but not gases were associated with mortality risks for the COPD cohort. The mean O₃ concentration was not provided (IQR was 11 ppb for 1-h max O₃). However, a Mexico City study by Téllez-Rojo et al. (2000) observed a significant association between COPD mortality and O₃, as well as PM₁₀, among patients living outside a medical unit. For a cumulative 5-day lag, an excess risk of 15.6% (95% CI: 4.0, 28.4) per 40 ppb increase in 1-h max O₃ was observed for COPD mortality. The mean 1-h max O₃ was 69.4 ppb (SD 17.2).

Goldberg et al. (2003) investigated associations between air pollution and daily mortality with congestive heart failure as the underlying cause of death in patients aged 65 years or more in Montreal, Quebec during the period of 1984 to 1993. The mean 24-h avg O₃ was 15.0 ppb (SD 8.8). Analysis was stratified into two groups, those whose underlying cause of death was congestive heart failure and those with a diagnosis of congestive heart failure one year before their death. No association was found between daily mortality for congestive heart failure and any pollutants. However, associations were found between daily mortality and coefficient of haze, SO₂, and NO₂ among patients who were classified as having congestive heart failure before death. In the case of O₃, positive risk estimates were observed for year-round and warm-season data; however, the results were not statistically significant. Although the 10-year study period for this data was long, the daily mean death counts for the specific subcategory chosen was relatively small (0.7/day for mortality with congestive heart failure as underlying cause of death

and 4.0/day for total mortality in patients previously diagnosed with congestive heart failure), thus limiting the power of the study.

In the meta-analysis by Bell et al. (2005), a combined estimate was obtained for the elderly population (age 64 years and older or 65 years and older) using 10 estimates from 9 studies. The posterior mean estimate for the elderly category (2.92% [95% PI: 1.34, 4.51] per 20 ppb increase in 24-h avg O₃) was larger than that from all ages (1.75% [95% PI: 1.10, 2.37] from 41 estimates). The results from the Bell et al. (2005) meta-analysis suggest that the elderly population may be particularly susceptible to O₃-related mortality.

Few studies have examined O₃-mortality effects for specific subpopulations. Among those that investigated the effect of air pollution in populations with underlying cardiopulmonary diseases, associations were not unique to O₃ but, rather, were shared with other pollutants. There is suggestive evidence that severe asthmatics may be susceptible to the mortality effects associated with NO₂ and O₃. In addition, the meta-analysis by Bell et al. (2005) suggests that the elderly population may be more affected by O₃.

7.4.9 Summary of Acute Ozone Effects on Mortality

- A substantial body of new data on acute mortality effects of O₃ has emerged since the 1996 O₃ AQCD. While uncertainties remain in some areas, it can be concluded that robust associations have been identified between various measures of daily O₃ concentrations and increased risk of mortality. Most of the single-pollutant model estimates from single-city studies fall in the range between 0.5 to 5% excess deaths per standardized increment. The corresponding summary estimates in large U.S. multicity studies ranged between 0.5 to 1%, with some studies noting heterogeneity across cities and studies. These associations could not be readily explained by confounding due to time, weather, nor copollutants. Differences between populations as well as model specifications likely contributed to some of the observed heterogeneity in risk estimates across studies.
- Most studies calculated O₃-mortality risk estimates using all-year data. The results from studies that conducted stratified analyses by season suggested that the O₃ risk estimates were larger in the warm season. Some of the risk estimates in the cool season were negative, possibly reflecting the negative correlation between low-level O₃ and PM (and/or other primary pollutants) during that season. Thus, even with adjustment for temporal trends, the O₃ risk estimates obtained for year-round data may be misleading. In locations with considerable seasonal variation, season-specific analyses may better elucidate the effect of O₃ on mortality.

- The majority of the available O₃-mortality risk estimates were computed for a single-day lag. Choosing the optimal lag out of several lags examined may bias the single-day risk estimate upward. However, recent findings from the largest U.S. 95 communities study indicated that a strong association between O₃ and mortality was observed with a 7-day distributed lag model. Thus, it is possible that the effect of acute O₃ exposure on mortality persists over several days. Since the parameters estimated from single-day lag versus multiday lag models are not the same, comparison of these results are difficult.
- Some studies examined specific subcategories of mortality, but most of these studies had limited statistical power to detect associations due to the small daily mortality counts. A recent meta-analysis indicated that there was a slightly greater risk of cardiovascular mortality compared to total mortality.
- Few studies examined the effect of O₃ on mortality in subpopulations with underlying cardiopulmonary diseases. Similar to cause-specific mortality, these population-specific studies had limited statistical power to detect associations.

7.5 EFFECTS OF CHRONIC OZONE EXPOSURE

7.5.1 Summary of Key Findings on Studies of Health Effects and Chronic Ozone Exposure from the 1996 Ozone AQCD

The 1996 O₃ AQCD concluded that there was insufficient evidence from the limited number of studies to determine whether long-term ambient O₃ exposures resulted in chronic health effects. However, the aggregate evidence suggested that chronic O₃ exposure, along with other environmental factors, could be responsible for health effects in exposed populations.

7.5.2 Introduction to Morbidity Effects of Chronic Ozone Exposure

Several new longitudinal epidemiologic investigations have yielded information on health effects of long-term O₃ exposures. Epidemiologic interest in investigating long-term effects has been motivated by several considerations. Animal toxicology studies carried out from the late 1980s onward demonstrated that long-term exposures can result in permanent changes in the small airways of the lung, including remodeling of the airway architecture (specifically the distal airways and centriacinar region) and deposition of collagen, as discussed earlier in Chapter 5. These changes result from the damage and repair processes that occur with repeated exposure. Indices of fibrosis also were found to persist after exposure in some of the studies. Collectively,

these findings provide a potential pathophysiologic basis for the changes in airway function observed in children in longitudinal studies. Seasonal ambient patterns of exposure may be of greater concern than continuous daily exposure. In the classical study by Tyler et al. (1988), young monkeys with seasonal exposure to O₃, but not those with daily exposure, experienced increases in total lung collagen content, chest wall compliance, and inspiratory capacity, suggesting a delay in lung maturation in seasonally-exposed animals.

Controlled human exposure studies clearly demonstrated acute inflammation in the lung at ambient exposure levels. Epidemiologic studies could examine whether repeated exposures over multiple episode periods and/or multiple years would lead to persistent inflammation and result in damage to the human lung, especially in the small, terminal bronchiolar regions where vulnerability is greatest. However, the challenges to addressing these issues in epidemiologic studies are formidable, and as a result there exists relatively limited literature in this area. Long-term O₃ concentrations tend to be correlated with long-term concentrations of other pollutants, making specific attribution difficult. Subtle pulmonary effects require health outcome measures that are sensitive, and must usually be directly collected from individual human subjects, rather than from administrative data bases. Although these factors make chronic studies difficult and expensive to conduct, efforts must be made to design studies with adequate power to examine the hypothesis being tested. Epidemiologic studies have the potential to provide important new insights on the links between chronic exposure to O₃ and the occurrence of human health effects.

This section reviews studies published from 1996 onward in which health effects were tested in relation to extended O₃ exposures ranging from several weeks to many years (Table AX7-6 in the Annex 7, Section AX7.1). The available literature falls into four general categories: (1) studies examining seasonal changes in lung function as related to O₃ exposures in peak season; (2) studies addressing smaller increases in lung function during childhood or decline of lung function beyond childhood in relation to long-term O₃ exposures; (3) studies addressing respiratory inflammation in high versus low exposure groups or time periods; and (4) studies addressing longitudinal and cross-sectional associations between long-term O₃ exposures and asthma development and prevalence.

7.5.3 Seasonal Ozone Effects on Lung Function

While it has been well-documented in both chamber and field studies that daily, multihour exposures to O₃ result in transient declines in lung function, much less is known about the effects of repeated exposures to O₃ over extended periods on lung function. Several new studies reported during the past decade have examined lung function changes over seasonal time periods with differing levels of O₃ exposures (Frischer et al., 1999; Horak et al., 2002a,b; Ihorst et al., 2004; Kinney and Lippmann, 2000; Kopp et al., 2000). The seasonal effects of O₃ are examined first in this section. The next section then discusses effects of O₃ exposures over years, as opposed to over seasons, in addition to multiyear analyses of seasonal studies.

In a large Austrian study, Frischer et al. (1999) carried out repeated lung function measurements in 1,150 school children (mean age 7.8 years) from nine towns that differed in mean O₃ levels. Lung function was measured in the spring and fall over a 3-year period from 1994 to 1996, yielding six measurements per child. Mean summertime 24-h avg O₃ concentrations ranged from 32.4 to 37.3 ppb during the three summers. Growth-related increases in lung function over the summer season were reduced in relation to seasonal mean O₃ levels. Ozone was associated with a change of -156.6 mL (95% CI: -209.5, -103.7) (central estimate: $-0.029 \text{ mL/day/ppb} \times 90 \text{ days/year} \times 3 \text{ years} \times 20 \text{ ppb}$) in FEV₁ increase for each 20 ppb increase in mean 24-h avg O₃ concentrations over the three summers and -129.6 mL (95% CI: -193.1, -66.1) over the three winters. When analyses were restricted to children who had spent the whole summer period in their community, the changes were greater, with an O₃-related -183.6 mL (95% CI: -278.9, -88.3) change in FEV₁ increase over three summers. Other pollutants (PM₁₀, SO₂, and NO₂) had less consistent associations with changes in lung function. Horak et al. (2002a,b) extended the study of Frischer et al. (1999) with an additional year of data and found that seasonal mean O₃ was associated with a negative effect on increases in lung function, confirming results from the previous 3-year study. In an editorial, Tager (1999) stated that the Frischer et al. (1999) data provided the first prospective evidence of an association between exposure to ambient air pollution and alterations in lung function in children. Tager further noted that the prospective study design represented a substantial improvement over data derived from cross-sectional studies and should be emulated. However, Tager also cautioned that it was difficult to attribute the reported effects to O₃ alone independently of copollutants.

Kopp et al. (2000) reported smaller increases in lung function among a cohort of 797 children from 10 communities in Austria and southwestern Germany who were exposed to high levels of ambient O₃ (mean O₃ concentration of 44 to 52 ppb from April to October 1994). Children residing in lower O₃ (24 to 33 ppb) areas experienced a 43 mL increase in FEV₁, whereas those in high O₃ areas only manifested a 16 mL increase during the summer of 1994. Similar results were found in data for the summer of 1995. In another study of 15 communities in Austria and southwestern Germany, Ihorst et al. (2004) examined 2,153 children with a median age of 7.6 years and reported summer pulmonary function results which indicated that significantly lower FVC and FEV₁ increases were associated with higher O₃ exposures in the summer, but not in the winter. Semi-annual mean O₃ concentrations ranged from 22 to 54 ppb during the summer and 4 to 36 ppb during the winter.

In a pilot study (Kinney and Lippmann, 2000), 72 nonsmoking adults (mean age 20 years) from the second year class of students at the U.S. Military Academy in West Point, NY provided two lung function measurements, one before and one after a 5-week long summer training program at four locations. There was a greater decline in FEV₁ among students at the Fort Dix location (78 mL) as compared to students at the other locations (31 mL). Ozone levels at Fort Dix averaged 71 ppb (mean of daily 1-h max O₃) over the summer training period versus mean values of 55 to 62 ppb at the other three locations. In addition to the higher mean O₃ level, Fort Dix had greater peak O₃ values (23 hours >100 ppb) compared to the other locations (1 hour >100 ppb). Ambient levels of other pollutants, PM₁₀ and SO₂, were relatively low during the study and did not vary across the four sites. Although the conclusions that can be drawn are limited by the small study size, the results appear to be consistent with a seasonal decline in lung function that may be due, in part, to O₃ exposures. An exploratory observation from this study was that there appeared to be a larger decline for those subjects who completed their post-summer lung function measurements within two weeks after returning from training compared to those measured 3 to 4 weeks after training, results consistent with some degree of rebound of function following the summer exposure period.

Collectively, the above studies indicate that seasonal O₃ exposure is associated with smaller increases in lung function in children. The study by Kinney and Lippman (2000) also provides some limited evidence suggesting that seasonal O₃ also may affect lung function in young adults.

7.5.4 Chronic Ozone Exposure Effects on Lung Function and Respiratory Symptoms

Lung capacity grows during childhood and adolescence as body size increases, reaches a maximum during the 20s, and then begins to decline steadily and progressively with age. There has long been concern that long-term exposure to air pollution might lead to slower growth in lung capacity, diminished maximally attained capacity, and/or more rapid decline in capacity with age. The concern arises by analogy with cigarette smoking, where it is well-documented that lung function declines more rapidly with age in a dose-dependent manner among adults who smoke cigarettes. Adults who stop smoking return to a normal rate of decline in capacity, although there is no evidence that they regain the capacity previously lost due to smoking (Burchfiel et al., 1995). Because O₃ is a strong respiratory irritant and is associated with acute lung function declines as well as inflammation and re-structuring of the respiratory airways, it seems plausible that there might be a negative impact of long-term O₃ exposures on lung function. Exposures that negatively affect increases in lung function during childhood, in particular, might have greater long-term risks. Thus, studies of effects on the rates of increases in lung function in children are especially important.

Several studies published over the past decade have examined the relationship between long-term O₃ exposure and lung function. The most extensive and robust study of respiratory effects in relation to long-term air pollution exposures among children in the United States is the Children's Health Study carried out in 12 communities of southern California starting in 1993 (Avol et al., 2001; Gauderman et al., 2000, 2002, 2004a,b; Peters et al., 1999a,b). The first cohort included children from the fourth, seventh, and tenth grades. A total of 3,676 students completed questionnaires regarding their lifetime residential histories, historic and current health status, residential characteristics, and physical activity. Among those students, 3,293 also performed pulmonary function tests at the time of enrollment. Peters et al. (1999a) examined the relationship between long-term (1986 to 1990) O₃ exposures and self-reports of respiratory symptoms and asthma in a cross-sectional analysis. Mean 1-h max O₃ levels from 1986 to 1990 ranged from 30.2 ppb to 109.2 ppb across the 12 communities. For outcomes of current asthma, bronchitis, cough, and wheeze, the reported odds ratios were 0.95 (95% CI: 0.70,1.29), 1.14 (95% CI: 0.84, 1.55), 0.98 (95% CI: 0.82, 1.17), and 1.08 (95% CI: 0.87, 1.35), respectively, per 40 ppb increase in 1-h max O₃. In another cross-sectional analysis of the relationship

between lung function at baseline and levels of air pollution in the community, there was evidence that annual mean O₃ levels were associated with decreased FVC, FEV₁, PEF, and FEF₂₅₋₇₅ (the latter two being statistically significant) among females but not males (Peters et al., 1999b).

Avol et al. (2001) examined 110 children from the first cohort who had moved from the participating communities in southern California to other communities to determine whether changes in air quality caused by relocation were associated with changes in annual increases in lung function. Negative effects of O₃ were observed for all spirometric parameters except FEV₁, but the associations were not as strong as those observed for PM₁₀.

To examine the association between long-term exposure to air pollution and changes in lung function, fourth graders (n = 1,759) from the first cohort were followed over eight years (Gauderman et al., 2000, 2004a,b). During the follow-up period from 1993 to 2001, there was an attrition of approximately 10% of subjects per year, resulting in 747 subjects being tested in 2001. The longitudinal pulmonary function data for each child was related to average air pollution levels in each study community using a multistage regression approach. During the monitoring period, there was substantial variation in the average levels of various air pollutants across the 12 communities, but relatively little year-to-year variation within each community. However, O₃ levels did not vary widely across the communities in comparison to the other pollutants. The mean annual average of 8-h avg O₃ concentrations ranged from approximately 30 ppb in Long Beach to 65 ppb in Lake Arrowhead from 1994 to 2000. Average levels of O₃ were not significantly correlated across communities with any other study pollutant. Analyses indicated that there was no evidence that either 8-h avg O₃ (10 a.m. to 6 p.m.) or 24-h avg O₃ was associated with any measure of lung function growth over a 4-year (age 10 to 14 years; Gauderman et al., 2000) or 8-year (age 10 to 18 years; Gauderman et al., 2004) period. However, most of the other pollutants examined (including PM_{2.5}, NO₂, acid vapor, and elemental carbon) were found to be significantly associated with reduced growth in lung function. Also, a clinically low FEV₁ (i.e., FEV₁ <80% of predicted value) at age 18 years was reported to be correlated with all pollutants examined except for O₃ (Figure 7-26).

The Children's Health Study enrolled a second cohort of fourth graders (n = 1,678) in 1996 (Gauderman et al., 2002). While the strongest associations with negative lung function growth were observed with acid vapors in this cohort, children from communities with higher 4-year

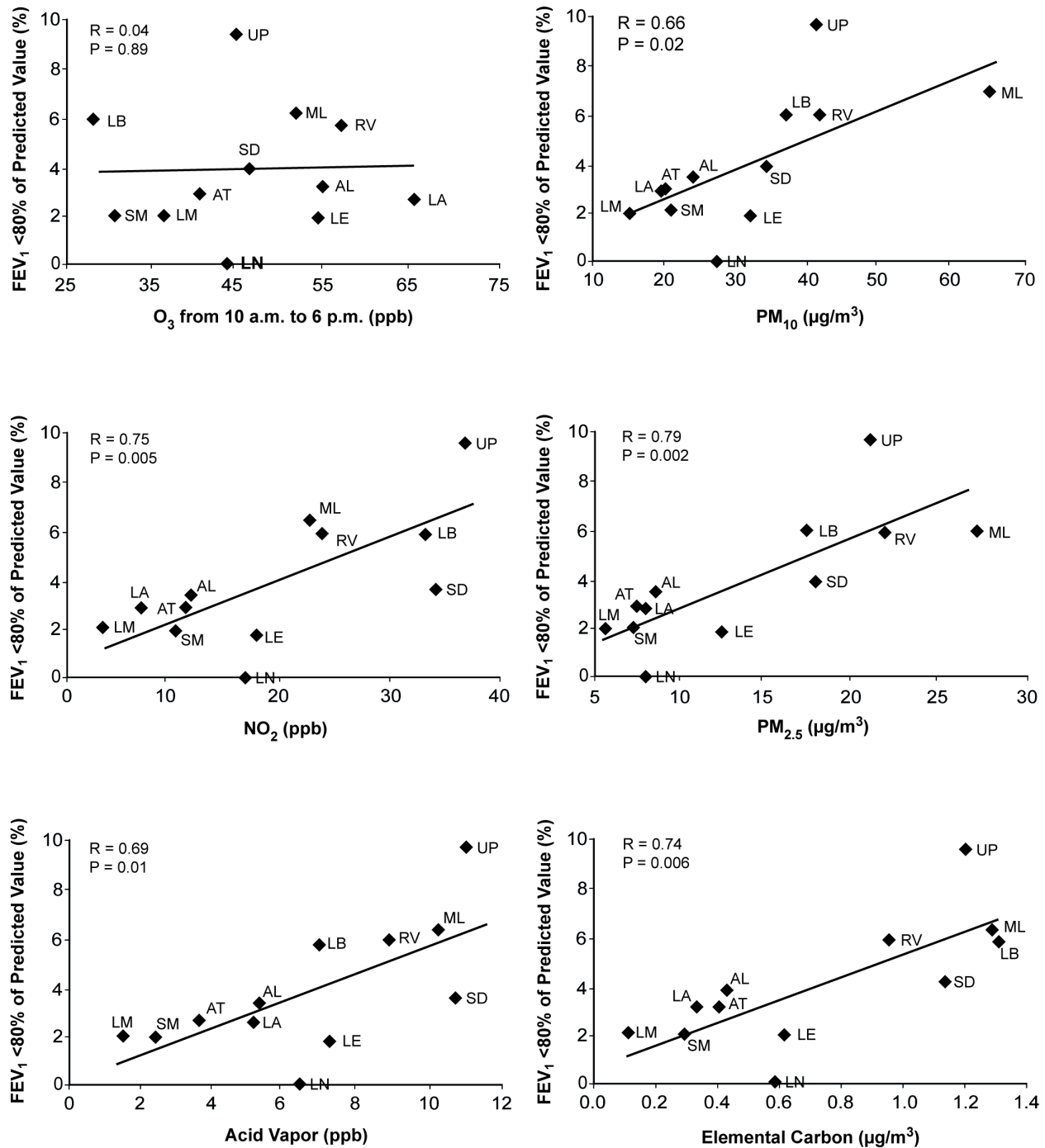


Figure 7-26. Proportion of 18-year-olds with a FEV₁ below 80% of the predicted value plotted against the average levels of pollutants from 1994 through 2000 in the 12 southern California communities of the Children’s Health Study.

AL = Alpine; AT = Atascadero; LA = Lake Arrowhead; LB = Long Beach; LE = Lake Elsinore; LM = Lompoc; LN = Lancaster; ML = Mira Loma; RV = Riverside; SD = San Dimas; SM = Santa Maria; UP = Upland

Source: Derived from Gauderman et al. (2004).

average O₃ levels also experienced smaller increases in various lung function parameters. The strongest effect of O₃ was on PEF, with children from the least-polluted community having a 1.21% (95% CI: 0.36, 2.06) greater increase in PEF compared to those from the most-polluted communities. In two-pollutant models, only 8-h avg O₃ and NO₂ were significant joint predictors of FEV₁ and maximal midexpiratory flow (MMEF). The range of mean annual average 8-h avg O₃ levels across the 12 communities was similar to that for the first cohort, 27 to 67 ppb.

In both cohorts of fourth graders, stratified analyses by time spent outdoors indicated a stronger association between long-term O₃ exposure and smaller increases in lung function during the 4-year follow-up in children who spent more time outdoors, as shown in Table 7-3. Jedrychowski et al. (2001) had reported a link between repeated respiratory symptoms and smaller lung function increases. Gauderman et al. (2002), therefore, suggested that the observation of reduced increases in lung function with increasing annual average air pollution might be a consequence of repeated acute respiratory events after short-term increases in pollutant levels. The findings of larger deficits in children who spend more time outdoors in the afternoon adds some support for this possibility.

Although results from the second cohort of children are supportive of an association, the definitive 8-year follow-up analysis of the first cohort (Gauderman et al., 2004a) provides little evidence that long-term exposure to ambient O₃ at current levels is associated with significant deficits in the growth rate of lung function in children. It should be noted, however, that O₃ exposure prior to the study period is unknown. In addition, there was only about a 2- to 2.5-fold difference in O₃ concentrations from the least to most polluted communities (mean annual average of 8-h avg O₃ ranged from 30 to 65 ppb), versus the ranges observed for the other pollutants (which had 4- to 8-fold differences in concentrations). Finally, the results from the stratified analyses by time spent outdoors indicate the potential for substantial misclassification of long-term O₃ exposure. The Children's Health Study, like most long-term epidemiologic studies, estimated O₃ exposure using centrally-located ambient monitors. Gonzales et al. (2003) and Künzli et al. (1997) evaluated the use of retrospective questionnaires to reconstruct past time-activity and location pattern information. Both studies found that questionnaires or activity diaries might improve the assessment of chronic exposure in epidemiologic studies.

Table 7-3. Difference in Annual Percent Increases in Lung Function from the Least to the Most Polluted Community in the Children's Health Study by Time Spent Outdoors^a

Lung Function Parameter	Cohort ^b	More Time Outdoors ^c	Less Time Outdoors ^c
		% Change (95% CI) ^d	% Change (95% CI) ^d
FVC	Cohort 1	-0.02% (-0.57, 0.54)	-0.04% (-0.45, 0.37)
	Cohort 2	-0.57% (-1.03, -0.09)	-0.06% (-0.76, 0.66)
FEV ₁	Cohort 1	-0.25% (-1.18, 0.68)	-0.05% (-0.58, 0.49)
	Cohort 2	-0.68% (-1.36, 0.00)	-0.29% (-1.02, 0.46)
MMEF	Cohort 1	-0.55% (-2.08, 1.01)	0.23% (0.89, 1.36)
	Cohort 2	-0.48% (-1.71, 0.78)	-0.80% (-2.07, 0.50)
PEF	Cohort 1	-0.77% (-2.03, 0.52)	0.25% (-0.65, 1.16)
	Cohort 2	-1.33% (-2.43, -0.24)	-0.71% (-1.71, 0.30)

^a Results are derived from Gauderman et al. (2002).

^b Cohort 1 includes children enrolled in 1993 as 4th graders and followed through 1997 (n = 1,457).

Cohort 2 includes children enrolled in 1996 as 4th graders and followed through 2000 (n = 1,678).

^c More or less time outdoors is based on reported time spent outdoors during weekday afternoons.

Subjects were split into the two groups on the basis of the median reported time outdoors within each cohort.

^d Percent change in lung function is per 30 ppb increase in 8-h avg O₃ (10 a.m.-6 p.m.).

In a study of 15 communities in Austria and Germany, Ihorst et al. (2004) found no associations between increases in lung function and mean summer O₃ levels for FVC and FEV₁ over a 3.5-year period, in contrast to the significant seasonal effects discussed in the earlier section above. Unlike the reduced increases in lung function parameters over the first two summers among children in high O₃ areas noted above, a greater increase was observed by Ihorst et al. during the third summer and no difference was observed during the fourth summer. The mean O₃ concentration during the first two summers across the 15 communities was similar to that during the last two summers, 36.6 ppb and 35.1 ppb, respectively. The authors concluded that medium-term effects on school children lung function were possibly present but were not detected over a 3- to 5-year period due to partial reversibility. A study by Frischer et al. (1999) showed results similar to the Ihorst et al. (2004) study. Although O₃ was related to smaller

increases in lung function when three years of data were analyzed collectively, the magnitude and direction of the effect changed throughout the years. Ozone was associated with a change of -34.0 mL (95% CI: $-58.7, -9.3$) in FEV₁ increase in the first year, compared to $+7.3$ mL (95% CI: $-20.8, 35.6$) in the third year for each 20 ppb increase in mean 24-h avg O₃. Mean summertime O₃ concentrations decreased slightly over the three years; the mean levels being 37.3 ppb, 35.4 ppb, and 32.4 ppb during the first, second, and third years, respectively.

A study by Gong et al. (1998b) examined lung function changes in 164 nonsmoking adults (mean age 45 years) from a high O₃ community in southern California, recruited from a cohort of 208 who had been tested on two previous occasions. In the earlier analysis by Detels et al. (1987), a significant decline in lung function was observed from 1977/1978 to 1982/1983. In contrast, Gong et al. (1998b) observed a slight increase in FVC and FEV₁ from 1982/1983 to 1986/1987. The mean annual average 1-h max O₃ level was 110 ppb from 1972 to 1982. For 1983 to 1989, the annual average 1-h max O₃ levels ranged from 103 to 134 ppb. A consistent decline in FEV₁/FVC ratio was observed at all three time points ($p < 0.0001$). Among the 45 subjects who further participated in the controlled exposure study (0.40 ppm O₃ over 2 hours with intermittent exercise), declines in FEV₁ and FVC were associated with acute O₃ exposure. However, the acute changes in lung function were not associated with long-term changes in lung function over a decade.

Evidence for a relationship between long-term O₃ exposures and decrements in maximally attained lung function was observed in a nationwide cohort of 520 first year students at Yale College in New Haven, CT (Galizia and Kinney 1999; Kinney et al., 1998). Each student performed one lung function test in the spring of their first year at college. Ozone exposures were estimated by linking 10-year average summer season (June to August) 1-h max O₃ levels at the nearest monitoring station to the residential locations reported each year from birth to the time of measurement. The mean 10-year average 1-h max O₃ was 61.2 ppb (range 13-185). Students who had lived four or more years in areas with long-term mean O₃ levels above 80 ppb had significantly lower FEV₁ (-3.07% [95% CI: $-0.22, -5.92$]) and FEF₂₅₋₇₅ (-8.11% [95% CI: $-2.32, -13.90$]) compared to their classmates with lower long-term O₃ exposures. Stratification by gender indicated much larger effect estimates for males than for females, which might reflect higher outdoor activity levels and corresponding higher O₃ exposure during childhood. Ozone was the only air pollutant examined in this study. The authors noted, therefore, that it was not

possible to determine whether, and to what extent, the observed effects are due to O₃ alone or O₃ in combination with unmeasured copollutants.

A similar study of 130 first year college freshmen at the University of California at Berkeley also reported significant effects of O₃ on lung function (Künzli et al., 1997; Tager et al., 1998). Enrollment was limited to students from either the San Francisco or Los Angeles, CA metropolitan areas. Lifetime monthly average 8-h avg O₃ (10 a.m.-6 p.m.) concentrations ranged from 16 to 33 ppb for San Francisco residents and 25 to 74 ppb for Los Angeles residents. After controlling for city of origin, long-term O₃ exposures were found to be associated with declines in FEF₂₅₋₇₅ and FEF₇₅ (forced expiratory flow after 75% of FVC has been exhaled). No effects were seen for PM₁₀ and NO₂. Künzli et al. (1997) noted that significant changes in these mid- and end-expiratory flow measures could be considered early indicators for pathologic changes that might ultimately progress to COPD, as evidenced by animal studies that showed that the primary site of O₃ injury in the lung was the centriacinar region (see Chapter 5). This study was repeated in 2000 to 2002, with 255 freshmen who were life-long residents of the Los Angeles or San Francisco areas (Tager et al., 2005). Lifetime monthly average 8-h avg O₃ levels ranged from 18 to 65 ppb for the entire cohort, with medians of 36 ppb for the men and 33 ppb for the women. In contrast to results from the first cohort, associations between long-term O₃ exposure and declines in FEF₂₅₋₇₅ and FEF₇₅ were not observed in the second cohort of freshmen. However, when the analysis was stratified by gender and an interaction term for intrinsic airway size (FEF₂₅₋₇₅/FVC ratio) was included in the model, lifetime exposure to O₃, as well as PM₁₀ and NO₂, was found to be associated with decreased FEF₂₅₋₇₅ and FEF₇₅ for both men and women. The adverse impact of O₃ exposure decreased with increasing intrinsic airway size. Also, in multipollutant models including all three pollutants, meaningful changes were not observed in the O₃ effect estimates, but the PM₁₀ and NO₂ estimates were reduced substantially.

The cross-sectional studies in college freshmen provide suggestive evidence that young adults who have grown up in high O₃ communities may have reduced lung function compared to those from low O₃ communities. However, attributing the effects specifically to O₃ is difficult, given that the effects of coexposures to other ambient air pollutants have not necessarily been adequately addressed in these cross-sectional long-term exposure studies. Thus, results of the longitudinal southern California Children's Health Study, as well as those from the European

studies, provide little evidence for impacts of long-term, relatively low-level O₃ exposures on lung function development in children. On the other hand, these studies appear to indicate likely strong effects on lung function growth for long-term exposures to PM_{2.5} and/or certain other copollutants.

7.5.5 Chronic Ozone Exposure and Respiratory Inflammation

As noted in Chapter 6, controlled human exposure chamber studies have demonstrated that brief (2 to 6.6 hours) exposures to O₃ while exercising result in inflammation in the lung, including the alveolar region where gas exchange takes place. This acute exposure effect is potentially important for effects of chronic exposure, because repeated inflammation can result in the release of substances from inflammatory cells that can damage the sensitive cells lining the lung. Over extended periods, repeated insults of this kind can lead to permanent damage to and restructuring of the small airways and alveoli. In addition, since inflammation is a fundamental feature of asthma, there is concern that O₃-induced inflammation can exacerbate existing asthma or perhaps promote the development of asthma among genetically pre-disposed individuals. Several studies are discussed next, examining different outcomes related to inflammation.

In a study by Kinney et al. (1996b), bronchoalveolar lavage fluids were collected in the summer and winter from a group of 19 adult joggers living and working on an island in New York City harbor. The mean 1-h max O₃ concentrations for a 3-month period were 58 ppb (maximum 110) in the summer and 32 ppb (maximum 64) in the winter. PM₁₀ and NO₂ concentrations were similar across the two seasons. There was little evidence for acute inflammation in bronchoalveolar lavage fluids collected during the summer as compared to that collected from the same subjects in the winter. However, there was evidence of enhanced cell damage, as measured by lactate dehydrogenase, in the summer lavage fluids. These results indicate that acute inflammatory responses may diminish with repeated exposures over the course of a summer (as also demonstrated by multiday chamber exposures; see Chapter 6, Section 6.9), but cell damage may be ongoing.

In a cross-sectional study by Frischer et al. (2001), urinary eosinophil protein was analyzed as a marker of eosinophil activation in 877 school children living in nine Austrian communities with varying ranges of O₃ exposure levels. The results indicated that O₃ exposure was

significantly associated with eosinophil inflammation after adjusting for gender, site, and atopy. The mean 30-day average O₃ concentration before sample collection was 31.6 ppb (5th % to 95th%: 11.8 to 51.5).

Pollution effects in the nose can be viewed as a potential surrogate measure for effects that may occur in the lungs, though doses to nasal tissues are usually higher for a given pollutant concentration. As discussed in Chapter 5, morphological effects of O₃ on the upper respiratory tract can indicate quantitative changes in the nasal transitional respiratory epithelium. The persistent nature of the O₃-induced mucous cell metaplasia in rats, as discussed in Chapter 5, suggests that O₃ exposure may have the potential to induce similar long-lasting alterations in the airways of humans. In a cross-sectional cohort study by Calderón-Garcidueñas et al. (1995), nasal lavage samples collected from children living in Mexico City (n = 38) were compared to those from children living in a clean coastal town (n = 28). In Mexico City, the 1-h avg O₃ concentrations exceeded 120 ppb for 4.4 h/day, while O₃ levels were not detectable in the coastal town. Mexico City children were examined four times within a 1-month period. Nasal cytologies revealed that all Mexico City children had abnormal nasal mucosae, including mucosal atrophy, marked decreases in the numbers of ciliated-type cells and goblet cells, and squamous metaplasia. Significantly higher nasal polymorphonuclear leukocyte counts (p < 0.001) and expression of human complement receptor type 3 (CD11b) (p < 0.001) were also observed in Mexico City children compared to the control children. Analyses using repeated measurements indicated that the inflammatory response did not appear to correlate with the previous day's O₃ concentration, suggesting that there might be a competing inflammatory mechanism at the bronchoalveolar level with structural injury following acute exposure.

Limited epidemiologic research has been conducted relating long-term O₃ exposure to inflammation. In the Mexico City study (Calderon-Garciduenas et al., 1995), specific attribution of the nasal abnormalities to long-term O₃ exposure is difficult, given the complex pollutant mixture present in the ambient air. However, the inflammatory changes such as increased eosinophil levels observed in the Austrian study (Frischer et al., 2001) would be consistent with known effects of O₃.

7.5.6 Risk of Asthma Development

Recent longitudinal cohort studies have reported associations between long-term O₃ exposures and the onset of asthma (McConnell et al., 2002; McDonnell et al., 1999). Significant associations between long-term O₃ exposure and new cases of asthma among adult males were observed in a cohort of nonsmoking adults in California (Greer et al., 1993; McDonnell et al., 1999). The Adventist Health and Smog (AHSMOG) study cohort of 3,914 (age 27 to 87 years, 36% male) was drawn from nonsmoking, non-Hispanic white California Seventh Day Adventists, who were surveyed in 1977, 1987, and 1992. To be eligible, subjects had to have lived 10 or more years within 5 miles of their current residence in 1977. Residences from 1977 onward were followed and linked in time and space to interpolate concentrations of O₃, PM₁₀, SO₄²⁻, SO₂, and NO₂. New asthma cases were defined as self-reported doctor-diagnosed asthma at either the 1987 or 1992 follow-up questionnaire among those who had not reported having asthma upon enrollment in 1977. During the 10-year follow-up (1977 to 1987), the incidence of new asthma was 2.1% for males and 2.2% for females (Greer et al., 1993). Ozone concentration data were not provided. A relative risk of 3.12 (95% CI: 1.16, 5.85) per 10 ppb increase in annual mean O₃ (exposure metric not stated) was observed in males, compared to a relative risk of 0.94 (95% CI: 0.65, 1.34) in females. In the 15-year follow-up study (1977-1992), 3.2% of the eligible males and a slightly greater 4.3% of the eligible females developed adult asthma (McDonnell et al., 1999). The mean 20-year average for 8-h avg O₃ (9 a.m. to 5 p.m.) was 46.5 ppb (SD 15.3) from 1973 to 1992. For males, the relative risk of developing asthma was 2.27 (95% CI: 1.03, 4.87) per 30 ppb increase in 8-h avg O₃. Once again, there was no evidence of an association between O₃ and new-onset asthma in females (relative risk of 0.85 [95% CI: 0.55, 1.29]). The lack of an association does not necessarily indicate no effect of O₃ on the development of asthma among females. For example, differences between females and males in time-activity patterns may influence relative exposures to O₃, leading to greater misclassification of exposure in females. None of the other pollutants (PM₁₀, SO₄²⁻, SO₂, and NO₂) were associated with development of asthma in either males or females. Adjusting for copollutants did not diminish the association between O₃ and asthma incidence for males. The consistency of the results in the two studies with different follow-up times, as well as the independent and robust association between annual mean O₃ concentrations and asthma incidence, provide supportive evidence that long-term O₃ exposure may be associated with the development of

asthma in adult males. However, because the AHSMOG cohort was drawn from a narrow subject definition, the representativeness of this cohort to the general U.S. population may be limited.

A similar study of incident asthma cases in relation to air pollutants (O₃, PM, NO₂, and acid vapors) among children was carried out in the Children's Health Study (McConnell et al., 2002). A total of 3,535 initially nonasthmatic children (ages 9 to 16 years at enrollment) were followed for up to 5 years to identify new-onset asthma cases. Communities were stratified by pollution levels, with six high-O₃ communities (mean 1-h max O₃ of 75.4 ppb [SD 6.8] over four years) and six low-O₃ communities (mean 50.1 ppb [SD 11.0]). Ozone concentrations were not strongly correlated with the other pollutants. New diagnoses of asthma were reported for 265 children during the follow-up period. Asthma risk was not higher for residents of the six high-O₃ communities versus residents of the six low-O₃ communities. However, within the high-O₃ communities, asthma risk was 3.3 (95% CI: 1.9, 5.8) times greater for children who played three or more sports as compared with children who played no sports. None of the children who lived in high-O₃ communities and played three or more sports had a family history of asthma. This association was absent in the low-O₃ communities (relative risk of 0.8 [95% CI: 0.4, 1.6]). No associations with asthma were seen for PM₁₀, PM_{2.5}, NO₂, or inorganic acid vapors. These results suggest effect modification by physical activity of the impacts of O₃ on asthma risk. The overall observed pattern of effects of sports participation on asthma risk was robust to adjustment for socioeconomic status, history of allergy, family history of asthma, insurance, maternal smoking, and body mass index. Playing sports may index greater outdoor activity when O₃ levels are higher and an increased ventilation rate, which may lead to increased O₃ exposure. It should be noted, however, that these findings were based on a small number of new asthma cases (n = 29 among children who played three or more sports) and were not based on a priori hypotheses. Future replication of these findings in other cohorts would lend greater weight to a causal interpretation.

Recent cross-sectional surveys have detected no associations between long-term O₃ exposures and asthma prevalence, asthma-related symptoms, or allergy to common aeroallergens in children after controlling for covariates (Charpin et al., 1999; Kuo et al., 2002; Ramadour et al., 2000). It should be noted that O₃ levels were quite low in all cases, with a range of 16 to 27 ppb for 8-h max O₃. The longitudinal study design, which observes new onset of asthma

prospectively, provides stronger evidence in relation to the question of O₃ effects on asthma development.

7.5.7 Respiratory Effects of Chronic Ozone Exposure on Susceptible Populations

Studies on the effect of long-term O₃ exposure on respiratory health have focused mostly on children, a potentially susceptible population. Seasonal O₃ exposure was associated with smaller increases in lung function and respiratory inflammation in children. Other studies have investigated additional groups of potentially susceptible individuals. In the Children's Health Study, McConnell et al. (1999) examined the association between O₃ levels and the prevalence of chronic lower respiratory tract symptoms in 3,676 southern California children with asthma. In this cross-sectional study, bronchitis, phlegm, and cough were not associated with annual mean O₃ concentrations in children with asthma or wheeze. All other pollutants examined (PM₁₀, PM_{2.5}, NO₂, and gaseous acid) were associated with an increase in phlegm but not cough. The mean annual average 1-h max O₃ concentration was 65.6 ppb (range 35.5 to 97.5) across the 12 communities.

In another Children's Health Study analysis, McConnell et al. (2003) evaluated relationships between air pollutants and bronchitic symptoms among 475 children with asthma. The mean 4-year average 8-h avg O₃ (10 a.m.-6 p.m.) concentration was 47.2 ppb (range 28.3 to 65.8) across the 12 communities. For a 1 ppb increase in 8-h avg O₃ averaged over 4 years, the between-community odds ratio was 0.99 (95% CI: 0.98, 1.01) compared to the within-community odds ratio of 1.06 (95% CI: 1.00, 1.12). The authors commented that if the larger within-community effect estimates were correct, then other cross-sectional (between-community) studies might have underestimated the true effect of air pollution on bronchitic symptoms in children. These differences might be attributable to confounding by poorly measured or unmeasured risk factors that vary between communities. PM_{2.5}, NO₂, and organic carbon also were associated with bronchitic symptoms. In two-pollutant models, the within-community effect estimates for O₃ were markedly reduced and no longer significant in some cases.

One recent study examined a susceptible group not examined before. Goss et al. (2004) investigated the effect of O₃ on pulmonary exacerbations and lung function in individuals over

the age of 6 years with cystic fibrosis (n = 11,484). The study included patients enrolled in the Cystic Fibrosis Foundation National Patient Registry, which contains demographic and clinical data collected annually at accredited centers for cystic fibrosis. For 1999 through 2000, the annual mean O₃ concentration, calculated from 1-h averages from 616 monitors in the U.S. EPA Aerometric Information Retrieval System (AIRS), was 51.0 ppb (SD 7.3). Exposure was assessed by linking air pollution values from AIRS with the patient's home ZIP code. No clear association was found between annual mean O₃ and lung function parameters. However, a 10 ppb increase in annual mean O₃ was associated with a 10% (95% CI: 3, 17) increase in the odds of two or more pulmonary exacerbations. Significant excess odds of pulmonary exacerbations also were observed with increased annual mean PM₁₀ and PM_{2.5} concentrations. The O₃ effect was robust to adjustment for PM₁₀ and PM_{2.5}, 8% (95% CI: 1, 15) increase in odds of two or more pulmonary exacerbations per 10 ppb increase in annual mean O₃.

In summary, few studies have identified and investigated potentially susceptible populations. Results from the Children's Health Study do not provide strong evidence that asthmatic children are particularly susceptible to long-term exposure to O₃. Ozone exposure was, however, shown to be associated with adverse respiratory health responses in individuals with cystic fibrosis.

7.5.8 Effects of Chronic Ozone Exposure on Mortality and Cancer Incidence

There is inconsistent and inconclusive evidence for a relationship between long-term O₃ exposure and increased mortality and cancer risk (see Table AX7-7 in Annex 7, Section AX7.1). In the Harvard Six Cities Study (Dockery et al., 1993), adjusted mortality rate ratios were examined in relation to long-term mean O₃ concentrations in six cities: Topeka, KS; St. Louis, MO; Portage, WI; Harriman, TN; Steubenville, OH; and Watertown, MA. Mortality rate ratios were adjusted for age, sex, smoking, education, and body mass index. Mean O₃ concentrations from 1977 to 1985 ranged from 19.7 ppb in Watertown to 28.0 ppb in Portage. Long-term mean O₃ concentrations were not found to be associated with mortality in the six cities. However, strong associations were observed between long-term mean concentrations of PM₁₅, PM_{2.5}, and sulfate particles and mortality.

In a large prospective cohort study of approximately 500,000 U.S. adults, Pope et al. (2002) examined the effects of long-term exposure to air pollutants on mortality (American

Cancer Society, Cancer Prevention Study II). All cause, cardiopulmonary, lung cancer, and all other cause mortality risk estimates for long-term O₃ exposure are shown in Figure 7-27. No consistent positive associations were observed between O₃ and mortality. The mortality risk estimates were larger when analyses were restricted to the summer months (July to September) when O₃ levels were generally higher. The mean summertime 1-h max O₃ level from 1982 to 1998 was 59.7 ppb (SD 12.8). The O₃-mortality risk estimates were positive for all-cause and cardiopulmonary mortality, with a marginally significant estimate for cardiopulmonary mortality in the summer months; but a nonsignificant negative O₃ risk estimate was observed for lung cancer mortality. In contrast, consistent positive and significant effects of PM_{2.5} were observed for both lung cancer and cardiopulmonary mortality.

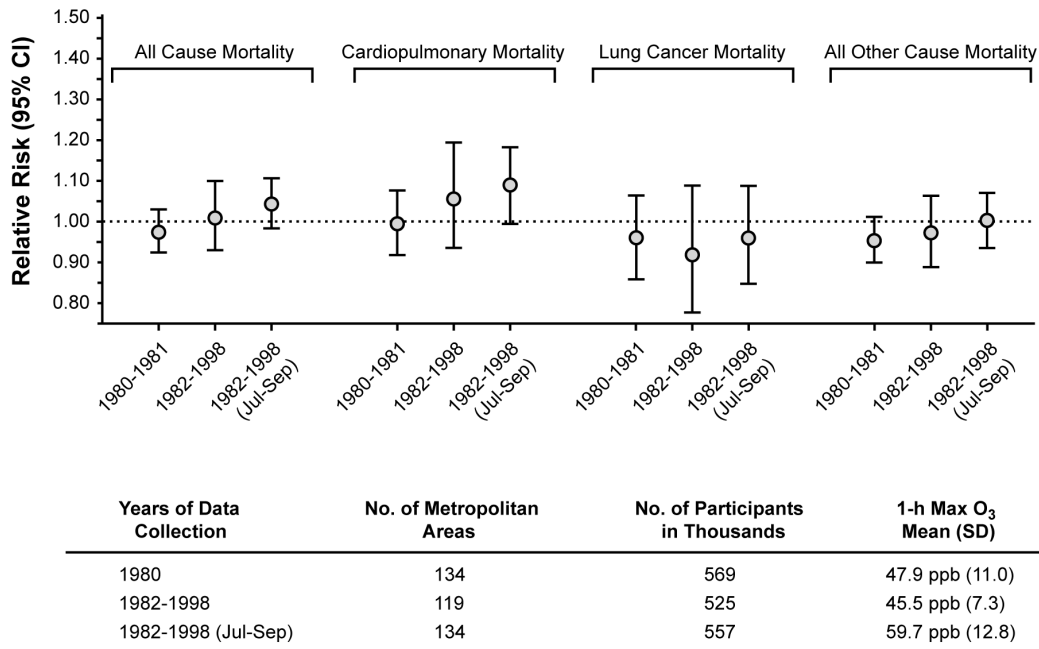


Figure 7-27. Adjusted O₃-mortality relative risk estimates (95% CI) by cause of mortality and time period of analysis per subject-weighted mean O₃ concentration in the Cancer Prevention Study II by the American Cancer Society.

Source: Derived from Pope et al. (2002).

Lipfert et al. (2000b, 2003) reported positive effects on all-cause mortality for peak O₃ exposures (95th percentile levels) in the U.S. Veterans Cohort study of approximately 50,000 middle-aged men recruited with a diagnosis of hypertension. The actual analysis involved smaller subcohorts based on exposure and mortality follow-up periods. Four separate exposure periods were associated with three mortality follow-up periods. The mean 95th percentile O₃ concentration ranged from 85 to 140 ppb in the four exposure periods. In a preliminary screening of regression results, Lipfert et al. (2000b) found a negative association for mean O₃ and a positive relationship for peak O₃; thus, peak O₃ was used in subsequent analyses. The mean of the peak values ranged from 85 to 140 ppb over the four exposure periods. For concurrent exposure periods, peak O₃ was positively associated with all-cause mortality, with a 9.4% (95% CI: 0.4, 18.4) excess risk per mean 95th percentile O₃ less estimated background level (not stated). When exposure periods preceding death were considered, no association between O₃ and mortality was observed (-0.2% [95% CI: -12.5, 12.1]). In a further analysis, Lipfert et al. (2003) reported the strongest positive association for concurrent exposure to peak O₃ for the subset of subjects with low diastolic blood pressure during the 1982 to 1988 period. Once again, the O₃ effect was diminished when exposure (1982 to 1988) preceded mortality (1989 to 1996).

A long-term prospective cohort study (AHSMOG) of 6,338 nonsmoking, non-Hispanic white individuals living in California examined the association between air pollutants and lung cancer incidence (Beeson et al., 1998). The mean monthly average 24-h avg O₃ level from 1973 to 1992 was 26.2 ppb (SD 7.7). Over the follow-up period of 1977 to 1992, 20 females (35% smokers, n = 7) and 16 males (37.5% smokers, n = 6) developed lung cancer. An association was observed between long-term O₃ exposure and increased incidence of lung cancer in males only. The relative risk for incident lung cancer among males was 3.56 (95% CI: 1.35, 9.42) for an interquartile range increase in hours per year (556 hours/year) when O₃ levels exceeded 100 ppb. A stronger association was seen for males who never smoked (4.48 [95% CI: 1.25, 16.04]) compared to those who smoked in the past (2.15 [95% CI: 0.42, 10.89]) (Beeson et al., 1998).

A related study by Abbey et al. (1999) examined the effects of long-term O₃ exposure on all-cause (n = 1,575), cardiopulmonary (n = 1,029), nonmalignant respiratory (n = 410), and lung cancer (n = 30) mortality in the same AHSMOG study population. A particular strength of this

study was the extensive effort devoted to assessing long-term air pollution exposures, including interpolation to residential and work locations from monitoring sites over time and space. No associations with long-term O₃ exposure were observed for all cause, cardiopulmonary, and nonmalignant respiratory mortality. However, effects of O₃ on lung cancer mortality paralleled the results of the previous study by Beeson and colleagues. That is, an association between lung cancer mortality and chronic O₃ exposure was observed in males only, with a relative risk of 4.19 (95% CI: 1.81, 9.69) for an interquartile range increase in hours per year (551 hours/year) when O₃ levels exceeded 100 ppb. The gender-specific O₃ effects may be partially attributable to differences in activity and time spent outdoors. The questionnaires indicated that males spent approximately twice as much time outdoors and performed more vigorous exercises outdoors, especially during the summer, compared to the females. However, the very small numbers of lung cancer deaths (n = 12 for males) raise concerns with regard to the precision of the effect estimate, as evidenced by the wide confidence intervals. The lack of an association of chronic O₃ exposure with other mortality outcomes, which had much larger samples sizes, also is of concern. A study by Pereira et al. (2005) provides limited evidence possibly suggestive of an association between O₃ and increase risk of cancer. Correlations between average air pollution data from 1981 to 1990 and cases of larynx and lung cancer in 1997 were assessed in communities of São Paulo, Brazil. Of all the pollutants examined (PM₁₀, NO₂, NO_x, SO₂, CO, and O₃), O₃ was best correlated with cases of larynx (r = 0.99, p = 0.01) and lung (r = 0.72, p = 0.28) cancer.

Only a few studies have examined the effect of chronic O₃ exposure on mortality outcomes and incidence of cancer. Consistent associations with long-term O₃ exposure have not been observed for all-cause and cardiopulmonary mortality. There is some very limited evidence suggestive of an association between chronic O₃ exposure and lung cancer incidence and mortality; however, uncertainty remains due to the small number of available studies and the very small numbers of lung cancer cases observed in most of the studies. Of particular note is the fact that the very large American Cancer Society study did not find any association between long-term O₃ concentrations and lung cancer. In addition, the weight of evidence from toxicologic studies does not support ambient O₃ as a pulmonary carcinogen in laboratory animal models (National Toxicology Program, 1994).

7.5.9 Effects of Ozone on Birth-Related Health Outcomes

In recent years, air pollution epidemiologic studies have examined impacts on birth-related endpoints, including intrauterine, perinatal, postneonatal, and infant deaths; premature births; intrauterine growth retardation; very low birth weight (weight <1500 grams) and low birth weight (weight <2500 grams); and birth defects. However, the majority of these studies did not examine the effect of O₃. In those limited studies that investigated O₃, no associations were found between O₃ and birth outcomes, with the possible exception of birth defects. The following is a synopsis of the literature on this topic.

Pereira et al. (1998) investigated the impacts of NO₂, SO₂, CO, O₃, and PM₁₀ on intrauterine mortality in São Paulo, Brazil during 1991 and 1992. Mean 1-h max O₃ was 34.8 ppb (SD 23.2). Intrauterine mortality was most significantly associated with NO₂, and less for SO₂ and CO. No association was found for O₃ or PM₁₀. Pereira et al. (1998) also sampled umbilical cord blood from healthy nonsmoking pregnant women soon after delivery in 1995 and analyzed the blood for carboxyhemoglobin. They found an association between carboxyhemoglobin levels and ambient CO after adjusting for passive smoking and weight, suggesting an impact of CO on the fetus. Also, Loomis et al. (1999) examined the association between several air pollutants (NO₂, SO₂, CO, O₃, PM_{2.5}) and infant mortality in Mexico City in the years 1993 to 1995. Mean 24-h avg O₃ was 44.1 ppb (SD 15.7). They reported that the strongest association was found for PM_{2.5} with a 3- to 5-day cumulative lag. They noted that infant mortality also was associated with NO₂ and O₃ at a 3- to 5-day lag, but not as consistently as PM_{2.5}. There have been other studies that examined possible air pollution effects on postneonatal mortality (Bobak and Leon, 1992; Bobak and Leon, 1999; Kaiser et al., 2004; Woodruff et al., 1997), but these studies did not examine O₃.

Ritz and Yu (1999) investigated the effects of ambient CO on low birth weight among children born in southern California between 1989 and 1993. They focused on CO, stating that “a biologic mechanism for fetal effects has been proposed for CO, but not for other air pollutants.” They found that exposure to higher levels of ambient CO during the last trimester was associated with an increased risk for low birth weight. Using available data, they also estimated last-trimester exposures for NO₂, O₃, and PM₁₀. The last trimester average O₃ level was 20.9 ppb (range 3.0 to 49.4). NO₂ and PM₁₀ were positively associated with CO, but O₃ was negatively associated with CO ($r = -0.65$). Ha et al. (2001) examined CO, NO₂, SO₂, O₃, and

TSP for their associations with low birth weight in Seoul, Korea during the years 1996 and 1997 (data were analyzed using Poisson GAM with default convergence parameters). They estimated first and third trimester exposures by averaging daily air pollution levels during the corresponding days for the registered births. The median 8-h avg O₃ levels were 22.4 ppb (IQR 13.6) for the first trimester and 23.3 ppb (IQR 16.1) for the third trimester. Ha et al. (2001) found that first trimester exposures of CO, NO₂, SO₂, and TSP were associated with increased risk of low birth weight, whereas O₃ was associated with a decreased risk. The opposite pattern was observed for third trimester exposures, with an increased risk of low birth weight found only for O₃. When exposures from both trimesters were examined simultaneously, the associations of first trimester exposures of CO, NO₂, SO₂, and TSP with increased risk of low birth weight remained; however, the association between third trimester O₃ exposure and low birth weight was diminished. Based on these results, Ha et al. concluded that exposures to CO, NO₂, SO₂, and TSP in the first trimester were risk factors for low birth weight. Note that neither of these studies examined the air pollution effect by season. Other studies that examined the associations between air pollution and low birth weight (Bobak, 2000; Bobak and Leon, 1999; Lin et al., 2001; Maisonet et al., 2001; Wang et al., 1997) found associations between low birth weight and either one or more of CO, SO₂, NO₂ and PM indices, but did not examine O₃ data. Collectively, these results do not provide any credible evidence of O₃ contributing to low birth weight.

Two studies by Dejmek et al. (1999, 2000) examined the relationship between ambient air pollution and risk of intrauterine growth retardation in a highly polluted area of Northern Bohemia (Teplice District). Both studies, however, focused on PM indices and did not analyze gaseous pollutants, such as O₃.

A few studies have examined the association between air pollution and premature births (Bobak, 2000; Ritz et al., 2000; Xu et al., 1995), but only Ritz et al. (2000) included O₃ in their analysis. Ritz et al. evaluated the effect of air pollution exposure during pregnancy on the occurrence of preterm birth in a cohort of 97,518 neonates born in southern California. CO, NO₂, SO₂, O₃, and PM₁₀ data measured at 17 air quality monitoring stations were used to estimate the average exposures for the first month and the last 6 weeks of pregnancy. The mean 8-h avg O₃ (9 a.m. to 5 p.m.) level was 36.9 ppb during both time periods. They found associations between PM₁₀ levels averaged for the last 6 weeks of pregnancy as well as PM₁₀ levels averaged over the first month of pregnancy. Similar but weaker associations were found

for CO, but no association was found for O₃. The reported correlation matrix indicated that O₃ was negatively correlated with CO ($r = -0.45$) and only weakly correlated with PM₁₀ ($r = 0.2$). Other results from Beijing, China (Xu et al., 1995) and the Czech Republic (Bobak, 2000) suggested that SO₂ and TSP were associated with preterm births. Considering that O₃ tends to be negatively correlated with winter-type pollutants, O₃ is unlikely to be an important risk factor for preterm births.

Ritz et al. (2002) evaluated the effect of air pollution on the occurrence of birth defects in neonates and fetuses delivered in southern California from 1987 to 1993 as ascertained by the California Birth Defects Monitoring Program. They averaged air pollution (CO, O₃, PM₁₀, and NO₂) levels measured at the assigned ambient station over the first, second, and third month of gestation. Conventional, polytomous, and hierarchical logistic regressions were used to estimate odds ratios for subgroups of cardiac and orofacial defects. Concentration-response relationships were observed for second month CO exposure on ventricular septal defects, and second month O₃ exposure on aortic artery and valve defects, pulmonary artery and valve anomalies, and conotruncal defects. The odds ratios observed for these outcomes were similar and quite large (e.g., the odds ratios comparing the highest [monthly 24-h avg mean 34.9 ppb] to lowest [monthly mean 6.4 ppb] O₃ quartiles ranged from 2.0 to 2.7), and were not sensitive in multipollutant models. Ritz et al. (2002) reported that they did not observe consistently increased risks and concentration-response patterns for NO₂ and PM₁₀ after controlling for the effects of CO and O₃. Results from this study contrast to those for other birth-related outcomes, in that both CO and O₃ (presumably negatively correlated pollutants) were associated with birth defects. Further, O₃ showed associations with more birth defect outcomes than did CO. It should be noted, however, that the concentration-response relationships were quite specific to exposures during the second month. Associations with third month exposures were often negative (though not significantly). Since both CO and O₃ show strong seasonal peaks, it is possible that seasonal confounding could have played some role in these associations. This is the only study to date that examined the relationship between air pollution and birth defects.

In summary, O₃ was not an important predictor of several birth-related outcomes including intrauterine and infant mortality, premature births, and low birth weight. Birth-related outcomes generally appear to be associated with air pollutants that tend to peak in the winter and are possibly traffic-related, most notably CO. The strong results for CO are consistent with its

ability to cross the placental barrier and the high affinity that hemoglobin in fetal blood has for binding with it. However, given that most of these studies did not analyze the data by season, seasonal confounding may have therefore influenced the reported associations. One study reported some results suggestive of associations between exposures to O₃ in the second month of pregnancy and birth defects, but further evaluation of such potential associations is needed.

7.5.10 Summary of Chronic Ozone Exposure Effects on Morbidity and Mortality

- In the past decade, important new longitudinal studies have examined the effects of chronic O₃ exposure on respiratory health outcomes, including seasonal declines in lung function, increases in inflammation, and development of asthma in children and adults. Seasonal O₃ effects on lung function have been reported in several studies. In contrast to supportive evidence from chronic animal studies, epidemiologic studies of inflammation, new asthma development, and longer-term lung function declines remain inconclusive.
- Few studies have investigated the effect of long-term O₃ exposure on mortality and cancer incidence. Uncertainties regarding the exposure period of relevance, differential effects by gender, and inconsistencies across outcomes raise concerns regarding plausibility. There is currently little evidence for a relationship between chronic O₃ exposure and increased risk of lung cancer or of mortality.
- A limited number of studies have examined the relationship between air pollution and birth-related health outcomes, including mortality, premature births, low birth weights, and birth defects. The most consistent associations with various birth outcomes were observed for CO, with very little credible evidence being found for any O₃ effects.

7.6 INTERPRETIVE ASSESSMENT OF THE EVIDENCE IN EPIDEMIOLOGIC STUDIES OF OZONE HEALTH EFFECTS

7.6.1 Introduction

In the 1996 O₃ AQCD, discussion of the available epidemiologic information focused primarily on individual-level camp and exercise studies and on studies of hospital admissions and emergency room visits. The field studies indicated concentration-response relationships of ambient air O₃ exposure with declines in pulmonary function, increases in respiratory symptoms, and exacerbation of asthma, especially in children. Numerous new studies provide additional

evidence for evaluating associations between O₃ exposure and the above respiratory health outcomes. The 1996 O₃ AQCD review of aggregate population time-series studies indicated an association between ambient O₃ concentrations and increased hospitalizations. A limited number of studies examined O₃-mortality relationships. The current O₃ AQCD presents results from time-series studies that have addressed previously unresolved issues regarding potential linkages between ambient O₃ concentrations and health outcomes, particularly mortality. Daily time-series studies minimize confounding by population characteristics (e.g., cigarette smoking, diet, occupation) by following the same population from day to day. However, confounders operating over shorter time scales can affect O₃ risk estimates in these studies.

In this section, issues and attendant uncertainties that affect the interpretation of available epidemiologic evidence regarding O₃ health effects are discussed. The use of various indices to represent O₃ exposure in epidemiologic studies is discussed first. Also, of interest is the issue of confounding by temporal factors, meteorological factors, and copollutants. The shape of the concentration-response function and heterogeneity of O₃ effects are also discussed briefly. All of these topics are important for characterizing and interpreting ambient O₃-health effects associations.

7.6.2 Ozone Exposure Indices

Three O₃ indices were most often used to indicate daily O₃ exposure: maximum 1-h average (1-h max); maximum 8-h average (8-h max); and 24-h average (24-h avg). The 8-h max O₃ concentration is a frequently used index in newer epidemiologic studies, as it relates most closely to the averaging time of the current U.S. EPA NAAQS. The O₃ exposure indices are highly correlated, as indicated in several studies. For example, in the 21 European multicities acute mortality study (Gryparis et al., 2004), 1-h max O₃ was found to be highly correlated with 8-h max O₃, with a median correlation coefficient of 0.98 (range 0.91 to 0.99). Among single-city studies, the 1-h max O₃ and 8-h max O₃ also were found to have correlation coefficients ranging from 0.91 to 0.99 in various areas, such as Atlanta, GA (Tolbert et al., 2000; White et al., 1994); southern New England (Gent et al., 2003); Ontario, Canada (Burnett et al., 1994); and Mexico City (Loomis et al., 1996; Romieu et al., 1995). In addition, 1-h max O₃ was highly correlated with 24-h avg O₃, as observed in the Mexico City study by Loomis et al. (1996) ($r = 0.77$) and in the Ontario, Canada study by Burnett et al. (1994) ($r = 0.87$).

All studies discussed in Sections 7.2 to 7.5 were examined for presentation of the three O₃ exposure indices. Several presented the concentration data and correlations among 1-h max, 8-h max, and 24-h avg O₃ ambient measures. Some presented the associated risk estimates of comparable analyses for the three exposure indices. No papers provided a statistical analysis comparing results from the different indices. The increments used in this document to standardize expressions of excess risks are 40 ppb for 1-h max O₃, 30 ppb for 8-h max O₃, and 20 ppb for 24-h avg O₃, as discussed in Section 7.1.3.2. Key findings derived from the assessed studies are discussed below, starting with two multicity mortality studies.

In the large U.S. 95 communities study by Bell et al. (2004), increases in O₃-associated daily mortality were estimated using all three O₃ indices. The mean 24 h avg O₃ level was appropriately 26 ppb across the 95 communities. Ozone concentrations for 1-h max O₃ and 8-h max O₃ were not provided. For the above-noted increments, the effect estimates calculated by Bell et al. (2004) using all available data were 1.34% (95% PI: 0.84, 1.85), 1.28% (95% PI: 0.88, 1.73), and 1.04% (95% PI: 0.54, 1.55) excess risk in mortality for 1-h max O₃, 8-h max O₃, and 24-avg O₃, respectively. A statistical test examining differences among these risk estimates indicated that there were no significant differences by exposure index. In the European study of 21 cities (two of the 23 cities evaluated did not have 8-h max O₃ data), the O₃-mortality effect estimate for the summer season was slightly smaller for the 8-h max O₃, 1.82% (95% CI: 0.99, 3.06) excess risk compared to the 1-h max O₃, 2.59% (95% CI: 1.32, 4.10) excess risk; however, the two risk estimates were not significantly different (Gryparis et al., 2004). The median 1-h max O₃ levels ranged from 44 to 117 ppb and the median 8-h max O₃ levels ranged from 30 to 99 ppb across all cities during the summer.

Several single-city mortality studies examined multiple O₃ exposure indices (Anderson et al., 1996; Dab et al., 1996; Sunyer et al., 2002; Zmirou et al., 1996; Borja-Aburto et al., 1997). These studies did not differentiate risk estimates by exposure index, because the results were considered to be similar. Hospital admission studies also provided limited data for O₃ index comparisons. Schouten et al. (1996) found similar O₃ effects on total respiratory hospitalizations from 8-h max O₃ and 1-h max O₃ in the summer. Both indices resulted in a 4.0% excess risk per standardized increment. For emergency department visits, the examples of Delfino et al. (1998b) and Weisel et al. (2002) indicated no difference in effect estimate when using various O₃ indices. Tolbert et al. (2000) noted an increase in emergency room visits of 4.0% per standard

deviation increase (approximately 20 ppb) for both 1-h max O₃ and 8-h max O₃ as being expected, since the correlation between the indices was 0.99.

Peak flow asthma panel studies generally used only one index; thus, there were limited data available for comparison. One respiratory symptom study (Gent et al., 2003) examined both 1-h max O₃ and 8-h max O₃ but noted no differences in the results. Only one FEV₁ panel study examined more than one O₃ exposure index. Chen et al. (1999) examined 1-h max O₃ and 24-h avg O₃ and reported a decrement in FEV₁ of -25.6 mL (95% CI: -49.1, -2.1) per 40 ppb increase in 1-h max O₃ and -13.6 mL (95% CI: -33.2, 6.0) per 20 ppb increase in 24-h avg O₃ in children at a 1-day lag. For 2- and 7-day lags, smaller differences were observed between the two indices. The effect estimates calculated using 1-h max O₃ and 24-h avg O₃ concentrations were not found to be significantly different for any of the lags examined.

Limited information is available by which to draw conclusions for comparison across the three O₃ indices of 1-h max O₃, 8-h max O₃, and 24-h avg O₃. Studies conducted in various cities have observed very high correlations among the daily O₃ indices. For the same distributional increment, the excess health risk estimates and significance of associations were generally comparable for the three O₃ indices across all outcomes. The high correlation among the indices presents a challenge in distinguishing the most appropriate measure for epidemiologic studies. Exploratory analyses using various O₃ exposure indices are valuable in understanding relationships. However, to address the issue of multiple hypothesis testing, hypotheses that are confirmatory and exploratory should be decided a priori and reported accordingly.

7.6.3 Confounding by Temporal Trends and Meteorologic Effects in Time-Series Studies

The challenge of analyzing acute O₃ effects in time-series studies is to avoid bias due to confounding by daily to seasonal temporal factors. On a seasonal scale, the analysis must remove the influence of the strong seasonal cycles that usually exist in both health outcomes and O₃. On a daily scale, weather factors and other air pollutants also may confound the association of interest. This section discusses the interpretation of effect estimates after adjusting for temporal trends and meteorologic effects.

7.6.3.1 Assessment of Ozone Effects after Adjusting for Temporal Trends and Meteorologic Effects

The relationship between O₃ and health outcomes are significantly affected by temporal trends and meteorological factors, namely temperature. Analyses of the association between health outcomes and O₃ concentrations using raw data, therefore, can be misleading. In Díaz et al. (1999), a U-shaped relationship was observed between mortality and O₃ concentrations, and the negative portion of the slope was likely due to the opposing seasonal cycles in mortality (high in winter) and temperature (low in winter).

Keatinge and Donaldson (2005) used two methods to identify confounding due to meteorological factors in their study of the mortality effect of air pollutants on adults aged 65 years or older during the warm season in Greater London. One involved GAM analyses controlling for up to seven weather covariates, and the other considered graphical comparisons to show interrelations of mortality with weather factors and pollutants. Both methods pre-adjusted for weather variables, which is a more stringent test of the air pollution effect as compared to co-adjustment. Few of the individual analyses found a significant O₃ effect, which tended to be small in comparison to the weather effect. Therefore, Keatinge and Donaldson (2005) concluded that, although air pollution may have a short-term adverse effect on health in London during the warm season, the results suggested that measures to prevent excess mortality in hot weather should focus on control of heat stress. Goldberg and Burnett (2003) reported a positive slope for the temperature-mortality relationship being fitted most tightly in the mild temperature range where mortality effects of temperature were not expected. While it is possible that temperature has mortality effects in the mild temperature range, daily fluctuations of air pollutants (especially O₃) are strongly influenced by weather conditions and ascribing the association between temperature and mortality entirely to effects of temperature may underestimate the effects of air pollution.

Sensitivity analyses to examine confounding of temperature on the O₃ effect were performed in the U.S. 95 communities study by Bell et al. (2004). In analyses excluding days with high temperature (that used a cutoff as low as 29 °C [85 °F]), the range of estimated effect sizes (1.00 to 1.10% excess risk in mortality per 20 ppb increase in 24-h avg O₃) did not differ from the effect estimate using all the data (1.04% excess risk). Effect estimates were found to be slightly higher at lower temperatures. In a related study of 19 U.S. cities, Huang et al. (2005)

examined the potential influence of heat waves, modeled as a natural cubic spline of the interaction between current temperature and average temperature in the three previous days, on the O₃ effect on cardiopulmonary mortality. The effect estimate was generally robust to inclusion of this interaction term, with only a slight decrease in the estimate from 2.52% (95% PI: 0.94, 4.10) to 2.23% (95% PI: 0.76, 3.75) excess risk per 20 ppb increase in 24-h avg O₃.

Bell et al. (2004) and Huang et al. (2005) also found that O₃ effects were robust to the selection of degrees of freedom for smoothing of temporal trends. Bell et al. observed that changing the degrees of freedom from 7 to 21 per year did not significantly affect the O₃-mortality estimates, with effect estimates ranging from 0.82 to 1.08% excess risk per 20 ppb increase in 24-h avg O₃ during the previous week. Huang et al. (2005) examined the sensitivity of summertime O₃ risk estimates to varying degrees of freedom from 4 to 16 per year. The extent of change in the risk estimates, though varied from city to city (graphically presented), was not substantial. Using more degrees of freedom in temporal trend fitting (i.e., controlling shorter temporal fluctuations) means ascribing more details of daily health outcomes to unmeasured potential confounders and possibly weakening real weather and air pollution effects. However, results from these large multicity studies indicated that O₃ effects were robust to aggressive smoothing of temporal trends.

Ito et al. (2005) examined sensitivity of O₃-mortality risk estimates to the extent of temporal trend adjustment and to alternative weather model specifications using data from seven U.S. cities (Cook County, IL; Detroit, MI; Houston, TX; Minneapolis, MN; New York City; Philadelphia, PA; and St. Louis, MO). They found that varying the degrees of freedom from 4 to 26 per year did not substantially or systematically affect the O₃-mortality estimates, except for Cook County where the percent excess O₃-mortality risk estimates were considerably reduced when the temporal adjustment term with 26 degrees of freedom was applied. Ito et al. (2005) noted that the O₃ risk estimates were generally more sensitive to alternative weather models (up to a 2-fold difference in percent excess risk) than to the degrees of freedom for temporal adjustment. Of the four weather models examined, the four-smoother model similar to the one used in the NMMAPS studies (Bell et al., 2004; Huang et al., 2005) generally resulted in the smallest effect estimates.

Schwartz (2005) examined the sensitivity of the O₃-mortality relationship to methods used to control for temperature. Initially, temperature lagged 0 and 1 day was controlled using

nonlinear regression splines with 3 degrees of freedom each. In a comparison analysis, control days were restricted to a temperature-matched subset. The effect estimates for all-year data using nonlinear regression splines (0.8% [95% CI: 0.1, 1.4] excess risk per 40 ppb increase in 1-h max O₃) and temperature-matched controls (0.9% [95% CI: 0.04, 1.8] excess risk) were not significantly different. Results were similar when restricting analysis to warm-season-only data.

Temporal cycles in daily hospital admissions or emergency department visits are often considerably more episodic and variable than is usually the case for daily mortality. As a result, smoothing functions that have been developed and tuned for analyses of daily mortality data may not work as well at removing cyclic patterns from morbidity counts. Two methods are commonly used to adjust for temporal trends. The pre-adjustment method involves applying the adjustment to both outcome and air pollution variables prior to the regression analysis. The co-adjustment method involves applying the adjustment as part of the regression analysis, by fitting a function of time while simultaneously fitting the regression effect of air pollution and weather factors. As shown in a hospital admissions study by Burnett et al. (2001), the co-adjustment approach may lead to biased air pollution effect estimates in cases where both outcome and pollution variables exhibit strong seasonal cycles (note that the data were analyzed using Poisson GAM with default convergence criteria). Using year-round data, pre-adjustment followed by regression analysis yielded a 14% (95% CI: 5, 24) increase in admissions per 40 ppb increase in 1-h max O₃ with a multiday lag of 0 to 4 days. The co-adjustment method resulted in a 7% (95% CI: 3, 11) decrease in admissions. When the authors limited the analysis to the warm season (May-August), both methods yielded similar results (32% [95% CI: 21, 44] versus 30% [95% CI: 17, 45] increase for co-adjustment and pre-adjustment, respectively), implying that stratification by season can remove a significant amount of the confounding seasonality (which also may include seasonally-varying population behavior and ventilation conditions). This finding may be important to consider in reviewing the acute O₃ mortality and morbidity literature, because the vast majority of studies published over the past decade have used the co-adjustment method. However, the use of pre-adjustment versus co-adjustment in time-series studies is an unresolved issue. More empirical research in different locales is needed to evaluate the merits of these two methods as far as O₃ is concerned and to determine what endpoints may be affected.

More sensitivity analysis of O₃ effect estimates to the extent of adjustment for temporal trends and meteorological factors is needed, but perhaps it is equally as important to evaluate the epidemiologic adequacy of a given adjustment. For example, do the fitted mortality series sufficiently depict influenza epidemics? Or, when larger degrees of freedom (e.g., 12 degrees of freedom per year) are used, what “unmeasured” confounders, other than weather and pollution, are the investigators trying to adjust? Even in PM studies that conducted sensitivity analyses, investigators rarely stated assumptions clearly or provided adequate discussions as to potential reasons for the sensitivity of results.

Given their relationship to health outcomes and O₃ exposure, adjusting for temporal trends and meteorologic factors is critical to obtain meaningful O₃ effect estimates. While the prevailing analytical approaches fit the data flexibly, the estimated effects of meteorologic variables and their impact on the adjusted O₃ effects are not adequately discussed. More work is needed in this area to reduce the uncertainty involved in the epidemiologic interpretation of O₃ effect estimates.

7.6.3.2 Importance of Season-Specific Estimates of Ozone Health Effects

Analysis of O₃ health effects is further complicated, as relationships of O₃ with other pollutants and with temperature appear to change across seasons. Moolgavkar et al. (1995) examined the relationship between daily mortality and air pollution by season in Philadelphia, PA for the period of 1973 to 1988. During the summer, there was a positive relationship between O₃ and TSP, as well as O₃ and SO₂. In contrast, the relationship of O₃ with TSP and SO₂ was inverse during the winter. Ozone showed positive associations only in the summer when the mean O₃ concentration was the highest. The effect of O₃ on mortality was negative (though not significantly) in the winter when the mean O₃ concentration was low. In the summer multipollutant model, O₃ was the only pollutant that remained significant. Similar results were found in another Philadelphia study by Moolgavkar and Luebeck (1996). Neither studies analyzed year-round data; therefore, the relationship between the excess risk estimates for all year and each season could not be compared. The results from these studies, however, suggest that year-round analyses may mask associations (positive or negative) that may exist in particular seasons.

Ito et al. (2005) examined O₃-mortality associations in seven U.S. cities, but also described the relationship between O₃ and PM for summer months (June through August) and winter months (December through February) in these cities (see Figure 7-28). The O₃-PM relationships were positive in the summer and negative in the winter in all of these cities, except in Houston, where the O₃-PM association was not clearly positive in the warmer months but positive in colder months. Ito et al. (2005) found that O₃-mortality associations were mostly weaker, null, or even negative in the winter compared to the summer in most of these cities. Once again, the exception was Houston where the cold season O₃-mortality association was positive and larger than those for year-round or warmer months. Findings from this study suggest the influence of seasonal O₃-PM relationships on O₃-mortality associations. It should be noted that the contrasts in O₃ risk estimates between warm and cold months varied across the alternative weather models examined, with the four-smoother model generally showing the least contrast.

In the analyses of the U.S. 90 cities data (of which 80 cities had O₃ data available) by Samet et al. (2000; reanalysis Dominici et al., 2003), the focus of the study was PM₁₀, but O₃ and other gaseous pollutants also were analyzed in single- and multiple-pollutant models. In the reanalysis (Dominici et al., 2003), O₃ was associated with an excess risk of mortality in analyses of all available data (0.4% [95% PI: 0.1, 0.7] excess risk per 20 ppb increase in 24-h avg O₃ at a 1-day lag) and summer only data (1.0% [95% PI: 0.5, 1.6]); however, a negative association was observed for the winter only analysis (-1.1% [95% PI: -2.2, 0.1]). A 2-fold greater effect was estimated using summer data compared to all available data. The mean 24-h avg O₃ levels using all available data ranged from 12 to 36 ppb across the 80 cities. Season-specific O₃ concentrations were not presented. It should be noted that the analyses by Samet et al. and Dominici et al. used a weather model specification that is more detailed than other studies in that it had multiple terms for temperature and dewpoint (these two variables are generally highly correlated). Thus, it is possible that the high concavity of O₃ with these weather covariates may have produced these conflicting results. Another possibility is that the apparent negative relationship between O₃ and mortality in the winter may have been due to confounding by PM. In the larger U.S. 95 communities study by Bell et al. (2004), the all-available data and warm-season-only analyses also indicated positive risk estimates (1.04% [95% PI: 0.54, 1.55] and 0.78% [95% PI: 0.26, 1.30] excess risk per 20 ppb increase in 24-h avg O₃, respectively, using a constrained distributed 7-day lag model), but the two estimates were similar in magnitude.

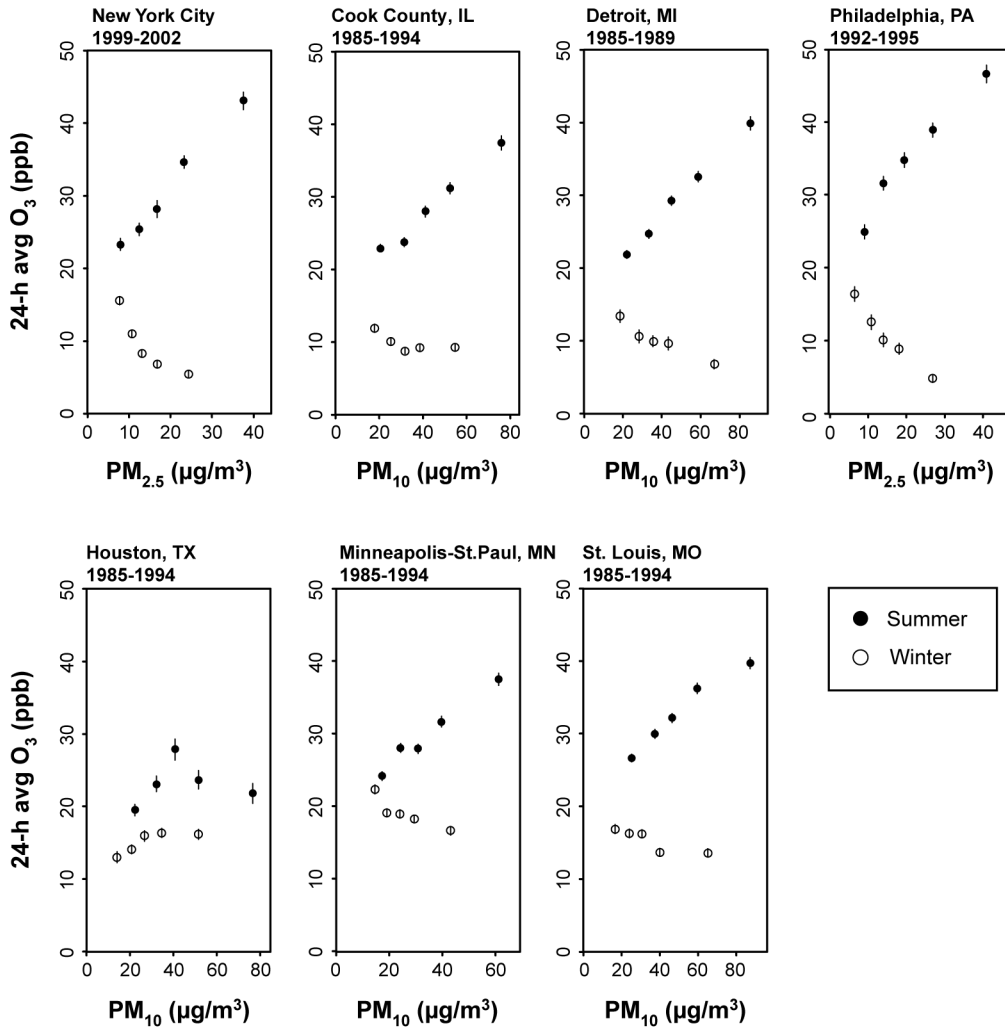


Figure 7-28. The relationship between PM and O₃ in the summer (June through August) and the winter (December through February) as sorted and averaged by quintiles of PM.

Source: Derived from Ito et al. (2005).

The median 24-h avg O₃ levels ranged from 14.4 to 37.3 ppb in the 55 communities with all-year data and from 20.4 to 36.2 ppb in the 40 communities with warm-season only data. Cold-season only analyses were not performed. Results from the U.S. 95 communities study appear to conflict with the strong seasonal variation observed in the U.S. 90 cities study. However, there are several differences between the two studies that might account for these results. First, the

U.S. 95 communities study nearly doubled the study period by extending the analysis by six additional years (1987 to 2000 versus 1987 to 1994) and it included 15 additional cities in addition to the original 80. Also, the warm seasons are characterized differently in the two studies. The U.S. 90 cities study defined summer as a three-month period of June through August, whereas the 95 communities study defined the warm season as a 7-month period of April through October. In addition, the results presented in the U.S. 90 cities study were from a single-day lag model (1-day lag) while the estimates from the 95 communities study were calculated using a constrained distributed 7-day lag model. Differences in seasonal O₃ effects observed in the two related studies might be attributable to some of these factors.

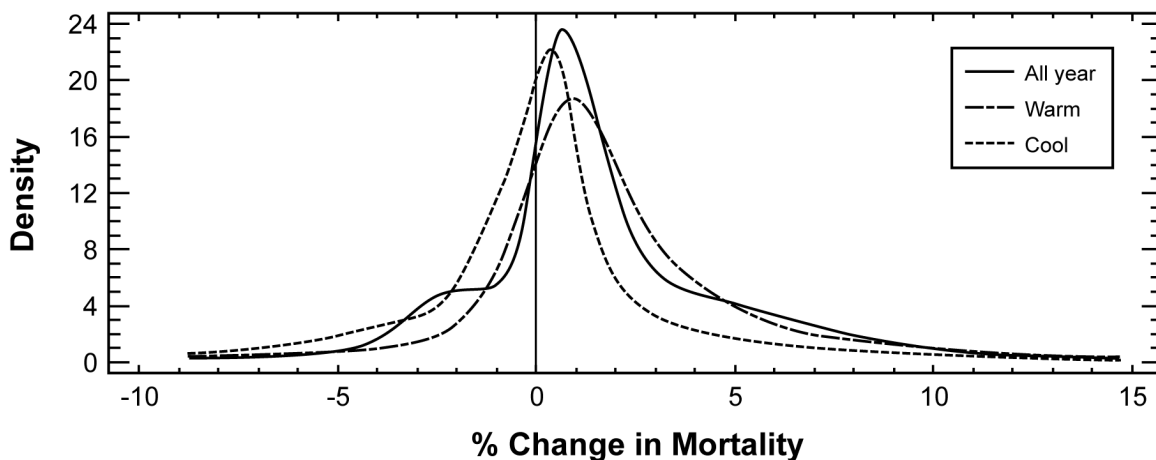
Many studies reported larger excess mortality risks in the warm (or summer) season than in the cool (or winter) season (see Figure 7-21 in Section 7.4.5). These studies showed cool-season risk estimates that were either smaller compared to warm-season estimates or slightly negative. Of the studies that analyzed data by season, only one study in Pittsburgh, PA (Chock et al., 2000) showed negative risk estimates for the summer. Ozone concentration data were not provided for the Chock et al. (2000) study. The studies that observed larger, positive associations between O₃ and mortality in warm seasons are consistent with the expectation that O₃, if harmful, should have a stronger association with health outcomes for the summer when exposure to O₃ is higher. However, the negative O₃-mortality associations seen for the winter suggest that further examination of this issue is required. Specifically, if O₃ levels in the winter are shown to be negatively associated with factors (e.g., PM) that are positively associated with mortality, then these potentially spurious negative O₃-mortality associations can be explained. Several examples of this phenomenon also exist in morbidity studies investigating the effect of O₃ on daily hospital admissions and emergency department visits (Anderson et al., 1998; Burnett et al., 2001; Prescott et al., 1998; Thompson et al., 2001).

Unlike the time-series studies examining outcomes of mortality, hospital admissions, and emergency department visits, most acute field studies did not perform year-round analyses. The acute field studies that examined the relationship between O₃ and lung function, respiratory symptoms, and inflammation focused primarily on the O₃ effect during the warm season when O₃ levels were expected to be high and subjects spent more time outdoors and were physically active.

There are seasonal (e.g., air conditioning use) or seasonally-modified (e.g., time spent outdoors, air exchange rates) factors that affect the relationship between ambient concentrations and personal exposures to O₃, as discussed in Section 3.9. The influence of combinations of these factors across seasons on air pollution health effects can become quite complex. For example, longer time spent outdoors in the summer may increase personal exposure to O₃ for some segment of the population, but the increased use of air conditioning may reduce exposures to ambient O₃ for those who spend much of their time indoors. In the meta-analysis by Levy et al. (2005), the combined risk estimate from the warm season was greater (3.38% [95% CI: 2.27, 4.42] per 40 ppb increase in 1-h max O₃) compared to the estimate from all-year data (1.64% [95% CI: 1.25, 2.03]). However, further analysis suggested that the O₃-mortality risk estimates were smaller in cities with high air conditioning prevalence. Seasonal factors such as these that influence the relationship between ambient concentrations and personal exposures make the interpretation of the concentration-response relationships obtained from analyses of year-round data less straightforward.

In some cities, O₃ is only monitored during the warm season. For example, 34% of the communities in the U.S. 95 communities study only collected O₃ data during the warm season (Bell et al., 2004). The cities with larger populations and/or higher O₃ concentrations generally collected year-round data. There is some concern that differential data availability may also contribute to the seasonal differences in O₃ health effects observed.

The potential influence of season on O₃ effect estimates was examined here using summary density curves. The O₃ effect observed in all-year data was compared to effects from warm-season and cool-season only data (Figures 7-29 and 7-30). Summary probability density curves were calculated to assess the effect estimates from the various studies (see Annex Section AX7.2 for further explanation of summary density curves). The summary density curves shown in Figures 7-29 and 7-30 were smoothed by multiplying a constant to the standard error of each effect estimate in the calculation of the individual distribution functions. Since the normal distribution is unimodal, this constant will oversmooth when the density is multimodal. In Figure 7-29, the summary density curves of O₃-associated all-cause (nonaccidental) mortality are presented (see Figure 7-21 in Section 7.4.5 for the effect estimates). The summary density curves are calculated using results from 14 studies that reported at least two of the three estimates. This figure indicates that 75% of the area under the density curve has a value greater

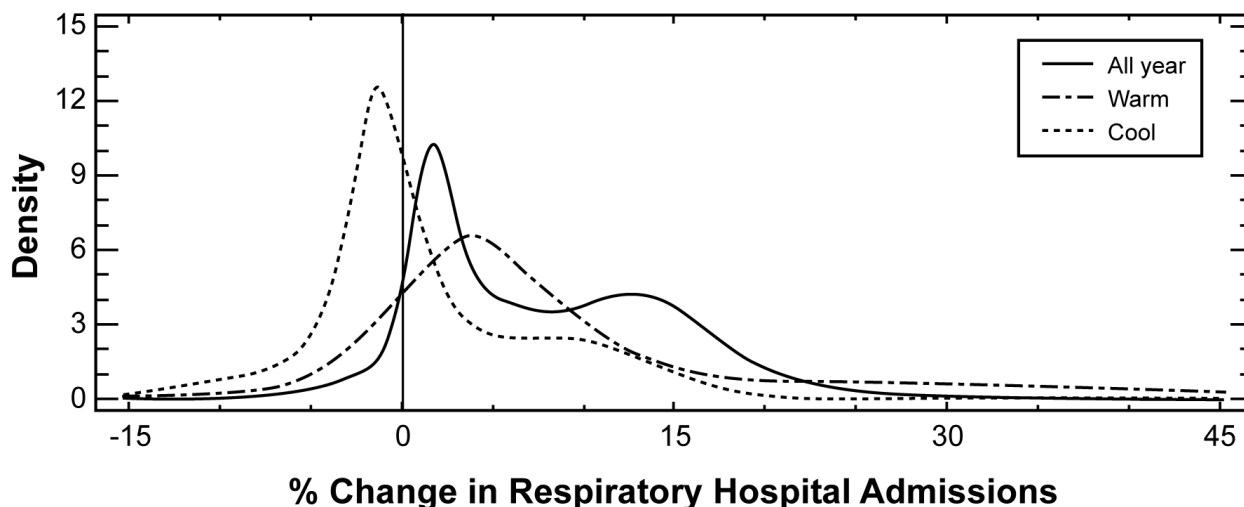


	All year	Warm season	Cool season
% area under the density curve and >0	75%	78%	62%
Mean (SD) effect estimates	1.2% (2.8%)	1.3% (2.6%)	0.1% (3.1%)
Mode effect estimates	0.7%	0.9%	0.4%

Figure 7-29. Summary density curves of the percent change in all cause mortality for all-year data and by season per standardized increment (see Section 7.1.3.2). Effect estimates from 14 studies have been included in the summary density curves (see Figure 7-21 in Section 7.4.5 for the effect estimates).

than zero for all-year data compared to 78% for warm-season data and 62% for cool-season data. Therefore, both all-year and warm-season data generally indicate a positive O₃ effect on mortality. The mean effect estimates for all-year data and warm-season only data are 1.2% (SD 2.8) and 1.3% (SD 2.6) excess risk in mortality per 40 ppb increase in 1-h max O₃, respectively. A slightly larger mode of effects is observed for warm-season data (0.9% excess risk) compared to all-year data (0.7%). The cool-season-only data indicate that there is no excess risk (mean 0.1% [SD 3.1]) associated with O₃ concentrations.

Similar observations are made when examining the O₃ effect on total respiratory hospital admissions (Figure 7-30). Six studies provided season-specific estimates as well as all-year results (see Figure 7-10 in Section 7.3.3 for the effect estimates). Once again, a large percent of the area under the summary density curve is greater than zero when using all-year and warm-season data, 92% and 84%, respectively, compared to cool-season data, 49%.



	All year	Warm season	Cool season
% area under the density curve and >0	92%	84%	49%
Mean (SD) effect estimates	6.5% (6.4%)	6.3% (9.1%)	0.8% (6.1%)
Mode effect estimates	1.8%	4.0%	-1.3%

Figure 7-30. Summary density curves of the percent change in total respiratory hospital admissions for all-year data and by season per standardized increment (see Section 7.1.3.2). Effect estimates from six studies have been included in the summary density curves (see Figure 7-10 in Section 7.3.3 for the effect estimates).

The mean O₃ effect estimates for warm-season-only data, 6.3% (SD 9.1) excess risk per 40 ppb increase in 1-h max O₃, and all-year analyses, 6.5% (SD 6.4) excess risk, are similar. A larger mode of effects is observed for warm-season data (3.9% excess risk) compared to all-year data (1.8% excess risk). A small O₃ effect (0.8% [SD 6.1] excess risk) is observed when using cool-season data only.

Integrating seasonal influences across the various health outcomes supports the view that O₃ effects are different in the cool and warm seasons, with greater effects observed during the warm season. Warm-season data should be used to derive quantitative relationships for the effect of O₃ on health outcomes, because the attenuation of O₃ exposure from ambient concentrations is likely to be much less during this season. However, studying summer data

only when all-year data are available weakens the power of the study, since less data are analyzed. In addition, increased adverse health outcomes are observed in the winter, some of which may be attributable to O₃. The O₃ effect in the wintertime may be masked by the effects of PM due to the negative correlation between these two variables (see Section 7.6.4.2 for further discussion). Therefore, analysis of all-year data may be improved by adjusting for PM indices in addition to adequate adjustment of meteorological factors and temporal trends.

Seasonality influences the relationship between O₃ and health outcomes, as it may serve as an indicator for time-varying factors, such as temperature, copollutant concentrations, infiltration, and human activity patterns. Given the potentially significant effect of season, O₃ effect estimates computed for year-round data need to be interpreted with caution. Finding small or no effects may simply reflect (a) the cancellation of positive associations in the summer and negative associations in the winter or (b) the presence of confounding due to the strong seasonal character of O₃ concentrations.

7.6.4 Assessment of Confounding by Copollutants

Potential confounding by daily variations in copollutants is another analytical issue to be considered. With respect to copollutants, daily variations in O₃ tend to not correlate highly with most other criteria pollutants (e.g., CO, NO₂, SO₂, PM₁₀), but they may be more correlated with secondary fine PM (e.g., PM_{2.5}, sulfates) measured during the summer months. Assessing the independent health effects of two pollutants that are correlated over time is not straightforward. If high correlations between O₃ and PM or other gaseous pollutants exist in a given area, then disentangling their relative individual contributions to observed health effects associations becomes very difficult. The changing relationship between O₃ and other copollutants also is at issue. Ito et al. (2005) described the relationship between O₃ and PM by season in seven urban U.S. cities: Cook County, IL; Detroit, MI; Houston, TX; Minneapolis-St. Paul, MN; New York City; Philadelphia, PA; and St. Louis, MO (see Figure 7-28). With the exception of Houston, TX, the O₃-PM associations were positive in the summer and negative in the winter. Relationships between O₃ and copollutants and the potential confounding of the O₃ effect by copollutants are discussed in the following section.

7.6.4.1 Relationship between Personal Exposure to Ozone and Copollutants

The correlation between ambient O₃ concentrations and ambient levels of PM, NO₂, SO₂, and CO measured at central monitoring sites has been reported in many studies; however, only a limited number of studies have examined the association between personal O₃ exposures and personal exposures to other copollutants. The relationship between personal exposure to O₃ and personal exposure to PM_{2.5} was examined by Chang et al. (2000) in various microenvironments in Baltimore, MD, using activity data from the National Human Activity Pattern Survey study (Klepeis, 1999). Activities were scripted to simulate activities performed by older adults (65+ years of age). Mean hourly personal O₃ exposures were 15.0 ppb (SD 18.3) in the summer and 3.6 ppb (SD 7.5) in the winter. Strong correlations were not observed between 1-h personal O₃ and PM_{2.5} concentrations in the various microenvironments, including the indoor residence (n = 91), other indoor environments (n = 53), outdoor near the roadways (n = 21), outdoor away from roads (n = 19), and in vehicles (n = 71). Spearman r ranged from -0.14 to 0.29 during the summer and -0.28 to 0.05 during the winter (SEs not provided, all correlation coefficients noted as statistically nonsignificant [p > 0.05]).

An issue of particular interest is the correlation between personal exposure to O₃ and personal exposure to the ambient component of PM_{2.5}. In a study of susceptible populations (older adults [n = 20], individuals with COPD [n = 15], and children [n = 21]) in Baltimore, MD, a total of 800 person-days of exposure data were collected for the following pollutants: PM_{2.5}, PM₁₀, O₃, NO₂, SO₂, elemental carbon, organic carbon, and volatile organic compounds (Sarnat et al., 2001). A subset of PM_{2.5} filters was analyzed for SO₄²⁻ concentration to estimate the personal exposure to PM_{2.5} of ambient origin. The mean ambient 24-h avg O₃ concentration was approximately 36 ppb. Sarnat et al. (2001) found that ambient 24-h avg O₃ concentrations and ambient 24-h avg PM_{2.5} levels, both measured at centrally located monitoring sites, were positively associated in the summer ($\beta = 0.84$ [95% CI: 0.56, 1.12], Spearman r = 0.67 [p < 0.05]) and negatively associated in the winter ($\beta = -0.67$ [95% CI: -0.91, -0.43], Spearman r = -0.72 [p < 0.05]). An association also was observed between ambient O₃ concentrations and personal PM_{2.5} of ambient origin, with a mixed regression effect estimate of $\beta = 0.37$ (95% CI: 0.25, 0.49) in the summer and $\beta = -0.36$ (95% CI: -0.41, -0.31) in the winter. However, no significant relationship was found between 24-h avg personal O₃ exposure

and personal exposure to $PM_{2.5}$ of ambient origin in either season ($\beta = 0.22$ [95% CI: $-0.06, 0.50$] in the summer and $\beta = -0.18$ [95% CI: $-0.39, 0.03$] in the winter).

In a related study conducted in Boston, MA, 20 healthy senior citizens and 23 school children were monitored during six sessions for a total of 714 person-days in both the summer and winter seasons (Sarnat et al., 2005). Personal SO_4^{2-} exposures were used as indicators of personal exposure to $PM_{2.5}$ of ambient origin. Mean ambient 24-h avg O_3 levels ranged from 22.7 to 31.6 ppb during the three summer sessions and 14.0 to 21.8 ppb during the three winter sessions. Similar to the earlier study conducted in Baltimore, ambient 24-h avg O_3 levels and ambient 24-h avg $PM_{2.5}$ levels were positively associated in the summer ($\beta = 0.51$ [95% CI: 0.34, 0.68]) but negatively associated in the winter ($\beta = -0.53$ [95% CI: $-0.22, -0.85$]). In addition, an association was observed between ambient O_3 concentrations and personal $PM_{2.5}$ of ambient origin, but only in the summer ($\beta = 0.24$ [95% CI: 0.20, 0.28]). However, in contrast to the Baltimore results, a significant association between 24-h avg personal O_3 exposure and personal exposure to $PM_{2.5}$ of ambient origin was observed both in the summer ($\beta = 0.35$ [95% CI: 0.22, 0.47]) and in the winter ($\beta = 0.07$ [95% CI: 0.01, 0.13]). Neither of the Sarnat et al. (2001, 2005) studies evaluated relationships between personal O_3 exposures and personal exposure to other gaseous pollutants.

Results from the two related Sarnat studies provide initial evidence that the relationship between personal O_3 exposure and personal exposure to $PM_{2.5}$ of ambient origin may differ by city/region and season. The authors noted that their results were for a small, nonrandom selection of subjects living in the eastern United States, therefore caution should be exercised in generalizing the results to other locations and cohorts. The results are likely to vary for locations depending on indoor air exchange rates and amount of time spent outdoors.

7.6.4.2 Assessment of Confounding Using Multipollutant Regression Models

Multipollutant regression models are generally used to determine whether the pollutant-specific effect is robust. However, because of the multicollinearity among O_3 and pollutants and the changing correlations by season, multipollutant models may not adequately adjust for potential confounding, especially when using year-round data. These limitations need to be considered when evaluating results from multipollutant models. Results from the U.S. 90 cities study, which included 80 cities with O_3 data, indicated that while the addition of PM_{10} in the

model did not substantially change the O₃-mortality risk estimates, slight declines in the O₃ effect were observed, as shown in Figure 7-31 (Samet et al., 2000; reanalysis Dominici et al., 2003). In the extended U.S. 95 communities study (Bell et al., 2004), the city-specific O₃-mortality effects were robust to adjustment for PM₁₀, as indicated by the nearly 1:1 ratio between estimates with and without PM₁₀ adjustment shown in Figure 7-32. This finding suggested that PM₁₀ generally did not confound the association between O₃ and mortality. Limited data were available to examine the potential confounding effect of PM_{2.5} on the O₃-mortality relationship. A weighted second-stage linear regression indicated that there was no association between long-term PM_{2.5} average and the community-specific O₃-mortality effect estimate. Several other mortality and morbidity studies have investigated confounding of O₃ risk estimates using multipollutant models with year-round data, and most have reported that O₃ effects were robust to adjustment for copollutants (see Figures 7-11 and 7-22 in Sections 7.3.3 and 7.4.6, respectively).

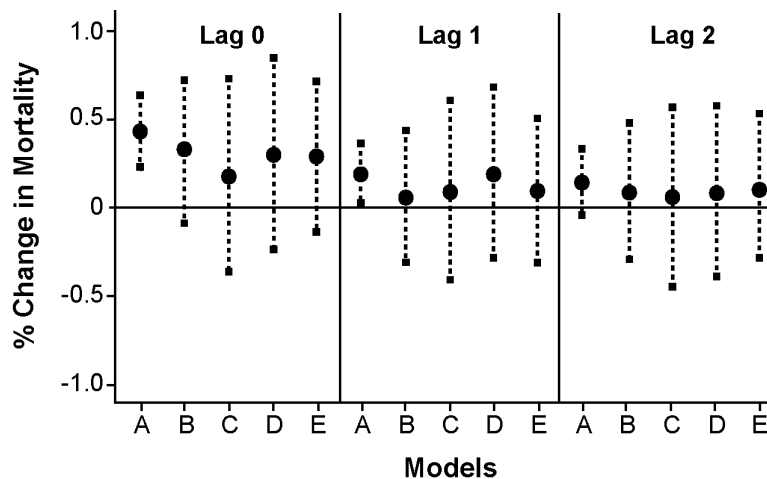


Figure 7-31. Posterior means and 95% probability intervals for the national average estimate of O₃ effects on total mortality from non-external causes per 10 ppb increase in 24-h avg O₃ at 0-, 1-, and 2-day lags within sets of 80 U.S. cities with pollutant data available. Models A = O₃ only; B = O₃ + PM₁₀; C = O₃ + PM₁₀ + NO₂; D = O₃ + PM₁₀ + SO₂; E = O₃ + PM₁₀ + CO.

Source: Derived from Dominici et al. (2003).

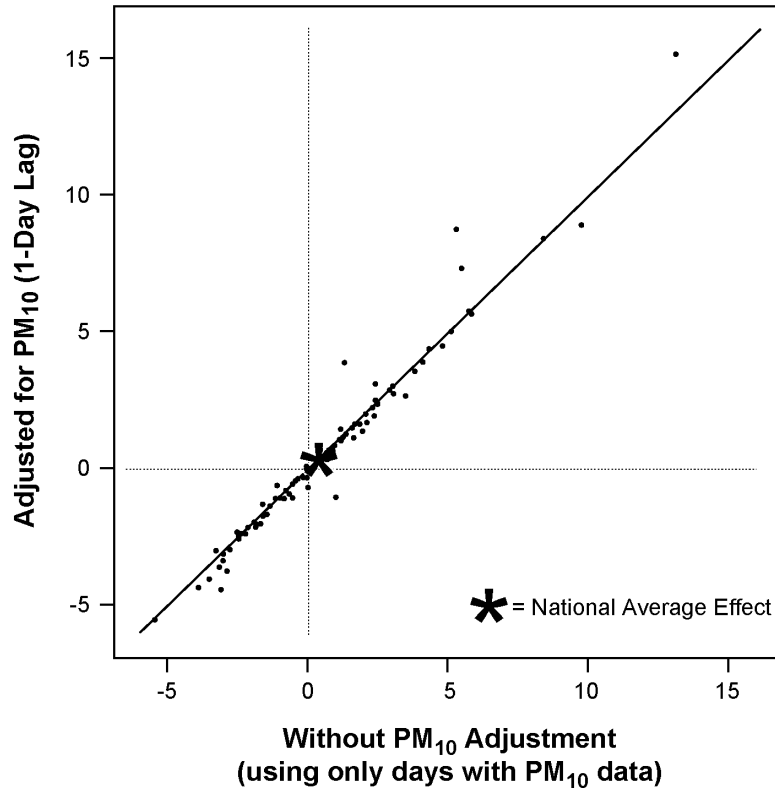


Figure 7-32. Maximum likelihood estimates of O₃-mortality for 95 U.S. communities, determined using a constrained distributed lag model for lags 0 through 6 days. Same data set was used for O₃ estimates with and without adjustment for PM₁₀.

Source: Derived from Bell et al. (2004).

The pollutant most correlated with O₃ in the summer is sulfate (which is in the fine particle size range), especially in the eastern United States. Therefore, the main potential confounders of interest for O₃ in the summer are PM_{2.5} and sulfate. Once again, the results from two-pollutant regression models with O₃ and sulfate (or PM_{2.5}) should be interpreted with caution because both of these pollutants are formed under the same atmospheric conditions and both are part of the “summer haze” pollution mix. A simple two-pollutant regression model does not address their possible synergistic effects, and the high correlation between the two pollutants may lead to unstable and possibly misleading results. In any case, most studies that analyzed O₃ with PM indices did not have PM_{2.5} data and very few examined sulfate data. A mortality study by Lipfert

et al. (2000a) found that all-year O₃ risk estimates were not affected by the addition of sulfate (3.2% versus 3.0% with sulfate per 45 ppb less background level of 1-h max O₃). Among studies with PM_{2.5} data, including Santa Clara County, CA (Fairley, 1999; reanalysis Fairley, 2003), Philadelphia, PA (Lipfert et al., 2000a), and Detroit, MI (Lippmann et al., 2000; reanalysis Ito, 2003), most examined copollutant models for year-round data only, but O₃-mortality risk estimates were not substantially affected by the addition of PM_{2.5}. An analysis of Philadelphia and Detroit data by season suggested that O₃-mortality risk estimates were not sensitive to adjustment for PM_{2.5} in all-year or seasonal analyses (Ito et al., 2005). Including both O₃ and PM_{2.5} in the regression models for these cities tended to attenuate both pollutants' coefficients, but not substantially. The models with both O₃ and PM_{2.5} in these cities often showed better fits (i.e., lower Akaike's Information Criteria) than those with either pollutant alone. Ito et al. (2003) concluded that these results suggest that O₃ and PM_{2.5} contribute independently to mortality.

Other studies have estimated O₃ health risks with copollutants in the model by season. Respiratory hospitalization studies conducted during the warm season in Canada observed consistent O₃ risk estimates with the inclusion of PM_{2.5} in the model (Burnett et al., 1997b, 2001). In one of these studies (Burnett et al., 1997b), the effect of O₃ also was adjusted for sulfate. With the addition of sulfate in the model, the risk estimate for O₃ on respiratory hospitalizations remained relatively stable, from a 14.4% (95% CI: 8.7, 20.5) excess risk to a 11.7% (95% CI: 5.6, 18.0) excess risk per 25 ppb increase in 12-h avg O₃ at a 1- to 3-day lag. In contrast, the effects for sulfate were reduced in half after adjusting for O₃. Amongst the mortality studies (see Figure 7-23 in Section 7.4.6), adjusting for copollutants, in particular PM indices, did not substantially change the warm-season O₃-mortality effect estimates, with both slight reductions and increases observed in the adjusted estimates. In the analysis using cool-season-only data, the O₃ effect estimates were generally negative, but none were statistically significant. The O₃ risk estimates all increased slightly with the adjustment of PM indices. The inverse relationship between O₃ and PM during the cool season most likely influenced the O₃-mortality effect estimates in the single-pollutant models.

In field studies, power to assess independent O₃ effects may be limited by small sample sizes and short follow-up times. Yet, the O₃ effect also was robust to the addition of copollutants in multipollutant models, with a few exceptions. For example, the effect of O₃ on PEF was not

robust to adjustments for PM_{2.5} and sulfate, in studies by Romieu et al. (1996) and Neas et al. (1999). In general, however, O₃ effects on respiratory symptoms (Romieu et al., 1996), lung function parameters (Brauer et al., 1996; Gold et al., 1999), and asthma medication use (Gent et al., 2003) were robust to inclusion of PM_{2.5}. Further, the effects for O₃ were observed to be stronger than those for PM.

Multipollutant regression analyses indicated that O₃ risk estimates, in general, were not sensitive to the inclusion of copollutants, including PM_{2.5} and sulfate. These results suggest that the effect of O₃ on respiratory health outcomes appears to be robust and independent of the effects of other copollutants.

7.6.5 Concentration-Response Function and Threshold

An important consideration in characterizing the public health impacts associated with O₃ exposure is whether the concentration-response relationship is linear across the full concentration range or instead shows nonlinearity. Of particular interest is the shape of the concentration-response curve at and below the level of the current 8-h standard of 80 ppb. There are limitations to identifying possible “thresholds” in air pollution epidemiologic studies, including difficulties related to the low data density in the lower concentration range, possible influence of measurement error, and individual differences in susceptibility to air pollution health effects. Despite these difficulties, the slope of the O₃ concentration-response relationship has been explored in several studies.

To examine the shape of the concentration-response relationship between O₃ and mortality, Gryparis et al. (2004) used meta-smoothing to combine smooth curves across the 23 European cities in a hierarchical model. The mean 1-h max O₃ concentrations ranged from 44 ppb in Tel Aviv to 117 ppb in Torino, Italy during the summer. For the summer period, while the estimated concentration-response curve did not appear to deviate significantly from linearity, there were indications of decreasing effects at lower exposures.

In the U.S. 95 communities study (Bell et al., 2004), effect estimates calculated using only days with 24-h avg O₃ levels less than 60 ppb were compared to those using all data. At a lag of 1 day, O₃ was associated with an excess risk of 0.36% (95% PI: 0.12, 0.60) per 20 ppb increase in 24-h avg O₃ using data from all days and only a slightly smaller risk of 0.30% (95% PI: 0.08, 0.54) when data were limited to days less than 60 ppb. These results suggest that if there is a

threshold, it must be notably lower than a 24-h avg O₃ of 60 ppb. In a more recent study by Bell et al. (2006; published online January 23, 2006), the shape of the concentration-response curve for the O₃-mortality relationship was evaluated in 98 U.S. urban communities for the period 1987 to 2000. The measure of exposure was the average of the same and previous days' ambient 24-h avg O₃ concentration. Analytical methods included linear, subset, threshold, and spline models. Results from all methods indicated that if a threshold did exist, it would have to be at low concentrations, less than a 24-h avg O₃ of 15 ppb.

Fairley (2003) reanalyzed the Santa Clara County mortality data using GAM with stringent convergence criteria and examined a new exposure index for O₃. He noted O₃ concentrations exceeding 60 ppb each hour and calculated a daily sum of these exceedances. Fairley's index incorporates measures of concentration and exposure duration; thus, this index represents a linear time-integrated concentration, also known as dosage. The O₃ index with the 60 ppb "threshold level" was found to be significantly associated with mortality in single-pollutant models as well as in multipollutant models. Two other "threshold levels" were examined, 40 ppb and 80 ppb. Both produced statistically significant results in single-pollutant models. These results suggest that the threshold for O₃-mortality effects, if it exists, is likely less than 40 ppb. The implication for thresholds in terms of the three standard indices (i.e., 1-h max, 8-h max, and 24-h avg) is unclear, but there may be an empirical relationship.

Vedal et al. (2003) observed that the annual mean 1-h daily max O₃ concentration of 27.3 ppb (SD 10.2) in Vancouver, Canada, was lower than that in any of the 80 NMMAPS cities (Samet et al., 2000); thus, a Vancouver study might provide a better focus on the shape of the O₃ concentration-response curve at lower levels. An O₃ effect was observed on total mortality at a 0-day lag during the summer. Ozone effects on respiratory mortality at a 2-day lag and cardiovascular mortality at a 0-day lag also were observed in the summer in this study. The effect of O₃ on mortality was robust in two-pollutant models. Vedal et al. (2003) concluded that O₃ concentrations were associated with adverse effects on mortality even at low levels. Although this study appears to support the argument that there are no threshold concentrations below which adverse effects cannot be detected, the results must be interpreted with caution as concerns remain. Vedal et al. (2003), for example, questioned if O₃, other gaseous pollutants, and PM may be acting as surrogate markers of pollutant mixes that contain more toxic compounds, since the low measured concentrations were unlikely, in their opinion, to cause the

observed effects. They further stated that measurement error and interference by meteorological factors might have contributed to the inability to detect a threshold.

Kim et al. (2004) investigated the presence of a threshold in O₃-mortality effects in Seoul, Korea by analyzing data using a log linear GAM (linear model), a cubic natural spline model (nonlinear model), and a B-mode splined model (threshold model). Models were stratified by season and adjusted for PM₁₀, long-term time trend, and meteorological variables. The mean 1-h max O₃ levels were 46.9 ppb (SD 22.5) during the summer and 21.3 ppb (SD 6.9) during the winter. Threshold values were estimated as 1-h max O₃ levels of 28 ppb when using all-year data and 45 ppb for summer-only data. None of the other pollutants examined, including PM₁₀, SO₂, NO₂, and CO, had a nonlinear association with mortality. Using summer-only data, the B-spline model resulted in an excess mortality risk of 7.1% (95% CI: 3.1, 11.2) per 40 ppb increase in 1-h max O₃, compared to an excess risk of 3.6% (95% CI, 0.5, 6.8) calculated using the log linear model. If a threshold truly exists, results from the Kim et al. (2004) study suggest that the use of log-linear models may underestimate the O₃ effect on mortality at levels above the threshold.

Other studies examining the effect of O₃ on mortality also have found suggestive evidence for a possible threshold level. In a London, England study (Anderson et al., 1996), an adjusted O₃-mortality bubble plot suggested that a threshold might exist around 50 ppb for 8-h avg O₃. A study by Simpson et al. (1997) in Brisbane, Australia observed a significant excess risk in mortality only in the highest quintile of O₃ exposure, which had a mean concentration of 42 ppb for 1-h max O₃.

Among several studies with morbidity outcomes, examination of the shape of the concentration-response function indicated evidence of an effect threshold. In a study of all-age respiratory hospital admissions in Toronto, Canada, effects of O₃ appeared to become apparent only above an approximate 30 ppb daily 1-h max O₃ (Burnett et al., 1997b). Also, Ponce de Leon et al. (1996) reported an indication of a threshold in the O₃ effect on hospitalizations in London, England at 40 to 50 ppb for 8-h max O₃ and 50 to 60 ppb for 1-h max O₃. In a study of emergency department visits for asthma in St. John, Canada, effects observed in the over 15 years age group were apparent only when data above the 95th percentile (75 ppb daily 1-h max O₃) were included (Stieb et al., 1996). However, other morbidity studies observed a monotonic increase in the concentration-response function, suggesting that there was no

threshold in O₃ effects on hospitalizations and emergency department visits (Burnett et al., 1997a; Jaffe et al., 2003; Petroeschevsky et al., 2001; Tenías et al., 1998).

In a field study by Mortimer et al. (2002), the associations of ambient O₃ levels with PEF and asthma symptoms were investigated in eight urban cities in the United States. The mean 8-h avg O₃ was 48 ppb, with less than 5% of days exceeding 80 ppb. Analysis performed using all data indicated that a 15 ppb change in 8-h avg O₃ was associated with decrements in PEF (-0.59% [95% CI: -1.05, -0.13]) and increased incidence of respiratory symptoms (odds ratio of 1.16 [95% CI: 1.02, 1.30]) over multiday lag periods. When data were restricted to days when ambient O₃ concentrations were less than 80 ppb, the O₃ effects persisted, with a significant PEF decline (-0.70% [95% CI: -1.29, -0.12]) and incidence of morning symptoms (odds ratio of 1.17 [95% CI: 1.01, 1.35]). A study by Chen et al. (1999) also found that there was no clear threshold in the O₃ effect on FEV₁ and FVC in Taiwanese school children. The 1-h max O₃ concentration ranged from 19.7 to 110.3 ppb.

The studies of both Brauer et al. (1996) and Korrick et al. (1998) demonstrate that exposure duration and exercise level, in addition to O₃ concentration, must be considered when evaluating thresholds. In the study by Brauer et al., the mean O₃ concentration during the 11-hour work shift was 26.0 ppb (SD 11.8). Workers experienced a change of -180.0 mL (95% CI: -227.0, -133.0) in FEV₁ levels the next morning per 40 ppb increase in 1-h max O₃. The hikers in the study by Korrick et al. (1998) were exposed to mean O₃ levels of 40 ppb (SD 12) over the duration of their hike (mean 8 hours). Korrick et al. observed a mean change of -62.5 mL (95% CI: -115.3, -9.7) in pre-hike to post-hike FEV₁ per 30 ppb increase in 8-h avg O₃ when all hikers were included in the analysis; however, when analysis was restricted to hikers with wheeze or asthma, a larger change of -182.5 mL (95% CI: -312.2, -52.9) was observed. In both studies, large reductions in lung function were observed in subjects exposed to relatively low levels of O₃ over multiple hours while active outdoors.

Other studies that may provide information concerning the concentration-response relationship are those that reported larger excess health risks in the warm (or summer) season than in the cold (or winter) season (as discussed in Section 7.6.3.2). During the cold season, when O₃ concentrations are generally low, no consistent O₃ effects were observed across the various health outcomes. These results appear to provide epidemiologic evidence of the presence of a threshold in O₃ health effects. However, other factors also may contribute to the

observed lack of an association between O₃ and health outcomes in the cold season. First, potential confounding by copollutants may mask the effect of O₃ during the cool season. For example, O₃ levels have been found to be inversely related to PM concentrations during the winter in some cities, e.g., New York City and Philadelphia (Ito et al., 2005). If PM is also associated with the health outcome of interest, inadequate control of confounding of the PM effect would result in negative or null associations for O₃ during the winter. In the few mortality studies that adjusted for PM in season-stratified analyses (see Figure 7-23), the cool-season O₃ risk estimates all increased slightly with the adjustment of PM indices; however, it should be noted that none achieved statistical significance. A second possible reason for the difference in O₃ effects observed by season may be the changing relationship between O₃ concentrations and personal exposure across seasons. The ambient O₃ levels are lower in the cold season, but people are likely to be exposed to even lower levels of O₃ during this season because of the shorter time spent outdoors and the longer time spent indoors with closed windows. Sarnat et al. (2005) observed that a 1 ppb increase in ambient O₃ concentration was associated with a 0.27 ppb (95% CI: 0.18, 0.37) increase in person O₃ exposure during the summer and 0.04 ppb (95% CI: 0.00, 0.07) increase during the winter.

A more “representative” concentration-response relationship may need to be examined in a summer-only data set as personal exposures tend to be better represented by ambient O₃ concentrations during that season. Even for summer data, however, an interpretation of the relationship is not straightforward because of the possible influence of the use of air conditioning (an effective remover of O₃). Greater use of air conditioning is expected on hot days when the O₃ level is higher, but the use of air conditioning may also vary from city to city and across social class within a city. Brauer et al. (2002) observed that surrogate measures of exposure (i.e., those from centrally-located ambient monitors) that were not highly correlated with personal exposures obscured the presence of thresholds in epidemiologic studies at the population level, even if a common threshold exists for individuals within the population.

Obscuring of thresholds would be even greater if thresholds varied across individuals (Brauer et al., 2002). Effects occur at the molecular level and are eventually manifested at the clinical level. One possible molecular mechanism for adverse physiologic effects from O₃ is its reaction with the carbon double bonds of lipids in the lung lining fluid (Levy et al., 2001). The molecular effects of O₃ are partially mitigated by the antioxidant defense system, which varies in

individuals with age, diet, genetic factors, and disease status. This raises the possibility that there may be a level below which O₃ would have few or no adverse effects as well as contributing to individual variability in the threshold level. The antioxidant defense system is one of many factors that contribute to the large intersubject variability in lung function responses to O₃ reported in human clinical studies (discussed in Sections 6.4 and 6.5). As individual sensitivities to O₃ health effect vary, a common threshold may not be observable at the population level in epidemiologic studies.

As noted earlier, demonstrating a clear threshold at a population level is difficult due to low data density in the lower O₃ concentration range, measurement error resulting from person-to-person differences in the relationship between personal exposures and monitored ambient concentrations, and individual differences in susceptibility to air pollution health effects. From 1990 to 2004, the 10th percentile values (which represent the lower concentration range) of the nationwide mean daily 8-h max O₃ concentrations were approximately 40 ppb during the warm season (May to September) (see Figure 3-17 in Section 3.2). While no fully confident conclusion can be made regarding the threshold issue from epidemiologic studies alone, the limited currently available evidence suggests that if a population threshold level exists in O₃ health effects, it is likely near the lower limit of ambient O₃ concentrations in the United States.

7.6.6 Heterogeneity of Ozone Health Effects

As described in Chapter 3 of this AQCD, O₃ concentrations tend to be more spatially variable than PM_{2.5} concentrations in urban areas. In addition, relative personal exposures to O₃ likely vary by region. The geographic variability in O₃ concentrations and personal exposures may contribute to the heterogeneity in observed O₃ health effects. The degree of influence of the geographic variability on heterogeneity in effects will vary by study, as study design affects different aspects of exposure (e.g., time period and duration of exposure).

More than 80% of the O₃-mortality estimates from the various studies conducted in North America, South America, Europe, and Australia were between 0.5 and 5% excess risk per 40 ppb increase in 1-h max O₃ using year-round data. In general, the O₃-mortality estimates were greater when using summer only data compared to year-round data. Though not all statistically significant, most of the O₃-mortality estimates were greater than zero, indicating a positive relationship between O₃ exposure and mortality. The O₃ risk estimates from the numerous

hospitalization and emergency department visit studies were generally larger in magnitude and more variable from study to study compared to the mortality studies. These differences in the O₃ effect estimates may be attributable to the greater variability in the outcome measure in hospitalization studies compared to mortality studies, as a result of the use of more subcategories of outcome and varying degrees of severity.

Three recent meta-analyses that included both U.S. and non-U.S. studies found consistent all-year combined point estimates: 1.75% (95% PI: 1.10, 2.40), 1.6% (95% CI: 1.1, 2.0), and 1.64% (95% CI: 1.25, 2.03) per 20 ppb increase in 24-h average O₃, for Bell et al. (2005), Ito et al. (2005), and Levy et al. (2005), respectively. Bell et al. further observed that the pooled estimate for U.S. studies (11 estimates), 1.69% (95% PI: 0.94, 2.78), was similar to the pooled estimate for the non-U.S. studies (30 estimates), 1.85% (95% PI: 0.94, 2.78). Levy et al. compared North American studies to European studies and also found nearly identical effect estimates.

As differences in study design, population, and data analysis may affect risk estimates, studies that were conducted in multiple cities using standardized methods were further examined to investigate the geographic heterogeneity of O₃ effects. Bell et al. (2004) conducted a time-series analysis of O₃ and mortality in 95 U.S. communities from 1987 to 2000. A 20-ppb increase in 24-h avg O₃ levels in the previous week was associated with an increase of 1.04% (95% PI: 0.54, 1.55) excess risk of mortality in the pooled analysis of 95 communities using all available data. The median 24-h avg O₃ concentrations varied from 14.4 ppb in Newark, NJ to 37.3 ppb in Bakersfield, CA. Intercommunity heterogeneity in O₃ effects was observed among the 95 communities (see Figure 7-17 of Section 7.4.3), which the authors noted as plausible given the community-specific differences in pollution characteristics, the use of air conditioning, time-activity patterns, and socioeconomic factors. One factor that appears to explain some of the intercity heterogeneity in the 95 U.S. communities study is long-term O₃ concentration. A weighted regression of the community-specific mortality risk from acute O₃ exposure and long-term average 24-h avg O₃ was significant ($p = 0.01$). These results indicated a higher excess risk of mortality per incremental change in 24-h avg O₃ in the previous week was observed in communities with lower long-term average O₃. Note that this analysis only includes the 40 communities with warm-season data. This relationship can be explained in a three parameter Gompertz concentration-response model of the cumulative percent excess risk in

mortality by long-term average 24-h avg O₃ levels (Figure 7-33). The method of estimating this curve is discussed in Annex Section AX7.3 along with limitations to inference concerning a threshold. The curve has shallower slopes above the median effective concentration (inflection point) than below that concentration point. The horizontal line is the 95% confidence interval on the long-term average 24-h avg O₃ associated with a 0.05% excess risk in mortality (EC_{0.05%}). The EC_{0.05%} is a risk that is believed to be low enough to be in the noise band. The Gompertz model presents a threshold of effects at low concentrations and a flat curve at higher concentrations. Thus, to the extent that the estimated concentration associated with a 0.05% excess risk is realistic, the fit of the model supports the observed relationship between greater mortality risk per incremental change in acute O₃ exposure at lower long-term O₃ levels.

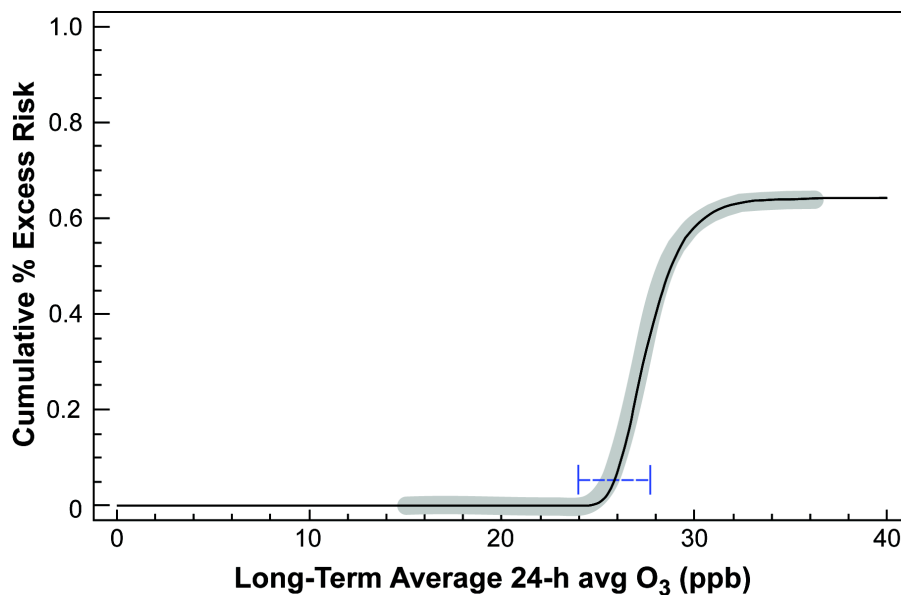


Figure 7-33. The fitted Gompertz model of the cumulative percent excess risk in mortality by long-term average 24-h avg O₃ concentrations using data from the 95 U.S. communities study (Bell et al., 2004). Only the 40 communities with warm-season data are included. Highlighted is the actual data range of the long-term average 24-h avg O₃ concentrations. The horizontal line is the 95% CI on the long-term average 24-h avg O₃ associated with a 0.05% excess risk in mortality (EC_{0.05%}).

Greater heterogeneity was observed in the European study of 23 cities in 14 countries (Gryparis et al., 2004). In the year-round analyses, only 8 of the 23 cities had positive O₃-mortality effect estimates. However, in the analyses using summer-only data, the risk estimates were positive in 19 of the 23 cities, with a range of 0.8 to 8% excess risk per 40 ppb increase in 1-h max O₃. Median 1-h max O₃ concentrations ranged from 44 ppb in Tel Aviv, Israel to 117 ppb in Torino, Italy during the summer. The heterogeneity may be attributable to the considerable variability among countries in factors that may influence the relationship between ambient O₃ concentrations and personal exposure to O₃, such as climate, use of air conditioning, personal activity patterns, and socioeconomic factors. In addition, the variability in the concentration and composition of copollutants by cities or countries may contribute to the heterogeneity in the O₃-mortality effects. For example, concentrations of NO₂ may vary widely by region, depending on the differences in traffic density.

Among the hospitalization studies, Burnett et al. (1997a) conducted the largest study of 16 Canadian cities. The mean daily 1-h max O₃ was 31 ppb (range 26-38) in the 16 cities. The pooled O₃ estimate was 5.6% (95% CI: 3.4, 7.9) excess risk in respiratory hospitalization per 40 ppb increase in 1-h max O₃ using warm-season data (April to December). The risk estimates were fairly homogenous across the 16 Canadian cities, ranging from 3.1% for Vancouver to 7.7% for Quebec City.

Anderson et al. (1997) investigated the association between O₃ and hospital admissions for COPD in five European cities (London, Paris, Amsterdam, Rotterdam, and Barcelona). The pooled effect estimate was 5.0% (95% CI: 2.6, 7.6) excess risk per 30 ppb increase in 8-h max O₃ for year-round data. Results from the APHEA study showed similar variability to that from the Burnett et al. (1997a) study. The year-round effects estimates were lower in the two Dutch cities (2.5% excess risk) compared to that in Paris (7.7% excess risk); however, analyses indicated that there was no significant heterogeneity in effects by city. The authors further noted that among the pollutants examined (O₃, BS, TSP, SO₂, and NO₂), O₃ had the most consistent and significant findings.

Among the field studies, various respiratory health outcomes were examined, including PEF, spirometric parameters, respiratory symptoms, and medication use. Only one field study investigated the O₃ effect in several locations (Mortimer et al., 2002). Mortimer et al. (2002) investigated the association of ambient O₃ concentrations with PEF and asthma symptoms in

asthmatic children living in eight urban cities in the United States: St. Louis, MO; Chicago, IL; Detroit, MI; Cleveland, OH; Washington, DC; Baltimore, MD; East Harlem, NY; and Bronx, NY. In the analysis pooling data from all eight cities, a 30 ppb increase in 8-h avg O₃ was associated with a decrement of -1.18% (95%CI: -2.10, -0.26) in morning PEF for a 5-day cumulative lag period. The percent changes in PEF were negative in all cities except for Baltimore, 0.49%. Among the other seven cities, the percent changes in PEF were quite homogenous, with values ranging from -1.08% for Washington, DC to -1.71% for St. Louis. A 30 ppb increase in 8-h avg O₃ also was associated with an increased incidence of morning symptoms in the pooled analysis (odds ratio of 1.35 [95% CI: 1.04, 1.69] for a 4-day cumulative lag period). In all cities except for St. Louis, there was an increase in the incidence of morning symptoms. In these cities, the odds ratios for incidence of morning symptoms were somewhat more varied than for the PEF measurements, ranging from 1.19 for Chicago to 2.96 for Detroit.

Most of the multicity and meta-analyses studies consistently found positive associations between O₃ and mortality. Consistent O₃ effects on hospitalizations and various respiratory health outcomes also were found. The observed heterogeneity of O₃ effects may be partially attributable to the use of centrally-located ambient monitors to assess exposure. There may be differences in relative personal exposures to O₃ due to varying factors, such as use of air conditioning and activity patterns, that affect the relationship between personal exposure and ambient concentrations. For example, Levy et al. (2005) found suggestive evidence that air conditioning prevalence was a predictor of heterogeneity in O₃ risk estimates in their meta-analysis. The variability in the concentration and composition of other pollutants present also may contribute to the heterogeneity of the effect of O₃ on health outcomes as confounding by copollutants may vary by region.

7.6.7 Health Effects of Ozone in Susceptible and Vulnerable Populations

In this section, the effects of O₃ on morbidity and mortality in potentially susceptible and vulnerable populations will be examined. In epidemiologic studies of O₃ health effects, the most widely studied subpopulation was asthmatics. Also of interest were the observed health effects of O₃ on different age groups, particularly children and the elderly. Other groups that are vulnerable to O₃ health effects are those that spend a lot of time outdoors at higher exertion

levels, such as outdoor workers. This section begins with a discussion of the O₃-related health effects in asthmatics.

7.6.7.1 Health Effects Associated with Ambient Ozone Exposure in Asthmatics

Epidemiologic studies of health effects from acute O₃ exposure in asthmatics have examined a range of outcomes: pulmonary function, respiratory symptoms, inflammation, emergency room visits, hospital admissions, and mortality. Chronic O₃ exposure studies have investigated similar outcomes, with the exception of emergency room visits and hospitalizations. Both are discussed in the earlier text. This subsection draws together this information to examine whether the evidence indicates that O₃ exposure impacts asthmatics.

In Germany and Mexico City, O₃ exposure was associated with a decline in FEV₁ in asthmatic adults and children (Höppe et al., 1995a, 2003; Romieu et al., 2002). Change in FEV₁ also was examined in a group of asthmatic hikers in Mount Washington, NH (Korrick et al., 1998). The mean hourly O₃ concentration during each hike was 40 ppb (range 21-74). Compared to the healthy subjects, the asthmatic subjects experienced a 4-fold greater decline in FEV₁ with the same exposure to O₃ (mean change of -1.08% [95% CI: -2.49, 0.33] versus -4.47% [95% CI: -7.65, -1.29] per 30 ppb increase in 8-h avg O₃). The results from the hiker study are consistent with those observed in a controlled human exposure study, which observed an approximately 2-fold greater decrement in FEV₁ among mild-to-moderate asthmatics versus nonasthmatic subjects performing light exercise during a 7.6-h exposure period at 0.16 ppm O₃ (Horstman et al., 1995).

PEF was examined in panels of asthmatic children in several field studies (see Figures 7-1 and 7-2). Collectively, most of the studies indicated decrements of morning PEF, though only a few estimates were statistically significant. One multicity study of eight urban areas in the United States observed O₃-related reductions in morning PEF that were not significant in each individual city (Mortimer et al., 2002); however, the analysis combining data from all eight cities indicated a significant decline in PEF with a cumulative lag of 1 to 5 days of O₃ exposure. The median 8-h avg O₃ levels ranged from 34 to 58 ppb across the eight cities. The odds ratio for the incidence of ≥10% decline in morning PEF was greater than one, which was discussed by the author as an indication that O₃ exposure might be associated with clinically important changes in

PEF in asthmatic children. The study examined 846 asthmatic children, the largest asthma panel study reported.

Mortimer et al. (2000) observed that the subpopulation of asthmatic children with a history of low birth weight or premature birth had greater O₃-associated declines in PEF (mean change of -3.66% [95% CI: -5.30, -2.02] per 30 ppb increase in 8-h avg O₃) than normal birth weight children (-0.60% [95% CI: -1.58, 0.38]). Low birth weight and prematurity are associated with reduced lung function, higher levels of airway reactivity and increased susceptibility to lung damage (Barker et al., 1993; Rona et al., 1993), which may explain why these factors are found to increase susceptibility to respiratory insults of air pollution in children.

Lung function parameters have been evaluated for clinical significance. A reversible 5 to 15% decline in FEV₁ in an individual may have clinical importance to asthma morbidity (American Thoracic Society, 1991; Lebowitz et al., 1987; Lippmann, 1988). The National Institutes of Health (1997) has stated that PEF below 80% of the personal best indicates a need for additional medication use in asthmatics. At a population level the mean changes in lung function attributable to O₃ exposure do not generally exceed 10% changes in FEV₁ or PEF per standardized increment of O₃. At an individual level, a subpopulation of susceptible asthmatics are likely experiencing clinically significant declines in lung function. Höpfe et al. (2003) investigated the effects of O₃ on the lung function of potential risk groups, including asthmatics, children, athletes, and the elderly. The mean afternoon (1 p.m.-4 p.m.) ½-h max O₃ levels on high O₃ days were 66.9 ppb (range 51-91) for the asthmatics and 65.2 ppb (range 43-88) for the children. The afternoon ½-h max O₃ levels in these two groups were similar to those experienced by the elderly and athletes. Consistent associations between O₃ and group means of lung function endpoints were not observed for the various risk groups. However, a potential pattern of O₃ sensitivity was observed when individual data was examined. About 20% of the asthmatics and children were regarded as O₃ responders (i.e., individuals with >10% change in FEV₁) compared to only 5% of the elderly and athletes. In these responders, a significant O₃ concentration-response relationship was observed in the regressions using repeated measurements from individuals. These results indicated that while the population as a whole was not reacting to O₃, susceptible individuals were experiencing clinically significant declines in lung function in response to O₃ exposure.

Respiratory symptom increases in asthma panels were examined in several field studies, some of which also examined PEF as discussed above. The outcome definition of symptoms varied among these studies. Collectively, the results are suggestive of a potential O₃ effect on respiratory symptoms, but the evidence is not strong in the available studies. Two U.S. studies that examined larger panels might be better to draw inferences from as the large sample size provided greater power to examine the effect of O₃ on respiratory symptoms. The eight U.S. urban cities study reported that morning symptoms in the 846 asthmatic children were most strongly associated with a 4-day cumulative lag of O₃ concentrations (Mortimer et al., 2002). A New England study examined 271 asthmatic children and observed an O₃ effect on a variety of respiratory symptoms at a lag of 1 day among the 130 subjects who used maintenance asthma medications (Gent et al., 2003). The mean 1-h max O₃ was 58.6 ppb (SD 19.0).

Only a few epidemiologic studies have examined airway inflammation in asthmatics. A Mexico City study indicated that supplementation with antioxidants may modulate the impact of O₃ exposure on the small airways of children with moderate-to-severe asthma (Romieu et al., 2002). The mean 1-h max O₃ was 102 ppb (SD 47). A related study indicated that asthmatic children with GSTM1 null genotype were found to be more susceptible to the impact of O₃ exposure on small airways (Romieu et al., 2004).

Emergency department visits for asthmatics have been examined in several studies and range from negative to positive results (see Figure 7-8 in Section 7.3.2). Studies of mostly year-long data tended to produce inconsistent results, with some finding negative estimates (Atkinson et al., 1999a; Castellsague et al., 1995; Thompson et al., 2001; Tobías et al., 1999). Warm-season studies tended to yield positive outcomes, as expected based on earlier discussions. Two studies in Atlanta, GA (Tolbert et al., 2000) and Valencia, Spain (Tenías et al., 1998) indicated positive effects in warm-season analyses. Further, a Canadian study, one of the larger studies conducted in the summertime, reported a large increase in asthma emergency department visits when the daily 1-h max O₃ concentration exceeded 75 ppb (Stieb et al., 1996). A three-city study in Ohio also indicated an increased risk of asthma visits during the summer (Jaffe et al., 2003). The mean 8-h max O₃ levels ranged from 50 to 60 ppb in the three cities.

Hospital admission studies that specifically examined asthmatics were fewer in number than those that examined total respiratory diseases. Associations were noted in all age groups in studies conducted in Seattle, WA (Sheppard, 2003), New Jersey (Weisel et al., 2002), Toronto,

Canada (Burnett et al., 1999), London, England (Anderson et al., 1998), Brisbane, Australia (Petroeshevsky et al., 2001), and Hong Kong (Wong et al., 1999a). However, several other studies, mostly examining the effect of O₃ on asthmatic children, did not observe a significant relationship (Gouveia and Fletcher, 2000a; Lin et al., 2003; Morgan et al., 1998; Nauenberg and Basu, 1999; Schouten et al., 1996).

Acute mortality related to asthma was examined in Barcelona, Spain (Saez et al., 1999; Sunyer et al., 2002). In the study by Sunyer et al. (2002), severe asthmatics with more than one asthma emergency visit were found to have the strongest mortality associations with O₃. The median 1-h max O₃ level was 35.8 ppb (range 3.4-146.1).

Recent reports from longitudinal cohort studies in California have reported associations between the onset of asthma and long-term O₃ exposures (Greer et al., 1993; McConnell et al., 2002; McDonnell et al., 1999). In adult studies, associations were seen in males but not females (Greer et al., 1993; McDonnell et al., 1999). Among children residing in high O₃ communities (mean 8-h avg O₃ of 59.6 ppb [range 55.8-69.0]), McConnell et al. (2002) observed that asthma risk was elevated for those who played three or more sports as compared with those who did not play sports. Playing sports may indicate outdoor activity and an increased ventilation rate which leads to increased dose of O₃. These outcomes would benefit from replication in other cohorts in regards to indicating weight of a causal interpretation.

A few studies provide limited discussion of concentration-response functions and thresholds. In the eight U.S. urban cities study of asthmatic children, the odds ratios for incidence of $\geq 10\%$ decline in morning PEF and incidence of morning symptoms when excluding days with 8-h avg O₃ greater than 80 ppb were nearly identical to those including data from all days (Mortimer et al., 2002). In the New England asthma panel study (Gent et al., 2003), some of the associations for symptoms occurred at 1-h max O₃ levels below 60 ppb. In the St. John, Canada study (Stieb et al., 2003), an effect of O₃ on emergency department visits was reported with evidence of a threshold somewhere in the range below a 1-h max O₃ of 75 ppb in the 15 years and over age group.

Overall, subjects with asthma have been examined across most health endpoints of interest. The results reported in these studies vary with some indicating a positive excess risk associated with O₃. While no endpoint in itself seems to indicate an unquestionable demonstration of an association, studies with adequate sample size and power consistently provide positive results,

especially during the summer months when higher O₃ levels occur. This view is strengthened as positive results are obtained cohesively across the varied outcomes. Therefore, based on the evidence, it seems prudent to consider asthmatics as a susceptible group that requires more protection from O₃ exposures than the general public.

7.6.7.2 Age-Related Differences in Ozone Effects

The American Academy of Pediatrics (2004) notes that children and infants are among the most susceptible to many air pollutants, including O₃. Eighty percent of alveoli are formed postnatally and changes in the lung continue through adolescence; the developing lung is highly susceptible to damage from exposure to environmental toxicants (Dietert et al., 2000). Children also have increased vulnerability as they spend more time outdoors, are highly active, and have a high minute ventilation, which collectively increase their dose (Plunkett et al., 1992; Wiley et al., 1991a,b). In addition to children, the elderly are frequently classified as being particularly susceptible to air pollution. The basis of the increased sensitivity in the elderly is not known but one hypothesis is that it may be related to changes in the respiratory tract lining fluid antioxidant defense network (Kelly et al., 2003).

Several mortality studies have investigated age-related differences in O₃ effects. Among the studies that observed positive associations between O₃ and mortality, a comparison of all age or younger age (≤ 65 years of age) O₃-mortality risk estimates to that of the elderly population (>65 years) indicates that, in general, the elderly population is more susceptible to O₃ effects (Borja-Aburto et al. 1997; Bremner et al., 1999; Gouveia and Fletcher 2000b; O'Neill et al., 2004; Simpson et al., 1997; Sartor et al., 1995; Sunyer et al., 2002). For example, a study by Gouveia and Fletcher (2000b) examined the O₃-mortality effect by age in São Paulo, Brazil. The mean 1-h max O₃ level was 35.1 ppb (SD 21.7). There were 151,756 deaths for all non-violent causes over the period of 1991 to 1993, of which 49% occurred in the elderly. Among all ages, O₃ was associated with a 0.6% (95% CI: -0.8, 2.0) excess risk in all cause mortality per 40 ppb increase in 1-h max O₃. In comparison, in the elderly population, the O₃-mortality risk estimate was nearly 3-fold greater, 1.7% (95% CI: 0.0, 3.3). Similarly, a Mexico City study found that O₃-mortality risk estimates were 1.3% (95% CI: 0.04, 2.6) and 2.8% (95% CI: 1.0, 4.6) per 20 ppb increase in 24-h avg O₃ concentration in all ages and the elderly, respectively (O'Neill et al., 2004). The mean 24-h avg O₃ level was 35.3 ppb (SD 11.0).

The meta-analysis by Bell et al. (2005) found a larger effect estimate for the elderly (2.92% [95% PI: 1.34, 4.51] per 20 ppb increase in 24-h avg O₃) than for all ages (1.75% [95% PI: 1.10, 2.37]). In the large U.S. 95 communities study (Bell et al., 2004), effect estimates were slightly higher for those aged 65 to 74 years, 1.40% (95% PI: 0.56, 2.25) excess risk per 20 ppb increase in 24-h avg O₃, compared to individuals less than 65 years and 75 years or greater, 1.00% (95% PI: 0.20, 1.85) and 1.04% (95% PI: 0.36, 1.75), respectively, using a constrained distributed 7-day lag model. Bell et al. (2004) notes that despite somewhat similar effect estimates, the absolute effect of O₃ is substantially greater in the elderly population due to the higher underlying mortality rates, which leads to a larger number of extra deaths for the elderly compared to the general population.

A few mortality studies examined another potentially susceptible age group, young children under the age of 5 years. The results were mixed, with one Mexico City study showing a lower risk of O₃-related all cause mortality in young children compared to all ages and the elderly (Borja-Aburto et al., 1997) and one São Paulo, Brazil study showing a greater risk in respiratory mortality in young children compared to the elderly (Gouveia and Fletcher, 2000b). Another study in Mexico City by Loomis et al. (1999) observed a positive but nonsignificant association between ambient O₃ concentrations and infant mortality (used Poisson GAM with default convergence criteria). Only a limited number of studies have focused on air pollution effects on mortality in children using a time-series approach. This is probably because the numbers involved are usually not adequate for such an analysis. Approximately 10% of mortality occurs in young children. Gouveia and Fletcher (2000b) noted that in the case of São Paulo, the mean number of daily deaths for respiratory causes in children was 2.2, which was unlikely to provide statistical power to detect the effects of air pollution even if they existed. In addition, there are other competing causes of mortality in young children, especially those in developing countries, which are together more important than air pollution.

With respect to age-specificity of associations between O₃ and acute respiratory hospitalizations or emergency department visits, no clear pattern emerges from recent studies. Associations have been reported for all ages (Anderson et al., 1997; Burnett et al., 1995, 1997b, 1999; Weisel et al., 2002), adults or elderly (Burnett et al., 1997a; Delfino et al., 1997b, 1998b; Moolgavkar et al., 1997; Schwartz et al., 1996; Yang et al., 2003), and children (Burnett et al., 2001; Gouveia and Fletcher, 2000a; Lin et al., 1999; Pönkä and Virtanen, 1996; Tolbert et al.,

2000; Yang et al., 2003). Interestingly, studies that have examined effects in multiple age strata often have seen effects only in non-pediatric strata (Delfino et al., 1997b, 1998b; Stieb et al., 1996; Jones et al., 1995). Several studies that focused on children did not report significant O₃ effects, though in some cases these studies are limited by small size, inadequate control of seasonal patterns, or very low O₃ levels (Lierl and Hornung, 2003; Lin et al., 2003; Thompson et al., 2001). If O₃ is causally related to exacerbations of respiratory diseases leading to hospital usage, one would expect to see effects most prominently among children, for whom asthma is more prevalent and O₃ exposures may be greater. However, once again, other competing causes and the small numbers of hospitalizations in children likely limit the ability to examine O₃-related health effects.

A few field studies compared the effect of O₃ in different age groups. Korrick et al. (1998) examined changes in FEV₁ and FVC related to O₃ exposure in a group of hikers ranging in age from 18 to 64 years, and found that there was no association between O₃ responsiveness and age. Brauer et al. (1996), in a study of berry pickers aged 10 to 69 years, also observed that subject age was not significantly associated with O₃-related changes in lung function. However, a study by Höpfe et al. (2003) observed that children, but not seniors (69 to 95 years of age), experienced a decline in lung function associated with O₃ exposure. Approximately 20% of the children and juvenile asthmatics experienced a greater than 10% change in FEV₁, compared to only 5% of the elderly population. The results by Höpfe et al. are consistent with the diminishing responses to O₃ exposure with increasing age observed in clinical studies. The clinical studies by Drechsler-Parks (1995) and Bedi et al. (1989) found that subjects aged 56 to 89 years had markedly reduced responses to O₃ exposure compared to young adults.

Many field studies focused on the effect of O₃ on the respiratory health of school children. In general, children experienced decrements in pulmonary function parameters, including PEF, FEV₁, and FVC (Castillejos et al., 1995; Chen et al., 1999; Gielen et al., 1997; Gold et al., 1999; Jalaludin et al., 2000; Mortimer et al., 2002; Romieu et al., 1996; Thurston et al., 1997). Increases in respiratory symptoms (Delfino et al., 2003; Gold et al., 1999; Neas et al., 1995; Romieu et al., 1996, 1997; Thurston et al., 1997) and asthma medication use (Delfino et al., 1996; Just et al., 2002; Ostro et al., 2001; Thuston et al., 1997) also were observed in children. These respiratory health effects were largely observed in asthmatic children. Ozone-associated lung function declines were found in healthy children as well.

Collectively, there is supporting evidence of age-related differences in susceptibility to O₃ health effects. The elderly population (>65 years of age) appear to be at increased risk of O₃-related mortality and hospitalizations, and children (<18 years of age) experience other potentially adverse respiratory health outcomes with increased O₃ exposure. One epidemiologic study also found that the lung function response to O₃ exposure may be diminished in elderly populations; this finding is further supported by evidence from clinical studies.

7.6.7.3 Vulnerability of Outdoor Workers and Others Who Participate in Outdoor Activities to Ozone Health Effects

The health effects of O₃ on outdoor workers and others who participate in outdoor activities have been investigated in various field studies. The most common endpoint examined in these studies is lung function. These individuals are typically exposed to high doses of O₃ as they spend long hours outdoors often at elevated exertion levels. Their increased vulnerability to O₃ health effects has been noted in epidemiologic studies.

Brauer et al. (1996) repeatedly measured spirometric lung function before and after outdoor summer work shifts over 59 days on a group of 58 berry pickers (mean age 44 years [range 10-69]) in Fraser Valley, British Columbia, Canada. Outdoor work shifts averaged 11 hours in duration. The mean ambient 1-h max O₃ was 40.3 ppb (SD 15.2) over the study period. Heart rates during the work shift averaged 36% higher than resting levels, indicating elevated exertion levels while working outdoors. The lung function changes experienced by the workers in this study are large compared to those from other field studies (see Table 7-1b). The next morning FEV₁ declined by 6.36% (95% CI: 4.70, 8.02) per 40 ppb increase in 1-h max O₃. These results indicate that extended exposures to O₃ at elevated exertion levels produce more marked effects on lung function.

Mexico City outdoor street workers (n = 47) were repeatedly monitored for lung function changes at the end of the work shift over a two-month period (Romieu et al., 1998). Workers were exposed to outdoor ambient O₃ levels for a mean of 7.4 hours during the day. The mean 1-h max O₃ was 123 ppb (SD 40). Among those who had never taken an antioxidant supplement, same day O₃ concentrations were associated with decreases in afternoon FEV₁. A mean change of -3.55% (95% CI: -6.28, -0.82) was observed per 40 ppb increase in 1-h max O₃.

Höppe et al. (1995a) examined forestry workers ($n = 41$) for O_3 -related changes in pulmonary function in Munich, Germany. The average time they spent outdoors was not presented. Ventilation rates were estimated from the average activity levels. Their ventilation rates were elevated at 40 L/min (compared to 15 L/min in the control group). When comparisons were made between high O_3 days (mean $\frac{1}{2}$ -h max O_3 of 64 ppb) and low O_3 days (mean $\frac{1}{2}$ -h max O_3 of 32 ppb), 59% of the forestry workers experienced a remarkable decrement in lung function (i.e., at least a 20% increase in specific airway resistance or at least a 10% decrease in FEV_1 , FVC, or PEF) on high O_3 days. None experienced an improvement in lung function. A change of -56.0 mL (95% CI: $-118.4, 6.4$) in FEV_1 was observed per 40 ppb increase in $\frac{1}{2}$ -h max O_3 .

In addition to forestry workers, Höppe et al. (1995a) also monitored athletes ($n = 43$) in the afternoon following a two-hour outdoor training period. Athletes had a fairly high ventilation rate of 80 L/min. Compared to the forestry workers a smaller percentage of athletes experienced a remarkable decrement in lung function, 14%, on high O_3 days; 19% of the athletes actually showed an improvement. Overall, a significant change in FEV_1 was observed, -60.8 mL (95% CI: $-115.2, -6.4$) per 40 ppb increase in $\frac{1}{2}$ -h max, in the athletes. In a subsequent study, Höppe et al. (2003) reanalyzed the results of the athletes after stratifying the spirometric data by time of day (morning versus afternoon) and at different lag periods (lags of 0 to 2 days). The reanalysis indicated that O_3 -related decrements were observed only with the afternoon FEV_1 at a 0-day lag, -1.26% (95% CI: $-2.63, 0.10$) change in FEV_1 per 40 ppb increase in $\frac{1}{2}$ -h max O_3 .

A study by Korrick et al. (1998) also examined the effects of multihour O_3 exposures on adults exercising outdoors. A total of 530 hikers (mean age 35 years [range 18-64]) of Mount Washington, NH performed spirometry before and after hiking for a mean of 8 hours (range 2-12). The mean of the hourly O_3 concentrations during the hike was 40 ppb (range 21-74). After the hike, all subjects combined experienced a small mean decline of 1.53% (95% CI: 0.24, 2.82) in FEV_1 per 30 ppb increase in the mean of the hourly O_3 concentrations during the hike. Ozone-related changes in lung function parameters were estimated after stratifying by hiking duration. Subjects who hiked 8 to 12 hours ($n = 265$) experienced a 2.07% (95% CI: 0.36, 3.78) decline in FEV_1 per 30 ppb increase in the mean of the hourly O_3 concentrations; those who hiked 2 to 8 hours ($n = 265$) experienced a smaller decline of 0.99% (95% CI: $-0.72, 2.70$).

Each hour hiked, which may reflect dose, was associated with a decline of 0.3% ($p = 0.05$) in FEV₁, after adjusting for O₃.

The O₃ effect attributable to exercise in children ($n = 40$) was investigated in a Mexico City study (Castillejos et al., 1995). These children were chronically exposed to moderate to high levels of O₃. The mean ambient 1-h max O₃ level during the study period was 112.3 ppb, with a maximum value of 365 ppb. Spirometry was performed by the children before and after a one-hour intermittent exercise session outdoors. Children were repeatedly tested up to 8 times. A 0.48% (95% CI: 0.24, 0.72) decline in FEV₁ was experienced by the children after exercising outdoors. However, stratified analyses indicated that significant changes were observed only with higher quintiles of O₃ exposure. At the highest quintile of exposure (183-365 ppb), a -2.85% (95% CI: -1.30, -4.40) change in FEV₁ was observed postexposure. Therefore, children exercising outdoors when ambient O₃ levels were high experienced declines in pulmonary function despite the repeated daily exposure to moderate and high levels of O₃ in Mexico City.

In the southern California Children's Health Study, a total of 3,535 initially nonasthmatic children (ages 9 to 16 years at enrollment) were followed for up to 5 years to identify new-onset asthma cases (McConnell et al., 2002). Communities were stratified by pollution levels, with six high-O₃ communities (mean 1-h max O₃ of 75.4 ppb [SD 6.8] over four years) and six low-O₃ communities (mean 50.1 ppb [SD 11.0]). Asthma risk was not found to be higher for residents of the six high-O₃ communities versus residents of the six low-O₃ communities. However, within the high-O₃ communities, asthma risk was 3.3 (95% CI: 1.9, 5.8) times greater for children who played three or more sports as compared with children who played no sports. This association was absent in the low-O₃ communities (relative risk of 0.8 [95% CI: 0.4, 1.6]). Thus, similar to the results observed in the Mexico City study by Castillejos et al. (1995), greater effects were seen for susceptible individuals who exercised more outdoors at high O₃ levels.

The studies discussed above indicate that prolonged exposure periods, combined with elevated levels of exertion or exercise, may magnify the effect of O₃ on lung function. Results from these studies are consistent with the earlier summer camp studies (Avol et al., 1990; Higgins et al., 1990; Raizenne et al., 1987, 1989; Spektor et al., 1988a, 1991) which also indicated large O₃-related changes in lung function parameters in children who spent long hours outdoors. The large observed effects further suggest that other components in the ambient air

pollution mixture may potentiate the effects of O₃. In conclusion, outdoor workers who spend, on average, 8 to 10 hours outdoors daily during the time of day when high peak O₃ concentrations are expected appear to be particularly vulnerable to O₃ health effects and may need protection from O₃ exposures.

7.6.8 Summary of Key Findings and Conclusions Derived from Ozone Epidemiologic Studies

In the previous 1996 O₃ AQCD, there was considerable evidence of O₃-related respiratory health effects from individual-level camp and exercise studies, as well as some consistent evidence from time-series studies of emergency room visits and hospitalizations. Since the 1996 document, more field studies have been conducted, with some emphasis on additional outcome markers such as respiratory symptoms and asthma medication use. Another significant addition to the current O₃ AQCD is the substantial number of short-term O₃ mortality studies. The recent publication of an analysis examining the relationship between O₃ and mortality in 95 U.S. communities (Bell et al., 2004) and three meta-analysis on O₃-mortality associations (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005) also contribute significantly to the evidence base. Considering the wide variability in possible study designs and statistical model specification choices, the reported O₃ risk estimates for the various health outcomes are in reasonably good agreement. In the case of O₃-mortality time-series studies, combinations of choices in model specifications (e.g., the number of weather terms and degrees of freedom for smoothing of mortality-temporal trends) alone may explain the extent of the difference in O₃ risk estimates across studies. As use of time-series studies to investigate air pollution effects has become more common, there has been a great effort to evaluate the issues surrounding these studies.

The epidemiologic studies discussed in this chapter provide important information on the associations between health effects and exposures of human populations to ambient O₃. A variety of oxidants in both the gaseous and particulate phases have not been examined in the available epidemiologic literature. The associations observed between ambient O₃ concentrations and health outcomes may represent O₃ effects, per se, or O₃ may be serving as a surrogate measure for the overall ambient photochemical oxidant mix.

In this section, conclusions regarding O₃ health effects from the epidemiologic evidence and the issues that may affect the interpretation of the effect estimates are briefly summarized.

A more integrative synthesis of all relevant information will be presented in Chapter 8 of this AQCD.

- (1) Field/panel studies of acute O₃ effects. Results from recent field/panel studies continue to confirm that short-term O₃ exposure is associated with acute decrements in lung function and increased respiratory symptoms, particularly in children and asthmatics. There is also suggestive evidence that O₃ is related to increased asthma medication use. Strong O₃ effects on lung function also were observed in outdoor workers. Taken together with the evidence from controlled human exposure studies, O₃ is likely causally related to the various respiratory health outcomes. The current evidence is much more limited, though suggestive, of a potential effect of O₃ on heart rate variability, ventricular arrhythmias, and the incidence of myocardial infarctions.
- (2) Acute O₃ effects on emergency department visits and hospitalizations. Large multicity studies, as well as many studies from individual cities have reported an effect of O₃ on respiratory hospital admissions. Studies using year-round data noted some inconsistencies; however, studies with data restricted to the summer or warm season, in general, indicated positive and robust associations between ambient O₃ concentrations and respiratory hospital admissions. Effects of O₃ on asthma emergency department visits also were observed during the warm season.
- (3) Acute O₃ effects on mortality. The majority of the studies suggest an elevated risk of all-cause mortality associated with acute exposure to O₃, especially in the summer or warm season when O₃ levels are typically high. Slightly greater O₃ effects were observed for cardiovascular mortality. Results from recent large U.S. multicity time-series studies provide the strongest evidence to-date for O₃ effects on acute mortality. Recent meta-analyses also indicate positive risk estimates that are unlikely to be confounded by PM; however, future work is needed to better understand the influence of model specifications on the risk coefficient.
- (4) Chronic O₃ exposure effects on morbidity and mortality. Fewer studies have investigated the effect of chronic O₃ exposure on morbidity and mortality. The strongest evidence is for seasonal effects of extended O₃ exposures on lung function in children, i.e., reduced lung function growth being associated with higher ambient O₃ levels. Longer-term studies investigating the association of chronic O₃ exposure on yearly lung function, asthma incidence, and respiratory symptoms are inconclusive. Chronic O₃-mortality studies observed inconsistent results across exposure periods and cause-specific mortality outcomes.
- (5) Exposure assessment. Exposure misclassification may result from the use of stationary ambient monitors to determine exposure in population studies. Although central ambient monitors do not explain the variance of individual personal exposures,

significant correlations are found between aggregate personal O₃ measurements and O₃ concentrations from ambient monitors. A simulation study indicated that the use of ambient monitor data will tend to underestimate the O₃ effect. A better understanding of the factors that affect the relationship between ambient concentrations and personal exposures will improve interpretation of the O₃ effect estimates.

- (6) Ozone exposure indices. The three most commonly used daily O₃ exposure indices, 1-h max O₃, 8-max O₃, and 24-h avg O₃, were found to be highly correlated in studies conducted in various regions. In addition, the effect estimates and significance of associations across all health outcomes were comparable when using the standardized distributional increment of 40 ppb, 30 ppb, and 20 ppb for mean 1-h max O₃, mean 8-h max O₃, and mean 24-h avg O₃, respectively.
- (7) Lag structures for O₃ exposure and effect. The lag time between O₃ exposure and effect may differ depending on various factors such as the specific health outcome of interest, the mechanism of effect, and preexisting health conditions. The majority of the studies found an immediate O₃ effect, with the strongest associations observed between health outcomes and O₃ exposure on the same day and/or previous day. Some studies found large cumulative effects of O₃ over longer lag periods, indicating that multiday lags also may be relevant for some health outcomes, including mortality and asthma symptoms in children.
- (8) Sensitivity to model specifications for temporal trends. Ozone effect estimates that were reported in studies whose main focus was PM often were calculated using the same model specifications as PM to adjust for temporal trends. While the sensitivity of the O₃ risk estimates to alternative model specifications has not been thoroughly investigated, limited evidence indicates that O₃ effects may be robust to various model specifications for temporal trend adjustment.
- (9) Sensitivity to model specifications for meteorological effects. Ozone risk estimates were generally more sensitive to alternative weather models than to varying degrees of freedom for temporal trend adjustment. In studies in which alternative weather models (e.g., quintile indicator model and four-smoother model) were considered, up to a factor of two difference in the O₃ risk estimates was observed. Further research is needed to reduce the uncertainties related to confounding by weather influences.
- (10) Influence of seasonal factors. An evaluation of the confounding effects of meteorologic factors and copollutants on O₃ risk estimates is complicated by their changing relationships with O₃ across seasons. In addition, seasonal or seasonally-modified factors (e.g., air conditioning use, time spent outdoors) complicate interpretation of all-year effect estimates as they affect the relationship between

ambient concentrations and personal exposures. Given the potentially significant influence of season, season-specific analyses are more informative in assessing O₃ health risks.

- (11) Confounding by copollutants. Multipollutant regression models often are used to adjust for confounding by copollutants. Although there is some concern regarding the use of multipollutant models given the varying concurrency across pollutants, results generally suggest that the inclusion of copollutants into the models do not substantially affect O₃ risk estimates. These findings indicate that effects of O₃ on various health outcomes are robust and independent of the effects of other copollutants.
- (12) Concentration-response function. In the limited mortality and morbidity studies that have specifically examined the O₃ concentration-response relationship, the evidence is inconclusive regarding the presence of an effect threshold. Factors such as exposure measurement error may reduce the ability to detect a threshold in population studies. The limited evidence suggests that if a population threshold exists in O₃ health effects, it is likely near the lower limit of ambient O₃ concentrations in the United States.
- (13) Heterogeneity of O₃ health effects. Consistent O₃ effect estimates generally were observed for mortality, hospitalizations, and other respiratory health outcomes in multicity studies. Some of the observed geographic heterogeneity in effects may be attributable to the differences in relative personal exposure to O₃, which is affected by factors such as air conditioning prevalence and activity patterns, as well as varying concentrations and compositions of copollutants present by region.
- (14) Ozone health effects in asthmatics. The effects of O₃ on asthmatics have been examined widely in both time-series studies and panel studies. Associations of O₃ with various respiratory health outcomes, including lung function declines, increased respiratory symptoms, and emergency department visits, were observed. These findings, along with the pathophysiologic understanding of asthma as a chronic inflammatory disease, indicate that asthmatics are likely a susceptible population that requires protection from O₃ exposures.
- (15) Age-related differences in O₃ health effects. Supporting evidence exists for heterogeneity in the effects of O₃ by age. The elderly population (>65 years of age) appears to be at greater risk of O₃-related mortality and hospitalizations compared to all age populations. In addition, potentially adverse respiratory health outcomes are associated with O₃ exposure in children (<18 years of age). Lung function responses to O₃ exposure are diminished in the elderly population, as observed in an epidemiologic study and numerous human clinical studies.

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8. INTEGRATIVE SYNTHESIS: OZONE EXPOSURE AND HEALTH EFFECTS

8.1 INTRODUCTION

This integrative synthesis is structured to provide a coherent framework for the assessment of health risks associated with human exposures to ambient surface-level (tropospheric) ozone (O₃) in the United States. The main goal of the chapter is to integrate newly available scientific information with key findings and conclusions from the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996a), so as to address issues central to the EPA's assessment of evidence needed to support the current review of the primary O₃ NAAQS. The integrated assessment of key findings and conclusions provided here and elsewhere in this document with regard to O₃ exposure and health effects will be drawn upon and their policy implications considered in an Ozone Staff Paper prepared by EPA's Office of Air Quality Planning and Standards (OAQPS). The analyses provided in that Staff Paper aim to "bridge the gap" between scientific assessments in this criteria document and judgments required of the EPA administrator in evaluating whether to retain or, possibly, to revise the current primary O₃ NAAQS. Other types of scientific information concerning ambient O₃ welfare effects (i.e., tropospheric O₃ effects on vegetation and ecosystems, relationships to surface-level solar UV flux/climate changes, and effects on man-made materials) are assessed in ensuing Chapters 9, 10, and 11. That information will also be considered in the OAQPS staff paper in posing options regarding the secondary O₃ NAAQS.

As discussed in Chapter 2 of this document, O₃ found in the earth's troposphere generally originates from photochemical reactions that involve the interaction of sunlight with precursor pollutants, especially nitrogen oxides (NO_x), carbon monoxide (CO), and volatile organic compounds (VOCs) such as hydrocarbons emitted by surface-level mobile and stationary sources and natural sources. Other photochemical oxidants, such as peroxyacetyl nitrate (PAN) and hydrogen peroxide (H₂O₂), are also generated along with O₃ by such atmospheric interactions. In addition to the tropospheric O₃ generated by these interactions, some O₃ is found near the earth's surface as the result of its downward transport from the stratosphere. However, in contrast to stratospheric O₃, which plays an important role in maintaining the habitability of the planet by shielding the Earth's surface from harmful solar ultraviolet (UV) radiation,

tropospheric O₃ at the surface can also exert adverse effects on humans, nonhuman animal species, and vegetation. As was the case for previous O₃-related NAAQS criteria revisions, the present criteria document focuses mainly on the assessment of health and welfare effects resulting from exposures to surface-level concentrations of tropospheric O₃, whereas less attention is accorded to the distinctly much more limited available information on other photochemical oxidants, e.g., PAN or H₂O₂.

Based on the criteria review completed in 1978, the original primary and secondary NAAQS set in 1971 for total photochemical oxidants were revised in 1979 to focus on O₃ as the indicator for new primary and secondary standards that were attained when the expected number of days per calendar year with daily maximum 1-h average O₃ concentrations (1-h max O₃) >0.12 ppm did not exceed one. The NAAQS for ambient O₃ were revised in 1997 by replacing the 1-h standards with an 8-h primary standard that is met when the 3-year average of the annual fourth highest daily maximum 8-h average concentration (8-h max O₃) is ≤0.08 ppm. The new 1997 primary standard was based on various scientific supportive data from experimental human exposure studies, animal toxicologic studies, and epidemiological studies, as assessed in the 1996 O₃ AQCD and in the 1996 O₃ Staff Paper (U.S. Environmental Protection Agency, 1996b).

8.1.1 Chapter Organization

In addition to providing the above brief background information regarding prior O₃ NAAQS reviews, this first section (8.1 Introduction) of the integrative synthesis chapter aims to orient the reader to the organization and content of the chapter. The next section (Section 8.2) focuses on air quality trends and current ambient O₃ levels to help provide context for the ensuing discussions of O₃ exposures and associated health effects. Subsequent sections then integrate newly available key scientific information assessed in Chapters 4 through 7 of this document, including integration of information on O₃ dosimetry, toxicological information derived from controlled human exposure and laboratory animal studies, and epidemiologic evidence.

These sections collectively address the following topics: (1) ambient O₃ exposures, personal exposures, and dosimetric considerations; (2) experimental studies on toxicological responses to acute O₃ exposures in humans (clinical studies) and both acute and chronic effects in animals; (3) epidemiological evidence for associations between ambient O₃ exposure of

human populations and various health effects, as well as the strength and robustness of these associations; (4) integration of the experimental and epidemiological evidence; (5) biological mechanisms and other evidence useful in judging the plausibility of adverse health effects being associated with human exposures to ambient O₃ levels encountered in the United States; and (6) delineation of susceptible and vulnerable populations likely at increased risk for O₃-related health effects and numbers of people potentially falling in such categories in the United States.

8.2 AMBIENT OZONE AIR QUALITY IN THE UNITED STATES

8.2.1 Current Ozone Concentrations and Spatial Patterns

The EPA has established “O₃ seasons,” during which ambient O₃ concentrations must be monitored in the United States and its territories. These seasons vary in length depending on location. The O₃ season extends all year in the Southwest. In most other areas of the country, O₃ is monitored typically from April to October. However, O₃ is monitored throughout the year in many urban areas, because O₃ is present year round not only in polluted areas but in clean areas as well. The median of the daily 8-h max O₃ in the United States, averaged over May to September from 2000 to 2004 for all U.S. counties, was 0.049 ppm. In 95% of all counties, the median of the daily 8-h max O₃ was less than 0.057 ppm. However, it should be noted that most monitors are located in the East and O₃ data are sparse throughout large areas of the West. Median values of daily 1-h max O₃ were typically much higher in large urban areas or in areas downwind of them. For example, in Houston, TX they approached 0.20 ppm during the same 2000-2004 period. Daily 1-h max O₃ concentrations were lower in the rest of the country, but were still above 0.12 ppm in many locations. Eight-hour daily maximum concentrations were not as high, but tend to be highly correlated with 1-h daily maximums.

Within individual Metropolitan Statistical Areas (MSAs), O₃ concentrations tend to be well correlated across monitoring sites, although spatial variations in concentrations can be substantial. In many city centers, O₃ concentrations tend to be lower than in either upwind or downwind areas, largely due to reaction of O₃ with NO emitted by motor vehicles. For example, much lower O₃ concentrations overall are found in downtown Los Angeles (e.g., in Lynwood) than at sites located further downwind (e.g., in San Bernadino). The much higher downwind levels are formed from photochemical reactions involving the urban emissions, including

products formed as the result of reactions titrating O₃ in the urban core. Thus, O₃ concentrations tend to be higher downwind of urban centers, and they decrease again in areas that are more remote from precursor sources. Likewise, surface-level O₃ can be depleted in rural areas close to NO emission sources, such as highways and powerplants.

8.2.2 Diurnal and Seasonal Variations

Ozone concentrations typically tend to peak in early to mid-afternoon in areas where there is strong photochemical activity and to peak later in the afternoon or during early evening in areas where transport is more important in determining the O₃ abundance. Summertime maxima in O₃ concentrations occur in those U.S. areas where substantial photochemical activity acts on O₃ precursors emitted as the result of human activities. Monthly maxima can occur anytime from June through August. However, springtime maxima are observed in some National Parks, mainly in the western United States, and at a number of other relatively unpolluted monitoring sites throughout the Northern Hemisphere. For example, the highest O₃ concentrations at Yellowstone National Park tend to occur during April and May. Typically, monthly minima tend to occur from November through February at polluted sites and during the fall at relatively remote sites.

8.2.3 Long-Term Trends

National attention started to be focused in the 1940s on O₃ and associated photochemical smog in the Los Angeles area. Prior to the adoption of stringent emissions controls to reduce O₃ precursors, peak O₃ levels were consistently higher in the Los Angeles area than are currently observed. For example, in 1958, peak O₃ concentrations measured in Los Angeles were about 0.6 ppm but have declined since then, although not at a steady rate. Peak O₃ levels of 0.2 to 0.5 ppm were still found at some locations in the Los Angeles basin during the 1970s. For example, on two days (October 13 and 14) during a 1978 episode, Tuazon et al. (1981) observed peak 1-h averaged O₃ values of nearly 0.4 ppm and nearly 0.5 ppm. Currently, peak 1-h and 8-h average O₃ concentrations are about 0.17 and 0.15 ppm in the Los Angeles basin (cf., Figures 3-10 and 3-11). High O₃ levels were also earlier found throughout the rest of the United States as well, but peak O₃ levels have also gradually declined across the country during the 1980s and 1990s. However, during one particularly hot summer (of 1988) in the East, peak

1-h O₃ concentrations of about 0.2 ppm were observed in many eastern U.S. cities (U.S. Environmental Protection Agency, 1990).

Nationwide, 2nd highest 1-h ozone concentrations in the United States have decreased dramatically during the past several decades, i.e., by ~29 percent from 1980 to 2003 and 16 percent from 1990 to 2003. Also, 4th highest 8-h O₃ concentrations decreased by ~21 percent since 1980 and 9 percent since 1990 (U.S. Environmental Protection Agency, 2003). Trends in metrics for evaluating compliance with the O₃ NAAQS (i.e., changes in the 4th highest O₃ concentration) can be found in EPA's "National Air Quality and Emissions Trends Reports." These reports indicate that the 4th highest O₃ concentrations are still decreasing nationwide, but the rate of decrease has slowed since 1990. However, such trends have not been uniform across the United States. In general, reductions in the O₃ metrics given above have been largest in New England and in states along the West Coast and smallest in midwestern states. Downward trends in California O₃ concentrations have been driven mainly by notable decreases in Southern California, with smaller reductions occurring in other areas. Trends in peak O₃ metrics do not necessarily reflect changes in O₃ concentrations across the middle of the distribution of ambient O₃ values. Of note, O₃ concentrations towards the center of its nationwide distribution have not shown much change (cf., Figures 3-17, 3-18, and Table AX3-9), and there are some indications that O₃ concentrations at the lower end of the distribution may even be increasing.

8.2.4 Interrelationships Between Ozone and Other Ambient Pollutants

Data on ambient concentrations of other oxidants (e.g., H₂O₂, PAN) and oxidation products (e.g., HNO₃, H₂SO₄) in the atmosphere are not nearly as abundant as they are for O₃. Because data for such species are usually obtained only as part of specialized field studies, it is difficult to relate observed ambient O₃ concentrations to ambient levels of other oxidant species or oxidation products. In general, such secondary species are expected to be at least moderately positively correlated with O₃, whereas, primary species are expected to be more highly correlated with each other than with secondary species (provided that the primary species originate from common sources in given areas). Measurements of gas phase oxidants obtained as part of the Southern Oxidants Study (SOS) found combined hydroperoxide (H₂O₂, CH₃OOH, and HOCH₂OOH) concentrations typically in the range of several ppb. Other SOS measurements likewise indicated combined concentrations of PAN, PPN, and MPAN in the range of several ppb.

Oxidants are also present in airborne cloud droplets, rain drops, and particulate matter (PM). A few measurements of reactive oxygen species (ROS), expressed as equivalent H_2O_2 , in ambient fine PM have indicated levels of less than 1% of those for ambient O_3 on a molar basis. However, it should be noted that these measurements are potentially subject to both positive and negative artifacts.

Relationships between ambient air O_3 and PM levels can be quite complex because PM is not a single distinct chemical species, but rather a mix of primary and secondary species. As an example of this complexity, O_3 concentrations were found to be positively correlated with $\text{PM}_{2.5}$ during the summer, but negatively correlated during the winter at Ft. Meade, MD (cf., Figure 3-21). Also, Ito et al. (2005) examined relationships between ambient O_3 and PM_{10} on a seasonal basis in several U.S. urban areas (cf., Figure 7-24). Seasonal O_3 -PM relationships similar to those observed at Ft. Meade were found, reflecting the dominant contribution of $\text{PM}_{2.5}$ to PM_{10} in the urban areas studied. Possibly contributing to higher correlations seen between O_3 and fine PM in the summer is the fact that O_3 can contribute to formation of submicron particles via interactions with various other atmospheric constituents present, such as terpenes and other biogenically derived hydrocarbons from trees, other vegetation, and wood products. Formation of ultrafine particles by this mechanism mostly occurs during summer days when temperatures and O_3 concentrations are sufficiently elevated to facilitate O_3 reactions with increased amounts of biogenic hydrocarbons emitted from vegetation. Bursts of ultrafine particle formation have been observed repeatedly in both urban and rural air. Woo et al. (2001), for example, reported rapid formation of ultrafine particles in the ambient air of Atlanta typically around noon in both summer and winter. The mechanisms underlying such ultrafine particle formation events may also involve other atmospheric reactions related to O_3 formation, such as nucleation of H_2SO_4 (produced by oxidation of SO_2) and, probably, NH_3 .

8.2.5 Policy Relevant Background (PRB) Ozone Concentrations

Policy relevant background (PRB) O_3 concentrations, i.e., background O_3 concentrations used for NAAQS-setting purposes, are those that would occur in the United States in the absence of anthropogenic emissions in continental North America (defined here as the United States, Canada, and Mexico). Such PRB O_3 concentrations include contributions from natural sources everywhere in the world and from anthropogenic sources outside these three countries. For the

purpose of informing O₃ NAAQS decisions, EPA focuses on assessing risks to human health and environmental effects from O₃ levels in excess of PRB concentrations. Issues concerning the methodology for estimating PRB O₃ concentrations are discussed in detail in Annex AX3, Section AX3.9.

Contributions to PRB O₃ include photochemical reactions involving natural emissions of VOCs, NO_x, and CO, as well as the long-range transport of O₃ and its precursors from outside North America and the stratospheric-tropospheric exchange (STE) of O₃ (see Annex AX2, Section AX2.3 for details regarding O₃-related STE processes). Natural sources of O₃ precursors include biogenic emissions, wildfires, and lightning. Biogenic emissions from agricultural activities are not considered in the formation of PRB O₃. However, estimates of PRB O₃ concentrations cannot be derived solely from measurements of O₃ at relatively unpolluted sites because of long-range transport from anthropogenic source regions within North America. It is impossible to determine sources of O₃ at a particular location without ancillary data that could be used as tracers of sources or to calculate photochemical production and loss rates for O₃. Thus, estimates of PRB O₃ concentrations are currently based on predictions generated by the global scale, three dimensional, chemical transport model GEOS-CHEM (Fiore et al., 2003).

Policy relevant background O₃ concentrations vary as a function of season, altitude, and total surface O₃ concentration, with PRB O₃ concentrations at the surface generally falling in the range of 0.015 to 0.035 ppm from 1300 to 1700 local time and tending to decline under weather conditions conducive to O₃ episodes. The PRB concentrations are highest during spring and decline into summer; and higher values also tend to occur at higher elevations during the spring due to contributions from hemispheric pollution and stratospheric intrusions. The contribution to surface O₃ by stratospheric intrusions is typically well below 0.020 ppm. Stohl (2001) and Sprenger et al. (2003) found that the maximum probability of stratospheric intrusions reaching the 800 hPa level (~1800 m) was less than 1% and that higher probabilities (1 to 2%, and 10%) applied for stratospheric intrusions penetrating to the 600 hPa level (~4100 m) and 500 hPa level (~5400 m), respectively. Thus, stratospheric O₃ intrusions only rarely contribute to elevated surface-level O₃ concentrations at low altitude sites but have a higher (albeit still low) probability of elevating them at high-altitude sites.

8.3 FACTORS AFFECTING HUMAN EXPOSURE TO AMBIENT OZONE

Human exposure to O₃ and related photochemical oxidants varies over time due to changes in their ambient concentrations and because people move between locations having notably different O₃ concentrations. Also, the amount of O₃ delivered to the lung is not only influenced by the ambient concentration but also by the individual's breathing route and rate. Thus, activity level is an important consideration in determining potential O₃ exposure and dose received.

The use of data from ambient air monitoring stations is still the most common surrogate for assigning exposure estimates in epidemiologic studies. Since the primary source of O₃ exposure is the ambient air, O₃ concentration data from outdoor community monitoring sites should provide a relative assignment of exposure with time, if O₃ concentrations are relatively uniform across the region; time-activity patterns are roughly the same across the study population; and housing characteristics (such as ventilation rates and O₃ sinks contributing to indoor O₃ decay rates) are relatively constant for the study area. However, because these types of factors often do vary across populations and locations, some error tends to be associated with estimates of the magnitude of O₃ exposure of large populations. Nevertheless, although substantial variability may exist among personal measurements, human exposure studies have observed that daily average personal O₃ exposures for the general population tend to be reasonably well correlated with monitored ambient O₃ concentrations. Therefore, ambient O₃ monitoring data appear to provide the most useful index of human O₃ exposure currently available to help characterize health outcomes associated with O₃ exposures of large population groups.

8.3.1 Personal Exposure

Personal O₃ concentrations have been measured for children, outdoor workers, and individuals with pulmonary diseases (populations potentially vulnerable to increased ambient O₃ exposure and/or susceptible to O₃ or other respiratory irritants). Outdoor workers can be expected to have somewhat higher O₃ exposures than other individuals, because they typically spend more time outdoors and often engage in prolonged moderate and heavy exertion activities. Children also tend to be more active outside and, therefore, often manifest a higher breathing rate than most adults. Given the higher percentages of time spent outdoors than most other population groups, personal exposures of both outdoor workers and children tend to be more highly correlated with ambient O₃ concentrations measured at community monitoring sites.

8.3.2 Indoor Concentrations

Apart from only a few specific indoor sources such as photo-copying machines, O₃ indoors is derived from the infiltration of ambient air from outdoors. Generally, O₃ enters indoor environments through infiltration from the outdoors and through building components, such as windows, doors, and ventilation systems. Ozone concentrations in indoor environments depend primarily on the outdoor O₃ concentration, outdoor/indoor infiltration and the air exchange rate (AER). Hence, indoor O₃ concentrations tend to reflect outdoor concentrations and are higher when outdoor O₃ is higher.

Once indoors, O₃ reacts on various surfaces and with airborne components of either indoor or outdoor origin. Because O₃ reacts indoors with surfaces and other contaminants, O₃ concentrations are typically lower indoors than outdoors. Gas phase reactions occurring indoors also produce other oxidants analogous to the production of photochemical smog. The extent and rate of production of these other species indoors is a function of indoor O₃ concentrations and the presence of other necessary precursors (i.e., VOCs), along with an optimal AER.

Several studies of O₃ measured in residences, schools, office buildings and museums found that typical O₃ concentrations varied across all such locations. Indoor O₃ levels generally varied in relationship to the AER in the indoor environment (increasing with higher AER) and were generally notably lower than outdoor ambient O₃ levels. For example, one study of O₃ levels indoors and outside of a school in New England found average O₃ concentrations of 40 ppb (0.040 ppm) outdoors and 20 ppb (0.020 ppm) indoors. With regard to O₃ levels in mobile source microenvironments, as is the case for other enclosed environments, O₃ concentrations depend on the extent of mixing of outdoor air into the vehicle cabin. Thus, if windows are kept open, O₃ concentrations inside the vehicle may approach outdoor values. But, if windows are kept closed and there is air conditioning, then interior values can be much lower than those outside, especially if recirculated air is used. For example, in one N.C. study of police cars with air conditioning and recirculated air, vehicle cabin O₃ levels (11.7 ppb average) were less than half those outside (28.3 ppb average at area outdoor monitoring sites).

Although O₃ concentrations may be reduced to lower levels once ambient O₃ enters indoor environments, the indoor O₃ may interact with other airborne substances of indoor or outdoor origin present indoors. For example, Wainman et al. (2000) found that O₃ reacts with d-limonene, a common component of air fresheners to produce submicron particles found mainly

to range in size from 0.1 to 0.3 μm . Wainman et al. noted that terpenes such as limonene are emitted by wood products; that they are used as solvents, as odorants in cleaning products, and as air fresheners; and, even though they are produced by vegetation outdoors, their indoor concentrations are often higher than outdoors because of their widespread uses. In addition to particle formation, Weschler (2004) noted that gas phase products (i.e., aldehydes and hydroperoxides) produced by O_3 reactions with terpenes and other unsaturated carbon compounds may also be of concern. During formation of such products, OH radicals are also generated that can react with compounds that do not react with O_3 . To the extent that building ventilation rates remain constant between days characterized by high and low ambient O_3 , the concentrations of these other secondary pollutants formed indoors will tend to be correlated with ambient O_3 . Thus, ambient O_3 concentrations measured outdoors at community monitoring sites and/or personal O_3 exposure monitor measurements may serve not only as indices of direct human exposure to O_3 per se, but also as surrogate indices of exposures to broader O_3 -containing ambient mixtures of photochemical oxidants and/or other pollutants.

8.4 SYNTHESIS OF AVAILABLE INFORMATION ON OZONE-RELATED HEALTH EFFECTS

The integrated synthesis of the latest available information on O_3 -related health effects poses an interesting challenge in view of the emergence of highly important new information since the 1996 O_3 AQCD. Such information includes new findings from:

- Dosimetry studies that further characterize factors potentially affecting regional distribution of O_3 in the respiratory tract of humans and laboratory animals and provide improved bases for animal-to-human extrapolations of experimentally-observed O_3 -induced health effects.
- Experimental toxicological studies using controlled human exposures and laboratory animals aimed at delineating exposure-response relationships and understanding potential biochemical mechanisms underlying toxic effects, pathology, and susceptibility.
- Epidemiological studies, reflecting progress in addressing many research needs identified during the last review, as well as raising new issues and reevaluating previously addressed issues that remain important in interpreting the body of epidemiological evidence and characterization of its strengths and limitations.

Previous criteria assessments, including the 1996 O₃ AQCD, found that experimental studies of controlled human and laboratory animal exposures to O₃ provided the most clear cut and compelling evidence with regard to characterizing O₃-related health effects. Based on extensive dosimetric and experimental data, as well as growing epidemiologic evidence available at the time, the 1996 O₃ AQCD arrived at a set of findings and conclusions regarding potential health effects of ambient O₃ exposure. In general, the existing evidence was such to warrant a high degree of confidence in those conclusions derived from experimental (controlled exposure) studies of acute O₃ exposure effects. Considerable confidence could also be placed in the emerging field/panel studies providing observational study results substantiating and extending the controlled exposure study findings. The controlled exposure and field/panel studies clearly demonstrated respiratory effects (e.g., reduced pulmonary function) among healthy and asthmatic children and adults acutely exposed for 1 to 8 h to 0.08 ppm O₃ while physically active. Other epidemiologic studies provided highly suggestive, although less conclusive, indications of increased morbidity (e.g., as indexed by emergency department [ED] visits, hospital admissions, etc.) and, possibly, mortality being associated with acute exposure of human populations to ambient O₃. Also, data from laboratory animal and controlled human exposure studies supported the hypothesis that coexposure to O₃ and other pollutants at low-effect levels can result in enhanced effects. However, the issue of exposure to copollutants was poorly understood, especially with regard to potential chronic effects.

Since the 1996 O₃ AQCD evaluations, further controlled human exposure studies have extended earlier findings of respiratory effects of acute exposures in exercising adults to O₃ concentrations ranging below 0.08 ppm for some sensitive subjects. Also, a more extensive database of air pollution epidemiologic studies has become available; and a subset of these new observational studies have reported a variety of O₃-related health effects associations, with newly reported evidence of ambient O₃ mortality relationships being of special interest. Based on physiological, biochemical, and molecular changes observed in controlled human exposure studies and animal toxicological studies, new evidence is now available by which to more fully evaluate the biological plausibility and extent of coherence for various health outcomes (such as respiratory and cardiovascular effects and mortality) reported in epidemiologic studies.

8.4.1 Integration of Experimental and Epidemiologic Evidence

8.4.1.1 Cross-Cutting Issues Relevant to Assessment/Interpretation of Ozone Health Effects

Three chapters in the current document provide detailed discussion of various experimental approaches utilized to evaluate O₃-related health effects. Chapter 4 discusses dosimetry issues pertinent to both animal and human exposure scenarios. Chapter 5 discusses the experimental studies of biochemical (cellular and molecular changes), physiological, and pathological observations in laboratory animals (including nonhuman primates, dogs, and rodent species) and in vitro studies using cell culture systems, in certain cases, on humans cells recovered from bronchoalveolar lavage fluid (BALF) after exposure to O₃. Chapter 6 evaluates studies on human volunteers exposed to O₃ which have investigated a variety of physiological and biochemical endpoints. In interpreting the results from the experimental approaches, one must consider the following issues: (1) exposure/dose considerations; (2) interpretation of results for animals and humans from relatively high dose O₃ exposures; (3) animal-to-human extrapolations, and (4) experimental results derived from single pollutant exposures in comparison to much more complex ambient exposures.

Earlier animal toxicology studies were carried out using relatively high O₃ exposure concentrations/doses that do not necessarily reflect “real-world” exposure scenarios. Those experiments were primarily aimed at understanding the pathophysiology associated with O₃ exposure in healthy animals, to help identify potential mechanisms(s) of action and to help validate health outcomes reported in epidemiologic studies. Since the 1996 O₃ AQCD, some of the human and animal studies have used ambient and/or near ambient doses. The majority of controlled chamber exposure studies on human volunteers mainly limited exposures to O₃ alone in comparison to sham (clean air) exposures, thus providing evidence concerning direct effects of O₃ per se versus more closely mimicking real-world atmospheric exposures to multipollutant mixes. Some limited data from laboratory animal and controlled human exposure studies reviewed in the 1996 O₃ AQCD pointed towards enhancement of some observed health effects by coexposures to O₃ and other air pollutants. Since then a few newer human clinical and laboratory animal air pollution studies have utilized various coexposure regimens to simulate more closely ambient exposure to air pollution mixtures; and the results from these studies are highly useful in developing better models to interpret toxicologic effects associated with O₃-containing ambient air pollutant mixes.

Interpretations of experimental studies of air pollution effects in animals, as in the case of environmental comparative toxicology studies, are affected by limitations associated with extrapolation models. The differences between humans and rodents with regard to O₃ absorption and distribution profiles based on breathing pattern, exposure dose, and differences in lung structure and anatomy (see Chapter 4 and 5 for details) all have to be taken into consideration. Also, in spite of a high degree of homology and the existence of a high percentage of orthologous genes across human and rodents (particularly mice), extrapolation of molecular alterations at the gene level is complicated by species-specific differences in transcriptional regulation. Given these molecular differences, there are large uncertainties associated with quantitative extrapolations at this time between laboratory animals and humans of observed O₃-induced pathophysiological alterations under the control of widely varying biochemical, endocrine, and neuronal factors.

Epidemiologic studies provide important additional observational information to our knowledge base regarding O₃-induced health effects. Of particular importance, such studies help (1) to ascertain the extent to which the types of health effects found in experimental studies occur in response to ambient exposures of human populations to O₃ and/or other copollutants; and (2) to delineate exposure-response relationships for more serious health effect outcomes (e.g., hospital admissions, mortality) that help to elucidate more fully the likely public health consequences of ambient O₃ exposures.

The ensuing sections both briefly summarize key dosimetry and health related findings derived from the 1996 O₃ AQCD and integrate those findings with new information obtained since 1996 from (a) human and animal experimental studies and (b) epidemiologic studies. Physiological, pathological, cellular and biochemical alterations induced by experimental O₃ exposures are evaluated and compared to human health effects associated with ambient O₃ exposures. Also, the influence of O₃-induced changes at cellular and molecular levels are integrated to help elucidate mechanistic bases for the observed physiological and pathological alterations. These research results are evaluated both to help assess the biological plausibility of health outcome associations observed in epidemiologic studies and to assess the coherence of the overall body of evidence relevant to O₃-related health outcome conclusions, to aid in drawing conclusions about the likelihood of causal relationships existing between ambient exposure to O₃ and various types of reported O₃-related health effects.

8.4.1.2 Dosimetry

The state-of-the-art of O₃ dosimetry, as described in 1996 O₃ AQCD, indicated consistency across data and models derived from in vivo human and animal studies, thus increasing the level of confidence in the development of dosimetric extrapolation models. Earlier dosimetry models predicted that the tissue dose of inhaled O₃ was greatest at the bronchoalveolar junction, the lung region experimentally shown to be most impacted by O₃. Ozone bolus inhalation studies in humans had indicated that inspired O₃ reaches the distal airways and alveoli of resting humans; and, with increased inspiratory flow rates due to exercise, O₃ penetrates deeper and in greater quantity to the distal regions of the lung. These findings were corroborated by observations of ¹⁸O₃ (oxygen-18-labeled ozone) in the BALF of humans and rats (Hatch et al., 1994). Based on BALF parameters monitored in this study, an exposure of 0.4 ppm O₃ in exercising humans appears to be approximately equivalent to an exposure of 2 ppm in resting rats.

Some O₃-induced acute responses compared well across species when controlled for dose, indicating that animals and humans (a) respond to O₃ in a dose-dependent manner, i.e., they exhibit increasing breathing frequency with an accompanying decrease in tidal volume (tachypnea), and (b) show similar changes in alveolar permeability as measured by protein in the BALF. These parallel changes in humans and animals were sufficiently homologous to suggest a common mode of action. The majority of lung function decrements were seen to subside with repeated exposures in both humans and animals, with analogous attenuation of certain (but not all) parameters measured in the BALF. The mechanisms associated with attenuation are unclear but may involve endogenous antioxidants. The significance of non-attenuated markers in BALF has been interpreted to relate to potential chronicity of O₃ effects.

During the past decade, no further reports have been published on O₃ uptake studies in animals, although several controlled human bolus and/or general O₃ uptake studies have provided refined data. The bolus uptake studies suggest that prior exposure to O₃ diminishes bolus uptake. In the 1996 O₃ AQCD, the effect of mode of breathing (oral or nasal) on O₃ uptake was thought to be minimal, with approximately equal uptake via the nose or mouth. Newer bolus dose studies have demonstrated that the uptake and regional respiratory tract distribution of O₃ is sensitive to mode of breathing (nasal uptake greater than oral) and to air flow rate (uptake decreases with increasing flow). Studies (Rigas et al., 1997, Bush et al., 2001) that evaluated the role of co-exposure with other oxidants such as NO₂ and SO₂ recognized the

importance in mediating O₃ toxicity of other oxidative environment components in the lung, e.g., radicals generated by O₃ interaction with the lipids in the airway and epithelial lining fluid (ELF).

New uptake studies (Ultman et al., 2004) carried out in controlled human clinical studies have observed gender-specific differences in the uptake of O₃, but these differences do not correlate well with spirometric responses. Rather, they appear to be related to breathing pattern and lung size, due to females having smaller lungs than males. Thus, a number of variables seem to affect O₃ uptake, notably including route of breathing, breathing pattern, gender, copollutants, and certain pre-exposure conditions. These differences are important in order to interrelate experimentally-demonstrated pathophysiological effects and epidemiologically-observed associations between ambient O₃ concentrations and health risks among human population groups.

8.4.2 Experimental Evidence for Ozone-Related Health Effects

Pulmonary Function

Numerous controlled human exposure, animal, and epidemiological studies assessed in the 1996 O₃ AQCD demonstrated alterations in various pulmonary function measures. Ozone inhalation for several hours (1 to 3 h) while physically active was shown to elicit both acute pathophysiologic changes and subjective respiratory tract symptoms. The pulmonary responses observed in healthy human subjects exposed to ambient O₃ concentrations included decreased inspiratory capacity; mild bronchoconstriction; rapid, shallow breathing pattern during exercise; and subjective symptoms of tracheobronchial airway irritation, including cough and pain during inspiration. Acute O₃ exposures were also found to cause decreases in forced vital capacity (FVC), forced expiratory volume in 1 s (FEV₁), peak expiratory flow (PEF), and increased airways resistance (SR_{aw}). The severity of symptoms and the magnitude of pulmonary response were reported to vary as a function of exposure concentration, duration, and level of exercise (which help to determine inhaled O₃ dose), as well as the individual sensitivity of exposed subjects and the extent of tolerance resulting from previous exposures. With regard to the latter, the 1996 O₃ AQCD noted that during repeated short-term exposures, some of the O₃-induced responses (inflammation and lung function decrements) are partially or completely attenuated with a differential attenuation profile. Over a 5-day exposure, both pulmonary and inflammatory

markers were found to attenuate by the fifth day of exposure, but markers of cell damage (discussed below under *Lung Injury*) do not attenuate and continue to increase.

Group mean data from numerous controlled human exposure (in healthy subjects 8 to 45 years of age, after 1 to 3 h exposure) and field studies indicated that, in general, statistically significant pulmonary function decrements beyond the range of normal measurement variability (e.g., 3 to 5% for FEV₁) occur (a) at >0.50 ppm O₃ when at rest; (b) at >0.37 ppm O₃ with light exercise (slow walking); (c) at >0.30 ppm O₃ with moderate exercise (brisk walking); (d) at >0.18 ppm O₃ with heavy exercise (easy jogging) and (e) at >0.16 ppm O₃ with very heavy exercise (running). Smaller group mean changes (e.g., <5%) in FEV₁ occurred at lower O₃ concentrations with very heavy exercise in healthy adults at 0.15 to 0.16 ppm O₃ and in healthy young adults at levels as low as 0.12 ppm. Also, pulmonary function decrements were seen in children and adolescents at O₃ concentrations of 0.12 and 0.14 ppm with heavy exercise.

Human studies reviewed in the 1996 O₃ AQCD that used longer duration (6- to 8-h) acute exposures with exercise (which better mimic multihour exposures to ambient O₃ that typify more prolonged elevated ambient O₃ levels often observed in U.S. urban areas) provided some of the strongest and most quantifiable concentration-response data on pulmonary function effects of acute O₃ exposure. Overall, the 1996 O₃ AQCD found that for healthy subjects performing moderate exercise during longer duration (6 to 8 h) acute O₃ exposures, group mean 5% decrements in FEV₁ were seen at relatively low O₃ levels, such as: 0.08 ppm O₃ after 5.6 h, 0.10 ppm O₃ after 4.6 h, and 0.12 ppm O₃ after 3 h. Some subjects in the earlier studies experienced FEV₁ decrements in excess of 15% with short-term (1- to 3-h) or longer-duration (6.6-h) acute O₃ exposures, suggesting potential wide interindividual variability in pulmonary function responses.

The few newly available controlled human exposure studies (since those assessed in the 1996 O₃ AQCD) that used near-ambient O₃ concentrations (\leq 0.12 ppm) are summarized in Appendix Table 8A-1; and these, like most controlled exposure studies to date, continue to indicate that considerable interindividual differences exist in the magnitude of responses to O₃. However, a given individual's lung function and, to a lesser extent, respiratory symptom responses to O₃ are reproducible over a period of time, indicating that some individuals are consistently more responsive than others to O₃. Figure 8-1A illustrates well the variability in FEV₁ responses in young healthy adults following a prolonged (6.6 h) exposure to O₃, as

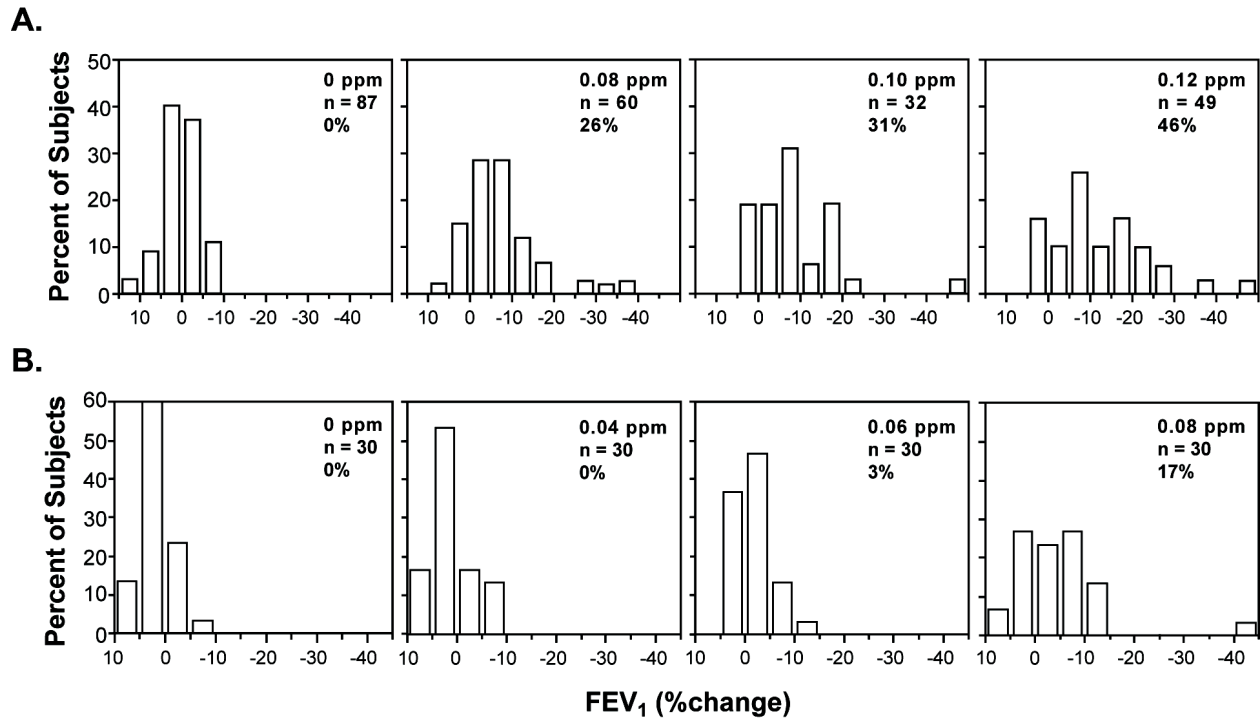


Figure 8-1A,B. Frequency distributions of FEV₁ changes following 6.6-h exposures to a constant concentration of O₃ or filtered air. During each hour of the exposures, subjects were engaged in moderate exercise for 50 minutes. With increasing O₃ concentration, the distribution of responses becomes asymmetric, with a few individuals exhibiting large FEV₁ decrements. Note that the percentage in each panel indicates the portion of subjects having a FEV₁ decrement in excess of 10%.

Source: Panel A, McDonnell (1996); Panel B, Adams (2002, 2006), pre- and post-FEV₁ data for each subject provided by author.

summarized in the 1996 O₃ AQCD. Referring to this figure, the group mean FEV₁ response following continuous (square-wave) exposure to 0.08 ppm O₃ was relatively small (between a 5 and 10% decrement). However, 18% of the exposed subjects had moderate FEV₁ decrements of 10 to 20%, and 8% experienced large FEV₁ decrements of greater than 20%. This serves to emphasize that, while group mean responses may be small and seem physiologically insignificant, some individuals can experience distinctly larger effects under similar O₃ exposure conditions. Newer data from Adams (2002, 2006), as illustrated in Figure 8-1B, demonstrate notable interindividual variability for O₃ exposure concentrations at and even below 0.08 ppm.

Following 6.6-h continuous exposure to 0.08 ppm O₃ under intermittent, moderate exercise conditions, the group mean FEV₁ decrement was 5%, but 17% of the subjects had greater than a 10% decrease in FEV₁. Following exposure to 0.06 ppm O₃, the group mean FEV₁ decrement was less than 2%. However, five subjects still had greater than a 5% decrease in FEV₁ whereas only one experienced this magnitude of effect following exposure to filtered air.

It should be noted that spirometry typically *improves* in healthy young adults with exercise exposures to filtered air (FA). The term “O₃-induced” was used in Chapter 6 and Annex 6 to designate effects that have been corrected for these filtered air responses. For healthy adults, an O₃-induced change in lung function is the difference between the *decrement* experienced with O₃ exposure and the *improvement* observed with filtered air exposure. The FEV₁ responses illustrated in Figure 8-1 were not corrected for the responses following filtered air exposures. For comparison to Figure 8-1B, O₃-induced FEV₁ responses from the Adams (2002, 2006) studies are illustrated in Figure 8-2. For the exposures to 0.04, 0.06, and 0.08 ppm O₃, an O₃-induced FEV₁ decrement of ≥10% was experienced by 7, 7, and 23% of subjects, respectively. Effects of a 0.04 ppm O₃ exposure were not apparent when simply comparing pre-post FEV₁ (Figure 8-1B), whereas they are evident when considering O₃-induced FEV₁ data (Figure 8-2). The distinction between an O₃-induced change and a post- versus preexposure change is particularly important in individuals with respiratory disease who may experience exercise-induced *decrements* in pulmonary function during both filtered air and O₃ exposures.

Other new studies (assessed in Chapter 6 and Annex 6 of this document) that evaluated responses in hundreds of subjects clearly indicate that O₃-related FEV₁ decrements and symptom responses decrease with age beyond young adulthood (18 to 20 years). For example, Hazucha et al. (2003) studied possible gender and age differences in O₃ responsiveness and found that young females lose O₃ sensitivity faster than young males, but the rate is about the same for both genders by middle age.

A few controlled human exposure studies that explored a triangular exposure profile of average O₃ concentrations between 0.08 to 0.12 ppm over 6.6 to 8 h (to more closely mimic the typical ambient O₃ exposure pattern) observed greater overall FEV₁ decrements with triangular exposures compared to constant or square-wave exposures (see Appendix Table 8A-1 and Annex Figure AX6-3). Furthermore, the peak FEV₁ decrements observed during triangular exposures exceed those observed during square-wave exposures. At a lower average O₃ concentration of

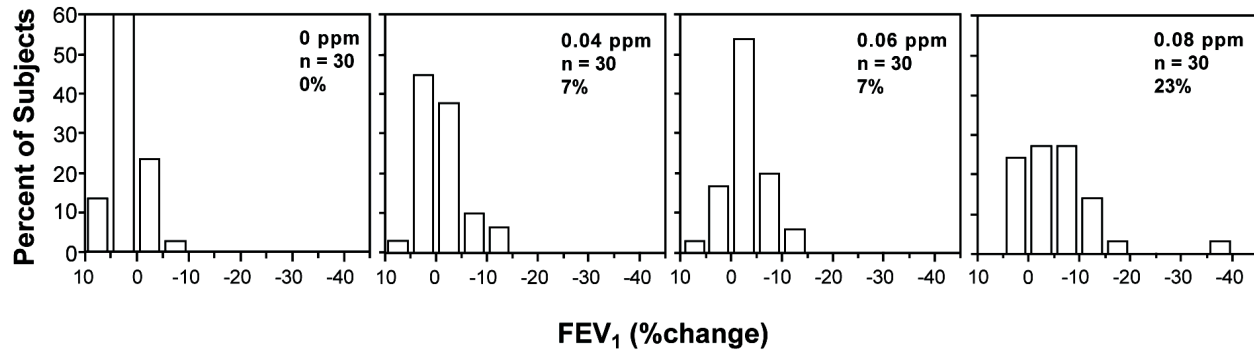


Figure 8-2. Frequency distributions of FEV₁ changes following 6.6-h exposures to a constant concentration of O₃ or filtered air. The FEV₁ changes following O₃ exposures have been corrected for filtered air responses, i.e., they are O₃-induced FEV₁ changes. These data should be compared with Figure 8-1B. Note that the percentage in each panel indicates the portion of subjects tested having FEV₁ decrements in excess of 10%.

Source: Adams (2002, 2006), pre- and post- FEV₁ data for each subject provided by author.

0.06 ppm, no temporal (i.e., hour by hour responses) differences were observed in FEV₁ responses between square-wave and triangular exposure profiles (Adams, 2003, 2006; Hazucha et al., 1992). These observations suggest the potential for somewhat enhanced O₃ effects on lung function in ambient exposure scenarios that typically involve gradually increasing daily O₃ exposure up to a peak in late afternoon and a subsequent gradual decline (i.e., the “triangular” exposure paradigm).

The development of effects is time-dependent during both exposure and recovery periods, with considerable overlap of evolving and receding effects. In healthy human subjects exposed to typical ambient O₃ concentrations (i.e., <0.12 ppm), spirometric responses largely resolve within a few hours (4 to 6 h) postexposure, but cellular effects persist for longer periods (~24 h). In such subjects, persisting small residual lung function effects are almost completely resolved within 24 hours; but in hyperresponsive individuals, the recovery takes longer (as much as 48 h) to return to baseline values. The majority of these responses are attenuated after repeated exposure, but such tolerance to O₃ is lost within one week postexposure. Unfortunately, no new data are available on pulmonary function changes in animals upon chronic exposure to O₃; but,

earlier research on repeated exposure of rats to an episodic profile of O₃ found small (but significant) decrements in lung function that were consistent with early indicators of focal fibrinogenesis in the proximal alveolar region.

In the 1996 O₃ AQCD, O₃-induced decreases in inspiratory capacity were hypothesized to result from neurogenic inhibition of maximal inspiration due to stimulation of C-fiber afferents, either directly or from O₃-induced inflammatory mediators. Earlier human studies (Coleridge et al., 1993; Hazucha and Sant'Ambrogio, 1993) reported a role for bronchial C-fibers and rapidly adapting receptors as primary vagal afferents responsible for O₃-induced changes in ventilatory rate and depth. As discussed in Chapter 6, the newer results of Passannante et al. (1998) also support C-fiber stimulation as a primary mechanism of the O₃-induced reduction in inspiratory capacity and suggest a role for nociceptive mechanisms. This neurogenic mechanism is also likely related to effects such as increased airway responsiveness and lung inflammation.

Only two studies evaluated lung function changes due to O₃ exposure in patients with preexisting respiratory diseases under experimental controlled exposure regimens. Studies using COPD patients found that O₃ exposure-induced minimal effects (nonsignificant decreases in lung function) in this population (Gong et al., 1997a). However, newer studies of asthmatics by Alexis et al. (2000) still continue to indicate that asthmatics are at least as sensitive, if not more, than healthy subjects, based on pulmonary function deficiencies detected by spirometric analyses.

In addition to effects of O₃ exposure on the large airways as indicated by spirometric responses, O₃ exposure also affects the function of the small airways and parenchymal lung tissue. Studies reported by Foster et al. (1993, 1997) that examined the effect of O₃ on ventilation distribution in healthy adult males suggest a prolonged O₃ effect on the small airways and ventilation distribution in some individuals. Animal toxicology studies have shown the centriacinar region (CAR) of the lung (the segment between the last conducting airway and the gas exchange region) to be a region highly susceptible to O₃-induced damage (epithelial cell necrosis and remodeling of respiratory bronchioles). Common pulmonary function tests do not measure acute changes in the small airways of the CAR. Identification of acute effects of O₃ in small airways, if any, would lend additional support for concerns about long-term effects of repeated O₃ exposures.

Airway Responsiveness

The 1996 O₃ AQCD recognized the induction of airway hyperresponsiveness (AHR) in humans on exposure to O₃, which usually resolves in 18 to 24 h after exposure in a majority of subjects, but may persist in some individuals for longer periods. Ozone-induced nonspecific AHR could have clinical implications in asthmatics, possibly putting them at potential increased risk for more prolonged bouts of bronchoconstriction in response to various triggering stimuli in the ambient air. Three new studies (discussed in Chapter 6) suggest that (1) subjects with asthma developed tolerance to repeated O₃ exposures in a manner similar to normal subjects; but there were more persistent effects of O₃ on airway responsiveness, which only partially attenuated when compared to filtered air control exposures (Gong et al., 1997b); (2) the enhancement of allergen responsiveness after O₃ exposure appears to be time-dependent, suggesting that the timing of allergen challenge in O₃-exposed subjects with allergic asthma is important (Jörres et al., 1996); and (3) subjects with rhinitis exhibit significant, clinically relevant decreases in pulmonary function in the early phase allergen response (Holz et al., 2002). These observations suggest that O₃ exposure may be a clinically important factor that can exacerbate responses to other ambiently-encountered bronchoconstrictor substances in individuals with preexisting allergic asthma and that its influence may be both immediate and persist for relatively long periods of time (a few days).

An extensive laboratory animal study database (see Chapter 5 Annex Tables) exploring the effects of acute, long-term, and repeated exposures to O₃, clearly indicates that induction of AHR occurs at relatively high O₃ concentrations. Only one study that utilized very low O₃ (0.05 ppm) concentrations found that acute exposure induced AHR in a few inbred strains of rat, suggesting a role for genetic susceptibility in this process (Depudyt et al., 1999). As seen with humans, acute changes in AHR do not persist upon long-term exposure of laboratory animals to near-ambient concentrations of O₃; and attenuation has been observed. Both human and animal studies indicate that airway responses are not associated with inflammation, but they do suggest a likely role for neuronal involvement.

Lung Inflammation, Permeability, and Biochemical Alterations

Respiratory tract inflammation and increased cellular permeability are the two most important biological markers of O₃-induced injury response mechanisms in both humans and

animals. These distinct, independent biological effects have been observed in all species studied in response to acute O₃ exposure.

Several controlled human exposure studies reviewed in the 1996 O₃ AQCD clearly showed that a single acute exposure of humans to moderate O₃ concentrations (≥ 0.08 to 0.1 ppm) while exercising at moderate to heavy levels disrupts the barrier function of the lung, initiating a cascade of responses indicative of increased lung permeability and pulmonary inflammation, as indexed by a variety of cellular and biochemical changes, e.g., increases in levels of polymorphic neutrophils (PMNs) and protein in lung fluid. Both the inflammatory response and increased lung permeability have been observed as early as 1 h and persisted for at least 18 h. The newer studies reviewed in this document (see Chapter 6) provide additional information on three different aspects of O₃-induced inflammatory responses, such as (1) intersubject variability; (2) differential attenuation profiles for different inflammatory markers; and (3) effects of repeated exposures.

Soluble mediators of inflammation (e.g., the cytokines IL-6 and IL-8), as well as arachidonic acid metabolites (e.g., PGE₂, PGF_{2 α} , thromboxane, and leukotrienes [LTs] such as LTB₄), have been measured in the BAL fluid of humans exposed to O₃. In addition to their role in inflammation, many of these compounds have bronchoconstrictive properties and may be involved in increased airway responsiveness following O₃ exposure. The time course for the inflammatory responses (including recruitment of neutrophils and other soluble mediators) is not clearly established, but differential attenuation profiles for many of these parameters are evident from the meta-analysis by Mudway and Kelly (2004) of 21 controlled human exposure studies.

Recent studies continue to support the observation made in the 1996 O₃ AQCD that repeated O₃ exposures in humans also induce ongoing cellular damage irrespective of attenuation of the inflammatory responses and lung function decrements (Devlin et al., 1997; Jörres et al., 2000). Devlin et al. (1997) studied inflammatory responses of humans repeatedly exposed to 0.4 ppm O₃ for 5 consecutive days. Several indicators of inflammation (e.g., PMN influx, IL-6, PGE₂, fibronectin) were attenuated after 5 days of exposure (i.e., values were not different from FA). Several other markers (LDH, IL-8, total protein, epithelial cells) did not show attenuation, indicating that tissue damage probably continues to occur during repeated exposure. Recovery of the inflammatory response occurred for some markers after 10 days, but some did not return to normal even after 20 days. When re-exposed 2 weeks later, changes in BALF indicated that

epithelial cells appeared to be fully repaired (Devlin et al., 1997). Kopp et al. (1999) observed inflammatory responses only after the first O₃ peak in summer; and their absence late in summer (even after exposure to higher levels of O₃) may be due to attenuation of such responses upon repeated O₃ exposures.

Numerous studies reported acute O₃-induced changes in lung epithelial permeability assessed by indirect assay (increased levels of albumin and protein in BALF). A few other studies demonstrated O₃-induced epithelial cell permeability through direct assessment of clearance of ^{99m}Tc-DTPA (technetium-99m labeled diethylene triamine pentaacetic acid). For example, Kehrl et al. (1987) showed increased ^{99m}Tc-DTPA clearance in healthy young adults at 75 minutes postexposure to 0.4 ppm O₃ for 2 h. More recently, Foster and Stetkiewicz (1996) have shown that increased ^{99m}Tc-DTPA clearance persists for at least 18 to 20 h post-O₃ exposure (130 min to average O₃ concentration of 0.24 ppm), and the effect is greater at the lung apices than at the base.

Laboratory animals, like humans, exhibit varying degrees of sensitivity to O₃ exposure (see Chapter 5 for detailed discussion); and this is evident even for the induction of pulmonary inflammation and permeability. New animal toxicology studies of O₃-induced inflammation assessed in Chapter 5 indicate that the lowest acute O₃ exposures that had an effect on mouse lung inflammation was 0.12 ppm for 24 hours. Shorter durations (8 h) required greater O₃ exposure (0.26 ppm) for effects on epithelial permeability but had no effect on inflammation. The lowest acute O₃ concentrations that had an effect on epithelial permeability or inflammation in the rat were 0.5 ppm for 3 hours or 0.12 ppm for 6 hours. Also, increased lung permeability and inflammation occurred in rabbits with O₃ exposures as low as 0.10 ppm for 2 h/day for 6 days. Subchronic exposures in animals suggest that permeability changes are transient (and species-dependent) and return to control levels even with continuing exposure. Chronic animal O₃ exposure studies suggest a role for persistent inflammation in O₃-induced alterations in lung structure and function. Significant remodeling of epithelium and underlying connective tissues in distal airways have been reported in rats exposed to 0.25 ppm O₃ (12 h/day for 6 wk) and in monkeys exposed to 0.2 ppm O₃ (8 h/day for 90 d). Various factors such as viral infection, chemotactants and oxidized matrix fragments are also implicated in the establishment and persistence of O₃-induced inflammation.

Animal toxicology and human in vitro studies that evaluated biochemical mediators implicated in injury and inflammation found alterations in the expression of cytokines, chemokines, and adhesion molecules, indicative of an ongoing active stress response as well as injury repair and regeneration processes. Both animal and human studies indicate cellular and biochemical changes associated with inflammation and increased permeability, but the relationship between these changes and possible airway remodeling is not known.

As discussed in the earlier section on dosimetry, interaction of O₃ with the lipid constituents of pulmonary surfactant has been proposed as one of the key mechanisms by which O₃ exerts its toxic effects. Experimental evidence clearly indicates a role for the interaction of O₃ with lipid constituents of the ELF and cell membranes and the generation of lipid ozonation products and secondary redox mediators in the initiation of site-specific cell-injury response cascades. One such lipid ozonation product, 4-hydroxynonenal, has been found to bind to proteins and increased protein adducts in human alveolar macrophages, suggesting a role for 4-hydroxynonenal in acute cell toxicity. Cholesterol, the most abundant neutral lipid in pulmonary surfactant is susceptible to attack by O₃, resulting in formation of multiple oxidized cholesterol products, e.g., cholesterol epoxide. A 20-fold increase in cholesterol epoxide in the BALF from mice exposed to 0.5 ppm O₃ for 3 h suggests a potential role for this oxidation product in O₃ toxicity (Pulfer et al., 2005). Species- and region-specific increases in lung xenobiotic metabolism have been observed in response to both short- and long-term O₃ exposure. It has been well recognized that antioxidants in the ELF confer some protection against O₃ toxicity. But even with environmentally relevant exposures, O₃ reactivity is not quenched. Further, antioxidant reactivity with O₃ is both species-specific and dose-dependent.

Lung Injury

Pulmonary histopathological observations reported in the 1996 O₃ AQCD suggested that similar types of alterations occurred in lung morphology in all laboratory animal species studied, including primates, upon short-term O₃ exposure. The cells in the CAR were recognized as a primary target, possibly because the CAR receives the greatest dose of O₃ delivered to the lower respiratory tract. With chronic O₃ exposure, structural changes were also observed in this region of the respiratory tract (the region typically found affected in most chronic airway diseases of the human lung). Simulated seasonal exposure studies in animals also suggested that seasonal O₃

exposures may have potential for cumulative impact over many years. Ciliated cells in the nasal cavity and airways and Type I epithelial cells in the gas-exchange region have also been identified as targets. Though acute O₃ exposure induces structural changes such as fibrosis in the CAR, these structural alterations appear to be partially transient, with recovery shortly post O₃ exposure; but the time for recovery is dependent on the species and the dose of O₃. Long-term or prolonged exposure to O₃ has been found to cause chronic lesions similar to early lesions of respiratory bronchiolitis, which have the potential to progress to fibrotic lung disease. Some of the morphological changes associated with long-term O₃ exposures, e.g., increases in hyperplastic epithelial cells, appear to reverse following cessation of O₃ exposure. However, in the underlying interstitium of the CAR, proliferation of fibroblasts creates excess noncellular matrices. These processes are only partially reversible and may progress following cessation of O₃ exposure. This suggests initiation of focal interstitial fibrosis, which can progress to nonreversible structural damage to lung tissue.

Reports of morphological changes following chronic O₃ exposures in animal studies (rodents and primates) published since the 1996 AQCD allude to the earlier findings assessed in that document. In rats, the effects of chronic ~0.5 ppm O₃ exposure included mucous cell metaplasia, hyperplasia of the nasal epithelium, increased mucosubstances, and increased Bcl-2 protein levels. In mice, lifetime exposures of 0.5 ppm O₃ were linked to similar outcomes. Taken together, the rodent studies suggest that O₃ exposure may have the potential to induce similar long-lasting alterations in human airways. A series of new studies that utilized infant rhesus monkeys and simulated seasonal ambient exposure (0.5 ppm 8 h/day for 5 days, every 14 days for 11 episodes) reported remodeling in the distal airways; abnormalities in tracheal basement membrane; eosinophil accumulation in conducting airways; and decrements in airway innervation, again confirming the potential greater injury due to seasonal exposure compared to continuous exposure alluded to in the 1996 O₃ AQCD.

Host Defense

Based on a small number of studies available at the time, the 1996 O₃ AQCD concluded that short-term O₃ exposure of laboratory animals and humans impairs alveolar macrophage (AM) clearance of viable and nonviable particles from the lungs and decreases the effectiveness of host defenses against bacterial lung infections in animals and perhaps humans. A single

controlled human exposure study reviewed in the 1996 O₃ AQCD found decrements in the ability of alveolar macrophages to phagocytose microorganisms upon exposure to 0.08 to 0.1 ppm O₃ for 6.6 h during moderate exercise (Devlin et al., 1991).

Other evidence for O₃-induced dysfunction of host defense components and for subsequent enhanced susceptibility to bacterial lung infection has been derived from laboratory animal studies. Acute exposures of 0.08 ppm (3 h) O₃ were found to result in the mortality of mice due to Streptococcal bacterial infection. Changes in antibacterial defenses appear to be dependent on exposure regimens, species and strain of test animal, species of bacteria, and age of animal (with young mice being more susceptible to the effects of O₃, for example). Animal toxicology studies indicated that acute O₃-induced suppression of alveolar phagocytosis and immune functions observed in animals appeared to be transient and attenuated with continuous or repeated exposures.

It has also been reported that O₃ exposures can interfere with AM-mediated clearance in the respiratory region of the lung and with mucociliary clearance of the tracheobronchial airways. Ozone-induced perturbations in the clearance process have been found to be dose-dependent, with low dose exposures accelerating clearance and high doses slowing the clearance process. Some respiratory tract regional- and species-specific differences have also been observed.

In vitro cultures of epithelial cells obtained from nonatopic and mild atopic asthmatics exposed to 0.01 to 0.1 ppm O₃, exhibited significantly increased permeability compared to cells from normal subjects, thus indicating a potential inherent susceptibility of cells from asthmatics for O₃-induced permeability. Newer in vitro cell culture studies of human bronchial epithelial cells indicate O₃-induced exacerbation of human rhinovirus type 16 infection (Spannhake et al., 2002); and new animal toxicology studies have shown O₃-induced modulation of cell-mediated immune responses affecting the onset and persistence of infection in rats (Cohen et al., 2001, 2002).

The available data at this time indicate that acute O₃ exposure has a potential to impair host defense capability, primarily by interfering with the functions of alveolar macrophages. Any impairment in macrophage function may lead to decreased clearance of microorganisms or nonviable particles. Compromised alveolar macrophage functions in asthmatics may increase their susceptibility to other O₃ effects, the effects of particles, and respiratory infections.

Cardiovascular Effects

Based on the paucity of then-available information, the 1996 O₃ AQCD did not accord much, if any, attention to possible O₃-related cardiovascular effects. However, since then, an emerging body of animal toxicology evidence is beginning to suggest that hematological and thermoregulatory alterations (in heart rate and/or core body temperature) may mediate acute O₃ cardiovascular effects. For example, it is beginning to be recognized that O₃-induced lung injury and permeability changes, as well as O₃-induced alterations in hemodynamics, may lead to cardiovascular system effects. Also, interactions of O₃ with ELF lipids and surfactants result in lipid ozonation products and reactive oxidant species (ROS) that have the potential to penetrate the epithelial barrier and to initiate toxic effects on the cardiovascular system.

Earlier studies in rats indicated a potential role for platelet activating factor (PAF) in the O₃-induced inflammatory response. Recent observations of O₃-induced generation of oxysterols and β-epoxides from cholesterol in surfactant also suggest that these lipid ozonation products (like lysophospholipids) might exhibit PAF-like activity and contribute to clotting and thrombolytic effects in the cardiovascular system.

Other studies carried out using isolated perfused rat lung model (Delaunois et al., 1998) indicate inhibition of pulmonary mechanical reactivity to bronchoconstrictors and persistent vasoreactivity of the vascular bed upon exposure to O₃ (0.4 ppm for 4 h). Newer studies have also now found that acute (less than 5 h) 0.1 ppm O₃ exposure caused decreased heart rate in young, but not old, rats (Arito et al., 1997).

Only one human experimental O₃ exposure study (Gong et al., 1998) evaluated potential cardiovascular effects in normal and hypertensive adult males. Various cardiovascular and hemodynamic parameters were monitored upon exposure to O₃ (0.3 ppm for 3 h) with intermittent exercise. No significant O₃-induced differences were observed in ECG, heart rate, or blood pressure in either normal or hypertensive subjects. An overall increase in myocardial work and impairment in pulmonary gas exchange was observed, however, that might be clinically important in patients with preexisting cardiovascular impairment with or without concomitant lung disease.

8.4.3 Biological Basis for O₃ Health Effects Assessment

The scientific knowledge base gained from animal toxicological studies and experimental studies of human volunteers in clinical settings (discussed in Chapters 4, 5, and 6 of this document) has notably expanded the knowledge base beyond that available at the time of 1996 O₃ AQCD. This section provides an interpretive integration of key findings derived from the experimental knowledge base that can aid in evaluating the biological plausibility for health effects observed in O₃-related epidemiological studies discussed later in this chapter.

Animal-to-Human Extrapolation Issues

The physiological and biochemical observations summarized in Table 8-1 provide an overview of the main types of acute O₃-induced health effects as demonstrated by toxicological studies of humans and animals. This table was generated from those experimental studies (see Annexes for Chapters 5 and 6 for experimental details) that utilized exposure regimens of varied concentration and duration that are environmentally relevant. As noted above in Section 8.4.1, observed acute O₃ effects are mostly transient (generally persisting ≤ 24 to 48 h) and attenuate over time. However, the time-line for resolution of many of these physiological and biochemical parameters in normal and human subjects with underlying cardiopulmonary diseases follow different profiles, as presented in Figure 8-3. Alterations in the cellular and molecular profiles observed in human airway epithelium upon acute exposure to O₃ evolve over time (Figure 8-3), and the knowledge of this profile is valuable in assessing biological plausibility to integrate across evidence for various health endpoints.

Basic similarities in physiological, biochemical, and pathological processes that exist between human and other mammalian species are derived from the high degree of genome sequence homology that exists across various species. This homology reinforces the significance of knowledge gained on the initiation, progression and treatment regimes for various disease processes across animal species. This homology is also apparent in acute O₃-induced effects, especially on the respiratory tract of human and animal species as presented in Table 8-1 and Figures 8-3 and 8-4. The commonality of phenomenon observed in humans and rats with regard to respiratory system effects (in terms of spirometry, ventilatory response, host defense and inflammation) and their attenuation adds strength to animal-human extrapolations. Such similarities observed at higher levels of cellular organization (neutrophilic inflammation,

Table 8-1. Acute O₃-Induced Physiological and Biochemical Changes in Human and Animals

Physiological/Biochemical Alterations	Human Exposure Studies^{1,2}	Animal Toxicology Studies^{3,4}
Pulmonary Function:	↓ FEV ₁ ↑ Frequency of breathing (rapid, shallow) ↓ FVC (cough, breathing discomfort, throat irritation, wheezing) Mild bronchoconstriction	↑ Frequency of breathing (rapid, shallow) ↓ FVC
Airway Responsiveness:	↑ (neuronal involvement) Change in lung resistance	↑ (vagal mediation) Change in lung resistance
Inflammation:	Yes ↑ inflammatory mediators	Yes ↑ inflammatory mediators
Reactive Oxygen Species:	↑	↑
Host Defense:	↑ particle clearance ↑ permeability ↓ AM phagocytosis	↑ particle clearance ↑ permeability ↓ clearance of bacteria ↑ severity of infection ↑ mortality & morbidity
Lung Injury:		
Morphology:	Yes	Yes
Susceptibility:	Age, Interindividual variability Disease status Polymorphism in certain genes being recognized	Species-specific differences Genetic basis for susceptibility indicated
Cardiovascular Changes:	Impairment in arterial O ₂ transfer Ventilation-perfusion mismatch (suggesting potential arterial vasoconstriction) ↑ rate pressure product ⁵ ↑ myocardial work ⁵	Heart rate ↓ core body temperature ↑ atrial natriuretic factor Role for platelet activity factor (PAF) indicated Increased pulmonary vascular resistance

¹ Controlled chamber exposure studies in human volunteers were carried out for a duration of 1 to 6.6 h with O₃ concentration in the range of 0.04-0.40 ppm with intermittent exercise. See text for discussion of O₃ levels shown to cause different types of effects listed.

² Data on some biochemical parameters obtained from in vitro studies of cells recovered from BALF.

³ Responses were observed in animal toxicology studies with exposure for a duration of 2 to 72 h with O₃ concentration in the range of 0.1 to 2.0 ppm.

⁴ Various species (mice, rat, guinea pigs and rabbit) and strains.

⁵ In hypertensive subjects.

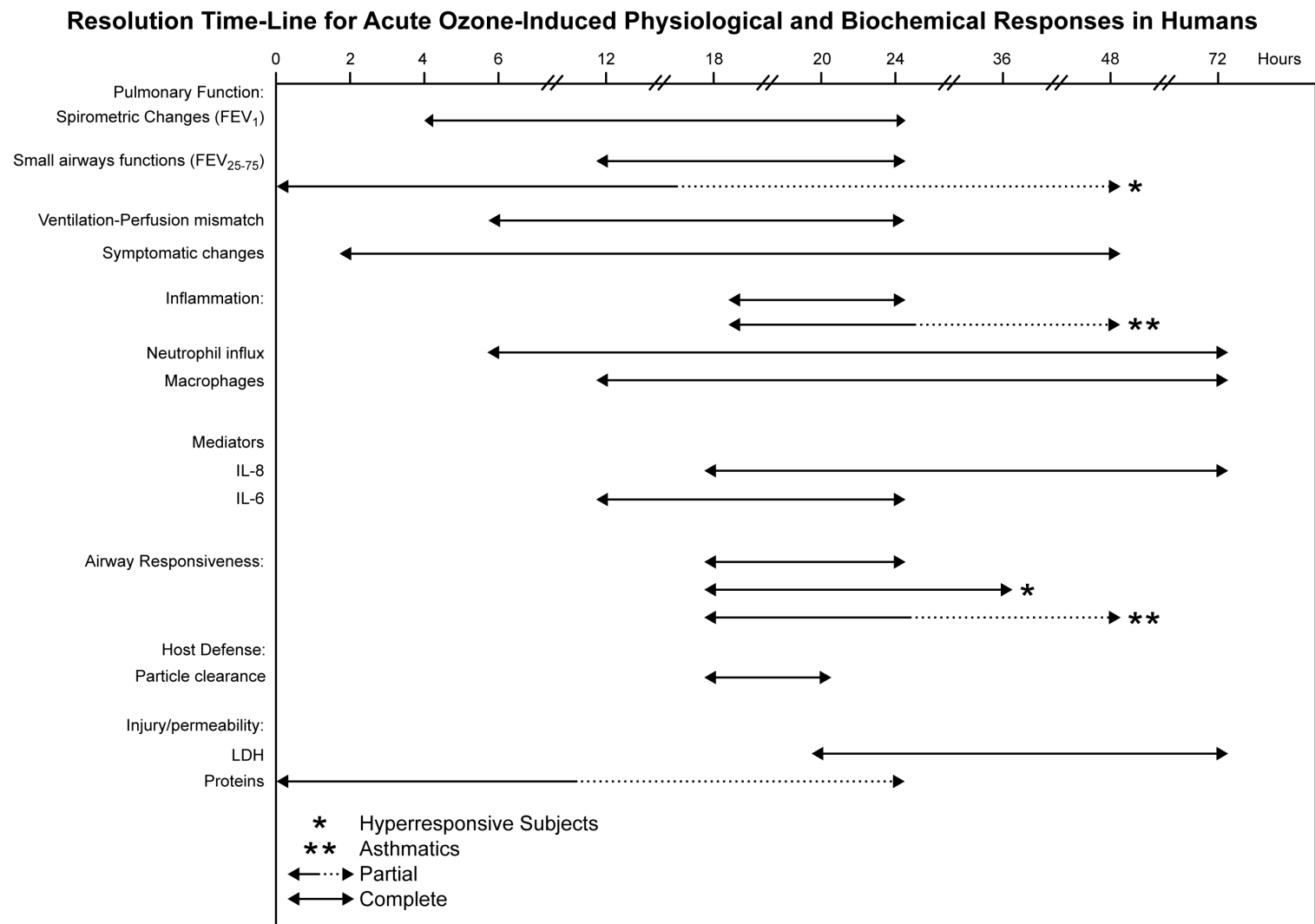


Figure 8-3. Resolution time-line for the respiratory, physiological, and biochemical parameters are derived from studies reported in Chapter 6 and Chapter 6 Annex.

Postulated Cellular and Molecular Changes in Human Airway Cells In Response to Acute Exposure to Ozone

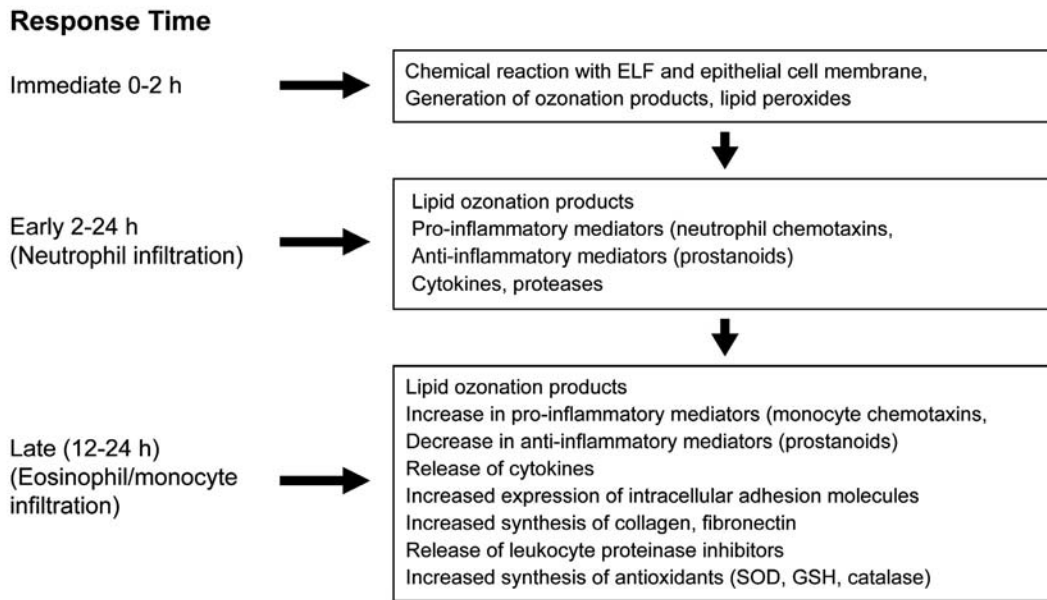


Figure 8-4. Acute (1-8 h) O₃ exposure-induced cellular and molecular changes and timelines for their resolution depicted here are derived from the data reported in Leikauf et al. (1995) and Mudway and Kelly (2000).

macrophage phagocytosis processes) have increased the value and importance of animal studies in generating important data that are impossible to collect in human studies but which may corroborate both human clinical and epidemiologic studies.

Extrapolation of results derived from laboratory animal studies to humans involves a combination of dosimetry, end point homology and species sensitivity, particularly in the case of exposure and health outcome analyses. However, existing extrapolation models have not yet been sufficiently validated to allow for highly confident quantitative animal-to-human extrapolation for O₃ effects. Still, qualitative extrapolation appears to be reasonable for some endpoints. For example, based on inflammatory markers in BALF, a 2 ppm O₃ exposure in nonexercising rats approximates to a 0.4 ppm exposure in exercising humans (Hatch et al., 1994). This observation lends support to the use of some of the animal toxicology data derived from relatively high O₃ concentration exposure regimens in understanding putative molecular changes likely to be associated with acute O₃ exposure in humans.

The time courses for induction and resolution of various types of O₃ effects experimentally induced in laboratory settings (see Figures 8-3 and 8-4) may provide a basis for reasonable projection of likely lag times for observing associations between ambient O₃ concentrations/exposures and analogous effects in epidemiologic studies of human populations. For example, it would not be unreasonable to expect that pulmonary function decrements could be detected epidemiologically within lags of 0 (same day) or 1 or 2 days following O₃ exposure, given the rapid onset of spirometric changes and their persistence for 24 to 48 hours among hyperresponsive human subjects in clinical studies. On the other hand, although asthmatic individuals may begin to experience symptoms soon after O₃ exposure, it may take a day or so, coupled with breathing difficulties secondary to O₃-induced hyperresponsiveness that may persist for 2 to 3 days, for members of that sensitive population group to seek medical attention. This could possibly be reflected by epidemiologic observations of significantly increased risk for asthma-related ED visits or hospital admissions with 1- to 3-day lags, or, perhaps, enhanced distributed lag risks (combined across 3 days) for such morbidity indicators. Analogously, one might project increased mortality within 0 to 3 day lags as a possible consequence of O₃-induced increases in prothrombotic agents, such as platelet aggregating factor (PAF) or lipid ozonation products (e.g., cholesterol epoxide) arising from inflammation cascades occurring within 12 to 24 hours of O₃ exposure (see Figure 8-4).

Similarly, the presence of apparent O₃-induced lesions in animals from chronic O₃ exposure studies (12 to 24 months) indicate morphological alterations that may analogously occur in humans with long-term (months, years) chronic exposure to relatively high O₃ levels, but specific O₃ concentrations and exposure patterns that may produce analogous alterations in human lungs remain to be substantiated.

8.4.4 Epidemiologic Evidence

Epidemiologic evidence available at the time of the 1996 O₃ AQCD indicated potential health effects associated with acute ambient O₃ exposures. The 1996 O₃ AQCD further stated that only suggestive epidemiologic evidence existed for health effects of chronic ambient O₃ exposure in the population, and this was partly due to an inability to isolate potential effects related to O₃ from those of other pollutants, especially PM (U.S. Environmental Protection Agency, 1996a). The availability of numerous recent epidemiologic studies, with some studies

designed explicitly to evaluate ambient O₃ exposure effects (as discussed in detail in Chapter 7 of this document), makes it possible to better assess the health effects of both acute and chronic ambient O₃ exposure. There exist a number of issues and uncertainties associated with the interpretation of O₃ health effects evidence in epidemiologic studies, which are discussed in Sections 7.1.3 and 7.6 of Chapter 7 in this document. These include the use of various indices to represent O₃ exposure, exposure measurement errors (i.e., differences between ambient concentrations and personal exposure), lag period between exposure and effect, potential confounding by temporal and meteorological factors, potential confounding by copollutants, the concentration-response function, and heterogeneity of O₃ health effects.

The recent epidemiological studies that have been conducted in areas across the United States and Canada, as well as in Europe, Latin America, Australia and Asia, are reviewed in Chapter 7 (for details see Annex to Chapter 7). Present discussions in this integrative synthesis chapter focus on findings from studies carried out in the United States and Canada. Important features of these epidemiologic studies (e.g., study location, study population, mean O₃ concentrations, health risk estimates) are summarized in Chapter 8 Appendix Tables 8A-2 to 8A-5. The studies are ordered in those tables by the 98th and 99th percentile values of the calculated 8-h max O₃ data for the full study period. The 98th and 99th percentile values were selected as they represent a high concentration that roughly approximates a 4th maximum concentration, depending on the study period length. For studies that did not have this data available, their ordering was approximated based on the mean O₃ concentrations observed in the study.

8.4.4.1 Acute Ozone Exposure Studies

Numerous epidemiological studies carried out over the past decade have added evidence to the knowledge base assessed in the 1996 O₃ AQCD, which included both (a) individual-level summer camp and exercise studies that established a relationship between ambient O₃ exposure and lung function decline and (b) aggregate population studies that suggested positive relationships for O₃-related respiratory morbidity endpoints (i.e., respiratory ED visits and hospitalizations). The new studies reviewed in Chapter 7 in this document include numerous field/panel studies and population studies from various regions in the United States and abroad. In field/panel studies on the effects of air pollution exposure, the most common health outcomes

measured were lung function and respiratory symptoms. The population studies examined daily ED visits, hospital admissions, and mortality data.

Field/Panel Studies of Acute Exposure Effects

Pulmonary Function and Respiratory Symptoms

Many of the new field/panel studies reviewed in Chapter 7 and the controlled human exposure studies reviewed in Chapter 6 of this document provide additional data supporting two key findings reported in the 1996 O₃ AQCD. One key finding was that acute O₃ exposure was associated with a significant decline in lung function parameters. Ozone-related lung function decrements were most notable in children, as indicated by the results from the meta-analysis of summer camp studies by Kinney et al. (1996). Similar responses were reported for children and adolescents exposed to O₃ in ambient air in the summer camp studies as were found with exposure to O₃ in clean air for 1 to 2 h while exercising in clinical studies. A second key finding was that adults who work or exercise outdoors were also found to be vulnerable to O₃-associated declines in lung function due to their greater exposure to O₃ during periods of increased physical activity.

In a number of newly available field/panel studies, FEV₁ was measured in panels of exercising children, outdoor workers, and adult hikers exposed to ambient O₃ while experiencing elevated exertion levels. Collectively, the results of the new studies (discussed in Section 7.2.3.1) confirm and extend the findings from analogous field/panel studies and experimental controlled human exposure studies assessed in the 1996 O₃ AQCD. Acute O₃ exposures prolonged over several hours and combined with elevated levels of exertion or exercise were found to have magnified effects on lung function, as evaluated in terms of FEV₁. Only studies carried out in the U.S. and Canada that evaluated lung function changes are discussed here (see Appendix Table 8A-2 for details).

Brauer et al. (1996) measured lung function in berry pickers in British Columbia, Canada. The mean ambient 1-h max O₃ was 40.3 ppb (SD 15.2). Significant O₃-related decreases in FEV₁ were observed in the morning and afternoon, but not across the day. Another field study on adult hikers by Korrick et al. (1998) observed that the mean FEV₁ decline in healthy hikers was marginal, but a 4-fold greater decline was observed in hikers with asthma and wheeze. Both studies, however, did not evaluate the possible role of confounding by copollutants. Other

studies from Germany and Mexico City provided additional supportive evidence that healthy adults with prolonged periods of exposure and exertion experienced declines in lung function.

Several panel studies that examined lung function changes in children, including a few U.S. studies (Neas et al., 1995, 1999; Linn et al., 1996), observed small declines in FEV₁ or in peak expiratory flow (PEF). It should be noted, however, that the group mean responses reported in panel or population studies may mask interindividual differences, e.g., larger lung function changes in sensitive populations (as discussed in Section 8.4.2 and in more detail in Chapter 6).

Panel studies that evaluated lung function decrements in asthmatic children collectively indicated O₃-induced decrements, although most of the individual-study estimates were not statistically significant. Results from a multicity study of asthmatic children from eight urban areas in the U.S. (Mortimer et al., 2002) suggest a small, but significant decline in PEF based on a multiday distributed lag model. Of all the pollutants examined, including O₃, PM₁₀, NO₂, and SO₂, only O₃ was found to be associated with morning PEF. The small change in lung function observed likely reflects the low mean O₃ levels across the eight cities studied (8-h avg O₃ of 48 ppb). Overall, acute ambient O₃ exposure was observed to be associated with lung function decrements at O₃ concentrations ranging down to a 98th percentile 8-h max of 55 to 60 ppb for potentially vulnerable and susceptible populations, as shown in Appendix Table 8A-2.

In addition to pulmonary function, the majority of asthma panel studies also evaluated the relationship between O₃ and respiratory symptoms in asthmatic children. The results obtained from these studies show some inconsistencies, with some indicating significant positive associations and other well-conducted, albeit relatively smaller, studies not finding such effects. Overall, however, the multicity study by Mortimer et al. (2002) and several credible single-city studies (e.g., by Gent et al., 2003) indicate a fairly robust positive association between ambient O₃ concentrations and increased respiratory symptoms in asthmatics.

Airway Inflammation

A few epidemiologic studies, conducted in Germany and Mexico City, examined the effect of acute ambient O₃ exposure on airway inflammation in children. Results indicated that associations were found largely on days or in study locations having the higher O₃ concentrations (1-h max O₃ of approximately 100 ppb) among those present. At locations with

lower O₃ levels, no association was detected between upper airway inflammation and ambient O₃ concentrations across the entire study period (Kopp et al., 1999). However, a more detailed analysis showed that the first significant O₃ episode of the summer was associated with increases in inflammatory markers, but subsequent and even higher O₃ episodes had no effect, suggesting an adaptive response.

Cardiophysiological Endpoints

A few recent air pollution epidemiology studies have also evaluated the potential effects of ambient O₃ on cardiophysiological endpoints. Based on the discussions presented in Chapter 7 (see Section 7.2.7), the limited data available at this time are suggestive of altered heart rate variability (HRV), ventricular arrhythmias, and MI incidence possibly being associated with acute O₃ exposures; but, overall, these findings must currently be considered inconclusive.

Population Studies of Acute Exposure Effects

Emergency Department Visits and Hospital Admissions

Many time-series studies reviewed in the 1996 O₃ AQCD indicated positive associations between O₃ air pollution and increased respiratory hospital admissions and ED visits, providing strong evidence for a relationship between O₃ exposure and increased exacerbations of preexisting respiratory disease in the general public at 1 h-max O₃ concentrations <0.12 ppm (120 ppb). Analyses of data for the northeastern United States suggested that O₃ air pollution was associated with a substantial portion (on the order of 10 to 20%) of all summertime respiratory hospital visits and admissions. Several new studies have been published in the past decade examining temporal associations between O₃ exposures and emergency department visits and hospital admissions for respiratory diseases.

Studies conducted in the United States and Canada examining the association between acute O₃ exposure and asthma ED visits are summarized in Appendix Table 8A-3. The risk estimates presented in Appendix Table 8A-3 include those using all-year and warm-season only O₃ data. A majority of the studies conducted in the U.S. and Canada examined the effects of O₃ during the warm season only. An analysis of respiratory disease ED visits in Atlanta, GA over an 8-year period indicated a significant positive association between ambient O₃ concentrations and asthma visits during the warm season (Peel et al., 2005). The mean 8-h max O₃ from March

to November was 55.6 ppb (SD 23.8) in this study. A study conducted in Saint John, Canada observed a positive association between O₃ and asthma emergency visits during the warm season in individuals 15 year or older at even lower O₃ concentrations (mean 1-h max O₃ of 41.6 ppb) (Stieb et al., 1996). However, several other studies, mostly using all-year data, reported no association between O₃ and ED visits for respiratory causes. In general, ambient O₃ concentrations were associated with ED visits for asthma in warm-season only analyses.

Relationships between ambient O₃ and respiratory-related hospital admissions have also been evaluated in various epidemiologic studies carried out in the United States and abroad. Results from studies in the U.S. and Canada are summarized in Appendix Table 8A-4. Some of these studies evaluated potential confounding by season or copollutants of the O₃ effects on respiratory hospitalizations. A multicity study by Burnett et al. (1997a) evaluated the relationship between ambient O₃ and respiratory hospitalizations in 16 Canadian cities over an 11-year period. Seasonal analyses indicated that O₃ effects were only observed during the warm season. As shown in Appendix Table 8A-4, most studies in the United States and Canada indicated a consistent positive association between ambient O₃ concentrations and respiratory hospital admissions in the warm season, including studies with 98th percentile 8-h max O₃ levels as low as about 50 ppb. Analyses of all-year data were mostly positive, but nonsignificant for either total respiratory or asthma hospital admissions. The studies that found positive associations based on analyses using various lag periods found the association to be strong with short lag periods (0 to 2 day), suggesting likely short-term O₃ effects on asthma hospital admissions.

Some new epidemiologic studies have also evaluated the associations between ambient O₃ concentrations and cardiovascular-related hospitalizations (see Section 7.3.4 for details). The epidemiologic evidence for cardiovascular morbidity is much more mixed than for respiratory morbidity, with only one of several U.S./Canadian studies showing statistically significant positive associations of cardiovascular hospitalizations with warm-season O₃ concentrations. Studies from Taiwan and Hong Kong also provided suggestive evidence of an association, but most of the available European and Australian studies (all of which conducted all-year O₃ analyses) did not find an association between short-term O₃ concentrations and cardiovascular hospitalizations.

Acute Effects of Ozone on Mortality

The 1996 O₃ AQCD noted that an association between O₃ concentrations and daily mortality for areas with high O₃ levels (e.g., Los Angeles) was suggested by a few studies, but the magnitude of such an effect was unclear. However, due to the limited number of available studies and uncertainties regarding weather model specifications, no conclusive evidence of O₃-mortality associations was found in the 1996 O₃ AQCD. In contrast, newly available large multicity studies designed specifically to examine the effect of O₃ on mortality have since provided much more credible information. Numerous recent air pollution epidemiologic studies have been published that examined acute O₃ exposure effects on all-cause mortality in the U.S. and elsewhere. Studies from the U.S. and Canada are presented in Appendix Table 8A-5.

Five new multicity studies, three conducted in the United States and two in Europe, found small, but very precise (extremely narrow 95% CIs) positive associations for increased mortality risk using all-year ambient O₃ data and warm-season only data. Most notably, the largest of the multicity studies conducted in the U.S. that evaluated the association between ambient O₃ concentrations and excess mortality risk, the NMMAPS 95 U.S. communities study by Bell et al. (2004), observed a significant positive association between O₃ and all-cause mortality using all available data (55 communities with all-year data and 45 communities with warm-season only data). When the data were restricted to the warm season only, a slightly smaller, but still significant, O₃ effect was observed. Analyses suggested that the O₃-related excess mortality risk estimate was not confounded by exposure to PM. Another multicity study by Schwartz (2005) examining the association between O₃ and all-cause mortality observed a slightly greater excess risk with warm-season only data compared to all-year data using a case-crossover study design.

Mortality risk estimates from single-city studies were also positive, but many were not statistically significant. However, as in the case of ED visits and hospitalizations, restricting data to the warm season resulted in mortality risk estimates that were more positive and stronger. Meta-analyses of single-city mortality studies (Bell et al., 2005; Ito et al., 2005; Levy et al., 2005) observed that O₃-mortality risk estimates from warm-season only analyses were approximately 2-fold greater than those from all-year analyses. As shown in Appendix Table 8A-5, the recent multicity and single-city studies generally show consistent positive and significant associations between acute O₃ exposure and all-cause mortality in studies with 98th percentile 8-h max O₃ values of 80 to 85 ppb and above.

Some studies examined specific subcategories of mortality; however, most of these studies had limited statistical power to detect associations due to the small daily counts for the specific causes of death. In studies using broad categories of death, such as cardiopulmonary mortality, significant associations with acute O₃ exposure have been observed. For example, an analysis of the NMMAPS data by Huang et al. (2005) examined the association between ambient O₃ concentrations and cardiopulmonary mortality in 19 large U.S. cities during the warm season. Ozone was positively associated with excess risk in cardiopulmonary mortality, even after adjusting for PM and heat waves. Also, the meta-analysis by Bell et al. (2005) examined the excess risk estimates for cardiovascular and respiratory causes and observed a slightly greater risk of cardiovascular mortality compared to all-cause and respiratory disease mortality. It is notable that only one single-city study, from among approximately 30 studies (including 2 multicity ones), did not find a positive association between ambient O₃ concentrations and increased cause-specific cardiovascular mortality, with both the multicity and several single-city associations being statistically significant. It should be noted that these analyses of cause-specific mortality do not provide direct evidence on possible causal mechanisms as they often do not incorporate information on the contributing causes of death. For example, if an individual has been suffering from a major cardiovascular disease, his/her death may be misclassified as cardiovascular, even if a respiratory condition causes the death.

8.4.4.2 Chronic Ozone Exposure Studies

There were only a limited number of studies reported in the 1996 O₃ AQCD that addressed potential health effects of long-term ambient O₃ exposures. The few available epidemiological studies that had attempted to associate chronic health effects in humans with long-term O₃ exposure provided very limited suggestive evidence that such a linkage may exist. Several longitudinal epidemiologic studies carried out during the past decade have further evaluated the potential effects of chronic (several weeks to many years) O₃ exposure on lung function, respiratory symptoms, lung inflammation, asthma prevalence, and birth defects. The strongest evidence is for seasonal effects of extended O₃ exposures on lung function in children, i.e., reduced lung function growth being associated with higher ambient O₃ levels. Based on the available data at this time, however, no clear conclusions can be drawn regarding the relationship between chronic O₃ exposure and the other health outcomes.

Few studies have evaluated effects of long-term O₃ exposure on the incidence of cancer and/or mortality. Uncertainties regarding possible exposure periods of relevance and, also, inconsistencies across mortality outcomes and gender raise concerns regarding plausibility. The largest and most representative U.S. study, by Pope et al. (2002), observed positive but nonsignificant associations between long-term O₃ concentrations and all-cause, cardiopulmonary, and lung cancer mortality. Thus, there is currently little evidence for potential relationships between chronic O₃ exposure and increased incidence of cancer and mortality.

8.4.4.3 Summary of the Epidemiologic Evidence

Assessment of the evidence from epidemiologic studies for various health outcomes has focused on several considerations that are important in forming judgments as to the likely causal significance of the observed associations. As discussed in Section 7.1.2, these include the *strength* of the epidemiologic evidence, including the magnitude and precision of reported O₃ effect estimates and their statistical significance, and the *robustness* of the effects associations, or stability in the effect estimates after considering a number of factors, including alternative models and model specifications, potential confounding by copollutants, as well as issues related to the consequences of measurement error. *Consistency* of the effects associations involves looking across the results of multiple- and single-city studies conducted by different investigators in different places and times.

In general, when associations are strong in terms of yielding large relative risk estimates, it is less likely that the association could be completely accounted for by a potential confounder or some other source of bias. With associations that yield small relative risk estimates, it is especially important to consider potential confounding and other factors in assessing causality. Across the range of different health outcomes, effect estimates for many health associations reported with O₃ are generally small in size and could thus be characterized as weak. For example, effect estimates for associations with mortality generally range from 0.5 to 5% increases per 40 ppb increase in 1-h max O₃ or equivalent, whereas associations for hospitalization range up to about 50% increases per standardized O₃ increment. Of particular note are several multicity studies that have yielded relative risk estimates for associations between short-term O₃ exposure and mortality or morbidity that, although small in size, have great precision due to the statistical power of the studies. For example, corresponding summary

estimates for mortality in large U.S. multicity studies ranged from 0.5 to 1.0% per standardized 24-h increment, with some notable heterogeneity across cities. Thus, such associations are strong relative to the precision of the studies; that is, the associations were strong enough to have been reliably measured by the studies such that many of the associations can be distinguished from the null hypothesis with statistical confidence (e.g., at $p \leq 0.05$).

As discussed in Section 7.6.4.2, the associations reported between short-term O₃ exposure and various health outcomes were generally robust to adjustment for copollutants, including PM_{2.5} and sulfates, which are correlated with O₃ concentrations in many areas in the United States. More limited evidence is available on the effects of different modeling strategies on associations with O₃; however, these studies indicate that O₃ associations may be robust to various model specifications for temporal trend adjustment. Risk estimates for O₃ were generally more sensitive to the use of alternative weather models than to adjustment for temporal trends, due to the temperature-dependent nature of O₃ formation in the atmosphere.

In considering results from the multicity studies and single-city studies in different geographic areas, there was general consistency in effects of short-term O₃ exposure on mortality, respiratory hospitalization and other respiratory health outcomes. Some variation in effects was observed and that may be attributable to differences in relative personal exposure to O₃, which is affected by factors such as air conditioning prevalence and activity patterns, as well as varying concentrations and composition of copollutants present in different regions. Thus, consideration of consistency or heterogeneity of effects is appropriately understood as an evaluation of the similarity or general concordance of results, rather than an expectation of finding quantitative results within a very narrow range.

Taken together, the epidemiologic evidence shows strength, robustness, and consistency in associations between short-term O₃ exposure and a range of respiratory morbidity health outcomes, including increased respiratory-related hospital admissions and asthma-related ED visits. There is also relatively strong epidemiologic evidence for associations between short-term O₃ exposure and all-cause mortality. There is less strong, but nevertheless highly suggestive evidence for effects related to cardiovascular morbidity. Moreover, consistently positive associations were found for O₃-related cardiovascular mortality across approximately 30 studies, with two well-conducted multicity studies in the United States and Europe yielding small but statistically significant positive associations. In contrast, generally inconclusive

evidence currently exists for associations between long-term chronic O₃ exposure and either morbidity or mortality.

8.5 ASSESSMENT OF POTENTIAL THRESHOLDS

A key issue in assessment of air pollution health effects is whether any thresholds can be discerned for various types of health effects. Animal toxicology studies reviewed in Chapter 5 appear to indicate a linear concentration-response relationship for O₃ exposure effects. Lowest concentrations found to have caused statistically significant health effect changes, e.g., altered pulmonary function, airway hyperresponsiveness, lung inflammation, increased lung tissue permeability, or altered host defense, in various laboratory animal species are listed in Appendix Table 8A-6. Some of these concentrations are rather high, but others appear to be more environmentally relevant, including at least one study where the induction of airway hyperresponsiveness was seen at O₃ levels distinctly below current ambient O₃ standards.

One recent controlled human exposure study, reviewed in Chapter 6, included exposures to low, clearly environmentally-relevant, O₃ concentrations for a group of healthy young adults. Adams (2006) reported that following a 6.6 h exposure to 0.08 ppm O₃, group mean O₃-induced FEV₁ decreases (-6.1%, square-wave; -7.0%, triangular) and symptom responses in healthy adults were statistically significantly greater than after 0.04 and 0.06 ppm O₃ exposures. During a 6.6 h exposure to 0.06 ppm O₃, group mean FEV₁ responses diverged from responses for filtered-air and 0.04 ppm O₃ by 5.6 h, but did not reach statistical significance at $p \leq 0.05$ by the end of the exposure period (Adams, 2006). Exposures to 0.04 ppm O₃ for 6.6 h produced group mean FEV₁ responses quantitatively similar to those observed for filtered-air exposures (Adams, 2002, 2006). These results suggest that acute exposure to O₃ can have effects in healthy young adults with exposure levels below 0.08 ppm O₃ (6.6 hr average) although no statistically significant effects were observed at the lowest exposure level of 0.04 ppm O₃. It is important to note, however, that there is considerable interindividual variability in responses, as was illustrated in Figures 8-1B and 8-2, and some evidence of FEV₁ decrements >10% for some subjects at 0.06 ppm. It is also important to note that this type of study generally includes healthy individuals and more subtle health measures, in contrast to population-based epidemiologic studies.

The potential for the existence of threshold levels has been explored in several recent epidemiologic studies, as described in more detail in Section 7.6.5. A U.S. multicity panel study by Mortimer et al. (2002) examined associations of ambient O₃ concentrations with changes in lung function and respiratory symptoms in asthmatic children. The mean 8-h avg O₃ (10 a.m.-6 p.m.) was 48 ppb, with less than 5% of study days exceeding 80 ppb. Analyses indicated that ambient O₃ was significantly associated with decrements in PEF, both when using all data available and after restricting data to days when 8-h avg O₃ concentrations were <80 ppb. A slightly larger effect estimate was observed using the restricted data set. Ozone-related increased incidence of respiratory symptoms also persisted after eliminating data when 8-h avg O₃ exceeded 80 ppb.

Examination of the shape of the concentration-response function in several time-series studies of O₃-related ED visits and hospitalizations has provided some indications of an effect threshold. In a study of ED visits for asthma in St. John, Canada, the O₃ effect observed in the >15 years age group was apparent only when 1-h max O₃ data above the 95th percentile value of 75 ppb were included (Stieb et al., 1996). In another study conducted in Toronto, Canada, the association between ambient O₃ and all-age respiratory hospital admissions only became apparent above a daily 1-h max O₃ level of approximately 30 ppb (Burnett et al., 1997b). In London, England, possible thresholds for O₃ effects on respiratory hospitalizations were observed at 40 to 50 ppb for 8-h max O₃ and 50 to 60 ppb for 1-h max O₃ (Ponce de Leon et al., 1996). However, other studies have reported a monotonic increase in the concentration-response function throughout the range of ambient O₃ concentrations, suggesting a lack of any evident threshold for O₃-related effects on respiratory hospitalizations and asthma ED visits (Burnett et al., 1997a; Jaffe et al., 2003; Petroeschovsky et al., 2001; Tenías et al., 1998).

In the 95 U.S. communities study, Bell et al. (2004) reported that the risk estimate for excess mortality associated with short-term O₃ exposure was slightly smaller but remained statistically significant when only using data from days with 24-h avg O₃ concentrations below 60 ppb. A more formal threshold analyses recently reported by Bell et al. (2006) for 98 U.S. communities, including the same 95 communities in Bell et al. (2004), indicated that if a population threshold existed for mortality, it would likely fall below a 24-h avg O₃ concentration of 15 ppb. Other analyses by Kim et al. (2004) investigating the presence of a threshold in O₃-mortality effects in Seoul, Korea estimated threshold values of 28 ppb in 1-h max O₃ when using

all-year data and 45 ppb for summer-only data. None of the other pollutants examined, including PM₁₀, SO₂, NO₂, and CO, had a nonlinear association with mortality. Results from the Kim et al. study suggested that if a threshold truly exists, the use of log-linear models may underestimate the O₃ effect on mortality at levels above the threshold.

Taken together, the available evidence from toxicologic, clinical and epidemiologic studies suggests that no clear conclusion can now be reached regarding possible threshold levels for O₃-induced effects. The controlled human exposure studies demonstrate notable variability in responsiveness among healthy subjects, even at low (≤ 0.08 ppm) O₃ exposure levels, making it infeasible to suggest any clear threshold for the heterogeneous general population from these results. As discussed in Section 7.6.5, there are limitations in epidemiologic studies that make discerning thresholds in populations difficult, including low data density in the lower concentration ranges, the possible influence of measurement error, and interindividual differences in susceptibility to O₃-related effects in populations. Nevertheless, the limited clinical and epidemiologic evidence suggests that if a population threshold level does exist, it is likely near the lower limit of ambient O₃ concentrations in the United States.

8.6 BIOLOGICAL PLAUSIBILITY AND COHERENCE OF EVIDENCE FOR OZONE-RELATED HEALTH EFFECTS

This section integrates findings from epidemiologic studies with toxicologic and mechanistic information obtained from controlled human exposure studies and animal toxicology studies for major health endpoints reported to be associated with either short- or long-term exposure to ambient O₃. The section focuses on evidence related to two key considerations for drawing conclusions about causality – biological plausibility and coherence of the evidence. Evaluation of the *biological plausibility* of the O₃-health effects associations observed in epidemiologic studies reflects consideration of both exposure-related factors and dosimetric/toxicologic evidence relevant to identification of potential biological mechanisms. Similarly, *coherence* of health effects associations reported in the epidemiologic literature reflects consideration of information pertaining to the nature of the various respiratory- and cardiovascular-related morbidity and mortality effects and physiological endpoints evaluated in human and animal toxicologic studies. The discussion in each subsection summarizes pertinent

key information and then discusses plausibility based on toxicological studies of effects attributed to O₃ exposures in epidemiologic studies.

For this assessment, the ensuing discussion on biological plausibility and coherence considers (a) the extent to which available epidemiological evidence logically ties together a range of relevant health endpoints (from cardiopulmonary physiological changes to morbidity to mortality) and (b) the extent to which available toxicological and biochemical evidence supports plausible causal relationships for observed epidemiological associations for specific types of health outcomes, i.e., how well do key epidemiologic findings compare with reasonably hypothesized or experimentally demonstrated biological mechanisms of action.

8.6.1 Acute Ozone Exposure-Induced Health Effects

Respiratory Health Effects

As noted in Section 8.4.3, several new epidemiologic (field/panel) studies show positive associations between short-term exposure to ambient O₃ and human respiratory effects. These health effects, as evaluated, include reductions in lung function, increased use of asthma medication, and increased hospitalization, especially among individuals with asthma or certain known cardiopulmonary diseases (see Chapter 7). The patterns of experimentally demonstrated physiological and biochemical alterations noted earlier (see Table 8-1 and Figures 8-3 and 8-4) support certain hypotheses regarding underlying pathological mechanisms in the development of respiratory effects reported in the epidemiologic studies. Some of these mechanisms (see Table 8-1) include (a) decrements in lung function (capacities and volume), (b) bronchoconstriction, (c) increased airway responsiveness, (d) airway inflammation, (e) epithelial injury, (f) immune system activation, (g) host defense and (h) individual sensitivity factors such as age, genetic susceptibility and the extent of tolerance resulting from previous exposures. The time sequence, magnitude, and overlap of these complex events, both in terms of development and recovery (see Figures 8-3 and 8-4), indicate the difficulties associated with the interpretation of biological plausibility associated with the cardiopulmonary health effects.

Controlled human exposure studies have clearly demonstrated the following three types of respiratory responses to acute O₃ exposures: (1) irritative cough and substernal chest pain upon inspiration; (2) decrements in FVC and FEV₁ due to decreased inspiratory capacity rather than airways obstruction and (3) neutrophilic inflammation of the respiratory tract. Increased

sensitivity (susceptibility) to these effects was observed even among a carefully selected homogeneous study population. The sources of this heterogeneity are not fully understood. As discussed in the earlier section, changes in baseline levels of various responses, the lag in the recovery phase and the role of residual defects in these mechanisms in hyperresponsive individuals suggest the potential for increased health effects in compromised individuals with preexisting cardiopulmonary diseases. Recent research has emphasized further characterization of the mechanisms and consequences of O₃-induced pulmonary function and inflammatory responses. In addition, animal studies indicate morphological changes associated with acute O₃ exposures.

Ozone-induced altered breathing patterns (rapid shallow breathing) observed in controlled human exposure studies and animals occur without significantly affecting minute ventilation, suggesting compensatory changes in breathing pattern. Such a shift in breathing pattern diminishes deep lung penetration of O₃. Breathing pattern is modulated by changes in peripheral mechanisms, such as direct or indirect stimulation of lung receptors and bronchial C-fibers. The activity of these afferents is integrated with input from sensory pathways and thus determines the depth and frequency of breathing. Stimulation of bronchial C-fibers along with inhibition of inspiration through local axon reflexes can induce neurogenic inflammation via tachykinins and other proinflammatory neuropeptides. Ozone-induced increases in the levels of the neuropeptide substance P observed in the BALF of human subjects suggests potential neurogenic involvement in O₃-induced increased vascular permeability, plasma protein extravasation, mucus secretion, and bronchoconstriction (Solway and Leff, 1991). Similar neurogenic involvement due to vagally mediated stimulation of C-fibers seen in animal toxicology studies also supports O₃-induced airway hyperresponsiveness observed in humans.

An extensive database of animal, human, and in vitro studies supports the conclusion that O₃ interacts with airway epithelial cell membranes and lining fluid to form lipid ozonation products and reactive oxygen species (ROS). These reactive products initiate a cascade of events leading to oxidative stress, injury, inflammation, airway epithelial damage and increased alveolar permeability to vascular fluids. Inflammation is the outcome of host response to injury and usually resolves completely. Continued irritant challenge may evolve into a chronic inflammatory state with simultaneous alterations in lung structure and function, leading to diseases such as fibrosis and emphysema, although neither fibrosis nor emphysema have yet

been observed with O₃ exposure. Continued inflammation can also alter the lung's ability to respond to infectious agents, allergens, and toxins. Acute inflammatory responses to O₃ exposure have been well documented experimentally in both humans and animals. As presented in Figure 8-4, the early inflammatory response to O₃-induced lung injury is apparent in human subjects within 3 h postexposure. This initial neutrophilic inflammatory response phase is characterized by increases in PMNs in BALF along with increased levels of inflammatory mediators such as interleukins, prostaglandins and complement component C3a. In vitro studies using human and animal lung cell culture systems have further examined the involvement of various inflammatory mediators and in some instances their downstream signaling pathways. The late inflammatory phase in the lung is characterized by increased levels of monocytes and eosinophils, as well as respective mediators such as cytokines, leukotrienes, proteinases, and ROS.

Disruption of the lung's blood barrier by O₃ results in vascular permeability changes and plasma protein extravasation. Analysis of BALF plasma influx markers such as albumin, other proteins, immunoglobulins, and epithelial cell damage markers such as LDH indicate O₃-induced lung epithelial injury. Ozone-induced lung injury and subsequent disruption of the airway epithelial barrier has been implicated in altered mucociliary clearance of particles observed in controlled human studies. Analogously, animal toxicology studies (see Chapter 5) have reported increased mortality to bacterial and viral infections subsequent to O₃ exposure and also increased clearance of particles with low O₃ exposure levels.

Controlled O₃ exposure studies of healthy humans have indicated a large degree of intersubject variability. The spirometric and symptomatic responses are highly reproducible within a given subject; but, within a group, pulmonary function can vary widely (see Figure 8-1) across different subjects. This interindividual variability is likely to be due to factors such as age, genetic background and antioxidant defenses, but as yet other factors remain to be characterized. Epidemiologic studies indicate a positive association between exposure to ambient O₃ and declines in lung function in children and those with preexisting respiratory diseases such as asthma. Increased incidence of ED visits and hospitalizations due to respiratory causes that have been reported in various epidemiologic studies (discussed in Chapter 7), as also supported by animal toxicology data, suggest a causal association with O₃. The epidemiologic associations become more apparent when the data are analyzed for the influence of seasonal

differences in ambient O₃ levels, with stronger evidence for associations in the warm season. Together, the evidence from the animal and human studies suggests that acute O₃ exposure is causally associated with respiratory system effects, including O₃-induced pulmonary function decrements, respiratory symptoms, lung inflammation, and increased lung permeability, airway hyperresponsiveness, increased uptake of nonviable and viable particles, and consequent increased susceptibility to PM-related toxic effects and respiratory infections.

Cardiovascular Health Effects

Only a few experimental studies of animals and humans have evaluated possible mechanisms or physiological pathways by which acute O₃ exposures may induce cardiovascular effects. Ozone induces lung injury, inflammation, and impaired mucociliary clearance, with a host of associated biochemical changes all leading to increased lung epithelial permeability. As discussed in Section 5.2.1, the generation of lipid ozonation products and reactive oxygen species in lung tissue can influence pulmonary hemodynamics and, ultimately, the cardiovascular system.

Recent *in vitro* studies of O₃ reactions with cholesterol in lung surfactant found consequent generation of highly reactive products such as oxysterols and β -epoxide in BALF isolated from rats exposed to 2.0 ppm O₃ for 4 h (Pulfer and Murphy, 2004). Additionally, both 5 β ,6 β -epoxycholesterol and its most abundant metabolite, cholestan-6-oxo-3 β ,5 α -diol, were shown to be cytotoxic to human lung epithelial (16-HBE) cells and to inhibit cholesterol synthesis. Studies (Pulfer et al., 2005) of mice exposed to 0.5, 1.0, 2.0, or 3.0 ppm O₃ for 3 h also demonstrated that these oxysterols were produced *in vivo*. These results suggest that this may be an additional mechanism of O₃ toxicity, including a pathway by which O₃ may play a possible role in atherosclerosis and other cardiovascular effects.

The presence of oxysterols in human atherosclerotic lesions implicates the oxidation of cholesterol in the pathogenesis of atherosclerosis, a well-known contributor to development of cardiovascular disease. Oxysterols may arise from different cholesterol oxidation mechanisms, (including free radical-mediated oxidations), and their unabated accumulation in macrophages and smooth muscle cells of arterial walls lead to formation of fatty streaks in advanced lesions. The presence of one of the O₃-induced oxysterols, secosterol, in endogenously formed arterial plaques (Wentworth et al., 2003) suggests that the oxysterols produced in the lung either due to

direct O₃ interaction with surfactant cholesterol or with oxidant radicals at the O₃-induced inflammation site may have potential involvement in the development of cardiovascular and myocardial diseases. In addition, the recent in vitro observation (Sathishkumar et al. 2005) of increased apoptosis (programmed cell death) induced by secosterol in H9c2 cardiomyocytes (heart cells) supports possible involvement of such biologically active oxysterols in O₃-induced cardiovascular effects observed in the epidemiologic studies. Also, the detection of oxysterols in the BALF of rats exposed to O₃ suggests their potential to be used as biomarkers of O₃ exposure. Demonstration of relationships between oxysterols of the type generated in lung surfactant with O₃ exposure and cardiovascular disease outcomes in clinical settings or epidemiologic studies would add considerable value to the experimental observations thus far reported in the animal toxicology studies.

Ozone-induced changes in heart rate, edema of heart tissue, and increased tissue and serum levels of atrial natriuretic factor (ANF) found with 8-h 0.5 ppm O₃ exposure in animal toxicology studies (Vesely et al., 1994a,b,c) also raise the possibility of potential cardiovascular effects of acute O₃ exposures. Such effects resulting from stimulation of airway irritant receptors, c-fiber activation, may result from either local or central nervous system involvement. Only one controlled human study (Gong et al., 1998) evaluated potential cardiovascular health effects of O₃ exposure and reported O₃-induced changes in alveolar-arterial oxygen transfer in subjects with hypertension. However, several other cardiovascular parameters (e.g., heart rate) evaluated in this study did not show any O₃-induced effects.

Animal toxicology studies have found both transient and persistent ventilatory responses with or without progressive decrease in heart rate (Arito et al., 1997). Observations of O₃-induced vasoconstriction in a controlled human exposure study by Brook et al. (2002) suggests another possible mechanism for O₃-related exacerbations of preexisting cardiovascular disease.

A few new field/panel studies of human adults have reported associations between ambient O₃ concentrations and changes in electrophysiologic indicators of cardiac function, e.g., heart rate variability (HRV). Also, some population time-series studies have suggested positive associations between acute O₃ exposure and cardiovascular hospitalizations in the warm season. These suggestive positive epidemiologic associations gain credibility and scientific support from

results of experimental animal toxicology and human clinical studies, as discussed above, which are indicative of plausible pathways by which O₃ may exert cardiovascular effects.

8.6.2 Chronic O₃ Exposure-Induced Health Effects

The effects of chronic O₃ exposure in humans were earlier addressed primarily with cross-sectional epidemiologic studies, as discussed in the 1996 O₃ AQCD. Due to lack of precise information on exposure, the possibility of selection bias, and the difficulty of controlling for confounders, the earlier study findings were considered to be inconclusive. Since 1996, several new cross-sectional epidemiological studies have evaluated potential associations between chronic exposure to O₃ and morbidity or mortality (see Chapter 7, Section 7.5). These studies suggest that seasonal exposure to O₃ may be related to changes in lung function in children. However, little evidence is available to support a relationship between chronic O₃ exposure and mortality or lung cancer incidence. There are no data available from controlled human chamber studies that evaluated chronic exposure regimens.

The lack of adequate data from epidemiologic and clinical studies in human has helped to focus attention much more so on results from chronic O₃ exposure studies in animals. Earlier chronic animal studies employed traditional exposure designs using chronic stable O₃ exposures to one or another single O₃ concentration over extended periods (weeks, months). More recent studies have attempted to incorporate design features that more closely mimic diurnal and seasonal patterns of O₃ exposure and realistic exposure concentrations. Studies of monkeys that compared these two designs reported greater airway pathology with the latter design. Persistent and irreversible effects observed in chronic animal toxicology studies indicate the need for further complementary human data from epidemiologic studies.

Animal toxicology data provide a clearer picture indicating that long-term O₃ exposure may have lasting effects. Chronic exposure studies in animals have reported biochemical and morphological changes suggestive of irreversible long-term O₃ impacts on the lung. Some of the studies in rats (0.5-1.0 ppm O₃ for 6 h/day) for 20 months and monkeys (0.61 ppm) for one year noted increased deposition of collagen and thickening of the CAR. Differences in the degree of this type of lung damage have been observed between continuous exposure and seasonal patterns of variations in O₃ exposure levels over time. A long-term study of infant rhesus monkeys exposed to simulated seasonal O₃ (0.5 ppm 8 h/day for 5 days every 14 days for 11 episodes)

resulted in remodeling in the distal airways, abnormalities in tracheal basement membrane, accumulation of eosinophils in conducting airways and decrements in airway innervation. Earlier studies in rats following seasonal episodic profiles also showed small, but significant, decrements in lung function that were consistent with focal fibrinogenesis in the proximal alveolar region. On the other hand, chronic O₃ exposures in a range of 0.5 to 1.0 ppm induce epithelial hyperplasia that disappears in a few days. The weight of evidence from the new experimental animal studies (using non-lifetime exposures) does not support ambient O₃ as being a pulmonary carcinogen.

Collectively, evidence from animal studies strongly suggests that chronic O₃ exposure is capable of damaging the distal airways and proximal alveoli, resulting in lung tissue remodeling leading to apparent irreversible changes. Compromised pulmonary function and structural changes due to persistent inflammation may exacerbate the progression and development of chronic lung disease. These findings offer some insight into potential biological mechanisms for the suggested association between seasonal O₃ exposure and reduced lung function development in children as observed in epidemiologic studies.

8.6.3 Mortality-Related Health Endpoints

An extensive analysis of population time-series studies that evaluated the air pollution related mortality risk estimates presented in Section 7-4 utilized data from single and multicity studies from around the world. Mortality risk estimates derived from single- and multicity studies in U.S. and Canada coupled with meta-analyses generally indicate associations between acute O₃ exposure and elevated risk for all-cause mortality, even after adjustment for the influence of season and PM. Several single-city studies that specifically evaluated the relationship between O₃ exposure and cardiopulmonary mortality also reported results suggestive of a positive association.

The epidemiology results outlined above for mortality suggest a pattern of effects that may be biologically germane to interpretation of its causality, but our knowledge about potential underlying mechanisms remains relatively limited and suggests a need for further experimental support. The majority of the physiological and biochemical parameters evaluated both in human clinical and animal toxicology studies (Table 8-1; Figure 8-3) suggest a relatively transient nature for O₃-induced biochemical perturbations. Most effects attenuate over time, depending on

the preexisting pathophysiology. However, one can hypothesize a generic pathway of O₃-induced lung damage, potentially involving oxidative lung damage with subsequent inflammation and/or decline in lung function leading to respiratory distress in some sensitive population groups (e.g., asthmatics) or, other plausible pathways noted below may lead to O₃-related contributions to cardiovascular effects that ultimately increase the risk of mortality.

Recent analysis of data from the third National Health and Nutrition Examination Followup study indicated that about 20% of the adult population have reduced FEV₁ values indicative of impaired lung function. The majority of these individuals have COPD, asthma or fibrotic lung disease (Mannino et al., 2003). These cardiopulmonary disease conditions are associated with persistent low-grade systemic inflammation. It has also been reported that patients with COPD are at increased risk for cardiovascular disease. Lung disease with underlying inflammation may also link to low-grade systemic inflammation associated with atherosclerosis. These effects in disease are independent of cigarette smoking (Sin et al., 2005). Lung function decrements in cardiopulmonary disease have also been associated with inflammatory markers such as C-reactive protein (CRP) in blood. In fact, at the population level, individuals with the lowest FEV₁ have the highest levels of CRP, while those with highest FEV₁ have the lowest values for CRP (Mannino et al., 2003; Sin and Man, 2003). The complex, physiological and biochemical perturbations that exist simultaneously (Figures 8-3 and 8-4) subsequent to acute exposure to O₃ may tilt the biological homeostasis mechanisms leading to adverse health effects in people with compromised cardiopulmonary systems.

Of much interest are several other types of newly available experimental data that support reasonable hypotheses that may help to explain findings of O₃-related increases in cardiovascular mortality observed in some epidemiological studies. These include the direct effects of O₃ in terms of its increasing platelet aggregating factor (PAF) in lung tissue that can then enter the general circulation and possibly contribute to increased risk of blood clot formation (as illustrated in Figure 5-3) and consequent increased risk of myocardial infarction (heart attack), cerebrovascular events (stroke), and/or associated cardiovascular-related mortality. Other O₃-induced effects may also contribute to increased risk of cardiovascular impacts. Ozone reactions with cholesterol in lung surfactant to produce epoxides and/or associated diol metabolites that are cytotoxic to lung and heart muscle cells and that contribute to atherosclerotic plaque formation in arterial walls represent another such pathway. Also, to the extent that O₃-induced

increases in lung permeability allow more ready entry of inhaled particulate matter (PM) into the blood stream, then O₃ exposure would enhance the risk for PM-related cardiovascular effects (see U.S. Environmental Protection Agency, 2004). An associated factor to consider is O₃'s ability to contribute to ultrafine PM formation in ambient air and indoor environments (as discussed in Sections 8.2.4 and 8.3.2). Thus, O₃ can both contribute to increased presence of fine particles and human exposure to them, as well as enhance uptake of inhaled fine PM and thereby presumably contribute to exacerbation of PM-induced cardiovascular effects in addition to those more directly induced by O₃ per se.

8.6.4 Health Effects of Ozone-Containing Pollutant Mixtures

The above-noted potential for O₃-related enhancements of PM formation, particle uptake, and exacerbation of PM-induced cardiovascular effects underscores the importance of considering contributions of O₃ interactions with other often co-occurring air pollutants to health effects due to O₃-containing pollutant mixes. Chapters 4, 5, and 6 provided a discussion of experimental studies that evaluate interactions of O₃ with other co-occurring pollutants. Some examples of important pollutant mixture effects noted there are highlighted below.

First, Chapter 4 noted some important interactive effects of coexposures to O₃ and NO₂ and SO₂, two other common gaseous copollutants found in ambient air mixes. That is, a study by Rigas et al. (1997) showed that continuous exposure of healthy human adults to SO₂ or to NO₂ increased inhaled bolus O₃ absorption, while continuous exposure to O₃ alone decreased bolus absorption of O₃. This suggests enhancement of O₃ uptake by NO₂ or SO₂ coexposure in ambient air mixes. Also, as noted in Chapter 6, another study by Jenkins et al. (1999) showed that asthmatics exhibited enhanced airway responsiveness to house dust mite following exposures to O₃, NO₂, and the combination of the two gases. Spirometric response, however, was impaired only by O₃ and O₃+NO₂ at higher concentrations. On the other hand, animal toxicology studies discussed in Chapter 5 that evaluated exposures to O₃ in mixture with NO₂, formaldehyde, and PM demonstrated additive, synergistic or antagonistic effects, depending on the exposure regimen and the specific health endpoints evaluated.

The results of the Jenkins et al. (1999) study also help to illustrate a more general phenomena of enhancement by O₃ exposure of various respiratory responses of sensitive individuals to allergens. Chapter 6 noted, for example, studies (a) by Peden et al. (1995)

showing O₃-induced increased response to nasal allergen challenge among allergic asthmatic subjects, and (b) by Michelson et al. (1999) showing promotion by 0.4 ppm O₃ exposure of inflammatory cell influx in response to nasal allergen challenge in asymptomatic dust-mite sensitive asthmatics. In addition, Jörres et al. (1996) demonstrated enhancement by 0.25 ppm O₃ exposure of airway responsiveness in mildly allergic asthmatics that was increased in response to an individual's historical allergen (grass and birch pollen, house dust mite, animal dander). These results were further extended by Holz et al. (2002) who showed that repeated daily exposure to 0.125 ppm O₃ for 4 days exacerbated lung function decrements (e.g., decreased FEV₁) in response to bronchial allergen challenges among subjects with preexisting allergic airway disease, with or without asthma. This suggests that O₃ exposure can place allergic nonasthmatic persons, as well as asthmatics, at increased risk for allergic respiratory effects. Consistent with and supporting the above findings are animal toxicology studies reviewed in detail by Harkema and Wagner (2005), which indicate that (a) O₃-induced epithelial and inflammatory responses in laboratory rodents are markedly enhanced by coexposure to inhaled biogenic substances (e.g., bacterial endotoxin or ovalbumin, an experimental aeroallergen) and (b) adverse airway effects of biogenic substances can be exacerbated by coexposure to O₃.

Also of much note is a newly emerging literature which indicates that O₃ can modify the biological potency of certain types of ambient PM, as shown by experimental tests. For example, as described in Chapter 5 (Section 5.4.2) reaction of diesel PM with 0.1 ppm O₃ for 48 h increased the potency (compared to non-exposed or air-exposed diesel PM) to induce neutrophil influx, total protein, and LDH in lung lavage fluid in response to intratracheal PM instillation in rats (Madden et al., 2000). However, the potency of carbon black particles was not enhanced by exposure to O₃, suggesting that O₃ reaction with organic components of the diesel PM were responsible for the observed increased diesel PM effects.

Potential interaction of O₃ with fine PM in aged rats was examined by Kleinman et al. (2000). In this study the effects of fine PM containing two common toxic constituents, ammonium bisulfate (ABS, 0.3 μm 70 μg/m⁻³) and elemental carbon (C, 0.3 μm 50 μg/m⁻³) and a mixture (ABS + C) with 0.2 ppm O₃ was evaluated on aged rat lung structure and macrophage function. Exposures of O₃, elemental carbon or ABS alone did not cause significant lung injury, lung tissue collagen content or respiratory burst activity. On the other hand, mixtures (ABS + C + O₃) caused significant lung injury as assessed by increased cell proliferation response in

lung epithelial and interstitial cells, loss of lung tissue collagen and increase in respiratory burst and phagocytic activity.

The majority of toxicological studies discussed earlier in this chapter evaluated effects of individual pollutants or simple mixtures of the constituents of urban smog mixtures, and these toxicology studies do not fully explain epidemiologic findings that have increasingly shown ambient O₃, other gaseous pollutants, and/or PM to be associated with various health effects at relatively low concentrations. In a recent report, Sexton et al (2004) utilized “smog chambers”, i.e., environmental irradiation chambers to generate synthetic photochemical oxidants mixtures similar to urban smog, and studied the toxicity of such mixtures on the inflammatory response of A549 cells in an in vitro exposure system. In this preliminary study, the authors found the simulated urban photochemical oxidant mixture generated with the addition of O₃ to have enhanced toxicity (as assessed by the expression of IL-8 mRNA). Additional toxicology studies using similar realistic air pollution smog mixtures in the future may provide more relevant biological understanding for the potential interactions that occur in the ambient air among various pollutants.

All of the above types of interactive effects of O₃ with other co-occurring gaseous and nongaseous viable and nonviable PM components of ambient air mixes argue for not only being concerned about direct effects of O₃ acting alone but also the need for viewing O₃ as a surrogate indicator for air pollution mixes which may enhance risk of adverse effects due to O₃ acting in combination with other pollutants. Viewed from this perspective, epidemiologic findings of morbidity and mortality associations with ambient O₃ concentrations extending to concentrations below 0.08 ppm become more understandable and plausible.

8.7 SUSCEPTIBLE AND VULNERABLE POPULATIONS, AND POTENTIAL PUBLIC HEALTH IMPACTS

Many factors such as age, gender, disease, nutritional status, smoking, and genetic variability may contribute to the differential effects of environmental pollutants, including O₃. Genetic factors, such as single nucleotide polymorphisms (SNPs) and developmental defects, can contribute to innate susceptibility, while acquired susceptibility may develop due to personal habits (smoking, diet, exercise) and other risk factors such as age, gender, pregnancy, and

copollutant exposures. In the 1996 O₃ AQCD, children, outdoor workers, and people with preexisting respiratory disease were identified as likely being more susceptible or vulnerable to effects of ambient O₃ exposure. However, the available toxicological and human data had not shown that males and females respond differently to O₃; nor were available data adequate to suggest differences in responsiveness to O₃ based on ethnic or racial background. Overall, then, the available information from animal toxicology and epidemiologic studies provided only relatively limited evidence by which to confidently identify likely susceptible groups and/or to associate specific factors as contributing to increased risk of O₃-related adverse health effects. Advances in available research results since 1996 have improved our ability to delineate likely susceptible or vulnerable populations at increased risk for O₃-induced health effects and to delineate factors contributing to such risk.

8.7.1 Preexisting Disease as a Potential Risk Factor

People with preexisting pulmonary disease are likely to be among those at increased risk from O₃ exposure. Altered physiological, morphological and biochemical states typical of respiratory diseases like asthma, COPD and chronic bronchitis may render people sensitive to additional oxidative burden induced by O₃ exposure. Based on studies assessed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996a), asthmatics appear to be at least as, or more, sensitive to acute O₃ exposure as healthy nonasthmatic subjects. The new results reviewed in Chapters 6 and 7 of this document from controlled exposure and epidemiologic studies also indicate that asthmatics are a potentially sensitive subpopulation for O₃ health effects.

A multicity study by Mortimer et al. (2002) support earlier observations that asthmatic children are particularly susceptible to ambient O₃. This association, based on decrements in lung function and exacerbation of pulmonary disease symptoms, suggests that O₃ exposures may result in increased use of medication in asthmatic children. A number of time-series epidemiologic studies have reported increased risk in study subsets of individuals with preexisting lung diseases, especially asthma, as potentially susceptible individuals. The epidemiologic studies of acute exposure to O₃ discussed in Section 8.4.2 indicate increased risk for exacerbation of respiratory disease symptoms during the warm season.

Several clinical studies reviewed in the 1996 O₃ AQCD on atopic and asthmatic subjects had suggested but not clearly demonstrated enhanced responsiveness to acute O₃ exposure

compared to healthy subjects. The majority of the newer studies reviewed in Chapter 6 indicate that asthmatics are as sensitive as, if not more sensitive than, normal subjects in manifesting O₃-induced pulmonary function decrements.

Ozone-induced increases in neutrophils, protein, and IL-8 were found to be significantly higher in the BALF from asthmatics compared to healthy subjects, suggesting mechanisms for the increased sensitivity of asthmatics. Similarly, subjects with allergic asthma exhibited increased airway responsiveness to inhaled allergens upon acute O₃ exposure. Consistent with these changes, it is suggested that asthmatics will be more sensitive to small airway effects of ambient O₃. Asthmatics present a differential response profile for cellular, molecular, and biochemical parameters (Figure 8-1) that are altered in response to acute O₃ exposure. Increases in O₃-induced nonspecific airway responsiveness incidence and duration could have important clinical implications for asthmatics.

Bronchial constriction following provocation with allergens presents a two-phase response. The early response is mediated by release of histamine and leukotrienes that leads to contraction of smooth muscle cells in the bronchi, narrowing the lumen and decreasing the airflow. In asthmatics, these mediators also cause accumulation of eosinophils, followed by production of mucus and a late-phase bronchial constriction and reduced airflow. Holz et al. (2002) reported an early phase response in subjects with rhinitis after a consecutive 4-day exposure to 0.125 ppm O₃ that resulted in a clinically relevant (>20%) decrease in FEV₁. Allergen challenge in mild asthmatics 24 h postexposure to 0.27 ppm O₃ for 2 h resulted in significantly increased eosinophil counts in BALF compared to healthy subjects (Vagaggini et al., 2002). Epithelial cells from mucosal biopsies of allergic asthmatics indicated significant increases in the expression of IL-5, IL-8 and GM-CSF, suggesting increased neutrophilic inflammation compared to healthy subjects (Bosson et al., 2003).

Several human exposure studies have shown differences between asthmatics and healthy human subjects with regard to PMN influx in BALF. In vitro studies (Schierhorn et al., 1999) of nasal mucosal biopsies from atopic and nonatopic subjects exposed to 0.1 ppm O₃ found significant differences in release of IL-4, IL-6, IL-8, and *TNF-α*. Another study by Schierhorn et al. (2002) found significant differences in the O₃-induced release of the neuropeptides neurokinin A and substance P for allergic patients in comparison to nonallergic controls, suggesting increased activation of sensory nerves by O₃ in the allergic tissues. Another study by

Bayram et al. (2002) using in vitro culture of bronchial epithelial cells recovered from atopic and nonatopic asthmatics also found significant increases in epithelial permeability in response to O₃ exposure. In addition, some controlled human O₃ exposure studies in asthmatics (Hiltermann et al., 1999; Scannell et al., 1996) reported increased secretion of IL-8, suggesting increased neutrophilic inflammation. Two studies (Jörres et al., 1996; Holz et al., 2002) observed increased airway responsiveness to repeated daily O₃ exposure to bronchial allergen challenge in subjects with preexisting allergic airway disease.

Newly available reports from controlled human exposure studies (see Chapter 6) utilized subjects with preexisting cardiopulmonary diseases such as COPD, asthma, allergic rhinitis, and hypertension. The data generated from these studies that evaluated pulmonary function changes in spirometry did not find clear differences between filtered air and O₃ exposure in COPD and asthmatic subjects. However, the new data on airway responsiveness, inflammation, and various molecular markers of inflammation and bronchoconstriction indicate that people with atopic asthma and allergic rhinitis comprise susceptible groups for O₃-induced adverse health effects. Collectively, these observations suggest that O₃ exposure exacerbates pre-existing allergic asthma. People with allergic asthma likely represent a sizable segment of the population reported to have increased symptoms of respiratory illness exacerbations, ED visits, and hospital admissions in epidemiologic studies.

Although controlled human exposure studies have not found evidence of larger spirometric changes in people with COPD relative to healthy subjects, this may be due to the fact that most people with COPD are older adults who would not be expected to have such changes based on their age. However, new epidemiological evidence indicates that people with COPD may be more likely to experience other effects, including emergency room visits, hospital admissions, or premature mortality. A study by Tenías et al. (2002) observed a significant positive association between O₃ and for COPD. Results from an analysis of five European cities indicated strong and consistent O₃ effects on unscheduled respiratory hospital admissions, including COPD (Anderson et al., 1997). Also, an analysis of a 9-year data set for the whole population of the Netherlands provided risk estimates for more specific causes of mortality, including COPD (Hoek et al., 2000, 2001; reanalysis Hoek, 2003); a positive, but nonsignificant excess risk of COPD-related mortality was found to be associated with short-term O₃ concentrations.

8.7.2 Age-Related Variations in Susceptibility/Vulnerability

Growth and development of the human respiratory system is not complete until 18 to 20 years of age; 80% of alveoli are formed postnatally, and changes in the lung continue through adolescence (Dietert et al., 2000). Because the developing lung is highly sensitive to damage from exposure to environmental toxicants, children and infants are likely to be among the most susceptible to detrimental effects from O₃ exposure. In addition, children experience enhanced O₃ vulnerability because they generally spend more time outdoors and are highly active, especially during the warm season when ambient O₃ concentrations are expected to be high and results in increased O₃ dose delivered to their lungs (Plunkett et al., 1992; Wiley et al., 1991a,b).

Controlled human exposure studies assessed in the 1996 O₃ AQCD indicated that children (generally aged 8 to 11 years) and adolescents (generally aged 12 to 18 years) exhibited, on average, greater spirometric responses to O₃ than middle-aged or older adults when exposed to comparable O₃ doses. However, less evidence of respiratory symptoms has been observed among healthy children in clinical studies of O₃ exposure. Such diminished symptomatic responses in children may put them at an increased risk for continued O₃ exposure. As reported in the 1996 O₃ AQCD, several summer camp panel studies collectively provided strong evidence for an association between acute O₃ exposure and lung function declines in mostly healthy, nonasthmatic children (aged 7 to 17 years) who spent long hours outdoors. Additional epidemiologic field studies observed that respiratory symptoms (or exacerbation of asthma) and decrements in PEF were associated with increased ambient O₃ concentrations, with greater responses found in asthmatic than in nonasthmatic children. No new human exposure studies investigating O₃ responses in children have been published since the last O₃ AQCD; however, the epidemiologic studies published during the last decade (see Section 7.2.3.1 for details) continue to indicate that children are at increased risk for to O₃-related respiratory health effects.

Many recent field studies published in the past decade have focused on the effects of acute O₃ exposure on the respiratory health of children. In general, children experienced O₃-related decrements in pulmonary function parameters, including PEF, FEV₁, and FVC (e.g., Mortimer et al., 2002; Thurston et al., 1997). Declines in lung function were observed in both healthy and asthmatic children following acute exposure to O₃, but respiratory symptoms were largely found only in asthmatic children. The O₃-related changes in lung function in children were often small, <5% change per 40 ppb increase in mean 1-h max O₃ or equivalent. However,

as stated earlier, the group mean responses reported may mask larger lung function changes in sensitive individuals. Höpfe et al. (2003) examined several susceptible populations, including children (aged 6 to 8 years) and juvenile asthmatics (aged 12 to 23 years), for changes in pulmonary function attributable to O₃ exposure in Munich, Germany. Using group mean values, consistent O₃ effects were not observed in both groups; but, when data was analyzed on an individual basis, a potential pattern of increased O₃ sensitivity was observed. About 20% of the children and asthmatics were found to be particularly susceptible to O₃ health effects (i.e., experienced greater than 10% change in FEV₁).

Several epidemiologic field studies have examined effects of long-term O₃ exposure on lung function in children. The results collectively indicate that seasonal O₃ exposure is associated with smaller increases in lung function in children (mean age 7 to 8 years). These results are supported by a long-term study of infant rhesus monkeys exposed to simulated seasonal O₃ patterns (0.5 ppm 8h/day for 5 days, every 14 days for 11 episodes) which found remodeling in the distal conducting airways as a result of damage and repair processes occurring with repeated O₃ exposure (Evans et al., 2003; Schelegle et al., 2003). However, epidemiologic evidence of an effect of longer-term (i.e., multiyear) O₃ exposure on lung function development in children has generally not yet been found, as most definitively indicated by the southern California Children's Health Study that examined children from 10 to 18 years of age (Gauderman et al., 2004).

In time-series studies that examined associations between ambient O₃ concentrations and ED visits, hospitalizations, and mortality in children, consistent results were not observed. This is likely due to the small number of daily deaths and hospitalizations among children and, also, the fact that the large number of competing causes of mortality and hospitalizations in children (which are together more important than air pollution) make it difficult to detect any excess risk attributable to ambient O₃ exposure.

The elderly are also often classified as being particularly susceptible to air pollution. The basis for increased O₃ sensitivity among older adults is not known, but one hypothesis is that it may be related to changes in the respiratory tract lining fluid antioxidant defense network (Kelly et al., 2003). Increased susceptibility of older adults to O₃ health effects is most clearly indicated in the newer mortality studies. In the meta-analysis by Bell et al. (2005), a comparison of O₃-mortality risk estimates by age indicates that, in general, the older adults population (>65 years

of age) is more susceptible to O₃ effects, with older adults experiencing a more than 50% greater excess risk in mortality compared to the all-age group. In the large U.S. 95 communities study (Bell et al., 2004), the O₃ effect on all-cause mortality was slightly larger for older adults compared to younger individuals (≤ 65 years of age). Bell et al. (2004) further noted that though the risk estimates for older adults were only slightly higher, the absolute effect of O₃ is substantially greater due to the higher underlying mortality rates in the older adult population (e.g., approximately 50% of deaths were in those ≥ 75 years), which leads to a larger number of extra deaths for older adults compared to the general population.

Lung function responses to O₃ exposure, however, have been found to be diminished in the older adult population. The study by Höpfe et al. (2003) observed that ambient O₃ exposure was not associated with lung function declines in seniors (69 to 95 years of age). While a greater than 10% change in FEV₁ was observed in 20% of the children and juvenile asthmatics, only 5% of the older adult population experienced such a change. The results by Höpfe et al. are consistent with the diminishing lung function responses to O₃ exposure with increasing age observed in clinical studies.

In summary, new epidemiologic research presented in Chapter 7 of this document continues to indicate that children, asthmatics in particular, constitute a sensitive group in epidemiologic studies of oxidant air pollution. Healthy children with prolonged exposure periods, combined with elevated levels of exertion or exercise also appear to be more vulnerable to O₃-related respiratory health effects. Older adults (>65 years of age) also have been shown to be susceptible to O₃ health effects, particularly O₃-related mortality. One epidemiologic study also found that the elderly experience diminished lung function responses to O₃ exposure, which is supported by evidence from clinical studies.

8.7.3 Vulnerability of Outdoor Workers and Others Who Participate in Outdoor Activities

The 1996 O₃ AQCD indicated that one population group that was shown to have increased responsiveness to ambient O₃ exposure consisted of exercising healthy and asthmatic individuals, including children, adolescents, and adults. The effects of O₃ on the respiratory health of outdoor workers and others who participate in outdoor activities have been investigated in several recent epidemiologic studies. These individuals may experience increased

vulnerability for O₃ health effects, because they are typically exposed to high doses of O₃ as they spend long hours outdoors often at elevated exertion levels.

In a group of berry pickers in Fraser Valley, Canada, large decrements in lung function (~5% decrease in FEV₁ per 40 ppb increase in 1-h max O₃) were associated with acute exposure to O₃ (Brauer et al., 1996). The mean ambient 1-h max O₃ was 40.3 ppb (SD 15.2) over the study period of June to August 1993. The berry pickers worked outdoors for an average of 11 h at elevated heart rates (on average, 36% higher than resting levels). These results indicate that extended exposures to O₃ at elevated exertion levels can produce marked effects on lung function among outdoor workers.

Höppe et al. (1995) examined forestry workers for O₃-related changes in pulmonary function in Munich, Germany. Ventilation rates, estimated from their average activity levels, were elevated. When comparisons were made between high O₃ days (mean ½-h max O₃ of 64 ppb) and low O₃ days (mean ½-h max O₃ of 32 ppb), 59% of the forestry workers experienced a notable decrement in lung function (i.e., at least a 20% increase in specific airway resistance or at least a 10% decrease in FEV₁, FVC, or PEF) on high O₃ days. None experienced improved lung function. This study also examined athletes following a 2-h outdoor training period in the afternoon yielding a ventilation rate double the estimate for the forestry workers. Though a significant association between ambient O₃ levels and decrements in FEV₁ was observed overall, a smaller percentage of the athletes (14%) experienced a notable decrement in lung function on high O₃ days compared to the forestry workers; and 19% of the athletes actually showed an improvement.

A large field study by Korrnick et al. (1998) examined the effects of multihour O₃ exposures (on average, 8 h) on adults hiking outdoors in Mount Washington, NH. The mean of the hourly O₃ concentrations during the hike was 40 ppb (range 21-74). After the hike, all subjects combined experienced a relatively small mean decline in FEV₁ (1.5% decrease per 30 ppb increase in mean hourly O₃ concentrations) during the hike. Ozone-related changes in lung function parameters were estimated. Stratifying the data by hiking duration indicated that individuals who hiked 8 to 12 h experienced a >2-fold decline in FEV₁ versus those only hiking 2 to 8 h.

Results from the above field studies are consistent with those from earlier summer camp studies (Avol et al., 1990; Higgins et al., 1990; Raizenne et al., 1987, 1989; Spektor et al., 1988,

1991), which also observed strong associations between acute O₃ exposure and decrements in lung function among children who spent long hours outdoors. In a recent analysis by the southern California Children's Health Study, a total of 3,535 initially nonasthmatic children (ages 9 to 16 years at enrollment) were followed for up to 5 years to identify new-onset asthma cases associated with higher long-term ambient O₃ concentrations (McConnell et al., 2002). Communities were stratified by pollution levels, with six high-O₃ communities (mean 1-h max O₃ of 75.4 ppb [SD 6.8] over four years) and six low-O₃ communities (mean 50.1 ppb [SD 11.0]). In the combined analysis using all children, asthma risk was not found to be higher for residents of the six high-O₃ communities versus those from the six low-O₃ communities. However, within the high-O₃ communities, asthma risk was more than 3 times greater for children who played three or more sports versus those who played no sports, an association not observed in the low-O₃ communities. Therefore, among children repeatedly exposed to higher O₃ levels, increased exertion outdoors (and resulting increased O₃ dose) was associated with excess asthma risk.

These field studies with subjects at elevated exertion levels support the extensive evidence derived from controlled human exposure studies. The majority of human chamber studies have examined the effects of O₃ exposure in subjects performing continuous or intermittent exercise for variable periods of time (see Chapter 6). Significant O₃-induced respiratory responses have been observed in clinical studies of exercising individuals. The epidemiologic studies discussed above also indicate that prolonged exposure periods, combined with elevated levels of exertion or exercise, may magnify O₃ effects on lung function. Thus, outdoor workers and others who participate in higher exertion activities outdoors during the time of day when high peak O₃ concentrations occur appear to be particularly vulnerable to O₃ effects on respiratory health.

8.7.4 Genetic Factors Affecting O₃ Susceptibility

In the 1996 O₃ AQCD, the potential influence of genetic factors on responses to O₃ exposure could not be thoroughly evaluated due to the very limited data then available. The human and toxicologic data were inadequate to suggest differences in O₃ responsiveness based on gender, ethnic, or racial background. New animal toxicology studies using various strains of mice and rats that have identified O₃-sensitive and resistant strains illustrate the importance of genetic background in determining O₃ susceptibility. Using subacute low exposure regimen

(0.3 ppm O₃, 48 h) studies on inbred strains that have been designated as inflammation prone or resistant, Kleeberger et al., (1997) identified the pro-inflammatory cytokine gene, *Tnf-α*, as a susceptibility gene. Further characterization of this model indicated a role for TNF receptors in O₃-induced pulmonary epithelial injury and inflammation (Cho et al., 2001). Studies of five inbred strains of mouse with differing response to O₃ exposure (acute high dose or low dose continuous exposure for 3 days), found a protective role for clara cell secretory protein against O₃-induced oxidative damage (Broeckeaert et al., 2003; Wattiez et al., 2003). The role for these genes and/or their orthologs in human susceptibility to O₃ exposure has yet to be examined.

Biochemical and molecular parameters evaluated in these toxicology experiments were used to identify specific loci on chromosomes and, in some cases, to relate the differential expression of specific genes to biochemical and physiological differences seen among these species. Using O₃-sensitive and O₃-resistant species, it has been possible to identify the involvement of AHR and inflammation processes in O₃ susceptibility. However, most of these studies were carried out using relatively high doses of O₃, making the relevance of these studies questionable for human health effects assessment. The molecular parameters identified in animal and human experimental studies may serve as useful biomarkers with the availability of suitable technologies and may, ultimately, be integrated with epidemiologic studies. The interindividual differences in O₃ responsiveness observed across a spectrum of symptoms and lung function responses do not yet allow identification of important underlying factors, except a significant role for age.

Apart from age at the time of exposure, controlled human exposure studies have also indicated a high degree of interindividual variability in some pulmonary physiological parameters. Genetic factors likely contribute to the substantial variability observed between individuals. Several recent human clinical and epidemiologic studies (Bergamaschi et al., 2001; Corradi et al., 2002; David et al., 2003; Romieu et al., 2004; Yang et al., 2005) have reported that genetic polymorphisms for antioxidant enzymes and inflammatory genes may modulate pulmonary function and inflammatory responses to O₃ exposure. Glutathione S-transferases (GSTs) play a major role in protecting cells against damage from reactive oxygen species by conjugating them with glutathione so that they can be rapidly eliminated. A common homozygous deletion polymorphism of the GSTM1 gene (GSTM1 null genotype) abolishes enzyme activity and may increase susceptibility to O₃-induced oxidative stress. Asthmatic

children with a genetic deficiency of GSTM1 were found to be more responsive to ambient O₃ exposure, as assessed by decrements in FEF₂₅₋₇₅, in a Mexico City study by Romieu et al. (2004). Antioxidant supplementation of vitamins C and E attenuated post-O₃ lung function response in these children. More specific genotyping has shown that O₃ responsiveness of asthmatic children may be related to the presence of variant Ser allele for a detoxifying enzyme (NQO1) induced in response to oxidative stress (David et al., 2003). The presence of at least one NQO1 Ser allele in asthmatic children with GSTM1 null genotype was found to have a protective effect against O₃ exposure. Also, polymorphism in *Tnf-α* has been implicated in the risk of O₃-induced lung function changes in healthy, mild asthmatics and individuals with rhinitis (Yang et al., 2005), which supports toxicologic evidence referred to earlier in this section.

The above observations suggest a potential role for these genetic markers in the innate susceptibility to O₃; however, the validity of these markers and their relevance in the context of prediction to population studies need additional validation. The lack of correlation between lung function and airway inflammatory responses to O₃ in healthy subjects, combined with the evidence of separate chromosomal loci for O₃-induced AHR and airway inflammation in inbred mice, suggest that these two responses are probably independently regulated.

8.7.5 Potential Public Health Impacts

Exposure to ambient O₃ is associated with a variety of health outcomes, including increased incidence of cough, reduction in lung function, increased inflammation, and increased hospital admissions and mortality. In protecting public health, a distinction must be made between health effects that are considered “adverse” and those that are not. What constitutes an adverse health effect varies for different population groups, with some changes in healthy individuals not being viewed as adverse but those of similar type and magnitude in other susceptible individuals with preexisting disease being seen as adverse.

8.7.5.1 Concepts Related to Defining of Adverse Health Effects

The American Thoracic Society (ATS) published an official statement on “What Constitutes an Adverse Health Effect of Air Pollution?” (ATS, 2000). This statement updated guidance for defining adverse respiratory health effects published fifteen years earlier (ATS,

1985), in order to take into account new investigative approaches used to identify the effects of air pollution and to reflect concern for impacts of air pollution on specific susceptible groups.

In the 2000 update, there is an increased focus on quality of life measures as indicators of adversity and, also, a more specific consideration of population risk. Exposure to air pollution that increases the risk of an adverse effect to the entire population is viewed as adverse, even though it may not increase the risk of any identifiable individual to an unacceptable level. For example, a population of asthmatics could have a distribution of lung function such that no identifiable single individual has a level associated with significant impairment; and exposure to air pollution could shift the distribution to lower levels that still do not bring any identifiable individual to a level that is associated with clinically relevant effects. However, this would be considered to be adverse because individuals within the population would have diminished reserve function and, therefore, would be at increased risk if affected by another agent.

Reflecting new investigative approaches, the ATS statement also describes the potential usefulness of research into the genetic basis for disease, including responses to environmental agents that provide insights into the mechanistic basis for susceptibility and provide markers of risk status. Likewise, biomarkers that are indicators of exposure, effect or susceptibility may someday be useful in defining the point at which one or an array of responses should be considered an adverse effect.

The 1996 O₃ AQCD provided information useful in helping to define adverse health effects associated with ambient O₃ exposure by describing the gradation of severity and adversity of respiratory-related O₃ effects, and those definitions are reproduced and presented here as Tables 8-2 and 8-3. The severity of effects described in those tables and the approaches taken to define their relative adversity still appear to be valid and reasonable even in the context of the new ATS statement (ATS, 2000).

As assessed in detail in earlier chapters of this document and briefly recapitulated in preceding sections of this chapter, exposures to a range of O₃ concentrations have been reported to be associated with increasing severity of several categories of health effects.

Respiratory effects associated with short-term O₃ exposures have been by far the most extensively studied and most clearly shown to be causally related to O₃ exposure. Such effects include the induction of pulmonary function decrements and respiratory symptoms demonstrated in response to controlled acute (1- to 8-h) O₃ exposures of human subjects and also observed

Table 8-2. Gradation of Individual Responses to Short-Term Ozone Exposure in Healthy Persons ^a

Functional Response	None	Small	Moderate	Large
FEV ₁	Within normal range ($\pm 3\%$)	Decrements of 3 to $\leq 10\%$	Decrements of >10 but $<20\%$	Decrements of $\geq 20\%$
Nonspecific bronchial responsiveness ^a	Within normal range	Increases of $<100\%$	Increases of $\leq 300\%$	Increases of $>300\%$
Duration of response	None	<4 hours	>4 hours but ≤ 24 hours	>24 hours
Symptomatic Response	Normal	Mild	Moderate	Severe
Cough	Infrequent cough	Cough with deep breath	Frequent spontaneous cough	Persistent uncontrollable cough
Chest pain	None	Discomfort just noticeable on exercise or deep breath	Marked discomfort on exercise or deep breath	Severe discomfort on exercise or deep breath
Duration of response	None	<4 hours	>4 hours but ≤ 24 hours	>24 hours
Impact of Responses	Normal	Normal	Mild	Moderate
Interference with normal activity	None	None	A few sensitive individuals choose to limit activity	Many sensitive individuals choose to limit activity

^a An increase in nonspecific bronchial responsiveness of 100% is equivalent to a 50% decrease in PD₂₀ or PD₁₀₀.

Source: This table is reproduced from the 1996 O₃ AQCD (Table 9-1, page 9-24) (U.S. Environmental Protection Agency, 1996a).

epidemiologically to be associated with ambient O₃ exposures in children and adults vigorously engaged in outdoor activities. The severity of the symptoms and magnitude of the pulmonary decrements, as previously noted, depend on the inhaled O₃ dose and individual O₃ sensitivity of the exposed persons. Controlled exposure chamber studies assessed in the 1996 O₃ AQCD provided very clear evidence that statistically significant reductions in lung function occurred in healthy adults in response to 6.6-h O₃ exposures as low as 0.08 ppm under moderate intermittent exercise conditions. Of considerable importance, whereas none of the subjects showed marked

Table 8-3. Gradation of Individual Responses to Short-Term Ozone Exposure in Persons with Impaired Respiratory Systems

Functional	None	Small	Moderate	Large
FEV ₁ change	Decrements of <3%	Decrements of 3 to ≤10%	Decrements of >10 but <20%	Decrements of ≥20%
Nonspecific bronchial responsiveness ^a	Within normal range	Increases of <100%	Increases of ≤300%	Increases of >300%
Airway resistance (SR _{aw})	Within normal range (±20%)	SR _{aw} increased <100%	SR _{aw} increased up to 200% or up to 15 cm H ₂ O/s	SR _{aw} increased >200% or more than 15 cm H ₂ O/s
Duration of response	None	<4 hours	>4 hours but ≤24 hours	>24 hours

Symptomatic	Normal	Mild	Moderate	Severe
Wheeze	None	With otherwise normal breathing	With shortness of breath	Persistent with shortness of breath
Cough	Infrequent cough	Cough with deep breath	Frequent spontaneous cough	Persistent uncontrollable cough
Chest pain	None	Discomfort just noticeable on exercise or deep breath	Marked discomfort on exercise or deep breath	Severe discomfort on exercise or deep breath
Duration of response	None	< 4 hours	>4 hours, but ≤24 hours	>24 hours

Impact of Responses	Normal	Mild	Moderate	Severe
Interference with normal activity	None	Few individuals choose to limit activity	Many individuals choose to limit activity	Most individuals choose to limit activity
Medical treatment	No change	Normal medication as needed	Increased frequency of medication use or additional medication	Physician or emergency room visit

^a An increase in nonspecific bronchial responsiveness of 100% is equivalent to a 50% decrease in PD₂₀ or PD₁₀₀.

Source: This table is reproduced from the 1996 O₃ AQCD (Table 9-2, page 9-25) (U.S. Environmental Protection Agency, 1996a).

decrements in FEV₁ measures (e.g., >10% decreases in FEV₁) at the end of the 6.6-h exposure to filtered air, a notable percentage did so with O₃ exposure. More specifically, the percentages showing >10% FEV₁ decrements at 0.08, 0.10, and 0.12 ppm O₃ were 26%, 31%, and 46%, respectively, as illustrated in Figure 8-1A. Newly available results from the Adams (2002, 2006) studies confirm the finding of a notable percentage (17 to 23%) of healthy adult subjects exhibiting >10% decrements in lung function indexed by FEV₁ measurements (as shown in Figure 8-1B and Figure 8-2). The newly available Adams (2002, 2006) results also further extend the earlier findings in terms of providing indications of >10% FEV₁ decrements among a small percentage (3 to 7%) following 6.6-h exposure of experimentally tested healthy adults to 0.06 ppm O₃ while engaged in moderate exercise (as also shown in Figures 8-1B and 8-2).

Given that comparable or enhanced pulmonary function decrements are observed in asthmatic children (accompanied by respiratory symptoms), it is reasonable to project that at least comparable percentages of such children would exhibit >10% FEV₁ decrements (and respiratory symptoms) at 0.06 or 0.08 ppm if tested under similar exercise conditions. The likelihood of notable lung function decrements (as indexed by >10% FEV₁ decrease) and/or respiratory symptoms occurring among asthmatic children in response to relatively low ambient O₃ exposures (i.e., <0.08 O₃ ppm, 1- to 8-h average) is further supported by the epidemiologic observations of the Thurston et al. (1997), Mortimer et al. (2002), and Höppe et al. (2003) studies discussed above in Section 8.7.2. The Höppe et al. (2003) study, in fact, suggests that about 20% of asthmatic children may be at risk of experiencing >10% FEV₁ decrements, which presumably could be sufficient to cause many of them to limit activities or increase medication use (as indicated in Table 8-3). Such projected potential impacts on asthmatic children in response to acute (1- to 8-h) ambient O₃ exposures of <0.08 ppm may possibly occur in the range of 0.06 to 0.07 ppm, based on the Höppe et al. (2003) study results. The observed lung function decrements, as well as any accompanying symptoms and/or increased medication use, would be consistent with and lend plausibility to the results of currently available time-series studies that found associations between ambient O₃ concentrations and increased asthma-related ED visits or hospitalizations, especially during the warm season.

8.7.5.2 Estimation of Potential Numbers of Persons in At-Risk Susceptible Population Groups in the United States

Although O₃-related health risk estimates may appear to be small, they may well be significant from an overall public health perspective due to large numbers of individuals in potential risk groups. Several subpopulations may be identified as having increased susceptibility or vulnerability to adverse health effects from O₃, including: older adults, children, individuals with preexisting pulmonary disease, and those with higher exposure levels, such as outdoor workers.

One consideration in the assessment of potential public health impacts is the size of various population groups that may be at increased risk for health effects associated with O₃-related air pollution exposure. Table 8-4 summarizes information on the prevalence of chronic respiratory conditions in the U.S. population in 2002 and 2003 (Dey and Bloom, 2005; Lethbridge-Çejku et al., 2004). Individuals with preexisting cardiopulmonary disease constitute a fairly large proportion of the population, with tens of millions of people included in each disease category. Of most concern here are those individuals with preexisting respiratory conditions, with approximately 11% of U.S. adults and 13% of children having been diagnosed with asthma and 6% of adults having COPD (chronic bronchitis and/or emphysema). Table 8-5 provides further information on the number of various specific respiratory conditions per 100 persons by age among the U.S. population during the mid-1990s. Asthma prevalence tends to be higher in children than adults.

In addition, subpopulations based on age group also comprise substantial segments of the population that may be potentially at risk for O₃-related health impacts. Based on U.S. census data from 2003, about 26% of the U.S. population are under 18 years of age and 12% are 65 years of age or older. Hence, large proportions of the U.S. population are included in age groups that are considered likely to have increased susceptibility and vulnerability for health effects from ambient O₃ exposure.

The health statistics data illustrate what is known as the “pyramid” of effects. At the top of the pyramid, there are approximately 2.5 million deaths from all causes per year in the U.S. population, with about 100,000 deaths from chronic lower respiratory diseases (Kochanek et al., 2004). For respiratory health diseases, there are nearly 4 million hospital discharges per year (DeFrances et al., 2005), 14 million ED visits (McCaig and Burt, 2005), 112 million ambulatory

Table 8-4. Prevalence of Selected Respiratory Disorders by Age Group and by Geographic Region in the United States (2002 [U.S. Adults] and 2003 [U.S. Children] National Health Interview Survey)

Chronic Condition/Disease	Adults (18+ Years)		Age (Years)				Region			
	Cases		18-44	45-64	65-74	75+	Northeast	Midwest	South	West
	(× 10 ⁶)	%	%	%	%	%	%	%	%	%
Respiratory Conditions										
Asthma	21.9	10.6	11.5	10.6	8.4	7.6	11	10.9	9.8	11.8
COPD										
Chronic Bronchitis	9.1	4.4	3.5	5.5	5.5	5.3	3.8	4	5.4	3.8
Emphysema	3.1	1.5	0.3	2	4.9	4.7	1.5	1.8	1.7	1.1
Chronic Condition/Disease	Children (<18 years)		Age (Years)			Region				
	Cases		0-4	5-11	12-17	Northeast	Midwest	South	West	
	(× 10 ⁶)	%	%	%	%	%	%	%	%	
Respiratory Conditions										
Asthma	9.1	12.5	7.5	14	14.7	14	13.5	11.8	11.2	

Source: Lethbridge-Çejku et al. (2004) for data on adults (18+ years); Dey and Bloom (2005) for data on children (<18 years).

Table 8-5. Acute Respiratory Conditions per 100 Persons/Year by Age Group in the United States (1996 National Health Interview Survey)

Type of Acute Condition	All Ages	Under 5 Years	5-17 Years	18-24 Years	25-44 Years	45+ Years		
						Total	45-64 Years	65+ Years
Respiratory Conditions	78.9	129.4	101.5	86	76.9	53.3	55.9	49
Common Cold	23.6	48.6	33.8	23.8	18.7	16.1	16.4	15.7
Other Acute Upper Respiratory Infections	11.3	13.1	15	16.1	11.6	7	7.5	6.1
Influenza	36	53.7	44.3	40.5	38.1	23.3	26.1	18.6
Acute Bronchitis	4.6	7.2 ^a	4.3	3.9 ^a	5.1	3.8	3.5	4.4 ^a
Pneumonia	1.8	3.9 ^a	1.7 ^a	1.4 ^a	1.3 ^a	2.0 ^a	0.9 ^a	3.8 ^a
Other Respiratory Conditions	1.7	2.9 ^a	2.4 ^a	0.4 ^a	2.0 ^a	1.1 ^a	1.5 ^a	0.5 ^a

^a Figure does not meet standard of reliability or precision.

Source: Adams et al. (1999).

care visits (Woodwell and Cherry, 2004), and an estimated 700 million restricted activity days per year due to respiratory conditions (Adams et al., 1999). Combining small risk estimates with relatively large baseline levels of health outcomes can result in quite large public health impacts. Thus, even a small percentage reduction in O₃ health impacts on cardiopulmonary diseases would reflect a large number of avoided cases.

Another key input for public health impact assessment is the range of concentration-response functions for various health outcomes. Epidemiologic studies have reported associations between short-term exposure to O₃ with mortality, hospitalizations for pulmonary diseases, ED visits for asthma, reduced lung function, and incidence of respiratory symptoms. Effect estimates for morbidity responses to short-term changes in O₃ tend to be larger and more variable in magnitude than those for mortality.

In addition to attribution of risks for various health outcomes related to O₃ and other copollutants, important considerations in assessing the impact of O₃ on public health include the size of population groups at risk, as well as the concentration-response relationship and potential

identification of threshold levels. Taken together, based on the above information, it can be concluded that exposure to ambient O₃ likely has a significant impact on public health in the United States.

8.8 SUMMARY AND CONCLUSIONS FOR OZONE HEALTH EFFECTS

This section summarizes the main conclusions derived from this integrated synthesis of information regarding health effects associated with ambient O₃ exposures. The conclusions are based on human clinical, animal toxicologic, and epidemiologic studies that have evaluated health effects associated with short-term, repeated, and long-term exposures to O₃ alone or in combination with other ambient pollutants. The controlled human exposure (or “clinical”) studies provide the clearest and most compelling evidence for human health effects directly attributable to acute exposures to O₃ per se. The evidence from human and animal toxicologic studies presented in Chapters 4, 5, and 6 are further useful in not only providing insights into possible mechanisms of action underlying different types of O₃-related health effects but, also, in helping to provide biological plausibility for health effects observed in epidemiologic studies assessed in Chapter 7. These empirical efforts are also aimed at identifying susceptible and vulnerable populations that are at potentially greater risk for effects of O₃ exposure. Overall, the new evidence generally supports and builds further upon key health-related conclusions drawn in the previous 1996 AQCD. The following conclusions integrate results from newly available studies with the scientific evidence assessed in that document.

1. Health Effects of Short-term Exposures to Ozone

The 1996 O₃ AQCD assessed a substantial body of evidence from toxicologic, human clinical, and epidemiologic studies. That AQCD concluded that short-term ambient O₃ exposure resulted in various respiratory health effects, including lung function decrements and increased respiratory symptoms in both healthy and asthmatic individuals exposed during moderate to heavy exercise to O₃ concentrations ranging down to the lowest levels (0.12 ppm for 1 h; 0.08 ppm for 6.6 to 8 h) tested in the available controlled human exposure studies. Such experimentally demonstrated effects were consistent with and lent plausibility to epidemiologic

observations highlighted in the 1996 AQCD of increases in daily hospital admissions and ED visits for respiratory causes. Epidemiologic evidence also provided suggestive evidence for an association between short-term O₃ exposure and mortality. However, there was essentially no evidence available in the 1996 O₃ AQCD regarding potential cardiovascular effects of short-term O₃ exposure. The newly-available evidence assessed in this revised O₃ AQCD notably enhances our understanding of short-term O₃ exposure effects, as summarized below, first in relation to respiratory morbidity endpoints and then cardiovascular effects and, lastly, mortality.

A. Respiratory Morbidity

Lung Function: Controlled exposure studies clearly demonstrate acute reversible decrements in lung function in healthy adults exposed to ≥ 0.08 ppm O₃ when minute ventilation and/or duration of exposure are increased sufficiently. On average, spirometric responses to O₃ exposure appear to decline with increasing age starting at approximately 18 years of age. There is considerable variability in responses between similarly exposed individuals, such that some may experience distinctly larger effects even when small group mean responses are observed. For example, healthy adults exposed to 0.08 ppm O₃ for 6.6 h with moderate exercise exhibited a group mean O₃-induced decrement in FEV₁ of about 6%, but a decrement of >10% was seen in 23% of these individuals. Also, exposure to 0.06 ppm O₃ caused >10% lung function decline in a small percentage (7%) of the subjects. Summer camp field studies conducted in southern Ontario, Canada, in the northeastern U.S., and in southern California have also reported lung function responses in pre-adolescent children associated with ambient O₃ levels.

With repeated acute O₃ exposures (0.12 to 0.45 ppm for 1 h) over several days, controlled exposure studies typically find that FEV₁ response to O₃ is enhanced on the second of several days of exposure, but spirometric responses become attenuated on subsequent days with these repeated exposures. However, this tolerance is lost after about a week without exposure. Animal toxicologic studies also provide extensive evidence that acute O₃ exposures alter breathing patterns so as to cause rapid shallow breathing (i.e., increased frequency and decreased tidal volume), an effect which attenuates after several days of exposure. Such results from controlled human exposure studies and animal toxicologic studies provide clear evidence of causality for the associations observed between acute (≤ 24 h) O₃ exposure and relatively small, but statistically significant declines in lung function observed in numerous recent epidemiologic

studies. Declines in lung function are particularly noted in children, asthmatics, and adults who work or exercise outdoors.

Respiratory Symptoms: Some young healthy adult subjects exposed in clinical studies to O₃ concentrations ≥ 0.08 ppm for 6 to 8 h during moderate exercise exhibit symptoms of cough and pain on deep inspiration. An increase in the incidence of cough has been found in clinical studies as low as 0.12 ppm in healthy adults during 1 to 3 h with very heavy exercise and other respiratory symptoms, such as pain on deep inspiration and shortness of breath, have been observed at 0.16 ppm to 0.18 ppm with heavy and very heavy exercise. These O₃-induced respiratory symptoms gradually decrease in adults with increasing age. With repeated O₃ exposures over several days, respiratory symptoms become attenuated, but this tolerance is lost after about a week without exposure. The epidemiologic evidence shows significant associations between acute exposure to ambient O₃ and increases in a wide variety of respiratory symptoms (e.g., cough, wheeze, production of phlegm, and shortness of breath) in asthmatic children. Epidemiologic studies also indicate that acute O₃ exposure is likely associated with increased asthma medication use in asthmatic children. On the other hand, an effect of acute O₃ exposure on respiratory symptoms in healthy children is not as clearly indicated by epidemiology studies, consistent with diminished symptom responses seen in healthy children in human clinical studies.

Airway Inflammation: Inflammatory responses have been observed subsequent to 6.6 h O₃ exposures to the lowest tested level of 0.08 ppm in healthy human adults. Some studies suggest that inflammatory responses may be detected in some individuals following O₃ exposures even in the absence of O₃-induced pulmonary function decrements in those subjects. With repeated O₃ exposures over several days, an attenuation of most inflammatory markers occurs. However, none of the several markers of lung injury and permeability evaluated show attenuation, indicating continued lung tissue damage during repeated exposure. Animal toxicologic studies provide extensive evidence that acute (1 to 3 h) O₃ exposures as low as 0.1 to 0.5 ppm can cause (1) lung inflammatory responses (typified by increased reactive oxygen species, inflammatory cytokines, influx of PMNs, and activation of alveolar macrophages); (2) damage to epithelial airway tissues, (3) increases in permeability of both lung endothelium and epithelium, and (4) increases in susceptibility to infectious diseases due to modulation of lung host defenses. Consistent with these experimental findings, there is also limited epidemiologic evidence

showing an association between acute ambient O₃ exposure and airway inflammation in children acutely exposed to ambient O₃ concentrations (1-h max O₃ of approximately 0.1 ppm). The extensive human clinical and animal toxicological evidence, together with the limited available epidemiologic evidence, is clearly indicative of a causal role for O₃ in inflammatory responses in the airways.

Airway Responsiveness: Controlled human exposure studies have found that acute O₃ exposure causes an increase in nonspecific airway responsiveness, as indicated by reductions in concentrations of methacholine or histamine required to produce a given decrease in FEV₁ or increase in SR_{aw}. Acute (2- or 3-h) O₃ exposure at 0.25 or 0.4 ppm of allergic asthmatic subjects, who characteristically already have somewhat increased airway responsiveness at baseline, was found to cause further increases in airway responsiveness in response to allergen challenges. Also, repeated daily exposure to 0.125 ppm O₃ for 4 days exacerbated lung function decrements in response to bronchial allergen challenges among persons with preexisting allergic airway disease, with or without asthma. Ozone-induced exacerbation of airway responsiveness persists longer and attenuates more slowly than O₃-induced pulmonary function decrements and respiratory symptom responses. Heightened airway responsiveness (reactivity) has also been observed in several laboratory animal species with acute exposures (1 to 3 h) to 0.5 to 1.0 ppm O₃. Ozone increases airway hyperreactivity to bronchoconstrictive agents (e.g., ovalbumin), and there is a temporal relationship between inflammatory cell influx and O₃-induced increases in airway reactivity. Several studies of sensitized laboratory animals showing O₃-induced increases in airway hyperreactivity are consistent with O₃ exacerbation of airway hyperresponsiveness reported in atopic humans with asthma. Airway responsiveness has not been widely examined in epidemiologic studies. However, the evidence from human clinical and animal toxicological studies clearly indicate that acute exposure to O₃ can induce airway hyperreactivity, thus likely placing atopic asthmatics at greater risk for more prolonged bouts of breathing difficulties due to airway constriction in response to various airborne allergens or other triggering stimuli.

Respiratory Hospital Admissions and Emergency Department Visits: Aggregate population time-series studies observed that ambient O₃ concentrations are positively and robustly associated with respiratory-related hospitalizations and asthma ED visits during the warm season. These observations are strongly supported by the human clinical, animal toxicologic, and epidemiologic evidence for lung function decrements, increased respiratory

symptoms, airway inflammation, and airway hyperreactivity. Taken together, the overall evidence supports a causal relationship between acute ambient O₃ exposures and increased respiratory morbidity outcomes resulting in increased ED visits and hospitalizations during the warm season.

B. Cardiovascular Morbidity

At the time of the 1996 O₃ AQCD, the possibility of O₃-induced cardiovascular effects was a largely unrecognized issue. Newly-available evidence has emerged since then which provides considerable plausibility for how O₃ exposure could exert cardiovascular impacts. This includes direct O₃ effects such as O₃-induced release from lung epithelial cells of platelet activating factor (PAF) that may contribute to blood clot formation that would increase the risk of serious cardiovascular outcomes (e.g. heart attack, stroke, mortality). Also, interactions of O₃ with surfactant components in epithelial lining fluid of the lung results in production of oxysterols and reactive oxygen species that may exhibit PAF-like activity contributing to clotting and/or exert cytotoxic effects on lung and heart cells. Other possible mechanisms may involve O₃-induced secretions of vasoconstrictive substances and/or effects on neuronal reflexes that may result in increased arterial blood pressure and/or altered electrophysiologic control of heart rate or rhythm. Consistent with the latter possibility, some field/panel studies that examined associations between O₃ and various cardiac physiologic endpoints have yielded limited epidemiologic evidence suggestive of a potential association between acute O₃ exposure and altered HRV, ventricular arrhythmias, and incidence of MI. Also, highly suggestive evidence for O₃-induced cardiovascular effects is provided by a few population studies of cardiovascular hospital admissions which reported positive O₃ associations during the warm season between ambient O₃ concentrations and cardiovascular hospitalizations. On the other hand, the only controlled human exposure study that evaluated effects of O₃ exposure on cardiovascular health outcomes found no significant O₃-induced differences in ECG, heart rate, or blood pressure in healthy or hypertensive subjects, but did observe an overall increase in myocardial work and impairment in pulmonary gas exchange. Also, some animal toxicological studies have shown O₃-induced decreases in heart rate, mean arterial pressure, and core temperature. Overall, then, this generally limited body of evidence is highly suggestive that O₃ directly and/or indirectly

contributes to cardiovascular-related morbidity, but much remains to be done to more fully substantiate links between ambient O₃ exposure and adverse cardiovascular outcomes.

C. Mortality

Numerous recent epidemiologic studies conducted in the United States and abroad have investigated the association between acute exposure to O₃ and mortality. Results from several large U.S. multicity studies as well as several single-city studies indicate a positive association between increases in ambient O₃ levels and excess risk of all-cause (nonaccidental) daily mortality. Determining cause-specific mortality is more difficult due to reduced statistical power by which to examine cause-specific associations and the lack of clarifying information on contributing causes of death. That is, attribution to one or the other of the more specific cardiopulmonary causes may underplay contributions of chronic cardiovascular disease to “respiratory” deaths (e.g., a heart attack victim succumbing to acute pneumonia) or vice versa. Nevertheless, consistent with observed O₃-related increases in respiratory- and cardiovascular-related morbidity, several newer multicity studies, single-city studies, and several meta-analyses of these studies have provided relatively strong epidemiologic evidence for associations between short-term O₃ exposure and all-cause mortality, even after adjustment for the influence of season and PM. In addition, consistently positive associations have been reported for O₃-related cardiovascular mortality across approximately 30 studies, with two well-conducted multicity studies in the United States and Europe yielding small, but statistically significant positive associations. Also, as discussed in Section 8.6, newly available experimental data from both animal and human studies provide evidence suggestive of plausible pathways by which risk of respiratory or cardiovascular morbidity and mortality could be increased by ambient O₃ either acting alone or in combination with copollutants in ambient air mixes. This overall body of evidence is highly suggestive that O₃ directly or indirectly contributes to non-accidental and cardiopulmonary-related mortality, but additional research is needed to more fully establish underlying mechanisms by which such effects occur.

2. Health Effects of Long-term Exposures to Ozone

In the 1996 O₃ AQCD, the available epidemiologic data provided only suggestive evidence that respiratory health effects were associated with chronic O₃ exposure. Animal toxicologic

studies indicated that chronic O₃ exposure caused structural changes in the respiratory tract, and simulated seasonal exposure studies in animals suggested that such exposures might have cumulative impacts. As summarized below, recent studies are generally consistent with the conclusions drawn in the previous 1996 AQCD.

A. Respiratory Morbidity

Lung Function: Recent epidemiologic studies observed that reduced lung function growth in children was associated with seasonal exposure to O₃; however, cohort studies investigating the effect of annual or multiyear O₃ exposure observed little clear evidence for impacts of longer-term, relatively low-level O₃ exposure on lung function development in children. The epidemiologic data, collectively, indicate that the current evidence is suggestive but inconclusive for respiratory health effects from long-term O₃ exposure.

Morphological Changes: Animal toxicologic studies continue to show chronic O₃-induced structural alterations in several regions of the respiratory tract including the centracinar region. Morphologic evidence from some recent studies using exposure regimens that mimic seasonal exposure patterns report increased lung injury compared to conventional chronic stable exposures. Infant rhesus monkeys repeatedly exposed to 0.5 ppm 8h/day O₃ for 11 episodes exhibited: (1) remodeling of the distal airways; (2) abnormalities in tracheal basement membrane; (3) eosinophil accumulation in conducting airways; and (4) decrements in airway innervation. Long-term O₃ exposure of rats to 0.5 or 1.0 ppm for 20 months resulted in upper respiratory tract mucus metaplasia and hyperplasia in the nasal epithelium (0.25 or 0.5 ppm, 8h/day, 7days/wk for 13 weeks). The persistent nature of these cytological changes raise the possibility of long-lasting alterations in human airways in response to chronic O₃ exposure, but it is highly uncertain as to what long-term patterns of exposure or O₃ concentrations in humans may be requisite to produce analogous morphological changes. Nor is it now possible to characterize the possible magnitude or severity of any such effects occurring in humans in response to ambient O₃ exposures at levels observed in the United States.

Incidence of Lung Cancer: The weight of evidence from recent animal toxicological studies and a very limited number of epidemiologic studies do not support ambient O₃ as a pulmonary carcinogen.

B. Mortality

Results from the few available epidemiologic studies are inconsistent regarding the association between long-term exposure to O₃ and mortality. There is little evidence to suggest a causal relationship between chronic O₃ exposure and increased risk for mortality in humans.

3. *Susceptibility or Vulnerability to Effects Associated with Exposure to Ozone*

Various factors have been shown to influence individuals' responses to environmental air pollutants. Factors that increase susceptibility to O₃-related effects include innate factors, such as genetic predisposition or developmental effects, or disease status. Other factors can lead to enhanced vulnerability to O₃-related effects, such as heightened exposures or activity patterns. In the 1996 O₃ AQCD, available evidence suggested that children, asthmatics, and outdoor workers were populations that may be more susceptible or vulnerable to effects of O₃ exposure. In addition, controlled human exposure studies also demonstrated a large variation in sensitivity and responsiveness to O₃ in studies of healthy subjects, but the specific factors that contributed to this intersubject variability were yet to be identified. Recent studies have built upon the evidence available in the previous review. Factors related to susceptibility or vulnerability to O₃ exposure-related effects are briefly summarized below:

People with Preexisting Pulmonary Diseases: Ozone-induced differential responses in lung function and AHR in people with allergic rhinitis suggest that asthmatics have potentially greater responses than healthy people with exposure to O₃. There is a tendency for slightly increased spirometric responses in mild asthmatics and allergic rhinitics relative to healthy young adults. Spirometric responses in asthmatics appear to be affected by baseline lung function, i.e., responses increase with disease severity. In addition, repeated O₃ exposure over several days has been shown to increase responsiveness to bronchial allergen challenge in subjects with preexisting allergic airway disease, with or without asthma. Asthmatics also show a significantly greater neutrophil response (18 h postexposure) than similarly-exposed healthy individuals. Epidemiologic studies have reported associations with a range of respiratory health outcomes in asthmatics, from decreases in lung function to hospitalization or ED visits for asthma, thus supporting this population group as being likely to experience increased risk for O₃-induced health effects. Although controlled human exposure studies have not found evidence of larger spirometric changes in people with COPD relative to healthy subjects, this may be due to

the fact that most people with COPD are older adults who would not be expected to have such changes based on their age. However, new epidemiologic evidence indicates that people with COPD may be more likely to experience other effects, including emergency room visits, hospital admissions, or premature mortality.

Age-related: Controlled human exposure studies have shown that lung function responses to O₃ varies with age, with responsiveness generally diminishing after about 18 to 20 years of age. Children and older adults thus have lesser respiratory symptoms with O₃ exposure than young healthy adults. Potentially increased O₃ doses can be received by individuals experiencing less severe respiratory symptoms. Evidence from newer epidemiologic studies supports the 1996 O₃ AQCD conclusions that children are more likely at increased risk for O₃-induced health effects. Notably, epidemiologic studies have indicated adverse respiratory health outcomes associated with O₃ exposure in children. In addition, recently published epidemiologic studies also suggest that older adults (aged ≥65 years) appear to be at excess risk of O₃-related mortality or hospitalization.

Heightened vulnerability due to greater exposures: Epidemiologic studies have provided some evidence to indicate that outdoor workers are more vulnerable to O₃-related effects, which is likely related to their increased exposure to ambient air pollution. Controlled human exposure studies clearly established differential biological response to O₃ based on physical activity (exertion). Epidemiologic studies also suggest that exercising (moderate to high physical exertion) children and adolescents appear to demonstrate increased responsiveness to ambient concentrations of O₃ and may be more likely to experience O₃-induced health effects. Animal studies show a similar impact of exercise on responsiveness to O₃.

Genetic susceptibility: Animal toxicologic studies provide supportive evidence to the observations of innate susceptibility. Various strains of mice and rats have demonstrated the importance, in general, of genetic background in O₃ susceptibility. Moreover, genetic and molecular characterization studies in laboratory animals identified genetic loci responsible for both sensitivity and resistance. Recent human clinical and epidemiologic studies also have shown that genetic polymorphisms for antioxidant enzymes and inflammatory genes (GSTM1, NQO1, and *Tmf-α*) may modulate the effect of O₃ exposure on pulmonary function and airway inflammation.

4. Health Effects of Ozone-Containing Pollutant Mixtures

The potential interaction of pollutant mixtures with O₃ is poorly understood and the animal studies reviewed in the 1996 O₃ AQCD reported additive, synergistic or even antagonistic effects depending on the exposure regimen and the endpoint studied. A few new controlled human exposure and animal toxicology studies reviewed in Chapters 4, 5, and 6 investigated health effects associated with O₃-containing pollutant mixtures of near ambient levels. As noted below, recent studies, although generally consistent with conclusions drawn in the 1996 O₃ AQCD, have added some new information, particularly with respect to interactions between O₃ and PM.

Controlled human exposure studies indicate that continuous exposure of healthy human adults to SO₂ or NO₂ increases bolus dose O₃ absorption, suggesting that co-exposure to other gaseous pollutants in the ambient air may enhance O₃ absorption. Other controlled human exposure studies that evaluated response to allergens in asthmatics (allergic and dust-mite sensitive) suggest that O₃ enhances response to allergen challenge. Consistent with these findings, animal toxicology studies also reported enhanced response to allergen on exposure to O₃.

A few other animal toxicology studies that exclusively investigated the co-exposure of PM and O₃ reported increased response (lung tissue injury, inflammatory and phagocytosis) to the mixture of PM + O₃ compared to either PM or O₃ alone. Recent investigations on the copollutant interactions using simulated urban photochemical oxidant mixes suggest the need for similar studies in understanding the biological basis for air pollutant mixture effects observed in epidemiologic studies.

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APPENDIX 8A

Summary of New Animal Toxicology, Human Clinical, and U.S./Canadian Epidemiologic Studies of Health Effects Associated with Ambient or Near-Ambient Ozone Exposures

Table 8A-1. Short-Term Ozone-Induced Health Effects Observed in Controlled Human Exposure Studies

Health Effects	Ozone Concentration	Exposure Duration/Activity	Subject Characteristics	Observable Effects	Reference
Pulmonary Function					
Short-term (1-2 h)	0.0-0.4	2 h rest or IE (4 × 15 min at $\dot{V}_E = 25$ or 35 L/min/m ² BSA)	Healthy NS 18 to 36 yr old mean age 24 yr	Statistical analysis of 8 experimental chamber studies conducted between 1980 and 1993 by the U.S. EPA in Chapel Hill, NC. Response decreased with age, was minimally affected by body size corrections, and was not more sensitive to O ₃ concentration than \dot{V}_E .	McDonnell et al. (1997)
Prolonged (6.6 h)	0.0-0.08	6.6 h IE (6 × 50min) $\dot{V}_E = 20$ L/min/m ² BSA	Healthy NS Males 23.5 ± 3.0 yrs Females 22.8 ± 1.2 yrs	FEV ₁ and symptom responses after 6.6 h exposure to 0.04 and 0.06 ppm not significantly different from FA.	Adams (2006)
	0.0-0.12	6.6 h IE (6 × 50min) $\dot{V}_E = 20$ L/min/m ² BSA	Healthy NS, 22.4 ± 2.4 yrs old	FEV ₁ and total symptoms after 6.6 h exposure to 0.04 ppm not significantly different from FA.	Adams (2002)
	(a) 0.08	6.6 h IE (6 × 50 min) $\dot{V}_E = 20$ L/min/m ² BSA	Healthy NS, 18 to 25 yr old	(a) FEV ₁ decreased 6.2% after 6.6 h in square-wave exposures. Total symptoms significantly increased at 5.6 and 6.6 h.	Adams (2003)
	(b) 0.08 (mean) varied from 0.03 to 0.15			(b) FEV ₁ decreased 5.6 to 6.2% after 4.6 to 6.6 h, respectively, in varied exposure; total symptoms significantly increased also after 4.6 to 6.6 h.	
Repeated	0.12	6.6 h 50 min exercise/10 min rest, 30 min lunch $\dot{V}_E = 38.8$ L/min	Healthy NS	FEV ₁ responses were maximal on first day of exposure (-13%), less on second day (-9%), absent thereafter. Symptoms only the first 2 d. Methacholine airway responsiveness was at least doubled on all exposure d, but was highest on the second day of O ₃ . Airway responsiveness was still higher than air control after 5 d of O ₃ exposure. Trend to lessened response, but it was not achieved after 5 d.	Folinsbee et al. (1994)

Table 8A-1 (cont'd). Short-Term Ozone-Induced Health Effects Observed in Controlled Human Exposure Studies

Health Effects	Ozone Concentration	Exposure Duration/Activity	Subject Characteristics	Observable Effects	Reference
Pulmonary Function (cont'd)					
Preexisting Disease	0.125 to 0.250	3 h IE (10 min rest, 15 min exercise on bicycle) $\dot{V}_E = 30$ L/min	Mild bronchial asthma 20-53 yr old	Mean early-phase FEV ₁ response and number of ≥20% reductions in FEV ₁ were significantly greater after 0.25 ppm O ₃ or 4 × 0.125 ppm O ₃ . Most of the ≥15% late-phase FEV ₁ responses occurred after 4 d of exposure to 0.125 ppm O ₃ , as well as significant inflammatory effects, as indicated by increased sputum eosinophils (asthma and allergic rhinitis) and increased sputum lymphocytes, mast cell tryptase, histamine, and LDH (asthma only).	Holz et al. (2002)
	0.125	3 h IE × 4 d	Allergic rhinitis 19-48 yr old		
	0.0 0.12	0.75 h IE (15 min exercise, 15 min rest) $\dot{V}_E = 40-46$ L/min	Asthmatics sensitive to SO ₂ 19 to 38 yrs old	No significant differences due to O ₃ between placebo and antioxidant supplement cohort in either spirometric responses or bronchial hyperresponsiveness to 0.1 ppm SO ₂ .	Trenga et al. (2001)
Copollutants					
SO ₂	0.10 to 0.12	1 h (mouthpiece) IE $\dot{V}_E \approx 30$ L/min 45-min exposure to air or O ₃ , followed by 15-min exposure to O ₃ or SO ₂	Allergic asthmatics, 12 to 18 yr old, medications withheld for at least 4 h before exposures	Prior exposure to O ₃ potentiated pulmonary function responses to SO ₂ ; decrements in FEV ₁ .	Koenig et al. (1990)
NO ₂	0.0 to 0.3	1 h (mouthpiece) IE $\dot{V}_E = 33$ L/min $\dot{V}_E = 35$ L/min	Healthy NS, 12 to 17 yr old Asthmatic 13 to 18 yr old	No significant differences in FEV ₁ and R _T between asthmatics and healthy, or between atmospheres and cohorts.	Koenig et al. (1988)

Table 8A-1 (cont'd). Short-Term Ozone-Induced Health Effects Observed in Controlled Human Exposure Studies

Health Effects	Ozone Concentration	Exposure Duration/Activity	Subject Characteristics	Observable Effects	Reference
Airway Hyperresponsiveness					
Preexisting Disease	0.125 0.250	3 h IE (10 min rest, 15 min exercise on bicycle) $\dot{V}_E = 30$ L/min 3h IE \times 4 d	Mild bronchial asthma Allergic rhinitis	Mean early-phase FEV ₁ response and number of $\geq 20\%$ reductions in FEV ₁ were significantly greater after 0.25 ppm O ₃ or 4 \times 0.125 ppm O ₃ . Most of the $\geq 15\%$ late-phase FEV ₁ responses occurred after exposure to 4 \times 0.125 ppm O ₃ , as well as significant inflammatory effects, as indicated by increased sputum eosinophils (asthma and allergic rhinitis) and increased sputum lymphocytes, mast cell tryptase, histamine, and LDH (asthma only).	Holz et al. (2002)
	0.12	45 min IE (exercise, rest, exercise) $\dot{V}_E = 3 \times$ resting	Physician diagnosed asthma; SO ₂ -induced airway hyperreactivity	The authors concluded O ₃ exposure increases bronchial responsiveness to SO ₂ in asthmatics and that antioxidant supplementation has a protective effects against this responsiveness, especially in the “more-severe” responders.	Trenga et al. (2001)
	0.12 Air antigen	1 h rest	Mild allergic asthma; 18 to 49 yr of age	No effect of O ₃ on airway response to grass or ragweed allergen.	Hanania et al. (1998)
	0.12	Rest	Atopic asthma	No effect of O ₃ on airway response to grass allergen.	Ball et al. (1996)
	0.10 to 0.40	1 h Light IE (2 \times 15 min on treadmill) $\dot{V}_E = 27$ L/min	Stable mild asthmatics with FEV ₁ $>70\%$ and methacholine responsiveness	No significant differences in FEV ₁ or FVC were observed for 0.10 and 0.25 ppm O ₃ -FA exposures or postexposure exercise challenge; 12 subjects exposed to 0.40 ppm O ₃ showed significant reduction in FEV ₁ .	Weymer et al. (1994)
Healthy Subjects	0.08 to 0.12	6.6 h IE at ≈ 39 L/min	Healthy NS, 18 to 32 yr old	33, 47, and 55% decreases in cumulative dose of methacholine required to produce a 100% increase in SRaw after exposure to O ₃ at 0.08, 0.10, and 0.12 ppm, respectively.	Horstman et al. (1990)

Table 8A-1 (cont'd). Short-Term Ozone-Induced Health Effects Observed in Controlled Human Exposure Studies

Health Effects	Ozone Concentration	Exposure Duration/Activity	Subject Characteristics	Observable Effects	Reference
Pulmonary Inflammation					
Short-term (1-2 h)	0.1	2 h Mild IE	Healthy subjects mean age ~30 yr	Markers of exposure in exhaled breath condensate, including increased 8-isoprostane, TBARS and LTB-4, and a marker of ROS-DNA interaction in peripheral blood leukocytes (8-OHdG), were increased in a sub-set of subjects bearing the wild genotype for NAD(P)H:quinone oxidoreductase and the null genotype for glutathione-S- transferase M1.	Corradi et al. (2002)
	0.12	2 h IE (15 min/30 min); (\dot{V}_E) \approx 20 L/min/m ² BSA	Healthy nonsmokers; mean age ~28 yr	Increase in the percentage of vessels expressing P-selectin in bronchial biopsies at 1.5 h postexposure. No changes in FEV ₁ , FVC, inflammatory cells or markers in BAL, or vessels expressing VCAM-1, E-selectin or ICAM-1 in biopsies.	Krishna et al. (1997)
Preexisting Disease	0.125 0.25	3 h exposures to both O ₃ concentrations and to FA; 3 h on four consecutive days to 0.125; study arms separated by >4 wks IE (15 min/30 min)	Allergic asthmatic and allergic rhinitic subjects; 19-53 yr of age	Repeated exposure caused increases in neutrophil and eosinophil numbers in both subject groups, as well as increased percentage and number of lymphocytes in the asthmatics.	Holz et al. (2002)
Cardiovascular					
Copollutants					
PM _{2.5}	0.0 to 0.12	2-2.5 h rest	Healthy NS 18 to 50 yr old	Neither systolic nor diastolic pressure has been affected by pollutants exposure despite a significant brachial artery constriction and a reduction in arterial diameter when compared to filtered air (p = 0.03).	Brook et al. (2002)

Table 8A-2. Effects of Acute O₃ Exposure on Lung Function in the U.S. and Canada

Reference, Study Location and Period	Study Population	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Change in Lung Function (95% CI) ^b		
			98th %	99th %	Range	Morning	Afternoon	Cross-day
Brauer et al. (1996) Fraser Valley, British Columbia, Canada Jun-Aug 1993	Berry pickers aged 10-69 yr (n = 58) repeatedly monitored over a 59-d period. Outdoor work shifts averaged 11 h in duration.	1-h max O ₃ : 40.3 SD 15.2	55	55	3-55	<u>FEV₁</u> : Lag 1: -6.4% (-8.0, -4.7)	<u>FEV₁</u> : Lag 0: -5.4% (-6.5, -4.3)	<u>FEV₁</u> : Lag 0: 0.0% (-1.7, 1.7)
Mortimer et al. (2000; 2002) Eight urban areas in the U.S.: Baltimore, MD; Bronx, NY; Chicago, IL; Cleveland, OH; Detroit, MI; East Harlem, NY; St. Louis, MO; Washington, DC Jun-Aug 1993	National Cooperative Inner City Asthma Study (NCICAS) cohort. Asthmatic children aged 4-9 yr (n = 846) repeatedly monitored over a 3-mo period.	8-h avg O ₃ (10 a.m.-6 p.m.): 48 SD not provided. Range of medians across cities: Approximately 34 to 58 (<5% of days exceeded 80).	64.3	66	28.8-66	<u>PEF</u> : Lag 1-5: All areas: -1.18% (-2.10, -0.26)	—	—
Ross et al. (2002) East Moline, IL and nearby communities May-Oct 1994	Mild and severe asthmatics aged 5-49 yr (n = 40) repeatedly monitored for a 5-mo period.	8-h max O ₃ : 41.5 SD 14.2 IQR 20	68.8	75	8.9-78.3	<u>PEF</u> : Lag 0-1: -0.96% (-1.78, -0.14)	<u>PEF</u> : Lag 0: -1.08% (-1.78, -0.37)	—
Naeher et al. (1999) Vinton, VA Summers 1995, 1996	Nonsmoking women aged 19-43 yr (n = 473) who recently delivered babies repeatedly monitored for a two-wk period.	24-h avg O ₃ : 34.87 SD 8.86 Range 8.74-56.63	74	79	13-87	<u>PEF</u> : Lag 1: -0.31% (-0.68, 0.07) Lag 1-3: -0.52% (-1.11, 0.07)	<u>PEF</u> : Lag 0: -0.36% (-0.73, 0.01) Lag 1-5: -1.11% (-1.88, -0.33)	—

Table 8A-2 (cont'd). Effects of Acute O₃ Exposure on Lung Function in the U.S. and Canada

Reference, Study Location and Period	Study Population	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Change in Lung Function (95% CI) ^b		
			98th %	99th %	Range	Morning	Afternoon	Cross-day
Korrick et al. (1998) Mount Washington, NH Summers 1991, 1992	Hikers aged 15-64 yr (n = 530) monitored before and after their hike. Hikes averaged 8 h in duration.	Avg of hourly O ₃ during each hike (approximately 8-h avg O ₃): 40 SD 12 Range 21-74	87	89	24-91	—	—	FEV ₁ : Lag 0: All hikers (n = 530): -1.53% (-2.82, -0.24) Respiratory disease status: Wheeze/asthma (n = 40): -4.47% (-7.65, -1.29) No wheeze/asthma (n = 490): -1.08% (-2.49, 0.33) Hours hiked: Hiked 2-8 h (n = 265): -0.99% (-2.70, 0.72) Hiked 8-12 h (n = 265): -2.07% (-3.78, -0.36)
Neas et al. (1995) Uniontown, PA Summer 1990	Symptomatic and asymptomatic 4th and 5th grade children (n = 83) who did not use any asthma medication during the previous year. Each child, on average, was monitored on 43 d.	All 12-h avg O ₃ (8 a.m.-8 p.m. and 8 p.m.-8 a.m.): 37.2 IQR 29.9 Maximum 87.5 Daytime 12-h avg O ₃ (8 a.m.-8 p.m.): 50.0 SD not provided.	96.5	98	15-98	—	PEF: Daytime 12-h avg O ₃ : Lag 0: -0.62% (-1.23, -0.01) Weighted by proportion of time spent outdoors during prior 12 h: Lag 0: -0.78% (-1.86, -0.31)	—

Table 8A-2 (cont'd). Effects of Acute O₃ Exposure on Lung Function in the U.S. and Canada

Reference, Study Location and Period	Study Population	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Change in Lung Function (95% CI) ^b		
			98th %	99th %	Range	Morning	Afternoon	Cross-day
Neas et al. (1999) Philadelphia, PA Jul-Sep 1993	Children aged 6-11 yr (n = 156) at two summer camps repeatedly monitored over 40 d.	12-h avg O ₃ (9 a.m.-9 p.m.): SW camp: 57.5 IQR 19.8 NE camp: 55.9 IQR 21.9	96.9	104.5	17.7-104.5	PEF: Lag 1: -0.74% (-1.54, 0.07) Lag 1-5: -0.76% (-2.65, 1.13)	PEF: Lag 0: -0.46% (-1.18, 0.27) Lag 1-5: -0.26% (-1.40, 0.88)	—
Delfino et al. (1997) Alpine, CA May-Jul 1994	Symptomatic asthmatics, children aged 10 to 15 yr (n = 13) and adults aged 24 to 47 yr (n = 9), repeatedly monitored for a 8-week period.	12-h avg O ₃ (8 a.m.-8 p.m.): Ambient: 64 SD 17 Range 34-103 Personal: 18 SD 14 Range 0-80 55% of personal O ₃ samples below limit of detection.	110	121	38-121	PEF: No effects observed using ambient or personal O ₃ concentrations. Effect estimates not presented.	PEF: No effects observed using ambient or personal O ₃ concentrations. Effect estimates not presented.	—
Linn et al. (1996) Three towns in California: Rubidoux, Upland, Torrance Fall-spring 1992-1993 and 1993-1994	School children (n = 269) repeatedly monitored for one week in fall, winter, and spring during their 4th and 5th grade school years.	24-h avg O ₃ : 23 SD 12 Range 3-53	150	164	2.5-192.5	FEV ₁ : Lag 1: -0.27% (-0.79, 0.24)	FEV ₁ : Lag 0: -0.19% (-0.76, 0.35)	FEV ₁ : Lag 0: -0.61% (-1.09, -0.14)

Table 8A-2 (cont'd). Effects of Acute O₃ Exposure on Lung Function in the U.S. and Canada

Reference, Study Location and Period	Study Population	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Change in Lung Function (95% CI) ^b		
			98th %	99th %	Range	Morning	Afternoon	Cross-day
Thurston et al. (1997) Connecticut River Valley, CT June 1991, 1992, 1993	Children aged 7-13 yr (total n = 166) with moderate-to-severe asthma repeatedly monitored over a 5-d period.	1-h max O ₃ : 1991: 114.0 1992: 52.2 1993: 84.6 1991-1993: 83.6	—	—	—	—	—	PEF: Lag 0: -1.2% (0.02, -2.4)

^aUsing O₃ data obtained for the study period in the location of the study, 8-h max O₃ concentrations were derived and statistics were calculated. The 98th and 99th percentile values for the full study period distribution are presented here (unless noted otherwise), along with the range (minimum-maximum) of concentrations. Since the time periods of the studies vary in length, from several weeks to over 10 yr, the 98th and 99th percentile values were selected for presentation here as a high study period concentration that roughly approximates a 4th maximum concentration, depending on the study period length.

^bPercent change in lung function per standard unit ppb O₃: 40 ppb for 1-h max O₃; 30 ppb for 8-h max O₃ or 8-h avg O₃; 25 ppb for 12-h avg O₃; and 20 ppb for 24-h avg O₃.

Table 8A-3. Effects of Acute O₃ Exposure on Asthma Emergency Department Visits in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Stieb et al. (1996) Saint John, New Brunswick, Canada May-Sep 1984-1992	1-h max O ₃ : Warm season: 41.6 Range 0-160 95th % 75	83	91	5-140.5	—	All ages: Lag 2: 9.3% (0.0, 18.7)
Cassino et al. (1999) New York City 1992-1995	24-h avg O ₃ : All yr: 17.5 IQR 14	83.3	88.8	3-114.6	All ages: All subjects: Lag 2: 8.3% (-0.8, 18.5)	—
					Note: Used Poisson GAM with default convergence criteria	
Wilson et al. (2005) Manchester, NH 1996-2000	8-h max O ₃ : Spring (Mar-May): 43.4 SD 9.7 Summer (Jun-Aug): 42.8 SD 14.6	Apr-Oct: 85	Apr-Oct: 93	Apr-Oct: 5-121	—	Apr-Sep: All ages: Lag 0: -3% (-14, 9) Age 0-14 yr: Lag 0: 6% (-25, 51) Age 15-64 yr: Lag 0: -6% (-21, 12)
Friedman et al. (2001) Atlanta, GA Jun-Sep 1996	1-h max O ₃ : Baseline (Jun 21-Jul 18 and Aug 5-Sep 1): 81.3 SD not provided. Intervention period (Jul 19-Aug 4): 58.6 SD not provided.	Jul 19-Aug 4: 85.8	Jul 19-Aug 4: 85.8	Jul 19-Aug 4: 20-85.8	—	Age 1-16 yr: Lag 0: 19% (-1, 43) Lag 0-2: 29% (2, 64)

Table 8A-3 (cont'd). Effects of Acute O₃ Exposure on Asthma Emergency Department Visits in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Jaffe et al. (2003) Columbus, OH Jun-Aug 1991-1996	8-h max O ₃ : Summer: 57 SD 16	98	106	25-117	—	Age 5-34 yr (medicaid recipients): Lag 3: 15.8% (-3.0, 36.8)
Jaffe et al. (2003) Cleveland, OH Jun-Aug 1991-1996	8-h max O ₃ : Summer: 50 SD 17	104	107	27-111	—	Age 5-34 yr (medicaid recipients): Lag 2: 3.0% (-8.7, 15.8)
Jaffe et al. (2003) Cincinnati, Cleveland, and Columbus, OH Jun-Aug 1991-1996	8-h max O ₃ : Summer: All three cities: Not reported.	104	108	24-124	—	Age 5-34 yr (medicaid recipients): All three cities: Lag 2 or 3: 9.3% (0.0, 19.1)
Jaffe et al. (2003) Cincinnati, OH Jun-Aug 1991-1996	8-h max O ₃ : Summer: 60 SD 20	106	116	24-124	—	Age 5-34 yr (medicaid recipients): Lag 2: 15.8% (0.0, 36.8)

Table 8A-3 (cont'd). Effects of Acute O₃ Exposure on Asthma Emergency Department Visits in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Wilson et al. (2005) Portland, ME 1998-2000	8-h max O ₃ : Spring (Mar-May): 43.7 SD 10.2 Summer (Jun-Aug): 46.1 SD 15.4	Apr-Oct: 108	Apr-Oct: 121	Apr-Oct: 15-142	—	Apr-Sep: All ages: Lag 0: 9% (3, 16) Age 0-14 yr: Lag 0: 13% (-5, 35) Age 15-64 yr: Lag 0: -3% (-14, 9) Age 65+ yr: Lag 0: -9% (-28, 16)
Tolbert et al. (2000) Atlanta, GA Jun-Aug 1993-1995	8-h max O ₃ : Summer: 59.3 SD 19.1	108.9	112.6	16.2-135.8	—	Age 0-16 yr: Lag 1: 6.1% (1.2, 11.3)
Peel et al. (2005) Atlanta, GA 1993-2000	8-h max O ₃ : Mar-Nov: 55.6 SD 23.8	Mar-Nov: 115	Mar-Nov: 124	Mar-Nov: 3-152	All available data (Mar-Nov): All ages: Lag 0-2: 2.6% (-0.5, 5.9)	All ages: Lag 0-2: 3.1% (0.2, 6.2)

^a Using O₃ data obtained for the study period in the location of the study, 8-h max O₃ concentrations were derived and statistics were calculated. The 98th and 99th percentile values for the full study period distribution are presented here (unless noted otherwise), along with the range (minimum-maximum) of concentrations. Since the time periods of the studies vary in length, from several weeks to over 10 yr, the 98th and 99th percentile values were selected for presentation here as a high study period concentration that roughly approximates a 4th maximum concentration, depending on the study period length.

^b Percent change in lung function per standard unit ppb O₃: 40 ppb for 1-h max O₃; 30 ppb for 8-h max O₃; and 20 ppb for 24-h avg O₃.

Table 8A-4. Effects of Acute O₃ Exposure on Total Respiratory and Asthma Hospital Admissions in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Yang et al. (2003) Vancouver, British Columbia, Canada 1986-1998	24-h avg O ₃ : All yr: 13.41 SD 6.61 IQR 9.74	42.7	47.3	1.1-71.9	<u>Total respiratory</u> : Age <3 yr: Lag 4: 50.4% (33.2, 71.4) Age 65+ yr: Lag 4: 28.5% (19.4, 40.5)	—
Burnett et al. (1997a) 16 Canadian cities 1981-1991	1-h max O ₃ : All yr: 31 95th % 60 Mean range across cities: 26-38 95th % 45-84 Apr-Dec: 32.9 95th % 64	Apr-Dec: 47.1	Apr-Dec: 51.3	Apr-Dec: 6.2-68.4	—	<u>Total respiratory</u> : Apr-Dec: All ages: Lag 1: 5.6% (3.4, 7.9)
Burnett et al. (1997b) Toronto, Ontario, Canada Summers 1992-1994	1-h max O ₃ : Summer: 41.2 IQR 22 12-h avg O ₃ : Summer: IQR 11.5	62	64	0-79	—	<u>Total respiratory</u> : 12-h avg O ₃ : All ages: Lag 1-3: 14.4% (8.7, 20.5)
Sheppard et al. (1999; reanalysis Sheppard, 2003) Seattle, WA 1987-1994	8-h max O ₃ : All yr: 30.4 IQR 20	65	73	2-100	<u>Asthma</u> : Age 0-64 yr: Lag 2: 10.7% (1.5, 20.1)	—

Table 8A-4 (cont'd). Effects of Acute O₃ Exposure on Total Respiratory and Asthma Hospital Admissions in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Lin et al. (2004) Vancouver, British Columbia, Canada 1987-1998	1-h max O ₃ : All yr: 28.02 SD 11.54 Range 1.93-105.50	—	—	—	<u>Asthma</u> : Age 6-12 yr: Males: Low SES: Lag 1: -35.5% (-52.4, -15.4) High SES: Lag 1: -17.8% (-39.6, 11.2) Females: Low SES: Lag 1: 32.6% (-7.9, 94.9) High SES: Lag 1: -22.5% (-48.9, 14.1)	—
Burnett et al. (1999) Toronto, Ontario, Canada 1980-1994	24-h avg O ₃ : All yr: 19.5 IQR 19	68.4	74.8	0.14-110.8	<u>Asthma</u> : All ages: Lag 1-3: 6.5% (3.7, 9.4)	—
Lin et al. (2003) Toronto, Ontario, Canada 1981-1993	1-h max O ₃ : All yr: 30.39 SD 17.87 Range 0-141	68.4	74.8	0.14-110.8	<u>Asthma</u> : Age 6-12 yr: Males: Lag 0: -7.8% (-22.6, 8.2) Females: Lag 0: -26.0% (-39.2, 8.2)	—
Burnett et al. (2001) Toronto, Ontario, Canada 1980-1994	1-h max O ₃ : Summer: 45.2 IQR 25	77.7	83.7	9-110.8	<u>Total respiratory</u> : Age <2 yr: Lag 0: 1.9% (-2.7, 6.8) Lag 0-4: 14.1% (4.9, 24.1)	<u>Total respiratory</u> : Age <2 yr: Lag 0: 6.7% (0.3, 13.6) Lag 0-4: 30.2% (16.9, 45.2)

Table 8A-4 (cont'd). Effects of Acute O₃ Exposure on Total Respiratory and Asthma Hospital Admissions in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Luginaah et al. (2005) Windsor, Ontario, Canada 1995-2000	1-h max O ₃ : All yr: 39.3 SD 21.4 Range 1-129	78	85	0-106	<u>Total respiratory:</u> All ages: Males: Lag 1: 5.4% (-10.5, 24.2) Females: Lag 1: -7.2% (-24.1, 13.5) Age 0-14 yr: Males: Lag 1: -7.6% (-33.4, 28.0) Females: Lag 1: 6.7% (-22.7, 47.0) Age 15-64 yr: Males: Lag 1: -5.6% (-43.5, 58.0) Females: Lag 1: -24.3% (-48.2, 10.5) Age 65+ yr: Males: Lag 1: 13.0% (-10.5, 42.8) Females: Lag 1: -7.5% (-29.4, 21.3)	—

Table 8A-4 (cont'd). Effects of Acute O₃ Exposure on Total Respiratory and Asthma Hospital Admissions in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Schwartz (1996) Spokane, WA Apr-Oct 1988-1990	1-h max O ₃ :	—	—	—	—	<u>Total respiratory:</u>
	Warm season: 41 IQR 12					Age 65+ yr:
	24-h avg O ₃ :					1-h max O ₃ : Lag 2: 40.3% (0.3, 96.1)
	Warm season: 29 IQR 9					24-h avg O ₃ : Lag 2: 21.4% (-5.8, 56.2)
						Note: Used Poisson GAM with default convergence criteria
Gwynn and Thurston (2001) New York City 1988-1990	24-h avg O ₃ :	90.6	106	6-125	<u>Total respiratory:</u>	—
	All yr: 22.1 IQR 14.1 Maximum 80.7				All ages: White: Lag 1: 1.1% (-0.4, 2.6) Non-white: Lag 1: 4.0% (2.5, 5.6)	
Schwartz et al. (1996) Cleveland, OH Apr-Oct 1988-1990	1-h max O ₃ :	91	99	5-120.3	—	<u>Total respiratory:</u>
	Warm season: 56 IQR 28					Age 65+ yr: Lag 1-2: 6.9% (1.5, 12.2)
Gwynn et al. (2000) Buffalo, NY 1988-1990	24-h avg O ₃ :	92.5	104	4.5-123	<u>Total respiratory:</u>	—
	All yr: 26.2 IQR 14.8 Maximum 87.6				All ages: Lag 1: 3.9% (1.8, 6.1)	
					Note: Used Poisson GAM with default convergence criteria	

Table 8A-4 (cont'd). Effects of Acute O₃ Exposure on Total Respiratory and Asthma Hospital Admissions in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Linn et al. (2000) Los Angeles, CA 1992-1995	24-h avg O ₃ : Winter: 14 SD 7 Spring: 32 SD 10 Summer: 33 SD 8 Fall: 15 SD 9	All yr: 98.8 Summer: 175	All yr: 106.9 Summer: 180	All yr: 4.6-143.2 Summer: 13.5-188	<u>Total respiratory:</u> Age 30+ yr: Lag 0: 1.61% (0.03, 3.22)	<u>Total respiratory:</u> Summer: Age 30+ yr: Lag 0: 1.21% (-1.53, 4.02)

^aUsing O₃ data obtained for the study period in the location of the study, 8-h max O₃ concentrations were derived and statistics were calculated. The 98th and 99th percentile values for the full study period distribution are presented here (unless noted otherwise), along with the range (minimum-maximum) of concentrations. Since the time periods of the studies vary in length, from several weeks to over 10 yr, the 98th and 99th percentile values were selected for presentation here as a high study period concentration that roughly approximates a 4th maximum concentration, depending on the study period length.

^bPercent change in lung function per standard unit ppb O₃: 40 ppb for 1-h max O₃; 30 ppb for 8-h max O₃; 25 ppb for 12-h avg O₃; and 20 ppb for 24-h avg O₃.

Table 8A-5. Effects of Acute O₃ Exposure on All-Cause Mortality in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Vedal et al. (2003) Vancouver, British Columbia, Canada 1994-1996	1-h max O ₃ : All yr: 27.3 SD 10.2 Range 3.1-75.1	40.5	47.3	1.6-58.7	—	All ages: Lag 0: 16.5% (5.3, 28.4) ^c
Villeneuve et al. (2003) Vancouver, British Columbia, Canada 1986-1999	24-h avg O ₃ : All yr: 13.4 Range 0.6-38.6	42.7	47.3	1.1-71.9	Age 65+ yr: Lag 0: 2.1% (-1.3, 5.3) Lag 0-2: 0.4% (-3.2, 4.4)	—
Fairley (1999; reanalysis Fairley, 2003) Santa Clara County, CA 1989-1996	8-h max O ₃ : All yr: 29 SD 15	67	74	2-105	All ages: Lag 0: 3.0% (-0.3, 6.4)	—
Moolgavkar (2003) Cook County, IL 1987-1995	24-h avg O ₃ : All yr: Median 18 Range 0.2-67 Summer: Median 28 Range 7-67	—	—	—	All ages: Lag 0: 0.28% (0.19, 0.36) Note: Used Poisson GAM with default convergence criteria.	All ages: Lag 0: 0.57% (0.42, 0.73) Note: Used Poisson GAM with default convergence criteria.
Ito and Thurston (1996) Cook County, IL 1985-1990	1-h max O ₃ : All yr: 38.1 SD 19.9	76	85.6	2.7-124	All ages: Lag 0-1: 3.9% (2.4, 5.8)	—
Lippmann et al. (2000; reanalysis Ito, 2003) Detroit, MI 1992-1994	24-h avg O ₃ : All yr: 25 IQR 18-30	80	85	4.3-101.3	All ages: Lag 0: 1.84% (-1.73, 5.53) Lag 0-3: 0.81% (-3.78, 5.63)	—

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Table 8A-5 (cont'd). Effects of Acute O₃ Exposure on All-Cause Mortality in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Chock et al. (2000) Pittsburgh, PA 1989-1991	1-h max O ₃ :	80	88.9	2.3-92.5	Age 0-74 yr: Lag 0: -1.5% (-5.6, 2.8)	Age 0-74 yr: Lag 0: -1.7% (-6.2, 3.1)
	Not reported.				Age 75+ yr: Lag 0: -1.8% (-6.0, 2.6)	Age 75+ yr: Lag 0: -0.6% (-5.3, 4.4)
Gamble (1998) Dallas, TX 1990-1994	24-h avg O ₃ :	81	86.3	2-98.7	All ages: Lag 1-2: 3.7% (0.8, 6.6)	All ages: Lag 1-2: 4.8%, p < 0.05
	All yr: 22 Range 0-160					
	Summer: 30 Range 0-160					
Lippmann et al. (2000; reanalysis Ito, 2003) Detroit, MI 1985-1990	24-h avg O ₃ :	81.5	88.7	2-123.5	All ages: Lag 0: 0.60% (-0.62, 1.83) Lag 0-3: 0.95% (-0.43, 2.35)	—
	All yr: 20.9 IQR 12.0-27.5					
Dockery et al. (1992) St. Louis, MO 1985-1986	24-h avg O ₃ :	—	—	—	All ages: Lag 1: 0.6% (-2.4, 3.6)	—
	All yr: 22.5 SD 18.5					
Dockery et al. (1992) Eastern Tennessee 1985-1986	24-h avg O ₃ :	—	—	—	All ages: Lag 1: -1.3% (-7.9, 5.8)	—
	All yr: 23.0 SD 11.4					

Table 8A-5 (cont'd). Effects of Acute O₃ Exposure on All-Cause Mortality in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Lipfert et al. (2000) 7 counties in Philadelphia, PA area 1991-1995	1-h max O ₃ : All yr: 44.76 SD 25.68	May-Sep: 88.8	May-Sep: 93.6	May-Sep: 2.3-116.6	All ages: Philadelphia: Lag 0-1: 2.49%, p < 0.055 4 counties in PA: Lag 0-1: 2.52%, p < 0.055 7 counties in PA and NJ: Lag 0-1: 2.84%, p < 0.055	—
Schwartz (2005) 14 U.S. cities 1986-1993	1-h max O ₃ : All yr: Median range 35.1 (Chicago, IL) to 60.0 (Provo, UT)	—	—	—	All ages: Lag 0: 0.76% (0.13, 1.40)	All ages: Lag 0: 1.04% (0.28, 1.77)
Samet et al. (2000; reanalysis Dominici et al., 2003) 90 U.S. cities (80 U.S. cities with O ₃ data) 1987-1994	24-h avg O ₃ : All available data: Mean range: Approximately 12 (Des Moines, IA) to 36 (San Bernardino, CA)	—	—	—	All available data: All ages: Lag 0: 0.84% (0.45, 1.24) Lag 1: 0.38% (0.05, 0.71)	All ages: Lag 1: 1.02% (0.46, 1.57)

Table 8A-5 (cont'd). Effects of Acute O₃ Exposure on All-Cause Mortality in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Bell et al. (2004) 95 U.S. communities 1987-2000	24-h avg O ₃ :	—	—	—	All available data:	All ages: Lag 0: 0.44% (0.16, 0.76)
	All available data from 95 communities: 26				All ages: Lag 0: 0.50% (0.24, 0.78) Lag 0-6: 1.04% (0.54, 1.55)	Lag 0-6: 0.78% (0.26, 1.30)
	55 communities with all year data: Median range: 14.38 (Newark, NJ) to 37.30 (Bakersfield, CA)				Age < 65 yr: Lag 0-6: 1.00% (0.20, 1.85)	
	40 communities with warm season only data: Median range: 20.41 (Portland, OR) to 36.15 (Memphis, TN)				Age 65-74 yr: Lag 0-6: 1.40% (0.56, 2.25)	
Klemm et al. (2004) Atlanta, GA 1998-2000	8-h max O ₃ :	—	—	6.63-124.41	Age 65+ yr: Lag 0-1: 4.2 (-2.4, 11.2)	—
	All yr: 47.03 SD 24.71					
Moolgavkar (2003) Los Angeles County, CA 1987-1995	24-h avg O ₃ :	—	—	—	All ages: Lag 0: 0.08% (0.01, 0.15)	All ages: Lag 0: 0.20% (0.06, 0.34)
	All yr: Median 24 Range 0.6-77 Summer: Median 36 Range 5-77				Note: Used Poisson GAM with default convergence criteria.	Note: Used Poisson GAM with default convergence criteria.
Ostro et al. (2000) Coachella Valley, CA 1989-1998	1-h max O ₃ :	—	—	—	All ages: -1% (-4, 3)	—
	All yr:				Note: Used Poisson GAM with default convergence criteria.	
	Indio: 62 Range 0-180 Palm Springs: 67 Range 0-190					

Table 8A-5 (cont'd). Effects of Acute O₃ Exposure on All-Cause Mortality in the U.S. and Canada

Reference, Study Location and Period	Mean O ₃ Levels (ppb)	Statistics for 8-h max O ₃ Air Quality Data (ppb) ^a			Standardized Percent Excess Risk (95% CI) ^b	
		98th %	99th %	Range	All Year	Warm Season
Moolgavkar et al. (1995) Philadelphia, PA 1973-1988	24-h avg O ₃ : Summer: 35.5 Range 1.3-159.0	—	—	—	—	All ages: Lag 1: 3.1% (1.8, 4.4)
Kinney et al. (1995) Los Angeles County, CA 1985-1990	1-h max O ₃ : All yr: 70 SD 41	115.3	130	5.4-156.1	All ages: Lag 1: 0.6% (0.0, 1.4)	—
Kinney and Özkaynak (1991) Los Angeles County, CA 1970-1979	1-h max O ₃ : All yr: Total oxidants (O _x): 75 SD 45	—	—	—	All ages: Lag 1: 0.79% (0.33, 1.26)	—
Ostro (1995) San Bernardino County and Riverside County, CA 1980-1986	1-h max O ₃ : Warm season: 140 Range 20-370	—	—	—	—	All ages: Lag 0: 0.8% (0.0, 2.0)

^a Using O₃ data obtained for the study period in the location of the study, 8-h max O₃ concentrations were derived and statistics were calculated. The 98th and 99th percentile values for the full study period distribution are presented here (unless noted otherwise), along with the range (minimum-maximum) of concentrations. Since the time periods of the studies vary in length, from several weeks to over 10 yr, the 98th and 99th percentile values were selected for presentation here as a high study period concentration that roughly approximates a 4th maximum concentration, depending on the study period length.

^b Percent change in lung function per standard unit ppb O₃: 40 ppb for 1-h max O₃; 30 ppb for 8-h max O₃; and 20 ppb for 24-h avg O₃.

^c Due to the low mean and relatively small variability in 1-h max O₃ concentrations in the study by Vedal et al. (2003), the standardized incremental change of 40 ppb may not be appropriate. The effect estimate in the warm season was 11.0% (95% CI: 3.6, 18.6) per mean level (27.3 ppb) increase in 1-h max O₃ at a 0-d lag.

Table 8A-6. Toxicological Effects of Acute Ozone Exposure in Animals

Effect	Species	O₃ Concentration and Duration of Exposure^a	References
Inflammation and increased epithelial/endothelial permeability	Rat	0.5 ppm × 3 h	Bhalla and Hoffman (1997)
	Mouse	0.26 ppm, 8 h/d, 5 d/wk × 1-90 d	Kleeberger et al. (2001)
	Dog	0.4 ppm × 6 h	Foster and Freed (1999)
Cardiovascular			
(↓ heart rate)	Rat	0.1 ppm × 5 h	Arito et al. (1997)
(↑ atrial natriuretic peptide)	Rat	0.5 ppm × 8 h	Vesely et al. (1994a,b,c)
Pulmonary Function			
(↑ AHR)	Rat	0.05 ppm × 4 h	Depuydt et al. (1999)
	Guinea Pig	0.3 ppm × 4 h	Segura et al. (1997)
(↓ V _t , ↑ f)	Rat ^b	0.1 ppm × 5 h	Arito et al. (1997)
Histological			
	Rat	0.2 ppm, 3-7 d	Dormans et al. (1999)
	Mouse	0.2 ppm, 3-7 d	Dormans et al. (1999)
	Guinea Pig	0.2 ppm, 3-7 d	Dormans et al. (1999)
Host Defense			
	Rat	0.8 ppm × 3 h	Dong et al. (1998)

^a Lowest reported ozone concentration from the current literature.

^b Effects on ventilation only occurred in young rats exposed to 0.1 ppm. No significant changes were seen between young and old animals at 0.3 and 0.5 ppm exposures.

9. ENVIRONMENTAL EFFECTS: OZONE EFFECTS ON VEGETATION AND ECOSYSTEMS

9.1 INTRODUCTION

A number of ozone (O₃) effect studies were published between 1996 and 2005, and they are reviewed in this document in the context of the previous O₃ air quality criteria documents (AQCDs) (U.S. Environmental Protection Agency, 1978, 1986, 1992, 1996b). The studies include multiple plant and tree species, multiple venues, and multiple research approaches from empirical data to process models. The research since 1996 continues to support and strengthen the conclusions from the previous O₃ AQCD, including that:

- (1) the entrance of O₃ into the leaf through the stomata is the critical step in O₃ effects;
- (2) current ambient concentrations in many areas of the country are sufficient to impair growth of numerous common and economically valuable plant and tree species;
- (3) effects can occur with only a few hourly concentrations above 80 ppb;
- (4) a plant's response to O₃ depends upon the cumulative nature of ambient exposure as well as the temporal dynamics of concentrations;
- (5) other environmental biotic and abiotic factors (e.g., insects, water, and nutrient availability, elevated CO₂, and temperature) are also influential to the overall impact of O₃ on plants and trees.

Research, to date, has continued to be focused at the species level with very few studies at the ecosystem level. Consequently, a high degree of uncertainty remains in our ability to assess the impact of O₃ on ecosystem services.

Although chamber exposures still dominate the effects literature, there has been a general shift away from chamber studies in favor of more field-based approaches. Field-based approaches include surveys of visible injury as well as physiological and growth studies using non-chambered free-air CO₂ exposure (FACE) systems. The FACE system studies published thus far have supported earlier observations of foliar injury and altered growth effects in open-top chambers (OTC) systems. Increased emphasis has also been placed on quantifying aspects

of O₃ uptake to better link ambient exposure monitoring with plant or tree response. Much of this progress has occurred in Europe in developing their O₃ air quality management tool, the “critical level.” The European research has developed exposure-response functions for several European crops and tree seedlings using OTC studies and for use in developing and testing models that simulate uptake. Evaluation of the above newly available information has added to our knowledge and provides new research directions, but it has not fundamentally altered the conclusions of the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996).

It is well known that O₃ is phytotoxic and that toxicity occurs only if O₃ or its reaction products reach the target tissues in the plant cell. Recent studies have provided an increased understanding of how O₃ interacts with the plant at the cellular level. This increased understanding of cellular-level O₃ effects has been translated into better models, more detailed schemata of how O₃ alters much of the basic metabolism of plants, and how to construct an index that more aptly captures the species, climate, and site factors that alter uptake. These results have and will continue to lead to better quantification of exposure and effect. However, the translation of these mechanisms into how O₃ is involved in altered cell metabolism and subsequent reductions in whole-plant productivity and ecosystem-level responses remains to be more fully resolved.

The ensuing sections of this chapter (Sections 9.2 to 9.8) are not intended to provide a comprehensive review of the environmental effects of O₃, but rather an assessment of key information published since the 1996 O₃ AQCD. More detailed discussion of the research since 1996 is provided in Chapter 9 Annex Sections AX9.1 to AX9.7 (in Volume 3 of this document). The framework for Chapter 9 follows the environmental effects chapter of the 1996 O₃ AQCD. First, an overview of various methodologies that have been, and continue to be, central to quantifying O₃ effects on vegetation is provided in Section 9.2 below (see Section AX9.1 for more detailed discussion). The adequacy of each methodology is discussed in the context of developing statistically robust data appropriate for assessing and predicting the risk of O₃ injury to vegetation resources. In Section 9.3, research is reviewed from the molecular to the biochemical and physiological levels in impacted plants, offering insight into the mode of action of O₃ (see also AX9.2). Then, the manner in which plants respond to O₃, as influenced by

numerous environmental biotic and abiotic factors, is next discussed (see also AX9.3). Quantifying these various modifiers is critical to scaling the response of individual plants to the community level and across varied landscapes and climates and is needed for regional-to-national assessments of risk. The development of indices of O₃ exposure or O₃ uptake is discussed in the context of their adequacy to realistically describe the ambient concentration-response relationships (see also AX9.4). Exposure-response relationships for a large number of crop species and cultivars, native vegetation, and tree species are also reviewed, tabulated, and compared to form the basis for an assessment of the potential risk from current levels of O₃ on vegetation resources (see also AX9.5). Available research by which to assess the impact of O₃ on ecosystems is also reviewed, along with data potentially available for estimating the loss of various ecosystem services (see also AX9.6). Finally, available research on the economic impact of O₃ effects on vegetation resources is briefly discussed (see also AX9.7).

9.2 METHODOLOGIES USED IN VEGETATION RESEARCH

Methodological advancements since 1996 have not fundamentally altered our understanding of O₃ effects on plants or ecosystems. Most of the new information confirms earlier conclusions and provides additional support for OTC use in assessing plant species and developing exposure-response relationships. A more in-depth discussion of this topic can be found in Annex Section AX9.1.

The majority of O₃ effects studies are fumigation studies that were conducted in controlled chambers, as noted in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996). That document noted that OTCs represented the best technology for determining statistically robust exposure-response models of O₃ and crop yield and plant biomass at that time. While OTCs are still the best method for conducting multiple, replicated controlled exposures of varying length and frequency for developing exposure-response relationships, several new approaches have been applied to O₃ effects research, most notably free-air exposure systems. FACE systems eliminate some concerns about closed or open-top chamber experiments including small plot size, altered microclimate within OTCs, and the effect of charcoal filtering on overall air quality

within OTCs. The FACE studies have, on the whole, confirmed what was already understood or hypothesized about how plants and plant assemblages respond to O₃. Some shortcomings of using free-air systems in O₃ research have also been identified, namely the relatively poor control of exposure levels, the presence of “hotspots” and the inability to decrease O₃ concentrations to below ambient levels when ambient concentrations are phytotoxic. Nonetheless, the application of FACE systems and other open-air systems to O₃-exposure research have greatly helped our scaling efforts and are, perhaps, the best current approach for studying the response of plant species mixtures to O₃ (Nussbaum and Fuhrer, 2000).

One of the advantages of the application of free-air systems to O₃ research is the ability to compare response of plants in open-field systems with results from OTCs. In particular, studies with quaking aspen (*Populus tremuloides* L.) performed in OTC and FACE systems and at sites along an ambient O₃ gradient have showed that O₃ foliar symptom expression is generally similar across these methodologies, supporting the previously observed level of variation among aspen clones in OTC studies (Isebrands et al., 2000, 2001; Karnosky et al., 1999). Recent exposure studies using the non-chambered FACE system with soybean cultivars in Illinois reported reductions in soybean yield of 15 to 25% (compared to ambient) in two year-long studies (Morgan et al., 2006). The results are similar to the reductions reported in multiple soybean studies conducted in the 1980s using OTC systems. Multiple-year exposure studies employing the FACE system reported foliar injury with reduced volume growth in aspen and maple similar to results reported from earlier OTC studies (Karnosky et al., 1999; Isebrands et al., 2001). Similar observations between the two exposure systems offer some corroborative support for the use of either system. Each has advantages and disadvantages as experimental tools and each can be used to effectively investigate O₃ effects. Extrapolation of the results from chamber studies depends on fully characterizing temperature, light, turbulence, and other chamber characteristics during exposures (Nussbaum and Fuhrer, 2000), but study design is equally important. Conducting studies with a large number of plant species across regions of the country where those species are indigenous is important in delineating regional climatic differences in order to reduce the uncertainty associated with extrapolating composited response

functions across regions and to identify relative risk to vegetation in relation to given O₃ exposure values (U.S. Environmental Protection Agency, 1996).

The lack of rural monitors continues to be a major problem in characterizing O₃ exposures in remote areas and complex landscapes so as to link effects to exposure in natural ecosystems. Since the 1996 O₃ AQCD, the use of passive samplers has expanded monitoring efforts to include remote areas that were previously uncharacterized. This has greatly enhanced our ability to link O₃ symptomology with elevated O₃ exposure in such remote areas. However, passive samplers do not capture the temporal dynamics of exposure. Therefore, passive samplers cannot substitute for active monitors when attempting to link exposure dynamics to plant response or when developing exposure- or dose-response relationships of much value as inputs for the standard setting process. To overcome this problem, Krupa et al. (2001, 2003) used models and data from a collocated O₃ monitor to estimate the underlying frequency distribution of hourly O₃ concentrations from passive samplers. Future development of passive monitor technology and data synthesis techniques holds promise, particularly as it is unlikely that extensive O₃ monitoring networks will be established in rural areas in the near future.

Exclusion methods that employ protective chemicals such as ethylenediurea (EDU) are the least disruptive of ambient culture conditions in the field, as noted in the 1996 O₃ AQCD. However, the level of protection afforded by EDU is site- and species-specific and is subject to local meteorologic conditions. In addition, new evidence suggests that EDU does not always provide greater protection at higher O₃ exposures and that the degree of protection by EDU largely depends on environmental conditions. Because of the variability observed in the level of protection provided and the fact that mechanisms of protection afforded by EDU and other exclusion methods are unknown, caution is needed in applying this approach to the study of O₃ effects in the field.

Advancements in biomonitoring have been made since the 1996 O₃ AQCD, primarily in the area of identification and symptom verification of O₃-sensitive species (Flagler, 1998; Krupa et al., 1998; Innes et al., 2001; Smith et al., 2003). The U.S. Department of Agriculture (USDA) Forest Service continues its program to monitor O₃ effects in forested ecosystems throughout the United States. Currently, 33 states participate in the program, which uses a grid system to

identify the location of plants showing foliar injury. Although results cannot be used for developing exposure-response relationships or for quantifying responses to O₃, they can provide an annual assessment and correlative information regarding the extent of O₃ injury occurring across many regions of the United States.

9.3 SPECIES RESPONSE AND MODE-OF-ACTION

Several steps in the process of O₃ uptake and toxicity are now better understood than in 1996. These advancements are important in refining hypotheses on O₃ uptake and mode of action on plants and in developing a flux-based index for use in quantifying response and, ultimately, for potential use in developing a secondary national ambient air quality standard (SNAAQS) for O₃. The new information available on the mode of action of O₃ is, in part, a result of improved molecular tools for following rapid changes that occur within the leaf (Ward et al., 1991; Pell et al., 1997; Sandermann, 2000). This new information is discussed in greater detail in Annex Section AX9.2.

Clearly, many changes occur within hours or possibly days following O₃ exposure (Sandermann, 1998). However, other O₃ effects take longer to occur (e.g., “carry-over” effects seen in the growing season following O₃ exposure) and tend to be most obvious only under exposure to low O₃ concentrations for long periods (Hogsett et al., 1989; Andersen et al., 1997; Langebartels et al., 1997). These low-exposure chronic effects have been linked to the senescence process or to some physiological response very closely linked to senescence (e.g., translocation, reabsorption, storage, and allocation of nutrients and carbon).

Langebartels et al. (1997) discussed “memory” or carry-over effects within the plant to explain sensitivity to frost in the winter following summertime O₃ exposure. Others have argued that this sensitivity is due to the nutrient status of the tree during the over-wintering phase of its life and to chronic (ongoing, less severe levels with fewer peaks at very high levels) exposure to ambient O₃ inducing (1) mineral nutrient deficiency; (2) alterations of normal metabolism, including translocation and allocation of carbohydrates and probably nitrogen; and/or (3) disturbance of normal transpiration and diurnal cycling, leading to water stress (Schmieden

and Wild, 1995). While generalized low nutrient concentrations within the foliage may not occur, localized deficiencies might. This is difficult to observe or to prove without a great deal of work on all portions of a tree and without a general hypothesis of what is occurring.

It is important to note that the dramatic strides made over the last few years in understanding the genetic makeup of plants, gene control, and signal transduction and control will accelerate in the future and translate into better models of the hypotheses listed above as well as into more detailed schemes of how O₃ alters basic plant metabolism. Thus, while our understanding of how O₃ interacts with the plant at the cellular level has dramatically improved (Assmann, 2003; Assmann and Wang, 2001; Rao and Davis, 2001), the translation of those mechanisms into how O₃ is involved with altered cell metabolism and the subsequent reductions in whole plant productivity and other physiological facts have not yet been fully achieved. As the understanding of wounding responses in plants and more information on genome details and varied plant mutants becomes available, the cellular and physiological responses of plants to O₃ exposures are slowly becoming clearer. However, more studies on a larger variety of species are needed before this type of information can be incorporated into indices of response and for consideration in developing SNAAQs.

9.4 MODIFICATION OF FUNCTIONAL AND GROWTH RESPONSES

It has been known for decades that several factors, both biotic and abiotic, alter plant response to O₃. However, only a few studies reported since the 1996 O₃ AQCD have improved our understanding of the role of these interactions in modifying plant O₃ response. Quantifying how these interactions alter plant O₃ response is a critical first step to reducing the uncertainty in extrapolating individual plant responses spatially or to higher levels of biological organization, e.g., ecosystems. Although the recent studies have not improved our ability to quantify the degree to which these factors modify plant O₃ response, they have reinforced the conclusions of the 1996 O₃ AQCD with regard to factors known to alter plant response to O₃. This new information is discussed in greater detail in Annex Section AX9.3.

In the area of biotic interactions, new evidence with regard to insect pests and diseases (see Docherty et al., 1997, and Flückiger et al., 2002 for recent reviews) has not altered the conclusions and uncertainties noted in the 1996 O₃ AQCD. Recent studies have supported the earlier conclusion that O₃ often increases the likelihood and success of insect attacks, but only with respect to chewing insects (e.g., Percy et al., 2002; Kopper and Lindroth, 2003). With the economically important group of sucking insects (e.g., aphids), no clear trends have been revealed in the latest studies (see reviews by Docherty et al., 1997; Flückiger et al., 2002). It seems that some insect problems could increase as a result of greater O₃ levels, but predicting the likelihood and severity of any particular O₃-plant-insect interaction is not possible at this time.

The situation is somewhat clearer with respect to interactions involving facultative necrotrophic plant pathogens, with O₃ exposure generally contributing to increased disease (Flückiger et al., 2002). With obligate biotrophic fungal, bacterial, and nematode diseases, however, twice as many reports indicate O₃-induced inhibitions as enhancements. This pattern is supported by the concept put forth by Dowding (1988) that pathogens that benefit from damage to cells are enhanced by pollution stress of their hosts, whereas those pathogens and pests that require healthy hosts are depressed by pollution stress. Despite frequent reports that infection by obligate biotrophs reduces the severity of O₃-induced foliar injury (e.g., Schraudner et al., 1996), such infection does not result in true “protection”, as the disease *per se* causes negative effects on the host plant. With obligate biotrophs, the nature of any interaction with O₃ is probably dictated by the unique, highly specific biochemical relationships between the pathogen and the host plant. At this time, therefore, although some diseases may become more widespread or severe as a result of exposure to O₃, it is still not possible to predict exactly which diseases are likely to present the greatest risks to crops and forests.

Recent studies of interactions between O₃ and root symbionts have supported conclusions put forth in the 1996 O₃ AQCD. Several studies have indicated that the functioning of tree root symbioses with mycorrhizae may be adversely affected by O₃ (e.g., Kytöviita et al., 2001), but there is also evidence that the presence of mycorrhizae may overcome O₃-enhanced root diseases (Bonello et al., 1993). Also, there is evidence that O₃ may encourage the spread of

mycorrhizae to the roots of uninfected trees. The role of O₃ in altering root symbionts, its interactions with soil organisms, and the subsequent feedback effects on plant growth represent one of the greatest areas of uncertainty in assessing the influence of O₃ on ecosystems and ecosystem services (Andersen, 2003).

The few recent studies of the impact of O₃ on intraspecific plant competition confirmed that grasses frequently show greater resilience than other types of plants. In grass-legume pastures, the leguminous species tend to suffer greater growth inhibition (Johnson et al., 1996; Nussbaum et al., 2000). The suppression of Ponderosa pine (*Pinus ponderosa* Laws.) seedling growth by blue wild-rye grass (*Elymus glaucus* Buckl.) was markedly increased by O₃ (Andersen et al., 2001). However, predicting the impact of O₃ on specific competitive situations, such as successional plant communities or crop-weed interactions, is not possible at this time.

Physical or abiotic factors play a large role in modifying plant response to O₃, and new information is available that supports the conclusions of the 1996 O₃ AQCD. Although some recent field studies have indicated that O₃ impact significantly increases with increased ambient temperature (Ball et al., 2000; Mills et al., 2000), other studies have indicated that temperature has little effect (Balls et al., 1996; Fredericksen et al., 1996). Temperature affects the rates of all physiological processes based on enzyme catalysis and diffusion; each process and overall growth (the integral of all processes) has a distinct optimal temperature range. It is important to note that a plant's response to changes in temperature will depend on whether it is growing near its optimum temperature for growth or near its maximum temperature (Rowland-Bamford, 2000). But temperature is very likely an important variable affecting plant O₃ response in the presence of the elevated CO₂ levels contributing to global climate change. In contrast, evidence continues to accumulate that O₃ exposure sensitizes plants to low temperature stress (Colls and Unsworth, 1992) and, also, that O₃ decreases below-ground carbohydrate reserves, which may lead to responses in perennial species ranging from rapid demise to impaired growth in subsequent seasons (i.e., carry-over effects) (Andersen et al., 1997).

Light, a component of the plant's physical environment, is an essential "resource" whose energy content drives photosynthesis and CO₂ assimilation. It has been suggested that increased light intensity may increase the O₃ sensitivity of light-tolerant species while decreasing that of

shade-tolerant species, but this appears to be an oversimplification with many exceptions. As pointed out by Chappelka and Samuelson (1998) and Topa et al. (2001), the interaction between O₃ sensitivity and light environment is complicated by the developmental stage as well as the light environment of individual leaves in the canopy.

Although the relative humidity of the ambient air has generally been found to increase the adverse effects of O₃ by increasing stomatal conductance (thereby increasing O₃ flux) abundant evidence also indicates that the ready availability of soil moisture results in greater O₃ sensitivity (Mills, 2002). The partial “protection” against the adverse effects of O₃ afforded by drought has been observed in field experiments and modeled in computer simulations (Broadmeadow and Jackson, 2000). There is also compelling evidence that O₃ can predispose plants to drought stress (Maier-Maercker, 1998). Hence, the response will depend to some extent upon the sequence in which the stressors occur; but, even though the nature of the response is largely species-specific, successful applications of model simulations have led to larger-scale predictions of the consequences of O₃-drought interactions. However, regardless of the interaction, the net result on short-term growth is negative; although in tree species, other responses (such as increased water use efficiency) could benefit long-term survival.

Mineral nutrients in the soil, other gaseous air pollutants, and agricultural chemicals constitute environmental chemical factors. The evidence regarding interactions with specific nutrients is still contradictory: some experimental evidence indicates that low general fertility increases sensitivity to O₃ (Whitfield et al., 1998; Landolt et al., 1997), while others have found less sensitivity with decreased fertility (Cardoso-Vilhena and Barnes, 2001). Simulation modeling of trees suggests that nutrient deficiency and O₃ may act less than additively.

Interactions of O₃ with other air pollutants have received relatively little attention since 1996 (see Barnes and Wellburn, (1998, and Fangmeier et al., 2002, for recent reviews). The situation with SO₂ remains inconsistent, but SO₂ seems unlikely to pose any additional risk to those related to other individual pollutants. With NO and NO₂, the situation is complicated by their nutritional value as a N source. More information is needed to predict the outcomes of different O₃-NO-NO₂ scenarios. The latest research into O₃-acid rain interactions has confirmed that, at realistic acidities, significant interactions are unlikely (Laurence et al., 1997; Momen

et al., 1997, 1999; Sayre and Fahey, 1999). A continuing lack of information precludes any generalizations about interactive effects of O₃ with NH₃, HF, or heavy metals. More evidence was reported for protective effects against O₃ afforded by the application of fungicides (Wu and Tiedemann, 2002).

Considerable emphasis during the last decade has been placed on research evaluating potential O₃ interactions with the components of global climate change: increased atmospheric CO₂, increased mean global temperatures, and increased surface level UV-B radiation. However, it must be noted that most of these studies have tended to regard increased CO₂ levels and increased mean temperatures as unrelated phenomena. Experiments into the effects of doubled CO₂ levels at today's mean ambient temperatures are of questionable value in trying to assess the impact of *climate change* on responses to O₃. To date, the limited experimental evidence and model simulations suggest that even though an enriched CO₂ atmosphere (~600 ppm) would more than offset the impact of O₃ on responses as varied as wheat (*Triticum aestivum* L.) yield or young Ponderosa pine growth, the concurrent increase in temperature would reduce, but probably not eliminate, the net gain (Batts et al., 1997; Van Oijen and Ewart, 1999; Constable et al., 1996). There is also some recent evidence that O₃ and ultraviolet radiation of wavelengths 280 to 320 nm (UV-B) interact in their effects on plant injury and photosynthesis (Schnitzler et al., 1999), but additional research is needed to fully understand how O₃ interacts with multiple climate change factors.

9.5 EFFECTS-BASED AIR QUALITY EXPOSURE INDICES

Exposure indices are metrics that relate measured plant damage (i.e., reduced growth) to monitored ambient O₃ concentrations over time to provide a consistent metric for reviewing and comparing exposure-response effects obtained from various studies. No new information is available since 1996 that alters the basic conclusions put forth in the 1996 O₃ AQCD. The 1996 AQCD (U.S. Environmental Protection Agency, 1996) focused on the research used to develop various exposure indices to quantify growth and yield effects in crops, perennials, and trees (primarily seedlings) and not foliar injury. The proposed indices included various functional and

statistical summaries of monitored hourly O₃ concentrations over designated time periods. The indices were developed through regression analyses of earlier exposure studies, which was accomplished by ordering the measured responses of growth and/or yield of crops and tree (seedling) species in response to O₃. The indices' development focused on consideration and inclusion of some, but not all, the factors that affect O₃ uptake and expression of effects (e.g., Lee et al., 1988).

In the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996b), it was concluded, based on the best available data, that those O₃ exposure indices that cumulated differentially weighted hourly concentrations were the best candidates for relating exposure to plant growth response. Greater weight was given to higher O₃ concentrations and daylight hours, and it was noted that the timing of peak concentrations and maximum plant conductance was critical in determining exposure impact on plants. Various weighting functions were used, including threshold-weighted (e.g., SUM06) and continuous sigmoid-weighted (e.g., W126) functions. Based on statistical goodness-of-fit tests, these cumulative, concentration-weighted indices could not be differentiated one from another. Additional statistical forms for O₃ indices have been discussed in Lee et al. (1988) and in Chapter 3 of this AQCD. A detailed discussion of effects-based O₃ exposure indices research since 1996 can be found in Annex Section AX9.4.

The few studies that have been published since the 1996 O₃ AQCD continue to support the earlier conclusions, including the importance of peak concentrations, and the duration and occurrence of O₃ exposures in altering plant growth and yield. In addition, a large body of new research, mostly out of Europe, addresses the need for an index related to the actual uptake of O₃ by the plant and the flux of O₃ from the atmosphere to the O₃-affected plant tissues. Despite additional research linking estimates of flux with plant response since 1996, information is still insufficient to identify a flux-based model that incorporates the necessary complexity across space and time to be non-site- or non-species-specific. Based on the current state of knowledge, exposure indices that cumulate and differentially weight the higher hourly average O₃ concentrations, but include the mid-level values, still represent the best approach for relating vegetation effects to O₃ exposure in the United States.

The new studies substantiated earlier conclusions on the role of exposure components including concentration, duration, and exposure patterns in determining plant growth response to O₃ (Yun and Laurence, 1999; Oksanen and Holopainen, 2001). Recent studies using different exposure patterns have confirmed earlier studies on the role of higher concentrations and exposure duration (Nussbaum et al., 1995). A role for higher concentrations is inferred based on improved air quality in regions in the western United States (Lefohn and Shadwick, 2000). For example, the O₃ reductions in the San Bernardino Mountain area since the late 1970s are associated with reductions in the higher hourly average O₃ concentrations, the number of hours of concentrations ≥ 0.95 ppm, and the cumulative concentration-weighted exposure index (Lee et al., 2003). The mid-range concentrations appeared to be relatively unchanged over the period of 1980 to 2000. General forest improvement has been reported following a decrease of O₃ along a decreasing gradient of exposure (Miller and Rechel, 1999; Arbaugh et al., 2003; Tingey et al., 2004). These studies suggest that the focus should be on the higher O₃ concentrations, while including the lower levels, when estimating the effects of O₃ precursor emission reduction strategies on vegetation. The area has also experienced increasing deposition of N over the same time period and, by many indicators, the soil is considered to be N-saturated (Fenn et al., 1996). The relative role of N, however, in the measured or simulated tree response in the area has not been quantified.

New studies have demonstrated the potential disconnection of peak events and maximal stomatal conductance at xeric to mesic sites in California (Grulke et al., 2002; Panek et al., 2002; Panek, 2004). In addition, a few studies have indicated that O₃ uptake during nighttime hours is greater than previously thought (Grulke et al., 2004; Massman, 2004), and a review of the literature suggests a large number of species exhibit some degree of conductance at night (Musselman and Minnick, 2000). These studies suggest the need for a reconsideration of cumulating exposure 24 h/day and not just during daylight hours in O₃ exposure index determinations. This lack of coincidence in temporal patterns of conductance and peak ambient concentrations introduces uncertainty in assessing the impact of O₃. The use of an exposure index that does not consider regionally unique climate and site factors modifying stomatal conductance may, as a result, under- or overestimate growth effects. The shortcomings of an

ambient exposure-based index is especially apparent when assessing the potential impact of O₃ across broad climatic regions of the United States or Europe. Various means to overcome this potential problem were addressed in several new studies, and one solution would be to add other components to the present statistical summaries of exposure indices (e.g., meteorological) to develop flux-based indices. However, the increased biological and meteorological information in these indices may make them more regional in their applicability.

A number of studies have taken a flux-based approach to improve upon the ambient air concentration-based (i.e., exposure indices) approach as a means to address the issue of assessing risk of O₃ across different climatic regions. The European acceptance and use of the flux-based critical values is, in part, a recognition of the landscape scaling problems associated with ambient exposure-based indices. A great deal of progress has occurred in developing and testing stomatal models that may be generally applicable across certain vegetation types (Emberson et al., 2000; Pleijel et al., 2000; Danielsson et al., 2003; Grünhage and Jäger, 2003; Matyssek et al., 2004). While a flux-based approach is preferred, a cautionary argument has been advanced in a few publications, based on the nonlinear relationship between O₃ uptake and foliar injury (growth was not assessed). The concern is that not all O₃ stomatal uptake results in a yield reduction, which depends to some degree on the amount of internal detoxification occurring with each particular species. Those species having high amounts of detoxification potential may, in fact, show little relationship between O₃ stomatal uptake and plant response (Musselman and Massman, 1999).

Given the current state of knowledge and the best available data, exposure indices that cumulate and differentially weight the higher hourly average concentrations and also include the mid-level values continue to offer the most defensible approach for use in developing response functions and comparing studies, as well as for defining future indices for vegetation protection. A large database exists that has been used for establishing exposure-response relationships; however, at this time, such a database does not exist for relating O₃ flux to growth response.

It is anticipated that, as the overlapping relationships of conductance, concentration, and defense mechanisms are better defined, the flux-based indices may be able to predict vegetation injury and/or damage across varied landscapes and climates with more accuracy than the

exposure-response models. However, that is not the case at this time. The translation of these indices from research and assessment tools to air quality standards has the additional need to be simple, understandable, and adaptive to a manageable monitoring program.

9.6 OZONE EXPOSURE-PLANT RESPONSE RELATIONSHIPS

Data published since 1996 continue to support the conclusions of previous O₃ AQCDs that there is strong evidence that ambient concentrations of O₃ cause foliar injury and growth and yield damage to numerous common and economically valuable plant and tree species. For annual vegetation, the data summarized in Table AX9-16 (see Annex Section AX9.5) show a range of growth and yield responses both within species and among species. Nearly all of these data were derived from OTC studies, with only two studies using open-air systems in the United Kingdom (Ollerenshaw and Lyons, 1999; Ollerenshaw et al., 1999). It continues to be difficult to compare studies that report O₃ exposure using different indices, such as AOT40, SUM06, W126, or 7-h or 12-h mean values. The AOT40, SUM06, and W126 indices are defined as follows:

AOT40: the seasonal sum of the difference between an hourly concentration above the threshold value of 40 ppb, minus the threshold value of 40 ppb;

SUM06: the seasonal sum of hourly concentrations at or above the threshold value of 60 ppb; and

W126: a sigmoid functional weighting of all hourly concentrations for the season.

When such index comparisons can be made, the results of recent research confirm earlier results assessed in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996).

A summary of earlier literature concluded that a 7-h, 3-month mean of 49 ppb O₃ corresponding to a SUM06 exposure of 26 ppm·h, would cause a 10% loss in 50% of 49 experimental cases (Tingey et al., 1991). Recent data summarized in Table AX9-16 support this conclusion and, more generally, indicate that ambient O₃ exposures can reduce the growth and yield of annual species. Some annual species such as soybean [*Glycine max* (L.) Merr.] are more sensitive, and

greater losses in such species may be expected (Table 9-16). Thus, the recent scientific literature supports the conclusions of the 1996 O₃ AQCD that ambient O₃ concentrations are reducing the yield of major crops in the United States.

Much research in Europe has used the cutoff-concentration weighted, cumulative-exposure statistic AOT40, and substantial effort has gone into developing “Level-1” critical levels for vegetation using this index. Based on regression analysis of 15 OTC studies of spring wheat, including one U.S. study and 14 studies from locations ranging from southern Sweden to Switzerland, an AOT40 value of 5.7 ppm·h was found to correspond to a 10% yield loss, and a value of 2.8 ppm·h corresponded to a 5% yield loss (Fuhrer et al., 1997). Because a 4 to 5% decrease could be detected with a confidence level of 99%, 3 ppm·h was selected by the European Union as the AOT40 critical level in 1996 (Kärenlampi and Skärby, 1996).

In addition to reductions in crop yield, O₃ may also reduce the quality or nutritive value of annual species. Many studies have found O₃ effects in various measures of plant organs that affect quality, with most of those studies focusing on characteristics important for food or fodder. These studies indicate that ambient O₃ may have economically important effects on the quality of crop and forage species. Previous O₃ AQCDs have concluded that visible symptoms on marketable portions of crops and ornamental plants can occur with seasonal 7-h mean O₃ exposures of 40 to 100 ppb (U.S. Environmental Protection Agency, 1978; 1986; 1992; 1996). The recent scientific literature does not refute this conclusion.

The use of OTCs may reverse the usual vertical gradient in O₃ that occurs within a few meters above the ground surface (see Annex Section AX9.1). This reversal suggests that OTC studies may, to some degree, overestimate the effects of an O₃ concentration as measured several meters above the ground. However, such considerations do not invalidate the conclusion of the 1996 O₃ AQCD that ambient O₃ exposures are sufficient to reduce the yield of major crops in the United States. Recent results from the non-chambered FACE systems, having different vertical gradients than chambered systems, report similar results for foliar injury, growth response and yield reductions as reported for the chambered OTC systems (Karnosky et al., 1999; Isebrands et al., 2001; Morgan et al., 2006).

As found for single-season agricultural crops, yields of multiple-year forage crops are reduced at O₃ exposures that occur over large areas of the United States. This result is similar to that reported in the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996). When species are grown in mixtures, O₃ exposure can result in the increased growth of O₃-tolerant species, which exacerbates the growth decrease of O₃-sensitive species. Because of this competitive interaction, the total growth of the mixed-species community may not be affected by O₃ exposure. However, in some cases, mixtures of grasses and clover species have shown significant decreases in total biomass growth in response to O₃ exposure in studies in the United States and in Sweden. In Europe, a provisional AOT40 critical level of 7 ppm·h over 6 months has been proposed by the European Union as a value to protect sensitive herbaceous perennial plant species from the adverse effects of O₃.

For deciduous tree species, recent evidence from FACE and OTC studies supports results observed in previous OTC studies. For example, a series of O₃-FACE studies was undertaken in Rhinelander, WI (Isebrands et al., 2000, 2001). These studies showed that O₃ symptom expression was generally similar in OTC and FACE studies and in studies at sites along an ambient O₃ gradient, supporting the previously observed variation among aspen clones obtained using OTCs (Karnosky et al., 1999). As has been observed in previous O₃ AQCDs, root growth is often found to be the most sensitive biomass response to O₃.

Results of studies since 1996 support the conclusion of the 1996 O₃ AQCD (U.S. Environmental Protection Agency, 1996) that deciduous trees are generally less sensitive to O₃ than are most annual plants, with the exception of a few very sensitive genera such as sensitive clones or genotypes of the *Populus* genera and sensitive species such as black cherry (*Prunus serotina* Ehrh.). However, the data presented in Table AX9-18 (see Annex Section AX9.5) suggest that ambient exposures that occur in the United States can sometimes reduce the growth of seedlings of deciduous species. Results from some multiyear studies have shown a pattern of increased effects in subsequent years (Hogsett et al., 1989; Anderson et al., 1997; Karlsson et al., 1995). In some cases, however, growth decreases due to O₃ have become less significant or even disappeared over time (Karlsson et al., 2002). While some mature trees show greater O₃

sensitivity than do seedlings in physiological parameters such as net photosynthetic rate, these effects may not translate into measurable reductions in biomass growth (Samuelson and Kelly, 2001). However, because even multiyear experiments do not expose trees to O₃ for more than a small fraction of their life span and because competition may, in some cases, exacerbate the effects of O₃ on individual species, determining O₃ effects on mature trees remains a significant challenge.

In Europe, a Level I critical level has been set for forest trees based on OTC studies of European beech (*Fagus sylvatica* L.) seedlings and is defined as an AOT40 value of 10 ppm·h for daylight hours for a 6-month growing season (Kärenlampi and Skärby, 1996). However, other studies show that other species, such as silver birch (*Betula pendula* Roth.), may be more sensitive to O₃ than beech (Pääkkönen et al., 1996).

As found for other tree species, various evergreen tree species and genotypes have widely varying O₃ sensitivities. Based on OTC studies with seedlings, major evergreen species in the United States are generally less sensitive than are most deciduous trees, and slower-growing evergreen species are less sensitive than are faster-growing species. There is evidence that plant interaction stress, such as competition stress, may increase the O₃ sensitivity of trees. As in studies of deciduous species, most experiments with evergreen species have only covered a very small portion of the life span of a tree and have been conducted with seedlings, so estimating effects for mature evergreens is difficult.

For all types of perennial vegetation, cumulative effects over more than one growing season may be important, and studies for only a single season may underestimate effects. Mature trees may be more or less sensitive to O₃ than are seedlings, depending on the species, but information on physiological traits can be used to predict such differences in specific cases. In some cases, mature trees may be more sensitive to O₃ than seedlings due to differences in gas exchange rates or growth rates, to greater cumulative exposures, or to the interaction of O₃ with other stressors.

9.7 EFFECTS OF OZONE EXPOSURE ON NATURAL ECOSYSTEMS

There is evidence that tropospheric O₃ is an important stressor of ecosystems, with documented impacts on the biotic condition, ecological processes, and chemical and/or physical nature of natural ecosystems (See Table AX9-22; Annex Section AX9.6; Figure 9-1). In turn, the effects of O₃ on individual plants and processes are scaled up through the ecosystem, affecting processes such as energy and material flow, inter- and intraspecies competition, and net primary productivity (NPP). Thus, effects on individual keystone species and their associated microflora and fauna, effects that have been shown experimentally, may cascade through the ecosystem to the landscape level, although this has not yet been demonstrated. By affecting water balance, cold hardiness, tolerance to wind and by predisposing plants to insect and disease pests, O₃ has the potential to impact the occurrence and severity of natural disturbances (e.g., fire, erosion). Despite the possible occurrence of such effects, there are essentially no instances where ecosystem-level, highly integrated studies have conclusively shown that O₃ is indeed altering ecosystem structure and/or function.

Systematic injury surveys demonstrate that foliar injury occurs to O₃-sensitive species in many regions of the United States (Chappelka et al., 1997; Campbell et al., 2000; Coulston et al., 2003; Smith et al., 2003) and Europe (Braun et al., 1999). However, the frequent lack of correspondence between foliar symptoms and growth effects means that other methods must be used to estimate the regional effects of O₃ on tree growth rates (e.g., Rebbeck, 1996; Kouterick et al., 2000). Investigations of the radial growth of mature trees in combination with data from many controlled studies with seedlings and a few studies with mature trees suggest that ambient O₃ is reducing the growth of mature trees in some locations (Somers et al., 1998). Studies using models based on tree physiology and forest stand dynamics suggest that modest effects of O₃ on growth may accumulate over time and may interact with other stressors (Laurence et al., 2001, 2003). For mixed-species stands, such models predict that overall stand growth rate is generally not likely to be affected. However, competitive interactions among species may change as a result of growth reductions of O₃-sensitive species (Weinstein et al., 2001). These results suggest that O₃ exposure over decades may alter the species composition of forests in some regions.

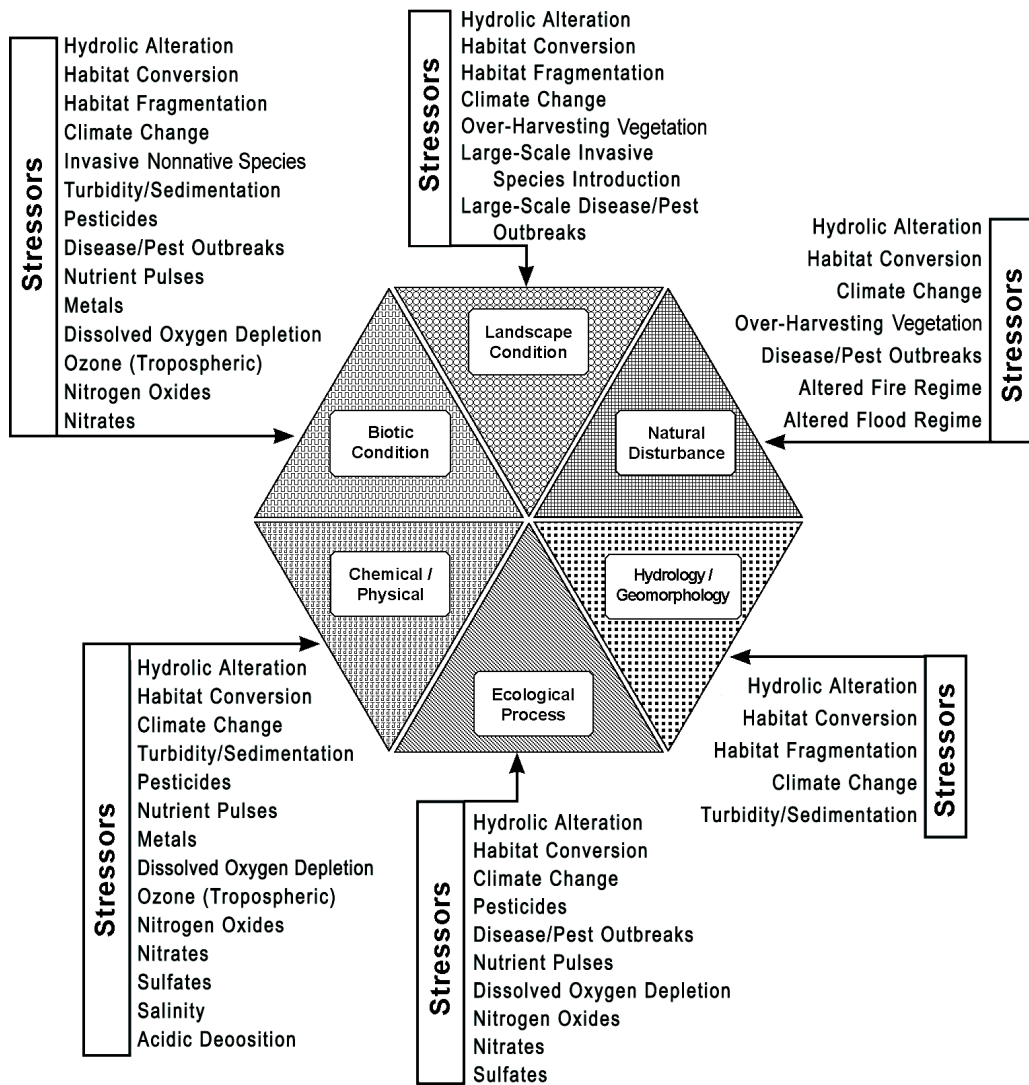


Figure 9-1. Common anthropogenic stressors and the essential ecological attributes they affect.

Source: Modified from Young and Sanzone (2002).

Despite the increased understanding of possible ecosystem effects of O₃, the database demonstrating and quantifying the degree to which O₃ is altering natural ecosystems is sparse. Much of the speculative O₃ impact on ecosystems must be inferred from a number of case studies of forest-plot field-based data reporting on a number of different species. One means to discuss our current knowledge is by listing the areas in which more information is needed, as shown below:

Ecosystem processes. Very little is known about the effects of O₃ on water, carbon, and nutrient cycling, particularly at the stand and community levels. Effects on below-ground ecosystem processes in response to O₃ exposure alone and in combination with other stressors are critical to projections at the watershed and landscape levels. Little is yet known about the effects of O₃ on structural or functional components of soil food webs or how these impacts could affect plant species diversity (Andersen, 2003).

Biodiversity and genetic diversity. The study of genetic aspects of O₃ impacts on natural ecosystems has been largely based on correlations, and it remains to be shown conclusively whether O₃ affects biodiversity or genetic diversity (Davison and Barnes, 1998; Pitelka, 1988; Winner et al., 1991). Studies of competitive interactions under elevated O₃ levels are needed (Laurence and Andersen, 2003). Reexaminations via new sampling of population studies to bring a time component into previous studies showing spatial variability in population responses to O₃ are also needed. These studies could be strengthened by modern molecular methodologies to quantify impacts on diversity.

Natural ecosystem interactions with the atmosphere. Little is known about feedbacks between O₃ and climate change on the production of volatile organic compounds (VOCs), which, in turn, could affect O₃ production (Fuentes et al., 2001). At moderate to high O₃ exposure sites, aberrations in stomatal behavior could significantly affect individual tree water balance in O₃-sensitive trees, and if the sensitive tree species is dominant, the hydrologic balance at the watershed and landscape levels could also be affected. This has not been addressed in any model, because O₃-exposure effects, if included at all in the modeling effort, have often assumed a linear relationship between assimilation and stomatal conductance. Interaction studies with other components of global change (i.e., warming, increasing atmospheric CO₂, N deposition) or with various biotic stressors are needed to better predict complex interactions likely to occur in the future (Laurence and Andersen, 2003). Whether O₃ will negate the positive effects of an elevated CO₂ environment on plant carbon and water balances is not yet known; nor is it known if these effects will scale up through the ecosystem. How O₃ affects the progress of pest epidemics and insect outbreaks as concentrations increase is also unclear (Ball et al., 1998). Information is needed with regard to the impact of O₃ on plant pest and insect reproductive

processes and reproductive development under realistic field or forest conditions, as well as examination of reproductive effects under interacting pollutants (Black et al., 2000).

Scaling. The vast majority of O₃ tree studies have been conducted with young, immature trees and with trees that have not yet formed a closed canopy. Questions remain as to the comparability of O₃ effects on juvenile versus mature trees, and on trees grown in the open versus those in a closed forest canopy, in a competitive environment (Chappelka and Samuelson, 1998; Kolb and Matyssek, 2001; Samuelson and Kelly, 2001). Merging the effects of O₃ across spatial scales is also difficult. Scaling responses of a single or a few plants to effects on communities and ecosystems are complicated matters that will require a combination of manipulative experiments with model ecosystems; community and ecosystem studies along natural O₃ gradients; and extensive modeling efforts to project landscape-level, regional, national and international impacts of O₃. Linking these various studies via impacts on common research quantification across various scales using measures of such factors as leaf area index or spectral reflective data, which could eventually be remotely sensed (Kraft et al., 1996; Panek et al., 2003), would provide powerful new tools for ecologists.

Identifying endpoints. In general, methodologies to determine the important values of services and benefits derived from natural ecosystems are lacking. Identifying and quantifying factors that could be used in comprehensive risk assessment for O₃ effects on natural ecosystems would increase societal awareness of the importance of protecting ecosystems (Heck et al., 1998).

9.8 ECONOMICS

Substantial progress has been made over the past two decades in our understanding of the effects of O₃ and other oxidants on vegetation, particularly for agriculturally important plant species (see Annex Section AX9.7 for a more detailed discussion). The physical and economic effects on agriculture are well documented and provide useful information for consideration in establishing air quality standards for crops (e.g., Spash, 1997).

Since the completion of the National Crop Loss Assessment Network (NCLAN) program in the late 1980s, the number of economic assessments of air pollution studies focusing on terrestrial ecosystems in general, and agriculture in particular, has declined. For example, 33 economic studies of O₃ and other air pollutant effects on U.S. crops were published in peer-reviewed journal outlets from 1980 to 1990 (Spash, 1997). However, in preparing this section of the current O₃ AQCD, only four peer-reviewed economic assessments addressing vegetation in the United States were found for the decade of 1991 to 2000. In addition, one peer-reviewed article (Kuik et al., 2000) was found dealing with agriculture in the Netherlands. Recent interest in global climate change and the potential effects of global warming on O₃ and other photochemical oxidants has renewed interest in the effects of air pollution on both managed and unmanaged terrestrial ecosystems (Adams et al., 1998). In addition, concern is growing regarding the effects of air pollutants on natural ecosystems and on the services they provide (Daily, 1997). Unfortunately, this interest has not yet translated into additional peer-reviewed publications addressing O₃ or other air pollutants' effects on ecosystems.

A study by Murphy et al. (1999) of the economic effects of tropospheric O₃ on U.S. agriculture is of note here, because it confirms the general magnitude of economic effects reported by the two key studies performed a decade earlier (Adams, 1985, 1986). Specifically, Murphy et al. (1999) evaluated benefits to eight major crops associated with several scenarios concerning the reduction or elimination of O₃ precursor emissions from motor vehicles in the United States. Their analysis reported a \$2.8 billion to \$5.8 billion (1990 dollars) benefit from complete elimination of O₃ exposures from all sources, i.e., ambient O₃ reduced to a background level assumed to be 0.025 to 0.027 ppm. While the analytical framework is similar to Adams et al. (1986) in the use of NCLAN-based yield response functions and a mathematical programming-based economic optimization model, the study is novel in its focus on the role of motor vehicle emissions of VOCs and nitrogen oxides (NO_x) in total anthropogenic O₃ levels. The study is also notable in its careful attention to federal farm program effects, particularly the deficiency payment component.

A number of recent studies of air pollutant effects on tree species have appeared in the literature. Some studies have reported changes in total biomass and focused on European species (Kurczynska et al., 1997). Other studies have assessed changes in composition of forest species (biodiversity) or forest health due to exposure to air pollutants (Bringmark and Bringmark, 1995; McLaughlin and Percy, 1999; Vacek et al., 1999). As noted previously, changes in forest biomass and composition are more difficult to value than marketable products. However, measures of forest composition or health have implications for an area of increasing policy concern, that being the effect of air pollutants and other environmental stressors on unmanaged (natural) ecosystems and the services they provide (Goulder and Kennedy, 1997; Pimentel et al., 1997). Considerable discussion has occurred among ecologists and economists as to the appropriate means for valuing these services (Carpenter and Dixon, 1985; Anderson, 1990; Common and Perrings, 1992). A number of conceptual articles have been published on this issue in both economic and ecological journals (Bergstrom, 1990; Suter, II, 1990; Castle, 1993; Pearce, 1993).

Effects on forests and natural ecosystems remain problematic because of limitations in biological response data and economic methods. The problem is even more acute for valuing natural ecosystem service flows. The current limitations surrounding forests and natural ecosystems present a rich research agenda. Areas of greatest potential value in terms of regional policymaking need to be prioritized. Such priority setting can be assisted by sensitivity analyses with existing economic models. By measuring the changes in economic effects arising from changes in key parameters, it is possible to identify those research data gaps most likely to affect economic values.

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10. THE ROLE OF TROPOSPHERIC OZONE IN UVB-RELATED HUMAN HEALTH OUTCOMES AND IN CLIMATE CHANGE

10.1 INTRODUCTION

In addition to directly affecting human health, vegetation and ecosystems, tropospheric ozone (O_3) influences the ground-level flux of solar ultraviolet (UV) radiation and alters the Earth's radiative balance by functioning as a greenhouse gas, therefore contributing to climate change. This chapter addresses these effects.

10.2 THE ROLE OF TROPOSPHERIC OZONE IN DETERMINING GROUND-LEVEL UV-B FLUX

Atmospheric O_3 plays a crucial role in reducing the exposure of living organisms to solar UV radiation. The stratospheric O_3 layer is responsible for nearly all of this shielding effect, as 90% or more of the total atmospheric burden of O_3 is located there. Specific quantification of the role that tropospheric O_3 plays in screening the earth's surface from harmful UV radiation has not been accomplished, to date.

Reasonable estimation of the importance of ground-level O_3 in US urban and suburban areas, i.e., the fraction of tropospheric O_3 subject to regulation by the US NAAQS, in altering human UVB exposure and the incidence of UVB-induced human diseases requires an explicit accounting for key factors that influence (a) ground-level flux in US urban and suburban areas, (b) the probability of UV-B exposure experienced by sensitive populations, and (c) links between exposure and the incidence rates of UVB-induced diseases. This section summarizes, in this order, available information on such factors.

10.2.1 Factors Governing Ultraviolet Radiation Flux at the Earth's Surface

Given its role in protecting life from harmful UV radiation, photochemical processes that alter the concentration of stratospheric O_3 are of particular concern to the global community. Scientific understanding of the significant losses of stratospheric O_3 due to chemistry with the

degradation products of the long-lived anthropogenic chlorinated- and fluorinated-hydrocarbons evolved during the 1970s and early 1980s. This realization led to the international treaty for the protection of stratospheric O₃, i.e., the 1987 Montreal Protocol on Substances that Deplete the Ozone Layer. The scientific community continues to track stratospheric O₃ levels and to document any changes in ground-level UV-B flux due to anthropogenic depletion. An outcome of these efforts is a limited body of literature that describes the effects of tropospheric pollutants, particulate matter (PM), clouds and O₃ on ground-level UV radiation flux.

The Montreal Protocol requires routine review of the latest scientific information available on the status of the O₃ layer and of UV radiation levels at the Earth's surface. The World Meteorology Organization (WMO) and U.N. Environmental Program (UNEP) are responsible for assessing the state of the science regarding the O₃ layer and for reporting on trends in surface UV radiation levels. The latest WMO/UNEP assessment was published in 2002 (WMO, 2002) and includes an evaluation of the role of the troposphere in determining UV flux at the earth's surface.

The mixture of gases, clouds, and particles that comprise the troposphere scatter and/or absorb incident solar radiation (see Figure 10-1). In general, these effects are greater in the troposphere than in the stratosphere, due to the higher atmospheric pressures, particle and cloud densities present there. Solar flux intensity has a temporal dependence, while radiative scattering and absorption have strong wavelength, pathlength, and/or particle concentration dependencies. These combine to create nonlinear effects on UV flux at the earth's surface. Thus, careful quantification of atmospheric absorbers and scatterers, along with a well-resolved description of the physics of these interactions, is necessary for predicting the impact of ground-level O₃ on UV-B flux. The following sections (10.2.1.1–10.2.1.4) summarize the relevant physics that govern solar flux and the nature of radiative interactions in the atmosphere. The sources for this information are a selection of atmospheric physics and chemistry texts, the WMO (2002) report, and other peer-reviewed literature on the role of the troposphere in defining UV surface flux.

10.2.1.1 UV Radiation: Wavelengths and Energies

The energy possessed by a photon is inversely proportional to its wavelength. For example, gamma rays, having wavelengths ~0.1 nm, are especially damaging high-energy

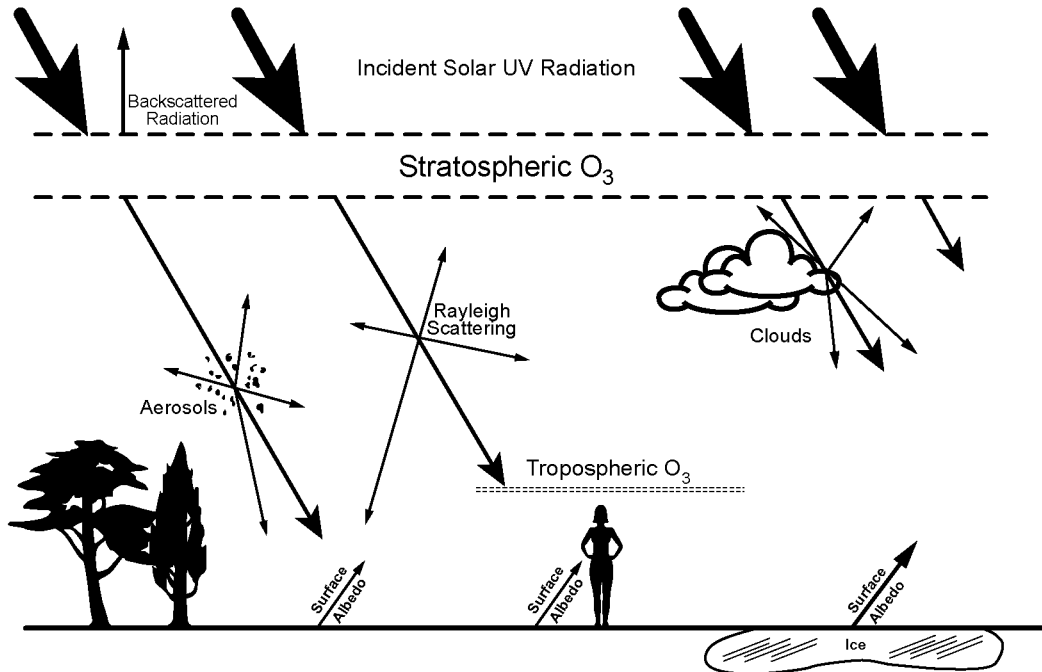


Figure 10-1. Complexity of factors that determine human exposure to UV radiation. In addition to the geophysical/atmospheric factors (e.g., stratospheric and tropospheric O₃, clouds, aerosols, and Rayleigh scattering) that affect the solar flux of UV radiation at surface level, there are human physical, behavioral and demographic factors that influence human exposure to UV radiation.

photons ($>10^6$ kJ einstein⁻¹) emitted during radioactive decay and by stellar activity.

Radiowaves, having wavelengths $\sim 10^8$ nm, are very low in energy ($\sim 10^{-3}$ – 10^{-8} kJ einstein⁻¹) and function as carriers for broadcast communications (Finlayson-Pitts and Pitts, 1986).

The wavelengths ranging between 50 and 400 nm in length are denoted “ultraviolet,” and are subdivided into classes, where UV-C corresponds to wavelengths <280 nm, UV-B refers to the range 280–320 nm, and UV-A to 320–400 nm. UV-C is almost entirely blocked by the Earth’s upper atmosphere, where it participates in photoionization and photodissociation processes. Solar UV-B radiation is absorbed or scattered in part within the atmosphere. UV-A radiation can be scattered but is not absorbed to any meaningful degree by atmospheric gases.

Both UV-B and UV-A photons contain sufficient energy to break (photolyze) chemical bonds in biomolecules and are associated with human health and ecosystem damage. However,

because UV-B is more energetic, it is capable of producing substantially more biological damage than UV-A.

10.2.1.2 Temporal Variations in Solar Flux at the Earth's Surface

The magnitude of the solar radiation flux entering the atmosphere depends upon long-term solar activity, sunspot cycle (11 years), solar rotation (27 days), and the position of the Earth in its orbit around the sun. At any given point on the Earth's surface, solar flux is dependent upon solar zenith angle, a quantity that varies with time of day, season, and latitude.

Solar Rotation and the Sunspot Cycle

A variety of changes in solar irradiance can be found in historical data, from 1700 to the present. Solanki and Fligge (2000) concluded that solar irradiance changes, on time-scales of days to centuries, can be attributed to variations in solar magnetic features. The maximum level of radiation (solar-max) differs from the minimum (solar-min) by as much as 10% for wavelengths near 160 nm. This peak-to-trough difference declines to ~1% for 300 nm (Salby, 1996).

On the decadal time scale, solar rotation and sunspot activity have the largest effects on overall solar irradiance. Since the last Maunder minimum in 1700, and consistent with increasing sunspot activity, solar irradiance has increased by ~3.0% for wavelengths in the UV-C range and by ~1.3% for wavelengths in the UV-B and UV-A ranges. Including visible wavelengths, Solanki and Fligge (2000) estimated that the overall increase in solar irradiance was ~0.3%. Rozema et al. (2001) pointed out that any increase in wavelengths <300 nm (UV-C) would initiate additional O₃ formation in the stratosphere. This suggests that any increase in UV-B and/or UV-A solar flux could, therefore, be offset by a more absorptive stratosphere.

The Position of the Earth with Respect to the Sun

The combined effects of the Earth's obliquity (the angle of the Earth's axis of rotation with respect to the plane of its orbit around the sun) and its precession (the rotation of the Earth's axis with respect to a perpendicular line through the plane of its solar orbit) yield variations of up to 30% in total summertime solar flux at the top of the earth's atmosphere, depending on latitude (Hartmann, 1994).

Zenith Angle: Latitude, Season, and Time of Day

The sun's relative elevation from a point on the earth's surface is measured with respect to the vertical and is known as its *solar zenith angle* (SZA). This angle varies hourly, seasonally, and with latitude. Diurnal and seasonal changes in SZA result in the largest changes in the magnitude of solar radiation flux at the surface, with larger zenith angles corresponding to lower solar flux. The largest natural fluxes occur in the tropical regions, where solar noon occurs at an SZA at or near 0° . Seasonal variation in solar flux ranges from small changes at the equator to very large changes at high latitudes. Diurnal variations in solar flux, from sunrise to sunset, show added wavelength dependence as a function of SZA. This wavelength dependence is a function of the concentrations of radiation scattering and absorbing gases and particles in the atmospheric pathlength traversed by the photon. These processes will be discussed further below.

10.2.1.3 Atmospheric Radiative Interactions with Solar Ultraviolet Radiation

When solar radiation enters the earth's atmosphere, it can be either scattered or absorbed by the gases and particles it encounters. The main mechanisms of atmospheric scattering are Rayleigh, in the case of scattering by gas molecules, and Mie, in the case of scattering by aerosol particles. The intensity of the Rayleigh effect, i.e., the extent to which the gas molecule is capable of perturbing the trajectory of the incident photon, depends on the size of the gas molecule in relation to the photon, and is maximized when the photon is comparable in size or smaller than the gas molecule. Light is scattered symmetrically with respect to the gas molecule, with similar forward and backward intensities. This effect explains the color of the clear sky, as blue photons are comparable in size to the dominant gases in the atmosphere (N_2 and O_2). With Mie scattering, the result of light encountering larger particles, photons are scattered with a strong forward tendency. This tendency explains why dense water clouds, i.e., cumulus clouds, appear brilliantly white.

The lower atmospheric pressures in the stratosphere mean fewer gas molecules are present that can absorb or scatter radiation. Stratospheric clouds and aerosols are also thinner and more dispersed than those in the troposphere. In the language of the radiative transfer literature, these conditions make the stratosphere a "single scattering" regime for UV radiation. The troposphere, due to its high gas and particle concentrations is referred to as a "multiple

scattering” regime. In practical terms, UV radiation traverses the stratosphere with a substantially lower probability of encountering a gas molecule or cloud or an aerosol particle than it would in the troposphere.

The multiple scattering of solar photons in the troposphere accounts for the “disproportionate” role that tropospheric O₃ is said to play in absorbing UV radiation versus stratospheric O₃ on a molecular per molecule basis (Brühl and Crutzen, 1989; Balis et al., 2002; Zerefos et al., 2002). The probability that an individual photon will encounter an O₃ molecule is increased when its effective pathlength through the atmosphere is increased with multiple scattering. This leads to an increase in the probability of absorption by O₃. This effect is well understood in principle, but its importance has not been quantified in relation to stratospheric O₃ or with respect to other tropospheric absorbers.

The importance of atmospheric pressure and particle concentrations is evident in the altitude dependence of solar flux. Solar flux has been observed to increase with altitude above sea level, consistent with an inverse relationship between cloud levels, air pollution concentrations, and intensity of Rayleigh scattering with altitude. A number of measurements of UV radiation have been taken at various altitudes and are reviewed by Xenopoulos and Schindler (2001). Increases in flux as a function of altitude are given as percent irradiance enhancement per 1000 m relative to sea level. The effect can range from 9 to 24% /1000 m, depending upon the altitude at which the measurement was taken (Xenopoulos and Schindler, 2001).

Similarly, this effect is seen as a function of SZA. For example, at very small zenith angles, near solar noon, photons have a shorter atmospheric pathlength to the surface and a smaller probability of encountering gas molecules, clouds, or particles. Conversely, photons that enter the atmosphere at larger zenith angles, i.e., increasingly tangential to the surface, are more likely to be scattered or absorbed.

Radiative Interactions in the Stratosphere: Absorption by Ozone

As noted earlier, the stratosphere contains 90% or more of the total column O₃. Ozone interacts with UV radiation by scattering the photon, or absorbing it. Photoabsorption by O₃ occurs with very high efficiency (Note: for a discussion of the metrics used and relative efficiencies of photoabsorption by gas-phase molecules, the reader is referred to text books such as Finlayson-Pitts and Pitts [1986]). After electronically-excited O₃ (O₃^{*}) is formed, it can

dissociate into ground-state oxygen, O₂, and an electronically excited oxygen radical, O(¹D) (see Reaction 10-1). At low pressures and with irradiation at 300 nm (UV-B range), O(¹D) is produced with a quantum efficiency near unity.



However, in the presence of other atmospheric gases, intermolecular collisions may also disperse the excess electronic energy of the O₃* molecule by transferring it to other molecules as vibrational, rotational, and translational energies, warming the atmosphere, and reducing the production of O(¹D) (see Reaction 10-2). The efficiency of the energy transfer process is a function of atmospheric pressure and the quantum mechanical properties of the colliding molecules. A deeper discussion of molecular energy transfer processes can be found in the text by Levine and Bernstein (1987).

The Dobson Unit (DU) is conventionally used for discussing stratospheric O₃ concentrations. One DU = 2.687 × 10¹⁶ molecules of O₃/cm². Alternatively, a DU corresponds to the column height in hundredths of a millimeter of O₃ at standard temperature and pressure (273 K and 1 atmosphere), integrated along the total height of the atmospheric column (Wayne, 2000).

As previously noted, the total O₃ column effectively prevents any UV-C from reaching the surface and reduces the penetration of UV-B to the surface, but it does little to attenuate the intensity of UV-A except at the shorter wavelengths close to the cutoff for UV-B. Figure 10-2 compares the solar flux above the atmosphere with ground-level flux. Cutchis (1974) calculated that with overhead sun, a 10% decrease in the O₃ column would lead to 20, 250, and 500% increases in flux at 305, 290, and 287 nm, respectively. These estimates have been supported by ground observations in Toronto, ON (49° N; Kerr and McElroy, 1993). Rapid changes of this magnitude appear to happen naturally. As seen in data collected by the Total Ozone Mapping Satellite (TOMS) (Figure 10-3), the total O₃ column undergoes wide natural variation on short timescales (Cockell, 2001).

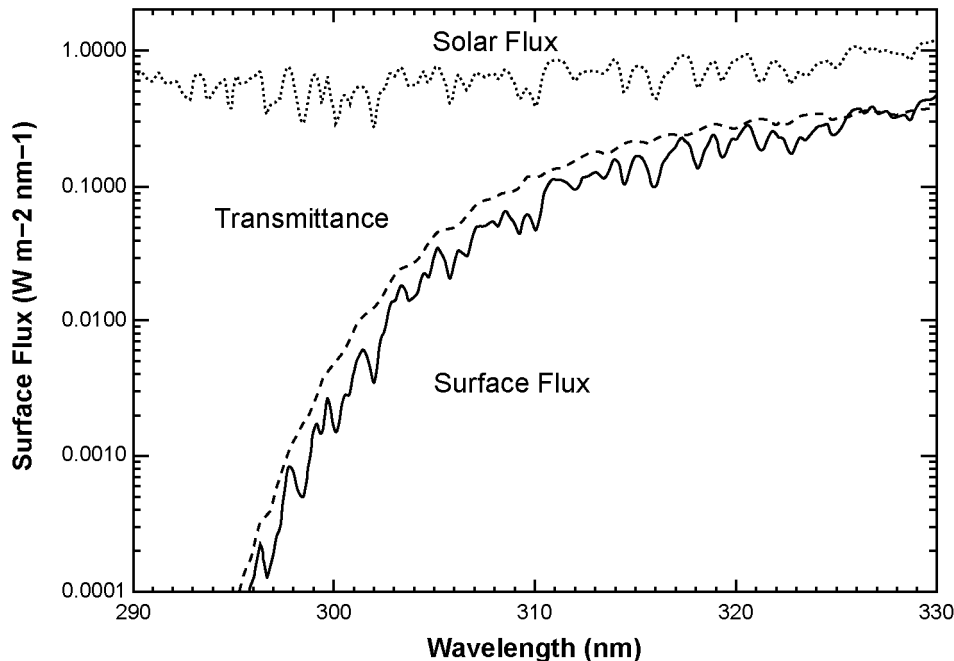


Figure 10-2. Comparison of solar flux above the atmosphere with flux at the Earth’s surface. The dotted line represents extraterrestrial solar flux measured by the satellite UARS SOLSTICE instrument (dotted line). The dashed line represents calculated atmospheric transmittance and the solid line is the calculated absolute flux of UV radiation for a solar zenith angle of 50°, total column O₃ of 275 DU, and a surface reflectivity of 8%. The fine structure on the surface flux trace results from Fraunhofer lines (absorption specific wavelengths within the solar atmosphere).

Source: Krotkov et al. (1998).

The WMO (2002) assessment reported that global average total column O₃ had declined by 3% from pre-1980 levels, due to the presence of anthropogenic O₃-depleting substances in the atmosphere. Ozone depletion has a strong latitude and seasonal dependence. The seasonality of total O₃ changes differs between the Northern and Southern Hemispheres. In the northern midlatitudes, total column O₃ declined by ~4% during the winter/spring seasons and by approximately half that amount in the summer/fall of the 1997–2001 time period, relative to pre-1980 total column O₃ levels. In southern midlatitudes, total column O₃ declined ~6% during all seasons, possibly tied to the enhanced photochemical losses associated with the meteorological dynamics peculiar to the antarctic region.

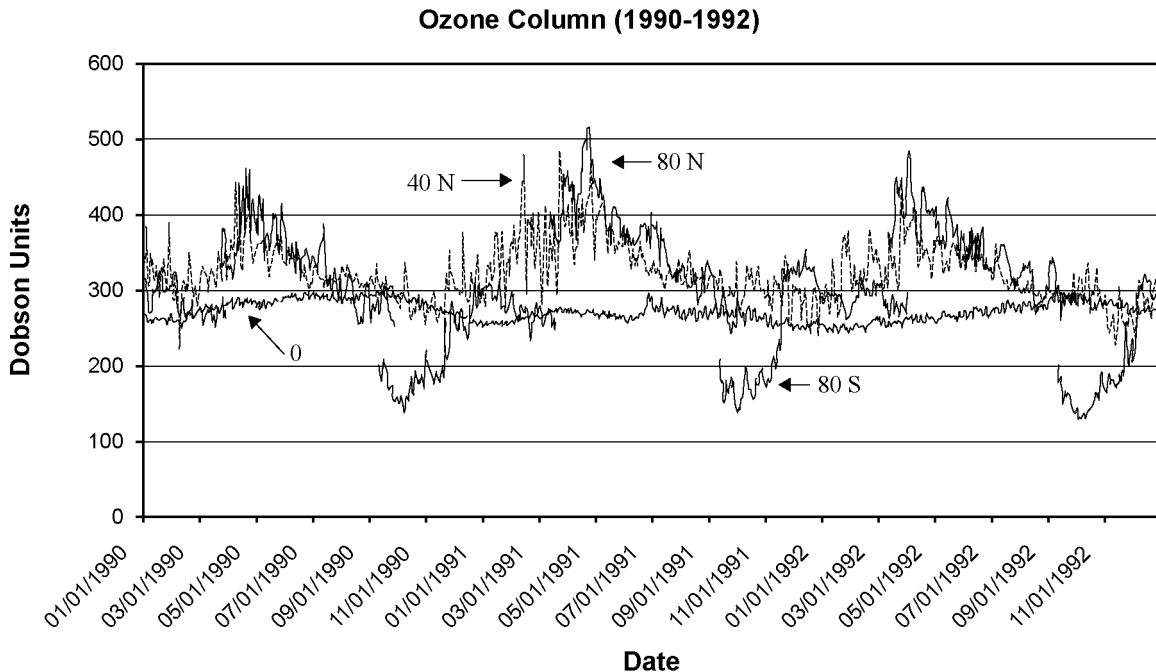


Figure 10-3. Ozone column abundances from the years 1990 to 1992 for 0, 40, and 80° N as well as 80° S. The data for 80° S are incomplete, but the graph shows the effects of the Antarctic O₃ hole on total column abundances at this latitude. The data for the Northern Hemisphere illustrate the natural variations in the O₃ column over time. The data are taken from the TOMS (Total Ozone Monitoring Satellite) data set (1979 to 1993).

Source: Cockell (2001).

Radiative Interactions in the Stratosphere: Scattering by Clouds and Particles

During periods of extreme cold in the stratosphere above the Earth’s poles, nacreous or polar stratospheric clouds (PSCs) form due to the condensation of sulfuric and nitric acids or water vapor. Like tropospheric water clouds, PSCs have the capacity to scatter incident solar radiation. Due to their infrequent appearance, location, and limited optical density, they are not, however, an important factor in determining ground-level UV flux for human health impacts assessment.

Stratospheric aerosols such as those produced by explosive volcanic eruption, i.e., Mt. Pinatubo, have the potential for playing a greater role in altering ground-level UV flux. In the case of Mt. Pinatubo, a large quantity of sulfur dioxide was injected into the stratosphere, which reacted to form a layer of sulfate aerosols. These aerosols increased the Earth’s albedo for

a period of years, reducing solar flux at the Earth's surface. Volcanic eruptions that lead to the injection of aerosols containing light-absorbing-material, such as crustal material (dust), can also reduce surface flux (Herman et al., 1997). However, these events are also too infrequent to be of importance in this discussion.

Radiative Interactions in the Troposphere: Gases, Particles, Clouds, and the Earth's Surface

The components of the troposphere that are of most importance in relation to the prediction of ground-level UV radiation flux are clouds, PM, and UV-absorbing trace gases. The UV-shielding efficiencies of these components are highly dependent on density (concentration), particle size, and the altitude at which they are present in the troposphere. All of these properties are sensitive to meteorology, which introduces an element of temporal dependency. The following section discusses some of the issues specific to each component.

Gases

UV-absorbing gases, including O₃, NO_x, VOCs, and SO₂, are vented into the upper troposphere, injected downwards from the stratosphere (O₃) or form in situ, i.e., O₃, VOC oxidation products (see Chapter 3). Ozone concentrations decrease with increasing altitude from the surface up to, roughly, the mid-troposphere, then increase up into the stratosphere. Figure 10-4 shows a series of O₃ vertical profiles for 4 sites within the continental U.S., i.e., plots of O₃ concentrations as a function of atmospheric pressure (correlating to altitude). The mean values of O₃ in the free troposphere reported in the literature range from ~50 to ~80 ppbv, with higher values occurring at the tropopause. For example, a series of ozonesonde soundings over France from 1976 to 1995 showed an O₃ increase from 48.9 ppbv in the 2.5 to 3.5 km layer to 56.5 ppbv in the 6.5 to 7.5 km layer (Ancellet and Beekmann, 1997).

Within the planetary boundary layer, photochemistry produces a diurnal rise and fall in O₃ and PM concentrations, especially in polluted urban settings. Temperature inversions that often occur in these settings prevent the upward mixing and dilution of ground-level O₃, also trapping primary and secondary PM within the boundary layer. These effects are described in Chapters 2 and 3 of this document. Such conditions substantially increase the UV absorptive capacity of the atmosphere, at the surface.

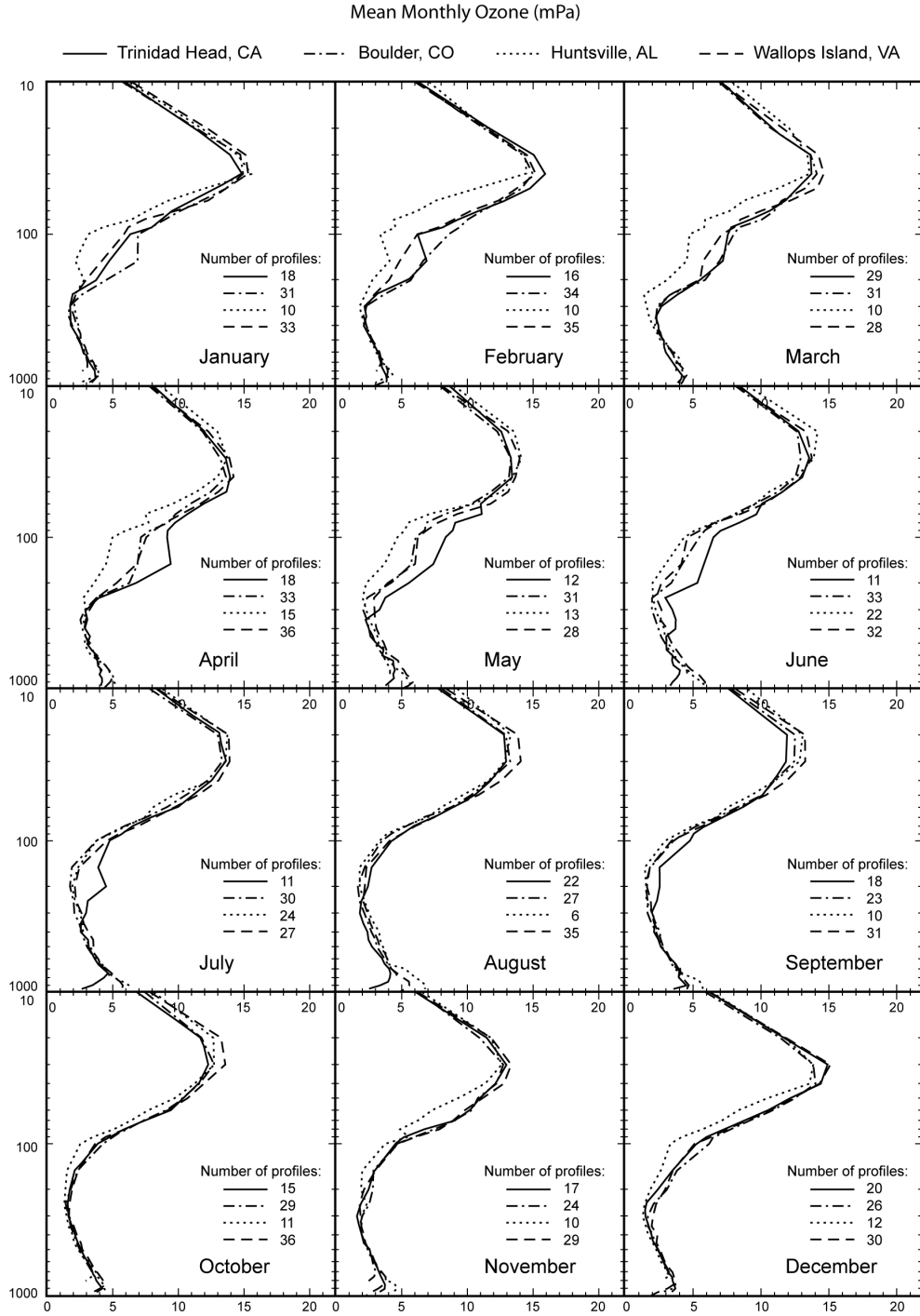


Figure 10-4. Monthly averaged vertical O₃ profiles (partial pressure in mPa) as a function of atmospheric pressure (in mBar) for Trinidad Head, CA (solid line); Boulder, CO (dot-dashed line); Huntsville, AL (dotted line); and Wallops Island, VA (dashed line). The number of launches at each site for each month are indicated on the charts.

Source: Newchurch et al. (2003).

Ultraviolet absorption by gases becomes significant under aerosol- and cloud-free conditions. Figure 10-5 shows a calculation by Krotkov et al. (1998) of the sensitivity, as a function of wavelength, of ground-level UV flux to a 1-DU decrease in total column O₃ under cloud- and aerosol-free conditions. (Note: 1 DU = 2.687×10^{16} molecules of O₃ per cm² or if calculated for the entire troposphere, 1 DU = 10.9 Tg(O₃) with 1 ppb of tropospheric O₃ = 0.65 DU).

A study in Chicago in 1991 and 1992, during which ambient O₃, broadband UV irradiance, and total sunlight were monitored (Frederick et al., 1993), found a significant negative correlation between the UV irradiance and ambient O₃ when the atmosphere was relatively free of clouds and haze. Although Frederick et al. (1993) estimated that a 10 ppbv reduction in O₃ was associated with a 1.3% increase in erythemally-weighted UV-B, they cautioned that this value had a comparatively large uncertainty (~1.2%, or nearly 100% of the estimated increase). Matthijsen et al. (2000) noted that scattering effects differ for UV versus total solar irradiance: UV radiation, even under clear sky conditions is mostly diffuse, whereas total solar is largely direct. This is consistent with Rayleigh scattering of UV wavelengths by N₂ and O₂.

Clouds

Clouds have the largest observed influence on solar irradiance—scattering as much as 100% of incoming direct beam radiation versus the 10–20% scattering that occurs on clear days due to Rayleigh and PM. The long term average for the U.S., based on measurements taken at various locations, show monthly average reductions in UV levels ranging from 10% to 50%, due to cloud cover (Madronich et al., 1995). An overall reduction in normalized global irradiance has been attributed, in part, to the increase in the extent (8%) and optical density of cirrus clouds observed since the middle of the last century (Trepte and Winkler, 2004).

The depth, particle size distribution, and coverage determine, in large part, the amount and wavelengths of radiation that clouds will scatter. In general, cloud effects are weaker for the UV wavelengths than for total solar radiation. Alados-Arboledas et al. (2003) observed that the cloud effect under overcast conditions for total solar radiation is 33% greater than for erythemally-weighted UV radiation.

Geometry is also an especially important factor: scattered or broken clouds can either slightly reduce or enhance irradiance due to scattering between clouds (WMO, 2002).

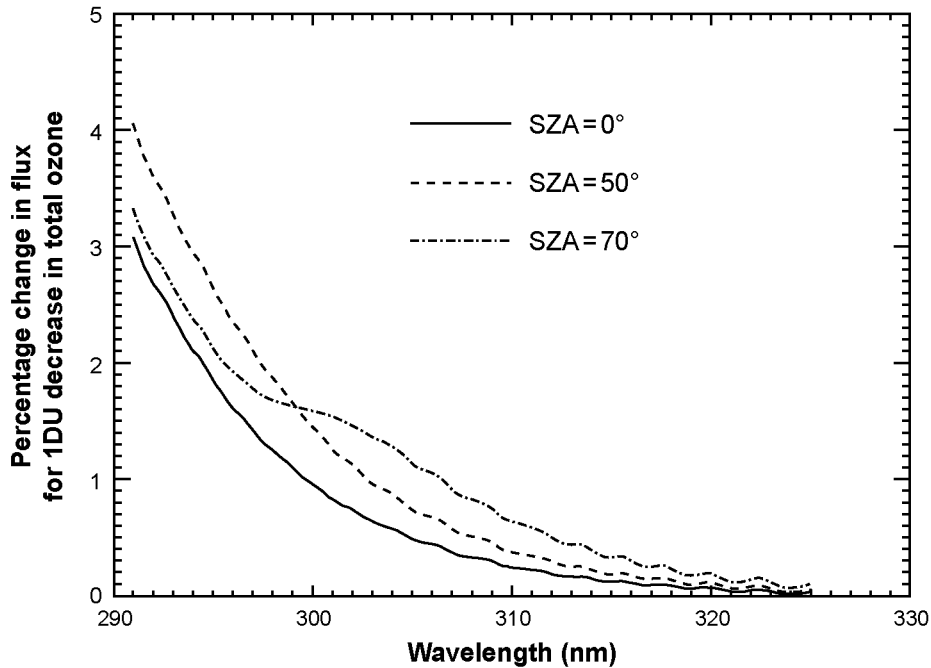


Figure 10-5. The sensitivity of ground-level UV flux to a 1 DU change in total column O₃, under clear sky conditions, as a function of solar zenith angle (SZA).

Source: Krotkov et al. (1998).

Palancar and Toselli (2004) estimated that broken clouds can increase noontime radiation by up to 34% when compared to calculated clear sky values. Mims and Fredrick (1994) reported observations of UV-B radiation enhancements of up to 20% over maximum noontime values due to scattering from cumulus clouds. Overcast conditions have been estimated, for SZAs averaging 50°, to reduce UV-B surface radiation to 29% of clear sky irradiances (Schafer et al., 1996). Schafer et al. (1996) also observed, as a percentage of clear sky values, 61% for 80–90% coverage, 75% for 60–70% coverage, 79% for 40–50% coverage, with frequent incidences of enhancement above clear sky levels due to reflections during partial cloud cover.

The altitude at which clouds are located also substantially alters their effectiveness in scattering radiation back into space. High clouds have been observed to be more effective, by a factor of 5, at scattering 300 nm wavelength radiation (SZA = 0°) than low clouds (Wen and Frederick, 1995). At SZA = 60°, 305 nm wavelength radiation scatters from high clouds

(8–10 km) at twice the efficiency of low clouds (1–2 km). The authors attribute this effect to the reduction in the atmospheric pathlength that UV photons must travel, and therefore a reduction in the odds of scattering due to other particle or gas interactions, before encountering the cloud.

Quantifying the effect of clouds on surface UV flux, therefore, requires detailed information on cloud depth, particle size distribution, geometry, altitude, and the position of the sun relative to the cloud as a function of time.

Particulate Matter

On a zonally averaged basis, PM does not contribute significantly to lower tropospheric absorption of UV radiation. However, in urban areas or other areas subject to high smog levels, such as areas adjacent to significant biomass combustion, PM may be the most important determinant of ground-level erythemal UV flux, excepting clouds (U.S. Environmental Protection Agency, 2004; WMO, 2002). Model-to-measurement comparisons of ground-level flux for Cordoba City (Argentina), the Aegean Sea (Greece) and Toronto (Canada), have shown 20%, 20% and 5–10% reductions, respectively (McKenzie et al., 2003). Liu et al. (1991) estimated similar potential reductions (20%) in UV-B radiation in the presence of large local PM concentrations.

Barnard et al. (2003) measured UV-B penetration to the surface at 3 sites: Rubidoux (California); the University of California, Riverside; and Mt. Gibbs in the North Carolina Blue Ridge Mountains. They found that UV-B is primarily sensitive, under cloud-free conditions, to O₃ values measured by satellite, which do not account for ground level O₃, and concentrations of PM and specifically PM containing black carbon. They found that ground level O₃ concentrations did not meet the 0.05 significance level in their model for urban environments. Barnard et al. (2003) further suggested that increases over the past 20 to 30 years in combustion-associated PM and black carbon may account for the inability to detect a surface trend in UV-B radiation caused by a known decrease in stratospheric O₃ over the Northern Hemisphere.

Surface Albedo

The Earth's surface, by absorbing or reflecting radiation, directly influences UV flux in the lower troposphere. The extent to which the surface reflects incident radiation back into the

atmosphere is known as its “albedo.” Ultraviolet flux is directly proportional to surface albedo (Wendisch and Mayer, 2003).

Surface albedo is a very strongly wavelength-dependent property. For example, fresh and wet snow reflect 60 to 90% of incident violet light, while soil and grass surfaces reflect <5% (Xenopoulos and Schindler, 2001). In their in situ measurement and modeling study of the vertical distribution of solar irradiance, Wendisch and Mayer (2003) found that surface albedos must be specifically measured in order to accurately simulate solar flux, due to the large variations that may occur within a given surface type.

Snow cover, even many kilometers from measurement sites is known to increase detected UV irradiances. Complicated interactions result when radiation is scattered by snow (or other bright surfaces) and backscattered or absorbed by atmospheric particles and clouds in the same vicinity (WMO, 2002).

10.2.1.4 Data Requirements for a Surface UV-B Climatology

A means of establishing the range of variability in UV-B at ground level for a populated area of interest would be to the development of a map of flux levels under typical seasonal conditions based on historical records. In the atmospheric sciences community, a map of this type is referred to as a “climatology.”

The WMO (2002) stated that, in principle, if the spatial distribution of all UV absorbers and scatterers were fully known, the wavelength and angular distribution of the UV irradiance at the Earth’s surface could be determined with model calculations. However, the very limited information available on the distribution of the primary components (i.e., clouds, particles, O₃, other UV-absorbing trace gases and surface albedo) makes detailed predictions impossible. Nearly all of the routine data on tropospheric O₃ concentrations in the U.S. is from ground-level O₃ monitors, such as those used to determine the attainment of the O₃ air quality standards. Such measurements, alone, are not sufficient information for making reliable estimates of ambient O₃ concentrations above the boundary layer—information needed to establish the vertical distribution of O₃ up to the tropopause. In an earlier assessment of the environmental effects of stratospheric O₃ depletion (WMO, 1999), the UNEP concluded that, in view of the high spatial and temporal variability of surface UV radiation and the difficulty in maintaining calibration within networks of UV monitoring instruments, satellite-based observations are necessary to

develop a satisfactory UV climatology. However, satellite-derived estimates of surface UV are limited by the availability of instruments in orbit, with available datasets comprising interpolations based upon a single satellite overpass per day for a given region.

Furthermore, the current limitation in all satellite retrievals is that backscattered radiation, from which flux rates are derived, are limited to atmospheric layers above the surface. Assumptions must be made concerning aerosol and ozone absorption below the atmospheric limit of detection by satellites (McKenzie, et al., 2001). Above densely populated or industrial areas, aerosol extinctions are large and reduce UV appreciably (Gonzalez et al., 2000). Complete assessment of the uncertainties in predictions of UV surface flux would require comparisons between sparse available ground-level observations and satellite data over longer periods of time and for different geographical locations. No such assessment has been reported in the scientific literature.

10.2.2 Factors Governing Human Exposure to Ultraviolet Radiation

An assessment of public health benefits due to the attenuation of UV-B radiation by surface-level O₃ requires appropriate consideration of: (1) the multiple factors that alter the flux of UV-B radiation at ground-level, as described above; (2) the factors that influence the extent of human exposure to UV-B radiation, particularly behavioral decisions; and (3) the effects of UV-B radiation exposure on human health. Consideration must also be given to the public health benefits from exposure to UV-B radiation. The present section outlines the most recent information on the determinants of exposure to UV-B radiation in human populations. Quantitative evaluation of human exposure to UV-B radiation is scientifically necessary to perform health risk assessment and to define subpopulations at risk for UV-B-related health effects.

10.2.2.1 Outdoor Activities

Exposure to solar UV radiation is related to one predominating factor: time spent outdoors during daylight hours. A large U.S. study was conducted using the EPA National Human Activity Pattern Survey (NHAPS) to assess UV radiation dose in Americans (Godar, 2001; Godar et al., 2001, 2003). The EPA NHAPS recorded the activity profiles of 9,386 Americans (age 0 to 60+ years) over a 24-month period to assess their exposure to various environmental

pollutants, including UV radiation. Available UV radiation was assessed using the EPA UV-monitoring program. Solar radiation in the UV-A and UV-B waveband regions were measured daily at a monitoring site in each quadrant of the U.S. There is considerable error associated with quantifying UV radiation dose from exposure surveys and four UV-monitoring sites across the country; however, the qualitative information regarding factors that increase human exposure to UV radiation is still of relevance. The EPA-UV monitoring network has since expanded to 21 sites, located in 14 U.S. national parks and 7 urban areas across the United States (<http://www.epa.gov/uvnet/>). A UV-B monitoring network by the U.S. Department of Agriculture is also available for the quantitative assessment of UV radiation exposure (<http://uvb.nrel.colostate.edu/UVB/>). This monitoring network has 30+ monitoring sites across the United States and three additional sites in Canada and New Zealand.

Godar et al. (2001) observed a strong seasonal preference for outdoor activities, with people spending the most time outdoors during the summer followed by spring, fall, and, lastly, winter. Because the solar erythemal (i.e., skin reddening) UV radiation dose is also highest during the summer, the estimated UV radiation dose of Americans was more than 10-fold greater in the summer compared to the winter season (Godar et al., 2001).

Vacationing at the beach in the summer was associated with higher UV radiation exposures (Godar et al., 2001; Thieden et al., 2001). Even after accounting for sunscreen use at the beach, the erythemal UV radiation doses were more than 40% higher during a 3-week beach vacation compared to a 3-week stay at home (Godar et al., 2001). Danish children and adolescents were found to receive >50% of their annual UV radiation dose while vacationing at European beaches (Thieden et al., 2004a). Sunbathing also was associated with increased annual UV radiation dose in the Canadian National Survey on Sun Exposure and Protective Behaviours (Shoveller et al., 1998). Among the 3,449 adults (age 25+ years) who completed the telephone household survey, 21% stated that they spent time actively sunbathing. In a Danish study with 164 participants, all children (age 1 to 12 years) and teenagers (age 13 to 19 years) as well as 94% of adults (age 20 to 76 years) had days with risk behavior (Thieden et al., 2004b). Teenagers, who had the highest number of risk-behaviors days, were found to have the highest annual UV radiation doses. Among teenagers, 76% (95% CI: 41, 98) of their UV radiation dose during the measurement period was received on risk-behavior days, as determined using personal electronic UV dosimeters and exposure diaries (Thieden et al., 2004b).

An Australian study examining time profiles of daily UV radiation exposure among 8th grade students observed that up to 47% of the daily UV radiation dose fell within the time periods when students were outdoors during school hours, sitting under shaded structures during lunch breaks and participating in routine outdoors or sports activities (Moise et al., 1999). Other studies also have found that participation in outdoor sports (e.g., basketball, soccer, golfing, swimming, cycling) significantly increased UV radiation exposure (Moehrle, 2001; Moehrle et al., 2000; Thieden et al., 2004a,b).

10.2.2.2 Occupation

Of the various factors that affect human exposure to UV radiation, occupation is also important. Approximately 5% of the American workforce work outdoors, as determined by the EPA NHAPS (Godar et al., 2001). On average, American indoor workers spend ~10% of their day outdoors. During their time outdoors, they are exposed to ~30% of the total ground-level UV flux, as measured by the EPA UV-monitoring program (Godar et al., 2001). Compared to indoor or in-home workers, outdoor workers are exposed to much higher levels of UV radiation (Kimlin et al., 1998a; Thieden et al., 2004a), frequently at levels that are above current exposure limits set by the International Commission on Non-Ionizing Radiation Protection (ICNIRP, 2004). For example, Thieden et al. (2004a) observed that the annual UV radiation dose, estimated using personal electronic UV dosimeters and exposure diaries, was ~70% higher for gardeners than indoor workers. The gardeners received the majority (55%) of their UV radiation dose on working days (Thieden et al., 2004a). Another study found that outdoor workers received three to four times the annual UV radiation exposure of indoor workers (Diffey, 1990). At-risk working populations include farmers (Airey et al., 1997; Schenker et al., 2002), fishermen (Rosenthal et al., 1988), landscapers (Rosenthal et al., 1988), building and construction workers (Gies and Wright, 2003), physical education teachers (Vishvakarman et al., 2001), mail delivery personnel (Vishvakarman et al., 2001), and various other workers who spend the majority of their day outdoors during peak UV radiation hours.

10.2.2.3 Age

Age may be a factor that influences human exposure to UV radiation. In a large U.S. study using the EPA NHAPS, the average UV radiation dose among American children (age

<12 years) was estimated to be slightly higher (~20%) than that of adolescents (age 13 to 19 years) (Godar, 2001). A large Canadian survey found that 89% of children (age <12 years) had 30 minutes or more of daily UV exposure compared to 51% for both adults (age 25+ years) and youth (age 15 to 24 years) (Lovato et al., 1998a, 1998b; Shoveller et al., 1998). In an English study (Diffey et al., 1996), UV radiation exposure was estimated in 180 children (age 9 to 10 years) and adolescents (age 14 to 15 years) using personal film badges and exposure records. Once again, children were found to have received higher UV radiation exposure compared to adolescents (Diffey et al., 1996). However, as discussed earlier, a Danish study found that the annual UV radiation dose in teenagers (age 13 to 19 years) was 14-24% higher compared to children (age 1 to 12 years) and adults (Thieden et al., 2004b). This increase in UV radiation dose in the Danish teenagers was attributed to their increased risk-behavior days. Therefore, age may affect human exposure to UV radiation by influencing other factors of exposure, such as outdoor activity and risk behavior.

Two studies examined lifetime UV radiation exposure among persons in the U.S. (Godar et al., 2003) and Denmark (Thieden et al., 2004b). Both studies observed that while there are slight differences in UV radiation dose by age, generally people receive fairly consistent UV doses at different age intervals throughout their lives.

10.2.2.4 Gender

Studies have indicated that females generally spend less time outdoors and, consequently, have lower UV radiation exposure compared to males (Gies et al., 1998; Godar et al., 2001; Shoveller et al., 1998). The U.S. study by Godar et al. (2001) observed that while both males and females had relatively consistent erythemal UV radiation doses throughout their lives, males consistently received higher overall UV doses compared to females at all age groups. Among all Americans, the lowest exposure to UV radiation was received in females during their child-raising years (age 22 to 40 years) (Godar et al., 2001). The highest exposure was observed in males aged 41 to 59 years in the U.S. study (Godar et al., 2001). A similar Canadian survey found that younger adult males had the greatest exposures to UV radiation (Shoveller et al., 1998).

10.2.2.5 Geography

In the U.S. study by Godar et al. (2001), erythematous UV radiation doses were examined in persons living in northern and southern regions. Northerners and southerners were found to spend an equal amount of time outdoors; however, the higher solar flux at lower latitudes significantly increased the annual UV radiation dose for southerners (Godar et al., 2001). The annual UV radiation doses in southerners were 25 and 40% higher in females and males, respectively, compared to northerners (Godar et al., 2001). Other studies also have shown that altitude and latitude influence personal exposure to UV radiation (Kimlin et al., 1998b; Rigel et al., 1999).

10.2.2.6 Protective Behavior

Protective behaviors such as using sunscreen (e.g., Nole and Johnson, 2004), wearing protective clothing (e.g., Rosenthal et al., 1988; Sarkar, 2004; Wong et al., 1996), and spending time in shaded areas (Moise et al., 1999; Parisi et al., 1999) have been shown to reduce exposure to UV radiation. In one study, the use of sunscreen was associated with extended intentional UV radiation exposure (Autier et al., 1999); however, a follow-up study indicated that sunscreen use increased duration of exposures to doses of UV radiation that were below the threshold level for erythema (Autier et al., 2000).

In a national study of U.S. youths aged 11 to 18 years, the most prevalent protective behavior was sunscreen use (39.2%) followed by use of a baseball hat (4.5%) (Davis et al., 2002). There were significant differences in the use of sunscreen by age group and gender, with the younger age group (age 11 to 13 years) and girls having greater likelihood (47.4 and 48.4%, respectively) of using sunscreen (Davis et al., 2002). The Canadian National Survey on Sun Exposure and Protective Behaviours observed that less than half of the adults (age 25+ years, n = 3,449) surveyed took adequate protective actions (Shoveller et al., 1998). Once again, children (age <12 years, n = 1,051) were most protected from exposure to UV radiation, with 76% using sunscreen and 36% avoiding the sun, as reported by their parents (Lovato et al., 1998a). However, the protection level was still not adequate, as indicated by the high 45% rate of erythema in children. Among Canadian youth (age 15 to 24 years, n = 574), protective actions from UV radiation exposure included wearing a hat (38%) and seeking shade and avoiding the sun between the peak hours of 11:00 a.m. to 4:00 p.m. (26%) (Lovato et al., 1998b).

The lowest prevalence of protective behavior among the youth was likely responsible for the highest proportion of erythema (68%) experienced in this age group. A Danish study observed that both children and teenagers applied sunscreen on more days than adults, but teenagers had the most days with erythema, due to their increased risk behavior (Thieden et al., 2004b). A survey in Switzerland of 1,285 individuals, including children and parents, indicated that sunscreen use was the protective action most commonly used, but only at the beach and not in routine daily exposure (Berret et al., 2002). In general, protective clothing and avoiding the sun were not highly used among these individuals to protect against UV-related health effects.

10.2.2.7 Summary of Factors that Affect Human Exposures to Ultraviolet Radiation

The factors that potentially influence UV radiation doses were discussed in the previous sections and include outdoor activities, occupation, age, gender, geography, and protective behavior. Results from the various studies indicate that the following subpopulations may be at risk for higher exposures to UV radiation:

- Individuals who engage in high-risk behavior, viz., sunbathing;
- Individuals who participate in outdoor sports and activities;
- Individuals who work outdoors with inadequate shade, e.g., farmers, fishermen, gardeners, landscapers, building and construction workers; and
- Individuals living in geographic areas with higher solar flux (i.e., lower latitudes [e.g., Honolulu, HI] and higher altitudes [e.g., Denver, CO]).

10.2.3 Factors Governing Human Health Effects due to Ultraviolet Radiation

Ultraviolet radiation occupies a specific region of the electromagnetic spectrum of wavelengths and can be further subdivided into three parts, UV-A (320 to 400 nm), UV-B (280 to 320 nm), and UV-C (200 to 280 nm). Most of the health risks associated with UV radiation exposure are wavelength dependent. Wavelengths <180 nm are of little practical biological significance as they are almost completely absorbed by the stratosphere (ICNIRP, 2004).

“Action spectra” of a given biological response to UV radiation across its spectral range are used to estimate exposure by weighting individual wavelength intensities according to the

associated response. The overall effectiveness of the incident flux at inducing the biological response of interest is computed by means of the relationship:

$$\text{effective irradiance} = \int_{\lambda} I_{\lambda} E_{\lambda} d\lambda \quad (10-3)$$

where I_{λ} and E_{λ} are, respectively, the irradiance and its relative effectiveness at wavelength λ .

Until 1980, it was generally thought that wavelengths <315 nm were responsible for the most significant adverse UV radiation health effects; however, recent studies have found that the longer wavelengths in the UV-A range also may produce adverse responses at substantially higher doses (ICNIRP, 2004). As UV-A radiation is not absorbed by O₃, health effects solely induced by UV-A exposure are not relevant in an analysis of public health risks/benefits associated with O₃-related UV attenuation. Therefore, this section focuses on the latest available information on the various adverse health effects associated with acute and chronic UV-B radiation exposure.

10.2.3.1 Erythema

Association Between Ultraviolet Radiation Exposure and Erythema

The most conspicuous and well-recognized acute response to UV radiation is erythema, or the reddening of the skin, which is likely caused by direct damage to DNA by UV-B and UV-A radiation (Matsumura and Ananthaswamy, 2004). Indirect oxidative damage also may occur at longer wavelengths (Matsumura and Ananthaswamy, 2004). Skin type appears to play a large role in the sensitivity to UV radiation-induced erythema. The Fitzpatrick classifications for skin types are: (1) skin type I – individuals with extremely sensitive skin that sunburns easily and severely, and is not likely to tan (e.g., very fair skin, blue eyes, freckles); (2) skin type II – individuals with very sensitive skin that usually sunburns easily and severely, and tans minimally (e.g., fair skin, red or blond hair, blue, hazel or brown eyes); (3) skin type III – individuals with sensitive skin that sunburns moderately and tans slowly (e.g., white skin, dark hair); (4) skin type IV – individuals with moderately sensitive skin that sunburns minimally and usually tans well (e.g., white or light brown skin, dark hair, dark eyes); (5) skin type V – individuals with minimally sensitive skin that rarely sunburns and tans deeply (e.g., brown skin); and (6) skin type VI – individuals with nonsensitive skin that never sunburns and tans

profusely (e.g., dark skin). Harrison and Young (2002) found that the perceptible minimal erythema dose was approximately twofold greater for individuals with skin type IV compared to skin type I, although there was considerable overlap in the minimal erythema dose among the four skin types. Waterston et al. (2004) further observed that within an individual, erythema response differed by body site (e.g., abdomen, chest, front upper arm, back of thigh). These differences were likely attributable to body site-specific variations in melanin pigmentation.

Kollias et al. (2001) investigated the change in erythema response following a previous exposure to UV radiation. Body sites that received a second exposure to UV radiation always showed a reduced erythema response compared to body sites with a single exposure, especially when the first exposure was at levels greater than the minimal erythema dose. The suppression of erythema was more pronounced when the second exposure was given 48 hours after the first. These findings support the well established notion that repeated exposures to UV radiation results in adaptation (e.g., stimulation of melanogenesis). Kaidbey and Kligman (1981) examined individuals with skin types I, II, and III, and found that multiple exposures to subthreshold doses of UV radiation at 24-hour intervals resulted in cumulative injury to the skin, as indicated by a lowering of the minimal erythema dose. These results suggest that a longer time period than 24 hours may be necessary to repair damage from a single exposure to UV radiation. Henriksen et al. (2004) also observed a lowering of the minimal erythema dose with repeated exposure at 24-hour intervals in 49 healthy volunteers with skin types II, III, IV, and V. However, adaptation was reached after the 4th consecutive exposure. Henriksen et al. further found that the change in threshold depended on skin type. After 4 days of repeated UV radiation, there was little change (10 to 20%) in the erythema threshold dose with repeated exposure to UV radiation in the fair-skinned individuals. Among the darker-skinned individuals, the minimal erythema dose was lowered by 40 to 50%. However, both the initial UV dose and the dose to erythema after four days of exposure was still higher in the dark-skinned persons.

A reference erythema action spectrum was adopted by the Commission Internationale de l'Eclairage (International Commission on Illumination, CIE) in 1987 (McKinlay and Diffey, 1987). The CIE erythema action spectrum indicates that UV-B radiation is orders of magnitude more effective per unit dose than UV-A radiation.

Risk of Erythema from Changes in Tropospheric O₃ Levels

There is no literature examining the risk of erythema associated with changes specifically in tropospheric or ground-level O₃ levels. The scientific studies, available to date, focus on the effects of a reduction in stratospheric ozone. One such study has assessed the effects of stratospheric O₃ depletion on the risk of erythema (Longstreth et al., 1998). The analysis by Longstreth et al. (1998) concluded that the risk of erythema would not appreciably increase with depletion of the stratospheric O₃ layer. This is due to the powerful adaptation of the skin to different levels of UV radiation, as evidenced by its ability to cope with changes in UV radiation by season (Van Der Leun and De Gruijl, 1993). Gradual exposure to increasing UV radiation from the winter to summer leads to decreased sensitivity of the skin. In midlatitudes, the UV-B radiation in the summer is 10-fold greater than in the winter. In contrast, the steady depletion of the O₃ layer has been estimated to result in an approximately 20% increase in UV-B over 10 years (Longstreth et al., 1998). The comparatively small increase in UV radiation throughout the years, therefore, would not significantly increase the risk of erythema. Tropospheric O₃ constitutes no more than 10% of total atmospheric O₃. Given that stratospheric O₃ depletion was unlikely to increase the risk of erythema, one could reasonably conclude that small changes in ground-level O₃ that take place with attainment of the O₃ NAAQS would also not result in increased risk.

10.2.3.2 Skin Cancer

According to the American Academy of Dermatology, one in five Americans develop skin cancer during their lifetime. The three main forms of skin cancer include basal cell carcinoma and squamous cell carcinoma, which are both nonmelanoma skin cancers, and malignant melanoma. Nonmelanoma skin cancers constitute more than one-third of all cancers in the U.S. and ~90% of all skin cancers, with basal cell carcinoma being approximately four times as common as squamous cell carcinoma (Diepgen and Mahler, 2002; ICNIRP, 2004). The incidence of malignant melanoma is much lower than nonmelanoma skin cancers. In 2004, more than one million cases of basal and squamous cell skin cancer are expected to be newly diagnosed, compared to 40,780 cases of melanoma (Jemal et al., 2004). However, melanoma has great metastatic potential and accounts for the majority of skin cancer deaths.

Exposure to UV radiation is considered to be a major risk factor for all three forms of skin cancer (Gloster and Brodland, 1996; Diepgen and Mahler, 2002; IARC, 1992). Ultraviolet radiation is especially effective in inducing genetic mutations and acts as both a tumor initiator and promoter. Keratinocytes have evolved DNA repair mechanisms to correct the damage induced by UV; however, mutations can occur, leading to skin cancers that are appearing with increasing frequency (Hildesheim and Fornace, 2004). The relationship between skin cancer and chronic exposure to UV radiation is further explored below, followed by discussion of the influence of O₃ on the incidence of skin cancer.

10.2.3.3 Ultraviolet Radiation Exposure and the Incidence of Nonmelanoma Skin Cancers

The incidence of all three types of cancers has been shown to rise with increasing UV radiation concentrations across the U.S. (De Gruijl, 1999); however, the most convincing evidence for a causal relationship exists between UV radiation and squamous cell carcinoma. Squamous cell carcinoma occurs almost exclusively on skin that is regularly exposed to the sun, such as on the face, neck, arms, and hands. The incidence is higher among whites in areas of lower latitudes, where solar flux is greater (Krickler et al., 1994). The risk of squamous cell carcinoma was shown to increase with life-long accumulated exposure to UV radiation in one cross-sectional study (Vitasa et al., 1990); however, increased risk was found to be associated only with exposure 10 years prior to diagnosis in a case-control study (Gallagher et al., 1995a). One of the major concerns with both types of studies is the potential for recall bias in reporting past UV radiation exposure by individuals already aware of their disease status.

Ultraviolet radiation also has been linked to basal cell carcinoma. Basal cell carcinoma is common on the face and neck (80-90%) but rarely occurs on the back of the hands (De Gruijl, 1999). While cumulative UV radiation exposure was not associated with an increased risk of basal cell carcinoma (Vitasa et al., 1990), increased risk was observed in individuals with greater recreational UV radiation exposure in adolescence and childhood (age <19 years) and individuals with a history of severe erythema in childhood (Gallagher et al., 1995b). Once again, consideration must be given to potential recall bias in assessing these results. Thus, there is suggestive evidence that UV radiation also plays a role in the development of basal cell carcinoma, but the etiologic mechanisms for squamous cell carcinoma and basal cell carcinoma likely differ. In an Australian study conducted in a subtropical community, the factors of having

fair skin, a history of repeated sunburns, and nonmalignant solar skin damage diagnosed by dermatologists were strongly associated with both types of nonmelanoma skin cancer (Green et al., 1996). The authors attributed the finding that outdoor occupation was not associated with nonmelanoma skin cancer to self-selection. Individuals with fair or medium complexions and a tendency to sunburn accounted for more than 80% of the community study sample; however, they were systematically underrepresented among outdoor workers (Green et al., 1996). Such self-selection bias might partly explain the lack of consistent quantitative evidence of a causal link between UV radiation and skin cancer in humans.

De Gruijl et al. (1993) assessed the action spectrum for nonmelanoma skin cancers using hairless albino mice. Human data are not available regarding wavelength dependence of the carcinogenicity of UV radiation. After adjusting for species differences, the Skin Cancer Utrecht-Philadelphia action spectrum indicated the highest effectiveness in the UV-B range with a maximum at 293 nm, which dropped to 10^{-4} of this maximum at the UV-A range above 340 nm (De Gruijl et al., 1993). The mutations commonly present in the *p53* tumor suppressor gene in individuals with squamous cell carcinoma and basal cell carcinoma are called the “signature” mutations of UV-B radiation (De Gruijl, 2002). UV-B radiation is highly mutagenic, because DNA is a chromophore for UV-B, but not for UV-A radiation (Ichihashi et al., 2003). Nevertheless, other studies have found that UV-A radiation, in addition to UV-B radiation, can induce DNA damage (Persson et al, 2002; Runger et al., 2000). DNA damage by UV-A is mediated by reactive oxygen species, making it indistinguishable from damage caused by other agents that generate reactive oxygen species (De Gruijl, 2002). Epidemiologic evidence of a carcinogenic effect of UV-A was found in a study of psoriasis patients receiving oral psoralen and UV-A radiation treatment (Stern et al., 1998). High-dose exposure to oral psoralen and UV-A radiation was associated with a persistent, dose-related increase in the risk of squamous cell cancer. Risk of basal cell cancer also was increased in those patients exposed to very high levels of UV-A radiation. Therefore, although UV-B radiation has long been considered the main culprit for nonmelanoma skin cancer, limited evidence suggest that UV-A radiation may also play a role.

Susceptible populations for nonmelanoma skin cancers include individuals with reduced capacity for nucleotide excision repair, the primary repair mechanism for UV radiation-induced DNA lesions (Ichihashi et al., 2003). At particular risk are individuals with xeroderma

pigmentosum, as they have defective nucleotide excision repair in all tissues (Kraemer, 1997; Sarasin, 1999). Skin type also largely affects susceptibility to skin cancer. Of the six skin phenotypes, the most sensitive individuals are those with skin types I and II, who have a fair complexion, blue or green eyes, and red or blond hair (Diepgen and Mahler, 2002). These individuals tend to sunburn easily, tan poorly, and freckle with sun exposure. A history of repeated sunburns also appears to increase the risk of both cancers, while sunburns during childhood are more associated with increased basal cell carcinoma (Gallagher et al., 1995b; Green et al., 1996).

Ultraviolet Radiation and the Incidence of Cutaneous Malignant Melanoma

From 1973 to 1994, the incidence rate of melanoma increased 120.5% along with an increased mortality rate of 38.9% among whites in the United States (Hall et al., 1999). The ICNIRP (2004) states that during the past 40 years or so, each decade has seen a twofold increase in the incidence of malignant melanoma in white populations, with increased incidence observed more prominently in individuals living in lower latitudes. Cutaneous malignant melanoma has a multifactorial etiology with environmental, genetic, and host factors (Lens and Dawes, 2004). The major environmental factor of malignant melanoma has been identified as UV radiation exposure (Diepgen and Mahler, 2002); therefore, the increased incidence of melanoma throughout the years might be partially attributable to changes in human activity patterns (e.g., increased outdoor activity) that influence UV exposure or increased UV radiation at the ground level. The risk of melanoma appears to depend on the interaction between the nature of the exposure and skin type (Lens and Dawes, 2004).

Fears et al. (2002) examined the association between invasive cutaneous melanoma and UV radiation in non-Hispanic whites using a case-control study design. Lifetime residential history was coupled with mid-range UV-B radiation flux measurements to reduce exposure misclassification and recall bias. A 10% increase in the average annual UV-B flux was significantly associated with a 19% (95% CI: 5, 35) increase in individual odds for melanoma in men and a 16% (95% CI: 2, 32) increase in women. Whiteman et al. (2001) conducted a systematic review of studies that examined the association between childhood UV radiation exposure and risk of melanoma. Researchers found that ecological studies assessing ambient sun exposure consistently reported higher risks of melanoma among people who resided in an

environment with high UV radiation during their childhood (Whiteman et al., 2001). The lack of consistency among the case-control studies was likely due to the varying methods used to assess UV radiation dose.

While the evidence is generally suggestive of a causal relationship between UV radiation and malignant melanoma, possibly conflicting data also has been observed. For example, the highest occurrence of malignant melanoma is on men's backs and women's legs, areas that do not have prolonged exposure to the sun (Rivers, 2004). This indicates that, unlike nonmelanoma skin cancers, malignant melanoma tends to occur in sites of intermittent, intense sun exposure (trunk and legs), rather than in areas of cumulative sun damage (head, neck, and arms) (Swetter, 2003). A study by Whiteman et al. (2003) observed that individuals with melanomas of the trunk had more melanocyte nevi and less solar keratoses compared to individuals with head and neck melanomas, suggesting that cutaneous melanomas may arise through two pathways, one associated with melanocyte proliferation and the other with chronic exposure to sunlight. Green et al. (1999) also found that melanomas of the soles and palms resembled other cutaneous melanomas in their association with sun exposure, but were distinguished from them by their strong positive associations with nevi on the soles.

The available data conflict with regard to the relative importance of UV-A versus UV-B in inducing melanomas. UV-A has a much higher flux rate at the Earth's surface, as it is not absorbed by O₃; and it is able to penetrate more deeply into the skin surface due to its longer wavelength. However, UV-B, as mentioned earlier, is much more energetic and, therefore, more effective in photochemically altering DNA. The individual roles of UV-A and UV-B in the development of cutaneous malignant melanoma have been examined in several studies. A case-control study of 571 patients and 913 matched controls found an elevated odds ratio of 1.8 (95% CI: 1.2, 2.7), after adjusting for skin type, hair color, raised nevi, and number of sunburns, for developing malignant melanoma in individuals who regularly used tanning beds, which typically are UV-A sources (Westerdahl et al., 2000). In a study by Setlow et al. (1993), an action spectrum using the tropical fish *Xiphophorus* indicated that UV-A range wavelengths were especially important in malignant melanoma induction. However, an action spectrum using the opossum *Monodelphis domestica* found that the potency of UV-A for melanoma induction was extremely low compared to that of UV-B (Robinson et al., 2000). A recent study by De Fabo et al. (2004) examined the differences in wavelength effectiveness using a

hepatocyte growth factor/scatter factor-transgenic mouse model. The epidermal tissue of these transgenic mice behaves similar to the human epidermis in response to UV exposure. Given the absence of a mammalian melanoma action spectrum, the standardized CIE erythema action spectrum was used to deliver identical erythemally effective doses. Only UV-B radiation was found to initiate mammalian cutaneous malignant melanoma. UV-A radiation, even at doses considered physiologically relevant, were ineffective at inducing melanoma (De Fabo et al., 2004). Overall, current evidence suggests that UV-B, and not UV-A, is the primary risk factor for malignant melanoma (ICNIRP, 2004).

The populations susceptible for malignant melanoma are similar to those for nonmelanoma skin cancers. Once again, individuals with xeroderma pigmentosum or a reduced capacity for nucleotide excision repair are at increased risk (Tomescu et al., 2001; Wei et al., 2003). Individuals with skin types I and II, or the fair-skin phenotype (blue or green eyes; blond or red hair; skin that freckles, sunburns easily, and does not tan), have increased susceptibility to malignant melanoma (Evans et al., 1988; Swetter, 2003; Veierød et al., 2003). However, the incidence of melanoma was also positively associated with UV radiation in Hispanics and blacks (Hu et al., 2004). Although the incidence of melanoma is much lower in Hispanics and blacks compared to whites, melanomas in these populations are more likely to metastasize and have a poorer prognosis (Black et al., 1987; Bellows et al., 2001). Among children, malignant melanoma appears to have similar epidemiologic characteristics to the adult form of the disease (Whiteman et al., 1997). Individuals with intermittent, intense sun exposure, particularly during childhood, were found to have increased risk of melanoma (Whiteman et al., 2001), in contrast to the association between cumulative exposure and increased risk of squamous cell carcinoma. One study found that a personal history of nonmelanoma skin cancer or precancer, higher socioeconomic status, and increased numbers of nevocytic nevi also were associated with increased incidence of melanoma (Evans et al., 1988).

Effect of Changes in Tropospheric O₃ Levels on Skin Cancer Incidence

The current evidence strongly suggests a causal link between exposure to UV radiation and the incidence of both nonmelanoma and melanoma skin cancer. Genetic factors, including skin phenotype and ability to repair DNA, affect an individual's susceptibility to skin cancer. Quantifying the relationship between UV radiation and skin cancer is complicated by the

uncertainties involved in the selection of an action spectrum and appropriate characterization of dose (e.g., peak or cumulative levels of exposure, childhood or lifetime exposures). In addition, there are multiple complexities in attempting to quantify the effect of tropospheric O₃ levels on UV-radiation exposure, as described in Section 10.2. The absence of published studies that critically examine increased incidence of skin cancer attributable to decreased tropospheric O₃ levels reflects the significant challenges in determining ground-level O₃-related changes in UV radiation exposure. An analysis by Lutter and Wolz (1997) attempted to estimate the effects of a nationwide 10 ppb reduction in seasonal average tropospheric O₃ on the incidence of nonmelanoma and melanoma skin cancers and cataracts. Their estimate, however, depended upon several simplifying assumptions, ranging from an assumed generalized 10 ppb reduction in O₃ column density, national annual average incidence rates for the two types of skin cancer, and simple, linear biological amplification factors. Further, the methodologies used in this analysis inherently have ignored area-specific factors that are important in estimating the extent to which small, variable changes in ground-level O₃ mediate long-term exposures to UV-B radiation. More reasonable estimates of the human health impacts of enhanced UV-B penetration following reduced surface O₃ concentrations require both (a) solid understanding of the multiple factors that define the extent of human exposure to UV-B at present and (b) well-defined and quantifiable links between human disease and UV-B exposure. The reader is referred to the U.S. EPA 2002 Final Response to Court Remand (Federal Register, 2003) for detailed discussions of the data and scientific issues associated with the determination of public health benefits resulting from the attenuation of UV-B by surface-level O₃.

In the absence of studies specifically addressing the reduction of tropospheric O₃ (by assuming that the key variable is total column O₃ density), inferences could be made concerning the effects of reduced tropospheric O₃-related increases in UV-B exposure on the basis of studies focused on stratospheric O₃ depletion. Several studies have examined the potential effect of stratospheric O₃ depletion on the incidence of skin cancer (De Gruijl, 1995; Longstreth et al., 1995; Madronich and De Gruijl, 1993; Slaper et al., 1996; Urbach, 1997). Note that several of the concerns expressed in relation to the Lutter and Wolz (1997) analysis are relevant here as well. Stratospheric O₃ depletion is likely to increase the ground-level UV-B flux, as O₃ absorbs radiation in that wavelength range with high efficiency. Because UV-B radiation is primarily implicated in the induction of skin cancer, especially among persons with skin phenotypes I and

II, there is concern that the depletion of the O₃ layer would result in significantly increased incidence of skin cancers.

Estimation of the increased risk in melanoma associated with stratospheric O₃ depletion cannot be done adequately due to the lack of a mammalian action spectrum for melanoma. Furthermore, the complexity of the UV-related induction mechanism of melanoma adds an additional layer of uncertainty to the calculations. The excess risk in nonmelanoma skin cancers associated with a decrease in stratospheric O₃ was estimated using the Skin Cancer Utrecht-Philadelphia action spectrum based on hairless albino mice (Longstreth et al., 1995).

Quantification of how much more UV radiation would reach ground level with each percentage decrease in O₃ required several assumptions: (1) annual doses were an appropriate measure; (2) personal doses were proportional to ambient doses; and, most notably, (3) each percentage decrease in O₃ was associated with a 1.2% increase in UV radiation. Next, the relationship between UV radiation and nonmelanoma skin cancer incidence was determined: each percent increase in annual UV radiation dose was estimated to cause a 2.5% increase in squamous cell carcinoma and 1.4% increase in basal cell carcinoma over a human lifetime. Incorporating all these factors, Longstreth et al. (1995) calculated that a sustained 10% decrease in stratospheric O₃ concentration would result in 250,000 additional nonmelanoma skin cancer cases per year. Madronich and De Gruijl (1993) noted that the largest percent of O₃-induced nonmelanoma skin cancer increases would be at high latitudes, where baseline incidence of skin cancer is usually small. Assuming a phaseout of primary O₃-depleting substances by 1996, as established by the Copenhagen Amendments in 1992, Slaper et al. (1996) estimated that the number of excess nonmelanoma skin cancers in the U.S. caused by O₃ depletion would exceed 33,000 per year (or approximately 7 per 100,000) around the year 2050.

However, estimating the increase in nonmelanoma skin cancer incidence attributable to the depletion of the stratospheric O₃ layer is marred by uncertainty. The following statement by Madronich and De Gruijl (1994) describes the uncertainty of estimating the effect of stratospheric O₃ depletion on the incidence of skin cancer:

Extrapolating trends and effects of UV into the future is very hypothetical due to uncertainties that arise from atmospheric chemistry, epidemiology, and related disciplines. The values that we calculated are one plausible measure of the magnitude of the O₃-UV effects... The timescales for atmospheric change and skin-cancer development are still far from certain: O₃ reductions are expected to continue well into next century, and the time between UV exposure and development of skin cancer is essentially unknown.

Therefore, much caution is necessary when assessing and interpreting the quantitative results of excess nonmelanoma skin cancer incidence due to stratospheric O₃ depletion. Although the effect of reductions in tropospheric or ground-level O₃ concentrations on skin cancer incidence has not been assessed, it would be expected to be much less compared to the effect from the depletion of the stratospheric O₃ layer, given that tropospheric O₃ makes up $\leq 10\%$ of the total atmospheric O₃.

10.2.3.4 Ocular Effects of Ultraviolet Radiation Exposure

Ultraviolet Radiation Exposure and Risk of Ocular Damage

Ocular damage from UV radiation exposures includes effects on the cornea, lens, iris, and associated epithelial and conjunctival tissues. Absorption of UV radiation differs by wavelength, with short wavelengths (<300 nm) being almost completely absorbed by the cornea, whereas longer wavelengths are transmitted through the cornea and absorbed by the lens (McCarty and Taylor, 2002). The most common acute ocular effect of environmental UV exposure is photokeratitis, also known as snowblindness, caused by absorption of short wavelength UV radiation by the cornea. The action spectrum indicated that maximum sensitivity of the human eye was found to occur at 270 nm (ICNIRP, 2004; Pitts, 1993). The threshold for photokeratitis in humans varied from 4 to 14 mJ/cm² for wavelengths 220 to 310 nm.

Exposure to longer wavelengths has been shown to cause both transient and permanent opacities of the lens, or cataracts. Extensive toxicologic and epidemiologic evidence supports the causal association between UV radiation and cataracts (Hockwin et al., 1999; McCarty and Taylor, 2002). Ultraviolet radiation-induced cataracts are hypothesized to be caused by oxidative stress leading to increased reactive species in the lens, which then causes damage to lens DNA and cross-linking of proteins. Exposure time to low-dose UV radiation was found to strongly influence cataract formation (Ayala et al., 2000). An action spectrum determined using young female rats indicated that the rat lens was most sensitive to 300 nm, correcting for corneal transmittance (Merriam et al., 2000). Oriowo et al. (2001) examined the action spectrum for cataract formation using whole cultured pig lenses. As pig lenses are similar in shape and size to the human lens, some inferences may be made. Results indicated that the 270 to 315 nm waveband was most effective in producing UV-induced cataracts in vitro. However, the

threshold values varied widely within that range, from 0.02 J/cm² for 285 nm to 0.74 J/cm² for 315 nm (Oriowo et al., 2001). At wavelengths >325 nm, the threshold levels were orders of magnitude larger, with a minimum threshold value of 18.7 J/cm².

An epidemiologic study examined the effects of UV radiation on cataract formation in watermen (e.g., commercial fishermen, boat workers) who worked on Chesapeake Bay, MD (Taylor et al., 1988). Among the 838 individuals surveyed in this study, 111 had cortical cataracts and 229 had nuclear cataracts. Results indicated that UV-B radiation was significantly associated with cortical, but not nuclear, cataract formation. For a given age, a doubling of cumulative UV-B exposure was associated with a 60% excess risk (95% CI: 1, 164) of cortical cataracts. No association was observed between cataracts and UV-A radiation in this outdoor-working population.

Risk of Ocular Damage from Changes in Tropospheric Ozone Levels

Cataracts are the most common cause of blindness in the world. McCarty et al. (2000) calculated that ocular UV radiation exposure accounted for 10% of the cortical cataracts in an Australian cohort of 4,744 individuals from both urban and rural areas. A study by Javitt and Taylor (1994-1995) found that the probability of cataract surgery in the U.S. increased by 3% for each 1° decrease in latitude. These results suggest that depletion of the stratospheric O₃ layer may increase UV radiation-induced cataract formation. After assuming a certain wavelength dependency along with several additional assumptions, every 1% decrease in the stratospheric O₃ layer was estimated to be associated with a 0.3 to 0.6% increase in cataracts (Longstreth et al., 1995). Longstreth et al. (1995) noted that this estimate has a high degree of uncertainty due to inadequate information on the action spectrum and dose-response relationships. Quantitative estimates have not been possible for photokeratitis, pterygium, or other UV-related ocular effects due to lack of epidemiologic and experimental data.

As is the case for all of the other UV-related health outcomes, there is no published information on the potential effects on cataract formation due to any changes in surface-level UV flux resulting from decreases in tropospheric O₃.

10.2.3.5 Ultraviolet Radiation and Immune System Suppression

Experimental studies have suggested that exposure to UV radiation may suppress local and systemic immune responses to a variety of antigens (Clydesdale et al., 2001; Garssen and Van Loveren, 2001; Selgrade et al., 1997). In rodent models, these effects have been shown to worsen the course and outcome of some infectious diseases and cancers (Granstein and Matsui, 2004; Norval et al., 1999). Granstein and Matsui (2004) stated that exposure to UV-B radiation caused immunosuppression in mice ultimately by releasing cytokines that prevent antigen-presenting cells from performing their normal functions and causing direct damage to epidermal Langerhans cells. Noonan et al. (2003) investigated UV skin cancer induction in two strains of reciprocal F1 hybrid mice and found that genetically determined differences in susceptibility to UV-induced immunosuppression was a risk factor for skin cancer. At high UV radiation doses, mice with greater susceptibility to immune suppression had a larger proportion of skin tumors compared to those with lower susceptibility (Noonan et al., 2003). In a study by Yoshikawa et al. (1990), development of contact hypersensitivity to dinitrochlorobenzene on irradiated buttock skin was examined. Individuals who failed to develop contact hypersensitivity were considered to be susceptible to UV-B radiation. Virtually all skin cancer patients (92%) were susceptible to UV-B radiation-induced suppression of contact hypersensitivity, compared to approximately 40% of healthy volunteers. Others studies have observed increased skin cancer in immune suppressed organ transplant patients (Caforio et al., 2000; Lindelöf et al., 2000). Collectively, results from these studies suggest that immune suppression induced by UV radiation may be a risk factor contributing to skin cancer induction (Ullrich, 2005).

There is also some evidence that UV radiation has indirect involvement in viral oncogenesis through the human papillomavirus (Pfister, 2003). Additional evidence of UV-related immunosuppression comes from an epidemiologic study of 919 patients with rare autoimmune muscle diseases from 15 cities on four continents with variable UV radiation intensity (Okada et al., 2003). Ultraviolet radiation was strongly associated with the prevalence of dermatomyositis, an autoimmune disease distinguished by the presence of photosensitive pathognomonic rashes (Okada et al., 2003). In patients with the human immunodeficiency virus, UV-B radiation lead to activation of the virus in their skin through the release of cytoplasmic nuclear factor kappa B (Breuher-McHam et al., 2001). In a study by Selgrade et al. (2001), UV-induced immunosuppression was examined in 185 subjects with different skin

pigmentations. To assess immune suppression, dinitrochlorobenzene was applied to irradiated buttock skin 72 hours after irradiation. Differences in sensitivity were unrelated to skin type based on the Fitzpatrick classification or minimal erythemal dose (Selgrade et al., 2001). However, erythemal reactivity, assessed by the steepness of the erythemal dose-response curve, was shown to be significantly associated with UV-induced immunosuppression. Only subjects with steep erythemal responses, which included individuals with skin types I through V, showed a dose-response relationship between UV exposure and immune suppression (Selgrade et al., 2001).

In other studies, UV radiation was associated with decreased autoimmune diseases. Several ecologic studies observed a decreased prevalence of multiple sclerosis, insulin-dependent diabetes mellitus, and rheumatoid arthritis in regions with lower latitude (i.e., higher UV radiation exposure) (Ponsonby et al., 2002). These results may be attributable to UV radiation-induced immunosuppression and UV-B-related production of vitamin D, which has immunomodulatory effects (Cantorna, 2000). The protective effects of UV radiation resulting from its active role in vitamin D production are further discussed in the next section.

Most action spectrum investigations have concluded that immunosuppression is caused most effectively by the UV-B waveband (Garssen and Van Loveren, 2001). The effects of UV-A on local and systemic immunosuppression have been unclear and inconsistent. There is some evidence that high doses of UV-A are protective of immunosuppression induced by UV-B exposure (Halliday et al., 2004). Given the variety of outcomes of immune suppression and possible mechanisms of effect, little detailed information exists on UV radiation action spectra and dose-response relationships. The available data are insufficient to conduct a critical risk assessment of UV radiation-induced immunosuppression in humans.

10.2.3.6 Protective Effects of Ultraviolet Radiation—Production of Vitamin D

Any risk assessment that attempts to quantify the consequences of increased UV-B exposure on humans due to reduced ground-level O₃ must include consideration of both negative and positive effects. A potential health benefit of increased UV-B exposure relates to the production of vitamin D in humans. Most humans depend on sun exposure to satisfy their requirements for vitamin D (Holick, 2004). UV-B photons are absorbed by 7-dehydrocholesterol in the skin, leading to its transformation to previtamin D₃, which is rapidly

converted to vitamin D₃. Vitamin D₃ is metabolized in the liver, then in the kidney to its biologically active form of 1.25-dihydroxyvitamin D₃. One minimal erythemal dose produces vitamin D equivalent to an oral dose of 20,000 IU vitamin D, which is 100 times the recommended dietary allowance for adults under 50 years of age (Giovannucci, 2005; Holick, 2004).

Vitamin D deficiency can cause metabolic bone disease among children and adults, and also may increase the risk of many common chronic diseases, including type I diabetes mellitus and rheumatoid arthritis (Holick, 2004). Substantial *in vitro* and toxicologic evidence also support a role for vitamin D activity against the incidence or progression of various cancers (Giovannucci, 2005; Studzinski and Moore, 1995). Large geographical gradients in mortality rates for a number of cancers in the U.S. are not explained by dietary or other risk factors; therefore, it has been hypothesized that some carcinomas may be due to latitude-related reduction in UV-B radiation exposure. Published literature indicates that solar UV-B radiation, by increasing vitamin D production, is associated with a reduced risk of cancer. Most of these studies used an ecologic study design, in which latitude gradient was examined in relation to cancer rates. Kimlin and Schallhorn (2004) observed that latitude was a valid predictor of vitamin D-producing UV radiation. The strongest evidence exists for an association between UV radiation and reduced risk of colorectal cancer (Giovannucci, 2005; Grant and Garland, 2004; Freedman et al., 2002). Several other studies also have found an inverse relationship between UV radiation and various other cancers, including cancer of the breast (Freedman et al., 2002; Garland et al., 1990; Gorham et al., 1990; Grant, 2002a; John et al., 1999), ovary (Freedman et al., 2002; Lefkowitz and Garland, 1994), and prostate (Freedman et al., 2002; Hanchette and Schwartz, 1992), as well as non-Hodgkin lymphoma (Hughes et al., 2004; Hartge et al., 1996). Eight other cancers (i.e., bladder, esophageal, kidney, lung, pancreatic, rectal, stomach, and corpus uteri) have been found to exhibit an inverse correlation between mortality rates and UV-B radiation (Grant, 2002b).

Using UV-B data from July 1992 and U.S. cancer mortality rates from 1970 to 1994, premature cancer deaths attributable to insufficient UV-B exposure were analyzed in an ecologic study (Grant, 2002b). The minimum mortality rate, which was determined as the value corresponding to the maximum UV-B dose, was used to calculate the number of premature deaths. This analysis observed that the annual number of premature deaths from various cancers

due to latitude-related reduction in UV-B exposure was 21,700 (95% CI: 20,400, 23,400) for white Americans; 1,400 (95% CI: 1,100, 1,600) for black Americans; and 500 (95% CI: 400, 600) for Asian Americans and other minorities. Uncertainty in the estimations of UV-B exposure limits the confidence for the estimates of excess cancer deaths attributable to insufficient exposure. Caution is required in interpreting results from ecologic data; however, no strong alternative explanation is indicated in the association observed between UV radiation and the decreased risk of cancer (Giovannucci, 2005). No study has assessed the decreased risk of cancer mortality resulting from increased UV radiation attributable to decreased tropospheric O₃ levels, but the change in risk is expected to be unappreciable.

In establishing guidelines on limits of exposure to UV radiation, the ICNIRP agreed that some low-level exposure to UV radiation has health benefits (ICNIRP, 2004). However, the adverse health effects of higher UV exposures necessitated the development of exposure limits for UV radiation. The ICNIRP recognized the challenge in establishing exposure limits that would achieve a realistic balance between beneficial and adverse health effects.

10.2.4 Summary and Conclusions for Ozone Effects on UV-B Flux

Latitude and altitude are primary variables in defining UV-B flux at the Earth's surface, immediately followed in importance by clouds, surface albedo, PM concentration and composition, and then by gas phase pollution. Of all of these, only latitude and altitude can be defined with small uncertainty in any effort to develop a UV climatology for use in a public health benefits analysis relevant to the areas not presently attaining the NAAQS for O₃. Cloud cover, and its effect on surface UV flux, continues to be extremely difficult to define and predict. Particulate matter and gas-phase tropospheric pollutants are subject to similarly high degrees of uncertainty in predicting their relative concentration distributions. Land cover and, consequently, surface albedos are highly variable at the geographic scales relevant to NAAQS attainment.

Within the uncertain context of presently available information on UV-B surface fluxes, a risk assessment of UV-B-related health effects would need to factor in human habits (e.g., daily activities, recreation, dress, and skin care) in order to adequately estimate UV-B exposure levels. Little is known about the impact of variability in these human factors on individual exposure to UV radiation. Furthermore, detailed information does not exist regarding the relevant type (e.g.,

peak or cumulative) and time period (e.g., childhood, lifetime, or current) of exposure, wavelength dependency of biological responses, and interindividual variability in UV resistance. Recent reports of the necessity of UV-B in the production of vitamin D—a vitamin in which many individuals are deficient—suggests that increased risks of human disease due to a slight excess in UV-B radiation exposure may be offset by the benefits of enhanced vitamin D production. However, as with other impacts of UV-B on human health, this beneficial effect of UV-B has not been studied in sufficient detail to allow for a credible health benefits assessment. In conclusion, the effect of changes in surface-level O₃ concentrations on UV-induced health outcomes cannot yet be critically assessed within reasonable uncertainty.

10.3 TROPOSPHERIC OZONE AND CLIMATE CHANGE

Water vapor, CO₂, O₃, N₂O, CH₄, CFCs, and other polyatomic gases present in the Earth's troposphere, trap infrared radiation emitted by the Earth's surface, leading to surface warming. This phenomenon is widely known as the “Greenhouse Effect” (Arrhenius, 1896), and the gases involved are known as “greenhouse gases” (GHGs). The term used for the role a particular atmospheric component, or any other component of the greater climate system, plays in altering the Earth's radiative balance is “forcing.” In the past decade, the global atmospheric sciences and climate communities have made significant progress in determining the specific role that atmospheric O₃ plays in forcing climate.

The Intergovernmental Panel on Climate Change (IPCC) was founded in 1988 by the World Meteorological Society (WMO) and the United Nations Environmental Program (UNEP) to support the work of the Conference of Parties (COP) to the United Nations Framework Convention on Climate Change (UNFCCC). Drawing from the global climate and atmospheric sciences community for its authors and reviewers, the IPCC produces reports containing thorough assessments of the available peer-reviewed science regarding the physical climate system, past and present climate, and evidence of human-induced climate change. This section summarizes the reviews of the available information on the forcing properties of tropospheric O₃ as provided by the IPCC Third Assessment Report (IPCC, 2001a) and also describes some of the more recent developments on the subject.

The projected effects of global climate change will be briefly explained to provide the context within which O₃ serves as a regional, and possibly global, anthropogenic pollutant. The concept of climate forcing is also explained, along with the factors that influence the extent of climate forcing by O₃. The section concludes with a summary of the various estimates that have been placed on the amount of globally averaged forcing due to O₃.

10.3.1 The Projected Impacts of Global Climate Change

The study of the atmospheric processes involved in global climate change, and its potential consequences for human health and global ecosystems, is an area of active research. The IPCC Third Assessment Report (TAR) is the most thorough evaluation currently available of the science concerning climate change. In addition to the first and second IPCC assessments in 1990 and 1995, along with other IPCC reports, earlier assessments included those conducted by the UNEP (1986), the WMO (1988), the U.S. Environmental Protection Agency (1987), and others (e.g., Patz et al., 2000a,b). The reader is referred to those documents for a complete discussion of climate change science. An abbreviated list of the IPCC conclusions to date and a short discussion of the potential impacts of climate change on human health and welfare is provided here to serve as the context for the discussion of the role of the increasing tropospheric O₃ concentration in climate change.

According to various historic and modern measurement records, atmospheric GHG concentrations have increased dramatically in the past century and have been attributed to human activities. The IPCC TAR describes the scientific theory and evidence linking increases in GHGs to human activities (IPCC, 2001a).

An increasing body of geophysical observations shows that the Earth is warming and that other climate changes are underway. These observations include the global surface temperature record assembled since the year 1860, the satellite temperature record begun in 1979, recorded changes in snow and ice cover since the 1950s, sea level measurements taken throughout the 20th century, and sea surface temperature observations recorded since the 1950s.

Observations (Levitus et al., 2005) show that ~84% of the total heating of the Earth System (oceans, atmosphere, continents, and cryosphere) over the last 40 years has gone into warming the oceans. Barnett et al. (2005) have reported the emergence of a clear pattern of ocean surface warming associated with anthropogenic GHGs. The authors constructed a model-based

fingerprint (i.e., a map of predicted changes in the vertical temperature profile of the Earth's six major oceans) and compared this map to the newly upgraded and expanded ocean temperature data set (Levitus et al., 2005). They concluded that the warming signal far exceeds what would be expected from natural variability, a finding that was in compelling agreement with GHG-forced model profiles. Other evidence of ocean warming includes a marked increase in the frequency, intensity, and persistence of the zonal atmospheric circulation shifts known as the El Niño-Southern Oscillation (ENSO) over the past 100 years. ENSO events occur when the tropical Pacific Ocean has accumulated a large, localized mass of warm water that interrupts cold surface currents along South America, altering precipitation and temperature patterns in the tropics, subtropics, and the midlatitudes.

IPCC (1998, 2001a) reports also describe the results of general circulation model (GCM) studies predicting that human activities will alter the climate system in a manner likely to lead to marked global and regional changes in temperature, precipitation, and other climate properties. These changes are expected to increase the global mean sea level as well as increase the number of extreme weather events such as floods and droughts, increased wind speeds and precipitation intensity of tropical cyclones, and changes in soil moisture. These predicted changes can be expected to directly impact human health, ecosystems, and global economic sectors (e.g., hydrology and water resources, food and fiber production) (IPCC, 1998, 2001b). Table 10-1 summarizes these projected impacts.

Wide variations in the course and net impacts of climate change in different geographic areas are expected. In general, the projected changes constitute additional stressors on natural ecosystems and human societal systems already impacted by increasing resource demands, unsustainable resource management practices, and pollution. Some of the predicted changes include alterations in ecological balances; in the availability of adequate food, water, clean air; and in human health and safety. Poorer nations can be expected to suffer the most, given their limited adaptive capabilities.

Although many regions are predicted to experience severe, possibly irreversible, adverse effects due to climate change, beneficial changes may also take place. For example, certain regions may benefit from warmer temperatures or increased CO₂ fertilization, e.g., U.S. West Coast coniferous forests, and some Western rangelands. Specific benefits may include reduced energy costs for heating, reduced road salting and snow-clearance costs, longer open-water

Table 10-1. Examples of Impacts Resulting From Projected Changes in Extreme Climate Events

Projected changes during the 21st Century in Extreme Climate Phenomena and their Likelihood^a	Representative Examples of Projected Impacts^b (all high confidence of occurrence in some areas^c)
<i>Simple Extremes</i>	
Higher maximum temperatures; more hot days and heat waves ^d over nearly all land areas (<i>very likely^a</i>)	<ul style="list-style-type: none"> • Increased incidence of death and serious illness in older age groups and urban poor • Increased heat stress in livestock and wildlife • Shift in tourist destinations • Increased risk of damage to a number of crops • Increased electric cooling demand and reduced energy supply reliability
Higher (increasing) minimum temperatures; fewer cold days, frost days, and cold waves ^d over nearly all land areas (<i>very likely^a</i>)	<ul style="list-style-type: none"> • Decreased cold-related human morbidity and mortality • Decreased risk of damage to a number of crops, and increased risk to others • Extended range and activity of some pest and disease vectors • Reduced heating energy demand
More intense precipitation events (<i>very likely^a</i> over many years)	<ul style="list-style-type: none"> • Increased flood, landslide, avalanche, and mudslide damage • Increased soil erosion • Increased flood runoff could increase recharge of some floodplain aquifers • Increased pressure on government and private flood insurance systems and disaster relief
<i>Complex Extremes</i>	
Increased summer drying over most midlatitude continental interiors and associated risk of drought (<i>likely^a</i>)	<ul style="list-style-type: none"> • Decreased crop yields • Increased damage to building foundations caused by ground shrinkage • Decreased water resource quantity and quality • Increased risk of forest fire
Increase in tropical cyclone peak wind intensities, mean and peak precipitation intensities (<i>likely^a</i> over some areas) ^e	<ul style="list-style-type: none"> • Increased risk to human life, risk of infections, disease epidemics, and many other risks • Increased coastal erosion and damage to coastal buildings and infrastructure • Increased damage to coastal ecosystems such as coral reefs and mangroves
Intensified droughts and floods associated with El Niño events in many different regions (<i>likely^a</i>) (see also under droughts and intense precipitation events)	<ul style="list-style-type: none"> • Decreased agricultural and rangeland productivity in drought- and flood-prone regions • Decreased hydropower potential in drought-prone regions
Increased Asian summer monsoon precipitation variability (<i>likely^a</i>)	<ul style="list-style-type: none"> • Increased flood and drought magnitude and damages in temperate and tropical Asia
Increased intensity of midlatitude storms (little agreement between current models) ^d	<ul style="list-style-type: none"> • Increased risks to human life and health • Increased property and infrastructure losses • Increased damage to coastal ecosystems

^aLikelihood refers to judgmental estimates of confidence used by TAR WGI: *very likely* (90-99% chance); *likely* (66-90% chance). Unless otherwise stated, information on climate phenomena is taken from the Summary for Policymakers, TAR WGI. TAR WGI = Third Assessment Report of Working Group 1 (IPCC, 2001a).

^bThese impacts can be lessened by appropriate response measures.

^cHigh confidence refers to probabilities between 67 and 95%.

^dInformation from TAR WGI, Technical Summary.

^eChanges in regional distribution of tropical cyclones are possible but have not been established.

Source: IPCC (2001b).

seasons in northern channels and ports, and improved agricultural opportunities in the northern latitudes, as well as in the Western interior and coastal areas. For further details about the projected effects of climate change on a U.S.-regional scale, the reader is also referred to several regionally-focused reports (MARAT, 2000; Yarnal et al., 2000; NERAG, 2001; GLRAG, 2000), as well as a report on potential impacts of climate change on human health (Bernard et al., 2001a,b). The IPCC report, “The Regional Impacts of Climate Change,” (IPCC, 1998) describes the projected effects of human-induced climate change on various regions of the globe, including Africa, the Arctic and Antarctic, the Middle East and arid Asia, Australasia, Europe, Latin America, North America, the small island nations, temperate Asia, and tropical Asia.

While current climate models can successfully simulate the present-day annual mean global climate and the seasonal cycles on a continental scale, they have been less successful on a regional scale. Clouds and humidity, essential factors in defining local and regional (sub-grid scale) climate, are significantly uncertain (IPCC, 2001a). Due to modeling uncertainties, both in reproducing regional climate and in predicting the future economic activity that will govern future GHG emissions, the projected impacts discussed above are also uncertain.

Findings from the U.S. Global Change Research Program (USGCRP) (NAST, 2000) and related reports illustrate the considerable uncertainties and difficulties in projecting likely climate change impacts at the regional or local scale. The USGCRP findings also reflect the mixed nature of projected potential climate change impacts, i.e., combinations of deleterious and beneficial effects, for U.S. regions and the variation of projected impacts across different regions. Difficulties in projecting region-specific climate change impacts are complicated by the need to evaluate the potential effects of regional- or local-scale changes in key air pollutants not only on global-scale temperature trends, but also on regional- or local-scale temperature and precipitation patterns. The EPA is currently leading a research effort that uses regional-scale climate models with the goal of identifying changes to O₃ and PM concentrations that may occur in a warming climate. An assessment of the results of this effort will be available by the next review of the O₃ NAAQS. This focused effort to determine the impact of a warming climate on criteria air pollution requires regional-scale models with improved skill in reproducing climate history and predicting change. Among the innovations being employed in this effort is the downscaling of global circulation model outputs to provide boundary conditions for model calculations at the regional scale (Liang et al., 2005). While focusing on projecting the impact of

a warming climate on regional O₃ concentrations, the effort applied to improving regional-scale modeling will also lead to improved estimates of current and projected future impacts of tropospheric O₃ on climate.

10.3.2 Solar Energy Transformation and the Components of the Earth's Climate System

Mass, in any form, has the capacity to interact with solar and terrestrial radiation, but the manner in which it interacts with radiation is governed by its particular physical form and/or molecular properties. Water provides one of the most interesting examples of how physical form affects radiative properties. In its gaseous form, water is the most important GHG present in the climate system, because of its ability to absorb long-wave terrestrial radiation. Conversely, in its frozen form as snow or sea ice, water plays a very important role in the climate system by scattering UV and visible solar radiation back to space, i.e., decreasing the Earth's net solar radiation receipts by increasing the Earth's reflective properties (albedo). In its liquid aerosol form as clouds, water also greatly increases the Earth's albedo. In its bulk liquid form as ocean water, it absorbs terrestrial radiation and represents the Earth's most important reservoir of heat energy.

The atomic composition and molecular structure of a gas determines the wavelengths of light it can absorb and, therefore, its role in defining the heat capacity of the atmosphere. Ozone and O₂ provide examples of the relative importance of these molecular properties. While these molecules are both composed solely of oxygen atoms, their bond structures are distinct. Ozone has a three-atom, bent molecular structure, giving it the capacity to absorb terrestrial (infrared) wavelengths – making it a GHG. At any altitude, i.e., in the stratosphere or troposphere, O₃ has the capacity to absorb UV radiation of 320 nm and shorter, further increasing the energy-absorbing capacity of the troposphere. Conversely, O₂, due to its diatomic, linear structure, is limited to absorbing very short-wave UV light—and does so at altitudes too high to influence the climate system significantly.

Each component of the climate system plays a role in absorbing, transforming, storing, dispersing, or scattering solar radiation. Weather is a tangible consequence of the transformation and dispersion of terrestrial radiation within the atmosphere. The term “weather” refers to the condition of the Earth's atmosphere at a specific time and place. It is defined by several specific

variables: the air temperature, air pressure, humidity, clouds, precipitation, visibility, and wind speed. The “climate” for a given place on the Earth’s surface is a long-term average of these variables accounting for daily and seasonal weather events. The frequency of extreme weather events is used to distinguish among climates that have similar averages (Ahrens, 1994).

Climate components (including GHGs, land, oceans, sea ice, land ice and snow, atmospheric particles, vegetation, clouds, etc.) all contribute to the Earth’s heat capacity, i.e., its ability to absorb and retain solar energy. Changes in the properties (or mass) of these components will “force” the climate system in one direction or the other, i.e., warmer versus cooler. The transformation of atmospheric O₂ into O₃ by way of air pollution chemistry, enhances the heat capacity of the atmosphere. The principles behind the important concept of climate forcing are further described below.

10.3.3 The Composition of the Atmosphere and the Earth’s Radiative Equilibrium

The “greenhouse effect” is the term given to the decreased rate of reemission of absorbed solar energy due to the heat-retaining properties of the Earth’s atmosphere. According to simple radiative transfer theory, at thermal equilibrium, the Earth’s temperature should be near $-15\text{ }^{\circ}\text{C}$. This is the temperature of a theoretical “black body” that is receiving and then reemitting 342.5 Wm^{-2} , i.e., the globally averaged amount of full-spectrum solar energy absorbed and then reemitted by the Earth as infrared terrestrial radiation per square meter. In fact, satellite observations well above the atmosphere indicate that the Earth’s average *planetary* temperature is remarkably close to its theoretical black body value at $-18\text{ }^{\circ}\text{C}$, a temperature at which liquid water ordinarily does not exist.

At its *surface*, however, the Earth’s average temperature is $+15\text{ }^{\circ}\text{C}$. The $+33\text{ }^{\circ}\text{C}$ temperature differential between the Earth’s planetary and surface temperatures is due to the presence of infrared (IR) radiation-absorbing components in the atmosphere such as water vapor, CO₂, CH₄, several other trace gases, and some types of particles and clouds.

The atmosphere, when cloud-free, is largely transparent in the solar wavelength range. A small fraction of this radiation is absorbed and reemitted as black body radiation by dark atmospheric particles (IPCC, 2001a). However, the majority of clouds and particles, in part,

offset the greenhouse effect by increasing the Earth's albedo, thereby decreasing the overall amount of solar radiation absorbed by the Earth system.

Ozone, SO₂ and NO₂ also absorb ultraviolet and near ultraviolet wavelengths, in addition to infrared radiation. Once absorbed by a gas molecule, the energy introduced by a photon may induce a photochemical reaction with the residual energy thermally exciting (heating) the products of the reaction. Alternatively, the energy introduced into the molecule by the photon may be dispersed amongst neighboring molecules via intermolecular collisions, or it may be reemitted in part as a lower energy (i.e., IR) photon.

Radiation from the sun or the Earth's surface that is absorbed by gases and particles is reemitted isotropically, i.e., it is equally likely to be emitted in all directions. Therefore, to a first approximation, half of the radiation trapped by the Earth's atmosphere is reflected back to its surface. A portion of this radiation is transformed into the heat energy that drives the atmospheric processes that form the basis of weather and climate. Radiation that is not absorbed by gases and aerosols reaches the Earth's surface where it is scattered (reflected) or absorbed, depending on the surface albedo.

Successful modeling of the Earth's climate and, therefore, the assessment of the extent of human-induced climate change and development of appropriate policy depend on the quality of available information on the relative efficiencies, amounts, and spatial and temporal distributions of the various radiatively active components of the atmosphere that absorb and/or reflect solar and terrestrial radiation, along with all the other nonatmospheric components of the Earth system.

10.3.3.1 Forcing of the Earth's Radiative Balance

As mentioned earlier, the commonly used measure of the relative influence of a given component of the climate system on the Earth's radiative balance is its radiative forcing (IPCC, 2001a; Houghton et al., 1990). Radiative forcing, in Wm⁻², is a quantity that was developed by the climate modeling community as a first-order-only means of estimating relative effects of individual anthropogenic and natural processes on the energy balance within the climate system.

When the effect of a particular component of the climate system is to reduce the amount of solar energy absorbed, usually by increasing the Earth's albedo, this component is said to provide a "negative" forcing, measured in Wm⁻². The convention assigns a positive value to the

forcing induced by climate system components that enhance the greenhouse effect or otherwise act to increase the heat-absorbing capacity of the Earth system. Purely reflective atmospheric aerosol, clouds, white rooftops, snow-covered land surfaces, and dense sea ice provide a negative forcing, while highly absorbing dark-colored atmospheric aerosols, GHGs, and increases in dark ocean surface area, due to the melting of sea ice sheets, positively force the climate system.

Global and regional climate are roughly defined by the balance between the large number of positive and negative forcings induced by the many different components of the Earth system and any changes in the properties of those components due to natural processes or anthropogenic activities. Following a perturbation or added forcing, such as an increase in GHG concentrations or modification the Earth's albedo through changes in land use, this balance is re-established via a complex redistribution of energy within the Earth system. Feedback mechanisms that are theorized but difficult to resolve at the quantitative level further complicate the prediction of the sensitivity of variables, such as surface temperature, to changes in forcing.

A simple example of a positive feedback would be melting sea ice. As sea ice melts with increasing ocean temperatures, the dark ocean surface that is revealed is more efficient at absorbing IR radiation, further increasing the rate of warming. A negative feedback would be the formation of clouds over a moist, warming surface. As clouds form, less radiation is available to warm the surface, leading to cooling. The role of feedbacks in determining the sensitivity of climate to changes in radiative forcing is described in detail in the IPCC TAR (IPCC, 2001a).

Discussions are presently underway within the climate community regarding a metric to replace forcing as the standard measure of climate impact—one that will account for more of the factors that determine the effectiveness of a specific change in altering climate. However, forcing remains the current standard (NRC, 2005).

The IPCC has reported estimated values for forcing by individual radiatively active gases, and by particle-phase components of the atmosphere that were derived primarily through expert judgment incorporating the results of peer-reviewed modeling studies. The forcing estimates, shown in Figure 10-6, are global averages attributed to known GHGs, including O₃, particles, anthropogenic cirrus clouds, land-use change, and natural solar flux variations. Uncertainty ranges are assigned to reflect the range of modeled values reported in those studies. The current

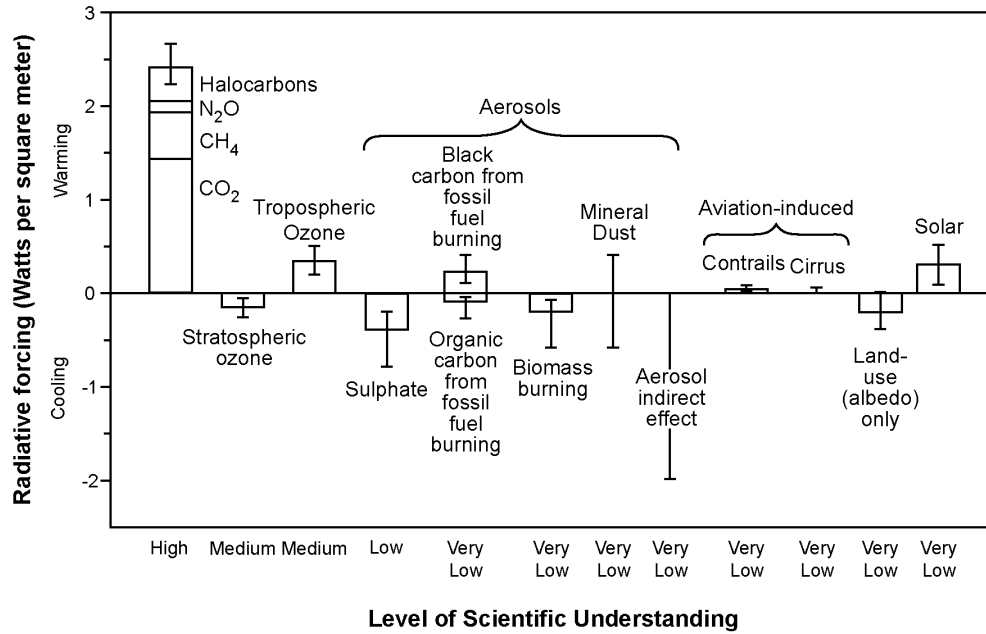


Figure 10-6. Estimated global mean radiative forcing exerted by gas and various particle phase species for the year 2000, relative to 1750.

Source: IPCC (2001a).

estimate of forcing due to long-lived, well-mixed, GHGs accumulated in the atmosphere from the preindustrial era (ca., 1750) through the year 2000 is $+2.4 \text{ Wm}^{-2} \pm 10\%$ (IPCC, 2001a). An indication of the level of confidence in each of these estimates is given along the bottom of this figure, again reflecting the expert judgment of the IPCC.

The IPCC reported a global average forcing value of $0.35 \pm 0.15 \text{ Wm}^{-2}$ for tropospheric O₃, based on model calculations constrained by climatological observations. The considerations and studies used to estimate this value are outlined below. Hansen and Sato (2001), accounting for uncertainties in pre-industrial emissions levels, more recently estimated a value of $0.5 \pm 0.2 \text{ Wm}^{-2}$ for forcing by O₃.

10.3.4 Factors Affecting the Magnitude of Climate Forcing by Ozone

The radiative properties of O₃ are distinct from those of other important GHGs in that it is capable of absorbing both UV and IR radiation. Furthermore, it is able to absorb long-wave radiation in a portion of the IR spectrum where water vapor does not absorb, i.e., the 9 to 10 mm

wavelength range, meaning that its ability to trap heat and force climate are unchanged by variations in humidity. Given its relatively short atmospheric lifetime in comparison to other GHGs, the distribution of tropospheric O₃ is highly variable in geographic extent and time. These properties enhance the prospect of attributing a unique geographic and time-dependent pattern or fingerprint to forcing by O₃.

Due to human activities, tropospheric O₃ is estimated to have provided the third largest increase in direct forcing since preindustrial times. It may also play a role in indirect forcing through its participation in the oxidative removal of other radiatively active trace gases, such as CH₄ and the HCFCs. The direct and indirect forcing that tropospheric O₃ imposes on the climate system depends upon its geospatial and temporal distribution, but it also depends upon its vertical position (altitude) in the atmosphere and the albedo of the underlying surface.

10.3.4.1 The Global Burden of Tropospheric Ozone

Little historical data exist that may be used to estimate the global O₃ burden prior to industrialization, although a few late 19th-century measurements suggest that O₃ has more than doubled in Europe during the 20th century. The insufficient data record on preindustrial tropospheric O₃ distributions introduces a major uncertainty in the estimation of the change in O₃-induced forcing since that period (IPCC, 2001a; Mickley et al., 2004a; Mickley et al., 2001; Shindell and Faluvegi, 2002).

Ozone reacts photochemically at time scales that are generally shorter than those for large-scale mixing processes in the atmosphere. Concentrated O₃ plumes evolve downwind of strong sources of its precursor pollutants: reactive nitrogen, CO, and non-methane hydrocarbons (NMHCs). The most important of these sources are midlatitude industrialized areas and tropical biomass burning. When viewed from above the atmosphere by satellite-borne spectrometers, O₃ enhancements appear as relatively localized air masses or regional-scale plumes, usually originating from industrialized areas or areas in which active biomass burning is underway. The IPCC (2001a) describes the efforts of several research teams who have analyzed data supplied by the satellite-borne Total Ozone Mapping Spectrometer (TOMS) and other remote-sensing instruments to map the global distribution of tropospheric O₃ and to attempt to identify processes that influence the global tropospheric O₃ budget (IPCC, 2001a). More recently, coincident observations of total O₃ by TOMS and the Solar Backscattered UV (SBUV) instrument were

used by Fishman et al. (2003) to construct well-resolved spatial and temporal maps of the regional distribution of tropospheric O₃. Their results were consistent with those reported by others, but with higher regional-scale resolution. They reported large O₃ enhancements in the southern tropics in austral spring and in the northern temperate latitudes in the summer. The regional nature of high O₃ concentrations was clearly visible in northeastern India, the eastern United States, eastern China, and west and southern Africa, each coincident with high population densities. Fishman et al. (2003) noted, as have the other groups cited above, significant interannual variability in the concentrations observed over these regions. *In situ* measurements of tropospheric O₃ concentrations range from 10 ppb over remote oceans to 100 ppb in both the upper troposphere and in plumes downwind from polluted metropolitan regions (IPCC, 2001a). Ground-level concentrations in urban areas are often >100 ppb. In the Southern Hemisphere, one of the largest sources of O₃ precursors is biomass burning. Biomass burning elevates O₃ on large spatial scales, particularly in the tropical Atlantic west of the coast of Africa and in Indonesia.

In its third assessment report, the IPCC estimates placed the global burden of tropospheric O₃ at a highly uncertain 370 Tg, equivalent to an average column density of 34 Dobson Units (1 DU = 2.687×10^{16} molecules/cm⁻²) or a mean concentration of 50 ppb (IPCC, 2001a). Accounting for differences in levels of industrialization between the hemispheres, the average column burden in the Northern Hemisphere is estimated to be 36 DU, with the Southern Hemisphere estimated to average 32 DU. Due to its rapid photochemistry, individual surface measurements of tropospheric O₃ cannot capture large-scale concentrations, nor will they represent the higher altitude concentrations. Dense surface and vertical measurements (ozonesondes) would be required to supplement available output from remote sensing instruments to provide the complete set of observations needed to derive a credible global O₃ budget. Such a measurement program appears, at present, to be impractical.

10.3.4.2 Background Concentrations versus Regionally-Oriented Ozone Enhancements

Vingarzan (2004) surveyed the air quality literature and reported that annual average background O₃ concentrations at ground level in the Northern Hemisphere appear to range between 20 and 45 ppb, depending upon geographic location, elevation, and the influence of local sources. Fiore et al. (2003) modeled the U.S. continental O₃ concentrations and found that

surface background levels overlap the lower end of the range reported by Vingarzan (2004), e.g., 15 to 35 ppb, with higher levels (40 to 50 ppb) arising at high-elevation sites due to the influence of the upper troposphere (See Chapter 3 and its associated annexes for a complete discussion of “policy relevant background [PRB]). Local- and regional-scale enhancements in O₃ may be thought of as roughly superimposed upon these background levels, with the exception of longer stagnation events in which preexisting background O₃ reacts away or is deposited as fresh O₃ is produced from locally-emitted precursors.

Lin et al. (2001) analyzed the EPA AIRS database for the 1980-1998 period and noted that O₃ concentrations have declined at the high end of the probability distribution, consistent with the effects of emissions controls, but had increased at the low end of the distribution by 3 to 5 ppb. They divided the monitoring data for the continental U.S. into 4 quadrants by geography and noted a pattern of increase for the Western states that might be attributed to the long-range transport of O₃ precursors from Asia. They found, however, that the Northeastern quadrant had the highest increase in the low end of the concentration probability distribution, which could not be reasonably attributed to transboundary transport of O₃ precursors.

While not representing an ideal source of information for assessing the climatic effects of O₃ within the continental United States, data from the large air-quality-focused ground-based monitoring network may be used to identify boundary-layer geospatial and temporal patterns in O₃ concentrations for comparison to regional-scale chemistry/climate models. Extensive analysis of data available within the EPA AQS database can be found in Chapter 3 of this document, including an analysis showing the diurnal O₃ concentration patterns for several large metropolitan areas with peak values ranging up to 160 ppb (Los Angeles). Lehman et al. (2004) analyzed the AQS database of daily 8-h maximum O₃ concentrations collected for 1,090 stations in the eastern half of the United States for the 1993 to 2002 period. They applied a rotated principle component analysis to a reasonably complete, spatially representative, nonurban subset of the database in order to identify coherent, regionally oriented patterns in O₃ concentrations. Five spatially homogenous regions were identified: the U.S. Northeast, Great Lakes, Mid-Atlantic, Southwest (including Alabama, Louisiana, Texas, Oklahoma), and Florida. The Mid-Atlantic region displayed the highest mean concentration (52 ppb) of all of the regions analyzed, followed by the Great Lakes, Southwest, and Northeast regions with around 47 ppb. The average concentration derived for Florida was 41 ppb. The authors found strong

correlations in measured concentrations among stations within the same region, suggesting that the geospatial patterns of pollutant emissions and meteorological activity may also have a regional orientation. These results suggest that these regions may define natural domains for regional scale modeling studies of the influence of O₃ (as well as PM) on climate.

10.3.4.3 Ozone Trends: Globally and in North America

For the Northern Hemisphere, weekly continuous data are available from 1970 for only nine stations in the latitude range 36° N to 59° N (IPCC, 2001a). Available tropospheric O₃ measurements do not reveal a clear trend in concentration, while trends in the stratosphere are more readily identified. Different trends are seen at different locations for different periods, consistent with regional changes in pollutant emission, especially NO_x. Logan et al. (1999) analyzed the composite record of mid-tropospheric O₃ abundance from the nine-station network. A plot of data is shown in Figure 10-7. While no clear trend appeared for 1980 through 1996, the average level for second half of this record (about 57 ppb) is clearly greater than for the first half (about 53 ppb). The trend may be consistent with changes in regional NO_x emission rates occurring due to pollution reduction efforts in developed countries being offset by increasing emissions in rapidly growing economies in Asia. The measurements shown in Figure 10-7 are concentrations observed at 4 to 7 km (mid-troposphere). Fewer locations have measured changes in the concentrations of O₃ as a function of altitude. Fewer still are locations that have collected and maintained data records prior to 1970. The absence of historical data on the vertical distribution of O₃ adds to the difficulty in estimating historical atmospheric burdens and trends in O₃-related climate forcing.

The IPCC (2001a) surveyed the results of published chemistry transport model (CTM) modelling studies (see Table 10-2) that estimated the global average increases in total column O₃ since the preindustrial era. Model estimates ranged from +7 to +12 DU. On the basis of these estimates, available measurements, and other analyses, the IPCC estimated that total column O₃ has increased by 9 DU, with a 67% confidence range of +6 to +13 DU. In some of the modelling studies, emissions scenarios predicted a further increase in column O₃ due to growing emissions of O₃ precursors. Fusco and Logan (2003) stated that, according to the models, increased NO_x emissions from fossil fuel combustion have had the greatest effect on O₃ in the lower troposphere since the 1970s. In addition, increases in background CH₄ have also contributed as

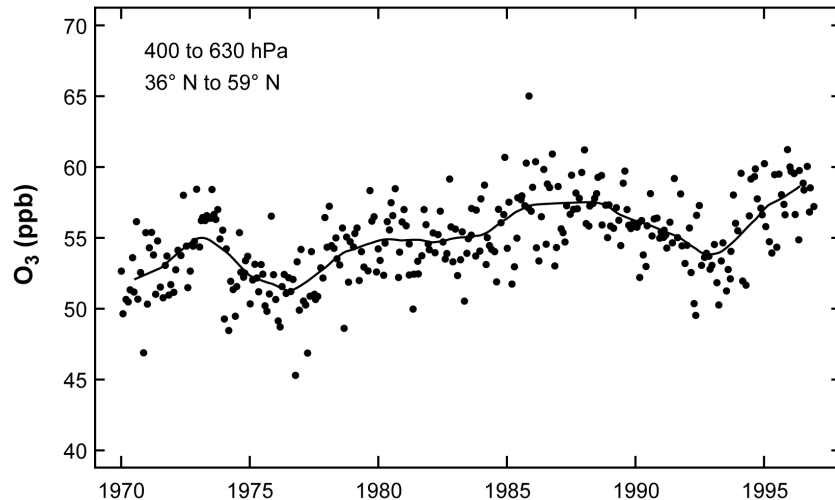


Figure 10-7. Mid-tropospheric O₃ abundance (ppb) in northern midlatitudes (36 °N-59 °N) for the years 1970 to 1996. Observations between 630 and 400 hPa are averaged from nine ozonesonde stations (four in North America, three in Europe, two in Japan), following the data analysis of Logan et al. (1999). Values are derived from the residuals of the trend fit, with the trend added back to allow for discontinuities in the instruments. Monthly data (points) are shown with a smoothed 12-month mean (line).

Source: IPCC (2001a).

Table 10-2. CTM Studies Assessed by the IPCC for its Estimate of the Change in Global and Total Column O₃ Since the Preindustrial Era

Estimated Change in Column O ₃ in DU	Model Used	References
7.9	<i>GFDL</i>	Haywood et al. (1998)
8.9	<i>MOZART-1</i>	Hauglustaine et al. (1998)
8.4	<i>NCAR/2D</i>	Kiehl et al. (1999)
9.5	<i>GFDL-scaled</i>	Levy et al. (1997)
12	<i>Harvard/GISS</i>	Mickley et al. (1999)
7.2	<i>ECHAM4</i>	Roelofs et al. (1997)
8.7	<i>UKMO</i>	Stevenson et al. (2000)
9.6	<i>UIO</i>	Berntsen and Isaksen (1999)
8	<i>MOGUNTIA</i>	Van Dorland et al. (1997)

Source: IPCC (2001a)

much as 20% to the increase in tropospheric O₃ in the northern latitudes. Given its longer atmospheric residence time, CH₄ can serve as an O₃ precursor at much longer distances from its source than can other O₃ precursors and, therefore, has a more uniform effect across the globe.

Fusco and Logan (2003) found a 10% increase in O₃ concentrations year-round over Canada, Europe, and Japan and a 20% increase for Japan and Europe during spring and summer. It was expected—but not found—that O₃ concentrations over Japan would increase in line with emissions from China. The authors suggested that convective activity over Asia is stronger than that seen over other industrialized areas of the globe. Such a meteorological characteristic would result in an injection of pollutants into the free troposphere, allowing long-range transport to North America. Their suggestion is supported by evidence of increasing background concentrations within the United States (Fiore et al., 1998).

NARSTO (2000), in its assessment of the available information on O₃ pollution in North America, stated that no single pattern for trends in O₃ over North America can be found in the available monitoring data. In the United States, the average 1-h concentration at surface monitoring sites decreased by 15% between 1986 and 1996, with most of the observed declines occurring in urban and urban-influenced locations. The largest declines occurred in Los Angeles, New York, and Chicago. Free tropospheric O₃ concentrations appeared to hold steady, or only declined slightly, from the 1980s forward.

In preparation for the IPCC TAR (IPCC, 2001a), research groups engaged in modeling global-scale tropospheric chemistry were invited to participate in a model intercomparison focusing on potential changes in the oxidative capacity of the atmosphere (OxComp), which included O₃ concentrations for the 2000 to 2100 period. Participating groups employed the IPCC A2p scenario, i.e., including the highest emissions levels, to calculate the geospatial distribution of O₃ up to 20 km. The predicted spatial distributions of O₃ were quite variable, but the predictions for total column O₃ density change fell within 9 DU of each other (11.4 to 20.5 DU) in all cases and that was considered to be encouraging by the authors. Fusco and Logan (2003) pointed out that several unresolved issues may limit the ability of models in reproducing observed trends in tropospheric O₃. Among these are the use of different meteorological inputs, photochemical reaction schemes, and predicted cloud cover—each contributing to different predictions in O₃ production and loss rates.

10.3.4.4 The Sensitivity of Ozone-Related Forcing Surface to Albedo

The characteristics of the surface underlying an O₃ enhancement play a role in the O₃ forcing effect. Highly reflective surfaces, such as light-colored deserts, sea ice and snow, scatter solar short wave (UV and visible) radiation. UV and visible radiation can then be absorbed, transformed into long-wave radiation, and reemitted in part back to the surface by tropospheric O₃. Studies by two groups, Hauglustaine et al. (1998) and Mickley et al. (1999), have shown that industrial pollution that has been transported to the Arctic induces a high, regional O₃-related forcing due to the highly reflective underlying ice and snow surface.

Liao et al. (2004) calculated that the maximum change in O₃-related top-of-the-atmosphere forcing occurs over high albedo regions in high northern latitudes. Surface forcing was calculated to be greatest at high northern latitudes as well as at dust-source regions, which also tend to have high surface albedos.

10.3.4.5 The Altitude Dependence of Forcing by Tropospheric Ozone

Altitude plays an important role in the forcing effect of tropospheric O₃ (IPCC, 1992; Gauss et al., 2003). The efficiency of IR absorption by O₃ depends upon its temperature—at atmospheric temperatures that are low, relative to the Earth's surface, it has the capacity to absorb more IR radiation than O₃ at temperatures close to that of the surface. While this temperature effect applies to all GHGs, it introduces a complication for estimating forcing by O₃, because O₃ is not homogeneously mixed within the troposphere. Ozone forcing estimates must account for these difficult-to-predict vertical inhomogeneities. However, as part of the OxComp modeling intercomparison, Gauss et al. (2003) found that the overall forcing by O₃ can be calculated within reasonable uncertainty simply on the basis of total column density.

10.3.4.6 Co-occurrence of Ozone with Particulate Matter

Analysis of the 2001 data from the AQS database showed infrequent co-occurrence of high PM_{2.5} and O₃ concentrations (Chapter 3 of this document). For those cases when O₃ production is high, in combination with high PM concentrations, there is a suggestion in the literature that heterogeneous chemistry on PM surfaces may lead to reduced gas-phase O₃. Liao et al. (2004) modeled heterogeneous chemistry taking place on PM, and found a significant

titration of O₃ and its NO_x precursors. The importance of this titration effect remains an open question, given the difficulty in obtaining in situ measurements to validate model calculations.

Liao et al. (2004) also estimated that forcing by BC, mineral dust, and organic carbon aerosols substantially offsets forcing by tropospheric O₃, yielding an overall negative globally averaged forcing at both the top of the atmosphere and at the Earth's surface. However, such estimates neglect the regional aspects of forcing by these individual pollutants. Elevated concentrations of these very different types of pollutants often appear independently of the others, such as with biomass burning plumes, Saharan dust, and organic aerosols associated with biogenic terpene emissions by forests. It is unlikely that a global average of the forcing effects of these individual pollutants will adequately capture their impacts on climate at the regional scale.

10.3.5 Estimated Forcing by Tropospheric Ozone

10.3.5.1 Direct Climate Forcing Due to Ozone

The inhomogeneous distribution of O₃ within the troposphere, coupled with the large uncertainty in the global O₃ budget, significantly complicates the matter of estimating the global average direct forcing due to O₃. The IPCC Third Assessment Risk (2001a) lists the results of several modeling studies that estimated the annual change in the relative forcing by O₃ from preindustrial times. It was noted that the differences among the estimates were most likely due to differences in predicted O₃ chemistry, including the emissions inventories used and the chemical process and transport mechanisms incorporated into the models, rather than by factors relating to radiative transfer. The IPCC intercomparison of the models and their results indicate that the uncertainties in estimated forcings due to O₃ have decreased since the IPCC Second Assessment Report (1996).

The O₃-related forcings estimated by studies considered by the IPCC (2001a) are listed in Table 10-3. Ten of the listed estimates are based on global CTM calculations. One study was constrained by a climatology derived from observations. Given the differences in calculated total column O₃ among the models, a normalized forcing (Wm⁻² per Dobson Unit of tropospheric O₃ change) is listed in addition to the absolute forcing (Wm⁻²) estimated by each model. Both clear sky (cloud-free) and total sky (including clouds) forcing estimates are listed.

Table 10-3. Tropospheric O₃ Change (O₃) in Dobson Units (DU) Since Preindustrial Times, and the Accompanying Net (SW plus LW) Radiative Forcings (Wm⁻²), After Accounting for Stratospheric Temperature Adjustment (using the Fixed Dynamical Heating Method). Estimates are Taken From the Published Literature. Normalized Forcings (norm.) Refer to Radiative Forcing per O₃ Change (Wm⁻² per DU)

Reference	Clear sky conditions			Total sky conditions	
	ΔO ₃	Net	Net (norm.)	Net	Net (norm.)
Berntsen et al. (1997) – [Reading model]	7.600	0.310	0.041	0.280	0.037
Stevenson et al. (1998)	8.700	0.391	0.045	0.289	0.033
Berntsen et al. (1997) – [Oslo model]	7.600	0.390	0.051	0.310	0.041
Haywood et al. (1998)	7.900	0.380	0.048	0.310	0.039
Kiehl et al. (1999)	8.400	0.379	0.045	0.320	0.038
Berntsen et al. (2000)	9.600	0.428	0.045	0.342	0.036
Brasseur et al. (1998)	—	—	—	0.370	—
Van Dorland et al. (1997)	8.070	0.443	0.055	0.380	0.047
Roelofs et al. (1997)	7.200	0.397	0.055	0.404	0.056
Lelieveld and Dentener (2000)	—	—	—	0.420	—
Hauglustaine et al. (1998)	8.940	0.511	0.057	0.426	0.048
Mean	8.224	0.403	0.049	0.343	0.042

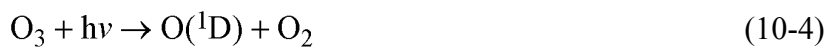
Source: IPCC (2001a).

The largest O₃-related forcings coincide with the strongest sources of tropospheric O₃, which the models predict occur in the northern midlatitude regions (40° to 50° N) and reach as much as 1 Wm⁻² in the summer as well as in the tropics, and are related to biomass burning. In general, the estimates are comparable in magnitude and show similarity in geographic distribution. For total sky conditions, the range in globally and annually averaged tropospheric O₃ forcing from all of these models is from 0.28 to 0.43 Wm⁻², while the normalized forcing is 0.033 to 0.056 Wm⁻² per DU. As expected, they are opposite in sign to the forcing estimated for sulfate aerosols, which scatter radiation. The range in normalized forcings emphasizes the differences in assumptions used by the different models. The

tropospheric O₃ forcing constrained by the observational climatology is 0.32 Wm⁻² for globally averaged, total sky conditions. As shown in Figure 10-6, the IPCC (2001a) concluded that 0.35 ± 0.15 Wm⁻² represents the most likely value for annually and globally-averaged forcing by tropospheric O₃. Not included here is the study by Hansen and Sato (2001), that evaluated forcing by O₃ with corrections made to the assumptions concerning pre-industrial O₃ concentrations and the effects of natural O₃ precursors, especially NO_x generated by lightning. Hansen and Sato (2001) concluded that a more likely range for globally averaged forcing by O₃ is 0.4 to 0.8 Wm⁻², with 0.5 Wm⁻² as their best estimate. Since the publication of the IPCC Third Assessment Report (2001a), new studies have been published that illuminate some of the regionally-relevant details associated with direct forcing by O₃ (Mickley et al., 2004a; Liao et al., 2004). Forcing by O₃, due to its capacity for absorbing solar UV as well as solar and terrestrial IR radiation, can be divided into “shortwave” forcing and “long-wave” forcing. These forcings occur under different conditions. Shortwave forcing can only take place during daytime, while long-wave forcing can occur at all hours as a function of the diurnally varying concentration of atmospheric O₃. As noted, earlier, unlike CO₂, the absorption spectrum for O₃ is distinct from that of water vapor—meaning that O₃ will absorb and reemit long-wave radiation without interference by water under high humidity conditions. Mickley et al. (2004a) reported that, according to their modeling study, surface temperature response to the predicted O₃ enhancement since the preindustrial period differs greatly from that of the CO₂ response, and that this difference can only be explained by the geographical distribution and absorption properties of O₃. Liao et al. (2004) estimated globally averaged top-of-the-atmosphere separate short- and long-wave forcings to O₃ to be 0.21 W/m² and 0.32 W/m², respectively.

10.3.5.2 Indirect Forcing Due to Ozone

Ozone has an indirect climate forcing effect due to its role in the oxidative removal of other reactive GHGs, including CH₄, hydrofluorocarbons (HFCs), and other reactive NMHCs. The primary actor in this effect is a second generation product of the photolysis of O₃, the hydroxyl (OH) radical. Hydroxyl radicals are produced by way of a pair of reactions that start with the photodissociation of O₃ by solar UV.



Reactions with OH are the primary removal mechanism for CH₄ and NMHCs as well as the pollutants NO_x and CO. Methane and CO are in especially high abundance in the global atmosphere. OH is estimated to react with these two gases within 1 second of its formation. In addition to CH₄, NO_x, CO, and the NMHCs, OH concentrations are controlled by local concentrations of H₂O (i.e., humidity) and the intensity of solar UV. Different atmospheric concentrations of the required precursors suggest that preindustrial OH concentrations were likely to have been different from present-day concentrations, but there is no consensus on the magnitude of this difference. Observations of global atmospheric concentrations of methylchloroform (CH₃CCl₃), a well-mixed tropospheric species that also reacts with OH, have been used to estimate OH abundances. Independent studies have shown overlapping trends for the period 1978 to 1994, but none of the trends are outside the given uncertainty ranges (0.5 ± 0.6%/year) (Prinn et al., 1995; Krol et al., 1998). The IPCC (2001a) reported a range of +5% to -20% for predicted changes in global OH abundances.

Given the difficulty in estimating global OH abundances in the past, present, and future, estimates of indirect forcing due to O₃ have been difficult to obtain and are highly uncertain.

10.3.5.3 Predictions for Future Climate Forcing by Anthropogenic Ozone

Not surprisingly, CTM modeling attempts to predict future precursor emissions and resulting O₃ abundances indicate that the largest future O₃-related forcings will be related to population growth and economic development in Asia (Van Dorland et al., 1997; Brasseur et al., 1998). The results of these modeling studies predict that the globally averaged total radiative forcing due to O₃ from preindustrial times 0.66 Wm⁻² will rise to 0.63 Wm⁻² by 2050. Chalita et al. (1996) predicted a change in the globally averaged radiative forcing from preindustrial times to 2050 of 0.43 Wm⁻². Stevenson et al. (1998) predicted an O₃-related forcing of 0.48 Wm⁻² in 2100. Applying the SRES scenario projecting the highest emissions out to the year 2100 (IPCC, 2000), the OxComp model intercomparison study yielded a projected O₃-induced forcing ranging from 0.40 to 0.78 Wm⁻². The authors concluded, given their prediction for

forcing by well-mixed GHGs of 5.6 Wm^{-2} , that O_3 would remain an important contributor to overall anthropogenic forcing well into the future. However, all of these predictions must be viewed with caution given the considerable uncertainties associated with the long-term economic activity projections required for such estimates.

10.3.6 The Impact of a Warming Climate on Atmospheric Ozone Concentrations

Evaluation of the potential impact of climate warming on U.S. air quality is currently underway. Initial modeling results reported by Mickley et al. (2004b) suggest that reduced cyclone frequency in a warmer climate will lead to increases in the severity of summertime pollution episodes. Cyclonic weather patterns are known to play an important role in ventilating pollution away from the surface. They note that compelling evidence is accumulating that the frequency of these cyclones has decreased over the past few decades. An early study by Jacob et al. (1993) found a correlation between O_3 concentrations and temperature was due to the effect of temperature on atmospheric chemistry, biogenic emissions, and stagnation.

10.3.7 Conclusion

The general consensus within the atmospheric sciences community, as represented by the United Nations Intergovernmental Panel on Climate Change (IPCC), is that human activities have a discernable effect on the Earth's climate. However, quantifying the extent of human-induced forcing on climate requires detailed information about human-induced change on the components of the Earth System that govern climate. Tropospheric O_3 is a well-known GHG, but information regarding its historical trends in concentration, its current and future atmospheric burden, and other critical details needed for estimating its direct and indirect forcing effects on the climate system are highly uncertain.

The IPCC has estimated that the globally averaged forcing due to O_3 is approximately $0.35 \pm 0.15 \text{ Wm}^{-2}$, with an updated value of $0.5 \pm 0.2 \text{ Wm}^{-2}$ provided by Hansen and Sato (2001). However, the most important role of O_3 in climate is likely to be at the regional scale, adjacent to the sources of its chemical precursors. This expectation is consistent with satellite observations of high regional scale column O_3 densities near large urban areas and large-scale biomass burning activity. Modeling studies evaluated by the IPCC have estimated that regional-

scale forcing due to O₃ can approach 1 Wm⁻², or as much as 40% of the globally averaged forcing due to the well-mixed GHGs. While more certain estimates of the overall importance of global-scale forcing due to tropospheric O₃ await further advances in monitoring and chemical transport modeling, the overall body of scientific evidence suggests that high concentrations of O₃ on the regional scale could have a discernable influence on climate, leading to surface temperature and hydrological cycle changes. Confirming this effect requires improvement in regional-scale modeling—an activity that is currently underway.

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11. OZONE EFFECTS ON MAN-MADE MATERIALS

Ozone (O₃) and other photochemical oxidants react with many economically important man-made materials, decreasing their useful life and aesthetic appearance. Some materials known to be damaged by O₃ include elastomers, fibers, dyes, and paints. This chapter provides a brief discussion of O₃ effects on man-made materials, including denoting of damage mechanisms and, where possible, concentration-response relationships. Much of what is known about O₃ effects on man-made materials is derived from research conducted in the 1970's, 1980's, and early 1990's, with very little new research on the subject having been conducted since then. Since only very limited new information has been published on O₃ effects on materials, this chapter mainly summarizes key information assessed in the previous 1996 Air Quality Criteria Document for Ozone and other photochemical oxidants (1996 O₃ AQCD) (U.S. Environmental Protection Agency, 1996) and provides detailed discussion of the very limited new information that has become available since then. In the ensuing sections, discussion is focused on O₃ effects on: elastomers (Sect 11.1); textiles and fabrics (11.2); dyes, pigments, and inks (11.3); artist's pigments (11.4); and surface coatings (11.5). Evaluation of the relevance and economic importance of O₃ materials damage information, as it affects productivity or cultural resources (such as museums), is beyond the scope of this chapter. The reader is referred to the previous criteria document (1996 O₃ AQCD) for more detailed discussion of the earlier studies summarized below.

11.1 ELASTOMERS

The elastomeric compounds, natural rubber and synthetic polymers and copolymers of butadiene, isoprene, and styrene, are particularly susceptible to even low levels of O₃. Elastomeric compounds are long chain unsaturated organic molecules. Ozone damages these compounds by breaking the molecular chain at the carbon-carbon double bond; a chain of three oxygen atoms is added directly across the double bond, forming a five-membered ring structure (Mueller and Stickney, 1970). The change in structure promotes the characteristic cracking of

stressed/stretched rubber called “weathering.” A 5% tensile strain will produce cracks on the surface of the rubber that increase in number with increased stress/stretching. The rate of crack growth is dependent on the degree of stress, the type of rubber compound, concentration, time of exposure, velocity, and temperature (Bradley and Haagen-Smit, 1951; Lake and Mente, 1992) (Gent and McGrath, 1965). Once cracking occurs, there is further penetration, additional cracking, and eventually mechanical weakening or stress relaxation (U.S. Environmental Protection Agency, 1996). Razumovskii et al. (1988) demonstrated the effect of O₃ on stress relaxation of polyisoprene vulcanizates. A decrease in stress (stress relaxation) is caused by O₃-induced cracks in exposed elastomers resulting in irreversible changes in the elastomer dimensions and decreased tensile strength.

To counteract O₃ effects on elastomers, antiozonants and wax are often added to elastomeric formulations during processing. An antiozonant is an additive used to protect a polymer against the effects of O₃-induced degradation and, hence, is used mainly in diene rubbers. Antiozonant protection works either (a) by providing a physical barrier to O₃ penetration via forming a thin surface film of an O₃-resisting wax or (b) by chemically reacting with O₃ or polymer ozonolysis products, as do aromatic diamines such as p-phenylene diamine derivatives. The antiozonant diffuses to the surface of the elastomeric material, where it reacts with O₃ faster than O₃ reacts to break the molecular chain and the carbon-carbon double bond, or the antiozonant diffuses to the surface of the material but is not reactive with O₃ and serves as a protective coating against O₃ attack. The antiozonant may also serve to scavenge O₃ while also providing protective film against O₃ attack (Andries et al., 1979; Lattimer et al., 1984).

Most studies of O₃ effects on elastomers were designed to evaluate the effectiveness of antiozonants in counteracting the rubber cracking produced by O₃ exposure. Consequently, many of the studies were conducted using O₃ concentrations notably higher than those typically found in the ambient air. Natural rubber strips exposed to high concentrations of O₃ (20,000 ppm) under stressed conditions cracked almost instantaneously and were broken within 1 sec. When the O₃ concentration was lowered (0.02 to 0.46 ppm), the time to required to produce cracks in the exposed rubber material was increased (Bradley and Haagen-Smit, 1951). Lake and Mente (1992) studied the effect of temperature on O₃-induced elastomer cracking and antiozonant protection on natural rubber, epoxidised natural rubber, and two acrylonitrile-butadiene copolymers under constant strain. Temperatures ranged from -20 °C to +70 °C.

The elastomers were exposed to 0.05 to 1,000 ppm O₃ for 16 h. Ozone cracking decreased at lower ambient temperatures; however, diffusing of both chemical and wax antiozonants also slowed at the lower temperatures. Cracking was slightly increased at the higher temperatures, but the antiozonants offered more protection.

Serrano et al. (1993) evaluated the appropriateness of using O₃-induced elastomer cracking to estimate ambient O₃ concentrations. Two vulcanized natural rubber compounds were exposed for 24 h to varying O₃ concentrations under stressed conditions. Ozone concentrations were 60, 80, 90, 100, and 120 ppb for durations of 2, 4, or 6 h. The 24 h average O₃ concentrations ranged from 31 to 57.5 ppb. There was a clear relationship between the 24-h average O₃ concentration and the distribution of crack length frequencies on the rubber surface. Table 11-1 gives the average 24-h O₃ concentration and lengths for two vulcanized natural rubber strips.

Table 11-1. Average 24-h Ozone Concentrations Producing the Highest Frequency of Cracks of a Certain Length in the Middle and Central Zones of the Rubber Test Strips

Crack Length (mm)	1% Antiozonant 4010NA #		0.5% Antiozonant 4010NA	
	Middle Zones	Central Zones	Middle Zones	Central Zones
0.05 - 0.10	37.5	37.5	40.0	42.5
0.10 - 0.15	45.0	48.0	48.0	53.0
0.15 - 0.20	48.0	≥57.5	≥57.5	≥57.5
0.20 - 0.40	≥57.5	≥57.5	≥57.5	≥57.5

Ozone concentrations given in ppb.

Adapted from Serrano et al. (1993).

11.2 TEXTILES AND FABRICS

Ozone can damage textiles and fabrics by methods similar to those associated with elastomers. Generally, synthetic fibers are less affected by O₃ than natural fibers; however, O₃ contribution to the degradation of textiles and fabrics is not considered significant (U.S. Environmental Protection Agency, 1996). A study reported by Bogaty et al. (1952) showed that O₃ affects moistened cloth more than dry cloth. Scoured cotton duck cloth and

commercially bleached cotton print cloth were exposed to 20 to 60 ppb O₃ for 1,200 h (50 days). The rate of deterioration was measured by the changes in cuprammonium fluidity values and the fabric breaking strength. At the end of the 1,200-h exposure, there was a 20% loss in breaking strength. Table 11-2 list the changes in cuprammonium fluidity values for both fabrics.

Table 11-2. Cuprammonium Fluidity of Moist Cotton Cloth Exposed to 20 to 60 ppb Ozone

	Duration of Exposure (h)	Cuprammonium Fluidity (rhes)
Duck Cloth	0	2.6
	200	2.8
	680	4.0
	960	6.8
	1200	9.5
Bleached Print Cloth	0	8.2
	200	8.7
	510	9.4
	650	12.0
	865	12.7
	1500	16.5

Adapted from Bogaty et al. (1952).

11.3 DYES, PIGMENTS, AND INKS

Ozone fading of textile dyes is diffusion-controlled; the rate of fading is controlled by the diffusion of the dye to the fiber surface. Many textile dyes react with O₃; however, the rate and severity of the O₃ attack is influenced by the chemical nature of the textile fiber and the manner in which the dye is applied. Ozone molecules break the aromatic ring portion of the dye molecule, oxidizing the dye (U.S. Environmental Protection Agency, 1996). In case of aromatic azo dyes, O₃ attacks the aromatic rings and electron rich nitrogen atoms (Matsui et al., 1988). Grosjean et al. (1987; 1988a,b) proposed a mechanism for reactions of O₃ with indigo, thioindigo, and dibromoindigo, alazarin, and curcumin dyes under dark conditions. Ozone attaches to the dye molecule at the unsaturated carbon = carbon bond. An O₃ adduct is formed (1,2,3-trioxolane), followed by scission of the carbon–carbon bond and the subsequent

formation of the corresponding Criegee biradical. A similar mechanism was proposed for the reaction of O₃ with triphenylmethane colorant Basic Violet 14. Ozone attacked Basic Violet 14 at the carbon-carbon unsaturated bond and at the carbon-nitrogen unsaturated bond under dark conditions. Other members of the group of triphenylmethane colorants with unsaturated carbon-carbon bonds also are expected to be subject to O₃ fading. Triphenylmethane colorants that are expected to be O₃-fugitive include the amino-substituted cationic dyes (Malachite Green, Brilliant Green, Crystal Violet, Pararosaniline Chloride, Methyl Green, and others) (Grosjean et al., 1989).

An indication that O₃ caused textile dye fading was first reported by Salvin and Walker (1955). The researchers found that the fading was primarily the result of the destruction of the blue dye molecule. Drapes made of acetate, Arnel, and Dacron and dyed with anthraquinone blue dye exhibited a decrease in shade that was not accompanied by the characteristic reddening caused by NO_x. Figures 11-1 and 11-2 demonstrate the effect of O₃ exposure on nylon 6 yarn colored with several blue dyes. Nylon samples inside the home were located on a wall away from sunlight. Outside nylon samples were placed on a covered patio or under the eaves of the house to minimize exposure to sunlight and rain. Ozone concentrations ranged from 2 to 5 ppb outside and 0 to 2 ppb inside. The percent change in dye color was determined monthly by extraction and analysis of the remaining dye or by instrumental measurement of the color change (Haylock and Rush, 1978).

11.4 ARTISTS' PIGMENTS

Several artists' pigments are sensitive to fading and oxidation by O₃ when exposed to concentrations found in urban areas (Shaver et al., 1983; Drisko et al., 1985; Whitmore et al., 1987; Whitmore and Cass, 1988; Grosjean et al., 1993). The organic pigments that are O₃ fugitive include alizarin red pigments containing lakes of the polycyclic aromatic compound 1,2-dihydroxyanthraquinone, blue-violet pigments containing substituted triphenylmethane lakes, indigo, and yellow coloring agents containing polyfunctional, polyunsaturated compounds such as curcumin (Grosjean et al., 1987). Because of the potential of O₃ to damage works of art, recommended limits on O₃ concentrations in museums, libraries, and archives are relatively low, ranging from 0.013 to 0.01 ppm.

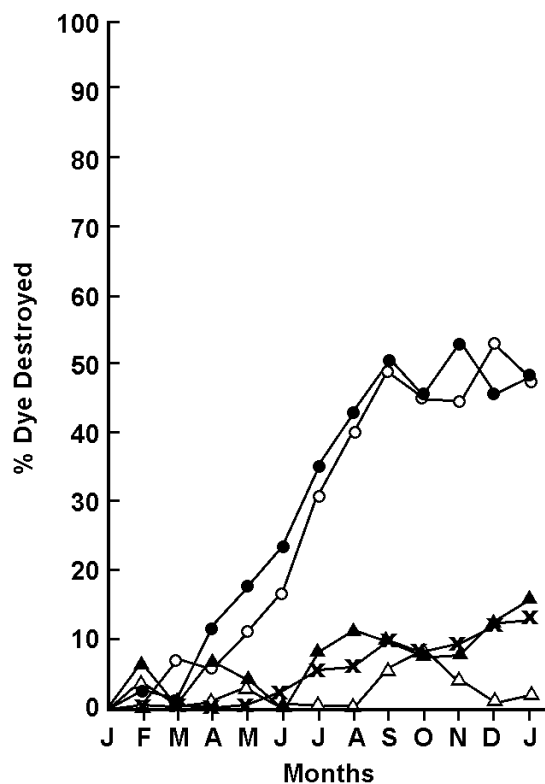


Figure 11-1. In-service fading of nylon 6 yarn inside house. ● = C.I. Disperse Blue 3; ○ = C.I. Basic Blue 22; ▲ = C.I. Acid Blue 27; x = C.I. Disperse Blue 56; △ = C.I. Acid Blue 232.

Source: Haylock and Rush (1978).

Experimental studies demonstrate a concentration \times time ($C \times T$) relationship between O_3 concentration, exposure time, and pigment fading. Cass et al. (1991) summarized some of the earlier research on the effects of O_3 on artists' pigments. In studies evaluating the effect of O_3 on organic and inorganic watercolors and traditional organic pigments, only the traditional organic pigments showed measurable fading from O_3 exposure. Of the inorganic pigments tested, only the arsenic sulfides showed O_3 -related changes. The pigments were exposed to 0.3 to 0.4 ppm O_3 for 3 mo in the absence of light, at 22 °C and 50% RH. The authors equated this exposure to a $C \times T$ of 6 to 8 years inside a Los Angeles museum with air conditioning but without a pollutant removal system.

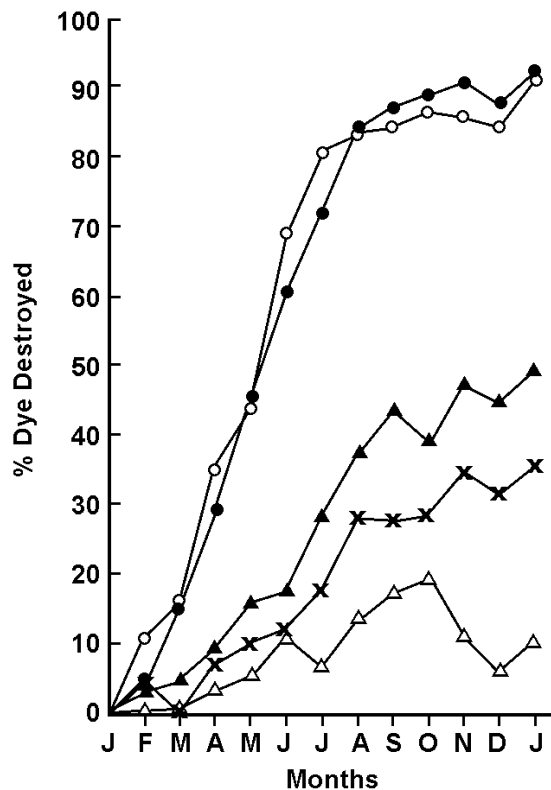


Figure 11-2. In-service fading of nylon 6 yarn outside house. ● = C.I. Disperse Blue 3; ○ = C.I. Basic Blue 22; ▲ = C.I. Acid Blue 27; x = C.I. Disperse Blue 56; △ = C.I. Acid Blue 232.

Source: Haylock and Rush (1978).

Whitmore and Cass (1988) studied the effect of O₃ on traditional Japanese colorants. Most of these compounds are insoluble metal salts that are stable in light and air. Suspensions or solutions of the colorants were airbrushed on hot-pressed watercolor paper or silk cloths. A sample of Japanese woodblock print also was included in the analysis. Samples were exposed to 0.4 ppm O₃ at 22 °C, 50% relative humidity, in the absence of light for 12 wk. Changes in reflectance spectra were used to evaluate the effect of O₃ exposure on colorant fading. Among the colorants tested on paper, curmin, indigo, madder lake, and lac lake were the most sensitive to O₃ exposure. Gamboge was relatively insensitive to O₃. The blue and green areas of the sample from the woodblock print was very reactive due to the indigo dye O₃ sensitivity. The

other colorants, red, yellow, and purple, showed very little sensitivity to O₃. The textiles dyes that reacted with O₃ were indigo, alone or in combination with several yellow dyes.

Ye et al. (2000) reported the rate of O₃ fading of traditional Chinese plant dyes. Twelve different colorants were applied to watercolor paper and silk and exposed to 0.4 ppm O₃ at 25 °C, at 50% RH, in the absence of light for 22 wks. Dye fading was greater when the colorant was applied to the watercolor paper compared to the silk cloth due to the darker initial depth of the shade, the greater saturation of the colorant throughout the cloth. Tumeric, gromwell, and violet on paper was particularly reactive. Tangerine peel was moderately reactive and sappan wood, dalbergia wood, Chinese gall, indigo, and Chinese yellow cork tree were slightly reactive to O₃. Black tea was not reactive to O₃. The colorants on silk samples showing color changes were gromwell, sappan wood, gardenia, tumeric, and violet. Figures 11-3 and 11-4 demonstrate the color change of the various colorants on watercolor paper and silk.

Artists' pigments also have exhibited fading when exposed to a mixture of photochemical oxidants. Grosjean et al. (1993) exposed 35 artists' pigments to a mixture of photochemical oxidants consisting of O₃, nitrogen dioxide (NO₂), and peroxyacetyl nitrate (PAN) for 12 wks. Weekly average photochemical concentrations were 200 ppb for O₃, 56 ± 12 to 99 ± 24 for NO₂, and 11 ± 3 to 18 ± 2 for PAN. All exposures were carried out at room temperature in the absence of light. To determine the effect of humidity on pigment fading, the relative humidity was increased from 46% after 8 weeks of exposure to 83% for a 2 week period and then returned to 46% for the remainder of the exposure.

Table 11-3 lists the artists' pigment and degree of fading. Eleven of the pigments tested exhibited negligible color change, 12 had small color changes, 3 had modest color changes, and 9 exhibited substantial color changes. Fading of Disperse Blue 3 and Reactive Blue 2 were likely the result of NO₂ exposure, and the fading of triphenylmethanes is consistent with exposure to nitric acid formed under high humidity conditions. Fading of the indigos was dominated by O₃ exposure and curcumin was faded by all of the photochemicals studied. Increasing the relative humidity resulted in a substantial color change for all of the pigments, with the exception of curcumin and indigo.

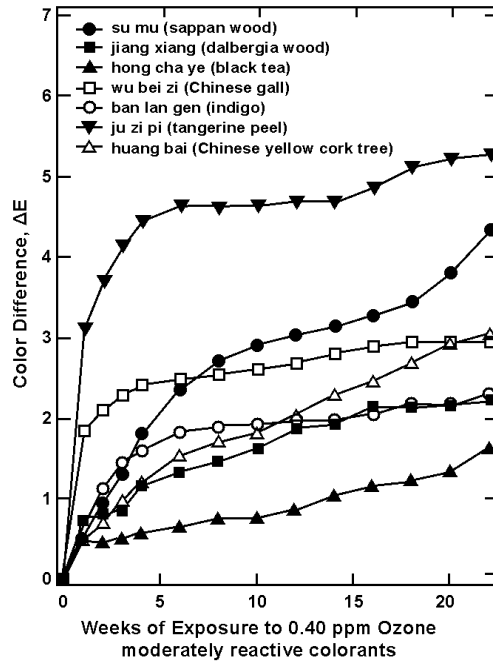
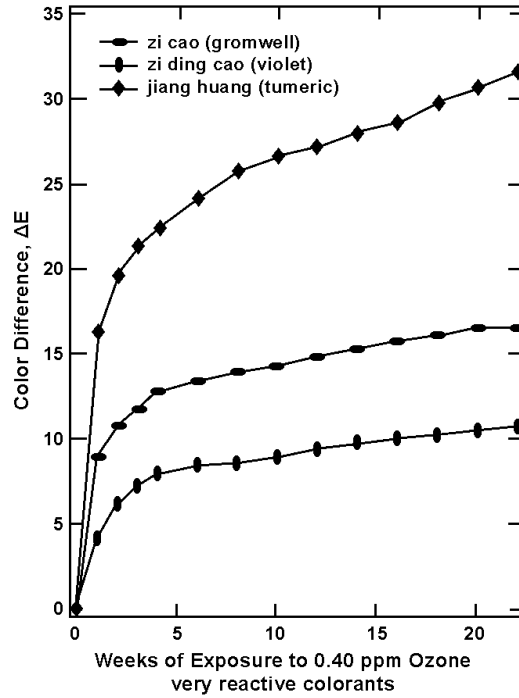


Figure 11-3. Observed color changes for natural colorant-on-paper systems during exposure to 0.40 ppm O₃ at 25 °C ± 1 °C, 50% RH, in the absence of light.

Source: Ye et al. (2000).

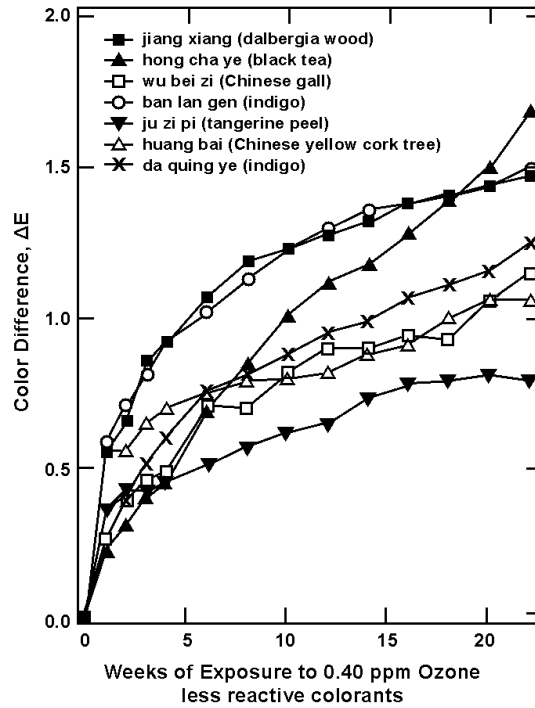
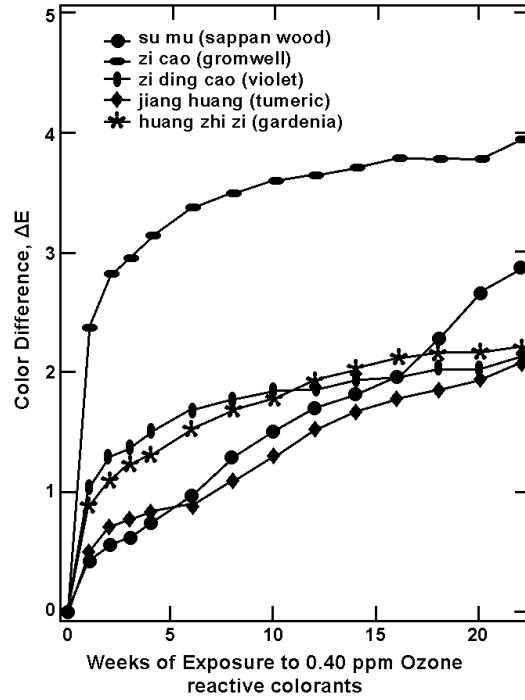


Figure 11-4. Observed color changes for natural colorant-on-site during exposure to 0.40 ppm O_3 at $25\text{ }^\circ\text{C} \pm 1\text{ }^\circ\text{C}$, 50% RH, in the absence of light.

Source: Ye et al. (2000).

Table 11-3. Color Change After 12 Weeks of Exposure to a Mixture of Photochemical Oxidants

Colorant*	Color Change (ΔE units)[†]	Chemical Functionality or Chemical Composition
Acid Red 37 (17045) [‡]	11.7 ± 0.5	Aminophenyl-substituted azo dye, sulfonate salt
Acid Yellow 65 [‡]	1.8 ± 0.5	Nitro- and phenyl-substituted azo dye, sulfonate salt
Alizarin Carmine	1.8 ± 0.2	Alizarin lake
Alizarin Crimson (Pigment Red 83)	1.4 ± 0.2	Alizarin lake
Aurora Yellow (77199)	0.5 ± 0.1	Cadmium sulfide
Basic Fuschin (42510) [‡]	33.4 ± 3.0	Amino-substituted triphenylmethane
Brilliant Green (42040) [‡]	20.6 ± 2.1	Amino-substituted triphenylmethane
Brown Madder	1.7 ± 0.1	Alizarin lake
Cadmium Yellow (77199)	0.4 ± 0.1	Cadmium sulfide
Carmine	1.8 ± 0.2	Lake of cochineal (substituted anthraquinone)
Chrome Yellow (77600) [‡]	1.7 ± 1.2	Lead chromate
Copper phthalocyanine (Pigment Blue 15)	1.0 ± 0.1	Copper phthalocyanine
Crimson Lake	3.5 ± 0.3	Alizarin lake
Curcumin (Natural Yellow 3)	15.2 ± 2.6	1,7 bis (4-hydroxy-3-methoxyphenyl)-1,6-heptadiene-3,5-dione
Disperse Blue 3	10.8 ± 0.1	Amino-substituted anthraquinone
French Ultramarine Blue	0.8 ± 0.3	
Gamboge (Natural Yellow 24)	0.4 ± 0.1	Gambogic acid
Hooker's Green Light	1.5 ± 0.4	Chlorinated copper phthalocyanine plus ferrous beta naphthol derivative
Indigo (a formulation)	1.1 ± 0.1	Alizarin lake plus lampblack plus copper phthalocyanine
Indigo carmine [‡]	14.0 ± 1.9	5,5-indigo disulfonic acid, sodium salt
Indigo (73000) [‡]	64.1 ± 4.5	
Mauve	3.6 ± 0.5	Lake of triphenyl methane (basic fuschin) plus copper phthalocyanine
New Gamboge	0.9 ± 0.1	Arylamide yellow (CI 11680) plus toluidine red

Table 11-3 (cont'd). Color Change After 12 Weeks of Exposure to a Mixture of Photochemical Oxidants

Colorant*	Color Change (ΔE units)[†]	Chemical Functionality or Chemical Composition
Pararosaniline base (42500) [‡]	25.6 ± 4.7	Amino-substituted triphenylmethane
Payne's Grey	1.0 ± 0.1	Alizarin lake plus prussian blue plus lampblack plus ultramarine blue
Permanent Magenta	1.1 ± 0.1	Quinacridone
Permanent Rose	2.0 ± 0.1	Quinacridone
Prussian Blue	0.7 ± 0.2 1.6 ± 0.3	Ferric ferrocyanide
Prussian Green	0.9 ± 0.2	Arylamide yellow plus prussian blue
Purple Lake	2.3 ± 0.3	Alizarin lake
Reactive Blue 2 (61211) [‡]	14.4 ± 1.1	Amino-substituted anthraquinone, sulfonate salt
Rose Carthane (12467)	0.8 ± 0.2	Arylamide (Pigment Red 10) plus xanthene (Pigment Red 90)
Rose Doré	2.0 ± 0.2	Quinacridone plus Yellow 3
Thioindigo Violet (73312) [‡]	1.9 ± 1.2	Chlorinated thioindigo
Winsor Yellow (11680)	0.5 ± 0.2	Arylamide yellow

* On watercolor paper unless otherwise indicated. Color Index (CI) names or CI numbers are given in parentheses.

[†] Mean ± one standard deviation for triplicate samples calculated from the parameters L^* , a^* , and b^* measured with the color analyzer.

[‡] On Whatman 41 paper.

Source: Grosjean et al. (1993).

11.5 SURFACE COATINGS

Ozone will act to erode some surface coatings (paints, varnishes, and lacquers). However, many of the available studies on O₃ degradation of surface coatings do not separate the effects of O₃ from those of other pollutants or environmental factors such as weather, humidity, and temperature. Campbell et al. (1974) attempted to demonstrate an O₃-related effect on oil house paint, acrylic latex coating, alkyd industrial maintenance coating, urea alkyd coil coating, and nitrocellulose/acrylic automotive paint. Painted test panels were exposed to 100 and

1,000 ppb O₃ in a xenon arc accelerated weathering chamber for up to 1,000 h. Using weight loss as a measure of O₃-induced erosion, the researchers concluded that all of the paints tested suffered degradation in the presence of O₃ and that the automotive finish suffered the most O₃-induced degradation. When O₃ degradation was measured using scanning electron microscopy, the oil house paint and latex coating samples showed erosion above that seen with clean air but only at the highest exposure level. No effects were noted for the automotive paint. The other painted surfaces were not evaluated.

Spence et al. (1975) studied the effect of air pollutants and relative humidity on oil based house paint, acrylic latex house paint, acrylic coil coating, and vinyl coil coating under laboratory conditions. Test panels were exposed in weathering chambers equipped with a xenon arc light for simulating sunlight to low and high levels of O₃ (0.08 and 0.5 ppm), sulfur dioxide (0.03 and 0.5 ppm), and nitrogen dioxide (0.05 and 0.5 ppm) and relative humidity (50 and 90%). Samples were exposed for a total of 1000 h. The exposure cycle consisted of 20 min of dew and 20 min of light. The effects of the exposure on the painted surfaces were measured by weight loss and loss in film thickness. The acrylic coil coating had the lowest erosion rate of the surface coatings tested. However, O₃ was the only pollutant that had a significant effect on the surface erosion. Sulfur dioxide and relative humidity were significant factors in the erosion of oil base house paints and vinyl coil coating. The findings for acrylic latex house paint were not reported.

11.6 CONCLUSIONS

Ozone and other photochemical oxidants react with many economically important man-made materials, decreasing their useful life and aesthetic appearance. Some materials known to be damaged by O₃ include elastomers, fibers and dyes, and paints. Most studies have been on single compounds rather than complex materials.

The elastomeric compounds, natural rubber and synthetic polymers and copolymers of butadiene, isoprene, and styrene, are particularly susceptible to even low concentrations of O₃. Ozone damages these compounds by breaking the molecular chain at the carbon-carbon double bond; a chain of three oxygen atoms is added directly across the double bond. The change in structure promotes the characteristic cracking of stressed/stretched rubber called “weathering.”

Tensile strain produces cracks on the surface of the rubber that increase in number with increased stress/stretching.

The rate of crack growth is dependent on the degree of stress, the type of rubber compound, O₃ concentration, time of exposure, O₃ velocity, and temperature. After initial cracking, there is further O₃ penetration, resulting in additional cracking and, eventually, mechanical weakening or stress relaxation.

Ozone can damage textiles and fabrics by methods similar to those associated with elastomers. Generally, synthetic fibers are less affected by O₃ than natural fibers; however, O₃ contribution to the degradation of textiles and fabrics is not considered significant .

Ozone fading of textile dyes is a diffusion-controlled process; the rate of fading is controlled by the diffusion of the dye to the fiber surface. Many textile dyes react with O₃. The rate and severity of the O₃ attack is influenced by the chemical nature of the textile fiber and the manner in which the dye is applied.

Several artists' pigments are also sensitive to fading and oxidation by O₃ when exposed to concentrations found in urban areas.

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